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Authors: Shifa Zhu, Tongxiang Cao, Yi Kong, Kui Luo, and Lianfen Chen

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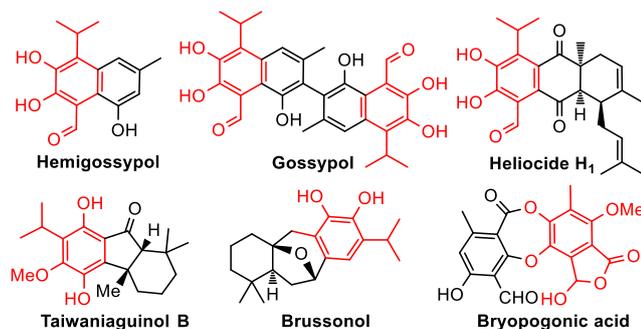
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Cascade Claisen Rearrangement: Rapid Synthesis of Polysubstituted Salicylaldehydes and Total Synthesis of Hemigossypol and Gossypol

Tongxiang Cao, Yi Kong, Kui Luo, Lianfen Chen, and Shifa Zhu*

Abstract: A cascade Claisen rearrangement of well-organized maltol propargyl ether for the construction of polysubstituted salicylaldehydes was reported. This reaction featured high atom economy (100%), as well as catalyst-free and gram-scale conditions. Based on this novel methodology, we have realized the total synthesis of hemigossypol, gossypol, and their analogues.

Polysubstituted salicylaldehydes and their derivatives are ubiquitous in pharmaceuticals, natural products and agrochemicals, such as hemigossypol, gossypol, helicoidine H₁, Taiwaniaguinol B, brussolon, and bryopogonic acid (Scheme 1).^[1] Among them, gossypol is found in flowers, seeds, roots and foliage of cotton plants, where it serves as a defense compound against insect pests and pathogens.^[2] It has attracted a lot of interest for its multiple pharmacological activities including spermicidal, antiparasitic, anticancer, and antiviral activities.^[3] Hemigossypol is the biosynthetic precursor of gossypol and has shown improved antifungal activity compared to gossypol.^[4] Despite the tremendous progress achieved in transition metal-catalyzed polyphenol synthesis,^[5] construction of these compounds in a concise manner still remains a difficult problem due to low selectivity^[6] and oxidant sensitivity (such as oxidative phenolic coupling and dearomatic reactivity),^[1d-e, 7] which may result in protecting-group or redox manipulation and multistep processes for the introduction of other substituents.



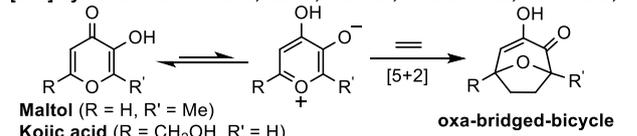
Scheme 1. Natural products containing polysubstituted phenol.

Traditionally, Diels-Alder^[8] and 6 π -electrocyclization^[9] reaction can assemble the cyclohexene skeleton, but further oxidation is commonly required to access the benzene structure. Dehydro-Diels-Alder and transition metal-catalyzed [2+2+2] reactions can form the benzene skeleton, but a mixture of isomers is often obtained, and the introduction of polyphenolic hydroxyl groups is

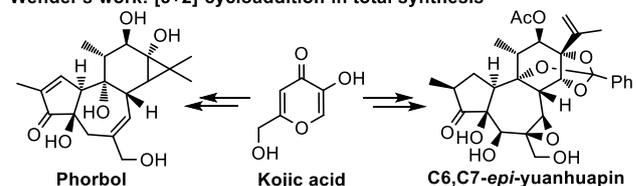
difficult.^[10] The complementary methods using a parent arene to introduce the desired functional groups by means of Friedel-Crafts reaction, S_NAr reaction, cross-coupling reaction or C(sp²)-H activation can serve as a good solution. However, these reactions usually need pre-activation, pre-functionalization of the arene, directing group assistance, or extensive reaction condition optimization for satisfactory selectivity and reactivity.^[11] Therefore, the development of efficient and practical methods for the rapid synthesis of polysubstituted salicylaldehyde derivatives would be a challenging but promising project.

