

Total Synthesis of (±)-Thebainone A by Intramolecular Nitronc Cycloaddition

Yuzhou Wang, André Hennig, Thomas Küttler, Christian Hahn, Anne Jäger, and Peter Metz*



Cite This: <https://dx.doi.org/10.1021/acs.orglett.0c00905>



Read Online

ACCESS |



Metrics & More

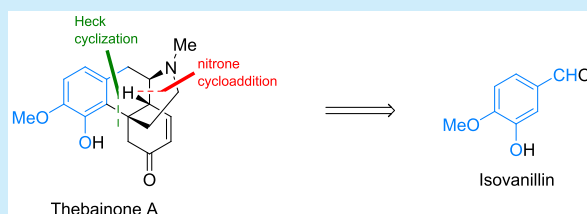


Article Recommendations



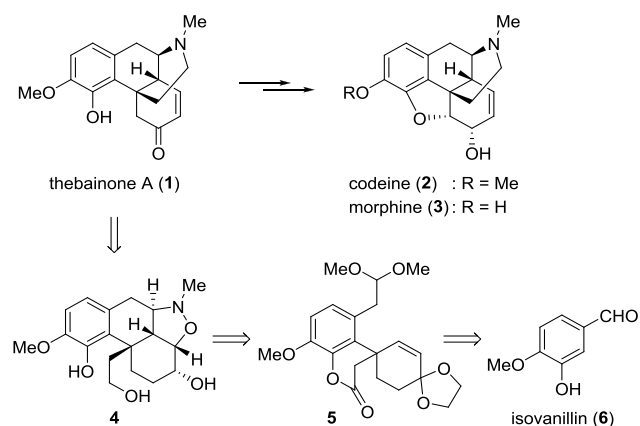
Supporting Information

ABSTRACT: Using an intramolecular nitronc cycloaddition and a Heck cyclization as the crucial transformations, a total synthesis of the racemic morphine alkaloid thebainone A was accomplished in 22 steps commencing with isovanillin.



The alkaloids morphine (3) and codeine (2) present in the opium poppy (*Papaver somniferum*) are among the most important drugs when the treatment of severe pain is necessary (Scheme 1). Gates succeeded in the first total synthesis of morphine (3) in 1952.¹ Since then, more than 30 total and formal syntheses have been published.^{2–4} Due to our interest in the application of intramolecular nitronc cycloadditions toward the synthesis of morphine alkaloids,^{5,6} we recently devised a novel access to thebainone A (1), which in turn is a known precursor for 2 and 3. This was already shown by Gates in his pioneering work.¹ A second total synthesis of 1 was described by Tius in 1992.⁷

Scheme 1. Retrosynthetic Analysis

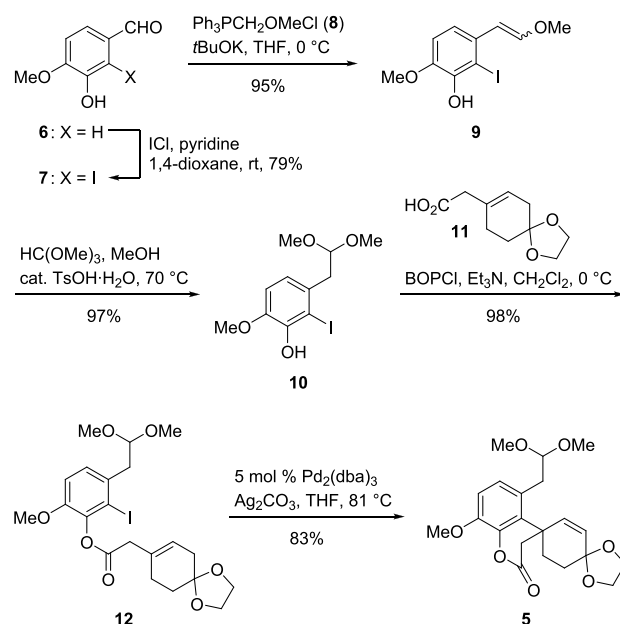


As depicted in Scheme 1, we envisioned to trace thebainone A (1) back to isoxazolidine 4. The heterocyclic moiety in 4 might be generated by a nitronc cycloaddition of a substrate derived from bisacetal 5. Ultimately, readily available isovanillin (6) serves as the starting material for the bisacetal 5, with the demanding construction of the

