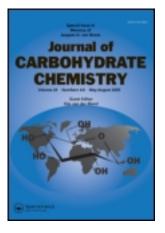
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Dirk Michalik & Klaus Peseke ^a Fachbereich Chemie, Universität Rostock, D-18051 Rostock, Germany Published online: 27 Feb 2008.

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SYNTHESES OF PYRAZOLE ISO-C-NUCLEOSIDES ☆

Dirk Michalik and Klaus Peseke*

Fachbereich Chemie, Universität Rostock, D-18051 Rostock, Germany

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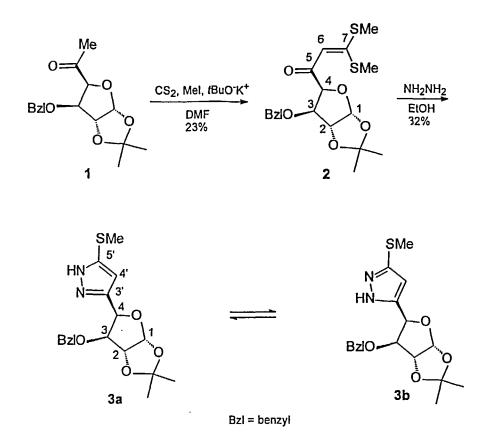
ABSTRACT

3-O-Benzyl-6-deoxy-1,2-O-isopropylidene- α -D-*xylo*-hexofuranos-5-ulose (1) and 3,6-dideoxy-1,2-O-isopropylidene- α -D-*glycero*-hex-3-enofuranos-5-ulose (6) reacted with carbon disulfide and methyl iodide under basic conditions to give the α oxoketene-*S*,*S*-acetals 2 and 7, respectively. Treatment of 2 and 7 with hydrazine hydrate yielded the pyrazole derivatives 3 and 8, respectively.

INTRODUCTION

Naturally occurring C-nucleosides such as pyrazofurin, showdomycin, oxazinomycin, and formycin B are important, in part, due to their antibacterial, antiviral, and antitumor properties.¹⁻³ The development of strategies for the formation of C-nucleoside analogues is a topic of current interest in organic synthesis.⁴ In recent years we have reported the preparation of 4,6-*O*-benzylidene-3(2)-[bis(methyl-thio)methylene]-3(2)-deoxy- α -D-*erythro*-hexopyranosid-2(3)-uloses by reaction of carbanions generated from corresponding deoxyuloses with carbon disulfide.^{5,6} These C-branched monosaccharides with push-pull functionality could be used as

th Dedicated to Professor Dr. *Ralf Miethchen* on the occasion of his 60th birthday.



Scheme 1

precursors for the syntheses of "inversed" C-nucleoside analogues (cf. 7,8). We describe herein the synthesis of such compounds with a pyrazole moiety based on tetrofuranos-4-yl and tetrenofuranos-4-yl substituted α -oxoketene-*S*,*S*-acetals from a de-oxyhexofuranosulose and the corresponding α , β -unsaturated ulose derivative.

RESULTS AND DISCUSSION

The 3-O-benzylated ulose 1 was synthesized starting from D-glucose.⁹⁻¹³ The preparation of the tetrofuranosyl substituted α -oxoketene-S,S-acetal 2 was performed using carbon disulfide, an excess of methyl iodide and potassium *tert*-butoxide as base in *N*,*N*-dimethylformamide and was carried out in a one-pot reaction without isolation

	C=0	-HC=	$=C(SMe)_2$	
2	190.3	109.9	166.6	
7	176.7	108.6	170.0	

Table 1. ¹³C chemical shifts of the sp²-carbon atoms of compounds 2 and 7 (in ppm).

