

Synthesis of N-Arylpyrazoles by Palladium-Catalyzed Coupling of Aryl Triflates with Pyrazole Derivatives

Shunsuke Onodera, Takuya Kochi, and Fumitoshi Kakiuchi

J. Org. Chem., **Just Accepted Manuscript** • DOI: 10.1021/acs.joc.9b00673 • Publication Date (Web): 26 Apr 2019

Downloaded from <http://pubs.acs.org> on April 26, 2019

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

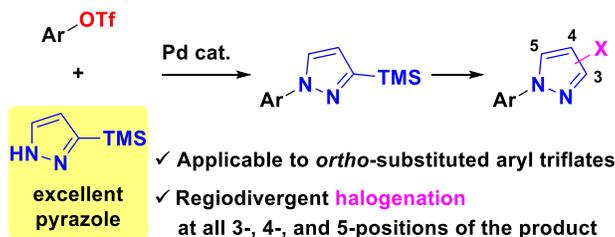
Synthesis of *N*-Arylpyrazoles by Palladium-Catalyzed Coupling of Aryl Triflates with Pyrazole Derivatives

Shunsuke Onodera, Takuya Kochi, and Fumitoshi Kakiuchi*

Department of Chemistry, Faculty of Science and Technology, Keio University, 3-14-1 Hiyoshi, Kohoku-ku, Yokohama, Kanagawa 223-8522, Japan.

kakiuchi@chem.keio.ac.jp

Table of Contents/Abstract Graphic



Abstract

A method for synthesis of *N*-arylpyrazoles by palladium-catalyzed coupling of aryl triflates with pyrazole derivatives is described. Using *t*BuBrettPhos as a ligand, the palladium-catalyzed C–N coupling of a variety of aryl triflates including *ortho*-substituted ones with pyrazole derivatives proceeded efficiently to give *N*-arylpyrazole products in high yields. 3-

1
2
3 Trimethylsilylpyrazole was found as an excellent pyrazole substrate for the coupling, and the
4 corresponding product, 1-aryl-3-trimethylsilylpyrazole, also served as a great template for
5
6 syntheses of *N*-arylpyrazole derivatives, as demonstrated by regioselective halogenation at 3-, 4-
7
8 and 5-positions of the pyrazole ring.
9
10

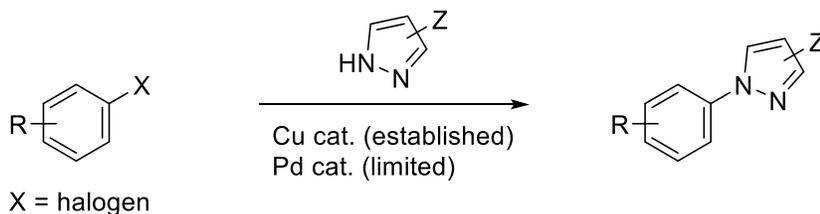
11
12
13
14
15 *N*-Arylpyrazoles are a class of important compounds because of their interesting biological
16 activities as agrochemicals and pharmaceuticals.¹ A variety of strategies have been developed for
17 preparation of *N*-arylpyrazoles,^{2,3} and one of the most efficient methods for the synthesis of *N*-
18 arylpyrazoles is transition-metal-catalyzed cross-coupling of aryl halides with 1*H*-pyrazoles
19
20
21
22
23
24 **(Figure 1)**.² Copper catalysts have been frequently used in the C–N coupling reaction to form *N*-
25 arylpyrazoles, and many researchers have developed new copper catalyst systems to improve the
26 reaction efficiency.⁴ On the other hand, synthesis of *N*-arylpyrazoles by palladium-catalyzed C–
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
N coupling of aryl halides with 1*H*-pyrazoles has been relatively less explored,^{5,6} and for
example, only a few examples of the coupling of *ortho*-substituted aryl halides has been reported
in publications.^{5c,5f}

40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
Aryl triflates are also common electrophiles in palladium-catalyzed C–N coupling reactions,
while copper catalysts have rarely been applied to the reaction of aryl triflates^{7,8} because of their
reluctance to react with this class of substrates.^{6a} The use of aryl triflates in the cross-coupling
reactions is sometimes advantageous because aryl triflates can be easily prepared from phenolic
compounds, which are more widely accessible than aryl halides.⁹ However, metal-catalyzed
coupling of aryl triflates with 1*H*-pyrazoles have been achieved only for limited sets of
substrates,⁸ and development of a general protocol for synthesis of *N*-arylpyrazoles by the C–N
coupling is still desired.¹⁰ During the course of our study of the rhodium-catalyzed deallylative

1
2
3 alkenylation reactions,¹¹ we envisioned that a method for the C–N coupling reactions of *ortho*-
4 substituted aryl triflates with a pyrazole derivative would provide an efficient route to prepare
5 substrates for the deallylative alkenylation. After screening of the reaction conditions, we found
6 that the palladium-catalyzed C–N coupling of an aryl triflates derived from an *ortho*-prenylated
7 phenol with 3-trimethylsilylpyrazole proceeded efficiently, but the generality of the C–N
8 coupling reaction was not investigated.
9
10
11
12
13
14
15
16

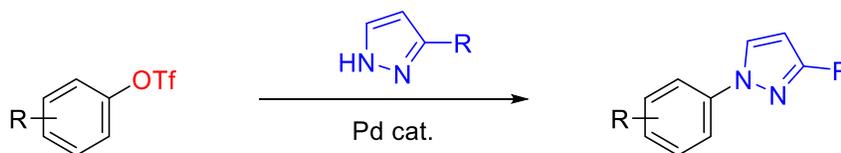
17 Herein, we report the details of the C–N coupling reaction to form *N*-arylpiperazines including
18 modification of the previous reaction conditions and wide substrate scope. The great utility of 3-
19 trimethylsilylpyrazole was also demonstrated by the excellent reactivity in the C–N coupling
20 reaction and the applicability of the coupling products for regiodivergent halogenation at all 3-,
21 4-, and 5-positions on the pyrazole ring.
22
23
24
25
26
27
28
29
30

31 **(A) Coupling of Aryl Halides with Pyrazoles (Well Known)**



40 **(B) This Work: Coupling of Aryl Triflates with Pyrazoles**

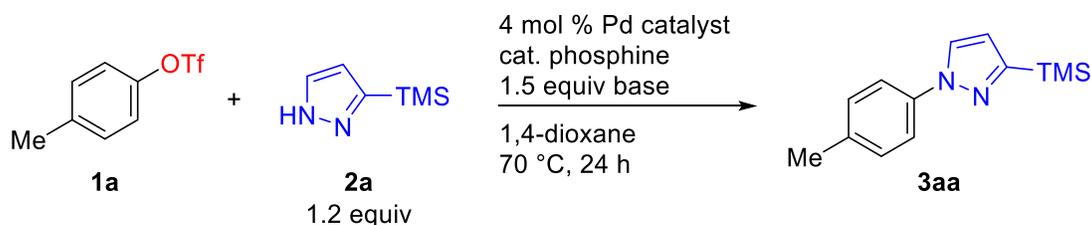
41 Aryl triflates/pyrazoles couplings have been reported only in limited cases.



- 49 ✓ Applicable to various substrates including *ortho*-substituted aryl triflates
50 ✓ When R = TMS, the products can be halogenated regioselectively
51 at all 3-, 4-, and 5-positions
52
53
54

55 **Figure 1. Metal-catalyzed C–N coupling of aryl (pseudo)halides with pyrazole derivatives.**
56
57
58
59
60

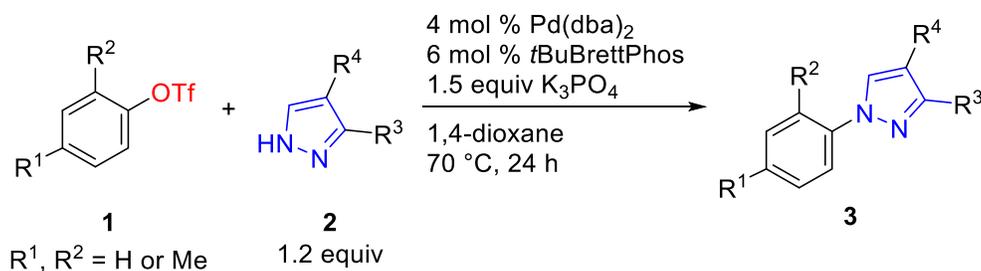
1
2
3 In our previous publication showing one example of the aryl triflate/pyrazole coupling, 10
4 mol % of 1/2 Pd₂(dba)₃ and 20 mol % of *t*BuBrettPhos were used to achieve 85% yield of the
5 product.¹¹ We began our investigation of the general protocol of the C–N coupling reaction by
6 modification of the previous protocol, especially to reduce the amounts of palladium catalysts
7 and ligands. We first examined the coupling of simple aryl triflate substrate **1a** with 3-
8 trimethylsilylpyrazole (**2a**) to see if the catalyst loading can be reduced. When the reaction of *p*-
9 tolyl triflate (**1a**) with **2a** was performed using 4 mol % of 1/2 Pd₂(dba)₃ and 8 mol % of
10 *t*BuBrettPhos in 1,4-dioxane in the presence of potassium phosphate as a base at 70 °C for 24 h,
11 the desired *N*-arylpyrazole product **3aa** was obtained in 96% NMR yield (Table 1, entry 1). The
12 use of Pd(dba)₂ instead of Pd₂(dba)₃ was also effective to give **3aa** in 94% NMR yield (entry 2).
13
14 Reduction of the amount of *t*BuBrettPhos was then investigated, and the reactions were found to
15 proceed efficiently using only 6 mol % of the ligand with either Pd₂(dba)₃ or Pd(dba)₂ (entries 3
16 and 4). Particularly, the reaction with 4 mol % of Pd(dba)₂ and 6 mol % of *t*BuBrettPhos
17 provided **3aa** in 98% NMR yield (entry 4). The use of other ligands such as *t*BuXPhos,
18 BrettPhos, and XPhos gave either lower yield or no detectable amount of **3aa** (entries 5-7). The
19 reaction using the *t*BuXPhos/NaO^tBu system, developed for aryl bromide/pyrazole coupling,^{5a}
20 did not give **3aa** and resulted in decomposition of the aryl triflate substrate (entry 8). The
21 reaction in toluene also provided **3aa** only in 58% NMR yield (entry 9). Therefore, the reaction
22 conditions shown in entry 4 were determined to be optimum and used for further examination.
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Table 1. Palladium-Catalyzed C–N Coupling of Aryl Triflate **1a** with Pyrazole Derivative**2a^a**

entry	Pd catalyst	phosphine	base	solvent	NMR yield
1	1/2 Pd ₂ (dba) ₃	8 mol % <i>t</i> BuBrettPhos	K ₃ PO ₄	1,4-dioxane	96%
2	Pd(dba) ₂	8 mol % <i>t</i> BuBrettPhos	K ₃ PO ₄	1,4-dioxane	94%
3	1/2 Pd ₂ (dba) ₃	6 mol % <i>t</i> BuBrettPhos	K ₃ PO ₄	1,4-dioxane	94%
4	Pd(dba) ₂	6 mol % <i>t</i> BuBrettPhos	K ₃ PO ₄	1,4-dioxane	98%
5	Pd(dba) ₂	6 mol % <i>t</i> BuXPhos	K ₃ PO ₄	1,4-dioxane	63%
6	Pd(dba) ₂	6 mol % BrettPhos	K ₃ PO ₄	1,4-dioxane	nd ^b
7	Pd(dba) ₂	6 mol % XPhos	K ₃ PO ₄	1,4-dioxane	nd ^b
8	Pd(dba) ₂	6 mol % <i>t</i> BuXPhos	NaO ^t Bu	1,4-dioxane	nd ^b
9	Pd(dba) ₂	6 mol % <i>t</i> BuBrettPhos	K ₃ PO ₄	toluene	58%

^aReaction conditions: **1a** (0.3 mmol), **2a** (0.36 mmol), Pd catalyst (0.012 mmol), phosphine (0.018 or 0.024 mmol), solvent (1.5 mL), 70 °C, 24 h. ^bNot detected.

