Compound	$m/z \ (I_{\rm rel} \ (\%))$		
	[M+SiMe <sub>3</sub> ] <sup>+</sup>	[M+SiMe <sub>3</sub> -NO <sub>2</sub> ] <sup>+</sup>	[M+SiMe <sub>3</sub> -XNO <sub>2</sub> ] <sup>+</sup>
1,1-Dinitroethane	193 (45.5)	147 (40.5)	146 (100)
1-Fluoro-1,1-dinitroethane	211 (99.5)	165 (61.8)	146 (100)
1-Chloro-1,1-dinitroethane	227 (67.9) 229 (24.4)	181 (100) 183 (42.7)	146 (67.5)

**Table 2.** CI mass spectra of 1,1-dinitroethane and 1-halo-1,1-dinitroethanes  $MeCX(NO_2)_2$  (with tetramethylsilane as the reagent gas)

Other routes of fragmentation, analogous to the reactions of decomposition of the protonated molecular ions of nitroalkanes (elimination of  $H_2O$  and HNO),<sup>2</sup> were not observed in the case of ion-adducts of mononitroalkanes with the trimethylsilyl cation.

Unlike those of mononitroalkanes, the CI mass spectra of dinitroethanes contain ion-adducts of only one type,  $[M+SiMe_3]^+$  (Table 2), which decompose with the loss of a stable NO<sub>2</sub><sup>-</sup> radical similarly to the corresponding protonated molecular ions (*cf.* Ref. 1). By analogy with the  $[M+H-NO_2]^+$  ions, the structure of MeC<sup>-</sup>XN<sup>+</sup>O<sub>2</sub> · SiMe<sub>3</sub> can be ascribed to the  $[M+SiMe_3-NO_2]^+$  ions.

The another reaction path of fragmentation of the  $[M+SiMe_3]^+$  ions of dinitroethanes, with the loss of an  $XNO_2$  (X = H, F, or Cl) molecule, has no analogs among the fragmentation processes of their protonated molecular ions.

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# Synthesis of $6\alpha$ -methyl- $16\alpha$ , $17\alpha$ -cyclohexanoprogesterone via $\gamma$ -methylenation of $16\alpha$ , $17\alpha$ -cyclohexanopregn-4-ene-3, 20-dione

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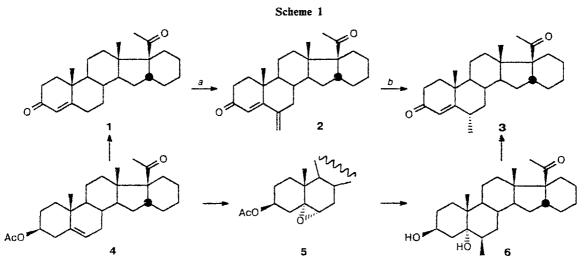
6-Methylene-16 $\alpha$ , 17 $\alpha$ -cyclohexanopregn-4-ene-3, 20-dione 2 has been synthesized by the reaction of  $\Delta^4$ -3-ketone 1 with CH<sub>2</sub>(OEt)<sub>2</sub> and POCl<sub>3</sub> in the presence of AcONa in 55% yield. Reduction of the product 2 in the presence of 5% Pd/C gives 6 $\alpha$ -methyl-16 $\alpha$ , 17 $\alpha$ -cyclohexanoprogesterone 3 in a yield exceeding 70%.

Key words: methylenation, pentarane, progestin.

 $6\alpha$ -Methyl-1 $6\alpha$ , 17 $\alpha$ -cyclohexanoprogesterone ( $6\alpha$ -methyl-D<sub>6</sub>-pentarane) (3) (Scheme 1) is a synthetic progestin that exhibits high progestational and contraceptive activities.<sup>1</sup> Earlier for the synthesis of compound 3, the succession of transformations  $4 \rightarrow 5 \rightarrow 6 \rightarrow 3$  (see Scheme 1) was used,<sup>2,3</sup> in which the key step was the

