Conversion of curcumin into heterocyclic compounds as potent anti-diabetic and antihistamine agents

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ABSTRACT

Potential biologically active derivatives of the Curcumin were prepared by the cyclocondensation reaction cyclohexanone 2, imino pyrimidine 3,pyrmidinones 4,7 thiopyrimidine 6 and 5, ,7pyranone when treated with acetylacetone, guanidine , urea ethylcyanoacetate , thiourea and ethylacetoacetate respectively. The structures of compounds (2-7) were elucidated by means of microanalysis as well as spectral measurements such as IR, ¹H-NMR,MS. The anti-diabetic potential of Curcumin derivatives were evaluated by assessing amylase inhibition assay, alsoinhibition of histamine releaseactivity of Curcumin derivatives were assessed by U937 human monocytes. The results for amylase inhibition activity revels that the Curcumin inhibits α -amylase in a concentration dependent manner. Compound 4 and 5 exhibited significant inhibitory activity against amylase enzyme and was comparable with that of acrabose. Also,compound5, 6 and 7 exhibited significant inhibitory activity against histamine. Our results concluded that Curcumin pyrmidinones and pyranone derivatives have highly effects as anti-diabetic and anti-histamine activities.

KEYWORDS: Curcumin, pyrimidine, pyranone, cyclohexanone derivatives, anti-diabetic, anti-histamine

INTRODUCTION

Curcumin, which imparts the yellow color to curry, is a natural product of the spice turmeric, Curcuma longa L (Zingiberaceae). Curcumin exhibits a variety of pharmacologic activities, including anti-inflammatory, anticancer, antioxidant, wound-healing, antimicrobial effects¹, antiallergic activity ^{2,3} and inhibits degranulation of the RBL-2H3 tumor mast cell line in culture^{4,5}.Also, it prevents biliary disorders, anorexia, coughs, diabetes, hepatic disorders, rheumatism, sinusitis, cancer and Alzheimer disease^{6,7}.

The ethyl acetate extract of *Curcuma longa* L. and curcumin were found to decrease histamine release from mast cells by blocking intracellular signalling events in mast cells. The anti-allergy activities of curcumin and curcumin-related compounds in relation to their antioxidant activities. Most of these compounds were shown to inhibit histamine release from RBL-2H3 cells induced by concanavalin A or a calcium ionophore⁸.

Flavonoid-rich fractions of plants have been reported to be effective as antihyperglycemic and antihyperlipidemic agents in animal models of diabetes⁹.Plants derived polyphenols such as quercetagetin, fisetin and quercetin, which belong to the flavonoid family have been shown to be effective inhibitors of mammalian alpha-amylase with IC50s in the order of micromolar¹⁰.Since curcumin [1, 7-bis (4-hydroxy-3-methoxyphenyl)-1, 6-heptadiene-3,5dione] could be viewed as a flavonoid of polyphenolic compounds, in this investigation we researched the inhibitory effect of curcumin on α -amylase and the effect of curcumin orally on blood glucose levels in STZinduced diabetic and normal rats¹¹.

Pyrimidines exhibit a wide broad as biologically active agents they showed Antioxidant ¹²,antifungal¹³, Antibacterial ¹⁴, antihypertensive ¹⁵, antitumor ¹⁶, activity. Also pyranone which is six membered oxygenated unsaturated ester heterocyclic compoundsthat shares chemical and physical properties reminiscent of alkene and aromatic compoundsshowed a wide range of biological activity as Antileukemic¹⁷antitumor¹⁸, anti-fungal¹⁹inhibiting, HIV protease ²⁰qualitiesalso it was found that In the last several decades, cyclohexenone derivatives have received considerable attention due to their wide range of applications. Cyclohexenone derivatives exhibit antitumor²¹ anti-bacterial ²², antimicrobial²³ agents this promoted the author to convert curcumin into this newly heterocyclic derivatives to enhance its biological activity. This promoted the author to convert curcumin into this newly heterocyclic derivatives to enhance its biological activity.