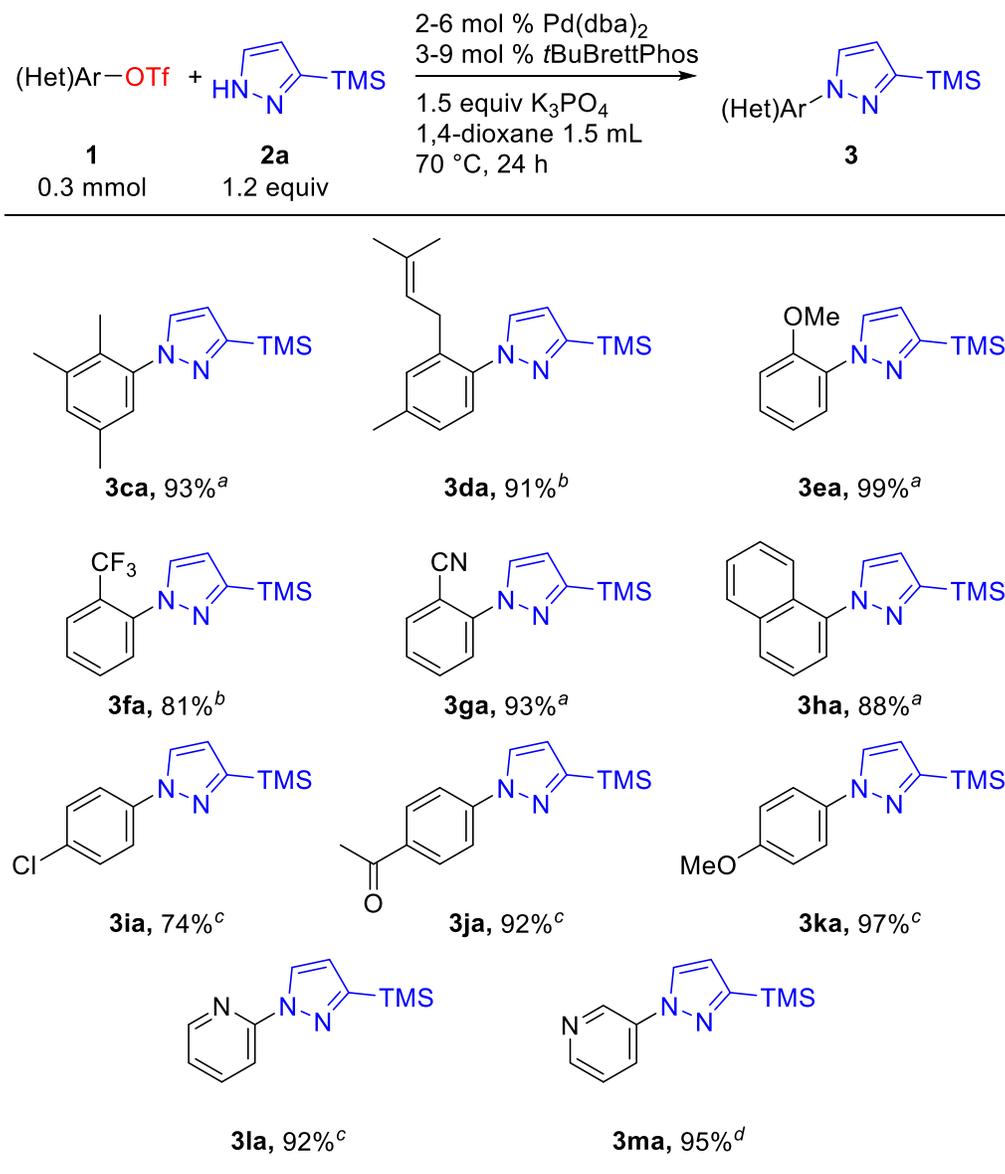
1
2
3 The C–N coupling was then examined for various sets of substrates by isolating the *N*-
4 arylpyrazole products (Table 2). Product **3aa** formed by the reaction of *p*-tolyl triflate (**1a**) with
5 **2a** under the standard conditions was isolated in 98% yield (entry 1), and in this case, reduction
6 of the catalyst loading to 2 mol % of Pd(dba)₂ and 3 mol % of *t*BuBrettPhos had little effect on
7 the yield (entry 2). The C–N coupling of **1a** with 3-substituted pyrazoles such as those bearing
8 *tert*-butyl (**2b**), phenyl (**2c**), and methyl (**2d**) groups gave the desired products **3ab-3ad** in high
9 yields (entries 3-5). In addition, pyrazoles with no substituents at the 3-position were also
10 applicable to this reaction, and the coupling of **1a** with pyrazole (**2e**) and 4-trimethylsilylpyrazole
11 (**2f**) provided the corresponding products **3ae** and **3af** in 84 and 69% yields, respectively (entries
12 6 and 7). The reaction of *o*-tolyl triflate (**1b**) with various pyrazoles was next examined. Under
13 the standard reaction conditions, 3-substituted pyrazoles **2a-2c** were coupled smoothly even with
14 the sterically-congested aryl triflate **1b** to give products **3ba-3bc** in excellent yields (entries 8-
15 10).¹² However, the reaction of **1b** with pyrazole derivatives possessing a smaller group such as
16 H or Me group, at the 3-position (**2d-2f**) resulted in low yields (entry 11-13). It is unclear why
17 the reactivity of the pyrazole derivatives with large substituents at the 3-position is higher than
18 those with a small group in this reaction, but as Buchwald and coworkers reported for a
19 palladium-catalyzed coupling reaction,¹³ formation of pyrazolyl-bridged palladium dimers,
20 which may inhibit the reaction, are less likely to occur, when the 3-position possesses a large
21 substituent.
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Table 2. Palladium-Catalyzed C–N Coupling of Aryl Triflates with 3-Substituted Pyrazole Derivatives^a**Derivatives^a**

entry	aryl triflate 1	R ¹	R ²	pyrazole 2	R ³	R ⁴	product 3	yield
1	1a	Me	H	2a	TMS	H	3aa	98%
2 ^b	1a	Me	H	2a	TMS	H	3aa	97%
3	1a	Me	H	2b	^t Bu	H	3ab	quant
4	1a	Me	H	2c	Ph	H	3ac	97%
5	1a	Me	H	2d	Me	H	3ad	90%
6	1a	Me	H	2e	H	H	3ae	84%
7	1a	Me	H	2f	H	TMS	3af	69%
8	1b	H	Me	2a	TMS	H	3ba	95%
9	1b	H	Me	2b	^t Bu	H	3bb	95%
10	1b	H	Me	2c	Ph	H	3bc	92%
11	1b	H	Me	2d	Me	H	3bd	25%
12	1b	H	Me	2e	H	H	3be	trace
13	1b	H	Me	2f	H	TMS	3bf	trace

^aReaction conditions: **1** (0.3 mmol), **2** (0.36 mmol), Pd(dba)₂ (0.012 mmol), *t*BuBrettPhos (0.018 mmol), 1,4-dioxane (1.5 mL), 70 °C. ^bPerformed with 0.006 mmol (2 mol %) Pd(dba)₂ and 0.009 mmol (3 mol %) *t*BuBrettPhos.

1
2
3 The C–N coupling reaction was applicable to aryl triflates possessing a variety of functional
4 groups (Table 3). The reaction of 2,3,5-trimethylphenyl triflate (**1c**) afforded 93% yield of
5 product **3ca**. A substrate possessing an *ortho*-prenyl group (**1d**) coupled with **2a** effectively to
6 give **3da** in 91% yield by using 6 mol % of the palladium catalyst. Aryl triflate **1e** bearing an
7 *ortho*-methoxy group showed excellent reactivity to give **3ea** in 99% yield. Substrates bearing
8 electron-withdrawing trifluoromethyl and cyano groups reacted with **2a** as well to afford the
9 corresponding *N*-arylpyrazoles **3fa** and **3ga** in 81% and 93% yields, respectively. 1-
10 Naphthyltriflate (**1h**) could also be applied for the C–N coupling reaction to form **3ha** in 88%
11 yield.¹⁴ The reactions of aryl triflates with various para substituents such as chloro, acetyl, and
12 methoxy groups proceed in the presence of only 2 mol % of the palladium catalyst to give the
13 corresponding *N*-arylpyrazoles **3ia-3ka** in 74-97% yields. The coupling of **1i** resulted in lower
14 yield than others, because further C–N coupling at the chlorinated carbon of **3ia** partially
15 occurred as a side reaction. The reaction of 2-pyridyl triflate (**1l**) with **2a** gave **3la** in 92% yield
16 using 2 mol % of the palladium catalyst. In the coupling of 3-pyridyl triflate (**1m**), pyrazole **2a**
17 was used as a limiting reagent because **3ma** and **2a** were difficult to separate by silica gel
18 column chromatography and 95% yield of **3ma** was obtained. In the reactions with 3-substituted
19 pyrazoles, two regioisomeric *N*-arylation products, 3- and 5-substituted *N*-arylpyrazoles could be
20 obtained theoretically, but only 3-substituted *N*-arylpyrazoles were observed in the coupling
21 reactions described here. The structures of the products were determined by ¹H NMR using a
22 coupling constant analysis method reported by Cristau and Taillefer.^{4b}
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Table 3. Palladium-Catalyzed C–N Coupling of Various Aryl Triflates with 2a

