Functionalization of 3-Chlorobenzaldehyde

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Abstract: 2-Substituted 3-chlorobenzaldehydes were prepared from the corresponding 2-(3-chlorophenyl)-1,3-dioxolanes using an *ortho*-lithiation strategy. The 6-chloro-2-formylbenzamide exists only in a ring form, but 6-chloro-2-formylbenzoic acid esters were isolated in both forms as open chain and cyclic tautomers. 7-Chloro-3-hydroxy-3*H*-isobenzofuran-1-one reacted with nucleophilic reagents at the carbonyl or quaternary carbon depending on the character of nucleophile.

Key words: 3-chlorobenzaldehyde, 2,3-dihydroisoindol-1-ones, 3*H*-isobenzofuran-1-ones, directed *ortho*-lithiation, ring-chain tautomers

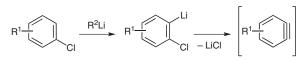
One of the aims of advanced organic synthesis is still the formation of regiospecific substituted benzene rings. Polysubstituted benzenes are used in many areas of pharmaceutical and agricultural chemistry as versatile intermediates.¹

We have demonstrated a general synthesis of 2-modified 3-chlorobenzaldehydes via *ortho*-lithiation² of cyclic acetal **1**. So far, Comins and co-workers reported in situ protection of aromatic aldehydes via the formation of α aminoalkoxides using *n*-BuLi and TMEDA.³ Carter and co-workers showed that aromatic aldimines could be *ortho*-lithiated and trapped with electrophiles and the products were converted to the corresponding benzaldehyde derivatives.⁴ Still the dimethyl acetals are more often used as aldehyde protecting group in *ortho*-lithiation reaction.⁵ In 1999, Yus and co-workers described naphthalene-catalyzed lithiation of 2-(chlorophenyl)-1,3dioxolanes by chlorine–lithium exchange.⁶

The aldehyde group was protected with ethylene glycol, because dioxolane ring has lower reactivity to alkyllithium reagents and oxygen atoms could coordinate with lithium ion. Thus, acetal functionality works as the lithiationdirecting group (LDG). The acetal group is considered to have a poor *ortho*-directing capacity, but can be easily transformed into the corresponding aldehydes by acid hydrolysis. On the other hand, chlorine atom is an electronaccepting group and it raises the acidity of hydrogen atom in 2-position.

Chlorine atom may cause difficulties connected with decreased stability of unsubstituted 2-chlorophenyllithiums

SYNTHESIS 2012, 44, 2200–2208 Advanced online publication: 19.06.2012 DOI: 10.1055/s-0031-1289784; Art ID: SS-2012-N0094-OP © Georg Thieme Verlag Stuttgart · New York $(R^1 = H)$ toward elimination of LiCl at -90 °C⁷ or -42 °C³ to form the intermediate benzyne (Scheme 1).



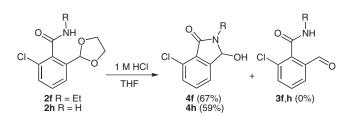
Scheme 1 Benzyne formation

The elimination temperature is highly dependent on the overall electronic effects induced in the ring by the substituents. Mortier and co-workers^{7c,d} have shown the elimination of LiCl in 2-chlorobenzoic acid between -50 and -30 °C. Meyers and co-workers^{7a} have described a benzyne generation from (3-chlorophenyl)oxazoline at 0 °C.

In our experiments, the elimination of LiCl did not proceed at -78 °C. At the same temperature, free 3-chlorobenzaldehyde did not form benzyne.³ Therefore, directed *ortho*-metalation of 2-(3-chlorophenyl)-1,3-dioxolane (1) was achieved by treatment with 1.5 equivalents of *n*-BuLi in THF at -78 °C. The resulting orange lithiated species easily reacted with a variety of electrophiles to give the *ortho*-substituted products **2a**–i. Some limitations of the reaction were observed. Thus, *ortho*-carboxylic- and *ortho*-methylsulfonyl-substituted 2-(3-chlorophenyl)-1,3dioxolanes **2j**,**k** were not obtained directly, but 2-fluorobenzaldehyde acetal **2i** was formed in low (19%) yield (Table 1). The low yield of 2-fluorination is explained due to spacial disturbances of used *N*-fluoro electrophile.

Newly formed acetals **2** were selectively deprotected under acidic conditions (aq 1 M HCl in THF) at reflux temperature to give the corresponding aldehydes **3** (Table 2).

The acetal protecting group of N-alkyl-2-formylbenzamides **2f**,**h** was cleaved in acidic media, but expected aldehydes **3f**,**h** were not obtained. Instead, the hydroxylactams **4f**,**h** were always isolated (Scheme 2).



Scheme 2 Cleavage of *N*-alkyl-2-formylbenzamides acetal protecting group

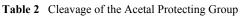
| $CI \xrightarrow{n-BuLi}_{THF, -78 °C} \left[CI \xrightarrow{Li}_{CI} \xrightarrow{O}_{O} \right] \xrightarrow{electrophile}_{2} CI \xrightarrow{E}_{O} \xrightarrow{O}_{C}$ | | | |
|---|--------------------------------------|----------------------|------------------------|
| Product | Electrophile | Е | Yield (%) |
| 2a | I ₂ | Ι | 56 |
| 2b | MeI | Me | 60 (1:1) ^a |
| 2c | MeSSMe | MeS | 66 |
| 2d | ClCO ₂ Et | CO ₂ Et | 68 |
| 2e | ClCON(Me) ₂ | CON(Me) ₂ | 32 |
| 2f | EtN=C=O | CONHEt | 42 |
| 2g | PrN=C=O | CONHPr | 44 |
| 2h | TMSN=C=O | CONH ₂ | 53 |
| 2i | (PhSO ₂) ₂ NF | F | 19 (10:3) ^a |
| 2j | CO ₂ | CO ₂ H | _ |
| 2k | MsCl | Ms | _ |

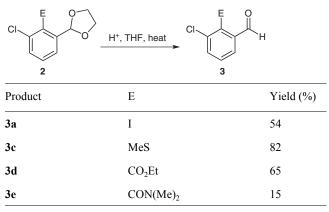
Table 1 Synthesis of 2-Substituted 3-Chlorobenzaldehyde Acetals

^a Ratio (product **2**:starting material **1**) detected by ¹H NMR analysis.

Open chain tautomers were characterized by IR and NMR spectroscopy. Ring structure of hydroxyisoindolinone **4f** was based on the X-ray crystallography (Figure 1). *N*-Al-kyl-3-chloro-2-amidobenzaldehydes **3** exist only as the cyclic tautomers **4**.