quaternary benzylic center being planned by a Heck reaction.⁸

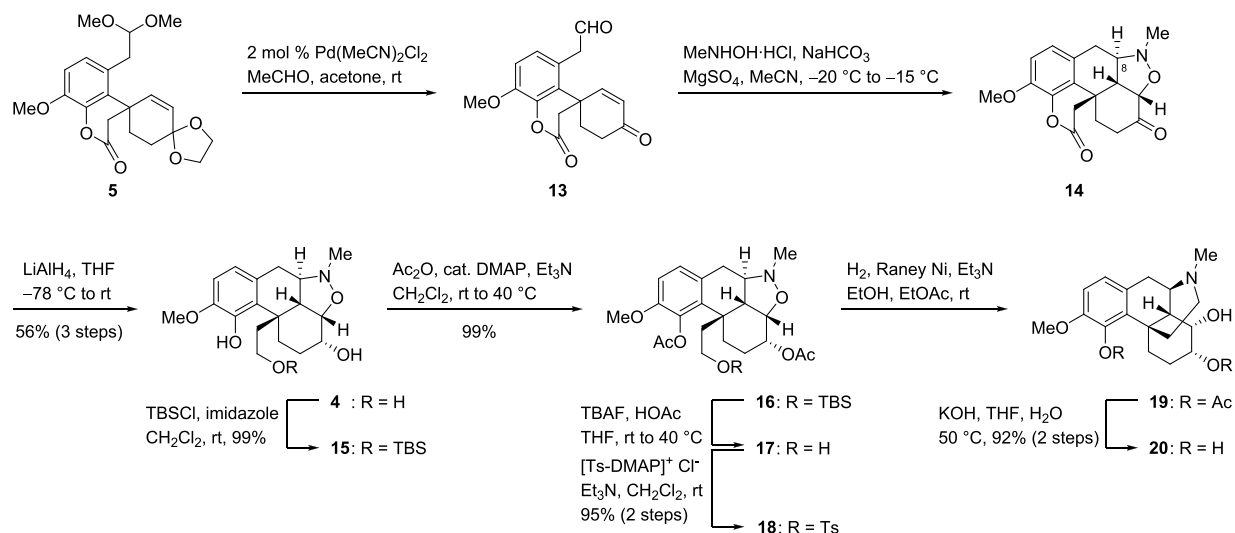
Already after the first four steps of the synthesis, all carbon atoms required for the basic framework of thebainone A (1) were assembled (Scheme 2). Iodination of isovanillin (6) to

Scheme 2. Conversion of Isovanillin (6) into Lactone 5



Received: March 11, 2020

Scheme 3. Synthesis of Morphinan 20 from Lactone 5



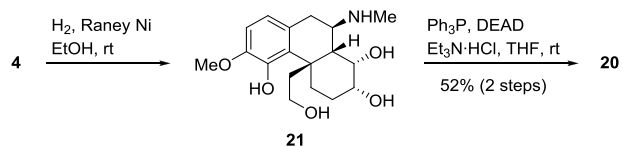
give derivative **7**⁹ was followed by a Wittig reaction with the ylide derived from phosphonium salt **8** to provide a diastereomeric mixture of enol ethers **9** and subsequent conversion of **9** into the known acetal **10**¹⁰ in very good yield.

By avoiding the use of protecting groups, the published routes from **6** to **10** could be streamlined to three steps. Esterification of phenol **10** with acid **11**^{8,11} through activation with BOPCI¹² led to ester **12** with high efficiency. After intense screening of reaction conditions, the Heck cyclization of **12** to give lactone **5** was best effected without any phosphine ligands^{8a} in the presence of silver carbonate.