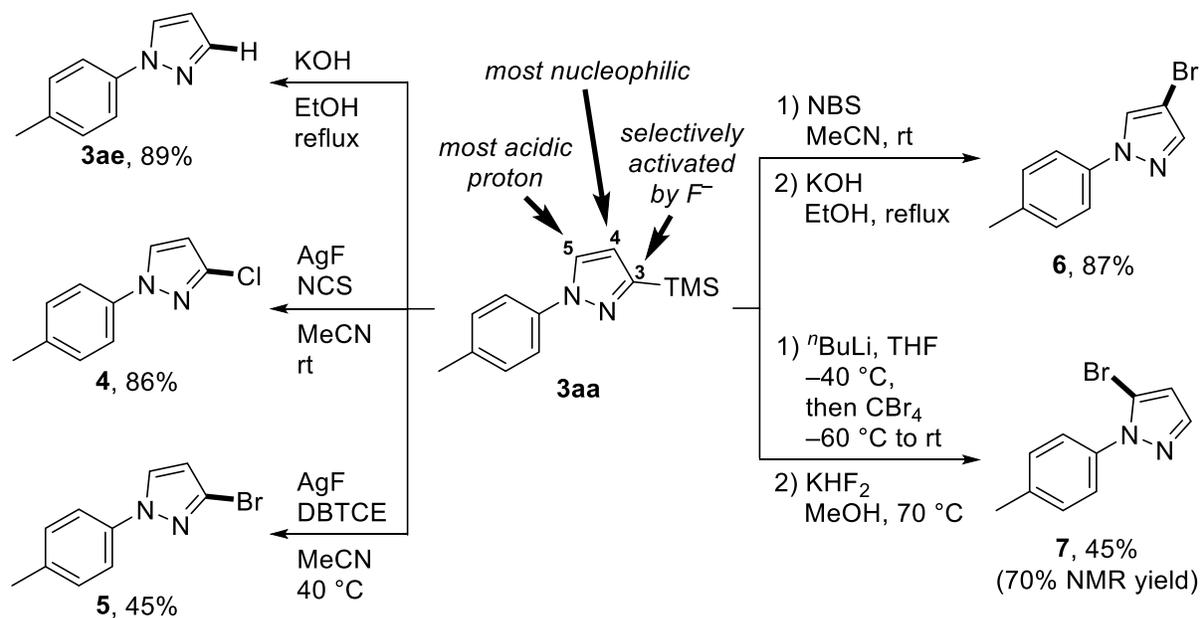
^aReaction conditions: **1** (0.3 mmol), **2a** (0.36 mmol), Pd(dba)₂ (0.012 mmol, 4 mol %), *t*BuBrettPhos (0.018 mmol, 6 mol %), K₃PO₄ (1.5 equiv), 1,4-dioxane (1.5 mL), 70 °C, 24 h.

^bPerformed with 0.018 mmol of Pd(dba)₂ (6 mol %) and 0.027 mmol of *t*BuBrettPhos (9 mol %).

^cPerformed with 0.006 mmol of Pd(dba)₂ (2 mol %) and 0.009 mmol of *t*BuBrettPhos (3 mol %).

1
2
3 ^dPerformed with 0.36 mmol of **1m**, 0.3 mmol of **2a**, 0.006 mmol of Pd(dba)₂ (2 mol %) and
4
5 0.009 mmol of *t*BuBrettPhos (3 mol %).
6
7
8
9
10

11
12 The 1-aryl-3-trimethylsilylpyrazole framework, formed by the C–N coupling with **2a**, found to
13
14 serve as a great template for the synthesis of various *N*-arylpyrazole derivatives (Figure 2). First,
15
16 deprotection of the trimethylsilyl group of **3aa** was readily accomplished by treatment with
17
18 ethanolic KOH solution to give **3ae** in 89% yield. Regioselective introduction of halogeno
19
20 groups at all 3-, 4-, and 5-positions were also possible, because each of these positions has
21
22 mutually different character and orthogonal reactivities.¹⁵ The 3-position of **3aa** has the
23
24 trimethylsilyl group and its removal under mild conditions would generate a carbanion at the 3-
25
26 position, which may react with electrophilic halogenating agents. Therefore, the reaction of **3aa**
27
28 with *N*-chlorosuccinimide (NCS) in the presence of silver fluoride in acetonitrile provided 1-
29
30 aryl-3-chloropyrazole **4** in 86% yield. The use of 1,2-dibromotetrachloroethane (DBTCE) instead
31
32 of NCS also gave the corresponding bromination product **5** in 45% yield. The 4-position of **3aa**
33
34 is the most nucleophilic, and direct reaction of *N*-bromosuccinimide (NBS) with **3aa**, followed
35
36 by deprotection of the trimethylsilyl group led to the formation of 1-aryl-4-bromopyrazole **6** in
37
38 87% yield. The 5-position of **3aa** bears the most acidic proton on the pyrazole ring. The reaction
39
40 of **3aa** with *n*-butyllithium and then with tetrabromomethane installed a bromo group at the 5-
41
42 position, and subsequent deprotection of the trimethylsilyl group formed 1-aryl-5-bromopyrazole
43
44 **7** in 45% yield (70% NMR yield).
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60



23 **Figure 2. Regiodivergent derivatization of 1-aryl-3-trimethylsilylpyrazole 3aa**

24
25
26
27
28
29 In summary, we developed a protocol for the C–N coupling between aryl triflates and pyrazole
30 derivatives using 2-6 mol % of a palladium catalyst and 3-9 mol % of *t*BuBrettPhos. The
31 reaction was applicable to various sets of substrates including sterically-congested aryl triflates
32 bearing ortho substituents and various reactive functional groups tolerated the reaction
33 conditions. The 1-aryl-3-trimethylsilylpyrazole framework, readily formed by the C–N coupling
34 developed here, can serve as a useful template for the synthesis of various *N*-arylpyrazole
35 derivatives, as demonstrated by regiodivergent halogenation at all 3-, 4-, and 5-positions.
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Experimental Section

Unless otherwise noted, all reactions were carried out under nitrogen, and commercial reagents were used as received. Pd₂(dba)₃ was purchased from Aldrich and used as received. Pd(dba)₂ was purchased from Tokyo Chemical Industry Co., Ltd. and used as received. *t*BuBrettPhos, *t*BuXPhos, BrettPhos and XPhos were purchased from Aldrich and used as received. K₃PO₄ was purchased from Junsei Chemical Co., Ltd. and dried by Kugelrohr prior to use. 1,4-Dioxane was purchased from FUJIFILM Wako Pure Chemical Corporation and used as received. Pyrazoles **2a**¹¹, **2b**¹⁶ and **2f**¹⁷ were prepared according to the literature procedures. Pyrazoles **2c** and **2d** were purchased from Tokyo Chemical Industry Co., Ltd. and used as received. Pyrazole **2e** was purchased from Nacalai Tesque, Inc. and used as received. Aryl triflates **1a**, **1b**, **1e-1k** were prepared according to the literature procedure using corresponding phenol derivatives.¹⁸ Heteroaryl triflates **1l**^{9b} and **1m**^{9b} were prepared according to the literature procedures. ¹H and ¹³C{¹H} spectra were recorded on a JEOL ECX-400, AL-400, or ALPHA-400 spectrometer. IR spectra were recorded on a JASCO FT/IR-410 infrared spectrometer. Flash chromatography was performed using a EPCLC-AI-580S (Yamazen Corporation) with silica gel 40 μm. ESI-MS was performed on a JEOL JMS-T100LCS.

Preparation of 2,3,5-Trimethylphenyl Trifluoromethanesulfonate (1c). To a solution of 2,4,5-trimethylphenol (2.72 g, 20.0 mmol) and 2.4 mL of pyridine in 50 mL dichloromethane at 0 °C was added Tf₂O (6.41 g, 22.7 mmol) dropwise, and the mixture was stirred for 2 h at room temperature. After this period, 1 M HCl aq was added to the mixture, which was then extracted three times with dichloromethane. The combined organic layers were washed with brine, dried over MgSO₄, filtered and concentrated. Kugelrohr distillation of the crude material afforded 4.79 g of **1c** (17.9 mmol, 90% yield) as a colorless oil; ¹H NMR (400 MHz, CDCl₃): δ 2.22 (s, 3H),

1
2
3 2.28 (s, 3H), 2.31 (s, 3H), 6.90 (s, 1H), 6.99 (s, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3): δ 12.5,
4
5 20.1, 20.8, 118.6 (q, $J = 320.0$ Hz), 119.1, 126.1, 130.5, 136.8, 139.4, 148.3; IR (neat): 2927 w,
6
7 2871 w, 1623 w, 1574 w, 1495 w, 1451 w, 1420, 1281 w, 1248 s, 1213 s, 1143 s, 1040 s, 1018
8
9 w, 968 m, 917 s, 853 m, 823 s, 766 w, 748 w, 711 w, 661 w, 610 m cm^{-1} ; HRMS (ESI-TOF)
10
11 m/z: $[\text{M}+\text{Na}^+]$ Calcd for $\text{C}_{10}\text{H}_{11}\text{F}_3\text{NaO}_3\text{S}^+$ 291.0273; Found 291.0265.