EXEPERIMENTAL SECTION

All melting points are uncorrected .IR spectra (KBr) were recorded with a Perkin Elmer Spectrum RXIFT-IR system .1HNMR were measured with a Varian Gemini 200 MHZ instrument using TMS as internal standard and mass spectra were measured with a Shimadzu GC-MS-QP 100 EX mass spectrometer.

Synthesis of 6-acetyl-3,5-bis(4-hydroxy-3-methoxystyryl)cyclohexa-2,4-dienone(2)

A mixture of **1** (0.01 mol) and acetylacetone (0.01 mol) in ethanol(50ml)was refluxed for6 hrs and left to cool. The separated solid was filtered off, washed with ethanol , dried and recrystallized from acetic acid. yield 85% and M.P 198 $^{\circ}$ C.

Analysis of 2 C₂₆H₂₄O₆ (432) (%) calcd C, 72.21; H, 5.59 Found: C, 72.18; H, 5.62

Synthesis of 4,4'-(1E,1'E)-2,2'-(2-imino-1,2-dihydropyrimidine-4,6-diyl)bis(ethene-2,1-diyl)bis(2-methoxyphenol) (3)

A mixture of **1** (0.01 mol) and guanidine.HCl (0.01 mol) in ethanol(50ml)was refluxed for 6 hrs and left to cool. The separated solid was filtered off, washed with ethanol , dried and recrystallized from acetic acid. yield 85% and M.P 204 $^{\circ}$ C.

Analysis of **3** C₂₂H₂₁N₃O₄ (391) (%) calcd C, 67.51; H, 5.41; N, 10.74 Found: C, 67.55; H, 5.40; N, 10.71

Synthesis of 4,6-bis((E)-4-hydroxy-3-methoxystyryl)pyrimidin-2(1H)-one (4)

A mixture of 1 (0.01 mol) and urea (0.01 mol) in ethanol(50ml)was refluxed for6 hrs and left to cool. The separated solid was filtered off, washed with ethanol, dried and recrystallized from ethanol. Yield78% and M.P 217 °C.

Analysis of **4** C22H20N2O5 (392) (%) calcd C, 67.34; H, 5.14; N, 7.14; Found: C, 67.40; H, 5.10; N, 7.12

Synthesis of 3-acetyl-4,6-bis((E)-4-hydroxy-3-methoxystyryl)-2H-pyran-2-one (5)

A mixture of 1 (0.01 mol) and ethylacetoacetate (0.01 mol) in ethanol (50ml)was refluxed for6 hrs and left to cool . The separated solid was filtered off, washed with H2O , dried and recrystallized from ethanol. Yield68% and M.P 253 $^{\circ}$ C.

Analysis of 5 C₂₅H₂₂NO₇ (434) (%) calcd C, 69.12; H, 5.10; Found C, 69.00; H, 5.23

Synthesis of 4,6-bis((E)-4-hydroxy-3-methoxystyryl)pyrimidine-2(1H)-thione (6)

A mixture of 1 (0.01 mol) and thiourea (0.01 mol) in DMF(50ml)was refluxed for6 hrs flitter while hot . The separated solid, washed with ethanol , dried and recrystallized from ethanol. Yield74% and M.P 228 °C.

Analysis of **6** C₂₂H₂₀N₂O₄S (408) (%) calcd C, 64.69; H, 4.94; N, 6.86; S, 7.85Found: C, 64.74; H, 5.00; N, 6.82; S, 7.83

Synthesis of5-(1-(4-bromophenyl)ethylidene)-4,6-bis((E)-4-hydroxy-3 methoxystyryl)-2oxo-5,6-dihydro-2H-pyran-3-carbonitrile (7)

Amixture of 1 (0.01 mol), (0.01 mol) of p-bromoacetophenone, and ethylcyanoacetate (0.01 mol) in sodium ethoxide (0.5 gm in20 ml) was refluxed for 5hrs. The solid product that separated after cooling and pouring into ice was filtered off, washed well with water, dried and recrystallized from EtOH. yield 77% and M.P 250 $^{\circ}$ C.

Analysis of 7 C₃₂H₂₆BrNO₆ (599) (%) calcdC, 64.01; H, 4.36; Br, 13.31; N, 2.33Found: C, 63.98; H, 4.38; Br, 13.29; N, 2.36.