Similar results with ring-chain tautomerism were observed in *o*-formylbenzoic amides existing as the ring tautomers, that is, hydroxylactam.^{8–11} Formation of the ring form was stimulated in conditions with rising electrophilicity of aldehyde group and increased nucleophilic character of nitrogen atom. Amides of *o*-formylated benzoic acids are shown to exist only in the cyclic form by IR and NMR spectroscopy.^{8,10,11} The presence of hydrogen bond between two molecules of cyclic tautomers (Scheme 3) was proved by IR spectroscopy.¹⁰





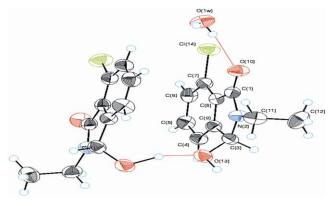
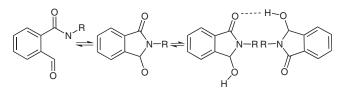
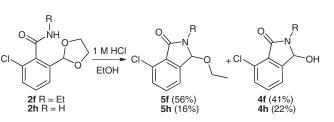


Figure 1 ORTEP view of 4f showing 50% probability displacement ellipsoids and atom labeling scheme

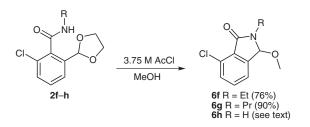


Scheme 3 Ring-chain tautomerism and association of 2-formylbenzamides

The *N*-alkyl-2-formylbenzamides 2f, **h** were refluxed in acidic media in polar protic solvents EtOH (Scheme 4) and MeOH (Scheme 5). In this case, nucleophilic substitution occurred to give products 5 and 6 in moderate to high yields. In all the cases aldehydes were not detected.



Scheme 4 Nucleophilic substitution of EtOH in isoindolinones



Scheme 5 Nucleophilic substitution of MeOH in isoindolinones

When EtOH was used as a solvent, additional hydroxylactams 4 together with alkoxylactams 5 were isolated. Less nucleophilic 2-chloro-6-(1,3-dioxolan-2-yl)benzamide (2h) gave a mixture of hydroxylactam 4h and ethoxylactam 5h in smaller yields than benzamide 2f (Scheme 4).

Reaction of amides 2f and 2g in MeOH in acidic medium produced methoxy derivatives 6f and 6g in high yields (76% and 90%, Scheme 5). An X-ray crystal structure (Figure 2) was obtained for 7-chloro-2-ethyl-3-methoxy-2,3-dihydro-1*H*-isoindol-1-one (6f).

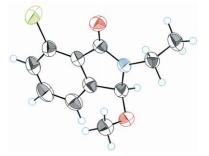
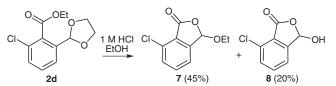


Figure 2 ORTEP view of 6f showing 50% probability displacement ellipsoids and atom-labeling scheme

In the case of 2-chloro-6-(1,3-dioxolan-2-yl)benzamide (**2h**), a mixture of 3 compounds, 3-methoxylactam **6h**, 2-chloro-6-dimethoxymethylbenzoic acid methyl ester, and traces of 2-chloro-6-formylbenzoic acid methyl ester were detected by ¹H NMR spectroscopy. Similar phenomenon with the intramolecular cyclization was observed for ace-tal **2d**. Thus, after removal of acetal function in EtOH and chromatographic separation only lactone type substances **7** and **8** were isolated (Scheme 6).

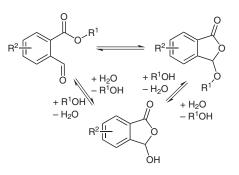
o-Formylbenzoic acids and esters exist as both tautomers, stable in neutral conditions. Transformations from tautomeric chain to ring form occur in acidic or basic media or



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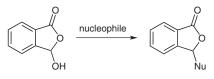
Scheme 6 Reaction of ethyl 2-chloro-6-(1,3-dioxolan-2-yl)benzoate (2d) with EtOH in acidic conditions

at elevated temperatures. Thus, esterification of carboxylic acid with alcohols in the presence of acidic catalysts (Fisher–Speier esterification) gives a mixture of isomers (Scheme 7).⁹ Ratios of products depend on the time of reaction (thermodynamic and kinetic control).



Scheme 7 Ring-chain tautomerism of formylbenzene-carboxylic acids and esters

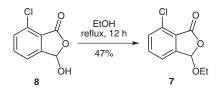
Wheeler and co-workers¹² and Kagan¹³ reported the presence of phthalaldehydic acid in the 3-hydroxyphthalide form both in solid state and in aqueous solution. Wheeler and co-workers¹² described phthalaldehydic acid reactions with a variety of alcohols forming 3-alkoxyphthalides (Scheme 8). Analogous is the reaction with mercaptans or thiophenols. With less hindered primary alkylamines and ammonia, 3,3'-iminodiphthalides were obtained as the only products. Sterically hindered primary alkylamines gave 3-alkylaminophthalides. The corresponding 3-amidophthalides were obtained from alkyl and aryl acid amides. Similarly 3-hydroxyphthalide reacts with urea or substituted ureas, alkyl carbamates, and carboxylic acid anhydrides (Scheme 8).¹²



Nu = OR, SR, NHR, NR¹R², NHCOR, OCOR, NHOCOR, NHCOR¹R²

Scheme 8 Reaction of phthalaldehydic acid with nucleophiles

Heating 7-chloro-3-hydroxy-2-benzofuran-1(3H)-one (8) in EtOH gave 7-chloro-3-ethoxyphthalide (7) as the product of nucleophilic substitution in the 3-position (Scheme 9).



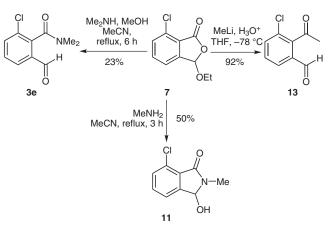
Scheme 9 Reaction of 7-chloro-3-hydroxy-2-benzofuran-1(3*H*)-one **(8)** with EtOH

Depending on the solvent used different products were obtained with primary amines. In polar nucleophilic solvent, a mixture of 3-alkoxylactam **10**, 3-alkylaminolactam **9**, and 3-hydroxylactam **11** together with unreacted starting material was obtained (Scheme 10).

Weaker nucleophiles such as *N*-methylurea did not react with 7-chloro-3-hydroxy-2-benzofuran-1(3H)-one (8) at position C3, but gave the product 12 (Scheme 10) after opening of lactone ring with subsequent lactam ring formation.

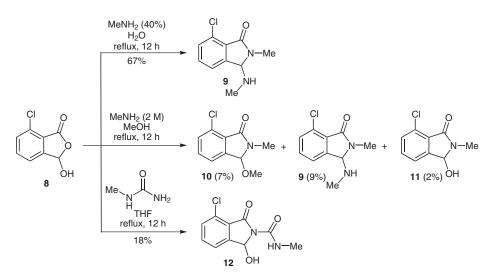
The reactivity of 3-ethoxyphthalide 7 with different nucleophiles is shown in Scheme 11. Thus, reaction with methylamine gave the 3-hydroxylactam 11 in good yield (50%). Another *ortho*-substituted aldehyde, 2-acetyl-3-chlorobenzaldehyde (13), was obtained by treating lactone 7 with MeLi in high isolated yield (92%). Reaction with dimethylamine was more complex and aldehyde 3e was isolated in low yield (23%).

In summary, a simple and efficient procedure of directed *ortho*-lithiation of 3-chlorobenzaldehyde has been demonstrated to provide trisubstituted benzaldehydes regio-specifically. When a deprotection of acetal functionality in substituted benzamides was performed, incompatibility of amide and aldehyde groups in the reaction products was shown; intramolecular cyclization took place and cyclic amide products were isolated.



