

Direct Synthesis of τ -Form Metal-Free Phthalocyanine from
Phthalodinitrile by Seeding Procedure

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Direct synthesis of τ -form metal-free phthalocyanine (τ -H₂Pc) was performed from phthalodinitrile by using of 1,8-diaza-bicyclo(5,4,0)undecene-7 (DBU) and crystal nucleus in alcohol.

The yield of τ -H₂Pc was investigated under various conditions.

Phthalocyanines have been recongnized as important organic materials for its photoconductivity. Well-known examples of phthalocyanines exhibiting polymorphism are metal-free phthalocyanine (H₂Pc) and copper phthalocyanine (CuPc). Both of them have two representative polymorphic form, the β -form is thermally stable and α -form is thermally unstable. In addition, the middle energy stages like as X¹⁾-form H₂Pc and δ , ϵ , R and X-form CuPc have been reported up to this time.

Recently, a new polymorph of metal-free phthalocyanine, designated as τ -H₂Pc, has been reported.^{2,3)} The τ - and X-H₂Pc were obtainable from the α -H₂Pc under appropriate conditions of crystalline conversion in the process of milling, that is, we could not produce the middle energy stages of H₂Pc by direct synthesis from phthalodinitrile. Baumann et al.⁴⁾ and Borodkin⁵⁾ studied α - or β -H₂Pc with an alkoxide, and Tomoda et al.^{6,7)} did it with an organic strong bases such as 1,8-diaza-bicyclo(5,4,0)undecene-7 (DBU) or 1,5-diaza-bicyclo(4,3,0)nonene-5 (DBN). So we tried direct synthesis of τ -H₂Pc by using of DBU and crystal nucleus (τ -H₂Pc), consequently, we were able to attain our purpose.

A typical experimental procedure is as follows. The τ -form H₂Pc was prepared by the crystal transformation of α -H₂Pc using wet milling method as previously described.²⁾ Phthalodinitrile (76.8 g, 0.6 mol) was dissolved in alcohol and then DBU (91.2 g, 0.6 mol) was added dropwise to the solution at the refluxing temperature or 100 °C (Table 1). The τ -H₂Pc as seeding crystals (2 g, 3.9×10^{-3} mol : prepared by wet milling) was added by slow degrees (1 g / 5 min), when the temperature reached 60 °C. After 18 h of heating, the products were collected by filtration, washed with water, and purified by extraction technique with acetone.

Crystalline form of the products were checked by X-Ray diffraction patterns using Rotafrex RU-200 (CuK α : Rigaku Denki Ltd.). The UV-Vis spectra were measured in a dispersion in tetrahydrofuran (THF) using Recording Spectrophotometer UV-365 (Shimadzu Seisakusho Ltd.). The X-Ray photoelectron spectra (XPS) were measured using Perkin-Elmer PHI-5400 ESCA system with MgK α X-Ray target. The shape of the particles were observed by scanning electron microscope (SEM). Table

1 shows the yield of H₂Pc from phthalodinitrile under various conditions.

The yield of H₂Pc increased with increasing the length of alkyl radical of alcohol and reaction temperature. In a secondary alcohol, the yield of H₂Pc was low. On the contrary, using methyl alcohol which was more than twice as effective as using ethyl or n-propyl alcohol. The yield was influenced by reaction temperature and the length of alkyl radicals. In the case of secondary or tertiary alcohol, the formation of τ -H₂Pc was little occurred except isopropyl alcohol. This results suggested that the formation of alkoxide ion was difficult. On occasion of methyl alcohol, the electric constant was larger than any other alcohol, consequently, the formation of alkoxide ion was promoted and a large amount of 1-alkoxy-3-iminoisoindolenine was formed. Thus a good yield was attained. In the case of 2-alkoxyethanol, the yield of τ -H₂Pc increased the same tendency, however, there were a little effective. Figure 1 shows the dependence of the yield of τ -H₂Pc using 2-butoxyethanol on the reaction time and reaction temperature. That value increased with an increase in reaction time and reaction temperature. When reaction temperature was under 100 °C, and added H₂Pc as seeding crystal, we could obtain the pure τ -H₂Pc. And, we could not obtain the τ -H₂Pc without adding seeding crystal in this work. The maximum yield of τ -H₂Pc with 2-butoxyethanol may therefore presume about 55wt%.

