

Synthesis of phosphodiesterase IVb inhibitors

2.* Stereoselective synthesis of hexahydro-3*H*-pyrrolo[1,2-*c*]imidazol-3-one and tetrahydro-1*H*-pyrrolo[1,2-*c*][1,3]oxazol-3-one derivatives**

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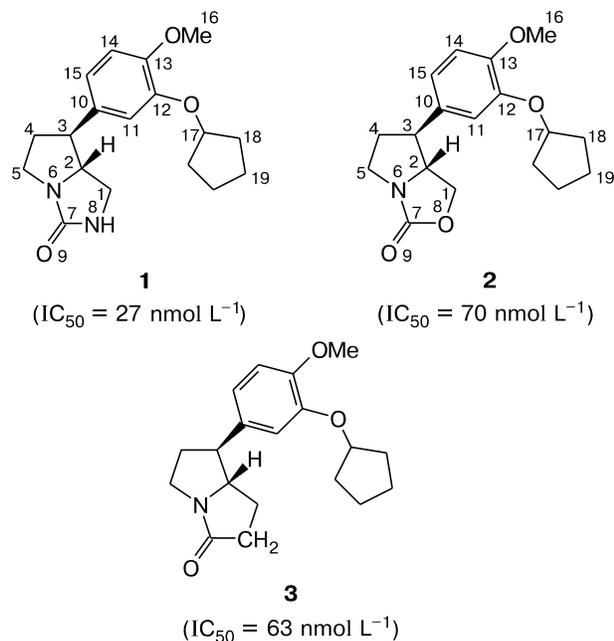
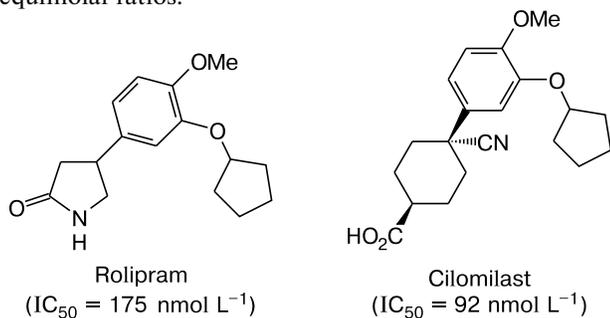
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The total stereoselective synthesis of two highly potent phosphodiesterase IVb inhibitors from nitroethane, isovanillin, and ethyl vinyl ether was developed. The compounds obtained are the derivatives of hexahydro-3*H*-pyrrolo[1,2-*c*]imidazol-3-one and tetrahydro-1*H*-pyrrolo[1,2-*c*][1,3]oxazol-3-one. The strategy proposed involves silylation of six-membered cyclic nitronates as a key step leading to 5,6-dihydro-4*H*-1,2-oxazines with the group CH₂FG (FG = N₃ or OH) at the C(3) atom.

Key words: stereoselective synthesis, nitro compounds, cyclic nitronates, silylation, pyrrolidines, rolipram, phosphodiesterase, pyrrolo[1,2-*c*]imidazol-3-ones, pyrrolo[1,2-*c*][1,3]oxazol-3-ones.

Phosphodiesterase (PDE) IVb inhibitors are considered to be promising drugs for the treatment of disorders of the central nervous system and respiratory diseases.¹ The well-known representatives are, e.g., rolipram (antidepressant)² and cilomilast (asthmatic drug).³ Bicyclic derivatives **1–3** proposed by GlaxoSmithKline Co.⁴ are vastly superior to rolipram and cilomilast in the PDE IVb inhibition constants IC₅₀.⁵ However, the syntheses of products **1–3** developed by GlaxoSmithKline Co. are nonstereoselective, yielding mixtures of diastereomers in nearly equimolar ratios.⁴

