

## Methyl Homologues of Methyl Jasmonate and Methyl Dihydrojasmonate (*Hedione*<sup>®</sup>) from Sorbyl Alcohol

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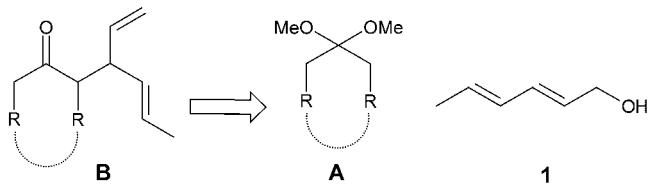
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Dedicated to *Günther Ohloff* on the occasion of his 80th birthday

Treatment of cycloalkanone dimethyl acetals **3–6** with sorbyl alcohol (= (2E,4E)-hexa-2,4-dien-1-ol; **1**) in the presence of acids afforded the novel cycloalkenones **8**, **9**, **11**, and **13** via a domino reaction (*Claisen* rearrangement with intramolecular ene reaction and *retro-ene* reaction). Cyclopentenone **8** was readily transformed into **14** and **15**, methyl homologues of racemic methyl jasmonate (**16**) and methyl dihydrojasmonate (= *Hedione*<sup>®</sup>; **17**), respectively. The organoleptic properties of **14** and **15** are also discussed.

**1. Introduction.** – As part of a screening program comprising the synthesis of new odorant compounds, we decided to study the acid-catalyzed reaction of sorbyl alcohol<sup>1)</sup> (**1**) with dimethyl acetals of general structure **A**, thus offering potential access to dienones **B** via *Claisen* rearrangement<sup>2)</sup> (*Scheme 1*).

*Scheme 1*



**2. Results and Discussion.** – 2.1. *Acid-Catalyzed Reaction of **1** with **2–6**.* Dimethyl acetals **2–6** (3 mol-equiv.), readily available from corresponding ketones by a standard acetalization procedure [1], were heated individually with **1** (1 mol-equiv.) in the presence of a catalytic amount of malonic acid (0.01 mol-equiv.)<sup>3)</sup> in an *Inox* autoclave at 200° during 48 h. Cooling of resulting product mixtures to room temperature was followed by bulb-to-bulb distillation *in vacuo*, whereby the compounds **7–13** (see

<sup>1)</sup> Sorbyl alcohol is the trivial name for (2E,4E)-hexa-2,4-dien-1-ol.

<sup>2)</sup> For the same strategy applied to prop-2-en-1-ol, see [1].

<sup>3)</sup> Other *Brønsted* acids such as, *e.g.*, citric acid, TsOH, and H<sub>2</sub>SO<sub>4</sub> could also be employed as catalyst.

*Table*) were isolated and spectroscopically characterized<sup>4</sup>). For *Entries 4* and *5*, preparative GC was necessary to separate the mixtures **10/11** and **12/13**.

Table. Acid-Catalyzed Reaction of Sorbyl Alcohol (**1**) with Dimethyl Acetals **2–6**<sup>a</sup>)

Entry	Substrate	Product(s)	Yield [%] <sup>b</sup> )
1			47
2			73
3			68
4			83
5			68

<sup>a</sup>) Reaction conditions: substrate (3 mol-equiv.), **1** (1 mol-equiv.), malonic acid (0.01 mol-equiv.), 200°, 48 h.

<sup>b</sup>) Isolated yield based on **1**. <sup>c</sup>) 2:1 Diastereoisomer mixture.