Biological assessment

All chemicals were received from Sigma-Aldrich (St. Louis, MO, USA).

1-Determination of α-amylase inhibitory activity:

We used the assay of colorimetric microplate to determined the activity of α -amylase inhibitory by utilizing a well-established protocol.²⁴ The reaction of enzyme consisted of 4 mg α -amylase of porcine pancreatic (type VI-B, ≥ 10 units/mg solid), 1.25 mM p-nitrophenyl- α -D-maltopentaoside (PNPG-5) and 1 mg of samples in the 96-well plate. The microplate reader was using to monitor the reaction of enzyme at 405 mn. The following equation was using to calculated the percentage of α -amylase inhibition:

Equation:

%inhibition = Abs Control – Abs Sample x 100

Abs Where:

Abs control = absorbance of control reaction.

Abs sample = absorbance of test sample.

The IC_{50} value was defined as the concentration of alpha-amylase inhibitor to inhibit 50% of its activity under the assay conditions.

In vitro Anti-Inflammatory assays:

Histamine Release Assay:

U937 human monocytes were used to study the effect of samples on histamine release. Approximately 50,000 U937 cells were plated in a 96-well cell culture plate (Corning Life Sciences, Lowell, MA, USA) and treated with various concentration (1000-7.81) of samples, in presence or absence of 20 nM Phorbol Myristate Acetate (PMA) (Sigma-Aldrich, St. Luis, MO, USA) for one h. The cell culture supernatants collected from either untreated control or treated cultures were clarified at 10,000 g for 5 min at 4°C and assessed for released histamine by a commercially available EIA kit (SPI-Bio, France)²⁵.

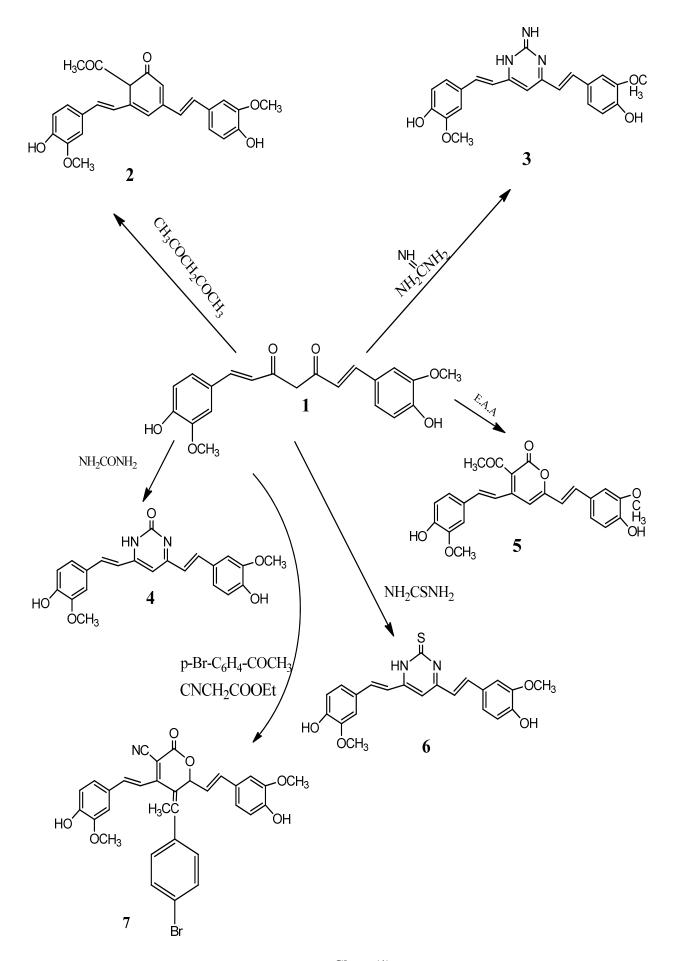


Chart (1) Biological and Pharmaceutical Bulletin Advance Publication

RESULTS AND DISCUSSION

The new derivatives were prepared following the reaction sequences depicted in Chart (1).

