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Synthesis of enantiomerically pure *cis*-2,4-disubstituted piperidines: extension of chiral homoenolate alkylations toward the preparation of nitrogen heterocycles

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Abstract

Enantiomerically pure cis-2,4-disubstituted piperidines were synthesized from chiral thiolactam 2 in six steps via a highly diastereoselective homoenolate alkylation which set the stereochemistry at the C4 carbon. In addition, two cleavage methods were used to cleave the lactam to the desired piperidines with a high level of diastereoselection at the newly created C2 stereocenter. © 1999 Published by Elsevier Science Ltd. All rights reserved.

Chiral piperidines occupy a unique place in many naturally occurring and biologically important compounds, and methods for their asymmetric syntheses have been reviewed recently.^{1,2} Our program utilizing chiral, non-racemic bicyclic lactams for the preparation of nitrogen-containing heterocycles, bicyclic lactams 1 and 2, have served as key sources for the preparation of 2-substituted piperidines [(-)-coniine 3, and cis-2,6-disubstituted piperidines including (+)-pinidinone³ 4].



We have recently described the use of homoenolates derived from chiral bicyclic lactams as useful precursors for the preparation of chiral, nonracemic 5-substituted cyclohexenones and various carbocycles.⁴ As an extension of this methodology, we felt that a variation of reductive cleavage conditions of the chiral auxiliary could produce 2,4-disubstituted piperidones and piperidines. Since very few general methods exist to install functionality at the 4-position of piperidine ring systems,^{5,6} we felt that this was a worthwhile endeavor.

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Treatment of thiolactam 2^7 with Meerwein's reagent⁸ followed by displacement with KCN in the presence of CuI and I₂ afforded α -cyanoenamine **5** in good overall yield (Scheme 1).⁹ Further treatment of the α -cyanoenamine with 1.2 equiv. of LiTMP in the presence of HMPA followed by addition of various electrophiles afforded the 4-substituted products **6** as *single diastereomers*. Aqueous acidic hydrolysis of the crude alkylated products in THF returned the lactam **6** in good overall yields over the two steps (Table 1). The absolute stereochemistry of the newly created 4-carbon of lactam **6a** was determined by conversion to a known 5-substituted cyclohexenone¹⁰ and the stereochemistry of the remaining examples (**6b–d**) was assigned by analogy.¹¹



Scheme 1.

Efforts were subsequently focused upon the simultaneous reductive cleavage of the ring C–O bond as well as the carbonyl group to reach the piperidine nucleus as reported earlier.¹² However, the use of bulky reducing agents (DibAl and RedAl) failed to reduce the ring C–O bond, while treatment with more reactive reducing agents (AlH₃ and BH₃) afforded poor to modest diastereomeric ratios (1:1–6:1) at the C6 stereocenter of the newly created piperidine ring. An alternative reductive strategy was sought to try to circumvent the poor diastereoselection observed at the C6 stereocenter.

Treatment of **6a–d** with RedAl (10 equiv.) in THF at rt afforded oxazolidines **7a–d¹³** which were directly subjected to hydrogenolysis conditions to afford the piperidine hydrochloride salts **8a,b** as single diastereomers (¹H and ¹³C NMR). While the absolute stereochemistry at C4 had been previously established,⁴ it remained to determine the relative stereochemistry between the C2 and C4 substituents of piperidine products **8a–d**. It is well documented that the C2 and C6 carbon resonances in the ¹³C NMR are diagnostic for the determination of different stereoisomers of 2,4-dimethyl piperidines.¹⁴ Piperidine **8a** exhibited chemical shifts of 52.1 and 46.8 ppm for C2 and C6, respectively, which were in excellent agreement with reported values (C2=52.1 and C6=46.7 ppm) for *cis*-2,4-dimethyl piperidine.^{14–16} Moreover, the literature values for C2 and C6 (46.1 and 41.6 ppm, respectively) for the *trans*-2,4-dimethyl piperidine lie outside the values observed for **8a–d**.¹⁴ In addition, piperidines **8b–d** exhibited almost identical chemical shift data for C2 and C6 as **8a** indicative of a *cis*-relationship between the two substituents (Scheme 2).

In order to more clearly understand the observed diastereoselection in the reduction step, various other reducing agents were examined. Treatment of the 6-methyl lactams **6b,d** under dissolving metal

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Entry	Electrophile	Yield (% 6)	Diast. Ratio ¹	
a	MeI	40	>20:1	
b	<i>n</i> -BuI	56	>20:1	
с	AllylBr	66	>20:1	
d	BnBr	76	>20:1	

Table 1Alkylation and hydrolysis of 5

¹Diastereomeric ratios determined by ¹H NMR and GC analysis.

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Scheme 2.

conditions (Ca/NH₃) followed by addition of Et₃SiH/TFA afforded the 4,6-disubstituted-2-piperidones **9b,d** in moderate yields (Scheme 3).¹⁷ Reduction of piperidones **9b,d** with LiAlH₄ in refluxing Et₂O afforded the desired piperidine hydrochloride salts **8b,d** whose physical properties (¹H NMR, ¹³C NMR, $[\alpha]_D^{23}$, and mp) were identical to those reported for the piperidine hydrochlorides prepared in Scheme 2.^{10c}



The observed diastereoselection in the creation of the C2 stereocenter in 8 via these two protocols can be explained by assuming that the bulky hydride reagent (Et₃SiH) can only approach the iminium ion derived from 6 from the *exo* face (top) or face severe steric interactions with the 4-substituent, R, which resides on the *endo* face. A similar argument has been presented in related systems.¹⁷

In summary, we have prepared chiral non-racemic 2,4-*cis*-disubstituted piperidines in several steps from thiolactam 2 by employing a highly diastereoselective homoenolate alkylation to 6. This is equivalent to substitution at the 4-position of the piperidine ring. Two different reduction methods have been successfully employed to cleave the ring C-O bond to ultimately set the stereochemistry at the C2 center of the piperidine ring. The inversion of the methyl group in 6 appears to be controlled by the 4-substituent on 6. Furthermore, a practical synthesis of chiral, non-racemic *cis*-4,6-disubstituted-2-piperidones 9 has been achieved. Additional studies to extend this methodology to more highly functionalized piperidine systems are currently underway and will be reported in due course.

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