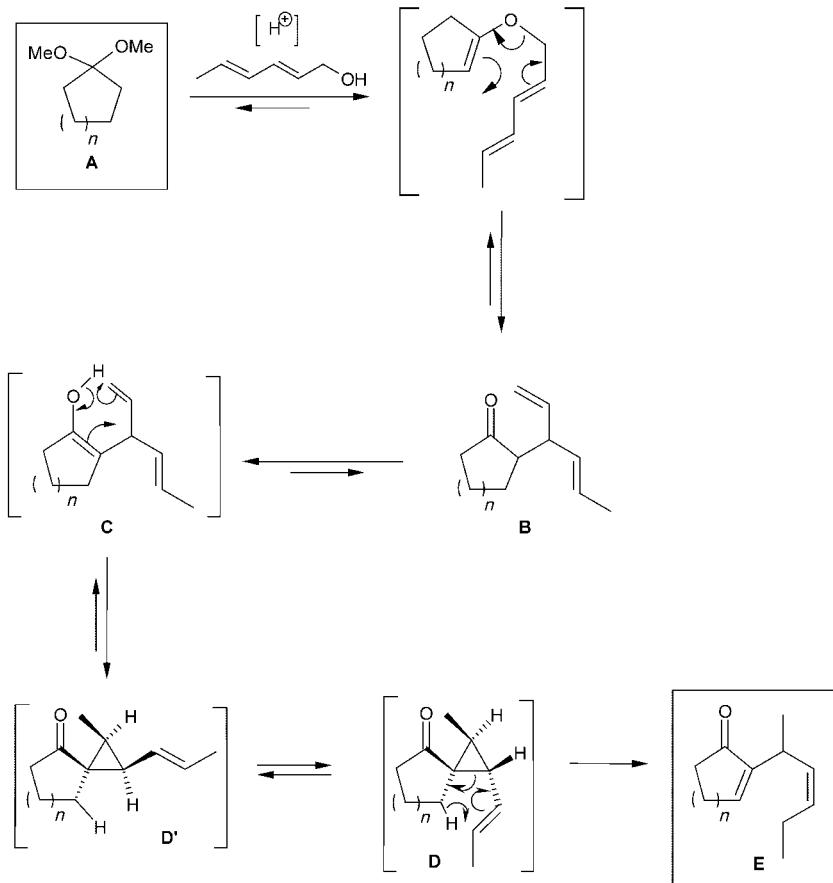
As can be seen from the *Table*, the acyclic substrate **2** gave only the expected *Claisen* rearrangement product **7**, as a 2:1 diastereoisomer mixture in 47% yield (*Entry 1*)<sup>5</sup>). In contrast, the cyclic substrates **3–6** afforded cycloalkenone **E**, either

<sup>4</sup>) All chiral compounds reported in this work are racemic and were fully characterized, see *Exper. Part.*

<sup>5</sup>) Despite this only moderate yield, no other compound (>5%) was detected by GC in the crude product mixture.

exclusively or as one of the principal products. Thus, **3** and **4** afforded **8** and **9** in 73% and 68% yield, respectively (*Entries 2 and 3*), whereas **5** and **6** afforded mixtures of the dienone **B** and the corresponding cycloalkenone **E**, *i.e.*, **10/11** 1:1 and **12/13** 2.4:1 in 83% and 68% yield, respectively (*Entries 4 and 5*). As shown in *Scheme 2*, the formation of the (*Z*)-configured **E** can be rationalized by invoking a domino reaction [2] in which the enol tautomer **C** of the initially formed *Claisen*-rearrangement product **B** undergoes an intramolecular ene reaction followed by a 1,5-H transfer (*retro-ene* reaction) of the intermediate cyclopropyl ketone **D'**). It is important to note that the stereoselectivity of the conversion of **C** to **D** and/or **D'** has no influence on the formation of **E**, due to the known epimerization (**D**  $\rightleftharpoons$  **D'**) of these systems [4]. The difference in behavior between **2** and **3–6** is almost certainly due to the (*Z*)-configuration of **C**, which favors the transition state of the subsequent ene reaction. However, in contrast to *Entries 2 and 3*, the requisite transition-state geometry in

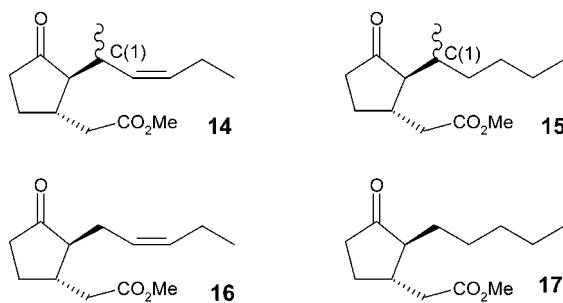
*Scheme 2. Mechanistic Proposal for the Formation of E from A (via B)*



<sup>6)</sup> For a literature precedent of an analogous domino reaction, see [3].