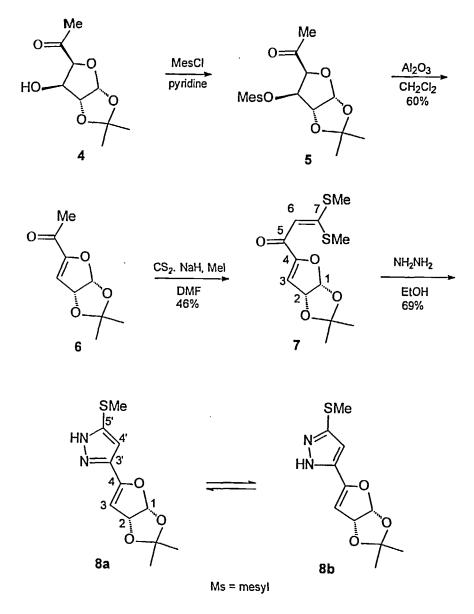
of the intermediate dithiolate (Scheme 1). Compound 2 could be isolated in 23% yield as a pale yellow syrup. The push-pull character of the α -oxoketene-*S*,*S*-acetal was confirmed by this yellow color and the characteristic alternating chemical shifts of the sp²-carbon atoms ¹⁴ (Table 1).

Formation of different unisolated side products was the reason for a low yield. The main side reaction was the β -elimination to form the corresponding α , β -unsaturated ulose 6 (Scheme 2). Another side reaction was caused by utilization of an excess of methyl iodide resulting in an additional methylation at C-4 and C-6. However, the yield of the desired tetrofuranosyl substituted α -oxoketene-*S*,*S*-acetal 2 is much lower if only a quantitative amount of methyl iodide was used.

 α -Oxoketen-*S*,*S*-acetals react easily with hydrazines to form pyrazoles.¹⁵ One thiomethyl group is displaced by a substitution reaction and ring closure occurs through attack of the hydrazino group on the carbonyl C-atom with elimination of water. The reaction of tetrofuranosyl substituted α -oxoketene-*S*,*S*-acetal **2** was performed in ethanol at room temperature using hydrazine hydrate. After 48 h the TLC indicated the absence of starting material. The *iso*-C-nuleoside **3** could be isolated in a yield of 32%.

The tetrofuranos-4-yl substituted pyrazole 3 can exist in tautomeric forms 3a and 3b but a decision between them was not possible. Due to the fast NH-proton exchange, the atoms C-3' and C-5' appear as broad signals in the ¹³C NMR spectrum. In compound 3 which can be seen as a C-nucleoside analogue, the furanose is linked to the heterocycle *via* a C-C single bond not at the anomeric position but at C-4.

Taking advantage of the easily occuring β -elimination of the ulose 1 to give the α , β -unsaturated ulose 6, we carried out the preparation of the corresponding α -oxoketene-*S*,*S*-acetal 7. In order to obtain the unsaturated ulose 6 in high yield we started from 6-deoxy-1,2-*O*-isopropylidene-3-*O*-mesyl- α -D-*xylo*-hexofuranos-5-ulose (5) pre-



pared by mesylation of the 6-deoxy-1,2-*O*-isopropylidene- α -D-*xylo*-hexofuranos-5-ulose (4). Compound 4 was obtained by a procedure described by Ohle et al.¹⁶⁻¹⁸ starting from D-glucose. Treatment of 5 with basic aluminium oxide in dichloromethane afforded the expected unsaturated sugar 6 in a yield of 60% (Scheme 2).

The reaction of the α,β -unsaturated ulose 6 with carbon disulfide and an excess of methyl iodide was performed in *N*,*N*-dimethylformamide using sodium hydride. The α -oxoketene-*S*,*S*-acetal 7 could be isolated in 46% yield as a pale yellow crystalline compound. The yield was much higher compared to the formation of **2** because no elimination reaction could occur.

The treatment of α -oxoketene-*S*,*S*-acetal 7 with hydrazine hydrate in ethanol at room temperature for 24 h yielded the moderately stable pyrazole 8 in 69% yield. Again, the assignment between the tautomeric forms 8a and 8b was impossible.