12
13
14 *Preparation of 4-Methyl-2-(3-methylbut-2-en-1-yl)phenol.* 4-Methyl-2-(3-methylbut-2-en-1-
15
16 yl)phenol was prepared according to the literature procedure.¹⁹ To a solution of *p*-cresol (4.32 g,
17
18 39.9 mmol) in 80 mL Et_2O was added Na (1.95 g, 84.8 mmol) portionwise, and the mixture was
19
20 stirred at room temperature for 30 min. Prenyl chloride (4.16 g, 39.8 mmol) was added to the
21
22 resulting mixture and heated to reflux for 1 h. After this period, the mixture was acidified with
23
24 0.1 N HCl aq and extracted three times with Et_2O . The combined organic layers were washed
25
26 with brine, dried over MgSO_4 , filtered and concentrated. Purification of the crude material by
27
28 silica gel column chromatography (hexane:AcOEt = 91:9) afforded 5.88 g of 4-methyl-2-(3-
29
30 methylbut-2-en-1-yl)phenol (33.4 mmol, 84% yield) as a colorless oil. The analytical data for 4-
31
32 methyl-2-(3-methylbut-2-en-1-yl)phenol are in good agreement with those reported in
33
34 literature.²⁰

35
36
37
38
39
40 *Preparation of 4-Methyl-2-(3-methylbut-2-en-1-yl)phenyl Trifluoromethanesulfonate (1d).* To
41
42 a solution of 4-methyl-2-(3-methylbut-2-en-1-yl)phenol (0.798 g, 4.53 mmol) and 0.6 mL of
43
44 pyridine in 20 mL dichloromethane at 0 °C was added Tf_2O (1.57 g, 5.56 mmol) dropwise, and
45
46 the mixture was stirred for 4.5 h at room temperature. After this period, 1 M HCl aq was added
47
48 to the mixture, which was then extracted three times with dichloromethane. The combined
49
50 organic layers were washed with brine, dried over MgSO_4 , filtered and concentrated.
51
52 Purification of the crude material by silica gel column chromatography (hexane:AcOEt = 98:2)
53
54
55
56
57
58
59
60

1
2
3 afforded 1.32 g of **1d** (4.28 mmol, 94% yield) as a colorless oil.; ^1H NMR (400 MHz, CDCl_3): δ
4 1.71 (s, 3H), 1.77 (d, $J = 1.2$ Hz, 3H), 2.34 (s, 3H), 3.38 (d, $J = 7.1$ Hz, 2H), 5.21-5.26 (m, 1H),
5
6 7.03-7.12 (m, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3): δ 17.8, 20.9, 25.7, 28.3, 118.6 (q, $J = 325$
7
8 Hz), 120.5, 120.8, 128.1, 131.4, 134.0, 134.4, 138.3, 145.9; IR (neat): 3014 m, 2969 s, 2917 s,
9
10 1860 s, 1612 m, 1507 s, 1448 s, 1376 s, 1347 m, 1260 s, 1203 s, 1149 m, 1108 s, 1040 w, 985 w,
11
12 922 m, 879 m, 848 m, 811 s, 787 s, 737 w, 717 w cm^{-1} ; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}^+]$ Calcd
13
14 for $\text{C}_{13}\text{H}_{15}\text{F}_3\text{NaO}_3\text{S}^+$ 331.0586; Found 331.0588.
15
16
17
18

19 **General Procedure A for Palladium-Catalyzed Amination of Aryl Triflates with Pyrazole**
20
21 **Derivatives.** In a glove box, $\text{Pd}(\text{dba})_2$, (2-6 mol %) *t*BuBrettPhos (3-9 mol %), K_3PO_4 (1.5
22
23 equiv) and 1,4-dioxane (1.5 mL) were placed in an oven-dried 10 mL Schlenk flask containing a
24
25 stirring bar. After taken out of the glove box, the mixture was heated with an oil bath whose
26
27 temperature was maintained at 120 °C for 5 min, and then cooled to room temperature. In the
28
29 glove box, aryl triflates **1** (0.3 mmol) and pyrazole derivatives **2** (0.36 mmol) were added to the
30
31 premixed solution, and the reaction mixture was stirred at 70 °C for 24 h. After this period, the
32
33 resulting mixture was diluted with dichloromethane, passed through a pad of Celite, and
34
35 concentrated in vacuo. Silica gel column chromatography of the crude material followed by
36
37 drying in vacuo afforded *N*-arylpirazoles **3**.
38
39
40
41

42 *1-(4-Methylphenyl)-3-(trimethylsilyl)-1H-pyrazole (3aa)*. General Procedure A was followed
43
44 with **1a** (72.9 mg, 0.303 mmol) and **2a** (50.0 mg, 0.356 mmol). Silica gel column
45
46 chromatography (hexane:AcOEt = 96:4) afforded 68.1 mg of **3aa** (0.296 mmol, 98% yield) as a
47
48 white solid: $\text{Mp} = 32\text{-}34$ °C; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.30 (s, 9H), 2.36 (s, 3H), 6.62
49
50 (d, $J = 2.7$ Hz, 1H), 7.28-7.30 (m, 2H), 7.72-7.76 (m, 2H), 8.26 (d, $J = 2.4$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$
51
52 NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$): δ -0.9, 20.8, 114.1, 119.6, 127.7, 130.6, 136.5, 139.1, 154.6.; IR
53
54
55
56
57
58
59
60

(KBr): 3143 m, 3116 w, 3052 m, 2956 s, 2898 s, 1906 w, 1707 w, 1609 m, 1594 m, 1523 s, 1488 s, 1457 m, 1408 s, 1381 w, 1354 s, 1308 s, 1292 , 1248 s, 1213 m, 1192 s, 1120 m, 1110 s, 1067 m, 1039 s, 971 s, 946 m, 840 s, 813 s, 754 s cm^{-1} ; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}^+]$ Calcd for $\text{C}_{13}\text{H}_{19}\text{N}_2\text{Si}^+$ 231.1312; Found 231.1310.

3-tert-Butyl-1-(4-methylphenyl)-1H-pyrazole (3ab). General Procedure A was followed with **1a** (71.5 mg, 0.298 mmol) and **2b** (43.0 mg, 0.346 mmol). Silica gel column chromatography (hexane:AcOEt = 95:5) afforded 63.8 mg of **3ab** (0.298 mmol, quant) as a white solid: Mp = 34.5-36 °C; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 1.33 (s, 9H), 2.34 (s, 3H), 6.37 (d, $J = 2.4$ Hz, 1H), 7.25-7.28 (m, 2H), 7.67-7.70 (m, 2H), 8.12 (d, $J = 2.7$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$) : δ 20.8, 30.8, 32.9, 104.7, 118.9, 127.7, 130.6, 135.9, 139.2, 163.6; IR (KBr): 3149 m, 3128 w, 3044 m, 2959 s, 2865 s, 1902 w, 1701 w, 1608 s, 1531 s, 1483 s, 1457 s, 1391 s, 1381 s, 1365 s, 1342 m, 1317 m, 1270 s, 1206 m, 1184 m, 1171 s, 1120 m, 1110 m, 1073 m, 1046 s, 982 s, 949 s, 836 s, 812 s, 755 s, 723 m cm^{-1} ; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}^+]$ Calcd for $\text{C}_{14}\text{H}_{19}\text{N}_2^+$ 215.1543; Found 215.1544.

3-Phenyl-1-(4-methylphenyl)-1H-pyrazole (3ac).²¹ General Procedure A was followed with **1a** (72.6 mg, 0.302 mmol) and **2c** (52.1 mg, 0.361 mmol). Silica gel column chromatography (hexane:AcOEt = 97:3) afforded 68.3 mg of **3ac** (0.292 mmol, 97% yield) as a white solid; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 2.38 (s, 3H), 6.94 (d, $J = 2.4$ Hz, 1H), 7.32-7.36 (m, 3H), 7.42-7.46 (m, 2H), 7.79-7.82 (m, 2H), 7.95-7.98 (m, 2H), 8.34 (d, $J = 2.4$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$) : δ 20.8, 105.5, 119.3, 126.4, 128.7, 129.2, 129.5, 130.7, 134.3, 136.7, 139.0, 153.0.

3-Methyl-1-(4-methylphenyl)-1H-pyrazole (3ad).²² General Procedure A was followed with **1a** (72.5 mg, 0.302 mmol) and **2d** (30.2 mg, 0.368 mmol). Silica gel column chromatography

(hexane:AcOEt = 97:3) afforded 46.6 mg of **3ad** (0.271 mmol, 90% yield) as a white solid; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 2.28 (s, 3H), 2.34 (s, 3H), 6.26 (d, $J = 2.4$ Hz, 1H), 7.24-7.28 (m, 2H), 7.65-7.68 (m, 2H), 8.12 (d, $J = 2.0$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$): δ 13.7, 20.8, 107.9, 118.9, 128.1, 130.6, 136.0, 139.1, 150.4.

1-(4-Methylphenyl)-1H-pyrazole (3ae).^{4b} General Procedure A was followed with **1a** (73.0 mg, 0.304 mmol) and **2e** (24.7 mg, 0.363 mmol). Silica gel column chromatography (hexane:AcOEt = 96:4) afforded 40.2 mg of **3ae** (0.254 mmol, 84% yield) as a pale yellow oil; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 2.36 (s, 3H), 6.48 (dd, $J = 2.5$ Hz, 1.8 Hz, 1H), 7.28-7.31 (m, 2H), 7.66 (d, $J = 1.6$ Hz, 1H), 7.70-7.74 (m, 2H), 8.26 (dd, $J = 2.5$ Hz, 0.6 Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$): δ 20.8, 108.1, 119.4, 127.6, 130.7, 136.6, 139.1, 141.3.

1-(4-Methylphenyl)-4-(trimethylsilyl)-1H-pyrazole (3af). General Procedure A was followed with **1a** (72.3 mg, 0.301 mmol) and **2f** (51.5 mg, 0.367 mmol). Silica gel column chromatography (hexane:AcOEt = 97:3) afforded 47.9 mg of **3af** (0.208 mmol, 69% yield) as a white solid: Mp = 51-53 °C; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.26 (s, 9H), 2.36 (s, 3H), 7.29 (d, $J = 8.2$ Hz, 2H), 7.67 (s, 1H), 7.72-7.75 (m, 2H), 8.26 (s, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$): δ -0.2, 20.8, 117.2, 119.5, 130.7, 132.3, 136.6, 138.9, 145.8; IR (KBr): 3112 w, 2954 s, 2897 m, 1527 s, 1424 w, 1400 m, 1359 m, 1326 m, 1299 w, 1249 s, 1214w, 1199 m, 1183 w, 1156 s, 1119 w, 1107 w, 1038 m, 986 w, 960 m, 833 s, 814 s, 753 s cm^{-1} ; HRMS (ESI-TOF) m/z: $[\text{M}+\text{H}^+]$ Calcd for $\text{C}_{13}\text{H}_{19}\text{N}_2\text{Si}^+$ 231.1312; Found 231.1313.

