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A Divergent Route to Nojirimycin Analogues from L-Serinal and 2-Acetylthiazole

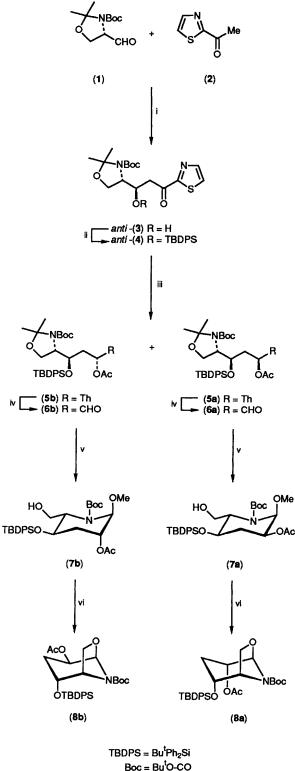
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A five-step synthesis of *O*,*N*-protected (+)-3-deoxynojirimycin (**7a**) and (+)-3-deoxymannojirimycin (**7b**) from *N*-Boc L-serinal acetonide (**1**) employing 2-acetylthiazole (**2**) as a masked α -hydroxypropanal β -anion synthon is described.

There is a growing interest in synthetic methodologies towards polyhydroxylated piperidines since these compounds have been shown to possess potent inhibitory activity against various glucosidases and mannosidases.¹ These compounds (aza sugars) are related to glucosides by substitution of the pyranose oxygen with the amino function and eventually deoxygenation of other positions. Among various 5-deoxyand 1,5-dideoxy-5-iminohexitols, nojirimycin² (5-amino-5deoxy-D-glucopyranose) and mannojirimycin³ (5-amino-5deoxy-D-mannopyranose) as well as their 1-deoxy derivatives, have been the target of several synthetic efforts,⁴ most of which involved elaborations of natural sugars. In view of interest in the structure and enzyme-inhibitory activity relationship, the demand for chemical modifications of these compounds has increased. Here we describe a short and facile synthetic sequence for the preparation of (+)-3-deoxynojirimycin (7a) and (+)-3-deoxymannojirimycin (7b) from the readily available and configurationally stable⁵ L-serine derived aldehyde (1) *via* thiazole-masked aminohexoses.⁶ The main feature of this method is the chain elongation of (1) into a three-carbon homologue employing 2-acetylthiazole (2) as a very effective equivalent to the α -hydroxypropanal β -anion synthon.⁷

Treatment of N-t-butoxycarbonyl L-serinal acetonide (1) in



Ac = MeCO Th = thiazol-2-yl

Scheme 1. Only the major isomers anti-(3) and anti-(4) are shown for convenience. Reagents and conditions: i, ButOLi, THF, -40 °C; ii, Bu^tPh₂SiCl, imidazole, dimethylformamide, room temp.; iii, NaBH₄, MeOH, 0°C, then Ac₂O-pyridine; iv, MeI, MeCN, reflux, then NaBH₄, MeOH, 0°C, then HgCl₂, H₂O-MeCN, room temp.; v, TsOH (Ts = $OSO_2C_6H_4Me_p$), MeOH, 50 °C; vi, TsOH, toluene, reflux.

tetrahydrofuran (THF) with the lithium enolate derived from 2-acetylthiazole (2) as described,⁷ produced the aldol (3) as a mixture of syn and anti diastereoisomers in 70% overall yield (Scheme 1). The protection of the hydroxy group with the t-butyldiphenylsilyl group, by treatment of this mixture with t-butyldiphenylchlorosilane, afforded the O-silyl derivatives[†] anti-(4) and syn-(4) (75%) which were separated by flash chromatography (silica, light petroleum-diethyl ether, 75:25) in ca. 80: 20 ratio. ‡ The reduction of anti-(4) in methanol with NaBH₄ and successive acetylation of the hydroxy group with acetic anhydride and pyridine afforded the differentially protected syn- and anti-1,3-diols (5a) and (5b) in 70:30 ratio§ by NMR spectroscopy and 88% overall yield. Stereochemical assignments for (5a) and (5b) followed their conversion into the corresponding di-O-acetates by desilylation and subsequent acetylation and comparison with authentic samples.⁷ After separation by flash chromatography (silica, diethyl ether-light petroleum, 60:40) the individual isomers (5a) and (5b) were converted into the aldehydes (6a) and (6b) (60-65%) by the one-pot thiazolyl-to-formyl deblocking sequence⁸ involving N-methylation, NaBH₄-reduction, and mercury-mediated hydrolysis. The cyclization of (6a) to the corresponding azapyranose was first examined. Wishing to preserve at this stage the N-Boc protection, the cleavage of the oxazolidine ring was effected by treatment of (6a) in methanol with toluene-p-sulphonic acid (reflux, 30 min), giving, after silica gel column chromatography (light petroleum-ethyl acetate, 70:30), the O,N-protected (+)-3-deoxynojirimycin (7a) (28%) and the 1,6-anhydro derivative (8a) (45%). Under milder conditions (50 °C, 40 min), compound (6a) gave (7a) as a single product (53%) which was successively converted to the 1,6-anhydro sugar (8a) (80%) by treatment with toluene-p-sulphonic acid in refluxing benzene for 30 min.

