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Synthesis of a tetrasaccharide related to the triterpenoid saponin isolated from *Schima noronhae*

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ABSTRACT

A concise synthesis of a tetrasaccharide related to the cell-growth inhibitory triterpenoid saponin isolated from *Schima noronhae* is reported. A late stage 2,2,6,6-tetramethylpiperidinyloxy (TEMPO)-mediated oxidation of a primary hydroxyl group to carboxylic acid has been achieved under phase-transfer conditions. Stereoselective glycosylations were carried out using thioglycoside or glycosyl trichloroacetimidate activation using sulfuric acid immobilized on silica (H₂SO₄–silica) in conjunction with *N*-iodosuccinimide and alone, respectively.

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Tetrahedron

1. Introduction

Saponins are glycosylated secondary metabolites synthesized by plants during their routine programme of growth and development.¹ They posses intense antifungal properties and are thus believed to act as the natural chemical barriers in plants against fungal attack.² In addition to their antifungal properties, saponins have been exploited as food crops, for example, legumes and oats or sources of drugs, for example, ginseng and liquorice making them commercially attractive for diverse reasons.³ However the details of the genetic machinery involved in the biosynthesis of these secondary metabolites have not yet been characterized. Despite the architectural diversity amongst the saponin structures isolated from different sources, one common feature that is shared by all saponins is the presence of a sugar chain attached to the 3-position of the aglycon moiety. These sugar chains are often branched and usually consist up to five sugar units. Normally the sugar units are selected from glucose, galactose, arabinose, xylose, rhamnose and glucuronic acid.⁴ It is believed that the glycosylation of the aglycon moiety occurs in the final stage of the saponin biosynthesis and the biological function of the saponin depends largely on the glycosylation pattern. Therefore, it is essential to elucidate the exact biosynthesis of the oligosaccharide chain and identify the enzymes involved in the whole process to realize the source of saponin bioactivity.⁵ For a better understanding of the glycosyltransferases involved and the order of events in the biosynthesis of saponins, the chemical synthesis of different sugar substrates will be useful.

Schima noronhae Reinw. ex Blume is a commonly grown plant in Southeast Asia and known as 'iju' in Japan.⁶ The plant is used as a source of timber and the leaves as fodder crops. They are also used in traditional medicines in Indonesia and Malaysia, for example, the astringent corollas are used to treat uterine disorders and hysteria and also as an ointment to alleviate the symptoms of smallpox.⁷ Recently, Ishibashi et al. reported two new triterpenoid saponins (Fig. 1) isolated from *S. noronhae* that showed potent cell-growth inhibitory activity against human cervical carcinoma cells (HeLa cells) and human colon carcinoma DLD1 cells.⁸ Herein we report the chemical synthesis of the tetrasaccharide side chain **1** (Fig. 1) of the triterpenoid saponins isolated from *S. noronhae* in the form of its propargyl glycoside.

2. Results and discussion

Careful consideration of the retrosynthetic analysis suggested that a 2+2 strategy would be best suited for the construction of the target tetrasaccharide. Rational protecting group manipulations on the commercially available monosaccharides followed by stereoselective glycosylations were planned to prepare the required disaccharide synthons. Further coupling would result in the formation of the tetrasaccharide framework. A late stage oxidation would furnish the required carboxylic acid moiety (Fig. 2). The choice of propargyl glycoside at the reducing end opens up the possibility of further glycoconjugate formation through click chemistry or other metal catalyzed multi-component reactions.

The synthesis of the disaccharide donor **8** was started from the known *p*-methoxyphenyl β -*p*-galactopyranoside **2**.⁹ The formation of the 4,6-benzylidene acetal using benzaldeylde dimethylacetal in the presence of catalytic CSA in acetonitrile¹⁰ followed by



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Figure 1. Structure of the triterpenoid and synthetic target.



Figure 2. Retrosynthetic analysis of the target oligosaccharides.

treatment with benzoyl chloride and pyridine in dichloromethane¹¹ afforded *p*-methoxyphenyl 4,6-*O*-benzylidene-2,3 di-*O*-benzoyl- β -D-galactopyranoside **3** in 81% yield. Hydrolysis of the benzylidene acetal using 80% AcOH at 80 °C¹² followed by selective protection of the primary hydroxyl group using TBDPS-Cl in pyridine¹³ afforded the acceptor, *p*-methoxyphenyl 2,3 di-*O*-benzoyl 6-*O*-tert-butyldiphenylsilyl β -D-galactopyranoside **4** in 87% yield. Acceptor **4** was then glycosylated with the known donor, *p*-tolyl 2,3,4-tri-O-acetyl-1-thio- α_{-L} -rhamnopyranoside **5**¹⁴ using NIS in the presence of H₂SO₄-silica to furnish the disaccharide **6** in 85% yield.¹⁵ Our previous experience with the oxidative cleavage of the *p*-methoxyphenyl group revealed that the OTBDPS moiety is not compatible with this transformation. Therefore, the TBDPS moiety was removed using Bu₄NF-THF¹⁶ to afford compound **7** in 87% yield and subsequently acetylated using Ac₂O in pyridine to afford the disaccharide **8** in 93% yield. Next, CAN-mediated oxidative

cleavage¹⁷ of the *p*-methoxyphenyl group followed by reaction with trichloroacetonitrile in the presence of DBU¹⁸ furnished the required disaccharide donor, 2,3,4-tri-O-acetyl- α -L-rhamnopyrano-syl-(1 \rightarrow 4)-6-O-acetyl-2,3-di-O-benzoyl- β -D-galactopyranosyl trichloroacetimidate **9** in 81% yield (Scheme 1).