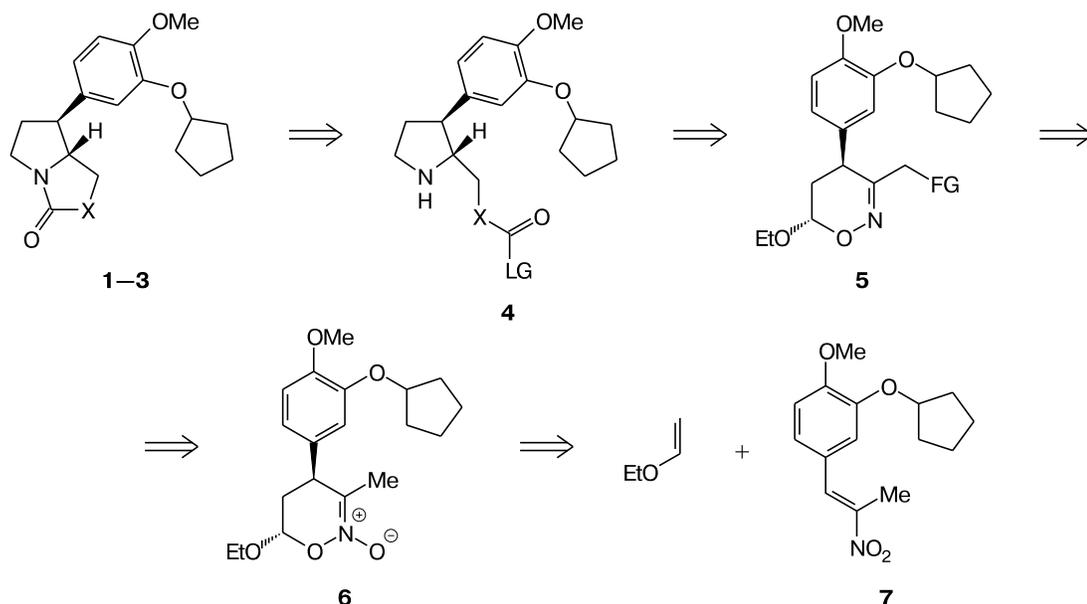


* For Part 1, see Ref. 1a.

** Dedicated to Academician of the Russian Academy of Sciences O. M. Nefedov on the occasion of his 80th birthday.

Recently,⁶ we have proposed a strategy for the stereoselective synthesis of 2,3-*trans*-disubstituted pyrrolidines from nitroethane, aldehydes, and vinyl ethers through in-

Scheme 1



X = NH (**1**), O (**2**), CH₂ (**3**)

LG stands for a leaving group; FG = N₃, OH, CH(CO₂Me)₂

intermediate formation of six-membered cyclic oxime ethers with a functionalized methyl group at the C(3) atom. The key step in this synthesis was silylation of six-membered cyclic nitronates.⁷ Obviously, this strategy can be employed for stereocontrolled synthesis of PDE IVb inhibitors **1–3** (Scheme 1). For instance, immediate precursors of products **1–3** are pyrrolidines **4**, which can be obtained by reduction of cyclic oxime ethers **5**. In turn, such oxime ethers can be derived from cyclic nitronate **6** according to our recent silylation-based procedures. Cyclic nitronate **6** is assembled *via* a [4+2] cycloaddition⁸ of nitroalkene **7** (accessible from isovanillin) to ethyl vinyl ether.

Following Scheme 1, we have earlier⁹ succeeded in the stereoselective synthesis of pyrrolizidinone **3**. In the present work, we pioneered in the total stereoselective synthesis of pyrroloimidazolone **1** and pyrrolooxazolone **2**.

The key intermediates in the synthesis of target products **1** and **2** are six-membered cyclic oxime ethers **5a** and **5b**, respectively (Scheme 2). They were obtained in two steps from nitronate **6** described earlier.⁹

For the synthesis of azide **5a**, nitronate **6** was transformed into bromomethyldihydrooxazine **8** according to a known method^{7a} using an excess of trimethylsilyl bromide and triethylamine. To replace the Br atom by azido group, compound **8** was treated with NaN₃ in aqueous acetone in the presence of catalytic amounts of NaI.¹⁰

For the synthesis of hydroxymethyldihydrooxazine **5b**, we used our recent procedure for rearrangement of Me₃Si-protected 1,2-oxazines.¹¹ To do this, nitronate **6**

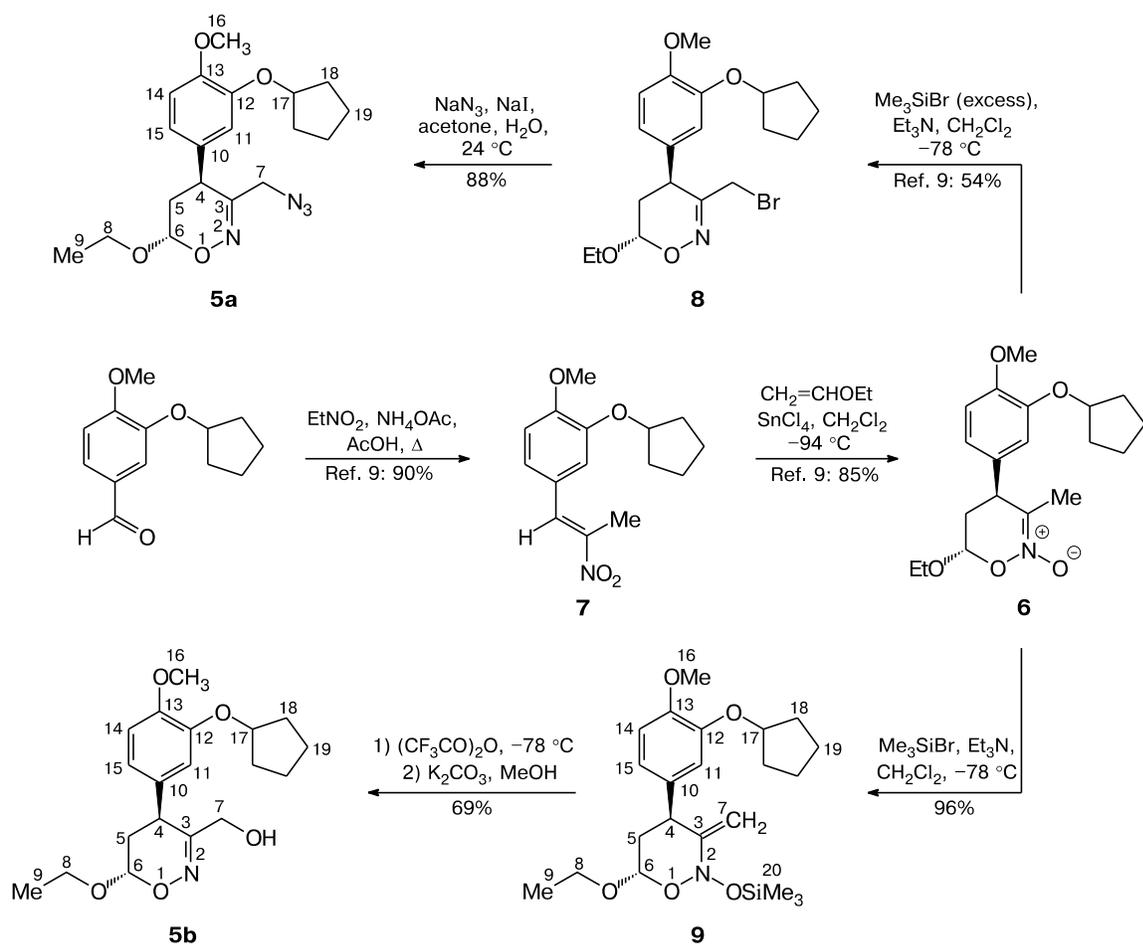
was initially transformed into cyclic compound **9** by monosilylation with trimethylsilyl bromide in the presence of triethylamine.^{7b} Treatment of compound **9** with trifluoroacetic anhydride followed by methanolysis of the reaction mixture gave product **5b** in good yield.