Treatment of curcumin with acetylacetone in ethanol afforded the cyclohexa-2,4dienone derivative (2) . Its . IR (KBr, cm⁻¹): 3300 (O-H) , 1722, 1692 (C=O); 1629 (C=C), 1150 (C-O). ¹H-NMR (DMSO-d6, δ ppm): 9.22(s,2H-OH),7.78-7.25 (m,6H Ar-H),; 6.44(s, 4H, CH=CH); 3.04(s, 6H, OCH₃);2.90(s, 3H, CO-<u>CH₃</u>) MS: molecular ion peak at m/e 432(8.15%); molecular fragments at m/e 77 (100%) for C₆H₅.

The condensation of curcumin with guanidine.HCl in ethanol formed -iminodropyrimidine(**3**)which showed IR (KBr, cm⁻¹): 3250 (O-H) absence of (C=O); 1614 (C=N), 1200 (C-O). ¹H-NMR (DMSO-d6, δ ppm):9.02(s,2H-OH), 7.78-6.85 (m,6H Ar-H),6.92(s, 4H, CH=CH); 4.06(s, 1H, C=NH); 4.96(s, 1H, NH), 3.54(s, 6H, OCH₃); MS: molecular ion peak at m/e 391(15.3%); molecular fragments at m/e 77 (100%) for C₆H₅¬‡.

While condensation with urea and thiourea gave pyrimidin-2(1H)-one (4), pyrimidine-2(1H)-thione(6) which spectral data gave the following data for (4) . IR (KBr, cm⁻¹): 3400 (OH/NH) , 1735(C=O); 1608 (C=N), 1200 (C-O). ¹H-NMR (DMSO-d6, δ ppm): 7.90-7.2(m,6H Ar-H), 5.01(s,2H-OH), 4.11(s,1H-NH),2.89(s, 1H, CH-CN); 6.03(s, 4H, CH=CH); 3.41(s, 6H, OCH₃); MS: molecular ion peak at m/e 392(8.3%); molecular fragments at m/e 77 (100%) for C₆H₅7;

And for **(6)**IR (KBr, cm⁻¹): 3420 (OH/NH) absence of (C=O); 1608 (C=N),1405(C=S), 1150 (C-O). ¹H-NMR (DMSO-d6, δ ppm):9.18(s,2H-OH), 7.97-6.89(m,6H Ar-H), 5.84(s, 4H, CH=CH),4.42(s,1H-NH),3.56(s, 1H, CH-CN);; 3.24(s, 6H, OCH₃); MS: molecular ion peak at m/e408(9.05%); molecular fragments at m/e 77 (100%) for C₆H₅¬‡

While the condensation of curcumin using active methylene group afforded the pyran-2-one(5) and pyran-3-carbonitrile (7) when treated with ethylacetoacetate and ethylcyanoacetate respectively

The spectral data of (5) showed . IR (KBr, cm⁻¹): 3356 (O-H) 1720 (C=O);1772(C=O lactam), 1620 (C=C), 1185 (C-O). ¹H-NMR (DMSO-d6, δ ppm): 9.12(s,2H-OH),7.97-7.01 (m,6H Ar-H), 5.75(s, 4H, CH=CH); 3.54(s, 6H, OCH₃), 2.27(s,3H,COCH₃),2.02(s,2H,CH₂); MS: molecular ion peak at m/e 436(24.5%); molecular fragments at m/e 77 (100%) for C₆H₅ \neg^{\ddagger}

While for (7) . IR (KBr, cm⁻¹): 3420 (OH) ,2150(c \equiv N), 1755(C=O); 1611 (C=N), 1210 (C-O). ¹H-NMR (DMSO-d6, δ ppm):9.07(s,2H-OH), 7.86-7.01(m,10H Ar-H),5.79(s, 4H, CH=CH); 3.52(s, 6H, OCH₃) , 2.78(s, 1H, CH-CN);; MS: molecular ion peak at m/e 599(11.02%); molecular fragments at m/e 77 (100%) for C₆H₅¬‡;

