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## Rh/Cu-catalyzed Ketone β-Functionalization via Merging Ketone Dehydrogenation and Carboxyl-directed C-H Alkylation

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**ABSTRACT**: An efficient Rh/Cu-catalyzed method has been developed for the direct  $\beta$ -arylation or alkenylation of ketones using (hetero)aryl or alkenyl carboxylic acids as coupling partners. This direct ketone β-functionalization reaction proceeded via merging Cu-catalyzed ketone dehydrogenative desaturation and Rh-catalyzed carboxyl-directed C-H alkylation, exhibited a broad substrate scope for both coupling partners. TEMPO proved to be essential for both dehydrogenation process and generation of the active Rh-catalyst for C-H activation.

KEYWORDS: Dehydrogenation • Cascade reaction • Ketone • Carboxylic acid • Alkylation

β-arylated saturated ketones are the structural motifs that are embedded in many natural products and bioactive compounds.1 Consequently, efficient methods for constructing such structures are always the goal sought after by chemists. Traditionally,  $\beta$ -arylated ketone architectures are constructed through the aldol condensation/hydrogenation two-step procedure,<sup>2</sup> or rhodiumcatalyzed Michael addition of aryl organometallic reagents to enones.<sup>3</sup> These conventional methods either require harsh reaction conditions such as strong bases that limit substrate scope, or employ the reactants that are often prepared by way of multi-step synthetic sequences, which prompts chemists to invent the efficient methods that produce *B*-arylated ketones in atom- and step-economical manner staring from simple reactants. Recently, a breakthrough has been achieved in the development of the bidentate directing group-assisted  $\beta$ -C(sp3)-H activation reaction of carbonyl compounds.<sup>4</sup> In this context, Yu and co-workers  $^{4a}$  have realized  $\beta$ -C(sp3)-H arylation reaction of ketones with aryl iodides as arylating reagents (Scheme 1a) by using amino acid as a transient directing group. MacMillan and co-workers have demonstrated that merging the organocatalysis and photoredox catalysis enabled β-C(sp3)-H arvlation reaction of aliphatic ketones using dicyanobenzene as an arylating reagent (Scheme 1b).<sup>5</sup>

On the other hand, due to the importance of olefins in organic synthesis, transition metal-catalyzed dehydrogenative desaturation of various compounds to generate olefins has recently attracted considerable interest.6-8 Moreover, efforts to achieve the tandem metal-catalyzed dehydrogenation/secondary olefin reaction sequence have led to the inventions of diverse interesting chemical transformations,<sup>9-13</sup> including Baudoin's Pd-catalyzed β-C-H ACS Paragon Plus Environment

arvlation of  $\alpha$ -substituted esters<sup>10a, b, c</sup> and N-Bocpiperidines with aryl halides,<sup>10d</sup> and Pihko's β-C-H indolation of  $\beta$ -keto esters.<sup>11</sup> As for the tandem sequence to access  $\beta$ -arylated ketone, Dong and Li have established the Pd-catalyzed β-arylation reaction of ketones with aryl iodides,<sup>12a</sup> diaryliodonium salts<sup>12b</sup> and arylboronic acids<sup>12c</sup> via ketone dehydrogenative desaturation respectively (Scheme 1b). Newhouse and co-workers have developed Pd-catalyzed the cascade ketone dehydrogenation/organocuprate conjugate addition process to realize  $\beta$ -functionalization or  $\alpha$ , $\beta$ -difunctionalization of cyclic ketones, including ketone β-arylation.<sup>13</sup> Very recently, our group has discovered a Cu-catalyzed radical-based ketone dehydrogenative desaturation process that could merge conjugate addition of nucleophiles to the newly formed enones to furnish β-functionalized saturated ketones.14 Herein, we report a novel method for facile synthesis of  $\beta$ arylated or  $\beta$ -alkenylated ketones via merging Cu(II)catalyzed ketone dehydrogenation and Rh-catalyzed carboxyl-directed ortho-C-H alkylation of (hetero)aryl- or alkenyl carboxylic acids with enone (Scheme 1c).<sup>15</sup>