For high stereoselectivity to be achieved, reductive transformation^{6a} of the dihydrooxazine ring into a pyrrolidine one was carried out in two steps (Scheme 3): successive hydride reduction of the C=N bond and catalytic hydrogenation of the N–O bond.^{6a}

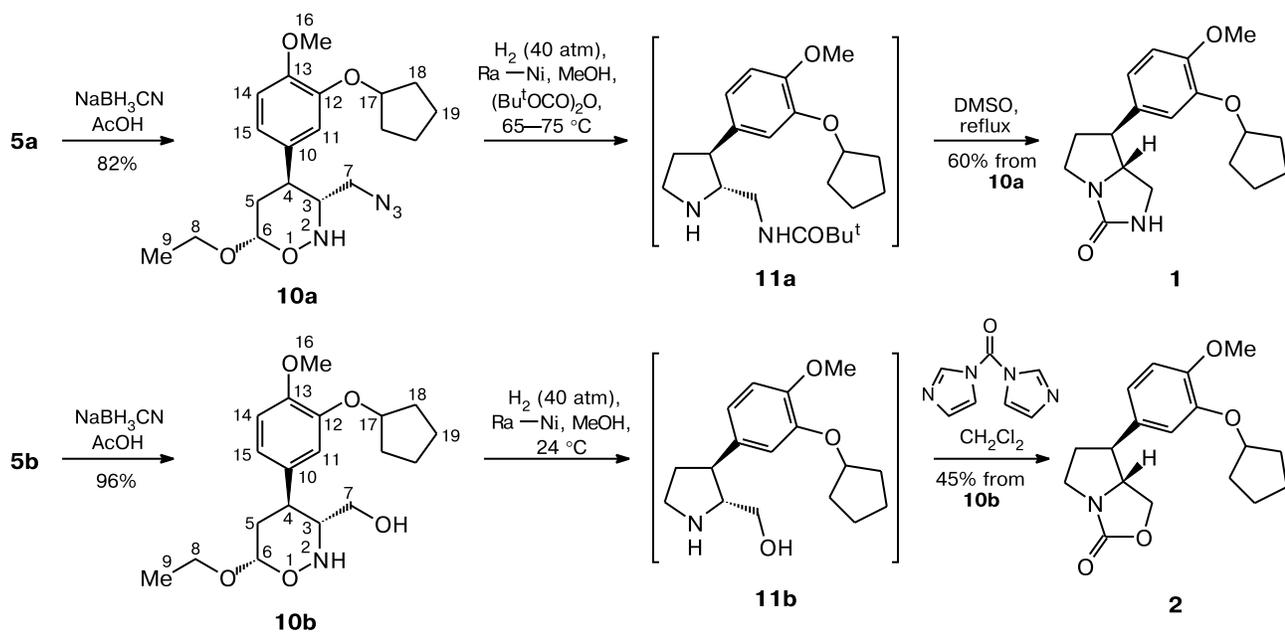
Reduction of dihydrooxazines **5a** and **5b** with sodium cyanoborohydride in acetic acid stereoselectively gave only tetrahydro-2*H*-1,2-oxazines **10a** and **10b**, respectively, with the required 3,4-*trans*-configuration of the substituents in the oxazine ring. Catalytic hydrogenation of compounds **10a** and **10b** on Raney nickel yielded, *via* the 1,2-oxazine→pyrrolidine transformation, products **11a** and **11b**, respectively (see Scheme 3). Oxazine **10a** was hydrogenated in the presence of di-*tert*-butyl dicarbonate (1 equiv.) for protection of the primary amino group in an intermediate resulting from the first-step reduction of the azido group. Pyrrolidine **11a** was transformed without further purification into the target product **1** in boiling DMSO (see Scheme 3). Cyclization of pyrrolidine **11b** into the target product **2** was carried out under the action of 1,1'-carbonyldiimidazole¹² in CH₂Cl₂.