α-Amylase inhibitory activity by curcumin derivatives

The α -amylase inhibiting activity of curcumin derivatives (2-7) at different concentrations (7.81-1000 μ M) was studied, and the results are depicted in the Table 1, fig 1. Compound 4 and 5 exhibited significant inhibitory activity against amylase enzyme and was comparable with that of acrabose and original curcumin. Compound 5 showed 18.34±1.3, 25.41±1.5, 35.21±0.58, 44.85±1.3, 51.78±1.5, 59.32±0.72, 62.50±58 and 71.63±1.2 % inhibition at 7.81, 15.63, 31.25, 62.5, 125, 250, 500 and 1000 μ M/l concentrations, respectively, IC₅₀ = 95.5. Compound 4 was found to possess pronounced inhibitory activity with 31.11±1.2, 36.25±1.5, 42.16±0.63, 57.46±0.58, 61.78±1.2, 65.24±1.5, 72.14±0.63 and 79.32±0.72 % inhibition at 7.81, 15.63, 31.25, 62.5, 125, 250, 500 and 1000 μ M/l concentrations, IC₅₀ = 47.26.Acarbose and original curcumin showed 86.32±0.63and 76.37±0.71 % inhibition at 1000 μ M/l concentration, IC₅₀ = 47.26mM) and Compound 5 (IC₅₀ = 95.5mM) are the most potent compared to other compounds. The inhibition of this compound may be due to pyrimidin and pyran rings¹²⁻¹⁹.

Our results agree with Mohamed, 2015^{26} he reported thatCurcumin inhibited α -amylase with an half maximal inhibitory concentration IC₅₀value 51.32 μ M. Curcumin at different concentrations i.e. 10, 50, 100 μ g showed dose dependent effect where as the standard Indomethacin at 50 μ g and 100 μ g exhibited much higher activity. Pancreatic and intestinal glucosidases are the key enzymes of dietary carbohydrates digestion and inhibitors of these enzymes may be effective in retarding glucose absorption. Pancreatic α -amylase enzyme inhibition potential of curcumin might be the most promising mechanism for the observed antidiabetic effect²⁷.

Sample conc. (µM)	7.81	15.63	31.25	62.5	125	250	500	1000	IC ₅₀
Control (acarbose)	37.81±1.20	40.75±1.50	48.84±1.20	59.31±1.50	60.17±0.63	69.37±1.20	80.14±0.58	86.32±0.63	34.71
Original curcumin	29.98±1.31	33.75±1.55	41.76±0.63	56.16±0.58	60.7±1.27	62.84±1.53	70.34±0.63	76.37±0.71	49.13
2	0	0	0	14.68±1.50	24.32±1.20	33.25±0.63	51.46±0.58	56.38±1.20	351.6
3	0	11.46±1.20	16.85±0.72	21.38±1.30	37.41±0.72	58.32±1.50	61.34±1.80	67.21±1.40	200.2
4	31.11±1.20	36.25±1.50	42.16±0.63	57.46±0.58	61.78±1.20	65.24±1.50	72.14±0.63	79.32±0.72	47.26
5	18.34±1.30	25.41±1.50	35.21±0.58	44.85±1.30	51.78±1.50	59.32±0.72	62.50±0.58	71.63±1.20	95.5
6	0	0	12.75±1.30	18.35±0.58	24.32±2.10	48.31±0.63	53.24±0.58	62.15±1.20	335.7
7	0	0	17.35±1.30	26.84±0.63	31.85±2.10	49.37±0.58	61.72±1.20	73.24±1.60	262.75

Table (1) α -amylase inhibitory of curcumin derivatives

All determinations were carried out in triplicate manner and values are expressed as the mean \pm SD. The IC₅₀ value is defined as the concentration of inhibitor to inhibit 50% of its activity under the assayed conditions.