Scheme 11 Reaction of 7-chloro-3-ethoxy-2-benzofuran-1(3*H*)-one (with nucleophiles

All nonaqueous reactions were performed under dry conditions in the atmosphere of argon, unless otherwise specified. Commercial grade reagents and anhyd solvents were used as received from suppliers and no attempts were made to purify or to dry these components further. TLC was performed using silica gel 60 F254 (Merck) plates. Visualization of TLC plates was made with short wave (254 nm) UV light. Flash column chromatography was carried out using silica gel 0.035–0.070 mm (Merck). ¹H, ¹³C, and ¹⁹F NMR spectra were obtained on 200 and 400 MHz NMR spectrometers (Varian Mercury 200 and Varian Mercury Plus 400) and are reported in ppm δ values, using TMS as an internal reference. Microanalyses were performed using a Carlo Erba-106 elemental analyzer. Melting points were determined in capillary tubes using a Gallenkamp apparatus. IR spectra were recorded on Perkin-Elmer IR spectrometer. For compounds 4f and 6f diffraction data were collected on a Bruker-Nonius KappaCCD diffractometer using graphite monochromated Mo-K α radiation ($\lambda = 0.71073$ Å). The crystal structures of **4f** and 6f were solved by the direct method and refined by full-matrix least squares. All non-hydrogen atoms were refined anisotropically. For further details, see Supporting Information and crystallographic data for 4f and 6f.14



Scheme 10 Reaction of phthalaldehydic acid 8 with N-nucleophiles

2-(3-Chlorophenyl)-1,3-dioxolane (1)⁶ [CAS # 64380-53-6]

A mixture of the commercially available 3-chlorobenzaldehyde (52.8 g, 376 mmol), ethylene glycol (32 mL, 564 mmol), and a catalytic amount of *p*-TsOH (50 mg) in toluene (50 mL) was refluxed in a Dean–Stark apparatus for 14 h. The solvent was evaporated and the residue was condensed (62–67 °C/0.15 mbar) to give the title compound as a colorless liquid; yield: 64.6 g (93%).

¹H NMR (200 MHz, CDCl₃): δ = 4.06 (m, 4 H, CH₂CH₂), 5.79 (s, 1 H, CH), 7.29–7.40 (m, 3 H, H-4, 5, 6), 7.46–7.53 (m, 1 H, H-2).

MS (EI, 70 eV): *m*/*z* (%) = 73.1 (50), 89.1 (25), 112.0 (25), 125.0 (12), 139.0 (47), 183.0 (100), 184.0 (30, [M]⁺).

Ortho-Functionalization of 3-Chlorophenyl Acetal 1; General Procedure

A solution of 2-(3-chlorophenyl)-1,3-dioxolane (1; 1.84 g, 10.0 mmol) in anhyd THF (20 mL) was cooled to -78 °C. Then, a hexane solution of 2.5 M *n*-BuLi (6.0 mL, 15.0 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added the electrophile (1.2–5.0 equiv). The mixture was warmed to -20 °C and sat. aq NH₄Cl (10 mL) was added. The mixture was extracted with MTBE (50 mL) and the organic layer was separated, washed with H₂O (2 × 25 mL) and brine (25 mL), and dried (Na₂SO₄). The residue obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc).

2-(3-Chloro-2-iodophenyl)-1,3-dioxolane (2a)

A solution of 1 (6.74 g, 36.5 mmol) in anhyd THF (40 mL) was cooled to -78 °C. Then, a hexane solution of 2.5 M *n*-BuLi (22 mL, 54.8 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added a solution of I₂ (27.8 g, 109.5 mmol) in THF (100 mL). The mixture was warmed to -20 °C and sat. aq Na₂S₂O₃ (50 mL) was added. After workup, the yellow oil obtained was purified by flash chromatography on silica gel (hexane–EtOAc, 10:1) affording the product as a colorless oil; yield: 6.40 g (56%).

IR (film): 3367, 3067, 1729, 1419, 1369, 1149, 1101–1091, 944, 788 $\rm cm^{-1}.$

¹H NMR (200 MHz, CDCl₃): δ = 4.00–4.10 (m, 4 H, CH₂CH₂), 5.99 (s, 1 H, CH), 7.28–7.36 (m, 1 H_{arom}), 7.40–7.52 (m, 2 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 60.5, 101.8, 106.9, 125.2, 129.1, 130.0, 139.6, 142.3.

MS (EI, 70 eV): m/z (%) = 73.1 (80), 182.0 (20), 309.0 (100, [M]⁺). HRMS (ESI): m/z [M + H]⁺ calcd for C₉H₈ClIO₂: 310.9336; found: 310.9316.

2-(3-Chloro-2-methylphenyl)-1,3-dioxolane (2b)

A solution of 1 (1.84 g, 10.0 mmol) in anhyd THF (20 mL) was cooled to -78 °C. Then, a hexane solution of 2.5 M *n*-BuLi (6.0 mL, 15.0 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added MeI (3.1 mL, 50.0 mmol). After workup, the yellow oil obtained was purified by flash chromatography on silica gel (hexane–EtOAc, 10:1) affording the product as a colorless oil; yield: 1.62 g (60%) (~1:1 mixture with starting material).

IR (film): 3343, 1723, 1444, 1385, 1267, 1149, 1097, 1018, 958, 787 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 2.45 (s, 3 H, CH₃), 3.97–4.18 (m, 4 H, CH₂CH₂), 5.97 (s, 1 H, CH), 7.09–7.19 (m, 1 H_{arom}), 7.34–7.52 (m, 2 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 15.4, 65.3, 101.8, 124.3, 126.5, 129.8, 134.7, 135.3, 137.6.

2-[3-Chloro-2-(methylsulfanyl)phenyl]-1,3-dioxolane (2c)

A solution of 1 (4.62 g, 25.0 mmol) in anhyd THF (40 mL) was cooled to -78 °C. Then, a hexane solution of 2.5 M *n*-BuLi (15 mL, 37.5 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added Me₂S₂ (11 mL, 125 mmol). After workup, the yellow oil obtained was purified by flash chromatography on silica gel (hexane–EtOAc, 10:1) affording the product as a colorless oil; yield: 3.81 g (66%).

IR (film): 3066, 1729, 1689, 1575, 1419, 1369, 1148, 1111, 964, 791 $\rm cm^{-1}.$

¹H NMR (200 MHz, CDCl₃): δ = 2.40 (s, 3 H, SCH₃), 4.01–4.20 (m, 4 H, CH₂CH₂), 6.46 (s, 1 H, CH), 7.32 (d, *J* = 7.8 Hz, 1 H_{arom}), 7.49 (dd, *J* = 7.8, 1.5 Hz, 1 H_{arom}), 7.56 (dd, *J* = 7.8, 1.5 Hz, 1 H_{arom}).

 ^{13}C NMR (50 MHz, CDCl₃): δ = 19.2, 65.5, 101.8, 125.0, 129.5, 130.7, 134.4, 140.5, 143.3.

MS (EI, 70 eV): *m/z* (%) = 73.1 (20), 108.0 (20), 138.0 (35), 171.0 (45), 182.0 (85), 215.0 (100), 230.0 (5, [M]⁺).

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₀H₁₁ClO₂S: 231.0247; found: 231.0224.

Ethyl 2-Chloro-6-(1,3-dioxolan-2-yl)benzoate (2d)

A solution of 1 (13.9 g, 75.0 mmol) in anhyd THF (50 mL) was cooled to -78 °C. Then, a hexane solution of 2.5 M *n*-BuLi (45 mL, 113 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added ethyl chloroformate (36.0 mL, 375 mmol). After workup, the colorless oil obtained was purified by flash chromatography on silica gel (hexane–EtOAc, 10:1) affording the product as a colorless oil; yield: 9.96 g (68%).