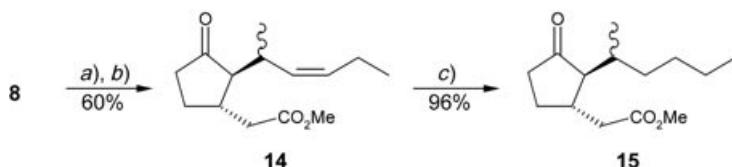
*Entries 4 and 5* is progressively harder to attain, due to the higher conformational strain of the seven- and eight-membered rings.

2.2. *Synthesis of 14 and 15 from 8.* With **8** in hand, we now had ready access to **14** and **15**, methyl homologues of ( $\pm$ )-methyl jasmonate (**16**)<sup>7</sup>) and ( $\pm$ )-methyl dihydrojasmonate (= *Hedione*)<sup>8</sup>; **17**) both well-known jasmine-type odorants [7].



Accordingly, treatment of **8** with dimethyl malonate under basic conditions followed by decarboxylative saponification [8] and subsequent esterification afforded **14** (diastereoisomer mixture at C(1)) in 60% overall yield. Catalytic hydrogenation then furnished **15** (diastereoisomer mixture at C(1)) in 96% yield, in a straightforward manner (*Scheme 3*).

*Scheme 3. Synthesis of 14 and 15 from 8.* Compounds **14** and **15** contained ca. 5–10% of the corresponding *cis*-diastereoisomer.



a) NaOMe, dimethyl malonate,  $-5^\circ$ , 5 d. b) NaOH, MeOH, then  $H_2SO_4$ , MeOH, reflux. c) 10% Pd/C, MeOH,  $H_2$ .

2.3. *Organoleptic Evaluation of 14 and 15.* The odor of **14** has a floral, jasminic note, is more indolic than *Hedione*®, and has a fruity-raspberry undertone. Compound **15** smells floral, *Hedione*®-like, but with a citrus connotation. It is finer, not as powerful, but more long-lasting than *Hedione*®.

We thank Dr. Pierre-Alain Blanc for the organoleptic evaluation, Dr. Roger L. Snowden for help in the preparation of the manuscript and Walter Thommen for the discussion of the NMR spectra.

### Experimental Part

*General.* Flash chromatography (FC): Silica gel 60, 35–70  $\mu$ m. Anal. GC: *Varian STAR 3400*; He as carrier gas; fused-silica capillary columns *SPB-1* and *Supelcowax*®, each 30 m  $\times$  0.25 mm i.d. with 0.25- $\mu$ m film. Prep. GC: *JAS-2000* system, column *HP-Wax*, 10 m  $\times$  0.1  $\mu$ m, 50–250°, 30°/min.  $^1H$ - and  $^{13}C$ -NMR Spectra: *Bruker*-

<sup>7)</sup> For the isolation of (–)-methyl jasmonate from jasmine oil, see [5]; for synthetic work, see [6].

<sup>8)</sup> Registered trade name of *Firmenich SA*.

*DPX-400 or AV-500* spectrometers;  $\delta$  in ppm downfield from SiMe<sub>4</sub>,  $J$  in Hz. GC-MS: *Hewlett-Packard-5890* or -6890 system equipped with a capillary column (30 m  $\times$  0.25  $\mu\text{m}$  i.d.), coupled with a *Hewlett-Packard MSD-5972* or -5973 quadrupole mass spectrometer; electron energy *ca.* 70 eV; in *m/z* (rel. int. in % of the base peak).