In a separate experiment, a known propargyl 4,6-*O*-benzylidene-3-*O*-(4-methoxybenzyl)- β -D-glucopyranoside **10**¹⁹ was coupled with a known, *p*-tolyl 2,3,4,6-tetra-*O*-acetyl-1-thio- β -D-glucopyranoside **11**¹⁴ using NIS in the presence of H₂SO₄-silica at -30 °C. Once the completion of the glycosylation was confirmed by TLC (*n*-hexane–EtOAc, 3:1), a slight excess of H₂SO₄-silica was added and the reaction was allowed to warm up to room temperature. The acid-labile *p*-methoxybenzyl group was hydrolysed¹⁹ to furnish the required disaccharide acceptor, propargyl 2,3,4,6-tetra-*O*-acetyl-1thio- β -D-glucopyranosyl-(1 \rightarrow 2)-4,6-*O*-benzylidene-3-*O*-(4methoxybenzyl)- β -D-glucopyranoside **12** in 83% overall yield (Scheme 2).



Scheme 1. Synthesis of the disaccharide donor.



Scheme 2. Synthesis of the disaccharide acceptor.

Final glycosylation between disaccharide donor **9** and the disaccharide acceptor **12** using H_2SO_4 -silica at 0 °C afforded the tetrasaccharide **13** in 87% yield. Hydrolysis of the benzylidene acetal using 80% AcOH at 80 °C followed by TEMPO oxidation²⁰ installed the required carboxylic acid moiety **14** in 78% overall yield. Finally, Zemplén de-O-acylation²¹ using NaOMe in MeOH furnished the target tetrasaccharide **1** in 89% yield (Scheme 3).



Scheme 3. Synthesis of the tetrasaccharide, oxidation and deprotection.

3. Conclusion

In conclusion, we have accomplished the total synthesis of the tetrasaccharide glycone part of the cell-growth inhibitory triterpenoid saponin isolated from *S. noronhae* as its propargyl glycoside. The propargyl glycoside at the reducing end will provide scope in making various types of glycoconjugates to assess the biological activity of the oligosaccharide in question.

4. Experimental

4.1. Preparation of H₂SO₄-silica

To a slurry of silica gel (10 g, 200–400 mesh) in dry diethyl ether (50 mL) was added commercially available concentrated H_2SO_4 (1 mL), and the slurry was shaken for 5 min. The solvent was evaporated under reduced pressure, resulting in free flowing H_2SO_4 -silica, which was dried at 110 °C for 3 h and then used for the reactions.

4.2. *p*-Methoxyphenyl 6-0-(*tert*-butyldiphenylsilyl)-2,3-di-Obenzoyl-β-p-galactopyranoside 4

To the mixture of known *p*-methoxyphenyl- β -D-galactopyranoside **2** (3.0 g, 10.4 mmol) in dry acetonitrile (25 mL) was added benzaldehydedimethyl acetal (2.4 mL, 15.8 mmol) followed by catalytic amount of CSA (pH: 3–4) at room temperature and the reaction mixture was allowed to stir for 30 min. After completion of the reaction as shown by TLC (CH₂Cl₂–MeOH; 15:1), the solution was neutralized with Et₃N and the solvent was evaporated in vacuo to afford *p*-methoxyphenyl 4,6-*O*-benzylidene- β -D-galactopyranoside as a crude product. To a slurry of this product in dry dichloromethane (30 mL) was added pyridine (3.4 mL, 41 mmol) and the mixture was placed in an ice bath. To the cold suspension was added benzoyl chloride (3.65 mL, 31.4 mmol); the exothermic reaction started immediately; the reaction mixture was allowed to stir at 0 °C for 30 min and then the temperature was raised to room temperature. The mixture was further allowed to stir for 3 h when TLC (n-hexane-EtOAc; 4:1) showed complete conversion of the starting material. The solvents were evaporated in vacuo and the residue was dissolved in dichloromethane (40 mL) and washed successively with $H_2O(2 \times 50 \text{ mL})$. The organic layer was collected, dried (Na₂SO₄) and evaporated to a syrup. The crude compound was purified by flash chromatography using *n*-hexane–EtOAc (4:1) to afford pure compound 3^{22} (4.9 g, 81%) as a white solid compound. $[\alpha]_{D}^{25} = +107$ (*c* 1.2, CHCl₃). HRMS calcd for C₃₄H₃₀O₉Na (M+Na)⁺: 605.1788; found 605.1785.