IR (film): 3422, 1734, 1445, 1271, 1153, 1124, 1095, 1062, 797 $\rm cm^{-1}.$

¹H NMR (200 MHz, CDCl₃): δ = 1.40 (t, *J* = 7.2 Hz, 3 H, CH₃), 3.99 (s, 4 H, CH₂CH₂), 4.43 (q, *J* = 7.2 Hz, 2 H, OCH₂), 6.03 (s, 1 H, CH), 7.29–7.43 (m, 2 H_{arom}), 7.48 (dd, *J* = 7.1, 1.8 Hz, 1 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 14.0, 61.7, 65.1, 101.3, 124.8, 130.0, 130.2, 131.4, 132.4, 137.9, 166.5.

HRMS (ESI): m/z [M + Na]⁺ calcd for C₁₂H₁₃ClO₄: 279.0400; found: 279.0435.

2-Chloro-6-(1,3-dioxolan-2-yl)-*N*,*N*-dimethylbenzamide (2e)

A solution of 1 (1.84 g, 10.0 mmol) in anhyd THF (20 mL) was cooled to -78 °C. Then, a hexane solution of 2.5 M *n*-BuLi (6.0 mL, 15.0 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added *N*,*N*-dimethylchloroformamide (2.8 mL, 30.0 mmol). After workup, the yellow oil obtained was purified by flash chromatography on silica gel (hexane–EtOAc, 19:1–16:1) affording the product as a yellowish oil, yield: 810 mg (32%).

IR (film): 3421, 1641, 1506, 1399, 1089, 944, 770 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 2.82 (s, 3 H, NCH₃), 3.15 (s, 3 H, NCH₃), 3.89–4.21 (m, 4 H, CH₂CH₂), 5.85 (s, 1 H, CH), 7.28–7.44 (m, 2 H_{arom}), 7.54 (dd, *J* = 7.2, 1.7 Hz, 1 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 34.4, 37.9, 65.3, 65.6, 101.1, 125.0, 129.6, 130.0, 130.5, 135.3, 136.7, 167.1.

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{12}H_{14}CINO_3$: 256.0740; found: 256.0699.

2-Chloro-6-(1,3-dioxolan-2-yl)-N-ethylbenzamide (2f)

A solution of 1 (4.62 g, 25.0 mmol) in anhyd THF (30 mL) was cooled to -78 °C. Then, a hexane solution of 2.5 M *n*-BuLi (18.8 mL, 30.0 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added isocyanatoethane (3.0 mL, 37.5 mmol). The residue obtained after filtration and evaporation of the solvent in vacuo was

crystallized from hexane–EtOAc to afford the product as a white powder; yield: 2.66 g (42%); mp 119–121 °C.

IR (Nujol): 3250, 2970-2835, 1660, 1645 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 1.26 (t, *J* = 7.3 Hz, 3 H, CH₃), 3.45–3.61 (m, 2 H, NCH₂), 3.96–4.18 (m, 4 H, CH₂CH₂), 5.88 (br s, 2 H, NH and CH), 7.29–7.45 (m, 2 H_{arom}), 7.54 (dd, *J* = 7.3, 1.7 Hz, 1 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 9.4, 29.6, 60.3, 95.9, 119.5, 124.8, 125.2, 126.1, 130.9, 131.7, 160.5.

Anal. Calcd for $C_{11}H_{12}CINO_3$: C, 56.37; H, 5.52; N, 5.48. Found: C, 56.00; H, 5.39; N, 5.52.

2-Chloro-6-(1,3-dioxolan-2-yl)-N-propylbenzamide (2g)

A solution of 1 (462 mg, 2.50 mmol) in anhyd THF (7 mL) was cooled to -78 °C. Then, a hexane solution of 1.6 M *n*-BuLi (1.3 mL, 3.00 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added 1-isocyanatopropane (0.71 mL, 7.50 mmol). The residue obtained after filtration and evaporation of the solvent in vacuo was crystallized from hexane–EtOAc to afford the product as a white powder, yield: 295 mg (44%); mp 118–119 °C.

IR (Nujol): 3267, 3072, 1641, 1554, 1307, 801 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 1.00 (t, *J* = 7.2 Hz, 3 H, CH₃), 1.65 (sext, *J* = 7.2 Hz, 2 H, CH₂CH₃), 3.44 (q, *J* = 7.2 Hz, 2 H, NCH₂), 3.90–4.19 (m, 4 H, CH₂CH₂), 5.87 (s, 1 H, CH), 7.28–7.47 (m, 2 H_{arom}), 7.53 (dd, *J* = 7.3, 1.6 Hz, 1 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 11.4, 22.7, 41.7, 65.5, 101.1, 124.7, 130.1, 130.4, 131.2, 136.1, 136.9, 165.8.

Anal. Calcd for C₁₁H₁₂ClNO₃: C, 57.89; H, 5.98; N, 5.19. Found: C, 57.46; H, 5.84; N, 5.46.

2-Chloro-6-(1,3-dioxolan-2-yl)benzamide (2h)

A solution of 1 (1.84 g, 10.0 mmol) in anhyd THF (20 mL) was cooled to -78 °C. Then, a hexane solution of 1.6 M *n*-BuLi (5.2 mL, 12.0 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added trimethylsilyl isocyanate (4.0 mL, 30.0 mmol). The residue obtained after filtration and evaporation of the solvent in vacuo was crystallized from hexane–EtOAc to afford the product as a white powder; yield: 1.21 g (53%); mp 100–103 °C.

IR (Nujol): 3327, 3187, 1668, 1609, 1368, 1091 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 3.96–4.20 (m, 4 H, CH₂CH₂), 5.90–6.20 (br s, 2 H, NH₂), 5.95 (s, 1 H, CH), 7.31–7.47 (m, 2 H_{arom}), 7.54 (dd, *J* = 7.3, 1.8 Hz, 1 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 65.5, 101.1, 124.9, 130.3, 130.6, 131.0, 135.0, 136.8, 167.9.

Anal. Calcd for $C_{11}H_{12}CINO_3$: C, 52.76; H, 4.43; N, 6.15. Found: C, 53.02; H, 4.41; N, 6.26.

2-(3-Chloro-2-fluorophenyl)-1,3-dioxolane (2i)

A solution of 1 (1.84 g, 10.0 mmol) in anhyd THF (20 mL) was cooled to -78 °C. Then, a hexane solution of 1.6 M *n*-BuLi (7.5 mL, 12.0 mmol) was added dropwise and the resulting red-brown solution was stirred for 45 min at this temperature. To this solution was added (PhSO₂)₂NF (3.78 g, 12.0 mmol). After workup, the colorless oil obtained was purified by flash chromatography on silica gel (hexane–EtOAc, 10:1–10:3) affording the product as a colorless oil; yield: 380 mg (19%) (mixed with ~15% of starting material).

IR (film): 3438, 2375, 1470, 1364, 1178, 1098, 944, 787 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 3.96-4.20 (m, 4 H, CH₂CH₂), 6.08 (s, 1 H, CH), 7.03-7.15 (m, 1 H_{arom}), 7.30-7.50 (m, 2 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 65.5, 98.8, 124.4, 124.5, 126.0, 126.1, 131.3, 158.0.