#### Scheme 1. The Metal-catalyzed Direct β-Csp<sup>3</sup>-H Arylation of Ketones



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Actually, the implementation of the tandem dehydrogenative desaturation/C-H alkylation of arene with enone to construct  $\beta$ -aryl saturated ketone represents a challenge for the following reasons: 1) the compatibility between the dehydrogenative desaturation and the C-H fuctionalization is required; 2) the  $\beta$ -aryl ketone products compete with ketone starting materials for dehydrogenation to form the overoxidized side-product;16 3) the expected alkyl-M intermediate in the C-H alkylation catalysis tends to undergo β-H elimination to form overoxidized side-products. One of our strategies to address these problems is the identification of a suitable catalyst system responsible for C-H alkylation with olefin, which is compatible with the Cucatalyzed dehydrogenative desaturation process.14 We hypothesized that if the C(sp3)-M bond of the alkyl-M intermediate generated is enough polar, this intermediate would prefer protonolysis over β-elimination. Since increasing the concentration of ketone could accelerate Cucatalyzed ketone dehydrogenation to enone, <sup>14a</sup>the use of excessive ketone starting material would suppress dehydrogenation of  $\beta$ -aryl ketone products.

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# Table 1. The Selected Result of the Optimization Studies on the Ketone $\beta$ -Arylation <sup>a</sup>

Ċ	+ COOH Cu(OA + TEM Tolu	tal-catalyst c) <sub>2</sub> (20 mol%) <u>/PO, Base</u> ene (0.1M)	COOH OMe +	ji	5 OMe
1a	2a	0, 2411, 142	3a	3aa	
Entry	Cat.(mol%)	Base (mol%)	TEMPO (equiv.)	Yield (%)	
				3a <sup>b</sup>	3aa
1	$Pd(OAc)_2(5)$		1.0	n.d	24
2	$[Ru(p-cym)Cl_2]_2(:$	5)	1.0	n.d	23
3	$[Rh(Cp^*)Cl_2]_2(5)$	)	1.0	n.d	35
4	$Rh(PPh_3)_3Cl(5)$		1.0	n.d	12
5	$[Rh(COD)Cl]_2(5)$	)	1.0	65	<5
6	$[Rh(COD)Cl]_2(3)$	)	1.0	52	<5
7	$[Rh(COD)Cl]_2(3)$	)	1.2	70	<5
8	$[Rh(COD)Cl]_2(3)$	CsF (10)	1.2	82	<5

<sup>a</sup>Reaction conditions: **1a** (0.6 mmol), **2a** (0.2 mmol), <sup>b</sup> yield as determined by NMR.

With these considerations in mind, we chose the reaction of propiophenone (1a) with *ortho*-methoxy benzoic acid (2a) as a model system for the optimization of reaction conditions (Table 1). Benzoic acids were used as arylating reagents because benzoic acids are readily available, versatile arylating reagents via decarboxylative crosscoupling reactions 17 or carboxyl-directed ortho-C-H functionalization reactions. <sup>18</sup> Our targeted reaction aimed at the carboxyl-directed C-H alkylation to produce 3a of which carboxyl group is able to undergo protodecarboxylation or various decarboxylative cross-coupling reaction, and therefore allow for the further elaboration of products. The undesired lactone by-product **3aa** came from overoxidation of 3a via further dehydrogenation of 3a and subsequent conjugate addition of carboxyl to enone. Initially, we screened a variety of metal catalysts in the presence of one 2,2,6,6-tetramethylpiperidine-1-oxyl equivalent of (TEMPO) and 20 mol% Cu(OAc)<sub>2</sub>, and found that the metal catalysts, which are regularly used for C-H activation reactions, allowed for the reaction of **1a** with **2a** to occur but gave overoxidized product **3aa** (entries 1-4). Gratifyingly, we obtained the targeted product in 65% yield using [Rh(COD)Cl]<sub>2</sub> (5 mol%) <sup>15b, 19</sup> as a catalyst (entry 5). Reducing loading of  $[Rh(COD)Cl]_2$  to 3 mol% led to a decrease in the yield of **3a** (entry 6). However, increasing the amount of TEMPO to 1.2 equivalents gave 70% yield with 3 mol%  $[Rh(COD)Cl]_2$  catalyst (entry 7). Then, we checked the effect of bases on the reaction outcome. Although acetate and carbonate salts have a negative effect on the reaction (see Table S1 in the Supporting Information), 10 mol% CsF could increase the yield to 82% (entry 8).

