

Tetrahedron Letters 41 (2000) 915-918

TETRAHEDRON LETTERS

Total synthesis of (±)-desoxycodeine-D: a novel route to the morphine skeleton

Jing-Ping Liou and Chen-Yu Cheng *

Institute of Pharmaceutical Sciences, College of Medicine, National Taiwan University, 1, Sec 1, Jen-Ai Road, Taipei, Taiwan 10018

Received 3 September 1999; revised 1 November 1999; accepted 19 November 1999

Abstract

A novel approach towards the construction of the morphine skeleton was demonstrated by a total synthesis of (\pm) -desoxycodeine-D (11) from 5,6,7,8-tetrahydroisoquinoline and isovanillin. The key steps are two consecutive Pd-catalyzed cyclizations and a Stevens rearrangement for the formation of ring B. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: analgesics; alkaloids; palladium; palladium compounds; rearrangements.

Morphine is a potent analgesic alkaloid with a rigid pentacyclic (ABCNO) structure. A number of total syntheses of morphine have been published since the first one achieved by Gates and Tschudi in 1952.¹ However, a more practical and stereoselective synthetic route for morphine alkaloids remains an attractive research goal for synthetic organic chemists.² In a previous publication,³ we have demonstrated an efficient synthesis of morphine ANO and ACNO fragments, namely spiro[benzofuran-3(2*H*),4'-piperidine] and octahydro-1*H*-benzofuro[3,2-*e*]isoquinoline, by intramolecular Heck reaction. We have now developed a novel approach for the construction of ring B in morphine, based on a Pd-catalyzed intramolecular *N*-benzylation followed by a Stevens rearrangement. The current strategy coupled with our previous method for the construction of the ACNO ring system has provided us an efficient access to the complete pentacyclic skeleton of morphine. Described here is a total synthesis of (\pm)-desoxycodeine-D (**11**),⁴ which serves to exemplify the applicability of the above methodology.



^{*} Corresponding author. Fax: 886-2-23512086; e-mail: cyc@ha.mc.ntu.edu.tw (C.-Y. Cheng)

^{0040-4039/00/\$ -} see front matter © 2000 Elsevier Science Ltd. All rights reserved. P11: S0040-4039(99)02188-7





H₃CO

CH₂OH

CO₂Et

85%

Br

5

CO₂Et

сно

CO₂Et

Βr

vi

88%

4

Scheme 1. Reagents and conditions: i. CH₃I, CH₂Cl₂, room temp.; ii. NaBH₄, MeOH, 0°C; iii. EtOOCCl, KHCO₃, ClCH₂CH₂Cl, reflux; iv. NaOH, MeOH, 0°C; v. 2-bromoisovanillin, DEAD, $(n-C_4H_9)_3P$, THF.; vi. NaBH₄, MeOH, 0°C; vii. Pd(OAc)₂, PPh₃, Et₃N, CH₃CN, 120–130°C; viii. TBDMSCl, imidazole, THF; ix. $(n-C_4H_9)_4N^+F^-$, THF, room temp.; x. NCS, PPh₃, THF, room temp.; xi. Pd(PPh₃)₄, Et₃N, CH₃CN, 120–130°C; xiii. CH₃I, CH₂Cl₂, room temp.; xiii. PhLi, ether, 0°C

As outlined in Scheme 1, the synthesis of **11** starts from readily available 5,6,7,8tetrahydroisoquinoline and isovanillin. Thus, 5-acetoxy-5,6,7,8-tetrahydroisoquinoline (**1**)^{5,6} derived from 5,6,7,8-tetrahydroisoquinoline was treated with iodomethane, followed by NaBH₄ reduction, to give the octahydroisoquinoline **2**.⁷ Compound **2** was treated with ethyl chloroformate followed by hydrolysis of the acetate group to give carbamate **3**. The condensation of compound **3** with 2-bromoisovanillin⁸ under Mitsunobu conditions⁹ provided compound **4**, which was reduced with NaBH₄ to give the benzyl alcohol intermediate **5**. The formation of the O-ring was achieved when compound **5** was subjected to Heck reaction conditions, and the tetracyclic (ACNO) compound **6**¹⁰ was obtained in 42% yield. The yield of the above intramolecular cyclization was significantly increased via prior protection of the alcohol function in **5** as a silyl ether. Compound **6** was then converted to the benzyl chloride **8** via treatment with *N*-chlorosuccinimide and triphenylphosphine. Our original plan for the construction of ring B was to utilize the documented Pd-catalyzed cyclization of benzyl halides containing alkenes.¹¹ However, when compound **8** was subjected to Heck reaction conditions $(Pd(PPh_3)_4, Et_3N, CH_3CN, 120^{\circ}C)$, instead of giving the anticipated compound **12**, an intramolecular *N*-benzylation occurred and provided the tertiary amine **9**.¹⁰ A likely mechanism for the formation of compound **9** from **8** is given in Scheme 2. Prompted by literature reports¹² of Stevens rearrangement of quaternary tetrahydroisoquinoline alkaloids, compound **9** was first converted into the corresponding *N*-methylammonium iodide **10**, which was then treated with PhLi in ether. To our gratification, compound **10** underwent the anticipated Stevens rearrangement, and provided (±)-desoxycodeine-D (**11**) in 83% yield.¹³