A solution of compound 3 (3.0 g, 5.1 mmol) in AcOH $-H_2O$ (9:1. 30 mL) was stirred at 85 °C for 2 h when TLC showed complete conversion of the starting material to a slower moving spot. The solvents were evaporated in vacuo and the residue was purified by flash chromatography using *n*-hexane–EtOAc (3:1) to afford a pure diol (2.4 g, 93%) as a white solid. To a solution of the diol²³ (2.4 g, 4.8 mmol) in dry pyridine (25 mL) was added TBDPS-Cl (1.8 mL, 7.15 mmol) and the solution was stirred for 12 h at room temperature. Solvents were evaporated in vacuo and the residual syrup was purified by flash chromatography using n-hexane-EtOAc (5:1) as eluent to afford pure compound 4 (3.0 g, 87%) as a white foam. $[\alpha]_{D}^{25} = +107$ (*c* 1.2, CHCl₃). ¹H NMR (CDCl₃, 300 MHz) δ: 8.03 (2d, 4H, J 8.6 Hz, COC₆H₅), 7.76-7.37 (m, 16H, ArH), 7.01, 6.76 (2d, 4H, J 8.7 Hz, C₆H₄OCH₃), 6.05 (dd, 1H, J 9.3, 7.9 Hz, H-2), 5.40 (dd, 1H, J 9.3, 3.1 Hz, H-3), 5.15 (d, 1H, J 7.9 Hz, H-1), 4.51 (br s, 1H, H-4), 4.06-3.88 (m, 3H, H-5, H-6^a, H-6^b), 3.76 (s, 3H, C₆H₄OCH₃), 2.98 (br s, 1H, OH), 1.11 (s, 9H, C(CH₃)₃). ¹³C NMR (CDCl₃, 75 MHz) δ: 166.0, 165.4 (COC₆H₅), 155.5, 151.4, 135.7(2), 135.6(2), 133.4, 133.1, 132.8, 132.5, 132.4, 129.9(3), 129.7(2), 129.5, 129.1(2), 128.4(2), 128.3(2), 127.8(3), 118.9(2), 114.4(2) (ArC), 101.3 (C-1), 74.5, 74.4, 69.6, 68.2, 63.6 (C-6), 55.6 (C₆H₄OCH₃), 26.8(3) (C(CH₃)₃), 19.1 (C(CH₃)₃). HRMS calcd for C₄₃H₄₉O₉SiNa (M+Na)⁺: 755.2653; found 755.2650.

4.3. p-Methoxyphenyl 2,3,4-tri-O-acetyl- α -L-rhamnopyranosyl-(1 \rightarrow 4)-2,3-di-O-benzoyl-6-O-tert-butyldiphenylsilyl- β -D-galac-topyranoside 6

A mixture of the acceptor 4 (1.3 g, 1.8 mmol), donor 5 (925 mg, 2.3 mmol) and MS 4 Å (2.0 g) in dry CH_2Cl_2 (25 mL) was stirred under an N2 atmosphere for 30 min. Next, NIS (675 mg, 3.0 mmol) was added followed by H_2SO_4 -silica (50 mg) and the mixture was allowed to stir at rt for 45 min when TLC (n-hexane-EtOAc, 3:1) showed complete conversion of the acceptor. The mixture was filtered through a pad of Celite[®] and the filtrate was washed successively with aq Na₂S₂O₃ (2×30 mL), saturated aq NaHCO₃ $(2 \times 30 \text{ mL})$ and H₂O (30 mL). The organic layer was collected, dried (Na₂SO₄) and filtered. The solvents were evaporated in vacuo and the residue was purified by flash chromatography using *n*-hexane-EtOAc (4:1) as eluent to afford pure disaccharide 6 (1.5 g, 85%) as a colourless gel. $[\alpha]_D^{25}=+97$ (c 1.0, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) *b*: 7.94 (d, 4H, J 8.0 Hz, ArH), 7.70 (m, 4H, ArH), 7.51– 7.32 (m, 12H, ArH), 6.96, 6.70 (2d, 4H, ArH), 5.94 (dd, 1H, J_{3',4'} 9.5 Hz, $J_{4',5'}$ 10.0 Hz, H-4'), 5.46 (dd, 1H, $J_{2',3'}$ 1.5 Hz, $J_{3',4'}$ 9.5 Hz, H-3'), 5.44 (m, 1H, H-2'), 5.35 (dd, 1H, J_{2,3} 9.5 Hz, J_{3,4} 3.5 Hz, H-3), 5.12 (d, 1H, J_{1,2} 9.5 Hz, H-1), 4.89 (t, 1H, J_{1,2}, J_{2,3} 9.5 Hz, H-2), 4.85 (br s, 1H, H-1'), 4.35 (br s, 1H, H-4), 4.08 (dd, 1H, J_{5,6a} 8.0 Hz, J_{6a,6b} 11.5 Hz, H-6a), 3.89–3.46 (m, 2H, H-5', H-6b), 3.74 (m, 1H, H-5), 3.71 (s, 3H, CH₂-C₆H₅-OCH₃), 2.02, 1.98, 1.94 (3s, 9H, $3 \times \text{COCH}_3$), 0.86 (d, 3H, $J_{5',6'}$ 6.0 Hz, C-CH₃). ¹³C NMR (CDCl₃,