Maltol and kojic acid, pyrone-containing natural products, have been widely used in oxidopyrylium-based [5+2] cycloaddition (Scheme 2).^[12] In addition to the synthetic methodology, Wender and co-workers also applied this method to the landmark total syntheses of phorbol and C6,C7-epi-yuanhuapin.^[13] In these [5+2] cycloaddition reactions, the pyrone moiety served as a five-carbon synthon to assemble the oxa-bridged bicyclic system. Herein we disclose an entirely new reaction type for pyrone, a cascade Claisen rearrangement of well-organized maltol propargyl ether for the rapid synthesis of polysubstituted salicylaldehyde through a cut-and-sew strategy,^[14] where the aromatic pyrone was torn apart and then fused into benzene ring. The reaction is a catalyst-free process and proposed to occur through the 1,6-Michael addition of *p*-quinone-methide (*p*-QM) intermediate. This protocol featured high atom economy (100%) and gram-scale conditions (Scheme 2).

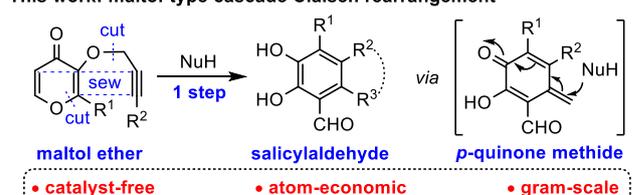
[5+2]-cycloaddition: Woods, Garst, Wender, Mascarenas, McDonald, Li



Wender's work: [5+2]-cycloaddition in total synthesis



This work: maltol-type cascade Claisen rearrangement



Scheme 2. Pyrone-based transformations.

First, maltol propargyl ether **1a**, which can easily be prepared from naturally occurring maltol and propargyl bromide in only one step with a 94% yield, was chosen as the model substrate for this investigation. Styrene was used as the nucleophile to trap the proposed *p*-QM intermediate (Scheme 2).^[15] As shown in Table 1,

[*] Mr. T. Cao, Mr. Y. Kong, Mr. K. Luo, Dr. L. Chen, and Prof. S. Zhu
School of Chemistry and Chemical Engineering
South China University of Technology, Guangzhou 510640 (China)
E-mail: zhuf@scut.edu.cn

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Table 1. Optimization of the reaction conditions.^[a]

Entry	Cat.	Add.	Sol.	T (°C)	2a (%)
1	Ph ₃ PAuCl	AgBF ₄	DCE	60	ND
2	Ph ₃ PAuCl	AgBF ₄	DCE	120	ND
3	Ph ₃ PAuCl	NaBARF	DCE	120	ND
4	Ph ₃ PAuCl	NaBARF	MeCN	150	53
5	-	-	MeCN	150	59
6	-	-	DCE	150	67 ^[b]
7	-	-	PhMe	150	64 ^[b]
8	-	-	PhCl	150	75 ^[b]

[a] Unless otherwise noted, the reaction was performed with **1a** (0.25 mmol) and styrene (0.88 mmol) for 13 h under the N₂ atmosphere. The yield was determined by ¹H NMR. NaBARF: Na[B(3,5-(CF₃)₂C₆H₃)₄]. [b] Isolated yield.

gold salts were initially utilized as the catalyst owing to their high efficiency in promoting the transformation of alkynes through activating the C≡C bond.^[16] However, no product was detected at 60 °C, and the reaction didn't work even at an elevated temperature of 120 °C with Ph₃PAuCl/AgBF₄ or Ph₃PAuCl/NaBARF as the catalyst (Table 1, entries 1-3). Encouragingly, a 1,6-Michael addition/Friedel-Crafts product **2a** was detected in 53% yield when the temperature was further raised to 150 °C with MeCN as the solvent (entry 4). However, the control reaction without the gold salt proceeded equally well, which indicated that this reaction is a non-catalytic thermolysis process (entry 5). In the absence of catalysts and additives, the yield of **2a** was further enhanced to 75% by variation of solvent (entries 6-8, see SI for more details).

With the optimized conditions (Table 1, entry 8) in hand, the substrate scope was then explored. As shown in Table 2, this cascade reaction was successfully extended to propargyl ethers **1** derived from different maltols. For example, the propargyl ethers from ethyl maltol and isopropyl maltol were transferred to the products **2b** and **2c** in 74% and 50% yields, respectively. It is noted

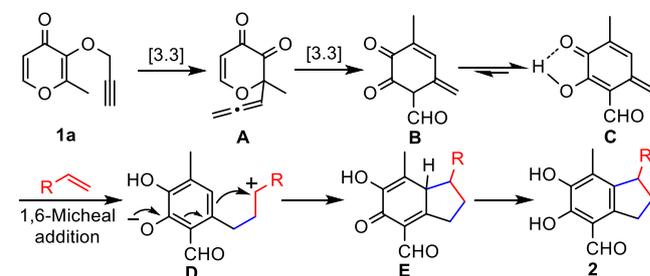
Table 2. Evaluation of various alkenes and maltol ethers.^[a]