† All compounds showed consistent NMR and IR spectral data with the assigned structure and gave satisfactory elemental analyses.

‡ Syn and anti O-silyl derivatives (4) are almost quantitatively and more easily separated than syn and anti aldols (3).

§ No substantial change of the level of diastereoselectivity was observed using other reducing agents [LiAlH4, LiAlH4-LiI, diisobutylaluminium hydride (DIBAL), Zn(BH₄)₂]

¶ Selected spectroscopic data: (7a), oil, $[\alpha]_{D}^{20}$ +25.6° (c 0.39, CHCl₃); $\begin{array}{c} IR \ \nu_{max} \ (CHCl_3) \ 3400, \ 1730, \ 1690 \ cm^{-1}; \ ^1H \ NMR \ (80 \ MHz, \\ C_6D_6-D_2O, 90 \ ^\circC), \\ \delta \ 1.20 \ (s, 9 \ H), \ 1.40 \ (s, 9 \ H), \ 1.85 \ (s, 3 \ H), \ 2.05 \ (t, \\ \end{array}$ 2 H, J 3.6 Hz), 3.22 (s, 3 H), 3.37 (d, 1 H, J 6.0 Hz), 3.41 (d, 1 H, J 5.6 Hz), 4.20 (m, 1 H), 4.41 (m, 1 H), 5.05 (dt, 1 H, J 3.6, 2.4 Hz), 5.50 (d, 1 H, J 2.4 Hz), 7.25 (m, 6 H), 7.80 (m, 4 H).

(8a), oil, $[\alpha]_{D}^{20}$ +4.3° (c 0.69, CHCl₃); ¹H NMR (80 MHz, C₆D₆, 63 °C) δ 1.22 (s, 9 H), 1.40 (s, 9 H), 1.55 (m, 2 H), 1.90 (s, 3 H), 2.80 (dd, 1 H, J 14.6, 1.4 Hz), 3.15 (dd, 1 H, J 14.6, 10.0 Hz), 3.35 (br, 1 H), 4.42 (br, 1 H), 4.75 (br, 1 H), 6.03 (d, 1 H, J 3.6 Hz), 7.22 (m, 6 H), 7.85 (m, 4 H).

 $(\mathbf{7b})$, oil, $[\alpha]_{D}^{20} + 15.0^{\circ} (c \ 1.02, \text{CHCl}_3)$, IR $\nu_{\text{max.}}$ (CHCl₃) 3400, 1730, 1690 cm⁻¹; ¹H NMR (300 MHz, CDCl₃), δ 1.08 (s, 4.5 H) and 1.10 (s, 4.5 H) (s, 9 H, in C₆D₆-D₂O, 90 °C, 80 MHz), 1.42 (s, 4.5 H) and 1.52 $(s, 4.5 \text{ H}) (s, 9 \text{ H}, \text{ in } C_6 D_6 - D_2 O, 90 \text{ }^\circ C, 80 \text{ MHz}), 1.92 (m, 1 \text{ H}), 2.08$ (m, 1 H), 2.14 (s, 3 H), 2.40 (dd, 0.5 H, J 8.8, 4.0 Hz), 3.2 (m, 1 H), 3.32 (m, 1 H), 3.38 (s, 1.5 H), and 3.45 (s, 1.5 H) (s, 3 H, in C₆D₆-D₂O, 90 °C, 80 MHz), 3.91 (t, 0.5 H, J 4.6 Hz), 4.13 (m, 2 H), 5.36 (t, 0.5 H, J 4.0 Hz), and 5.40 (t, 0.5 H, J 4.0 Hz) (still multiplet in $C_6D_6-D_2O$, 90 °C, 80 MHz), 5.52 (d, 0.5 H, J 3.5 Hz), and 5.68 (d, 0.5 H, J 3.5 Hz) (d 1 H, J 3.5 Hz in C₆D₆-D₂O, 90 °C, 80 MHz), 7.40 (m, 6 H), 7.68 (m, 4 H).