For the key nitron cycloaddition, the two protected carbonyl functions of **5** first had to be deblocked. This was cleanly achieved with the mild Lewis acid Pd(MeCN)₂Cl₂ in acetone (Scheme 3).¹³ It was not necessary to purify the labile keto aldehyde **13** on silica gel. Thus, nitron formation and subsequent intramolecular cycloaddition were carried out with crude **13** and *N*-methylhydroxylamine. Running this reaction at low temperature effected a high diastereoselectivity in the formation of the desired isoxazolidine **14** (**14**:*8-epi-14* ca. 10:1). Since this keto lactone was also rather unstable, it was immediately reduced with a large excess of lithium aluminum hydride to give dihydroxy phenol **4** in good overall yield from **13**. The relative configuration of **4** was confirmed by X-ray diffraction analysis. An efficient transformation of the primary alcohol of **4** into a tosylate without competing phenol ether formation succeeded by chemoselective silylation to give silyl ether **15** followed by double acetylation and desilylation of the resultant diacetate **16** to provide alcohol **17** and finally tosylation.¹⁴ Tosylate **18** was then subjected to reductive cleavage of the heterocycle with concomitant intramolecular nucleophilic substitution to generate piperidine **19**.^{5b} Subsequent saponification of **19** provided dihydroxy phenol **20** that was also characterized by X-ray diffraction analysis.

We also investigated a potentially shorter route from isoxazolidine **4** to morphinan **20** using a Mitsunobu reaction of trihydroxy phenol **21** that was readily obtained from **4** by hydrogenolytic cleavage of the N–O bond (Scheme 4). In order to avoid a competing intramolecular alkylation of the phenol oxygen, triethylamine hydrochloride was added.¹⁵

Scheme 4. Transformation of Isoxazolidine 4 to Morphinan 20 by Mitsunobu Reaction

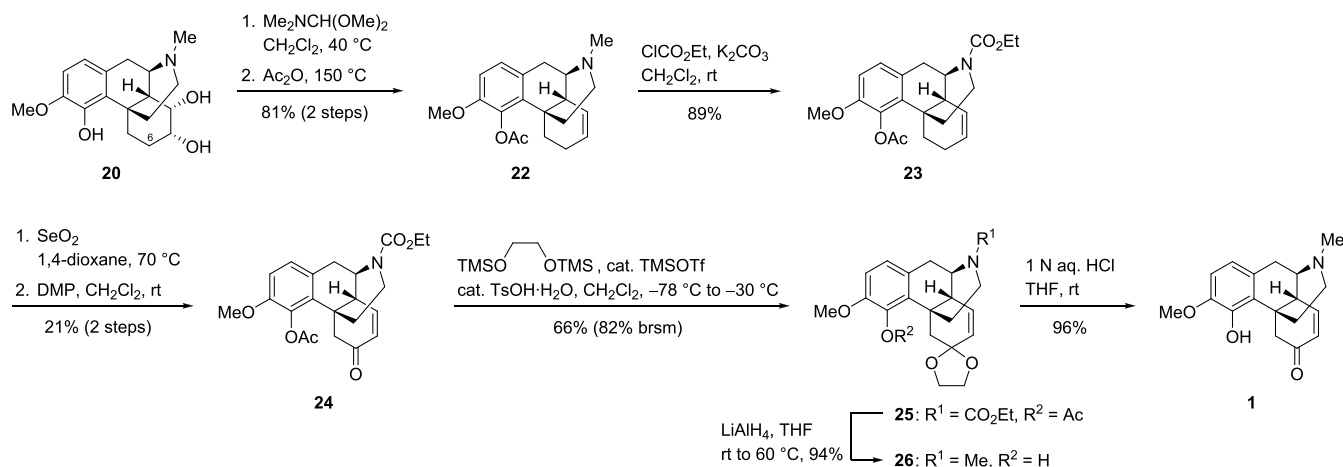


While we sometimes observed the formation of morphinan **20** in good yield, this reaction turned out to be rather capricious, and we could not find suitable conditions for the conversion of **21** to **20** in a reproducible fashion. Nevertheless, even though some more steps are required for the sequence leading from **4** to **20** in Scheme 3, its overall yield (75%) is far better than the best result obtained according to Scheme 4.