2a , R = Me, 75% 77% (1 g scale) 70% (10 g scale)	2d-p	2d (R ³ = 4-Me, 71%) 2e (R ³ = 4- ^t Bu, 70%) 2f (R ³ = 4-OMe, 52%) 2g (R ³ = 4-F, 56%) 2h (R ³ = 4-Cl, 58%) 2i (R ³ = 4-Br, 61%) 2j (R ³ = 4-NO ₂ , 39%, X-ray) 2k (R ³ = 4-CF ₃ , 42%) 2l (R ³ = 3-Br, 53%) 2m (R ³ = 2-Br, 56%) 2n (R ³ = 2-Cl, 63%) 2o (R ³ = H, R ¹ = Me, 56%) 2p (R ³ = H, R ¹ = Ph, 63%)
2b , R = Et, 74% (1.8 g scale)		
2c , R = ⁱ Pr, 50%		
2q , 35%	2r , ND	2s , 85%
2t , 54% (X-ray)	2u , 56%	2v ^[b] , 48% (X-ray)

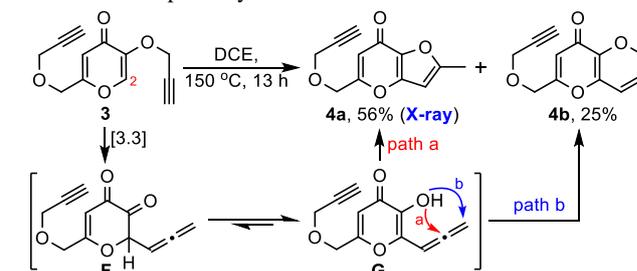
[a] Reaction conditions: **1** (0.25 mmol) and alkene (0.88 mmol), under N₂; [b] estrone-derived styrene (0.2 mmol) and **1a** (0.7 mmol).

that the gram-scale reactions proceeded smoothly as well, giving the desired products in good yields (**2a-b**: 1-10 gram scale). Furthermore, the reactions of maltol propargyl ether **1a** with a variety of alkene substrates as nucleophiles were carried out (**2d-v**). In addition to styrene, various styrene derivatives effectively reacted with **1a**, furnishing **2d-n** in 39-71% yields. Both electron-rich and electron-poor styrene derivatives functioned well to afford the desired fully substituted salicylaldehydes. The results indicated that electron-rich styrenes were better substrates and gave the products in higher yields. When gem-substituted styrenes were used, the products with a quaternary carbon center were formed in moderate yields (**2o-p**). The sterically congested spiro-compound was tolerated as well, albeit with a lower yield (**2q**). However, the simple aliphatic alkene did not work (**2r**). The more electron-rich conjugated diene and enyne were investigated as well, furnishing the corresponding products in good yields (**2s-u**). It's worth mentioning that estrone-derived styrene was also a good substrate for this reaction, delivering the desired product **2v** in 48% yield. In addition, we have tried both allyl trimethylsilane and allyl boronic acid pinacol for this reaction, but no desired products were observed. Internal alkyne ethers were also treated with styrene under the standard conditions, however, it only led to complicated mixtures (see details in SI). The structure of compounds **2j**, **2t**, and **2v** was confirmed by X-ray diffraction analysis.

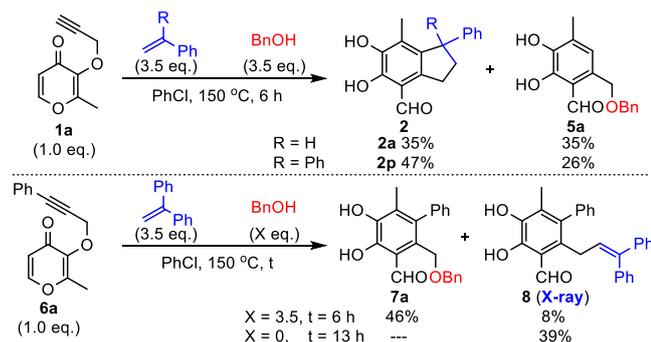
A plausible mechanism is proposed in Scheme 3. The initial dearomatic propargylic-Claisen rearrangement establishes a well-organized 1,5-ene-allene intermediate **A**, which then undergoes a second allenic-Claisen rearrangement to furnish the aldehyde **B**. After tautomerization, the key intermediate *p*-QM **C** is formed and trapped by an intermolecular 1,6-Michael addition of an alkene to give **D**, followed by an intramolecular Friedel-Crafts reaction and 1,5-hydrogen shift to deliver the desired product **2**.