The structures of novel intermediate products **5a,b**, **9**, and **10a,b** were confirmed by 1D (¹H and ¹³C) and 2D NMR spectroscopy (COSY and HSQC), high-resolution

Scheme 2



Scheme 3



mass spectrometry, and elemental analysis. The spectroscopic characteristics of the target products **1** and **2** agree well with the literature data.⁴

To sum up, starting from nitroethane and isovanillin, we performed the stereoselective seven-step syntheses of the known PDE IVb inhibitors **1** and **2** in overall 18 and 22% yields, respectively. The previously described schemes for their syntheses are nonstereoselective⁴ and inferior in yields: 2% over nine steps for product **1** and 5% over six steps for product **2**.

Experimental

Catalytic hydrogenation was carried out in a steel autoclave (Parr) with external heating and stirring. Column chromatography was carried out on silica gel (Merck Kieselgel 40–60 μm 60A). 1D and 2D NMR spectra were recorded on a Bruker AM-300 spectrometer (300.13 MHz). Chemical shifts are referenced to the residual signal of the solvent.¹³ High-resolution mass spectra were recorded on a MicroTOFF instrument. Elemental analysis was performed at the analytical laboratory of the N. D. Zelinsky Institute of Organic Chemistry (Russian Academy of Sciences). Melting points were measured on a Kofler hot stage and are given uncorrected.

Acetic acid was recrystallized twice. Dichloromethane, MeCN, Et₃N, ethyl vinyl ether, and Me₃SiBr were distilled over CaH₂; DMSO (Sigma-Aldrich), SnCl₄ (Aldrich), NaBH₃CN (Acros), NaN₃ (Aldrich), (CF₃CO)₂O (Aldrich), (Bu^tOCO)₂O (Aldrich), Raney nickel (Acros, 50% suspension in water), and 1,1'-carbonyldiimidazole (Acros) were used as purchased. Nitronate **6**, nitroalkene **7**, and bromooxazine **8** were prepared as described earlier.⁹

(4S*,6S*)-4-(3-Cyclopentyloxy-4-methoxyphenyl)-6-ethoxy-3-methylidene-2-trimethylsilyloxy-1,2-oxazinane (9). Triethylamine (0.19 mL, 1.3 mmol) and TMSBr (0.16 mL, 1.2 mmol) were successively added under argon at –78 °C to a stirred solution of oxazine *N*-oxide **6** (0.389 g, 1.11 mmol) in CH₂Cl₂ (2.2 mL). The reaction mixture was kept at –78 °C for 48 h, diluted with hexane (5 mL), and poured into the two-phase system hexane (40 mL)–H₂O (20 mL). The organic layer was washed with a solution of NaHSO₄ (0.12 g) in water (20 mL) and with brine (2×20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. The resulting labile nitroso acetal **9** (0.449 g, 96%) as a colorless oil was further used without purification. ¹H NMR (CDCl₃), δ : 0.26 (s, 9 H, H₃C(20)); 1.27 (t, 3 H, H₃C(9), *J* = 7.3 Hz); 1.57–1.67 (m, 2 H) and 1.78–1.97 (m, 6 H) (H₂C(18) and H₂C(19)); 2.10 (ddd, 1 H, H_{eq}C(5), *J* = 13.2 Hz, *J* = 5.5 Hz, *J* = 5.5 Hz); 2.23 (ddd, 1 H, H_{ax}C(5), *J* = 13.2 Hz, *J* = 9.5 Hz, *J* = 4.4 Hz); 3.59 (dq, 1 H, HC(8), *J* = 9.5 Hz, *J* = 7.3 Hz); 3.85 (s, 3 H, H₃C(16)); 3.89–4.14 (m, 3 H, HC(4), HC(7), HC(8)); 4.74–4.79 (m, 1 H, HC(17)); 4.95 (br.s, 1 H, HC(7)); 5.10 (t, 1 H, HC(6), *J* = 4.4 Hz); 6.83 (br.s, 3 H, HC(11), HC(14), HC(15)). ¹³C NMR (CDCl₃), δ : –0.8 (C(20)); 15.0 (C(9)); 24.0 (C(19)); 32.8, 36.6 (C(5), C(18)); 40.5 (br, C(4)); 56.1 (C(16)); 64.1 (C(8)); 80.4 (C(17)); 97.4 (br, C(7)); 99.3 (C(6)); 111.9, 115.9, 120.6 (C(11), C(14), C(15)); 133.0 (br, C(10)); 147.5, 149.0 (C(12), C(13)); 158.4 (br, C(3)).