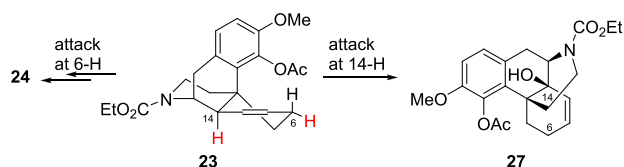
Completion of the synthesis of thebainone A (**1**) is shown in Scheme 5. In order to activate the C-6 position for introduction of the ketone function, the *cis* diol moiety of **20** was reductively eliminated by formamide acetal pyrolysis¹⁶ with simultaneous esterification of the phenol to give olefin **22**. Prior to allylic oxidation, the tertiary amine was converted to ethyl carbamate **23**.¹⁷ Subsequent reaction of **23** with selenium dioxide¹⁸ and further oxidation of the resultant allylic alcohol to the ketone with the Dess–Martin periodinane (DMP)^{3b,8a,18c,d} afforded enone **24** in moderate overall yield. We first opted for a reduction/oxidation sequence in order to convert **24** to **1**. However, treatment of **24** with lithium aluminum hydride to give α -thebainol¹⁹ led to partial overreduction of the enone, and reoxidation of the highly sterically hindered allylic alcohol in α -thebainol featuring a free phenol could not be achieved by Swern oxidation. Thus, ketone **24** was protected as the dioxolane **25** without migration of the double bond²⁰ under mild reaction conditions.²¹ Finally, lithium aluminum hydride reduction of the carbamate¹⁷ and acetate function in **25** gave rise to acetal **26** that was deprotected uneventfully to provide ($\pm$)-thebainone A (**1**).

The relatively low yield of the allylic oxidation of **23** at C-6 is probably due to an unfavorable conformation of the C ring cyclohexene (Scheme 6). All of the tetracyclic morphine precursors that have been successfully subjected to selenium

Scheme 5. Completion of the Synthesis of Thebainone A (1)



Scheme 6. Allylic Oxidation of Cyclohexene 23



dioxide oxidation adopt a boat conformation of the C ring with axial orientation of $6\beta\text{-H}$.¹⁸ In contrast, 23 lacking the tetrahydrofuran E ring features a half-chair conformation of the C ring. As a consequence, the sterically more accessible $6\beta\text{-H}$ is oriented equatorially, while 14-H is properly aligned with the adjacent π -system for a hetero ene reaction with selenium dioxide. Indeed, the C-14 hydroxylated morphinan 27 was isolated as another major product of this reaction.²²

With this shortest total synthesis of thebainone A (1) so far, a synthesis of codeine (2) and morphine (3) is formally completed, too. However, the known route from 1 to 2 and 3 still leaves room for improvement.^{1,23} Work along this line as well as further optimization of the allylic oxidation of 23 is currently under investigation.

■ ASSOCIATED CONTENT

■ Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c00905>.

Experimental procedures, spectroscopic data, ^1H and ^{13}C NMR spectra (PDF)

Accession Codes

CCDC 1968788–1968790 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

■ AUTHOR INFORMATION

Corresponding Author

Peter Metz – Fakultät Chemie und Lebensmittelchemie, Organische Chemie I, Technische Universität Dresden, 01062

Dresden, Germany; orcid.org/0000-0002-0592-9850;
Email: peter.metz@chemie.tu-dresden.de

Authors

Yuzhou Wang – Fakultät Chemie und Lebensmittelchemie, Organische Chemie I, Technische Universität Dresden, 01062 Dresden, Germany

André Hennig – Fakultät Chemie und Lebensmittelchemie, Organische Chemie I, Technische Universität Dresden, 01062 Dresden, Germany

Thomas Küttler – Fakultät Chemie und Lebensmittelchemie, Organische Chemie I, Technische Universität Dresden, 01062 Dresden, Germany

Christian Hahn – Fakultät Chemie und Lebensmittelchemie, Organische Chemie I, Technische Universität Dresden, 01062 Dresden, Germany

Anne Jäger – Fakultät Chemie und Lebensmittelchemie, Organische Chemie I, Technische Universität Dresden, 01062 Dresden, Germany

Complete contact information is available at: <https://pubs.acs.org/doi/10.1021/acs.orglett.0c00905>

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

This work was supported by the Deutsche Forschungsgemeinschaft (ME 776/22-1).