EXPERIMENTAL

General procedures. Melting points were measured with a Boëtius apparatus and are corrected. Specific rotations were determined with a Polar LµP polarimeter. IR spectra were recorded with a Nicolet 205 FT-IR spectrometer. ¹H NMR (300.13 and 250.13 MHz, respectively) and ¹³C NMR (62.90 MHz) spectra were recorded on Bruker instruments ARX 300 and AC 250, respectively, with CDCl₃ as solvent. The calibration of spectra was carried out by means of solvent peaks (CDCl₃: δ ¹H= 7.25; δ ¹³C= 77.0). The ¹³C NMR signals were assigned by DEPT and/or ¹³C ,¹H correlations. The mass spectra were recorded on an AMD 402/3 spectrometer (AMD Intectra GmbH). For chromatography Merck silica gel 60 (230-400 mesh) was used. TLC was performed on silica gel 60 GF₂₅₄ (Merck) with detection by using UV-light and charring with sulfuric acid. Elemental analysis were performed on a Leco CHNS-932 instrument.

3-O- Benzyl- 6-deoxy- 1,2-O-isopropylidene- 7-S-methyl- 7-C-methylthio-7thio- α -D-xylo-hept-6-enofuranos-5-ulose (2). A solution of 3-O-benzyl-6-deoxy-1,2-O-isopropylidene- α -D-xylo-hexofuranos-5-ulose (1,⁹⁻¹³ 0.8 g, 2.7 mmol), carbon disulfide (0.3 mL, 5.0 mmol) and methyl iodide (0.6 mL, 9.6 mmol) in N,N-dimethylformamide (15 mL) was cooled to 0 °C. Then potassium *tert*-butoxide (1.2 g, 11.1 mmol) was added. The mixture was stirred for 20 min at 0 °C and 40 min at room temperature, then poured into ice water and extracted with chloroform. The combined organic layers were washed with water, dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (toluene/ethyl acetate 15:1) to give **2** as a yellow syrup (0.25 g, 23%); $R_f = 0.33$ (toluene/ethyl acetate 7:1); $[\alpha]_D^{24}$ -92.7 (*c* 1.0, chloroform); IR (capillar) 1639 (CO) cm⁻¹; ¹H NMR (250.13 MHz, CDCl₃): δ 7.40-7.10 (m, 5H, Ph); 6.54 (s, 1H, H-6); 6.05 (d, 1H, $J_{1,2} = 3.7$ Hz, H-1); 4.70 (d, 1H, $J_{3,4} = 3.4$ Hz, H-4); 4.55 (d, 1H, $J_{2,3} = 0$ Hz, H-2); 4.50 (s, 2H, CH_2 Ph); 4.31 (d, 1H, H-3); 2.49, 2.41 (2s, 6H, 2 SCH₃); 1.47, 1.31 (2s, 6H, C(CH₃)₂); ¹³C NMR (62.90 MHz, CDCl₃): δ 190.3 (CO); 166.6 (*C*(SCH₃)₂); 137.4 (i-Ph); 128.3, 127.8 (*o*-, *m*-Ph); 127.8 (*p*-Ph); 112.2 (*C*(CH₃)₂); 109.9 (C-6); 105.7 (C-1); 84.8 (C-4); 83.6 (C-3); 82.6 (C-2); 73.0 (CH₂); 26.9, 26.4 (C(*C*H₃)₂); 17.1, 14.9 (2 SCH₃). MS, e.i. (*m*/*z*): 397 [M+H]⁺. HRMS: Calcd for C₁₉H₂₄O₅S₂: 396.10651. Found: 396.10764.