(4S*,6S*)-3-Azidomethyl-4-(3-cyclopentyloxy-4-methoxyphenyl)-6-ethoxy-5,6-dihydro-4*H*-1,2-oxazine (5a). A solution

of NaN₃ (0.249 g, 3.8 mmol) and NaI (0.035 g, 0.23 mmol) in water (2.5 mL) was added to a solution of bromooxazine **8** (0.410 g, 1.0 mmol) in acetone (12 mL).⁹ The reaction mixture was stirred at room temperature for 24 h and concentrated. The residue was preadsorbed on silica gel (~2 g) and then chromatographed on silica gel with gradient elution in hexane–EtOAc (10 : 1 → 5 : 1). The resulting colorless oil crystallized on cooling, m.p. 38–41 °C. The yield of compound **5a** was 0.330 g (88%), *R*_f 0.68 (EtOAc–hexane, 1 : 1). Found (%): C, 60.76; H, 7.27; N, 14.86. C₁₉H₂₆N₄O₄. Calculated (%): C, 60.95; H, 7.00; N, 14.96. HRMS, *m/z*: 375.2030 [MH]⁺, calculated for [MH]⁺: 375.2027. ¹H NMR (CDCl₃, COSY, HSQC), δ : 1.27 (t, 3 H, H₃C(9), *J* = 7.0 Hz); 1.57–1.66 (m, 2 H, HC(19)); 1.81–1.97 (m, 6 H, H₂C(18), HC(19)); 2.14 (ddd, 1 H, H_{ax}C(5), *J* = 13.9 Hz, *J* = 12.5 Hz, *J* = 2.2 Hz); 2.31 (ddd, 1 H, H_{eq}C(5), *J* = 13.9 Hz, *J* = 7.3 Hz, *J* = 1.8 Hz); 3.48 (d, 1 H, HC(7), *J* = 13.2 Hz); 3.65 (m, 1 H, HC(8)); 3.7 (dd, 1 H, HC(4), *J* = 12.5 Hz, *J* = 7.3 Hz); 3.82 (d, 1 H, HC(7), *J* = 13.2 Hz); 3.93 (s, 3 H, H₃C(16)); 3.91 (m, 1 H, HC(8)); 4.76 (m, 1 H, HC(17)); 5.20 (dd, 1 H, HC(6), *J* = 2.2 Hz, *J* = 1.8 Hz); 6.70 (d, 1 H, HC(11), *J* = 1.5 Hz); 6.75 (dd, 1 H, HC(15), *J* = 8.1 Hz, *J* = 1.5 Hz); 6.85 (d, 1 H, HC(14), *J* = 8.1 Hz). ¹³C NMR (CDCl₃, HSQC), δ : 15.0 (C(9)); 24.0 (C(19)); 32.8 (C(5), C(18)); 34.4 (C(4)); 52.5 (C(7)); 56.1 (C(16)); 63.8 (C(8)); 80.5 (C(17)); 95.9 (C(6)); 112.6, 114.9, 120.5 (C(11), C(14), C(15)); 131.1 (C(10)); 148.2, 149.6 (C(12), C(13)); 157.2 (C(3)).

(4S*,6S*)-4-(3-Cyclopentyloxy-4-methoxyphenyl)-6-ethoxy-3-hydroxymethyl-5,6-dihydro-4*H*-1,2-oxazine (5b). Tri-fluoroacetic anhydride (0.105 mL, 0.74 mmol) was added under argon at –78 °C to a stirred solution of nitroso acetal **9** (0.312 g, 0.74 mmol) in CH₂Cl₂ (3.7 mL). The reaction mixture was stirred at –78 °C for 1 h. Then K₂CO₃ (0.19 g) and MeOH (0.8 mL) were successively added. The reaction mixture was stirred without cooling for 2.5 h and poured into the two-phase system EtOAc (60 mL)–brine (20 mL). The organic layer was washed with brine (20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. The product was isolated by column chromatography with gradient elution in hexane–EtOAc (3 : 1 → 1 : 1). The resulting faintly yellow oil solidified upon storage, m.p. 97–99 °C. The yield of the target oxazine **5b** was 0.179 g (69%), *R*_f 0.39 (EtOAc–hexane, 1 : 1). HRMS, *m/z*: 350.1962 [MH]⁺, calculated for [MH]⁺: 350.1962. ¹H NMR (CDCl₃), δ : 1.26 (t, 3 H, H₃C(9), *J* = 7.3 Hz); 1.53–1.68 (m, 2 H, HC(19)); 1.73–1.95 (m, 6 H, H₂C(18), HC(19)); 2.11 (ddd, 1 H, HC_{ax}(5), *J* = 13.9 Hz, *J* = 12.5 Hz, *J* = 2.9 Hz); 2.26 (ddd, 1 H, H_{eq}(5), *J* = 13.9 Hz, *J* = 7.3 Hz, *J* = 2.2 Hz); 2.56 (br.s, 1 H, HO); 3.59 (dd, 1 H, HC(4), *J* = 12.5 Hz, *J* = 7.3 Hz); 3.66 (dq, 1 H, HC(8), *J* = 9.5 Hz, *J* = 7.3 Hz); 3.84 (s, 3 H, H₃C(16)); 3.91 (dq, 1 H, HC(8), *J* = 9.5 Hz, *J* = 7.3 Hz); 3.98 (s, 2 H, HC(7)); 4.76 (m, 1 H, HC(17)); 5.20 (br.s, 1 H, HC(6)); 6.68 (d, 1 H, HC(11), *J* = 1.9 Hz); 6.73 (dd, 1 H, HC(15), *J* = 8.1 Hz, *J* = 1.9 Hz); 6.83 (d, 1 H, HC(14), *J* = 8.1 Hz). ¹³C NMR (CDCl₃), δ : 15.0 (C(9)); 24.0 (C(19)); 32.6, 32.7 (C(5), C(18)); 34.2 (C(4)); 56.1 (C(16)); 62.6, 63.9 (C(7), C(8)); 80.5 (C(17)); 96.1 (C(6)); 112.4, 114.8, 120.6 (C(11), C(14), C(15)); 131.0 (C(10)); 148.1, 149.6 (C(12), C(13)); 159.0 (C(3)).