125 MHz) δ : 169.9, 165.7, 164.9 (3 × COCH₃), 165.7, 164.9 (2 × COPh), 155.3, 151.3, 135.6(2), 135.5(2), 133.4, 133.0, 132.9, 132.8, 130.0(2), 129.9, 129.8, 129.7(2), 129.4, 128.5(2), 128.4, 128.2(2), 127.8(3), 127.7, 118.3(2), 114.4(2), 100.8 (C-1), 99.2 (C-1'), 75.6, 74.2, 73.8, 70.9, 70.3, 70.2, 69.6, 68.8, 67.3, 63.2 (C-6), 55.5 (CH₂-C₆H₅-OCH₃), 26.7 [C(CH₃)₃], 20.9, 20.8, 20.7 (3 × COCH₃), 17.4 (C-CH₃). HRMS calcd for C₅₅H₆₀O₁₆SiNa (M+Na)⁺: 1027.3548; found 1027.3542.

4.4. p-Methoxyphenyl 2,3,4-tri-O-acetyl- α -L-rhamnopyranosyl-(1 \rightarrow 4)-2,3-di-O-benzoyl- β -D-galactopyranoside 7

To a solution of compound 6 (1.4 g, 1.4 mmol) in dry THF (10 mL), Bu₄NF in THF (2.8 mL, 2.8 mmol) was added and the solution was allowed to stir at rt for 6 h. Then the solvents were evaporated to dryness. The crude product thus obtained was purified by flash chromatography using *n*-hexane–EtOAc (2:1) to afford pure compound **7** (930 mg, 87%) as a light yellow syrup. $[\alpha]_{D}^{25} = +85$ (*c* 1.0, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) δ : 7.93–7.31 (m, 10H, ArH), 6.93, 6.75 (2d, 4H, ArH), 5.94 (dd, 1H, J_{3',4'} 8.0 Hz, J_{4',5'} 10.5 Hz, H-4'), 5.48 (dd, 1H, $J_{2',3'}$ 3.0 Hz, $J_{3',4'}$ 8.0 Hz, H-3'), 5.47 (m, 1H, H-2'), 5.38 (dd, 1H, J_{2,3} 9.5 Hz, J_{3,4} 3.5 Hz, H-3), 5.19 (d, 1H, J_{1,2} 8.0 Hz, H-1), 4.97 (t, 1H, J_{1,2}, J_{2,3} 8.5 Hz, H-2), 4.43 (d, 1H, J_{1',2'} 3.0 Hz, H-1'), 4.02–3.97 (m, 3H, H-5', H-6a, H-6b), 3.81 (m, 1H, H-5), 2.05, 1.95, 1.94 (3 × COCH₃), 1.20 (d, 3H, J_{5'.6'} 6.5 Hz, C-CH₃). ¹³C NMR (CDCl₃, 125 MHz) δ: 170.0, 169.6, 169.3 (3 × COCH₃), 165.7, 165.0 (2 × COPh), 155.5, 151.0, 133.5, 133.1, 129.9(2), 129.7(2), 129.3, 128.5(2), 128.4, 128.3(2), 118.5(2), 114.5(2) (ArC), 100.9 (C-1), 99.4 (C-1'), 74.8, 74.1, 73.6, 70.9, 69.5, 69.4, 68.9, 68.0, 61.2 (C-6), 55.5 (OC₆H₅OCH₃), 20.7, 20.6, 20.5 (3 \times COCH₃), 17.3 (C–CH₃).

4.5. *p*-Methoxyphenyl 2,3,4-tri-O-acetyl- α -L-rhamnopyranosyl-(1 \rightarrow 4)-6-O-acetyl-2,3-di-O-benzoyl- β -D-galactopyranoside 8