3-*O*-Benzyl- 1,2-*O*-isopropylidene- 4*C*- (5-methylthiopyrazol-3-yl)- α-D-xylotetrofuranose (3). A solution of 2 (96 mg, 0.24 mmol) and hydrazine hydrate (0.3 mL, 6 mmol) in ethanol (15 mL) was stirred for 2 days at room temperature and then concentrated. The residue was purified by column chromatography (toluene/ethyl acetate 4:1) to yield 3 as a colourless syrup (28 mg, 32%); $R_f = 0.26$ (toluene/ethyl acetate 2:1); $[\alpha]_D^{24}$ -52.5 (*c* 1.0, chloroform); ¹H NMR (250.13 MHz, CDCl₃): δ 7.30-7.10 (m, 5H, Ph); 6.25 (s, 1H, H-4'); 6.01 (d, 1H, $J_{1,2} = 3.8$ Hz, H-1); 5.26 (d, 1H, $J_{3,4} =$ 2.9 Hz, H-4); 4.69 (d, 1H, $J_{2,3} = 0$ Hz, H-2); 4.47 (d, 1H, $J_{CH(a),CH(b)} = 11.6$ Hz, $CH_2Ph(a)$); 4.30 (d, 1H, $CH_2Ph(br)$); 4.01 (d, 1H, H-3); 2.47 (1s, 3H, SCH₃); 1.52. 1.34 (2s, 6H, C(CH₃)₂); ¹³C NMR (62.90 MHz, CDCl₃): δ 145.2, 140.8 (2 x br, C-3', C-5'); 136.7 (i-Ph); 128.5, 127.8 (o-, m-Ph); 128.1 (p-Ph); 112.0 (*C*(CH₃)₂); 106.1 (C-4'); 104.6 (C-1); 83.1 (C-3); 82.8 (C-2); 75.0 (C-4); 72.5 (CH₂); 26.8, 26.2 (C(CH₃)₂); 16.9 (SCH₃). MS, e.i. (*m*/*z*): 362 [M]⁺. HRMS: Calcd for C₁₈H₂₂N₂O₄S: 362.13004. Found: 362.12882.

6-Deoxy-1,2-O-isopropylidene-3-O-mesyl- α -D-xylo-hexofuranos-5-ulose (5). A solution of 6-deoxy-1,2-O-isopropylidene- α -D-xylo-hexofuranosid-5-ulose (4,¹⁶⁻¹⁸ 1.14 g, 5.76 mmol) in pyridine (20 mL) was cooled to 0 °C. Then methanesulfonyl chloride (0.52 mL, 6.72 mmol) was added. The mixture was stirred for 30 min at 0 °C and afterwards for an additional 24 h at room temperature, then poured into ice water and extracted with chloroform. The combined organic layers were washed with water, 10% hydrochloric acid, NaHCO₃ solution and water, dried (Na₂SO₄) and concentrated. The residue was crystallized from ethanol and yielded **5** as colourless needles (1.23 g, 76%): mp 83-85 °C; $R_f = 0.61$ (toluene/ethyl acetate 1:1); $[\alpha]_D^{23}$ -100.8 (*c* 1.0, CHCl₃), IR (KBr) 1737 (CO) cm⁻¹; ¹H NMR (250.13 MHz, CDCl₃): δ 6.08 (d, 1H, $J_{1,2} = 3.7$ Hz, H-1); 5.14 (d, 1H, $J_{3,4} = 3.4$ Hz, H-4); 4.81 (d, 1H, $J_{2,3} = 0$ Hz, H-2); 4.71 (d, 1H, H-3); 2.96 (s, 3H, SO₂CH₃); 2.29 (s, 3H, CH₃); 1.48, 1.32 (2s, 6H, C(CH₃)₂); ¹³C NMR (62.90 MHz, CDCl₃): δ 205.2 (CO); 113.1 (*C*(CH₃)₂); 105.5 (C-1); 83.8 (C-3); 83.1 (C-2); 83.0 (C-4); 37.9 (SO₂CH₃); 28.1 (CH₃); 26.7, 26.2 (C(*C*(H₃)₂). MS, e.i. (*m*/*z*): 265 [M-CH₃]⁺.