In summary, we have demonstrated a novel synthetic route to the morphine skeleton, starting from 5,6,7,8-tetrahydroisoquinoline and isovanillin. Notable features of the synthesis include an unexpected Pd-catalyzed intramolecular *N*-benzylation and the efficient formation of ring B via a Stevens rearrangement. Further work towards a total synthesis of (–)-morphine is currently underway in our laboratory.

Acknowledgements

This research was supported by the National Science Council of the R.O.C. under grant no. NSC 87-2113-M002-027.

References

- For reviews on the total synthesis of morphine alkaloids, see: (a) Hudlicky, T.; Butora, G.; Fearnley, S. P.; Gum, A. G.; Stabile, M. R. In *Studies in Natural Products Chemistry*; Atta-ur-Rahman, Ed.; Elsevier: Amsterdam, 1996; pp. 43–154. (b) Maier, M. In *Organic Synthesis Highlights II*; Waldmann, H., Ed.; VCH: Weinheim, 1995; pp. 357–369. (c) Szántay, G.; Dörnyei, G. In *The Alkaloids*; Cordell, G. A.; Brossi, A., Eds.; Academic Press: New York, 1994; Vol. 45, p. 127 ff.
- For more recent studies on the total synthesis of morphine, see: (a) Parsons, P. J.; Penkett, C. S.; Shell, A. Chem. Rev. 1996, 96, 195–206. (b) Butora, G.; Hudlicky, T.; Fearnley, S. P.; Gum, A. G.; Stabile, M. R.; Abboud, K. Tetrahedron Lett. 1996, 37, 8155–8158. (c) Mulzer, J.; Dürner, G.; Trauner, D. Angew. Chem., Int. Ed. Engl. 1996, 35, 2830–2832. (d) Mulzer, J.; Bats, J. W.; List, B.; Opatz, T.; Trauner, D. Synlett 1997, 441–444. (e) White, J. D.; Hrnciar, P.; Stappenbeck, F. J. Org. Chem. 1997, 62, 5250–5251. (f) Trauner, D.; Bats, J. W.; Werner, A.; Mulzer, J. J. Org. Chem. 1998, 63, 5908–5918.
- 3. Cheng, C. Y.; Liou, J. P.; Lee, M. J. Tetrahedron Lett. 1997, 38, 4571–4574.
- 4. Small, L. F.; Mallone, J. E. J. Org. Chem. 1940, 5, 350-354.
- 5. Cheng, C. Y.; Hsin, L. W.; Liou, J. P. Tetrahedron 1996, 52, 10935-10944.
- 6. Boyd, D. R.; Davies, R. J. H.; Hamilton, L.; McCullough, J. J. J. Chem. Soc., Perkin Trans. 1 1992, 31–35.
- 7. All compounds synthesized were characterized by ¹H, ¹³C, IR, and mass spectrometric analysis. Chemical yields refer to isolated, purified products.
- 8. Hazlet, S. E.; Brotherton, R. J. J. Org. Chem. 1962, 27, 3253-3256.
- 9. Mitsunobu, O. Synthesis 1981, 1–28.