Compound 7 (900 mg, 1.2 mmol) was dissolved in pyridine (8 mL) followed by the addition of Ac₂O (4 mL) and the solution was allowed to stir at rt for 1 h. The solvents were evaporated and the residue was purified by flash chromatography using *n*-hexane-EtOAc (3:1) to give pure disaccharide 8 (880 mg, 93%) as a white foam. $[\alpha]_{D}^{25} = +108$ (*c* 1.1, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) δ: 7.97-7.34 (m, 10H, ArH), 6.96, 6.78 (2d, J 9.0 Hz, OC₆H₄OMe), 5.95 (dd, 1H, J_{1,2} 8.0 Hz, J_{2,3} 10.5 Hz, H-2), 5.49 (dd, 1H, J_{1',2'} 2.5 Hz, J_{2',3'} 3.0 Hz, H-2'), 5.45 (dd, 1H, J_{2,3} 10.5 Hz, J_{3,4} 3.0 Hz, H-3), 5.39 (dd, 1H, J_{2',3'} 3.5 Hz, J_{3',4'} 8.5 Hz, H-3'), 5.14 (d, 1H, J_{1,2} 8.0 Hz, H-1), 4.99 (t, 1H, J_{3',4'}, J_{4',5'} 8.5 Hz, H-4'), 4.88 (d, 1H, J_{1',2'} 2.5 Hz, H-1'), 4.47 (dd, 1H, J_{5,6a} 6.5 Hz, J_{6a,6b} 11.0 Hz, H-6a), 4.38 (br d, 1H, H-4), 4.26 (dd, 1H, J_{5,6b} 6.0 Hz, J_{6a,6b} 11.5 Hz, H-6b), 4.07 (bt, 1H, H-5), 4.02 (m, 1H, H-5'), 3.75 (s, 3H, C₄H₄OCH₃), 2.11, 2.06, 1.95, 1.90 (4s, 12H, 4 × COCH₃), 1.20 (d, 3H, J_{5'.6'} 6.0 Hz, C-CH₃). ¹³C NMR (CDCl₃, 125 MHz) *δ*: 170.4, 170.0, 169.5, 169.3 $(4 \times \text{COCH}_3)$, 165.7, 164.9 $(2 \times \text{COC}_6\text{H}_5)$, 155.6, 151.1, 133.6, 133.1, 130(2), 129.7(2), 129.3, 128.6(2), 128.3(2), 118.8(2), 114.4(2) (ArC), 101.1 (C-1), 99.6 (C-1'), 74.3, 73.5, 71.9, 70.9, 69.6, 69.3, 68.9, 67.7, 62.5, 55.6 (C₆H₄OCH₃), 20.8, 20.7, 20.6(2) $(4 \times \text{COCH}_3)$, 17.3 (C-CH₃). HRMS calcd for C₄₁H₄₄O₁₇Na (M+Na)⁺: 831.2476; found 831.2474.

4.6. p-Methoxyphenyl 2,3,4-tri-O-acetyl- α -L-rhamnopyranosyl-(1 \rightarrow 4)-6-O-acetyl-2,3-di-O-benzoyl- β -D-galactopyranosyl tri-chloroacetimidate 9

To a solution of compound **8** (850 mg, 1.1 mmol) in CH_3CN-H_2O (9:1, 20 mL), CAN (1.2 g, 2.2 mmol) was added and the solution was stirred at room temperature for 30 min after which TLC (*n*-hexane–EtOAc, 2:1) showed complete conversion of the starting

material to a slower moving spot. Solvents were evaporated in vacuo, and the residue was dissolved in CH₂Cl₂ (20 mL) and washed with H₂O (2 × 20 mL). The organic layer was separated, dried (Na₂SO₄) and evaporated to a syrup. The crude hemiacetal thus obtained was dissolved in dry CH₂Cl₂ (20 mL), after which CCl₃CN (550 μ L, mmol) was added followed by DBU (mL, mmol) and the solution was stirred at room temperature for 2 h when TLC (*n*-hexane–EtOAc, 3:1) showed complete conversion of the starting material to a faster moving spot. The solvents were evaporated in vacuo and the residue was purified by flash chromatography using 4:1 (*n*-hexane–EtOAc) as eluent to afford pure trichloroacetimidate derivative **9** (755 mg, 81%) as a colourless gel. The material was used directly for further glycosylation.