¹⁹F NMR (200 MHz, CDCl₃): $\delta = -122.1$ (t, J = 6.5 Hz).

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3-Chloro-2-iodobenzaldehyde (3a)

To a solution of **2a** (5.82 g, 18.7 mmol) in THF (20 mL) was added aq 1 M HCl (20 mL) and the mixture was refluxed for 2 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (30 mL) and the organic layer was separated, washed with H₂O (2×20 mL) and brine (20 mL), and dried (Na₂SO₄). Filtration and evaporation of the solvent in vacuo gave a solid, which was recrystallized from hexane as colorless needles; yield: 2.70 g (54%); mp 73–75 °C.

IR (Nujol): 3062, 2809, 2370, 1700, 1685, 1409, 1230, 1138, 753 $\rm cm^{-l}.$

¹H NMR (200 MHz, CDCl₃): δ = 7.41 (t, *J* = 7.8 Hz, 1 H_{arom}), 7.67–7.78 (m, 2 H_{arom}), 10.13 (s, 1 H, CHO).

¹³C NMR (50 MHz, CDCl₃): δ = 105.1, 128.0, 129.5, 134.5, 138.1, 140.6, 196.0.

MS (EI, 70 eV): *m*/*z* (%) = 75.1 (65), 110.0 (45), 138.0 (70), 236.9 (20), 265.9 (100, [M]⁺).

Anal. Calcd for C_7H_4 ClIO: C, 31.55; H, 1.51. Found: C, 31.52; H, 1.77.

3-Chloro-2-methylbenzaldehyde (3b)¹⁵

To a solution of $2\dot{b}$ (560 mg, 2.82 mmol) in THF (15 mL) was added aq 1 M HCl (15 mL) and the mixture was refluxed for 4 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (25 mL) and the organic layer was separated, washed with H₂O (2 × 20 mL) and brine (20 mL), and dried (Na₂SO₄). Filtration and evaporation of the solvent in vacuo afforded the product as a yellowish oil; yield: 278 mg (81%) (mixed with ~30% of starting material).

 ^1H NMR (200 MHz, CDCl₃): δ = 2.72 (s, 3 H, CH₃), 7.27–7.37 (m, 1 H_{arom}), 7.56–7.64 (m, 1 H_{arom}), 7.69–7.77 (m, 1 H_{arom}), 10.28 (s, 1 H, CHO).

MS (EI, 70 eV): m/z (%) = 125.1 (60), 153.3 (100), 154 (75, [M]⁺).

3-Chloro-2-(methylsulfanyl)benzaldehyde (3c)¹⁶ [CAS # 1033574-03-6]

To a solution of **2c** (231 mg, 1.0 mmol) in THF (5 mL) was added aq 1 M HCl (5 mL) and the mixture was refluxed for 5 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (10 mL) and the organic layer was separated, washed with H₂O (2 × 10 mL) and brine (10 mL), and dried (Na₂SO₄). The residue obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, 10:1–10:2) to afford the product as a yellowish oil; yield: 103 mg (82%).

¹H NMR (200 MHz, CDCl₃): δ = 2.46 (s, 3 H, SCH₃), 7.35–7.46 (m, 1 H_{arom}), 7.71 (dd, *J* = 7.8, 1.6 Hz, 1 H_{arom}), 7.83 (dd, *J* = 7.8, 1.6 Hz, 1 H_{arom}), 10.77 (s, 1 H, CHO).

MS (EI, 70 eV): m/z (%) = 63.0 (40), 75.0 (80), 108.0 (60), 125.0 (100), 142.9 (40), 157.0 (30), 170.9 (40), 186.0 (100, [M]⁺).

Ethyl 2-Chloro-6-formylbenzoate (3d) [CAS # 1049677-64-6]

To a solution of **2d** (2.57 g, 10.0 mmol) in THF (10 mL) was added aq 1 M HCl (10 mL) and the mixture was refluxed for 4 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (25 mL) and the organic layer was separated, washed with H₂O (2×20 mL) and brine (20 mL), and dried (Na₂SO₄). The residue obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, 6:1) to afford the product as a colorless oil; yield: 1.39 g (65%).

IR (film): 2981, 1777, 1740, 1705, 1458, 1366, 1267, 1067, 956, 932 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 1.42 (t, *J* = 7.2 Hz, 3 H, CH₃), 4.51 (q, *J* = 7.2 Hz, 2 H, OCH₂), 7.54 (t, *J* = 7.7 Hz, 1 H_{arom}), 7.67 (dd, *J* = 7.7, 1.2 Hz, 1 H_{arom}), 7.79 (dd, *J* = 7.7, 1.2 Hz, 1 H_{arom}), 9.97 (s, 1 H, CHO).

MS (EI, 70 eV): *m/z* (%) = 75.0 (90), 110.9 (40), 139.0 (100), 167.0 (95), 183.0 (100), 211.9 (10, [M]⁺).

2-Chloro-6-formyl-N,N-dimethylbenzamide (3e)

To a solution of 2e (810 mg, 3.17 mmol) in THF (15 mL) was added aq 1 M HCl (15 mL) and the mixture was refluxed for 4 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (25 mL) and the organic layer was separated, washed with H₂O (2 × 20 mL) and brine (20 mL), and dried (Na₂SO₄). The residue obtained after filtration and concentration of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, 6:1) to afford the product as a yellowish oil; yield: 102 mg (15%).

IR (film): 3303, 3068, 2977, 1768, 1702, 1626, 1603, 1593, 1455, 1210, 1181, 1079, 909, 748 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 2.83 (s, 3 H, NCH₃), 3.21 (s, 3 H, NCH₃), 7.50 (t, *J* = 7.8 Hz, 1 Harom), 7.67 (dd, *J* = 7.8, 1.1 Hz, 1 H_{arom}), 7.85 (dd, *J* = 7.8, 1.1 Hz, 1 H_{arom}), 9.98 (s, 1 H, CHO).

 ^{13}C NMR (50 MHz, CDCl₃): δ = 34.6, 37.6, 128.9, 130.1, 131.9, 134.2, 134.9, 137.5, 166.0, 189.6.

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{10}H_{10}CINO_2$: 212.0478; found: 212.0516.

7-Chloro-2-ethyl-3-hydroxy-2,3-dihydro-1*H***-isoindol-1-one (4f)** To a solution of **2f** (128 mg, 0.50 mmol) in THF (2.5 mL) was added aq 1 M HCl (2 mL) and the mixture was refluxed for 12 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (2×5 mL) and the organic layer was separated, washed with H₂O (2×5 mL) and brine (5 mL), and dried (Na₂SO₄). The residue obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, 5:1) to afford the product as a white powder; yield: 66 mg (67%); mp 109–112 °C.

IR (Nujol): 3346, 2977, 1683, 1674, 1608, 1461, 1419, 1203, 1064, 780, 732 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 1.21 (t, *J* = 7.2 Hz, 3 H, CH₃), 3.26–3.51 (m, 2 H, NCH₂), 3.53–3.74 (m, 1 H, CH), 5.72 (d, *J* = 11.2 Hz, 1 H, OH), 7.33 (dd, *J* = 7.4, 1.5 Hz, 1 H_{arom}), 7.42–7.57 (m, 2 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 8.1, 28.7, 74.9, 116.9, 122.3, 125.5, 126.1, 127.8, 141.3, 160.0.