(3*R,4*S**,6*S**)-3-Azidomethyl-4-(3-cyclopentyloxy-4-methoxyphenyl)-6-ethoxy-1,2-oxazinane (10a)**. Sodium cyanoborohydride (0.127 g, 2.05 mmol) was added to a stirred solution of azidomethyloxazine **5a** (0.250 g, 0.67 mmol) in acetic acid (3 mL). The reaction mixture was vigorously stirred at room

temperature for 1 h and poured into the system EtOAc (150 mL)—K₂CO₃ (150 mL). The aqueous layer was washed with EtOAc (50 mL). The combined organic layer was successively washed with a saturated aqueous solution of K₂CO₃ (70 mL), water (100 mL), and brine (100 mL), dried over Na₂SO₄, and concentrated *in vacuo*. The residue was preadsorbed on silica gel (~2 g) and then chromatographed on silica gel with gradient elution in hexane—EtOAc (10 : 1 → 5 : 1 → 3 : 1). The yield of compound **10a** was 0.207 g (82%), colorless oil, *R*_f 0.71 (EtOAc—hexane, 1 : 1). Found (%): C, 60.74; H, 7.56; N, 14.18. C₁₉H₂₈N₄O₄. Calculated (%): C, 60.62; H, 7.50; N, 14.88. HRMS, *m/z*: 377.2169 [MH]⁺, calculated for [MH]⁺: 377.2183. ¹H NMR (CDCl₃, COSY, HSQC), δ: 1.30 (t, 3 H, H₃C(9), *J* = 7.0 Hz); 1.55–1.67 (s, 2 H, HC(19)); 1.78–1.97 (m, 6 H, H₂C(18), HC(19)); 1.94 (ddd, 1 H, H_{eq}C(5), *J* = 13.2 Hz, *J* = 5.9 Hz, *J* = 1.4 Hz); 2.04 (ddd, 1 H, H_{ax}C(5), *J* = 13.2 Hz, *J* = 10.3 Hz, *J* = 2.9 Hz); 3.02 (ddd, 1 H, H_{ax}C(4), *J* = 11.0 Hz, *J* = 10.3 Hz, *J* = 5.9 Hz); 3.16 (dd, 1 H, HC(7), *J* = 12.5 Hz, *J* = 5.1 Hz); 3.27 (ddd, 1 H, H_{ax}C(3), *J* = 11.0 Hz, *J* = 5.1 Hz, *J* = 2.2 Hz); 3.33 (dd, 1 H, HC(7), *J* = 12.5 Hz, *J* = 2.2 Hz); 3.58 (dq, 1 H, HC(8), *J* = 9.5 Hz, *J* = 7.3 Hz); 3.83 (s, 3 H, H₃C(16)); 3.85 (dq, 1 H, HC(8), *J* = 9.5 Hz, *J* = 7.3 Hz); 4.75 (m, 1 H, HC(17)); 4.89 (dd, 1 H, HC(6), *J* = 2.9 Hz, *J* = 1.4 Hz); 5.39 (br.s, 1 H, HN(2)); 6.69–6.75 (m, 2 H, HC(11), HC(15)); 6.83 (d, 1 H, HC(14), *J* = 8.1 Hz). ¹³C NMR (CDCl₃, HSQC), δ: 15.2 (C(9)); 24.0 (C(19)); 32.8 (C(18)); 36.3 (C(5)); 37.3 (C(4)); 51.0 (C(7)); 56.1 (C(16)); 61.5 (C(3)); 63.6 (C(8)); 80.4 (C(17)); 98.0 (C(6)); 112.5, 114.5, 119.3 (C(11), C(14), C(15)); 133.7 (C(10)); 147.9, 149.1 (C(12), C(13)).