4.7. Propargyl 2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 2)$ -4,6-O-benzylidene- β -D-glucopyranoside 12

A mixture of acceptor **10** (1.3 g, 3.0 mmol), donor **11** (1.8 g, 3.9 mmol) and MS 4 Å (g) in dry CH₂Cl₂ (20 mL) was stirred under nitrogen for 30 min at -30 °C. Next, NIS (1.1 g, 5.1 mmol) was added followed by H₂SO₄-silica (50 mg) and the mixture was stirred for another 45 min when TLC (n-hexane-EtOAc, 3:1) showed complete consumption of the acceptor. Then the temperature was slowly raised to rt (1 h) to cleave the *p*-methoxybenzyl group. The mixture was filtered through a pad of Celite[®] and the filtrate was washed successively with aq $Na_2S_2O_3$ (2 \times 25 mL), saturated aq NaHCO₃ (2 \times 25 mL) and H₂O (25 mL). The organic layer was collected, dried (Na₂SO₄) and filtered. The solvents were evaporated in vacuo and the residue was purified by flash chromatography using *n*-hexane-EtOAc (3:1) as an eluent to afford pure disaccharide **12** (1.6 g, 83%) as a white foam. $[\alpha]_{D}^{25} = +98$ (*c* 1.0, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) *δ*: 7.46–7.33 (m, 5H, ArH), 5.52 (s, 1H, CHPh), 5.21 (t, 1H, J_{2',3'}, J_{3',4'} 9.5 Hz, H-3'), 5.11 (t, 1H, J_{3',4'}, *J*_{4',5'} 9.5 Hz, H-4'), 5.03 (dd, 1H, *J*_{1',2'} 8.0 Hz, *J*_{2',3'} 9.5 Hz, H-2'), 4.86 (d, 1H, $J_{1',2'}$ 8.0 Hz, H-1'), 4.68 (d, 1H, $J_{1,2}$ 8.0 Hz, H-1), 4.46, 4.34 (2dd, J 2.5 Hz, 15.0 Hz, 0-CH₂-C=CH), 4.32 (m, 1H, H-6^a), 4.27 (dd, 1H, $J_{5',6a'}$ 3.0 Hz, $J_{6a',6b'}$ 11.5 Hz, H-6a'), 4.21 (dd, 1H, $J_{5',6b'}$ 4.5 Hz, J_{6a'.6b'} 11.5 Hz, H-6b'), 3.91 (m, 1H, H-3), 3.78 (m, 1H, H-5'), 3.75 (t, 1H, J_{3.4}, J_{4.5} 9.5 Hz, H-4), 3.53-3.44 (m, 3H, H-2, H-5, H-6b), 3.14 (br d, 1H, J_{2,OH} 2.5 Hz, OH), 2.51 (t, 1H, J 2.5 Hz, O-CH₂−C≡CH), 2.08, 2.04, 2.02, 1.98 (4s, 12H, 4 × COCH₃). ¹³C NMR $(CDCl_3, 125 \text{ MHz}) \delta$: 170.7, 170.5, 170.2, 169.4 $(4 \times COCH_3)$, 136.8, 129.2, 128.2(2), 126.2(2) (ArC), 101.7 (CHPh), 101.3 (C-1), 100.9 (C-1'), 83.5, 79.7, 78.4, 75.5, 72.6, 72.0(2), 71.9, 68.5, 68.1, 66.0, 62.0, 53.4, 20.8, 20.7, 20.6, 20.5 $(4 \times \text{COCH}_3)$. HRMS calcd for C₃₀H₃₆O₁₅Na (M+Na)⁺: 659.1952; found 659.1957.

4.8. Propargyl 2,3,4-tri-O-acetyl- α -L-rhamnopyranosyl- $(1 \rightarrow 4)$ -6-O-acetyl-2,3-di-O-benzoyl- β -D-galactopyranosyl- $(1 \rightarrow 3)$ -4,6-O-benzylidene-2-O-(2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyl)- β -D-glucopyranoside 13

A mixture of donor **9** (750 mg, 0.9 mmol), acceptor **12** (480 mg, 0.75 mmol) and MS 4 Å (1.0 g) in dry CH₂Cl₂ (15 mL) was stirred under N₂ for 30 min. Next, H₂SO₄–silica (30 mg) was added and the solution was stirred for 45 min at 0 °C when TLC (*n*-hexane–EtOAc) showed complete conversion of the acceptor. The mixture was neutralized with Et₃N and filtered through a pad of Celite[®]. The filtrate was evaporated and the crude product was purified by flash chromatography using *n*-hexane–EtOAc (4:1) to afford pure tetrasaccharide **13** (865 mg, 87%) as a white foam. [α]_D²⁵ = +79 (*c* 0.9, CHCl₃). ¹H NMR (CDCl₃, 500 MHz) δ : 8.05–7.32 (m, 15H, ArH), 5.68 (dd, 1H, $J_{3'',4''}$ 9.5 Hz, $J_{4'',5''}$ 10.0 Hz, H-4''), 5.59 (s, 1H, *CH*Ph), 5.49 (m, 1H, H-2''), 5.32 (m, 3H, H-2'', H-3', H-3''), 5.20 (t, 1H, $J_{3'',4''}$, $J_{4'',5''}$ 9.5 Hz, H-4''), 5.08 (m, 2H, H-2''', H-3'''), 4.94 (t, 1H, $J_{1',2'}$, $J_{2',3'}$ 8.5 Hz, H-2'), 4.87 (br s, 1H, H-1''), 4.86 (d,