1. *Dimethyl Acetals 2–6: General Procedure.* To a stirred soln. of the ketone (2 mol), trimethyl orthoformate (5 mol), and MeOH (700 ml), TsOH (500 mg) was added at r.t. After 4 days, the mixture was poured onto ice, worked up with Et<sub>2</sub>O, and washed with sat. NaHCO<sub>3</sub> soln. and brine, then distilled.

*3,3-Dimethoxypentane (2)* [9]: 115 g (63%). B.p. 120–124°. <sup>1</sup>H-NMR: 0.82 (*t*,  $J$  = 7.3, 6 H); 1.60 (*q*,  $J$  = 7.3, 4 H); 3.15 (*s*, 6 H). MS: 144 (<0.5,  $M^+$ ), 103 (100), 101 (60), 69 (11), 57 (43), 45 (28).

*1,1-Dimethoxycyclopentane (3)* [10]: 172 g (66%). B.p. 138–140°. <sup>1</sup>H-NMR: 1.6–1.8 (*m*, 8 H); 3.2 (*s*, 6 H). MS: 130 (1,  $M^+$ ), 101 (100), 99 (48), 67 (42), 57 (22), 41 (10).

*1,1-Dimethoxycyclohexane (4)* [11]: 199 g (69.4%). B.p. 75–78°/40 mbar. <sup>1</sup>H-NMR: 1.35–1.7 (*m*, 10 H); 3.2 (*s*, 6 H). MS: 144 (5,  $M^+$ ), 113 (45), 101 (100), 81 (31), 55 (13), 43 (11).

*1,1-Dimethoxycycloheptane (5)* [12]: 255 g (74%). B.p. 85–89°/36 mbar. <sup>1</sup>H-NMR: 1.55 (br. *s*, 8 H); 1.78 (*m*, 4 H); 3.16 (*s*, 6 H). <sup>13</sup>C-NMR: 21.7 (*t*); 29.2 (*t*); 36.1 (*t*); 47.8 (*q*); 104.4 (*s*). MS: 158 (8,  $M^+$ ), 127 (46), 115 (12), 101 (100), 95 (20), 55 (19).

*1,1-Dimethoxycyclooctane (6)* [13]: 254 g (74%). B.p. 85–89°/36 mbar. <sup>1</sup>H-NMR: 1.55 (*s*, 10 H); 1.76 (br. *s*, 4 H); 3.14 (*s*, 6 H). <sup>13</sup>C-NMR: 21.3 (*t*); 24.6 (*t*); 28.2 (*t*); 30.3 (*t*); 47.7 (*q*); 103.8 (*s*). MS: 172 (6,  $M^+$ ), 141 (48), 101 (100), 88 (44), 67 (26), 55 (25), 41 (22).

2. *Acid-Catalyzed Reaction of Sorbyl Alcohol (1) with the Dimethyl Acetals 2–6: General Procedure.* A mixture of **1** (1 mol-equiv.), acetal (3 mol-equiv.), and malonic acid (100 mg) was heated with stirring in an autoclave (*Bergauf*; 100 ml, 100 mbar) to 200° during 48 h. The mixture was then bulb-to-bulb distilled under vacuum.

( $\pm$ )-(E)-5-Ethenyl-4-methyloct-6-en-3-one (**7**): 4 g (47%). B.p. 120°/14 mbar. <sup>1</sup>H-NMR: 0.98–1.04 (*m*, 6 H); 1.58–1.69 (*m*, 3 H); 2.33–2.47 (*m*, 2 H); 2.55 (*quint.*,  $J$  = 8.2, 1 H); 2.9 (*quint.*,  $J$  = 8.2, 1 H); 4.94–5.07 (*m*, 2 H); 5.22–5.78 (*m*, 3 H). <sup>13</sup>C-NMR: 7.49, 7.90 (*2q*); 14.49, 14.54 (*2q*); 18.00 (*q*); 35.59, 35.72 (*2t*); 49.95, 50.21 (*2d*); 50.30 (*d*); 115.07, 115.08 (*2t*); 126.49, 127.30 (*2d*); 130.82, 131.24 (*2d*); 138.58, 139.17 (*2d*); 214.40 (*s*). MS: 166 (2,  $M^+$ ), 151 (9), 137 (30), 109 (11), 81 (100), 67 (18), 57 (43), 41 (27), 29 (23).