1-(2-Methylphenyl)-3-(trimethylsilyl)-1H-pyrazole (3ba). General Procedure A was followed with **1b** (73.0 mg, 0.304 mmol) and **2a** (50.6 mg, 0.361 mmol). Silica gel column chromatography (hexane:AcOEt = 98:2 to 97:3) afforded 66.4 mg of **3ba** (0.288 mmol, 95% yield) as a colorless oil; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.29 (s, 9H), 2.24 (s, 3H), 6.60 (d, J

1
2
3 NMR (400 MHz, (CD₃)₂CO): δ 2.26 (s, 3H), 2.27 (s, 3H), 6.24 (d, *J* = 2.4 Hz, 1H), 7.27-7.36 (m,
4 4H), 7.72 (d, *J* = 2.4 Hz, 1H); ¹³C{¹H} NMR (100 MHz, (CD₃)₂CO) : δ 13.7, 18.5, 106.6, 126.5,
5
6 127.3, 128.5, 132.07, 132.11, 133.9, 141.3, 149.7.
7
8
9

10 *1-(2,4,5-Trimethylphenyl)-3-(trimethylsilyl)-1H-pyrazole (3ca)*. General Procedure A was
11 followed with **1c** (82.0 mg, 0.306 mmol) and **2a** (51.2 mg, 0.365 mmol). Silica gel column
12 chromatography (hexane:AcOEt = 98:2 to 97:3) afforded 73.8 mg of **3ca** (0.286 mmol, 93%
13 yield) as a colorless oil; ¹H NMR (400 MHz, (CD₃)₂CO): δ 0.28 (s, 9H), 1.98 (s, 3H), 2.30 (s,
14 6H), 6.58 (d, *J* = 2.4 Hz, 1H), 6.97 (s, 1H), 7.08 (s, 1H), 7.77 (d, *J* = 2.0 Hz, 1H); ¹³C{¹H} NMR
15 (100 MHz, (CD₃)₂CO) : δ -0.8, 14.1, 20.3, 20.7, 112.6, 125.2, 130.1, 131.2, 131.7, 136.1, 138.8,
16 141.3, 153.4; IR (neat): 2956 m, 2921 w, 2898 w, 2864 w, 1617 w, 1580 w, 1494 m, 1452 w,
17 1385 w, 1354 w, 1302 w, 1248 m, 1157 m, 1124 w, 1050 w, 976 w, 940 w, 842 s, 755 m, 718 w,
18 698 w, 632 w cm⁻¹; HRMS (ESI-TOF) *m/z*: [M+H⁺] Calcd for C₁₅H₂₃N₂Si⁺ 259.1625; Found
19 259.1626.
20
21
22
23
24
25
26
27
28
29
30
31
32

33 *1-[4-Methyl-2-(3-methylbut-2-en-1-yl)phenyl]-3-(trimethylsilyl)-1H-pyrazole (3da)*. General
34 Procedure A was followed with **1d** (95.7 mg, 0.310 mmol) and **2a** (50.9 mg, 0.363 mmol). Silica
35 gel column chromatography (hexane:AcOEt = 97:3) afforded 84.2 mg of **3da** (0.282 mmol, 91%
36 yield) as a colorless oil; ¹H NMR (400 MHz, (CD₃)₂CO): δ 0.28 (s, 9H), 1.52 (s, 3H), 1.62 (d, *J*
37 = 1.2 Hz, 3H), 2.36 (s, 3H), 3.25 (d, *J* = 7.4 Hz, 2H), 5.08-5.13 (m, 1H), 6.58 (d, *J* = 2.4 Hz, 1H),
38 7.11-7.13 (m, 1H), 7.17-7.19 (m, 2H), 7.78 (d, *J* = 2.4 Hz, 1H); ¹³C{¹H} NMR (100 MHz,
39 (CD₃)₂CO) : δ -0.8, 17.8, 21.1, 25.8, 30.7, 112.7, 123.7, 126.8, 127.9, 131.6, 131.8, 132.7, 138.0,
40 138.5, 138.9, 153.7; IR (neat): 3026 w, 2958 s, 2924 s, 2857 m, 1611 w, 1510 s, 1439 m, 1376 w,
41 1353 w, 1304 m, 1248 s, 1184 s, 1154 w, 1122 w, 1099 w, 1058 m, 1035 m, 971 s, 913 w, 843 s,
42 755 s cm⁻¹; HRMS (ESI-TOF) *m/z*: [M+H⁺] Calcd for C₁₈H₂₇N₂Si⁺ 299.1938; Found 299.1938.
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 *1-(2-Methoxyphenyl)-3-(trimethylsilyl)-1H-pyrazole (3ea)*. General Procedure A was followed
4 with **1e** (76.0 mg, 0.297 mmol) and **2a** (50.3 mg, 0.359 mmol). Silica gel column
5 chromatography (hexane:AcOEt = 97:3) afforded 72.3 mg of **3ea** (0.293 mmol, 99% yield) as a
6 colorless oil; ¹H NMR (400 MHz, (CD₃)₂CO): δ 0.29 (s, 9H), 3.93 (s, 3H), 6.58 (d, *J* = 2.4 Hz,
7 1H), 7.05-7.10 (m, 1H), 7.22 (dd, *J* = 8.4 Hz, 1.4 Hz, 1H), 7.30-7.35 (m, 1H), 7.79 (dd, *J* = 7.8
8 Hz, 1.6 Hz, 1H), 8.21 (d, *J* = 2.7 Hz, 1H); ¹³C{¹H} NMR (100 MHz, (CD₃)₂CO) : δ -0.9, 56.3,
9 112.9, 113.4, 121.7, 125.5, 128.5, 130.7, 132.2, 151.6, 153.4; IR (neat): 2956 m, 2898 w, 2839 w,
10 1598 w, 1509 s, 1484 w, 1470 m, 1420 w, 1352 w, 1321 w, 1301 m, 1283 m, 1246 s, 1184 m,
11 1161 w, 1124 m, 1049 m, 1027 m, 970 m, 948 w, 842 s, 794 w, 753 s, 698 w, 672 w, 632 w cm⁻¹;
12 HRMS (ESI-TOF) m/z: [M+Na⁺] Calcd for C₁₃H₁₈N₂NaOSi⁺ 269.1081; Found 269.1081.
13
14
15
16
17
18
19
20
21
22
23
24
25

26 *1-[2-(Trifluoromethyl)phenyl]-3-(trimethylsilyl)-1H-pyrazole (3fa)*. General Procedure A was
27 followed with **1f** (89.2 mg, 0.303 mmol) and **2a** (50.0 mg, 0.356 mmol). Silica gel column
28 chromatography (hexane:AcOEt = 96:4) afforded 69.3 mg of **3fa** (0.244 mmol, 81% yield) as a
29 pale yellow oil; ¹H NMR (400 MHz, (CD₃)₂CO): δ 0.29 (s, 9H), 6.63 (d, *J* = 2.4 Hz, 1H), 7.60-
30 7.62 (m, 1H), 7.68-7.73 (m, 1H), 7.80-7.85 (m, 1H), 7.90-7.93 (m, 2H); ¹³C{¹H} NMR (100
31 MHz, (CD₃)₂CO) : δ -1.0, 113.5, 124.3 (q, *J* = 272.8 Hz), 126.0 (q, *J* = 31.3 Hz), 128.1 (q, *J* =
32 5.3 Hz), 129.5, 129.7, 132.4 (q, *J* = 1.9 Hz), 134.1, 140.0, 154.9; IR (neat): 2958 m, 2899 w,
33 1609 m, 1588 m, 1509 s, 1464 s, 1422 w, 1355 w, 1316 s, 1273 m, 1250 s, 1187, 1136 s, 1113 s,
34 1077 m, 1054 s, 1039 s, 970 m, 843 757 s, 700 m, 677 w, 647 w, 632 m; HRMS (ESI-TOF) m/z:
35 [M+H⁺] Calcd for C₁₃H₁₆F₃N₂Si⁺ 285.1029; Found 285.1030.
36
37
38
39
40
41
42
43
44
45
46
47
48

49 *2-[3-(Trimethylsilyl)-1H-pyrazol-1-yl]benzotrile (3ga)*. General Procedure A was followed
50 with **1g** (76.4 mg, 0.304 mmol) and **2a** (49.7 mg, 0.354 mmol). Silica gel column
51 chromatography (hexane:AcOEt = 95:5) afforded 68.4 mg of **3ga** (0.283 mmol, 93% yield) as a
52
53
54
55
56
57
58
59
60

1
2
3 colorless oil; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.33 (s, 9H), 6.73 (d, $J = 2.4$ Hz, 1H), 7.55-7.59
4 (m, 1H), 7.82-7.88 (m, 2H), 7.92-7.94 (m, 1H), 8.33 (d, $J = 2.4$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100
5 MHz, $(\text{CD}_3)_2\text{CO}$) : δ -1.1, 106.6, 114.7, 117.7, 124.7, 128.2, 130.4, 134.9, 135.6, 142.7, 156.4;
6
7 IR (neat): 3112 w, 2957 m, 2898 w, 1601 m, 1579 m, 507 s, 1487 m, 1461 m, 1417 w, 1318 s,
8
9 1299 m, 1250 s, 1185 m, 1164 w, 1113 w, 1067 w, 1044 m, 1023 m, 968 s, 946 w, 844 s, 758 s,
10
11 700 w, 667 m, 633 m cm^{-1} ; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}^+]$ Calcd for $\text{C}_{13}\text{H}_{15}\text{N}_3\text{NaSi}^+$
12
13 264.0928; Found 264.0927.
14
15
16
17
18

19 *1-(Naphthalen-1-yl)-3-(trimethylsilyl)-1H-pyrazole (3ha)*. General Procedure A was followed
20 with **1h** (81.5 mg, 0.295 mmol) and **2a** (50.1 mg, 0.357 mmol). Silica gel column
21 chromatography (hexane:AcOEt = 96:4) afforded 69.6 mg of **3ha** (0.261 mmol, 88% yield) as a
22 colorless oil; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.33 (s, 9H), 6.72 (d, $J = 2.0$ Hz, 1H), 7.53-7.64
23 (m, 4H), 7.88-7.91 (m, 1H), 8.02-8.05 (m, 2H), 8.06 (d, $J = 2.4$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100
24 MHz, $(\text{CD}_3)_2\text{CO}$) : δ -0.8, 113.3, 123.6, 124.4, 126.2, 127.4, 127.8, 129.0, 129.4, 129.9, 132.7,
25 135.4, 138.6, 154.6; IR (neat): 3055 w, 2956 m, 2897 m, 1560 w, 1577 w, 1513 m, 1491 m, 1466
26 w, 1428 m, 1395 m, 1301 m, 1249 s, 1196 w, 1171 w, 1120 m, 1020 w, 1003 w, 968 m, 934 w,
27 842 s, 800 s, 771 s, 757 s, 698 w, 663 w, 632 m cm^{-1} ; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}^+]$ Calcd for
28 $\text{C}_{16}\text{H}_{18}\text{N}_2\text{NaSi}^+$ 289.1132; Found 289.1130.
29
30
31
32
33
34
35
36
37
38
39
40
41

42 *1-(4-Chlorophenyl)-3-(trimethylsilyl)-1H-pyrazole (3ia)*. General Procedure A was followed
43 with **1i** (78.3 mg, 0.300 mmol) and **2a** (50.1 mg, 0.357 mmol). Silica gel column
44 chromatography (hexane:AcOEt = 98:2) afforded 55.7 mg of **3ia** (0.222 mmol, 74% yield) as a
45 colorless oil; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.30 (s, 9H), 6.66 (d, $J = 2.7$ Hz, 1H), 7.50-7.54
46 (m, 2H), 7.89-7.92 (m, 2H), 8.35 (d, $J = 2.4$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$): δ -
47 1.0, 114.7, 121.0, 128.0, 130.2, 131.7, 140.0, 155.6; IR (neat): 2957 m, 2898 w, 1598 m, 1505 s,
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 1433 w, 1407 w, 1356 w, 1314 m, 303 m, 1249 s, 1220 w, 1191 s, 1095 m, 1067 w, 1036 m,
4
5 1011 w, 969 s, 947 m, 842 s, 753 s, 699 w, 633 m cm^{-1} ; HRMS (ESI-TOF) m/z : $[M+H^+]$ Calcd
6
7 for $\text{C}_{12}\text{H}_{16}\text{ClN}_2\text{Si}^+$ 251.0766; Found 251.0765.