**Scheme 3.** Proposed mechanism.

To investigate the necessity of the substituent at the C2-position, a propargyl ether of kojic acid **3** was then synthesized (Scheme 4). Consistent with Elmore's results,^[17] the furo[3,2-*b*]pyrone **4a** was formed in 56% yield when the reaction was conducted in the absence of trapping reagent. At the same time, an unexpected product chromone **4b** was also obtained in 25% yield. Compounds **4a** and **4b** might come from the cyclization of the rearomatized allene intermediate **G**. These results demonstrated the importance of the anchor group at C2-position, which might block the rearomatization pathway.

**Scheme 4.** Investigation of the substituent effect at C2-position.

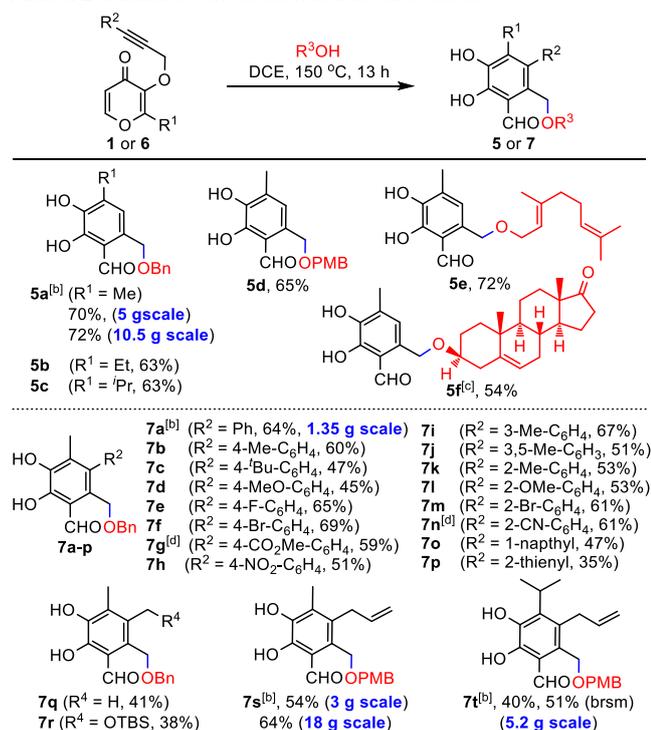
In order to trap the cationic intermediate **D**, benzyl alcohol, which might act as a nucleophile, was added to the reaction mixture of **1a** and styrene. A mixture of **2** and **5a** was ultimately furnished rather than the desired three-component adducts. Fortunately, when internal alkyne ether **6a** was used, **7a** and **8** were obtained in 46% and 8% yields, respectively. The yield of **8** was increased to 39% when no alcohol was added. **8** might come from the trapping of **6a**-derived *p*-QM intermediate by 1,1-diphenylethylene to form a relatively stable cationic intermediate, followed by an elimination process. Although we failed to directly trap **D**, the generation of **8** demonstrated that a **D**-like cationic intermediate was involved in the cascade process (Scheme 5).



Scheme 5. Attempts to trap the cationic intermediate.

Based on the above observation, we then moved to trap the proposed *p*-QM intermediate with different alcohols, aiming for penta- and hexa-substituted salicylaldehyde derivatives, which might be used as a key precursor for the total synthesis of hemigossypol and gossypol. Considering the benzyl ether could be easily deprotected, benzyl alcohol (Bn-OH) was initially tested as the nucleophile. As expected, the desired pentasubstituted salicylaldehyde products **5a-c** were produced in 63–72% yields from terminal propargyl ethers **1** derived from different maltols.

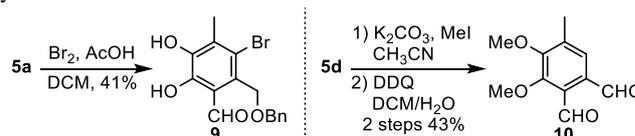
Table 3. Evaluation of various alcohols and maltol ethers.^[a]



[a] Reaction conditions: **1** or **6** (0.2 mmol) and alcohol (0.7 mmol), DCE, 150 °C, 13 h, under N₂; [b] PhCl as solvent; [c] prasterone (0.2 mmol) and **1a** (0.7 mmol); [d] 140 °C, 24 h; Bn: benzyl group; PMB: *p*-methoxybenzyl.