Anal. Calcd for $C_{10}H_{10}CINO_2 \cdot 0.5H_2O$: C, 54.43; H, 5.03; N, 6.35. Found: C, 54.51; H, 5.00; N, 6.25.

7-Chloro-3-hydroxy-2,3-dihydro-1*H*-isoindol-1-one (4h)

To a solution of **2h** (114 mg, 0.50 mmol) in THF (2.5 mL) was added aq 1 M HCl (2 mL) and the mixture was refluxed for 12 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (2×5 mL) and the organic layer was separated, washed with H₂O (2×5 mL) and brine (5 mL), and dried (Na₂SO₄). The residue obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, gradient from 2:1 to 1:2) to afford the product as a white powder; yield: 54 mg (59%); mp 164–168 °C.

IR (Nujol): 3345, 3287, 1683, 1465, 1087 cm⁻¹

¹H NMR (400 MHz, DMSO- d_6): $\delta = 5.82$ (d, J = 9.2 Hz, 1 H, CHOH), 6.37 (d, J = 9.2 Hz, 1 H, OH), 7.47–7.53 (m, 2 H, H-5, 7), 7.56–7.63 (m, 1 H, H-6), 8.99 (s, 1 H, NH).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 77.2, 123.2, 127.9, 129.5, 130.9, 133.9, 150.2, 166.5.

HRMS (ESI): $m/z [M + H]^+$ calcd for C₈H₆ClNO₂: 184.0165; found: 184.0217.

Anal. Calcd for C₈H₆ClNO₂: C, 52.34; H, 3.29; N, 7.63. Found: C, 51.57; H, 3.42; N, 6.97.

7-Chloro-3-ethoxy-2-ethyl-2,3-dihydro-1*H***-isoindol-1-one (5f)** To a solution of **2f** (115 mg, 0.45 mmol) in EtOH (3 mL) was added aq 1 M HCl (2 mL) and the mixture was refluxed for 6 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (7 mL) and the organic layer was separated, washed with H₂O (2 × 5 mL) and brine (10 mL), and dried (Na₂SO₄). The residue obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, gradient from 2:1 to 1:1) to afford the products **5f** (60 mg, 56%) as a yellowish oil and **4f** (40 mg, 41%) as a white powder.

IR (film): 3286, 3080, 2977, 1777, 1725, 1691, 1606, 1464–1385, 1202, 1085, 1063, 797 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): $\delta = 1.14$ (t, J = 7.0 Hz, 3 H, CH₃), 1.27 (t, J = 7.3 Hz, 3 H, CH₃), 2.87–3.07 (m, 1 H, NCH), 3.11–3.26 (m, 1 H, NCH), 3.34 (sext, J = 7.3 Hz, 1 H, OCH), 3.82 (sext, J = 7.3 Hz, 1 H, OCH), 3.82 (sext, J = 7.3 Hz, 1 H, OCH), 5.85 (s, 1 H, CH), 7.40–7.55 (m, 3 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 8.5, 10.0, 29.8, 52.8, 79.5, 117.6, 123.4, 127.8, 127.9, 128.0, 139.6, 160.5.

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₂H₁₄NClO₂: 240.0791; found: 240.0773.

7-Chloro-3-ethoxy-2,3-dihydro-1H-isoindol-1-one (5h)

To a solution of **2h** (114 mg, 0.50 mmol) in EtOH (3 mL) was added aq 1 M HCl (2 mL) and the mixture was refluxed for 6 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (7 mL) and the organic layer was separated, washed with H₂O (2 × 5 mL) and brine (10 mL), and dried (Na₂SO₄). The colorless oil obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, gradient from 2:1 to 1:1) to afford products **5h** (17 mg, 16%) and **4h** (20 mg, 22%) as white powders.

Mp 116-120 °C.

IR (Nujol): 3284, 1720, 1606, 1463, 1202, 1120, 1091, 768 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 1.21 (t, J = 7.1 Hz, 3 H, CH₃), 3.31–3.39 (m, 1 H, OCH), 3.51–3.60 (m, 1 H, OCH), 5.92 (s, 1 H, CHN), 7.19 (s, 1 H, NH), 7.43–7.56 (m, 3 H_{arom}).

¹³C NMR (50 MHz, CDCl₃): δ = 15.2, 60.7, 82.5, 122.3, 127.9, 131.5, 131.6, 133.4, 145.8, 167.8.

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{10}H_{10}CINO_2$: 212.0478; found: 212.0511.

Anal. Calcd for $C_{10}H_{10}CINO_2$: C, 56.75; H, 4.76; N, 6.62. Found: C, 56.21; H, 4.52; N, 6.17.

7-Chloro-2-ethyl-3-methoxy-2,3-dihydro-1*H*-isoindol-1-one (6f)

Compound **2f** (2.00 g, 7.80 mmol) was dissolved in 3.75 M AcCl– MeOH (15 mL) and the mixture was refluxed for 8 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (2×35 mL) and the organic layer was separated, washed with H₂O (2×35 mL) and brine (35 mL), and dried (Na₂SO₄). The yellow oil obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, gradient from 5:1 to 5:3) to afford the product **6f** as a white powder; yield: 1.34 g (76%); mp 80–82 °C.

IR (Nujol): 2960–2840, 1780, 1460, 1380, 1355, 1310, 1070, 930 $\rm cm^{-l}.$

¹H NMR (200 MHz, CDCl₃): δ = 1.32 (t, *J* = 7.1 Hz, 3 H, CH₃), 2.90 (s, 3 H, OCH₃), 3.25–3.35 (m, 1 H, NCH), 3.75–3.90 (m, 1 H, NCH), 5.85 (s, 1 H, CH), 7.45–7.55 (m, 3 H_{arom}).

¹³C NMR (100 MHz, CDCl₃): δ = 13.3, 34.6, 49.2, 84.9, 122.0, 129.3, 131.5, 131.7, 132.8, 143.0, 160.5.

Anal. Calcd for C₁₁H₁₂ClO₂: C, 58.55; H, 5.36; N, 6.21. Found: C, 58.36; H, 5.13; N, 5.91.

7-Chloro-3-methoxy-2-propyl-2,3-dihydro-1*H*-isoindol-1-one (6g)

Compound **2g** (94 mg, 0.35 mmol) was dissolved in 3.75 M AcCl– MeOH (2 mL) and the mixture was refluxed for 12 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (2×5 mL) and the organic layer was separated, washed with H₂O (2×5 mL) and brine (5 mL), and dried (Na₂SO₄). The residue obtained after filtration and evaporation of the solvent in vacuo was purified by flash chromatography on silica gel (hexane–EtOAc, 2:1) to give the product as a colorless oil; yield: 76 mg (90%).

IR (film): 3078, 2934, 2829, 1708, 1606, 1462, 1407, 1371, 1200, 1088, 1074, 807 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): $\delta = 0.97$ (t, J = 7.4 Hz, 3 H, CH₃CH₂), 1.50–1.82 (m, 2 H, CH₂CH₃), 2.89 (s, 3 H, OCH₃), 3.11–3.32 (m, 1 H, NCH₂), 3.63–3.83 (m, 1 H, NCH₂), 5.85 (s, 1 H, CHO), 7.40– 7.57 (m, 3 H_{arom}).