8
9
10 *1-(4-[3-(Trimethylsilyl)-1H-pyrazol-1-yl]phenyl)ethan-1-one (3ja)*. General Procedure A was
11 followed with **1j** (79.6 mg, 0.297 mmol) and **2a** (51.2 mg, 0.365 mmol). Silica gel column
12 chromatography (hexane:AcOEt = 95:5 to 90:10) afforded 70.6 mg of **3ja** (0.273 mmol, 92%
13 yield) as a colorless oil; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.32 (s, 9H), 2.61 (s, 3H), 6.71 (d, J
14 = 2.4 Hz, 1H), 8.01-8.04 (m, 2H), 8.10-8.14 (m, 2H), 8.47 (d, J = 2.4 Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR
15 (100 MHz, $(\text{CD}_3)_2\text{CO}$) : δ -1.1, 26.6, 115.1, 119.0, 128.3, 130.7, 135.5, 144.2, 156.4, 196.7; IR
16 (neat): 3135 w, 3114 w, 3003 w, 297 m, 2898 w, 1683 s, 1604 s, 1519 s, 1493 w, 1435 s, 1411 s,
17 1360 s, 1308 s, 1266 s, 1193 s, 1177 m, 1114 w, 1065 w, 1034 m, 967 s, 944 m, 842 s, 757 s,
18 724 w cm^{-1} ; HRMS (ESI-TOF) m/z : $[M+\text{Na}^+]$ Calcd for $\text{C}_{14}\text{H}_{18}\text{N}_2\text{NaOSi}^+$ 281.1081; Found
19 281.1082.
20
21
22
23
24
25
26
27
28
29
30
31

32
33 *1-(4-Methoxyphenyl)-3-(trimethylsilyl)-1H-pyrazole (3ka)*. General Procedure A was followed
34 with **1k** (76.0 mg, 0.297 mmol) and **2a** (50.0 mg, 0.356 mmol). Silica gel column
35 chromatography (hexane:AcOEt = 97:3) afforded 70.9 mg of **3ka** (0.288 mmol, 97% yield) as a
36 white solid: Mp = 39-41 $^\circ\text{C}$; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{CO}$): δ 0.29 (s, 9H), 3.84 (s, 3H), 6.60
37 (d, J = 2.0 Hz, 1H), 7.02-7.06 (m, 2H), 7.74-7.78 (m, 2H), 8.20 (d, J = 2.4 Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$
38 NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$) : δ -0.9, 55.8, 114.0, 115.2, 121.1, 127.7, 135.0, 154.3, 159.0; IR
39 (KBr): 3114 w, 3058 m, 2959 m, 2838 m, 1935 w, 1905 w, 1871 w, 1698 w, 1667 w, 1608 m,
40 1521 s, 1446 m, 1311 s, 1296 s, 1247 s, 1183 s, 1114 m, 1068 m, 1041 s, 971 s, 950 m, 847 s,
41 755 s, 696 m, 634 s cm^{-1} ; HRMS (ESI-TOF) m/z : $[M+H^+]$ Calcd for $\text{C}_{13}\text{H}_{19}\text{N}_2\text{OSi}^+$ 247.1261;
42 Found 247.1263.
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 2-[3-(Trimethylsilyl)-1H-pyrazol-1-yl]pyridine (**3la**). General Procedure A was followed with
4
5 **1l** (69.4 mg, 0.306 mmol) and **2a** (50.3 mg, 0.359 mmol). Silica gel column chromatography
6
7 (hexane:AcOEt = 97:3) afforded 61.4 mg of **3la** (0.282 mmol, 92% yield) as a colorless oil; ¹H
8
9 NMR (400 MHz, (CD₃)₂CO): δ 0.32 (s, 9H), 6.65 (d, *J* = 2.7 Hz, 1H), 7.30 (ddd, *J* = 7.2 Hz, 4.9
10
11 Hz, 1.2 Hz, 1H), 7.96 (ddd, *J* = 8.3 Hz, 7.3 Hz, 1.9 Hz, 1H), 8.06 (dt, *J* = 8.4 Hz, 0.9 Hz, 1H),
12
13 8.44 (ddd, *J* = 4.8 Hz, 1.7 Hz, 0.9 Hz, 1H), 8.61 (d, *J* = 2.4 Hz, 1H).; ¹³C{¹H} NMR (100 MHz,
14
15 (CD₃)₂CO) : δ -1.1, 113.1, 114.3, 122.4, 127.5, 139.7, 149.1, 152.6, 156.7; IR (neat): 3065 w,
16
17 3017 w, 2957 s, 2898 w, 1614 m, 1595 s, 1578 s, 1497 s, 1471 s, 1457 s, 1410 w, 1310 s, 1249 s,
18
19 1223 w, 1197 s, 1145 m, 1099 w, 1048 s, 1035 m, 992 w, 966 s, 952 m, 842 s, 777 s, 759 s, 739
20
21 w, 717 m cm⁻¹; HRMS (ESI-TOF) *m/z*: [M+H⁺] Calcd for C₁₁H₁₆N₃Si⁺ 218.1108; Found
22
23 218.1110.

24
25
26
27
28 3-[3-(Trimethylsilyl)-1H-pyrazol-1-yl]pyridine (**3ma**). General Procedure A was followed
29
30 with **1m** (81.2 mg, 0.357 mmol) and **2a** (41.8 mg, 0.298 mmol). Silica gel column
31
32 chromatography (hexane:AcOEt = 92:8 to 80:20) afforded 61.4 mg of **3ma** (0.282 mmol, 95%
33
34 yield) as a colorless oil; ¹H NMR (400 MHz, (CD₃)₂CO): δ 0.32 (s, 9H), 6.71 (d, *J* = 2.4 Hz, 1H),
35
36 7.51 (ddd, *J* = 8.4 Hz, 4.7 Hz, 0.6 Hz, 1H), 8.24 (ddd, *J* = 8.4 Hz, 2.5 Hz, 1.6 Hz, 1H), 8.43 (d, *J*
37
38 = 2.4 Hz, 1H), 8.51 (dd, *J* = 4.7 Hz, 1.6 Hz, 1H), 9.13 (d, *J* = 2.4 Hz, 1H).; ¹³C{¹H} NMR (100
39
40 MHz, (CD₃)₂CO) : δ -1.0, 114.9, 124.8, 126.7, 128.2, 137.5, 141.3, 148.1, 156.2; IR (neat): 3106
41
42 w, 3045 w, 2957 m, 2898 w, 1587 s, 1502 s, 1479 m, 1458 m, 1430 m, 1359 w, 1311 s, 1250 s,
43
44 1181 m, 1121 w, 1101 w, 1074 w, 1048 m, 1019 m, 969 s, 947 m, 843 s, 805 m, 755 s, 718 m,
45
46 704 s cm⁻¹; HRMS (ESI-TOF) *m/z*: [M+H⁺] Calcd for C₁₁H₁₆N₃Si⁺ 218.1108; Found 218.1108.

47
48
49 Deprotection of **3aa** to Form 1-(4-Methylphenyl)-1H-pyrazole (**3ae**).^{4b} A solution of **3aa** (34.6
50
51 mg, 0.150 mmol) in 20% ethanolic KOH (1.5 mL) was refluxed for 12 h under air. After this
52
53
54
55
56
57
58
59
60

1
2
3 period, water was added to the mixture, which was then extracted three times with AcOEt. The
4 combined organic layers were washed with brine, dried over MgSO₄, filtered and concentrated.
5
6 Silica gel column chromatography (hexane:AcOEt = 95:5) of the crude material afforded 21.0
7
8 mg (0.133 mmol, 89% yield) of **3ae** as a pale yellow oil.
9
10

11
12 *Desilylative Chlorination to Form 3-Chloro-1-(4-methylphenyl)-1H-pyrazole (4).*²⁴ In a glove
13 box, **3aa** (23.2 mg, 0.101 mmol), *N*-chlorosuccinimide (40.5 mg, 0.303 mmol), AgF (37.9 mg,
14 0.299 mmol), and 0.2 mL of MeCN were placed in an oven-dried sealed tube containing
15 magnetic stirring bar. The mixture was stirred in the dark at room temperature for 20 h. After this
16 period, the resulting mixture was diluted with dichloromethane, passed through a pad of Celite,
17 and concentrated in vacuo. Silica gel column chromatography (hexane:AcOEt = 99:1) of the
18 crude material afforded 16.8 mg (0.0872 mmol, 86% yield) of **4** as a white solid: Mp = 77-79
19 °C ; ¹H NMR (400 MHz, (CD₃)₂CO): δ 2.37 (s, 3H), 6.50 (d, *J* = 2.7 Hz, 1H), 7.32 (d, *J* = 7.8 Hz,
20 2H), 7.67 (d, *J* = 8.6 Hz, 2H), 8.31 (d, *J* = 2.7 Hz, 1H); ¹³C{¹H} NMR (100 MHz, (CD₃)₂CO) : δ
21 20.8, 107.6, 119.3, 130.2, 130.8, 137.4, 138.4, 141.4; IR (KBr): 3164 m, 3119 m, 3049 m, 2920
22 m, 2855 w, 1902 w, 1720 w, 1607 m, 1518 s, 1499 s, 1431 m, 1408 s, 1369 s, 1350 s, 1278 w,
23 1216 m, 1200 s, 1110 m, 1046 s, 970 m, 942 s, 862 w, 817 s, 757 s cm⁻¹; HRMS (ESI-TOF) m/z:
24 [M+H⁺] Calcd for C₁₀H₁₀ClN₂⁺ 193.0527; Found 193.0527.
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41

42 *Desilylative Bromination to Form 3-Bromo-1-(4-methylphenyl)-1H-pyrazole (5).*²⁴ In a glove
43 box, **3aa** (23.1 mg, 0.100 mmol), 1,2-dibromo-1,1,2,2-tetrachloroethane (163 mg, 0.501 mmol),
44 AgF (63.3 mg, 0.499 mmol), and 0.2 mL of MeCN were placed in an oven-dried sealed tube
45 containing magnetic stirring bar. The mixture was stirred in the dark at 40 °C for 20 h. After this
46 period, the resulting mixture was diluted with dichloromethane, passed through a pad of Celite,
47 and concentrated in vacuo. Silica gel column chromatography (hexane:AcOEt = 98:2) of the
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 crude material afforded 10.7 mg (0.0451 mmol, 45% yield) of **5** as a white solid: Mp = 69-71
4 °C ; ¹H NMR (400 MHz, (CD₃)₂CO): δ 2.37 (s, 3H), 6.58 (d, *J* = 2.4 Hz, 1H), 7.32 (d, *J* = 8.6 Hz,
5 2H), 7.68 (d, *J* = 8.6 Hz, 2H), 8.27 (d, *J* = 2.7 Hz, 1H); ¹³C{¹H} NMR (100 MHz, (CD₃)₂CO) : δ
6 20.8, 111.0, 119.3, 128.1, 130.3, 130.8, 137.5, 138.4; IR (KBr): 3154 w, 3127 w, 2919 m, 2854
7 w, 1610 w, 1594 w, 1524 s, 1495 m, 1407 m, 1364 s, 1342 s, 1266 w, 1214 w, 1193 m, 1119 w,
8 106 w, 1045 s, 956 s, 939 m, 849 w, 835 w, 809 s, 749 s cm⁻¹; HRMS (ESI-TOF) m/z: [M+H⁺]
9 Calcd for C₁₀H₁₀BrN₂⁺ 237.0022; Found 237.0021.