(3*R,4*S**,6*S**)-4-(3-Cyclopentyloxy-4-methoxyphenyl)-6-ethoxy-3-hydroxymethyl-1,2-oxazinane (10b)**. Sodium cyanoborohydride (0.109 g, 1.73 mmol) was added under argon to a stirred solution of oxazine **5b** (0.170 g, 0.49 mmol) in acetic acid (3.1 mL). The reaction mixture was stirred at room temperature for 80 min and poured into the two-phase system EtOAc (60 mL)—saturated aqueous Na₂CO₃ (30 mL). The aqueous layer was washed with EtOAc (2×20 mL). The combined organic layer was washed with saturated aqueous Na₂CO₃ (20 mL), water (20 mL), and brine (2×20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. The product was isolated by column chromatography with gradient elution from CHCl₃ to CHCl₃—MeOH (20 : 1). The yield of the target tetrahydrooxazine **10b** was 0.165 g (96%), faintly yellow oil. For analytical purposes, the product was recrystallized from hexane—EtOAc (2 : 1), m.p. 87–89 °C, *R*_f 0.49 (CHCl₃—MeOH, 10 : 1). Found (%): C, 64.59; H, 8.37; N, 3.99. C₁₉H₂₉NO₅. Calculated (%): C, 64.93; H, 8.32; N, 3.99. ¹H NMR (CDCl₃), δ: 1.30 (t, 3 H, H₃C(9), *J* = 7.0 Hz); 1.52–1.67 (m, 2 H, HC(19)); 1.72–2.12 (m, 8 H, H₂C(18), H₂C(19), H₂C(5)); 2.95 (ddd, 1 H, H_{ax}C(4), *J* = 11.3 Hz, *J* = 11.3 Hz, *J* = 4.4 Hz); 3.21–3.35 (m, 2 H) and 3.45–3.55 (m, 1 H) (HC(3) and HC(7)); 3.58 (dq, 1 H, HC(8), *J* = 9.5 Hz, *J* = 7.0 Hz); 3.81 (s, 3 H, H₃C(16)); 3.85 (dq, 1 H, HC(8), *J* = 9.5 Hz, *J* = 7.0 Hz); 4.76 (m, 1 H, HC(17)); 4.90 (br.s, 1 H, HC(6)); 6.68–6.75 (m, 2 H) and 6.81 (d, 1 H, *J* = 8.8 Hz) (HC(11), HC(14), HC(15)). ¹³C NMR (CDCl₃), δ: 15.2 (C(9)); 24.0 (C(19)); 32.8, 36.5, 37.0 (C(4), C(5), C(18)); 56.2 (C(16)); 61.5, 63.7, 77.3 (C(3), C(7), C(8)); 80.5 (C(17)); 98.3 (C(6)); 112.5, 114.6, 119.4 (C(11), C(14), C(15)); 134.2 (C(10)); 147.8, 149.1 (C(12), C(13)).

(7*S,7*aR**)-7-(3-Cyclopentyloxy-4-methoxyphenyl)hexahydro-3*H*-pyrrolo[1,2-*c*]imidazol-3-one (1)**. Raney nickel as

a 50% suspension in water (~0.2 mL) was washed with MeOH (4×2 mL) and added in MeOH (0.5 mL) to a solution of oxazine **10a** (0.058 g, 0.154 mmol) and di-*tert*-butyl dicarbonate (0.033 mL, 0.154 mmol) in MeOH (2.0 mL). The mixture was hydrogenated in an autoclave (65–75 °C, H₂, *p* = 40 atm) for 4.5 h, filtered through a short pad of Celite to remove the catalyst, and concentrated *in vacuo*. The residue was dissolved in DMSO (3.0 mL). The resulting solution was gently refluxed under argon for 30 min. Then the solvent was removed *in vacuo* (100 °C, 1 Torr) and the residue was chromatographed on silica gel with gradient elution from hexane—EtOAc (10 : 1 → 5 : 1 → 3 : 1 → 1 : 1) to EtOAc. The resulting oily product was recrystallized from Et₂O. The yield of the target compound **1** was 0.029 g (60%), m.p. 139–141 °C (*cf.* Ref. 4: m.p. 118–120 °C), *R*_f 0.54 (EtOAc—MeOH, 3 : 1). Found (%): C, 68.39; H, 7.73; N, 8.67. C₁₈H₂₄N₂O₃. Calculated (%): C, 68.33; H, 7.65; N, 8.85. HRMS, *m/z*: 317.1866 [MH]⁺, calculated for [MH]⁺: 317.1860. ¹H NMR (CDCl₃), δ: 1.56–1.71 (m, 2 H, HC(19)); 1.79–1.94 (m, 6 H, H₂C(18), HC(19)); 2.05 (dddd, 1 H, HC(4), *J* = 12.8 Hz, *J* = 11.9 Hz, *J* = 10.1 Hz, *J* = 9.2 Hz); 2.38 (dddd, 1 H, HC(4), *J* = 12.8 Hz, *J* = 10.1 Hz, *J* = 8.2 Hz, *J* = 4.6 Hz); 2.74 (ddd, 1 H, HC(3), *J* = 11.0 Hz, *J* = 10.1 Hz, *J* = 9.2 Hz); 3.33 (dd, 1 H, HC(1); *J* = 9.9 Hz, *J* = 9.9 Hz); 3.51 (dd, 1 H, HC(1), *J* = 9.9 Hz, *J* = 7.7 Hz); 3.26–3.31 (m, 1 H) and 3.64–3.75 (m, 2 H) (HC(2), H₂C(5)); 3.82 (s, 3 H, H₃C(16)); 4.76 (m, 1 H, HC(17)); 5.55 (br.s, 1 H, HN(8)); 6.72 (s, 1 H, HC(11)), 6.73 (d, 1 H, *J* = 7.3 Hz) and 6.83 (d, 1 H, *J* = 7.3 Hz) (HC(14), HC(15)). ¹³C NMR (CDCl₃, JMOD), δ: 24.0, 32.7, 34.4, 41.3, 45.2 (C(1), C(4), C(5), C(18), C(19)); 48.5, 56.2, 66.1 (C(2), C(3), C(16)); 80.6 (C(17)), 112.4, 114.7, 119.6 (C(11), C(14), C(15)); 131.7 (C(10)); 147.9, 149.3 (C(12), C(13)); 165.8 (C(7)). The spectroscopic characteristics are in full agreement with the literature data⁴ for compound **1**.