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**Reagents and conditions:** a.  $CH_2(OEt)_2$ ,  $CHCl_3$ ,  $POCl_3$ , NaOAc; 60–75 °C; yield 55%; b. 5% Pd/C, cyclo-C<sub>6</sub>H<sub>10</sub>, EtOH; 28 h; dilute HCl; yield 73%.

ring opening of  $5\alpha$ , $6\alpha$ -oxide 5 with methylmagnesium iodide; in this case, the reaction gave exclusively  $6\beta$ -methyldiol **6** while the sterically hindered 20-carbonyl group in molecule of 5 was not involved in the reaction.<sup>2</sup> However, this method of introducing the 6-methyl group required the preliminary protection of the 20-carbonyl group when it was used for other progestins of the D'-pentarane series, for example,  $16\alpha$ , $17\alpha$ -cyclobutanoprogesterone.<sup>4</sup>

In this communication we report an alternative synthesis of 6a-methyl-16a,17a-cyclohexanoprogesterone 3. The key intermediate in this synthesis is 6-methylene derivative 2, which was obtained by  $\gamma$ -methylenation of  $\Delta^4$ -3-oxopentarane 1 by the Wiechert method.<sup>5</sup> It should be noted that such a one-step procedure for introducing a methylene group into steroid conjugated ketones has been rather widely used for the synthesis of methylene steroids of various classes, 5-7 and the yields of the final product were 60-65%. The starting conjugated ketone 1 was obtained by Oppenauer oxidation of the corresponding  $3\beta$ -hydroxy- $\Delta^5$ -derivative.<sup>2</sup> The interaction of compound 1 with reagents  $X-CH_2OR$  (X = OR, Cl, OAc, R = Me, Et) and POCl<sub>3</sub> in the presence of AcONa led directly to 6-methylene derivative 2. Formaldehyde diethylacetal (ethylal) proved to be the best reagent in this case. When a mixture of ketone 1 with an excess of ethylal, POCl<sub>3</sub>, and AcONa in anhydrous  $CHCl_3$  was boiled, compound 2 was obtained; the yield was 55% after chromatographic purification. When ethylal was replaced by methylal, the yield of dione 2 decreased to 20-25%; when chloromethyl methyl ether was used, the obtained compound 2 was contaminated by chlorinated unidentified by-products.

Reduction of the 6-methylene group of dione 2 to the target product 3 in the presence of 5% Pd/C and cyclohexene as a hydrogen donor proceeds via the step of isomerization of the exocyclic double bond in molecule 2 into ring *B* to give the intermediate  $\Delta^{4.6}$ -6-methyl derivative.<sup>8</sup> The latter is further reduced into a mixture of stereoisomeric  $6\alpha,\beta-\Delta^4$ -3-ketones, which are easily transformed into  $6\alpha$ -methylpentarane 3 by treatment with an acid. The course of the reaction was monitored by UV spectroscopy:  $\lambda_{max}$  changed from 260 nm (6-methylene- $\Delta^4$ -3-ketone) via  $\lambda_{max} = 285$  nm ( $\Delta^{4.6}$ -3-ketone) to  $\lambda_{max} = 245$  nm ( $\Delta^4$ -3-ketone).

Thus, this communication describes a two-step method for the synthesis of  $6\alpha$ -methyl- $16\alpha$ , $17\alpha$ -cyclohexanoprogesterone from its 6-desmethyl precursor. In many cases this method can compete with the alternative methods for the introduction of a 6-methyl group into pentaranes.

#### Experimental

Melting points were determined on a Boetius heating plate. <sup>1</sup>H NMR spectra were recorded in CDCl<sub>3</sub> on a Bruker WM-250 spectrometer. UV spectra were obtained on a Unicam SP-700 instrument. The qualitative analysis of the mixtures was carried out by TLC on Silufol L UV-254 plates (Czech Republic). Columns with Woelm silica gel were used for preparative chromatography.