( $\pm$ -2-[*(2Z)*-1-Methylpent-2-enyl]cyclopent-2-en-1-one (**8**): 6 g (73%). B.p. 98°/16 mbar. <sup>1</sup>H-NMR: 0.95 (*t*,  $J$  = 7.5, 3 H); 1.14 (*d*,  $J$  = 7.1, 3 H); 1.66–2.659 (*m*, 6 H); 3.5 (*quint.*,  $J$  = 7.1, 1 H); 5.31 (*dd*,  $J$  = 10, 10, 1 H); 5.37 (*dt*,  $J$  = 10, 7, 1 H); 7.3 (*m*, 1 H). <sup>13</sup>C-NMR: 14.3 (*q*); 20.4 (*q*); 20.7 (*t*); 26.4 (*t*); 28.3 (*d*); 34.9 (*t*); 131.6 (*d*); 131.8 (*d*); 150.4 (*s*); 156.2 (*d*); 209.0 (*s*).

( $\pm$ -2-[*(2Z)*-1-Methylpent-2-enyl]cyclohex-2-en-1-one (**9**): 8.5 g (68%). B.p. 108°/13 mbar. <sup>1</sup>H-NMR: 0.94 (*t*,  $J$  = 7.5, 3 H); 1.07 (*d*,  $J$  = 6.8, 3 H); 1.90–2.45 (*m*, 8 H); 3.7 (*quint.*,  $J$  = 7.5, 1 H); 5.25 (*dd*,  $J$  = 10, 10, 1 H); 5.32 (*dt*,  $J$  = 6.8, 10, 1 H); 6.73 (*m*, 1 H). <sup>13</sup>C-NMR: 14.3 (*q*); 20.8 (*t*); 21.1 (*q*); 22.9 (*t*); 26.1 (*t*); 30.3 (*d*); 131.1 (*d*); 133.0 (*d*); 143.5 (*d*); 144.2 (*s*); 198.6 (*s*). MS: 178 (16,  $M^+$ ), 163 (11), 149 (100), 135 (9), 121 (12), 107 (10), 93 (28), 79 (22), 67 (12), 55 (37), 41 (20).

( $\pm$ -2-[*(2E)*-1-Ethenylbut-2-enyl]cycloheptanone (**10**) and ( $\pm$ -2-[*(2Z)*-1-Methylpent-2-enyl]cyclohept-2-en-1-one (**11**): 1.60 g (83%). B.p. 130/13 mbar. Separation of the isomers was effected by prep. GC.

Data of **10**<sup>9</sup>: <sup>1</sup>H-NMR: 4.95–5.05 (*m*, 2 H); 5.30–5.52 (*m*, 2 H); 5.65–5.80 (1 H). <sup>13</sup>C-NMR: major isomer: 18.0 (*q*); 24.8 (*t*); 28.3 (*t*); 28.4 (*t*); 29.5 (*t*); 43.5 (*t*); 49.3 (*d*); 56.8 (*d*); 114.8 (*t*); 127.4 (*d*); 130.2 (*d*); 139.8 (*d*); 215.7 (*s*); minor isomer: 18.0 (*q*); 24.8 (*t*); 28.3 (*t*); 28.4 (*t*); 29.5 (*t*); 43.5 (*t*); 49.5 (*d*); 56.8 (*d*); 115.8 (*t*); 126.3 (*d*); 131.5 (*d*); 138.7 (*d*); 215.5 (*s*). MS: major isomer: 192 (8,  $M^+$ ); 177 (5), 163 (9), 149 (63), 81 (100), 55 (24), 41 (36); minor isomer: 192 (8,  $M^+$ ), 177 (5), 193 (9), 149 (62), 81 (100), 55 (22), 41 (33).