10
11
12 *4-Bromination of 3aa Followed by Deprotection to Form 4-Bromo-1-(4-methylphenyl)-1H-*
13 *pyrazole (6).* In a glove box, **3aa** (22.9 mg, 0.0994 mmol), *N*-bromosuccinimide (19.8 mg, 0.111
14 mmol), and 0.2 mL of MeCN were placed in an oven-dried sealed tube containing magnetic
15 stirring bar. The mixture was stirred at room temperature for 1 h, and the resulting mixture was
16 concentrated in vacuo. Then a solution of the crude material solution in 20% ethanolic KOH (1.5
17 mL) was refluxed for 8 h under air. After this period, water was added to the mixture, which was
18 then extracted three times with AcOEt. The combined organic layers were washed with brine,
19 dried over MgSO₄, filtered and concentrated. Silica gel column chromatography (hexane:AcOEt
20 = 97:3) of the crude material afforded 20.6 mg (0.0869 mmol, 87% yield) of **6** as a white solid:
21 Mp = 88-90 °C ; ¹H NMR (400 MHz, (CD₃)₂CO): δ 2.37 (s, 3H), 7.32 (d, *J* = 8.2 Hz, 2H), 7.70-
22 7.72 (m, 3H), 8.44 (s, 1H); ¹³C{¹H} NMR (100 MHz, (CD₃)₂CO) : δ 20.8, 95.7, 119.4, 128.3,
23 130.8, 137.5, 138.5, 141.7; IR (KBr): 3112 s, 3044 m, 2918 m, 1697 w, 1608 w, 1525 s, 1509 s,
24 1426 m, 1398 s, 1380 s, 1335 s, 1244 m, 1214 w, 1193 m, 1150 m, 1109 m, 1033 s, 950 s, 850 s,
25 816 s cm⁻¹; HRMS (ESI-TOF) m/z: [M+H⁺] Calcd for C₁₀H₁₀BrN₂⁺ 237.0022; Found 237.0024.

26
27
28 *5-Bromination of 3aa Followed by Deprotection to Form 5-Bromo-1-(4-methylphenyl)-1H-*
29 *pyrazole (7).*²⁵ To a solution of **3aa** (23.4 mg, 0.102 mmol) in 1 mL of THF at -40 °C was added
30
31
32

1
2
3 ⁿBuLi in hexane (1.50 M, 0.150 mmol) dropwise and the mixture was stirred for 2 h. The
4
5 resulting mixture was then cooled to -60 °C, and tetrabromomethane in THF solution was added
6
7 dropwise (1.00 M, 0.250 mmol) to the mixture, which was stirred for 1 h at this temperature and
8
9 for another 1 h at room temperature. After this period, water was added to the reaction mixture,
10
11 which was then extracted with three times with Et₂O. The combined organic layers were washed
12
13 with brine, dried over MgSO₄, filtered, concentrated and purified by silica gel column
14
15 chromatography (hexane:AcOEt = 98:2) to give a mixture of **3aa** and the 5-bromination product.
16
17 To a round bottom flask charged with the obtained mixture was added KHF₂ (39.8 mg, 0.510
18
19 mmol) and MeOH 4 mL, and the mixture was stirred for 3 h at 70 °C. After this period, water
20
21 was added to the resulting mixture, which was extracted three times with AcOEt. The combined
22
23 organic layers were washed with brine, dried over MgSO₄, filtered and concentrated. Silica gel
24
25 column chromatography (hexane:AcOEt = 97:3) of the crude material afforded a mixture of **3ae**
26
27 and **7**, and the NMR yield of **7** (0.0709 mmol, 70% NMR yield) was determined by ¹H NMR
28
29 analysis. Gel permeation chromatography (GPC) of the mixture afforded 10.8 mg of pure **7**
30
31 (0.0456 mmol, 45% yield) as a pale yellow oil: ¹H NMR (400 MHz, (CD₃)₂CO): δ 2.42 (s, 3H),
32
33 6.58 (d, *J* = 2.0 Hz, 1H), 7.35-7.37 (m, 2H), 7.42-7.45 (m, 2H), 7.68 (d, *J* = 2.0 Hz, 1H); ¹³C{¹H}
34
35 NMR (100 MHz, (CD₃)₂CO): δ 21.1, 111.0, 113.0, 126.3, 130.3, 137.7, 139.4, 142.0; IR (neat):
36
37 3040 w, 2922 w, 2855 w, 1518 s, 1407 m, 1391 s, 1237 w, 1178 w, 1111 w, 1040 w, 1020 w,
38
39 967 m, 917 m, 872 w, 820 m, 799 w, 772 m, 711 w cm⁻¹; HRMS (ESI-TOF) *m/z*: [M+H⁺] Calcd
40
41 for C₁₀H₁₀BrN₂⁺ 237.0022; Found 237.0026.
42
43
44
45
46
47
48
49
50

51 Supporting Information

52
53
54
55
56
57
58
59
60

1
2
3 The Supporting Information is available free of charge on the ACS Publications website at
4
5 <http://pubs.acs.org>.

6
7 Copies of ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra (PDF)

11 12 **Corresponding Author**

13
14 *E-mail: kakiuchi@chem.keio.ac.jp

15 16 **ORCID**

17
18 Takuya Kochi: 0000-0002-5491-0566

19
20 Fumitoshi Kakiuchi: 0000-0003-2605-4675

21 22 **Notes**

23
24 The authors declare no competing financial interest.

25 26 27 28 29 30 **Acknowledgements**

31
32 This work was supported in part by JSPS KAKENHI Grant Numbers JP17K19126,
33
34 JP16H04150 and JP15H05839 (Middle Molecular Strategy). T. K. is also grateful for support by
35
36 JSPS KAKENHI Grant Number JP16H01040 (Precisely Designed Catalysts with Customized
37
38 Scaffolding). S. O. is also grateful for support by the Research Grant of Keio Leading-edge
39
40 Laboratory of Science & Technology.

41 42 43 44 45 46 **References and Notes**

47
48 (1) (a) Ansari, A.; Ali, A.; Asif, M.; Shamsuzzaman. Review: Biologically Active Pyrazole
49
50 Derivatives. *New. J. Chem.* **2017**, *41*, 16-41. (b) Faria, J. V.; Vegi, P. F.; Miguita, A. G. C.; dos
51
52 Santos, M. S.; Boechat, N.; Bernardino, A. M. R. Recently Reported Biological Activities of
53
54

1
2
3 Pyrazole Compounds. *Bioorg. Med. Chem.* **2017**, *25*, 5891-5903. (c) Devendar, P.; Qu, R.-Y.;
4 Kang, W.-M.; He, B.; Yang, G.-F. Palladium-Catalyzed Cross-Coupling Reactions: A Powerful
5
6 Tool for the Synthesis of Agrochemicals. *J. Agric. Food Chem.* **2018**, *66*, 8914-8934.
7
8

9
10 (2) (a) Fustero, S.; Sánchez-Roselló, M.; Barrio, P.; Simón-Fuentes, A. From 2000 to Mid-
11
12 2010: A Fruitful Decade for the Synthesis of Pyrazoles. *Chem. Rev.* **2011**, *111*, 6984-7034.
13

14
15 (3) For recent examples: (a) Romero, N. A.; Margrey, K. A.; Tay, N. E.; Nicewicz, D. A. Site-
16
17 Selective Arene C–H Amination via Photoredox Catalysis. *Science* **2015**, *349*, 1326-1330. (b)
18
19 Gonda, Z.; Novák, Z. Transition-Metal-Free *N*-Arylation of Pyrazoles with Diaryliodonium Salts.
20
21 *Chem. Eur. J.* **2015**, *21*, 16801-16806. (c) Muzalevskiy, V. M.; Rulev, A. Y.; Romanov, A. R.;
22
23 Kondrashov, E. V.; Ushakov, I. A.; Chertkov, V. A.; Nenajdenko, V. G. Selective, Metal-Free
24
25 Approach to 3- or 5-CF₃-Pyrazoles: Solvent Switchable Reaction of CF₃-Ynones with
26
27 Hydrazines. *J. Org. Chem.* **2017**, *82*, 7200-7214. (d) Tian, M.; Shi, X.; Zhang, X.; Fan, X.
28
29 Synthesis of 4-Acylpyrazoles from Saturated Ketones and Hydrazones Featured with Multiple
30
31 C(sp³)–H Bond Functionalization and C–C Bond Cleavage and Reorganization. *J. Org. Chem.*
32
33 **2017**, *82*, 7363-7372. (e) Niu, L.; Yi, H.; Wang, S.; Liu, T.; Liu, J.; Lei, A. Photo-Induced
34
35 Oxidant-Free Oxidative C–H/N–H Cross-Coupling Between Arenes and Azoles. *Nat. Commun.*
36
37 **2017**, *8*, 14226. (f) Shao, Y.; Zheng, H.; Qian, J.; Wan, X. In Situ Generation of Nitrilimines
38
39 from Aryldiazonium Salts and Diazo Esters: Synthesis of Fully Substituted Pyrazoles under
40
41 Room Temperature. *Org. Lett.* **2018**, *20*, 2412-2415. (g) You, G.; Wang, K.; Wang, X.; Wang,
42
43 G.; Sun, J.; Duan, G.; Xia, C. Visible-Light-Mediated Nickel(II)-Catalyzed C–N Cross-Coupling
44
45 in Water: Green and Regioselective Access for the Synthesis of Pyrazole-Containing
46
47 Compounds. *Org. Lett.* **2018**, *20*, 4005-4009. (h) Thombal, R. S.; Lee, Y. R. Synergistic Indium
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 and Silver Dual Catalysis: A Regioselective [2 + 2 + 1]-Oxidative N-Annulation Approach for
4 the Diverse and Polyfunctionalized *N*-Arylpyrazoles. *Org. Lett.* **2018**, *20*, 4681-4685.