#### Scheme 2. The Scope of Carboxylic Acid <sup>a</sup>



<sup>a</sup>Reaction conditions: **1a** (0.6 mmol), **2** (0.2 mmol), All isolated yields. <sup>b</sup>**1a** (0.8 mmol), TEMPO (2.4 equiv.).

With the optimized reaction conditions in hand, we evaluated the substrate scope of carboxylic acids (Scheme 2). Ortho-substituted benzoic acids gave good-to-excellent vields and variation of the ortho-substituents of these benzoic acids did not significantly affect the reaction outcomes (**3a-3g**, **3t**). Although nonsubstituted benzoic acids gave di-substituted product (**3q**), the reaction of 3-*t*-butyl benzoic acid occurred only on the less sterically hindered position (3h). A series of multi-substituted benzoic acids could be smoothly transformed into the corresponding products in good yields (3i-3m, 3r, 3u). Heteroaryl carboxylic acids were also suitable substrates as illustrated by **3n** and **3o**. Interestingly, this protocol was amenable to carboxyldirected alkenylic C-H alkylation of alkenyl carboxylic acids (**3v-3x**), further highlighting the generality of this protocol.

Next, we explored the substrate scope of saturated ketones (Scheme 3). A variety of electronically diverse functionalities on 4-positions of phenyl rings in propiophenones were well tolerated to give the  $\beta$ -arylation ketones in excellent yields (**4a-4f** and **4j**). Changing the substitution position to meta-position of phenyl ring in propiophenones had no effect on the reaction yields (**4g-4i**). Propiophenones containing multi-substituted phenyl rings were also suitable for this transformation (**4k-4m**). Additionally, heteroaromatic ketones such as 2-propionylfuran and 2-propionylthiophene participated in the targeted  $\beta$ -arylation reaction without side reactions occurring at their reactive C-2 or C-3 positions of heteroaromatic rings (**4o** and **4p**). As exemplified in the cases of **4q-4t**, aliphatic

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ketones also underwent  $\beta$ -arylation reaction which occurred exclusively on less sterically hindered ethyl moiety. This selective  $\beta$ -arylaton of aliphatic ketones circumvented the formidable synthesis of the alkyl vinyl ketones used in Ru-catalyzed hydroarylation of olefins, and provided a solution to the selectivity issue encountered in the synthesis of  $\beta$ -aryl ketone via aldol reaction of unsymmetrical ketone.

Scheme 3. The Scope of Saturated Ketones<sup>a</sup>



<sup>a</sup>Reaction conditions: **1** (0.6 mmol), **2** (0.2 mmol). All isolated yields. <sup>b</sup> Cu(OAc)<sub>2</sub> (40 mol%) was used.

Our Rh/Cu catalyzed method also enabled one-step gram-scale synthesis of a kind of matrix metalloproteinase inhibitor in 71% yield using commercially available starting materials. In sharp contrast, this matrix metalloproteinase inhibitor was previously synthesized by way of a six-step synthetic route starting from 3-(1bromnaphthalen-2yl) propan-1-ol (Scheme 4).<sup>20</sup>

Scheme 4. One Step Synthesis of a kind of matrix metalloproteinase inhibitor



Scheme 5. One-pot procedure for Rh/Cu-catalyzed ketone β-arylation and protodecarboxylation<sup>a</sup>



 $^aReaction$  conditions: 1 (0.6 mmol), 2 (0.2 mmol), 120 °C, 24 h, N2. All isolated yields.

To test the feasibility of further transformation of these ketone  $\beta$ -arylation products, we investigated one-pot twostep procedure for Rh/Cu catalyzed carboxyl-directed C-H alkylation and subsequent protodecarboxylation. We are pleased to find that adding AgOAc (1 equiv.),  $K_2CO_3$  (1 equiv.) and HOAc to the reaction system after Rh/Cu catalysis process allowed for protodecarboxylation of these  $\beta$ arylation products to generate 1,3,5-trisubstituted benzenes as final products in synthetically useful yields (Scheme 5).