Data of **11**: <sup>1</sup>H-NMR: 0.93 (*t*,  $J$  = 7.3, 3 H); 1.06 (*d*,  $J$  = 7.0, 3 H); 1.65–1.78 (*m*, 4 H); 2.05 (*m*, 2 H); 2.35 (*m*, 2 H); 2.54 (*m*, 2 H); 3.7 (*m*, 1 H); 5.2 (*dd*,  $J$  = 9.5, 10.5, 1 H); 5.32 (*dt*,  $J$  = 7.5, 10.5, 1 H); 6.36 (*t*,  $J$  = 6.4, 1 H). <sup>13</sup>C-NMR: 14.3 (*q*); 20.7 (*t*); 21.0 (*q*); 21.6 (*t*); 24.9 (*t*); 27.3 (*t*); 33.3 (*d*); 42.9 (*t*); 131.4 (*d*); 133.1 (*d*); 137.6 (*d*); 206.0 (*s*). MS: 192 (<0.5,  $M^+$ ), 178 (1), 177 (5), 163 (100), 135 (19), 121 (15), 107 (19), 93 (35), 79 (37), 67 (25), 55 (32), 41 (35).

( $\pm$ -2-[*(2E)*-1-Ethenylbut-2-enyl]cyclooctanone (**12**) and ( $\pm$ -2-[*(2Z)*-1-Methylpent-2-enyl]cyclooct-2-en-1-one (**13**): 1.40 g (68%). B.p. 145°/13 mbar. Separation of the isomers was effected by prep. GC.

Data of **12**<sup>9</sup>: Major isomer: <sup>1</sup>H-NMR: 5.00–5.07 (*m*, 2 H); 5.23–5.30 (*m*, 1 H); 5.32–5.42 (*m*, 1 H); 5.6–5.68 (*m*, 1 H). <sup>13</sup>C-NMR: 18.1 (*q*); 24.8 (*t*); 25.0 (*t*); 25.8 (*t*); 27.8 (*t*); 31.3 (*t*); 43.8 (*t*); 50.5 (*d*); 54.2 (*d*); 115.8 (*t*); 126.4 (*d*); 131.2 (*d*); 139.2 (*d*); 219.5 (*s*). MS: 206 (5,  $M^+$ ), 191 (2), 177 (5), 163 (10), 149 (29), 135 (5), 121 (5),

<sup>9</sup>) 3:1-Mixture of diastereoisomers.

107 (10), 93 (11), 81 (100), 67 (11), 55 (23), 41 (30). Minor isomer:  $^1\text{H-NMR}$ : 4.91–4.97 (*m*, 2 H); 5.22–5.30 (*m*, 1 H); 5.43–5.52 (*m*, 1 H); 5.60–5.68 (*m*, 1 H).  $^{13}\text{C-NMR}$ : 18.0 (*q*); 24.8 (*t*); 25.0 (*t*); 25.9 (*t*); 27.8 (*t*); 31.3 (*t*); 43.8 (*t*); 50.5 (*d*); 54.5 (*d*); 114.9 (*t*); 127.3 (*d*); 131 (*d*); 139.4 (*d*); 219.6 (*s*). MS: 206 (7,  $M^+$ ), 191 (2), 177 (4), 163 (12), 149 (33), 135 (5), 121 (4), 107 (11), 98 (11), 81 (100), 55 (6), 41 (33).