■ REFERENCES

- (1) (a) Gates, M.; Tschudi, G. *J. Am. Chem. Soc.* **1956**, 78, 1380–1393. (b) Gates, M.; Tschudi, G. *J. Am. Chem. Soc.* **1952**, 74, 1109–1110.
- (2) Recent reviews: (a) Chambers, S. A.; DeSousa, J. M.; Huseman, E. D.; Townsend, S. D. *ACS Chem. Neurosci.* **2018**, 9, 2307–2330. (b) Haider, S.; Chittiboyina, A. G.; Khan, I. A. *Curr. Org. Chem.* **2018**, 22, 148–164. (c) Li, Q.; Zhang, H. *Youji Huaxue* **2017**, 37, 1629–1652. (d) Reed, J. W.; Hudlicky, T. *Acc. Chem. Res.* **2015**, 48, 674–687. (e) Rinner, U.; Hudlicky, T. *Top. Curr. Chem.* **2012**, 309, 33–66. (f) Chida, N. *Top. Curr. Chem.* **2011**, 299, 1–28.
- (3) For total and formal syntheses of morphine and codeine since 2018, see: (a) Zhang, Q.; Zhang, F.-M.; Zhang, C.-S.; Liu, S.-Z.; Tian, J.-M.; Wang, S.-H.; Zhang, X.-M.; Tu, Y.-Q. *Nat. Commun.* **2019**, 10, 2507. (b) Brousseau, J.; Xolin, A.; Barriault, L. *Org. Lett.*

2019, 21, 1347–1349. (c) Rautschek, J.; Jäger, A.; Metz, P. *Org. Lett.* **2018**, 20, 832–835.

(4) For selected recent syntheses of further morphine alkaloids, see: (a) Makarova, M.; Endoma-Arias, M. A. A.; Dela Paz, H. E.; Simionescu, R.; Hudlicky, T. *J. Am. Chem. Soc.* **2019**, 141, 10883–10904. (b) Lipp, A.; Selt, M.; Ferenc, D.; Schollmeyer, D.; Waldvogel, S. R.; Opatz, T. *Org. Lett.* **2019**, 21, 1828–1831. (c) Endoma-Arias, M. A. A.; Makarova, M.; Dela Paz, H. E.; Hudlicky, T. *Synthesis* **2019**, 51, 225–232. (d) Park, K. H.; Chen, D. Y.-K. *Chem. Commun.* **2018**, 54, 13018–13021. (e) Lipp, A.; Ferenc, D.; Gütz, C.; Geffe, M.; Vierengel, N.; Schollmeyer, D.; Schäfer, H. J.; Waldvogel, S. R.; Opatz, T. *Angew. Chem.* **2018**, 130, 11221–11225; *Angew. Chem., Int. Ed.* **2018**, 57, 11055–11059. (f) Park, K. H.; Chen, R.; Chen, D. Y.-K. *Chem. Sci.* **2017**, 8, 7031–7037.

(5) See ref 3c and: (a) Rautschek, J.; Metz, P. *Heterocycles* **2017**, 95, 1106–1120. (b) Erhard, T.; Ehrlich, G.; Metz, P. *Angew. Chem.* **2011**, 123, 3979–3981; *Angew. Chem., Int. Ed.* **2011**, 50, 3892–3894.

(6) See also: (a) Endoma-Arias, M. A. A.; Hudlicky, J. R.; Simionescu, R.; Hudlicky, T. *Adv. Synth. Catal.* **2014**, 356, 333–339. (b) Parsons, P. J.; Penkett, C. S.; Shell, A. J. *Chem. Rev.* **1996**, 96, 195–206. (c) Chandler, M.; Parsons, P. J. *J. Chem. Soc., Chem. Commun.* **1984**, 322–323.

(7) Tius, M. A.; Kerr, M. A. *J. Am. Chem. Soc.* **1992**, 114, 5959–5966.

(8) (a) Varin, M.; Barré, E.; Iorga, B.; Guillou, C. *Chem. - Eur. J.* **2008**, 14, 6606–6608. (b) Guillou, C.; Beunard, J.-L.; Gras, E.; Thal, C. *Angew. Chem.* **2001**, 113, 4881–4882; *Angew. Chem., Int. Ed.* **2001**, 40, 4745–4746.