(7*S,7*aR**)-7-(3-Cyclopentyloxy-4-methoxyphenyl)tetrahydro-1*H*-pyrrolo[1,2-*c*][1,3]oxazol-3-one (2)**. Raney nickel as a 50% suspension in water (~1 mL) was washed with MeOH (4×2 mL) and added, under a layer of MeOH (1.5 mL), to a solution of tetrahydrooxazine **10b** (0.15 g, 0.43 mmol) in MeOH (2.5 mL). The mixture was hydrogenated in a steel autoclave (24 °C, H₂, *p* = 40 atm) for 6 h and filtered through a short pad of Celite. The Celite filter was washed with MeOH (50 mL). The filtrate was concentrated *in vacuo*. The residue (0.104 g) was dissolved in CH₂Cl₂ (3.0 mL) and 1,1'-carbonyldiimidazole (0.063 g, 0.39 mmol) was added with stirring under argon. The reaction mixture was kept for 72 h and poured into the two-phase system EtOAc (50 mL)—0.07 *M* aqueous HCl (20 mL). The organic layer was washed with brine (2×15 mL), dried over Na₂SO₄, and concentrated *in vacuo*. The product was isolated by column chromatography with gradient elution in hexane—EtOAc (2 : 1 → 1 : 1). The yield of the target compound **2** was 0.061 g (45%), colorless oil. For analytical purposes, the product was recrystallized from hexane—EtOAc (2 : 1), m.p. 99–101 °C (*cf.* Ref. 4: m.p. 97–98 °C), *R*_f 0.37 (hexane—EtOAc, 1 : 1). HRMS, *m/z*: 340.1517 [MNa]⁺, calculated for [MNa]⁺: 340.1519. ¹H NMR (CDCl₃, COSY), δ: 1.55–1.72 (m, 2 H) and 1.79–1.98 (m, 6 H) (H₂C(18), H₂C(19)); 2.15 (dddd, 1 H, HC(4), *J* = 12.7 Hz, *J* = 11.7 Hz, *J* = 9.5 Hz, *J* = 8.8 Hz); 2.49 (dddd, 1 H, HC(4), *J* = 12.7 Hz, *J* = 10.3 Hz, *J* = 8.1 Hz, *J* = 2.2 Hz); 2.79 (ddd, 1 H, HC(3), *J* = 10.3 Hz, *J* = 9.5 Hz, *J* = 7.3 Hz); 3.48 (ddd, 1 H, HC(5), *J* = 11.0 Hz, *J* = 9.5 Hz, *J* = 2.2 Hz); 3.74 (ddd, 1 H, HC(5), *J* = 11.0 Hz, *J* = 8.8 Hz, *J* = 8.1 Hz); 3.82–3.88 (m, 1 H,

HC(2)); 3.84 (s, 3 H, H₃C(16)); 4.22 (dd, 1 H, HC(1), $J = 9.5$ Hz, $J = 2.9$ Hz); 4.43 (dd, 1 H, HC(1), $J = 9.5$ Hz, $J = 8.1$ Hz); 4.73–4.81 (m, 1 H, HC(17)); 6.70–6.77 (m, 2 H) and 6.85 (d, 1 H, $J = 8.1$ Hz) (HC(11), HC(14), HC(15)). ¹³C NMR (CDCl₃, HSQC), δ : 24.0 (C(19)); 32.8 (C(18)); 34.5 (C(4)); 45.8 (C(5)); 49.2 (C(3)); 56.2 (C(16)); 65.8 (C(2)); 66.2 (C(1)); 80.7 (C(17)); 112.5, 114.5, 119.4 (C(11), C(14), C(15)); 130.4 (C(10)); 148.1, 149.7 (C(12), C(13)); 161.7 (C(7)). The spectroscopic characteristics are in full agreement with the literature data⁴ for compound **2**.

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