X-Ray diffraction patterns are shown in Fig.2. The pattern of direct synthesis (a) was quite alike as that of wet milling (b). UV-Vis absorption spectra are shown in Fig. 3. The spectra of (a) and (b) showed quite alike peaks due to scattering. The wavelength of absorption peaks were almost same. The absorbance of (b) was weaker than (a), however, the Q-band transition at ca. 655 and 690 nm were very sensitive. The Q-band transition is a $\pi - \pi^*$ transition of lower energy. Consequently, τ -H₂Pc of direct synthesis is more sensitive and oriented of nearest neighbor Pc molecule in THF. The split of absorption peaks in the Q-

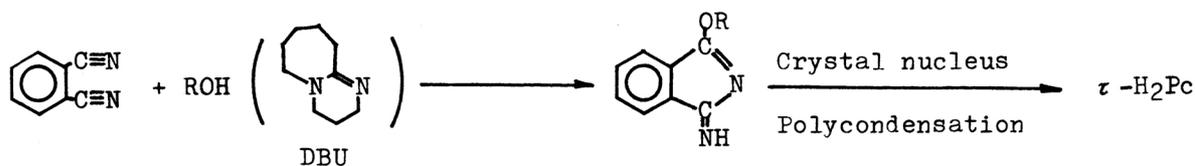


Table 1. Yield of τ -H₂Pc from phthalodinitrile under various conditions

Alcohol	Temp °C	Time h	Yield wt%
CH ₃ OH	bp (64.5)	18	50
C ₂ H ₅ OH	bp (78.3)	18	24
(CH ₃) ₂ CHOH	bp (82.4)	18	15
n-C ₃ H ₇ OH	bp (97.0)	18	26
n-C ₆ H ₁₃ OH	100	18	48
CH ₃ O(CH ₂) ₂ OH	100	18	36
C ₂ H ₅ O(CH ₂) ₂ OH	100	18	35
C ₄ H ₉ O(CH ₂) ₂ OH	100	18	55
C ₆ H ₁₃ O(CH ₂) ₂ OH	100	18	60

band are because the reduction of symmetry in τ -H₂Pc molecule from D_{4h} to D_{2h} by removal of the molecule center. The XPS of N1s for τ -H₂Pc are shown in Fig. 4.

There are two kinds of peaks which indicates at least two chemically different nitrogen atoms.⁸⁾ The spectrum of (b) has two peaks with almost equal intensity.

The higher energy peak of (a) is smaller than the lower energy peak. Namely, the spectra of (a) and (b) suggested the bonded structure with hydrogen of central nitrogen for D_{4h} and bridged structure for D_{2h},

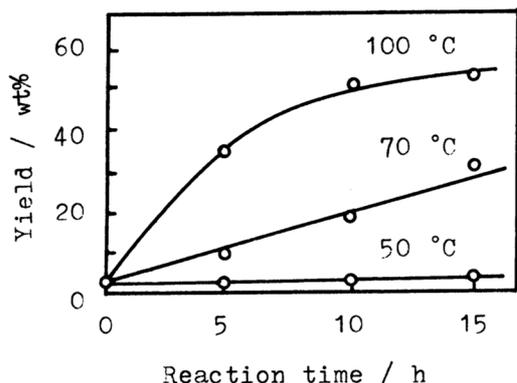


Fig. 1. Dependence of the yield of τ -H₂Pc on the reaction time and reaction temperature.

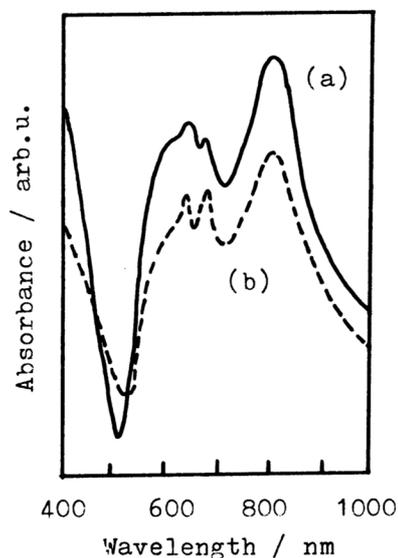


Fig. 3. UV-Vis absorption spectra of τ -H₂Pc. (a) wet milling
(b) direct synthesis
cf. Table 1 100 °C x 18 h
Solvent : butoxyethanol