(**8b**), oil, $[\alpha]_{D}^{20}$ +20.1° (c 4.7, CHCl₃); ¹H NMR (300 MHz, CDCl₃), δ 1.08 (s, 9 H), 1.52 (s, 9 H) 1.60 (ddd, 1 H, J 14.5, 10.5, 4.3 Hz), 1.95 (br. dd, 1 H, J 14.5, 6.0 Hz), 2.06 (s, 3 H), 3.51 (d, 1 H, J 8.7 Hz), 3.58 (dd, 1 H, J 8.7, 5.3 Hz), 3.88 (dt, 1 H, J 3.7, 1.7 Hz), 4.42 (br. s, 1 H), 5.12 (ddd, 1 H, J 10.5, 6.0, 1.6 Hz), 5.70 (br. s, 1 H), 7.42 (m, 6 H), 7.70 (m, 4 H).

This is consistent with the α -anomeric form and ${}^{1}C_{4}$ conformation of the amino-pyranose (7a) and its conversion to (8a) by inversion to the ${}^{4}C_{1}$ conformation and cyclization by an internal $S_{N}2cA$ reaction. Similarly, starting from the aldehyde (6b), the O,N-protected (+)-3-deoxymannojirimycin (7b)¶ (57%) was synthesised and transformed into the 1,6-anhydro derivative (8b)¶ (95%). This provides a five-step synthesis of aza sugars of the nojirimycin family in which the new asymmetric centres are created in subsequent steps by exploiting the 2S configuration of L-serine and the thiazole ring is employed as an effective formyl group equivalent. Application of this technology to the *de novo* synthesis of various aza sugars now becomes of interest.

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References

G. W. J. Fleet, *Tetrahedron Lett.*, 1985, 26, 5073; G. W. J. Fleet,
S. K. Namgoong, C. Barker, S. Baines, G. S. Jacob, and B. Winchester, *ibid.*, 1989, 30, 4439, and references cited therein.

- 2 S. Inouye, T. Tsuruoka, T. Ito, and T. Niida, *Tetrahedron*, 1968, 23, 2125.
- 3 T. Niwa, T. Tsuruoka, H. Goi, Y. Kodama, J. Itoh, S. Inouye, Y. Yamada, T. Niida, M. Nobe, and Y. Ogawa, J. Antibiot., 1984, 37, 1579.
- 4 For recent syntheses see, H. Iida, N. Yamazaki, C. Kibayashi, J. Org. Chem., 1987, 52, 3337; G. W. J. Fleet, N. G. Ramsden, and D. R. Witty, Tetrahedron Lett., 1988, 29, 2871; Y. Tsuda, Y. Okuno, and K. Kanemitsu, Heterocycles, 1988, 27, 63; E. Kappes and G. Legler, J. Carbohydr. Chem., 1989, 8, 371; C. H. von der Osten, A. J. Sinskey, C. F. Barbas, III, R. L. Pederson, Y.-F. Wang, and C.-H. Wong, J. Am. Chem. Soc., 1989, 111, 3924; B. Rajanikanth and R. Seshadri, Tetrahedron Lett., 1989, 30, 755; N. Chida, Y. Furuno, and S. Ogawa, J. Chem. Soc., Chem. Commun., 1989, 1230.
- 5 P. Garner and S. Ramakanth, J. Org. Chem., 1986, 51, 2609.
- 6 A. Dondoni, *Phosphorus, Sulphur, and Silica*, 1989, **43**, 25; A. Dondoni, G. Fantin, M. Fogagnolo, and P. Pedrini, *J. Org. Chem.*, 1990, **55**, 1433.
- 7 A. Dondoni, G. Fantin, and M. Fogagnolo, *Tetrahedron Lett.*, 1989, **30**, 6063.
- 8 A. Dondoni, G. Fantin, M. Fogagnolo, and A. Medici, Angew. Chem., Int. Ed. Engl., 1986, 25, 835; A. Dondoni, G. Fantin, M. Fogagnolo, A. Medici, and P. Pedrini, J. Org. Chem., 1989, 54, 693.