6-Methylene-16a, 17a-cyclohexanopregn-4-ene-3, 20-dione (2). A suspension of compound 1 (1 g), anhydrous AcONa (1 g), and freshly distilled POCl<sub>3</sub> (30 mL) in 25 mL of anhydrous CHCl<sub>3</sub> was stirred under argon for 45 min at 65-70 °C. A saturated aqueous solution of sodium carbonate (100 mL) was then added dropwise to the reaction mixture cooled to 18-20 °C and stirred for an additional 1.5 h. The aqueous layer was separated, and the organic layer was thoroughly washed with water until the reaction became neutral and dried with Na<sub>2</sub>SO<sub>4</sub>. The crystalline residue obtained after removal of the solvent was chromatographed on a column. Dienedione 2 (0.57 g, 55%), m.p. 181-185 °C (ether-hexane) was isolated by elution with a heptane—ether mixture (4 : 1). UV,  $\lambda_{max}/nm$ : 260 ( $\epsilon$  11200). <sup>1</sup>H NMR,  $\delta$ : 0.72 (s, 3 H, 18-Me); 1.09 (s, 3 H, 19-Me); 2.14 (s, 3 H, 21-Me); 3.0 (m, 1 H, C(16)H); 4.94, 5.06 (2 m, 2 H, C(6)CH<sub>2</sub>); 5.92 (s, 1 H, C(4)H). Further elution gave the starting ketone 1 (0.22 g).

 $6\alpha$ -Methyl-16 $\alpha$ ,  $17\alpha$ -cyclohexanopregn-4-ene-3, 20-dione (3). A mixture of a solution of product 2 (0.45 g) in 25 mL of EtOH, 5% Pd/C (0.3 g), and 2.5 mL of freshly distilled cyclohexene was boiled with stirring for 28 h. During this period cyclohexane was added three times ( $3\times1.5$  mL). The catalyst was filtered off and washed with ethanol. The filtrate was acidified with dilute HCl, and most of the ethanol was removed *in vacuo*. Ice was added to the residue, and the white powder that precipitated was filtered off, washed with water on a filter, and air dried. Chromatographic purification afforded enedione 3 (0.33 g, 73%), m.p. 175-178 °C, which gave no melting point depression with an authentic sample.<sup>9</sup>

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## Bis-para-semiquinoid type double cyclopalladation in the series of six-membered two-nitrogen bridged annulene-dihydroannulene ligands

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The cyclometallation reaction known in the series of annulene heteroorganic ligands was extended to six-membered dihydroannulenes with a dinitrogen bridge. Double cyclopalladation of 4-methyl-4-trichloromethylcyclohexa-2,5-dien-1-one azine yielded the first representative of a new class of cross-conjugated diazadipalladatetracycles, viz., 5,10-bis(acetylacetonato)-2,7-dimethyl-2,7-bis(trichloromethyl)-2,7-dihydro-4b,9b-diaza-5,10-dipalladaindeno[2,1-a]indene, isolated as a diastereomer mixture of the achiral meso-form (*E*-isomer) and a racemate (*Z*-isomer). This reaction offers a method for transition metal-mediated activation of non-reactive C—H bonds at position 2 of cyclohexa-2,5-dienylidene systems and a route toward the very rare chiral polyheteroelement system with rotational symmetry.

Key words: azines of  $\alpha,\beta$ -unsaturated carbonyl compounds, 2,5-cyclohexadienones, use of Pd<sup>II</sup> derivatives in fine organic synthesis, double cyclopalladation, unsaturated metallaheterocycles, organometallic compounds, stereoisomerism, chirality.

The reaction of cyclometallation (also known as *ortho*-metallation) producing  $\pi,\pi$ -conjugated diazametalladi- and tetracycles (1<sup>1</sup>, 2<sup>2</sup>; cf. also Ref. 3) was discovered at the end of the 1960s for the benzenoid compound, azobenzene 3, in which both six-membered rings are aromatic and the N atoms are linked by a  $\pi$ -bond (cf. the recently discovered reaction of double cyclometallation of azines of benzaldehyde<sup>4</sup> and benzophenone<sup>5</sup>). In the present work we for the first time extended the annulene type of transformation to a series of six-membered two-nitrogen bridged dihydroannulenes, azines of 4,4-disubstituted 2,5-cyclohexadienones (4).

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