To gain an insight into the Rh/Cu catalyzed  $\beta$ -arylation of ketone, we conducted preliminary mechanism investigations (See Supporting Information for details). By means of a H/D exchange experiment, we initially explored the effect of TEMPO on [Rh(COD)Cl]2-mediated carboxyldirected ortho C-H activation. In the presence of D<sub>2</sub>O (20 equiv.) and TEMPO (1.2 equiv.), 30% of C-H bond ortho to carboxyl group in 2-methoxy-benzoic acid was deuterated, indicating that the reversible C-H cyclorhodation occurred. Control experiments revealed that only negligible H/D exchanges occurred in the absence of TEMPO, suggesting that TEMPO played a key role in the [Rh(COD)Cl]2mediated C-H activation step. Moreover, TEMPO was essential for [Rh(COD)Cl]<sub>2</sub>-catalyzed carboxyl-directed C-H alkylation of benzoic acids with enone, consistent with H/D exchange experiments. In light of these observations, we reasoned that TEMPO would promote the generation of the active Rh catalyst for C-H bond activation, likely by oxidation of Rh(I) to Rh(III) species as proposed by Studer and co-workers <sup>21</sup> for the Rh-catalyzed oxidative crosscoupling reactions with TEMPO as the only terminal oxidant. Moreover, we conducted the reaction of 1(3methoxyphenyl)propan-1-one with [D5]benzoic acid under standard reaction conditions and found that 45% of deuterium on ortho-position of [D5]benzoic acid was incorporated to  $\alpha$ -position of  $\beta$ -arylated ketone product, and 35% to  $\beta$ -position.

#### Scheme 6. Proposed Mechanism.



The identification of enone intermediate in the reaction system and the experiments described by Scheme *S1-S3* supported that the Rh/Cu-catalyzed ketone  $\beta$ -arylation occurred via the in-situ ketone dehydrogenation and C-H alkylation with the enone. Scheme 6 shows a proposed mechanism for the Rh-catalyzed carboxyl-directed *ortho* C-H alkylation with the enone. Initially, the catalytically active Rh(III)(TEMPO)<sub>2</sub>Ln I is generated via oxidation of [Rh(COD)Cl]<sub>2</sub> by TEMPO (2 equiv.). After carboxylate coordinates to Rh(III) catalyst I, *ortho* C-H metalation proceeds via a concerted metalation/deprotonation mechanism (CMD) to form aryl-Rh(III) species II, <sup>22</sup> in which TEMPO<sup>-</sup>

anion ligand accepts proton by its nitrogen atom to promote C-H metalation. Olefin insertion into aryl-Rh bond leads to formation of alkyl-Rh intermediate III that undergoes protonolysis via proton transfer from the nitrogen atom of coordinating TEMPO ligand to the carbon atom of C-Rh bond to release product VI. Meanwhile, Product VI is also produced through protonolysis of isomeric alkyl-Rh intermediate V, which is generated from intermediate III via β-H elimination and subsequent re-insertion process,<sup>15b, 23</sup> since deuterium atoms were incorporated to both  $\alpha$  and  $\beta$  positions of  $\beta$ -arylated ketone product in the deuterium-labeling experiment. The fact that 80% of ortho-deuterium atoms of [D5]benzoic acid were incorporated into the final product implicates that after accepting proton in the C-H metalation process, TEMPO still coordinates to Rh atom in the form of zwitterion, and therefore makes it easy to deliver proton to the proximal Rh-C bond of alkyl-Rh intermediate.

In summary, we have developed the Rh/Cu catalyzed direct  $\beta$ -arylation or alkenylation of ketones with aryl or alkenyl carboxylic acids as coupling partners via combination of Cu/TEMPO-promoted ketone dehydrogenation process with Rh-catalyzed carboxyl-directed C-H alkylation with enone. Both aryl alkyl ketones and dialkyl ketones underwent  $\beta$ -functionalization reactions with a broad range of aryl carboxylic acids as well as alkenyl carboxylic acids with good selectivity in good-to-excellent yields. The strategy to merge the dehydrogenative desaturation and the C-H activation reaction would bring about discoveries of various new transformations due to versatile reactivity of olefins and diversity of C-H activation reactions.

#### ASSOCIATED CONTENT

**Supporting Information**. Experiment procedures, characterization data, <sup>1</sup>H, <sup>13</sup>C and <sup>19</sup>F spectra for compounds, mechanism study. This material is available free of charge via the Internet at http://pubs.acs.org.

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#### Notes

The authors declare no competing financial interests.

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SYNOPSIS TOC

R ÇOOH Cu(II) Rh(I) Ö COOH R TEMPO R = aryl, alkyl

- 52 examples, up to 91% yield Compatibility and cooperation between dehydrogenation and C-H alkylation Preferential dehydrogenation of staring materials to avoid overoxidation of products