- 10. Selected spectral and analytical data: **6**: ¹H NMR (200 MHz, CDCl₃) δ 1.22 (t, *J*=7.0 Hz, 3H), 1.58 (m, 1H), 1.75 (m, 1H), 1.98–2.15 (m, 3H), 2.27 (m, 1H), 2.44–2.56 (m, 1H), 3.00–3.20 (m, 1H), 3.70–3.90 (m, 1H), 4.05–4.15 (m, 4H), 4.45–4.62 (m, 3H), 5.95 (m, 1H), 6.69 (d, *J*=8.4 Hz, 1H), 6.82 (d, *J*=8.3 Hz, 1H); ¹³C NMR (50 MHz) δ 15.1, 21.6, 27.7, 35.1, 40.0, 48.4, 49.4, 56.2, 61.7, 62.3, 88.8, 110.0, 111.6, 122.9, 129.9, 130.9, 132.2, 133.6, 144.7, 148.8, 156.2; HRMS *m/e* calcd for C₂₀H₂₅NO5⁺: 359.1733, found: 359.1732. **9**: ¹H NMR (400 MHz, CDCl₃) δ 1.65–1.75 (m, 1H), 1.80–1.92 (m, 1H), 1.93–1.98 (m, 1H), 2.09–2.21 (m, 1H), 2.23–2.31 (m, 2H), 3.17 (m, 2H), 3.53 (d, *J*=17 Hz, 1H), 3.74 (d, *J*=17.1 Hz, 1H), 3.81 (s, 3H), 4.16 (d, *J*=17.4 Hz, 1H), 4.28 (d, *J*=17.4 Hz, 1H), 4.63 (m, 1H), 5.42 (m, 1H), 6.49 (d, *J*=8.2 Hz, 1H), 6.66 (d, *J*=8.1 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 1.62–1.36 (m, 1H), 1.81–1.91 (m, 2H), 1.99–2.10 (m, 3H), 2.45 (s, 3H), 2.52–2.65 (m, 3H), 3.28 (d, *J*=18.1 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 1.32–1.36 (m, 1H), 1.81–1.91 (m, 2H), 1.99–2.10 (m, 3H), 2.45 (s, 3H), 2.52–2.65 (m, 3H), 3.28 (d, *J*=8.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 1.32–1.36 (m, 1H), 1.81–1.91 (m, 2H), 1.99–2.10 (m, 3H), 2.45 (s, 3H), 2.52–2.65 (m, 3H), 3.28 (d, *J*=8.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 1.32–1.36 (m, 1H), 1.81–1.91 (m, 2H), 1.99–2.10 (m, 3H), 2.45 (s, 3H), 2.52–2.65 (m, 3H), 3.28 (d, *J*=8.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 1.32–1.36 (m, 1H), 1.81–1.91 (m, 2H), 1.99–2.10 (m, 3H), 2.45 (s, 3H), 2.52–2.65 (m, 3H), 3.28 (d, *J*=8.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 21.0, 26.5, 27.8, 34.8, 41.8, 43.3, 46.2, 53.3, 56.4, 61.7, 90.1, 113.2, 118.0, 119.1, 127.3, 132.4, 137.0, 143.3, 143.5; HRMS *m/e* calcd for C₁₈H₂₁NO₂⁺: 283.1573, found: 283.1579.
- 11. Wu, G. Z.; Lamaty, F.; Negishi, E. I. J. Org. Chem. 1989, 54, 2507-2508.
- (a) Kametani, T.; Kobari, T.; Fukumoto, K.; Fujihara, M. J. Chem. Soc. C 1971, 1796–1800. (b) Ito, K.; Furukawa, H.; Iida, T.; Lee, K. H.; Soine, T. O. J. Chem. Soc., Chem. Commun. 1974, 1037–1038. (c) Kametani, T.; Huang, S. P.; Ujiie, A.; Ihara, M.; Fukumoto, K. Heterocycles 1976, 4, 1223–1228. (d) Kametani, T.; Ujiie, A.; Huang, S. P.; Ihara, M.; Fukumoto, K. J. Chem. Soc., Perkin Trans. 1 1977, 394–397. (e) Kametani, T.; Huang, S. P.; Koseki, C.; Ihara, M.; Fukumoto, K. J. Org. Chem. 1977, 42, 3040–3046.
- 13. The ¹H NMR data is in agreement with that reported in the literature. See: Kshirsagar, T. A.; Portoghese, P. S. *J. Org. Chem.* **1998**, *63*, 1706–1708.