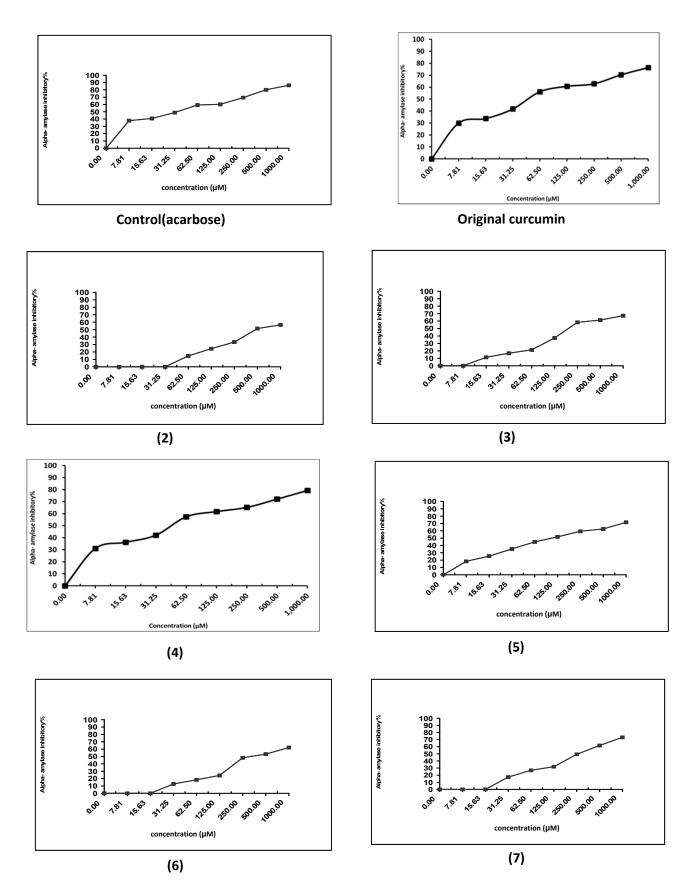


Fig (1): α-amylase inhibitory effect of control (acrabose), newly prepared curcumin derivatives 2-7

Inhibition of histamine release by curcumin derivatives

Sample conc. (µM)	0	7.81	15.63	31.25	62.5	125	250	500	1000	IC ₅₀
Original curcumin	13.11±0.32	17.56±1.40	26.26±1.31	43.17±1.20	53.33±0.62	60.19±1.35	64.29±0.60	69.35±1.34	73.37±1.51	52.26
2	0	0	0	0	13.45±1.2	21.35±1.5	34.13±2.2	49.28±1.5	58.32±0.58	539.8
3	0	9.25	17.36	24.32	33.85±1.6	46.34±1.5	52.41±0.58	55.62±1.2	61.85±0.63	200.37
4	0	6.38±1.5	13.25±0.85	19.35±1.2	31.25±0.72	46.31±0.72	53.18±0.58	59.32±1.2	71.34±1.5	192.13
5	0	29.35±2.5	36.82±2.1	41.85±1.6	58.41±1.3	62.41±0.58	68.72±1.2	74.65±2.1	79.31±1.8	46.6
6	9.35±1.2	11.32±0.63	28.41±0.72	52.34±1.2	56.32±0.58	61.91±0.72	66.81±1.5	70.14±2.1	73.25±1.6	29.7
7	0	23.58±1.5	36.28±1.2	52.13±1.2	54.32±0.63	60.17±1.2	63.25±0.58	76.34±1.2	82.31±1.5	29.15

Table (2) histamine release of curcumin derivatives

All determinations were carried out in triplicate manner and values are expressed as the mean \pm SD. The IC₅₀ value is defined as the concentration of inhibitor to inhibit 50% of its activity under the assayed conditions.

The histamine inhibiting activity of curcumin derivatives (2-7) at different concentrations (7.81-1000 μ M) was studied, and the results are depicted in the Table 2, fig 2. Compound 5, 6 and 7 exhibited significant inhibitory activity against histamineand was comparable with that oforiginal curcumin. Compound 5 showed 29.35±2.5, 36.82±2.1, 41.85±1.6, 58.41±1.3, 62.41±0.58, 68.72±1.2, 74.65±2.1 and 79.31±1.8 % inhibition at 7.81, 15.63, 31.25, 62.5, 125, 250, 500 and 1000 μ M/l concentrations, respectively, IC₅₀ = 46.6. Compound 6 and 7 were found to possess pronounced inhibitory activity with (11.32±0.63 and 23.58±1.5), (28.41±0.72 and 36.28±1.2), (52.34±1.2 and 52.13±1.2), (56.32±0.58 and 54.32±0.63), (61.91±0.72 and 60.17±1.2), (66.81±1.5 and 63.25±0.58), (70.14±2.1 and 76.34±1.2) and (73.25±1.6 and 82.31±1.5) % inhibition at 7.81, 15.63, 31.25, 62.5, 125, 250, 500 and 1000 μ M/l concentrations, respectively. IC₅₀ = 29.7 and 29.15, respectively.Original curcumin showed 73.37±1.51 % inhibition at 1000 μ M/l concentration, IC₅₀ =52.26.Compounds5