(9) Markovich, K. M.; Tantishaiyakul, V.; Hamada, A.; Miller, D. D.; Romstedt, K. J.; Shams, G.; Shin, Y.; Fraundorfer, P. F.; Doyle, K.; Feller, D. R. *J. Med. Chem.* **1992**, 35, 466–479.

(10) (a) Koizumi, H.; Yokoshima, S.; Fukuyama, T. *Chem. - Asian J.* **2010**, 5, 2192–2198. (b) Uchida, K.; Yokoshima, S.; Kan, T.; Fukuyama, T. *Heterocycles* **2009**, 77, 1219–1234. (c) Uchida, K.; Yokoshima, S.; Kan, T.; Fukuyama, T. *Org. Lett.* **2006**, 8, 5311–5313.

(11) Still, W. C.; Lewis, A. J.; Goldsmith, D. *Tetrahedron Lett.* **1971**, 12, 1421–1424.

(12) Diago-Meseguer, J.; Palomo-Coll, A. L.; Fernández-Lizarbe, J. R.; Zugaza-Bilbao, A. *Synthesis* **1980**, 547–551.

(13) Lipshutz, B. H.; Pollart, D.; Monforte, J.; Kotsuki, H. *Tetrahedron Lett.* **1985**, 26, 705–708.

(14) Guibe-Jampel, E.; Wakselman, M.; Raulais, D. *J. Chem. Soc., Chem. Commun.* **1980**, 993–994.

(15) Anderson, J. C.; Chapman, H. A. *Org. Biomol. Chem.* **2007**, 5, 2413–2422.

(16) Villemin, D. *Synthesis* **1987**, 154–155.

(17) (a) Taber, D. F.; Neubert, T. D.; Rheingold, A. L. *J. Am. Chem. Soc.* **2002**, 124, 12416–12417. (b) Rice, K. C.; Iijima, I.; Silverton, J. V. *Heterocycles* **1977**, 6, 1157–1165.

(18) See refs 3b, 8a, and: (a) Li, J.; Liu, G.-L.; Zhao, X.-H.; Du, J.-Y.; Qu, H.; Chu, W.-D.; Ding, M.; Jin, C.-Y.; Wei, M.-X.; Fan, C.-A. *Chem. - Asian J.* **2013**, 8, 1105–1109. (b) Stork, G.; Yamashita, A.; Adams, J.; Schulte, G. R.; Chesworth, R.; Miyazaki, Y.; Farmer, J. J. *J. Am. Chem. Soc.* **2009**, 131, 11402–11406. (c) Trost, B. M.; Tang, W.; Toste, F. D. *J. Am. Chem. Soc.* **2005**, 127, 14785–14803. (d) Trost, B. M.; Tang, W. *J. Am. Chem. Soc.* **2002**, 124, 14542–14543.

(19) Goto, K.; Yamamoto, I. *Proc. Jpn. Acad.* **1953**, 29, 457–459.

(20) Tada, H.; Kobayashi, M.; Sawa, Y. K. *Tetrahedron Lett.* **1969**, 10, 1805–1808.

(21) (a) Chochrek, P.; Kurek-Tyrlik, A.; Michalak, K.; Wicha, J. *Tetrahedron Lett.* **2006**, 47, 6017–6020. (b) Tsunoda, T.; Suzuki, M.; Noyori, R. *Tetrahedron Lett.* **1980**, 21, 1357–1358.

(22) Allylic alcohol 27 was isolated after SeO₂ oxidation in 21% yield. In addition, an allylically transposed alcohol SI-3 (see the Supporting Information) was formed in ca. 10% yield that was

probably derived from 27. The relative configuration at C-14 of 27 was elucidated by 2D NMR measurements after reduction with lithium aluminum hydride to give the corresponding N-methylphenol (see the Supporting Information).

(23) See also: (a) Weller, D. D.; Rapoport, H. *J. Med. Chem.* **1976**, 19, 1171–1175. (b) Coop, A.; Rice, K. C. *Heterocycles* **1999**, 50, 39–42.