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1H, $J_{1'',2''}$ 9.0 Hz, H-1''), 4.85 (d, 1H, $J_{1''',2'''}$ 9.0 Hz, H-1'''), 4.54 (d, 1H, $J_{1,2}$ 7.0 Hz, H-1), 4.33 (m, 2H, OCH₂C=CH), 4.31 (br d, 1H, H-4'), 4.30–3.90 (m, 8H, H-3, H-5, H-6a, H-6b, H-6a', H-6b', H-6a''', H-6b'''), 3.76 (t, 1H, $J_{3,4}$, $J_{4,5}$ 9.0 Hz, H-4), 3.72 (m, 1H, H-5'''), 3.54 (m, 1H, H-5''), 3.42 (m, 1H, H-5'), 2.43 (t, 1H, J 1.5 Hz, OCH₂C=CH), 2.10, 2.07, 2.06(2), 2.00, 1.98, 1.94, 1.93 (8 × COCH₃), 1.15 (d, 3H, J 6.5 Hz, C–CH₃). ¹³C NMR (CDCl₃, 125 MHz) δ : 170.8, 170.2, 170.1, 170.0, 169.6, 169.5, 169.4, 169.3 (8 × COCH₃), 165.8, 164.9 (2 × COPh), 137.1, 133.6, 133.2, 130.1(2), 129.9(2), 129.4, 129.2, 128.5(2), 128.4(2), 128.3(2), 128.2, 126.0(2) (ArC), 101.3 (CHPh), 100.11 (C-1'''), 99.8 (C-1'), 99.2 (C-1), 98.9 (C-1''), 80.8, 79.5, 78.6,77.5, 75.1, 73.9, 73.5, 73.0, 71.9, 71.8, 71.4, 71.0, 70.3, 69.5, 69.1, 68.8, 68.4, 67.7, 65.5, 62.6, 61.8, 60.4, 56.3, 29.7, 21.0, 20.8, 20.7(2), 20.6(2), 20.5, 20.3 (8 × COCH₃), 17.4 (C–CH₃). HRMS calcd for C₆₄H₇₂O₃₀Na (M+Na)⁺: 1343.4006; found 1343.4002.

4.9. Propargyl α -L-rhamnopyranosyl- $(1 \rightarrow 4)$ - β -D-galactopyranosyl- $(1 \rightarrow 3)$ -2-O- $(\beta$ -D-glucopyranosyl)- β -D-glucopyranosiduronic acid 1

A solution of compound 13 (850 mg, 0.65 mmol) in 80% aq AcOH (15 mL) was stirred at 80 °C for 2 h until TLC (n-hexane-EtOAc, 3:1) showed complete conversion of the starting material to a slower moving spot. The solvents were evaporated in vacuo and the syrupy residue was dissolved in 15 mL CH₂Cl₂. To this solution, 3 mL of H₂O was added followed by aq NaBr (1 M 0.3 mL), aq tetrabutylammonium bromide (1 M, 0.6 mL), TEMPO (27 mg) and saturated aq NaHCO₃ (1.6 mL) at 0 °C. To the resulting mixture, aq NaOCl (2.0 mL) was added and the mixture was allowed to stir for 1.5 h when the temperature was raised to room temperature. At this point TLC showed complete conversion of the starting material to a faster moving spot, presumably the corresponding aldehyde derivative. The mixture was neutralized with 1 M HCl (as required) to keep the pH of the mixture at 6-7. Then tert-butanol (9 mL), NaOCl₂ (600 mg in 2.3 mL H₂O) and NaH₂PO₄ (750 mg in 6 mL H₂O) were added and the mixture was allowed to stir at room temperature for another 4 h when TLC showed complete conversion. The mixture was diluted with saturated NaH₂PO₄ and the product was extracted with EtOAc. The organic layer was dried (Na₂SO₄) and evaporated. The crude product thus obtained was purified by flash chromatography using n-hexane-EtOAc (1:1) to n-hexane-EtOAc (1:4) to afford compound 14 (625 mg, 78%). To a solution of compound 14 (300 mg, 0.24 mmol) in dry MeOH (10 mL), NaOMe in MeOH (0.5 M, 1 mL) was added and the solution was stirred at room temperature for 6 h. Excess NaOMe was neutralized with DOWEX 50 W H⁺ resin, filtered through a cotton plug and the filtrate was evaporated to give pure target tetrasaccharide 1 (150 mg, 89%) as an amorphous white solid. $[\alpha]_{D}^{25} = +63$ (*c* 0.7, H₂O). ¹H NMR (D₂O, 500 MHz) δ : 5.09 (d, 1H, $J_{1'',2''}$ 1.5 Hz, H-1"), 4.74 (d, 1H, J_{1',2'} 7.5 Hz, H-1'), 4.70 (m, 2H, H-1, H-1"'), 4.39 (m, 2H, OCH₂C≡CH), 4.02 (dd, 1H, *J*_{1",2"} 1.5 Hz, *J*_{2",3"} 3.0 Hz, H-2"), 3.95 (br d, 1H, H-4'), 3.91-3.36 (m, 16H, H-2, H-2', H-2", H-3, H-3', H-3", H-3"', H-4, H-4", H-5, H-5", H-5"', H-6a', H-6b', H-6a'', H-6b"'), 3.34 (t, 1H, $J_{3'',4''}$, $J_{4''',5''}$ 8.5 Hz, H-4"'), 3.26 (m, 1H, H-5'), 2.85 (t, 1H, J 2.0 Hz, OCH₂C=CH), 1.19 (d, 3H, J 6.5 Hz, C-CH₃). ¹³C NMR (D₂O, 125 MHz) δ: 173.0 (COOH), 102.9 (C-1'), 102.4 (C-1^{///}), 101.8 (C-1), 99.7 (C-1^{//}), 83.0, 79.6, 78.5, 76.7, 76.0, 75.7, 75.6, 75.4, 73.6, 73.3, 71.7, 71.0, 70.2, 70.0, 69.5, 69.3, 61.5, 60.6, 56.9, 48.9, 23.1, 16.5 (C-CH₃). HRMS calcd for C₂₇H₄₂O₂₁Na (M+Na)⁺: 725.2116; found 725.2113.