The reaction was easily scaled up to 10-gram quantities without loss of yield (**5a**). *p*-Methoxybenzyl alcohol (PMB-OH) was also a good nucleophile for this reaction, affording the desired PMB-protected ether **5d** in 65% yield. Naturally occurring alcohols geraniol and prasterone were also employed to trap the transient Michael acceptor, leading to the products **5e** and **5f** in 72% and 54% yields, respectively. What's more, fully substituted salicylaldehyde products **7a-t** were furnished with internal propargyl maltol ethers **6** as the substrates and Bn-OH or PMB-OH as the nucleophile. As shown in table 3, the reactions proceeded smoothly when the propargyl group was capped with different aryl groups (**7a-p**). It seems that the reaction was not very sensitive to the electronic properties of the aryl groups (**7a-n**), with the product yields ranging from 45% to 69%. The structure of product **7g** was unambiguously confirmed by X-ray diffraction analysis. The naphthyl-, thienyl- and alkyl-substituted alkynes were tolerated as well, giving the desired products (**7o-r**) in slightly lower yields. Products **7s** and **7t** bearing an allylic handle group, aimed at the total synthesis of gossypol and its analogues, were obtained on a large scale in an acceptable yield.

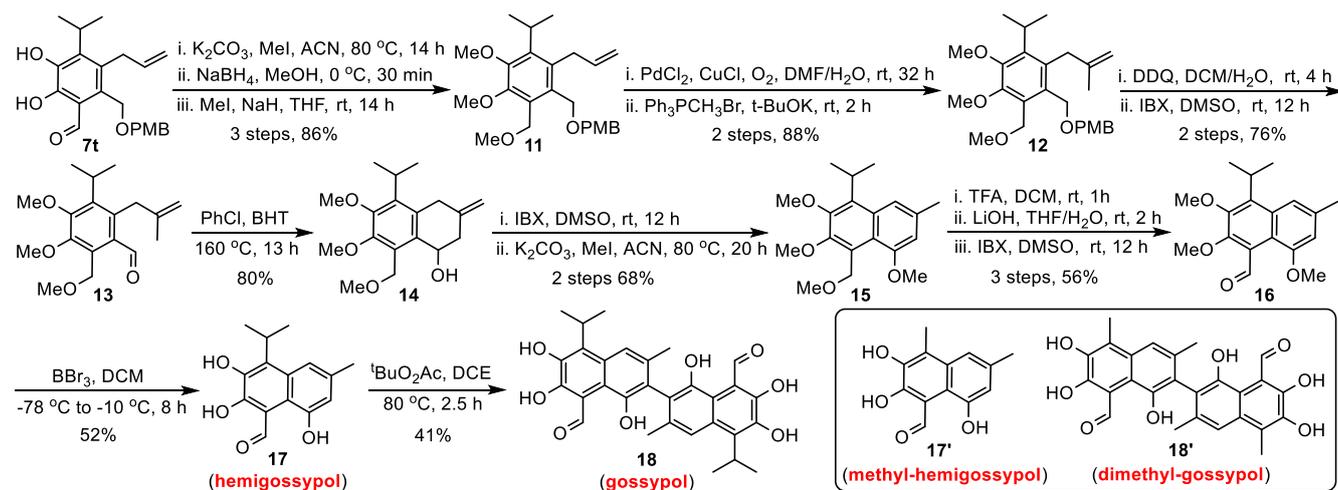
In the efforts to demonstrate the synthetic utility of this procedure, we took advantage of the highly substituted parent benzene for further transformations. As shown in Scheme 6, product **5a** was transferred to the bromo-salicylaldehyde **9** in 41% yield with Br₂/HOAc as the brominating reagent. Furthermore, the deprotection and oxidation of **5d** gave rise to aldehyde **10** in 43% yield.



Scheme 6. Further transformation of products **5**.