¹³C NMR (100 MHz, CDCl₃): δ = 11.6, 21.4, 41.4, 85.3, 110.1, 122.04, 129.2, 131.5, 131.8, 132.8, 143.0, 165.5.

HRMS (ESI): m/z [M + H]⁺ calcd for C₁₂H₁₄ClNO₂: 240.0791; found: 240.0777.

7-Chloro-3-ethoxy-2-benzofuran-1(3H)-one (7) and 7-Chloro-3-hydroxy-2-benzofuran-1(3H)-one (8) 17

To a solution of 2-chloro-6-(1,3-dioxolan-2-yl)benzoate (2d; 2.40 g, 9.30 mmol) in EtOH (10 mL) was added aq 1 M HCl (10 mL) and the mixture was refluxed for 12 h. After cooling to r.t., the mixture was neutralized with sat. aq NaHCO₃. To the mixture was added EtOAc (25 mL) and the organic layer was separated, washed with H_2O (2 × 20 mL) and brine (20 mL), and dried (Na₂SO₄). Filtration and concentration in vacuo gave a residue, which was washed with MTBE (7 mL) and filtered to afford 8. The filtrate was concentrated and purified by flash chromatography on silica gel (hexane–EtOAc, gradient from 10:1 to 10:3) to furnish a further amount of 8 and the product 7 as white powders.

7

Yield: 880 mg (45%); mp 102-107 °C.

IR (Nujol): 2960–2840, 1780, 1380, 1355, 1310, 1070, 930 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 1.32 (t, *J* = 7.1 Hz, 3 H, CH₃CH₂), 3.80–4.05 (m, 2 H, CH₂CH₃), 6.31 (s, 1 H, CHO), 7.45–7.68 (m, 3 H, H-4, 5, 6).

¹³C NMR (50 MHz, CDCl₃): δ = 15.1, 66.1, 100.7, 121.9, 132.2, 133.1, 135.3, 160.1.

Anal. Calcd for $C_{10}H_9ClO_3$: C, 56.49; H, 4.27. Found: C, 56.37; H, 4.25.

8 [CAS # 70097-44-8]

Yield: 350 mg (20%); mp 112–116 °C.

IR (Nujol): 3380, 2920, 2850, 1750, 1735, 1080 cm⁻¹.

¹H NMR (200 MHz, DMSO- d_6): $\delta = 6.62$ (d, J = 8.6 Hz, 1 H, CHO), 7.60–7.85 (m, 3 H), 8.23 (d, J = 8.6 Hz, 1 H, OH).

¹³C NMR (50 MHz, DMSO- d_6): δ = 96.7, 122.7, 130.6, 131.5, 136.1, 137.6, 150.0, 165.4.

Anal. Calcd for $C_8H_5ClO_3$: C, 52.06; H, 2.73. Found: C, 51.74; H, 2.46.

7-Chloro-2-methyl-3-(methylamino)-2,3-dihydro-1*H*-isoindol-1-one (9)

A mixture of **8** (92 mg, 0.50 mmol) and 40% MeNH₂ in H₂O (3 mL) was heated to 80 $^{\circ}$ C and stirred for 12 h. Concentration in vacuo

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gave a yellow oil, which was purified by flash chromatography on silica gel (CH₂Cl₂–MeOH, 25:1) to give the product as a white powder; yield: 67 mg (64%); mp 112–114 °C.

IR (Nujol): 3340, 3075, 2800, 1709, 1673, 1607, 1462, 1421, 1393, 1199, 1178, 1027, 937, 750 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 1.99 (s, 3 H, CH₃NH), 2.08 (br s, 1 H, NH), 3.05 (s, 3 H, CH₃N), 5.22 (s, 1 H, CHNH), 7.37–7.53 (m, 3 H, H-4, 5, 6).

¹³C NMR (100 MHz, CDCl₃): δ = 26.0, 27.1, 73.8, 121.6, 129.2, 130.7, 131.0, 132.3, 145.3, 165.6.

Anal. Calcd for $C_{10}H_{11}CIN_2O$: C, 57.02; H, 5.26; N, 13.30. Found: C, 56.86; H, 5.12; N, 12.86.

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{10}H_{11}CIN_2O$: 211.0638; found: 211.0698.

7-Chloro-3-methoxy-2-methyl-2,3-dihydro-1*H*-isoindol-1-one (10) and 7-Chloro-3-hydroxy-2-methyl-2,3-dihydro-1*H*-isoin-dol-1-one (11)

A mixture of $\hat{8}$ (92 mg, 0.50 mmol) and 2 M MeNH₂ in MeOH (3 mL) was warmed to 80 °C and stirred for 12 h. Concentration in vacuo gave a brown oil, which was purified by flash chromatography on silica gel (hexane–EtOAc, gradient from 5:1 to 3:1). Three fractions were collected.

10

Compound **10** was isolated from the first fraction; yield: 7 mg (7%); colorless oil.

IR (film): 3290, 3090, 2980, 1788, 1737, 1681, 1616, 1452, 1392, 1202, 1074, 784 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 2.91 (s, 3 H, CH₃N), 3.08 (s, 3 H, CH₃O), 5.73 (s, 1 H, CHO), 7.40–7.56 (m, 3 H, H-5, 6, 7).

¹³C NMR (50 MHz, CDCl₃): δ = 26.5, 49.2, 86.8, 121.9, 129.1, 131.4, 131.7, 132.8, 142.8, 165.4.

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{10}H_{10}CINO_2$: 212.0478; found: 212.0491.

11

Compound **11** was isolated from the second fraction; yield: 3 mg (2%); white powder; mp 133–135 $^{\circ}$ C.

IR (Nujol): 3290, 2920, 2840, 1685, 1610, 1080 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 3.01 (s, 3 H, CH₃N), 3.55 (d, *J* = 12.0 Hz, 1 H, OH), 5.59 (d, *J* = 12.0 Hz, 1 H, CHOH), 7.31 (dd, *J* = 7.5, 1.5 Hz, 1 H, H-6), 7.46 (dd, *J* = 7.5, 7.3 Hz, 1 H, H-5), 7.53 (dd, *J* = 7.3, 1.5 Hz, 1 H, H-4).

¹³C NMR (50 MHz, CDCl₃): δ = 26.2, 82.4, 122.0, 127.5, 130.9, 131.4, 133.1, 146.2, 165.1.

HRMS (ESI): $m/z [M + H]^+$ calcd for C₉H₈ClNO₂: 198.0322; found: 198.0361.

9

Compound **9** was isolated from the third fraction; yield: 10 mg (9%); white powder.

4-Chloro-1-hydroxy-*N*-methyl-3-oxo-1,3-dihydro-2*H*-isoin-dole-2-carboxamide (12)

A suspension of **8** (92 mg, 0.50 mmol) and *N*-methylurea (41 mg, 0.55 mmol) in THF was refluxed for 12 h. The product precipitated after cooling to r.t. as a white powder; yield: 19 mg (18%); mp 178–181 °C.

IR (Nujol): 3355, 3283, 1773, 1653, 919 cm⁻¹.