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References

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- 11. Williams, J. M.; Richardson, A. C. Tetrahedron **1967**, 23, 1369.
- 12. Gent, P. A.; Gigg, R. J. Chem. Soc., Perkin Trans. 1 1974, 1446-1455.
- 13. Limberg, G.; Thiem, J. Carbohydr. Res. 1995, 275, 107-115.
- 14. Mukhopadhyay, B.; Kartha, K. P. R.; Russell, D. A.; Field, R. A. J. Org. Chem. 2004, 69, 7758–7760.
- 1. Price, K. R.; Johnson, I. T.; Fenwick, G. R. C. R. C. *Crit. Rev. Food Sci. Nutr.* **1987**, *26*, 27–133.
- Papadopoulou, K.; Melton, R. E.; Legget, M.; Daniels, M. J.; Osbourn, A. E. Proc. Natl. Acad. Sci. U.S.A. 1999, 96, 12923–12928.
- Haralampidis, K.; Trojanowska, M.; Osbourn, A. E. Adv. Biochem. Eng./Biotechnol. 2002, 75, 31–49.
- Hostettmann, K. A.; Marston, A. Saponins. Chemistry and Pharmacology of Natural Products; Cambridge University Press: Cambridge, UK, 1995.
- (a) Paczkowski, C.; Wojciechowski, Z. A. Phytochemistry 1994, 35, 1429–1434;
 (b) Wojciechowski, Z. A. Phytochemistry 1975, 14, 1749–1753.
- Yoshida, T.; Chou, T.; Nitta, A.; Okuda, T. Chem. Pharm. Bull. 1991, 39, 2247–2251.
- (a) Perry, L. M. Medicinal Plants of East and Southeast Asia; MIT Press: MA, 1980;
 (b) Burkill, I. H. A Dictionary of the Economic Products of the Malay Peninsula;
- Ministry of Agriculture and Co-operatives: Kuala Lumpur, 1966. 8. Ohtsuki, T.; Miyagawa, T.; Koyano, T.; Kowithayakorn, T.; Kawahara, N.; Goda,
- Y.; Ishibashi, M. J. Nat. Prod. **2008**, 71, 918–921. 9. Ohlsson, J.; Magnusson, G. Carbohydr. Res. **2000**, 329, 49–55.
- 10. Minoru, I.; Koichi, F.; Shoichi, K. *Synlett* **2002**, 1409–1416.

- (a) Roy, B.; Pramanik, K.; Mukhopadhyay, B. *Glycoconjugate J.* 2008, 25, 157–166; (b) Dasgupta, S.; Pramanik, K.; Mukhopadhyay, B. *Tetrahedron* 2007, 63, 12310–12316; (c) Mandal, S.; Mukhopadhyay, B. *Tetrahedron* 2007, 63, 11363–11370; (d) Dasgupta, S.; Mukhopadhyay, B. *Eur. J. Org. Chem.* 2008, 34, 5770–5777; (e) Verma, P. R.; Mukhopadhyay, B. *Carbohydr. Res.* 2010, 345, 432–436; (f) Roy, B.; Field, R. A.; Mukhopadhyay, B. *Carbohydr. Res.* 2009, 344, 2311–2316; (g) Verma, P.; Mukhopadhyay, B. *Carbohydr. Res.* 2009, 344, 2554–2558.
- 16. Neumann, K. W.; Tamura, J.-I.; Ogaya, T. Glycoconjugate J. 1996, 13, 933–936.
- 17. Classon, B.; Garegg, P. J.; Samuelson, B. Acta. Chem. Scand. Ser. B. 1984, 38, 419-422.
- 18. Kerékgyártó, J.; Szurmai, Z.; Lipták, A. Carbohydr. Res. 1993, 245, 65-80.
- 19. Rajput, V. K.; Mukhopadhyay, B. J. Org. Chem. 2008, 73, 6924-6927.
- 20. Huang, L.; Teumelsan, N.; Huang, X. Chem. Eur. J. 2006, 12, 5246-5252.
- 21. Zemplén, G.; Gerecs, A.; Hadácsy, I. Ber. Dtsch. Chem. Ges. 1936, 69, 1827-1830.
- 22. Zhang, Z.; Magnusson, G. J. Org. Chem. 1996, 61, 2383-2393.
- Abdu-Allah, H. H. M.; Tamanaka, T.; Yu, J.; Zhuoyuan, L.; Sadagopan, M.; Adachi, T.; Tsubata, T.; Kelm, S.; Ishida, H.; Kiso, M. J. Med. Chem. 2008, 51, 6665–6681.