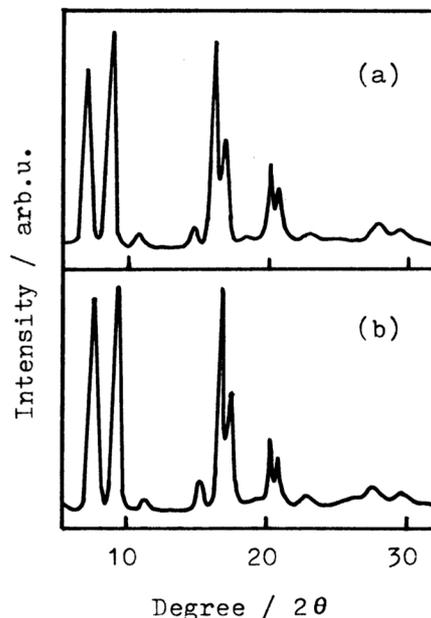


Fig. 2. X-Ray diffraction patterns of τ -H₂Pc. (a) wet milling
(b) direct synthesis
cf. Table 1 100 °C x 18 h
Solvent : butoxyethanol

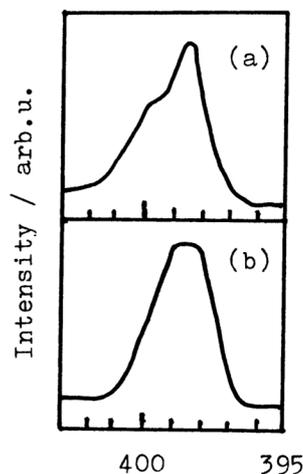


Fig. 4. The nitrogen 1s spectra of τ -H₂Pc. (a) wet milling
(b) direct synthesis
cf. Table 1 100 °C x 18 h
Solvent : butoxyethanol

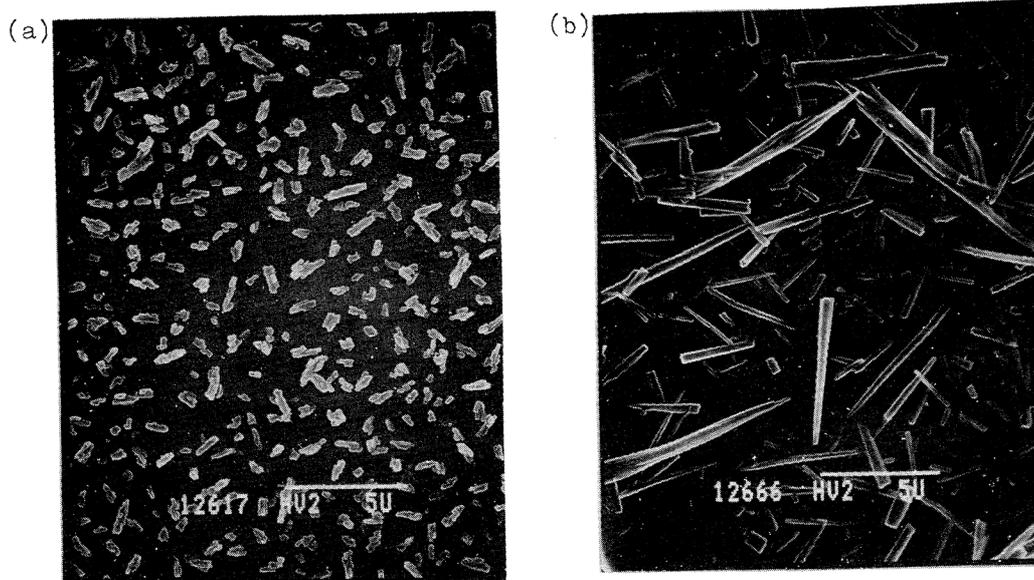


Fig. 5. Scanning electron microscope photographs of τ -H₂Pc.
 (a) wet milling (b) direct synthesis
 The bars in the photographs represent 5 μ m.

respectively. These results were supported the results of UV-Vis spectra. SEM photographs are shown in Fig. 5. The particle size by wet milling (a) are uniform, however, the particles by direct synthesis are rod-like and various size. Therefore, the crystallinity of τ -H₂Pc by direct synthesis may be relatively high because of its mild synthetic conditions. Further investigation of the relationship between synthetic conditions and crystallinity containing in the process of crystal growth in τ -H₂Pc and now in progress.

The authors are grateful to Mr. Kakuta (Hitachi Ltd.) for valuable suggestions, and to Mr. Ito, Mr. Morohashi, Miss Hotta and Mr. Sano for their assistance of these experiments.

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(Received September 29, 1987)