¹H NMR (400 MHz, DMSO- d_6): $\delta = 2.61$ (d, J = 4.3 Hz, 3 H, CH₃NH), 6.24 (d, J = 4.3 Hz, 1 H, NH), 6.94 (d, J = 9.9 Hz, 1 H, CHOH), 7.47 (d, J = 9.9 Hz, 1 H, OH), 7.56 (dd, J = 7.6, 0.8 Hz, 1 H, H-5), 7.66 (dd, J = 8.0, 7.6 Hz, 1 H, H-6), 7.76 (dd, J = 8.0, 0.8 Hz, 1 H, H-7).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 26.8, 83.0, 122.8, 124.0, 131.0, 131.6, 136.2, 149.7, 157.3, 166.3.

HRMS (ESI): m/z [M + H]⁺ calcd for $C_{10}H_9ClN_2O_3$: 241.0380; found: 241.0421.

2-Acetyl-3-chlorobenzaldehyde (13)

A solution of 7-chloro-3-ethoxy-2-benzofuran-1(3*H*)-one (7; 75 mg, 0.35 mmol) in anhyd THF (5 mL) was cooled to -78 °C. Then, a hexane solution of 1.6 M MeLi (0.24 mL, 0.39 mmol) was added dropwise and the resulting orange solution was warmed to -60 °C and aq 1 M HCl (1 mL) was added. The mixture was warmed to r.t. and sat. aq NaHCO₃ (3 mL) was added. To the mixture was added EtOAc (7 mL) and the organic layer was separated, washed with H₂O (2 × 5 mL) and brine (10 mL), and dried (Na₂SO₄). Filtration and evaporation of the solvent in vacuo gave a red oil, which was purified by flash chromatography on silica gel (hexane–EtOAc, 3:1) to afford the product as a yellowish oil; yield: 59 mg (92%).

IR (film): 3385, 1771, 1750, 1717, 1699, 1462, 1243, 1186, 1150, 1031, 835, 765 cm⁻¹.

¹H NMR (200 MHz, CDCl₃): δ = 2.59 (s, 3 H, CH₃), 7.54 (dd, *J* = 8.1, 7.4 Hz, 1 H, H-5), 7.66 (dd, *J* = 8.1, 1.2 Hz, 1 H, H-4), 7.79 (dd, *J* = 7.4, 1.2 Hz, 1 H, H-6), 9.93 (s, 1 H, CHO).

¹³C NMR (50 MHz, CDCl₃): δ = 31.1, 130.2, 130.6, 131.2, 134.3, 135.2, 141.5, 190.0, 201.8.

HRMS (ESI): $m/z [M + H]^+$ calcd for C₉H₇ClO₂: 183.0213; found: 183.0240.

Supporting Information for this article is available online at http://www.thieme-connect.com/ejournals/toc/synthesis.

References

(a) Hardcastle, I. R.; Ahmed, S. U.; Atkins, H.; Calvert, A. H.; Curtin, N. J.; Farnie, G.; Golding, B. T.; Griffin, R. J.; Guyenne, S.; Hutton, C.; Källblad, P.; Kemp, S. J.; Kitching, M. S.; Newell, D. R.; Norbedo, S.; Northen, J. S.; Reid, R. J.; Saravanan, K.; Willems, H. M. G.; Lunec, J. *Bioorg. Med. Chem. Lett.* 2005, *15*, 1515. (b) Wacker, D. A.; Varnes, J. G.; Malmstrom, S. E.; Cao, X.; Hung, C.-P.; Ung, T.; Wu, G.; Zhang, G.; Zuvich, E.; Thomas, M. A.; Keim, W. J.; Cullen, M. J.; Rohrbach, K. W.; Qu, Q.; Narayanan, R.; Rossi, K.; Janovitz, E.; Lehman-McKeeman, L.; Malley, M. F.; Devenny, J.; Pelleymounter, M. A.; Miller, K. J.; Robl, J. A. *J. Med. Chem.* 2007, *50*, 1365. (c) Wang, W.; Cha, X.-X.;

Reiner, J.; Gao, Y.; Qiao, H.-L.; Shen, J.-X.; Chang, J.-B. *Eur. J. Med. Chem.* **2010**, *45*, 1941. (d) Lüthy, C.; Zondler, H.; Rapold, T.; Seifert, G.; Urwyler, B.; Heinis, T.; Steinrücken, H. C.; Allen, J. *Pest. Manag. Sci.* **2001**, *57*, 205.

- (2) Reviews on *ortho*-lithiation: (a) Beak, P.; Meyers, A. I. Acc. Chem. Res. 1986, 19, 356. (b) Snieckus, V. Chem. Rev. 1990, 90, 879. (c) Epsztajn, J.; Jozwiak, A.; Szczesniak, A. K. Curr. Org. Chem. 2006, 10, 1817.
- (3) Comins, D. L.; Brown, J. D. J. Org. Chem. 1984, 49, 1078.
- (4) Flippin, L. A.; Muchowski, J. M.; Carter, D. S. J. Org. Chem. 1993, 58, 2463.
- (5) (a) Plaumann, H. P.; Keay, B. A.; Rodrigo, R. *Tetrahedron Lett.* **1979**, *20*, 4921. (b) Napolitano, E.; Giannone, E.; Fiaschi, R.; Marsili, A. *J. Org. Chem.* **1983**, *48*, 3653.
 (c) Costa, P. R. R.; da Silva, A. J. M.; Vasconcellos, M. L. A. A.; Lopes, C. C.; Lopes, R. S. C. *Synlett* **1996**, 783.
- (6) Huerta, F. F.; Gomez, C.; Yus, M. *Tetrahedron* **1999**, *55*, 4043.
- (7) (a) Pansegrau, P. D.; Rieker, W. F.; Meyers, A. I. J. Am. Chem. Soc. 1988, 110, 7178. (b) Caster, K. C.; Keck, C. G.; Walls, R. D. J. Org. Chem. 2001, 66, 2932. (c) Gohier, F.; Mortier, J. J. Org. Chem. 2003, 68, 2030. (d) Gohier, F.; Castanet, A.-S.; Mortier, J. Org. Lett. 2003, 5, 1919.
- (8) Flitsch, W. Chem. Ber. 1970, 103, 3205.
- (9) Valters, R. E. Russ. Chem. Rev. 1974, 43, 665.
- (10) Valtere, S. P.; Zariņa, Z. E.; Karlivāns, G. A.; Valters, R. E. Latv. PSR Zināt. Akad. Vēstis, Ķīm. Sēr. 1978, 575; Chem. Abstr. 1979, 90, 54316j.
- (11) Bowden, K.; Hiscocks, S. P.; Perjessy, A. J. Chem. Soc., Perkin Trans. 2 **1998**, 291.
- (12) Wheeler, D. D.; Young, D. C.; Erley, D. S. J. Org. Chem. 1957, 22, 547.
- (13) Kagan, J. J. Org Chem. 1967, 32, 4060.
- (14) The crystallographic data are deposited with the Cambridge Crystallographic Data Centre as Supplementary Publication Number CCDC 864340 (4f) and 864341 (6f), respectively. Copies of the data can be obtained, free of charge at www.ccdc.cam.ac.uk/conts/retrieving.html or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44(1223)336033, E-mail: deposit@ccdc.cam.ac.uk.
- (15) Mitchell, R. H.; Lai, Y.-H. J. Org. Chem. 1984, 49, 2534.
- (16) Lundstedt, T.; Seifert, E.; Lek, P.; Boman, A. WO 2008071980, **2008**; *Chem. Abstr.* **2008**, *149*, 79349.
- (17) Gohier, F.; Castanet, A.-S.; Mortier, J. Synth. Commun. 2005, 35, 799.