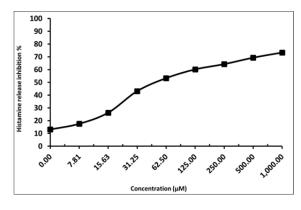
(IC₅₀= 46.6 μ M), 6 (IC₅₀ = 29.7 μ M) and 7 (IC₅₀ = 29.15 μ M) are the most potent compared to other compounds. The inhibition of this compound may be due to pyrimidin and pyran rings¹²⁻¹⁹.

Curcumin has been reported to have anti-allergic effects and can inhibit the release of histamine from mast cells²⁸. The histamine release from compound 48/80-treated RPMCs was reduced in a dose-dependent manner by curcumin (58 and 80% inhibition at 25 and 50 μ M, respectively).Curcumin inhibits mast cell-mediated anaphylactoid responses by suppressing histamine release from RPMCs²⁹.

Histamine levels were found to be raised by 8fold from the baseline upon induction with PMA. Curcumin when treated at different concentration 10,250 and 1000 ng mL-1 showed decrease in the percentage rise in histamine levels by 82.70, 67.51 and55.69% from U937 cells³⁰. Curcumin also suppressed the IgE-dependent PSA reaction in a mast cell-dependent in vivo model of systemic allergic reaction³¹ with a potency equivalent to that of the H1 histamine antagonist³².

Conclusion

From this study we have synthesized new pyrimidine, pyranone, cyclohexanone derivatives from curcumin and proved its structure by different spectral data .This newly prepared compounds had showed remarkable biological activity as antidiabatic and antihistamine. The Curcumin derivatives (4 and 5) showed a prominent inhibition of α -amylase with IC₅₀=47.26 and 95.5, respectively. On the other hand, Curcumin derivatives (5, 6 and 7) exhibited significant inhibitory activity as antihistamine.



Original curcumin

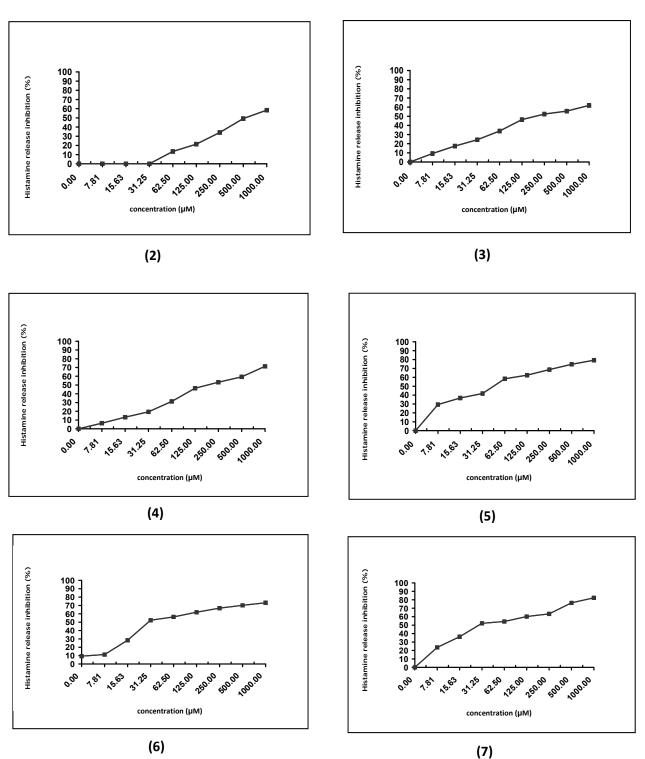


Fig (2): histamine release of curcumin derivatives (2-7)

Conflict of Interest

The authors declare no conflict of interest.

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