To further illustrate the practicability of this unique methodology, we began the journey of the total synthesis of hemigossypol, gossypol and their analogues (Scheme 7). The synthesis started with methylation and reduction of fully substituted salicylaldehyde **7t** to furnish **11** in 86% yield over three steps. After Wacker oxidation and a Wittig reaction of **11**, the alkene **12** was obtained in 88% yield. Deprotection of **12** with DDQ and subsequent oxidation with IBX gave aldehyde **13** in 76% yield. An efficient intramolecular Alder-ene reaction occurred for the aldehyde **13** to assemble the bicyclic tetrahydronaphthol **14** in 80% yield, which differed from the previous synthetic strategy by using a Friedel-Crafts reaction^[18-19] to construct the naphthalene skeleton. Afterward, oxidation and methylation took place to generate naphthalene **15** in 68% total yield. According to the reported protocol,^[18c] the methyl ether **15** was selectively deprotected, followed by IBX oxidation to provide **16**, which was then globally demethylated by BBr₃ to deliver hemigossypol **17** in 52% yield. The endgame to complete the total synthesis of gossypol **18** was achieved by treating **17** with ^tBuO₂Ac under nitrogen at 80 °C for 2.5 hours in 41% yield. The spectral data of synthetic **17** and **18** was in full agreement with those reported for these natural products. It's worth mentioning that the methyl analogues of hemigossypol **17'** and gossypol **18'** were also obtained from polysubstituted salicylaldehyde **7s** following a similar procedure (see SI for details).

Previously several groups realized the total synthesis of gossypol. In 1958, Edwards reported the first synthesis using a late-stage formylation strategy (9 steps).^[18a, 19a] In 1997, Meyers achieved the first asymmetric total synthesis of (*S*)-(+)-gossypol (23 steps and 8.71% yield), highlighted by a chiral oxazoline-induced diastereoselective Ullmann coupling.^[18b] Recently, Wang developed a practical route to gossypol from commercially available



Scheme 7. Total synthesis of hemigossypol, gossypol and their analogues.

carvacrol (19 steps and 6.67% yield), which featured an oxidative phenolic dimerization.^[18c] Our synthesis with polysubstituted salicylaldehyde **7t** as the starting material successfully preinstalled all the required functional groups on the phenyl ring, which are otherwise difficult to access.^[18c,19h] Although the inevitable protection/deprotection of these groups resulted in a slightly lengthy procedure (15 steps and 3.73% yield), the intramolecular carbonyl-ene reaction for the rapid synthesis of the polysubstituted naphthol skeleton added another highlight to this total synthesis.

In summary, we have disclosed a novel reaction type for pyrone, the cascade Claisen rearrangement of well-organized maltol propargyl ether for the rapid synthesis of polysubstituted salicylaldehyde through a cut-and-sew strategy, where the aromatic pyrone was torn apart and then fused into benzene ring. This reaction is a catalyst-free process and proposed to go through the cascade dearomatic propargylic-Claisen rearrangement/allenylic-Claisen rearrangement/1,6-Michael addition. It featured high atom economy (100%) and easy scale-up (up to 18-gram quantities). Based on this methodology, we also realized the total synthesis of hemigossypol, gossypol, and their analogues, which were highlighted by maltol-type cascade Claisen rearrangement and intramolecular Alder-ene reaction. Benefiting from the obvious advantages, this method holds great potential for the synthesis of polyphenolic natural products.

Acknowledgments

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Keywords: Claisen rearrangement • gram-scale • atom economy • catalyst-free • gossypol

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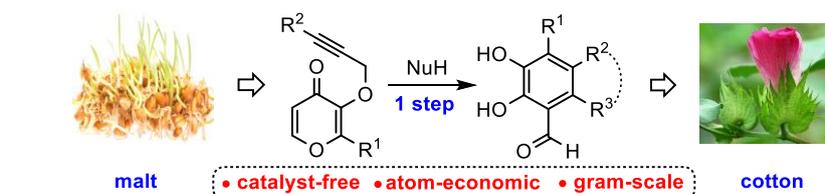
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Cascade Claisen Rearrangement

Tongxiang Cao, Yi Kong, Kui Luo,
Lianfen Chen, and Shifa Zhu*

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Cascade Claisen Rearrangement: Rapid
Synthesis of Polysubstituted
Salicylaldehydes and Total Synthesis of
Hemigossypol and Gossypol



A cascade Claisen rearrangement of well-organized maltol propargyl ether for the construction of polysubstituted salicylaldehydes was reported. This reaction featured high atom economy (100%), as well as catalyst-free and gram-scale conditions. Based on this novel methodology, we have realized the total synthesis of hemigossypol, gossypol, and their analogues.

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