*Data of 13:*  $^1\text{H-NMR}$ : 0.93 (*t*,  $J = 7, 3$  H); 1.09 (*d*,  $J = 7, 3$  H); 5.12 (br. *dd*,  $J = 11, 11, 1$  H); 5.36 (*dt*,  $J = 7, 11$ , 1 H); 5.75 (*dt*,  $J = 1.3, 7.4, 1$  H).  $^{13}\text{C-NMR}$ : 14.3 (*q*); 20.0 (*q*); 20.4 (*t*); 22.1 (*t*); 22.8 (*t*); 28.9 (*t*); 29.4 (*t*); 36.3 (*d*); 43.7 (*t*); 129.0 (*d*); 131.8 (*d*); 132.0 (*d*); 142.6 (*s*); 213.6 (*s*). MS: 206 (7,  $M^+$ ), 191 (8), 177 (100), 163 (28), 149 (20), 135 (19), 121 (22), 107 (32), 93 (53), 81 (58), 67 (42), 55 (71), 41 (67).

( $\pm$ )-*Methyl 2-[2Z)-1-Methylpent-2-enyl]-3-oxocyclopentaneacetate* (**14**). To a cooled ( $-15^\circ$ ) mixture of dimethyl malonate (50 g, 0.4 mol) and Na (0.17 g, 7 mmol) was added a soln. of **8** (10 g, 61 mmol), dimethyl malonate (7 g), and MeOH (13 ml) with stirring under  $\text{N}_2$ . After complete addition, the mixture was stirred at  $-5^\circ$  for 5 days, then neutralized with AcOH (4 g, 20 mmol), diluted with  $\text{Et}_2\text{O}$ , washed with  $\text{H}_2\text{O}$ , and distilled at 110–120°/0.4 mbar. The distillate (17.8 g, 98%) was then refluxed in  $\text{H}_2\text{O}$  (200 ml) in the presence of KOH (8 g, 143 mmol) for 5 h. After cooling, the mixture was washed with pentane ( $2 \times 100$  ml), the aq. soln. acidified with ice/ $\text{H}_2\text{SO}_4$  and extracted with  $\text{Et}_2\text{O}$ , the  $\text{Et}_2\text{O}$  phase evaporated, and the residue esterified with MeOH (150 ml) and conc.  $\text{H}_2\text{SO}_4$  soln. (5 ml) at r.t. during 5 days. Then the MeOH was distilled off, the remaining mixture diluted with  $\text{Et}_2\text{O}$ , and the  $\text{Et}_2\text{O}$  soln. washed with brine and distilled: **14** (8.6 g, 60%) as a diastereoisomer mixture. B.p. 170°/0.6 mbar. MS: 238 (25,  $M^+$ ), 220 (32), 165 (54), 149 (23), 147 (33), 135 (24), 123 (26), 109 (47), 107 (26), 95 (28), 93 (33), 83 (100), 82 (41), 67 (29), 55 (80), 41 (31); the MS of the isomers are identical.

( $\pm$ )-*Methyl 2-(1-Methylpentyl)-3-oxocyclopentaneacetate* (**15**). A mixture of **14** (3 g, 12 mmol), 10% Pd/C (50 mg), and MeOH (50 ml) was hydrogenated at r.t. and 1 atm and then filtered. The filtrate was evaporated and the residue purified by FC (heptane/AcOEt 95 : 5): **15** (1.4 g, 46%) as an isomer mixture. B.p. 150°/0.6 mbar.  $^{13}\text{C-NMR}$ : isomer A: 14.1 (*q*); 16.6 (*q*); 22.8 (*t*); 27.4 (*t*); 30.1 (*t*); 32.7 (*d*); 34.2 (*t*); 36.2 (*d*); 39.0 (*t*); 39.5 (*t*); 51.6 (*q*); 58.5 (*d*); 172.6 (*s*); 220.0 (*s*); isomer B: 14.1 (*q*); 16.9 (*q*); 22.8 (*t*); 27.4 (*t*); 30.1 (*t*); 32.9 (*d*); 34.2 (*t*); 34.8 (*d*); 38.6 (*t*); 39.9 (*t*); 51.6 (*q*); 58.5 (*d*); 172.7 (*s*); 219.3 (*s*). MS: 240 (2,  $M^+$ ), 209 (3), 183 (6), 167 (10), 156 (42), 109 (7), 96 (7), 83 (100), 55 (19), 41 (20); the MS of both isomers are identical.

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