Anal. Calcd for C₁₀H₁₆O₇S (280.3): C, 42.85; H, 5.75; S, 11.44. Found: C, 42.80; H, 5.74; N; 7.47; S; 11.50.

3,6-Dideoxy-1,2-*O*-isopropylidene-α-D-*glycero*-hex-3-enofuranos-5-ulose (6). A mixture of 5 (0.4 g, 1.43 mmol) and aluminium oxide (11 g, activity AI) in dichloromethane (30 mL) was stirred vigorously for 30 min, then filtered through Celite, concentrated to yield 5 as a colourless syrup (0.16 g, 60%); $R_f = 0.32$ (toluene/ethyl acetate 7:1); $[\alpha]_D^{23}$ +24.9 (*c* 1.0, CHCl₃); ¹H NMR (300.13 MHz, CDCl₃): δ 6.15 (d, 1H, $J_{1,2} = 5.3$ Hz, H-1); 5.99 (d, 1H, $J_{2,3} = 2.7$ Hz, H-3); 5.36 (dd, 1H, H-2); 2.33 (s, 3H, CH₃); 1.44, 1.42 (2s, 6H, C(CH₃)₂); ¹³C NMR (62.90 MHz, CDCl₃): δ 190.9 (C-5); 156.0 (C-4); 113.4 (*C*(CH₃)₂); 109.1 (C-3); 106.6 (C-1); 83.2 (C-2); 28.2 (CH₃); 27.9, 27.6 (C(*C*H₃)₂). MS, e.i. (*m*/*z*): 184 [M]⁺. HRMS: Calcd for C₉H₁₂O₄: 184.07356. Found: 184.06958.

3,6-Dideoxy- 1,2-O-isopropylidene- 7-S-methyl- 7-C-methylthio- 7-thio-α-Dglycero-hepto-3,6-dienofuranos-5-ulose (7) A suspension of sodium hydride (80%, 37 mg, 1.1 mmol) in heptane (2 mL) was stirred for 10 min. The solvent was decanted after the sodium hydride had settled. Then toluene (1 mL) was added and stirred for another 5 min. Afterwards a solution of 6 (100 mg, 0.54 mmol), carbon disulfide (0.065 mL, 1.09 mmol) and methyl iodide 3.47 mmol) (2.17 mL, in N,N-dimethylformamide (5 mL) was added. The mixture was stirred for 1 h, then poured into ice water and extracted with chloroform. The combined organic layers were washed with water, dried (Na₂SO₄) and concentrated. The residue was purified by chromatography (toluene/ethyl acetate 7:1). Recrystallization from column

ethanol/water yielded 7 as pale yellow crystals (72 mg, 46%): mp 96 °C; $R_f = 0.37$ (toluene/ethyl acetate 7:1); $[\alpha]_D^{23}$ +25.1 (*c* 0.5, CHCl₃), IR (KBr) 1602 (CO) cm⁻¹; ¹H NMR (250.13 MHz, CDCl₃): δ 6.25 (s, 1H, H-6); 6.15 (d, 1H, $J_{1,2} = 5.2$ Hz, H-1); 5.97 (d, 1H, $J_{2,3} = 2.5$ Hz, H-3); 5.35 (dd, 1H, H-2); 2.51, 2.50 (2s, 6H, 2 SCH₃); 1.44, 1.43 (2s, 6H, C(CH₃)₂); ¹³C NMR (62.90 MHz, CDCl₃): δ 176.7 (C-5); 170.0 (*C*(SCH₃)₂); 157.5 (C-4); 112.7 (*C*(CH₃)₂); 108.6 (C-6); 106.8 (C-1); 106.4 (C-3); 82.2 (C-2); 28.0, 27.7 C(CH₃)₂); 17.3, 15.1 (2 SCH₃). MS, e.i. (*m*/*z*): 288 [M]⁺. HRMS: Calcd for C₁₂H₁₆O₄S₂: 288.04901. Found: 288.04762.

3-Deoxy-1,2-*O*-isopropylidene-4-*C*-(5-methylthiopyrazol-3-yl)-α-D-glycerotetr-3-enofuranose (8). A solution of 7 (15 mg, 0.052 mmol) and hydrazine hydrate (3 mL, 0.061 mmol) in ethanol (3 mL) was stirred for 24 h at room temperature and then concentrated. The residue was purified by column chromatography (toluene/ethyl acetate 2:1) to give 8 as a colourless syrup (9 mg, 69%); R_f = 0.32 (toluene/ethyl acetate 2:1); ¹H NMR (250.13 MHz, CDCl₃): δ 6.47 (s, 1H, H-4'); 6.18 (d, 1H, $J_{1,2}$ = 5.2 Hz, H-1); 5.56 (d, 1H, $J_{2,3}$ = 2.6 Hz, H-3); 5.43 (dd, 1H, H-2); 2.49 (s, 3H, SCH₃); 1.46, 1.45 (2s, 6H, C(CH₃)₂); ¹³C NMR (62.90 MHz, CDCl₃): δ 150.4, 149.9 (2br, C-3', C-5'); 112.8 (*C*(CH₃)₂); 106.2, 105.6 (C-1, C-4'); 98.4 (C-3); 83.8 (C-2); 28.0, 27.8 (C(*C*H₃)₂); 17.1(SCH₃). HRMS: Calcd for C₁₁H₁₄O₃N₂S: 254.07251. Found: 254.07118.

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REFERENCES

- 1. F. Nicotra, Topics in Current Chemistry, 187, 55 (1997).
- R. J. Suhadolnik, *Nucleosides as Biological Probes*, Wiley-Interscience, New York, (1979).
- J. Goodchild in *Topics in Antibiotic Chemistry*, Vol. 6, The Biochemistry of Nucleoside Antibiotics, P. G. Sammes, Ed., E. Horwood, Chichester, 1982, p 99.
- D. E. Levy and C. Tang, *The Chemisry of C-Glycosides*, Pergamon, Elsevier Science Ltd., Oxford, 1995.

SYNTHESIS OF PYRAZOLE ISO-C-NUCLEOSIDES

- 5. K. Peseke, H. Feist and E. Cuny, Carbohydr. Res., 230, 319 (1992).
- 6. K. Peseke, G. Thiele and M. Michalik, Liebigs Ann., 1633 (1995).
- 7. J. M. Tronchet, Les Colloques de l'INSERM, INSERM, 81, 117 (1978).
- 8. H. H. Baer and I. Gilron, Carbohydr. Res., 164, 486 (1987).
- 9. K. M. Sun and B. Fraser-Reid, Synthesis, 28 (1982).
- 10. V. Salas-Reyes, Z. Naturforsch., 50b, 1537 (1995).
- 11. D. E. Kiely, H. Walls Jr. and R. L. Black, Carbohydr. Res., 31, 387 (1973).
- 12. M. L. Wolfrom and S. Hanessian, J. Org. Chem., 27, 1800 (1962).
- 13. M. L. Wolfrom and S. Hanessian, J. Org. Chem., 27, 2107 (1962).
- 14. M. Michalik, K. Peseke and R. Radeglia, J. Prakt. Chem., 327, 103 (1985).
- 15. R. K. Dieter, Tetrahedron, 42, 3029 (1986).
- 16. E. Recondo and H. Rinderknecht, Helv. Chim. Acta, 43, 1653 (1960).
- 17. R. L. Whistler, R. E. Gramera and A. Park, J. Org. Chem., 28, 3230 (1963).
- 18. H. Ohle and R. Deplanque, Ber. Dtsch. Chem. Ber., 66, 12 (1933).