7
8 (4) For seminal examples: (a) Antilla, J. C.; Baskin, J. M.; Barder, T. E.; Buchwald, S. L.
9 Copper-Diamine-Catalyzed *N*-Arylation of Pyrroles, Pyrazoles, Indazoles, Imidazoles, and
10 Triazoles. *J. Org. Chem.* **2004**, *69*, 5578-5587. (b) Cristau, H.-J.; Cellier, P. P.; Spindler, J.-F.;
11 Taillefer, M. Mild Conditions for Copper-Catalysed *N*-Arylation of Pyrazoles. *Eur. J. Org. Chem.*
12 **2004**, 695-709. See also ref 2.

13
14
15
16
17
18
19 (5) (a) Anderson, K. W.; Tundel, R. E.; Ikawa, T.; Altman, R. A.; Buchwald, S. L.
20 Monodentate Phosphines Provide Highly Active Catalysts for Pd-Catalyzed C–N Bond-Forming
21 Reactions of Heteroaromatic Halides/Amines and (H)N-Heterocycles. *Angew. Chem., Int. Ed.*
22 **2006**, *45*, 6523-6527. (b) Fernando, D. P.; Jiao, W.; Polivkova, J.; Xiao, J.; Coffey, S. B.; Rose,
23 C.; Londregan, A.; Saenz, J.; Beveridge, R.; Zhang, Y.; Storer, G. E.; Vrieze, D.; Erasga, N.;
24 Jones, R.; Khot, V.; Cameron, K. O.; McClure, K. F.; Bhattacharya, S. K.; Orr, S. T. M.
25 Spiroazetidine-Piperidine Bromoindane as a Key Modular Template to Access a Variety of
26 Compounds via C–C and C–N Bond-Forming Reactions. *Tetrahedron Lett.* **2012**, *53*, 6351-6354.
27
28 (c) Lennartz, P.; Raabe, G.; Bolm, C. Palladium-Catalyzed C–H Bond Acetoxylation: An
29 Approach to *ortho*-Substituted Hydroxy[2.2]paracyclophane Derivatives. *Adv. Synth. Catal.*
30 **2012**, *354*, 3237-3249. (d) Veisi, H.; Heravi, M. R. P.; Hamelian, M. SBA-15-Functionalized
31 Melamine-Pyridine Group-Supported Palladium(0) as an Efficient Heterogenous and Recyclable
32 Nanocatalyst for *N*-Arylation of Indoles through Ullmann-type Coupling Reactions. *Appl.*
33 *Organometal. Chem.* **2015**, *29*, 334-337. (e) Anitha, P.; Manikandan, R.; Viswanathamurthi, P.
34 Palladium(II) 9,10-Phenanthrenequinone *N*-Substituted Thiosemicarbazone/Semicarbazone
35 Complexes as Efficient Catalysts for *N*-Arylation of Imidazole. *J. Coord. Chem.* **2015**, *68*, 3537-
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 3550. (f) Touré, B. B.; Giraldes, J.; Smith, T.; Sprague, E. R.; Wang, Y.; Mathieu, S.; Chen, Z.;
4
5 Mishina, Y.; Feng, Y.; Yan-Neale, Y.; Shakya, S.; Chen, D.; Meyer, M.; Puleo, D.; Brazell, J. T.;
6
7 Straub, C.; Sage, D.; Wright, K.; Yuan, Y.; Chen, X.; Duca, J.; Kim, S.; Tian, L.; Martin, E.;
8
9 Hurov, K.; Shao, W. Toward the Validation of Maternal Embryonic Leucine Zipper Kinase:
10
11 Discovery, Optimization of Highly Potent and Selective Inhibitors, and Preliminary Biology
12
13 Insight. *J. Med. Chem.* **2016**, *59*, 4711-4723.

14
15
16
17 (6) (a) Beletskaya, I. P.; Cheprakov, A. V. The Complementary Competitors: Palladium and
18
19 Copper in C–N Cross-Coupling Reactions. *Organometallics* **2012**, *31*, 7753-7808. (b) Ruiz-
20
21 Castillo, P.; Buchwald, S. L. Applications of Palladium-Catalyzed C–N Cross-Coupling
22
23 Reactions. *Chem. Rev.* **2016**, *116*, 12564-12649.

24
25
26 (7) Hammoud, H.; Schmitt, M.; Bihel, F.; Antheaume, C.; Bourguignon, J.-J. Direct
27
28 Guanidinylation of Aryl and Heteroaryl Halides via Copper-Catalyzed Cross-Coupling Reaction.
29
30 *J. Org. Chem.* **2012**, *77*, 417-423.

31
32
33 (8) A copper-catalyzed C–N bond forming reaction using an aryl triflates derived from a
34
35 calix[4]arene with a 1*H*-pyrazole has been reported, but the assistance of the proximal hydroxy
36
37 groups was indispensable; Rawat, V.; Press, K.; Goldberg, I.; Vigalok, A. Straightforward
38
39 Synthesis and Catalytic Applications of Rigid *N,O*-Type Calixarene Ligands. *Org. Biomol. Chem.*
40
41 **2015**, *13*, 11189-11193.

42
43
44 (9) (a) Vila, C.; Hornillos, V.; Giannerini, M.; Fañanás-Mastral, M.; Feringa, B. L. Palladium-
45
46 Catalysed Direct Cross-Coupling of Organolithium Reagents with Aryl and Vinyl Triflates.
47
48 *Chem. Eur. J.* **2014**, *20*, 13078-13083. (b) Seganish, W. M.; DeShong, P. Preparation and
49
50 Palladium-Catalyzed Cross-Coupling of Aryl Triethylammonium Bis(catechol) Silicates with
51
52 Aryl Triflates. *J. Org. Chem.* **2004**, *69*, 1137-1143.

1
2
3 (10) Amination of aryltriflates with a 4-aminopyrazole and a 4-nitropyrazole catalyzed by
4 Pd₂(dba)₃/*t*BuXPhos has been reported recently: Babu, S.; Bhattacharyya, A.; Hwang, S.; Jani,
5 M.; Moon, Y.-C.; Sydorenko, N. Preparation of Thiadiazole and Pyridazine Derivatives Useful
6 for Treating Huntington's Disease. WO 2017100726, June 15, 2017.
7
8
9

10
11 (11) Onodera, S.; Ishikawa, S.; Kochi, T.; Kakiuchi, F. Direct Alkenylation of Allylbenzenes
12 via Chelation-Assisted C–C Bond Cleavage. *J. Am. Chem. Soc.* **2018**, *140*, 9788-9792.
13
14

15 (12) When the reaction of **1a** with **2a** was carried out in the presence of equal amount of
16 *t*BuBrettPhos to Pd(dba)₂ (2 mol %), the product **3aa** was isolated in 94% yield, and the reaction
17 of **1b** with 4 mol % *t*BuBrettPhos also proceeded efficiently to give **3ba** in 91% yield. These
18 results showed that the amount of *t*BuBrettPhos could be reducible, but to obtain reproducible
19 results, we continued to use slight excess of *t*BuBrettPhos to Pd(dba)₂ in the reaction.
20
21
22
23
24
25
26
27

28 (13) Düfert M. A.; Billingsley, K. L.; Buchwald, S. L. *J. Am. Chem. Soc.* **2013**, *135*, 12877-
29 12885.
30
31
32

33 (14) Substrates possessing bis-ortho substituents such as methyl and methoxy groups were not
34 tolerated in the reaction, and only trace amounts of the desired products were obtained.
35
36

37 (15) Janin, Y. L. Preparation and Chemistry of 3/5-Halogenopyrazoles. *Chem. Rev.* **2012**, *112*,
38 3924-3958.
39
40
41

42 (16) Bertogg, A.; Hintermann, L.; Huber, D. P.; Perseghini, M.; Sanna, M.; Togni, A.
43 Substrate Range of the Titanium TADDOLate Catalyzed Asymmetric Fluorination of Activated
44 Carbonyl Compounds. *Helvetica Chimica Acta* **2012**, *95*, 353-403.
45
46
47
48

49 (17) Pejic, M.; Popp, S.; Bolte, M.; Wagner, M.; Lerner, H.-W. Functionalized Pyrazoles as
50 Agents in C–C Cross-Coupling Reactions. *Zeitschrift für Naturforschung B*, **2014**, *69*, 83-97.
51
52
53
54
55
56
57
58
59
60

1
2
3 (18) Qin, L.; Ren, X.; Lu, Y.; Li, Y.; Zhou, J. Intermolecular Mizoroki-Heck Reaction of
4 Aliphatic Olefins with High Selectivity for Substitution at the Internal Position. *Angew. Chem.,*
5
6 *Int. Ed.* **2012**, *51*, 5915-5919.

7
8
9
10 (19) Yamada, S.; Ono, F. Katagiri, T.; Tanaka, J. Selective Ortho Alkylation of Phenols in the
11 Presence of Metallic Sodium. *Nippon Kagaku Kaishi*, **1980**, 733-737.

12
13
14 (20) Ricardo, C. L.; Mo, X.; McCubbin, J. A.; Hall, D. G. A Surprising Substituent Effect
15 Provides a Superior Boronic Acid Catalyst for Mild and Metal-Free Direct Friedel-Crafts
16 Alkylations and Prenylations of Neutral Arenes. *Chem. Eur. J.* **2015**, *21*, 4218-4223.

17
18
19 (21) Panda, N.; Jena, A. K. Fe-Catalyzed One-Pot Synthesis of 1,3-Di- and 1,3,5-
20 Trisubstituted Pyrazoles from Hydrazones and Vicinal Diols. *J. Org. Chem.* **2012**, *77*, 9401-9406.

21
22 (22) Alex, K.; Tillack, A.; Schwarz, N.; Beller, M. Zinc-Catalyzed Synthesis of Pyrazolines
23 and Pyrazoles via Hydrohydrazination. *Org. Lett.* **2008**, *10*, 2377-2379.

24
25 (23) Chen, C.-Y.; Huang, Y.-Y.; Su, W.-N.; Kaneko, K.; Kimura, M.; Takayama, H.; Wong, F.
26 F. Palladium-Catalyzed Dehalogenation of 5-Halopyrazoles. *J. Heterocyclic Chem.* **2012**, *49*,
27 183-189.

28
29 (24) Kuznetsov, A.; Onishi, Y.; Inamoto, Y.; Gevorgyan, V. Fused Heteroaromatic
30 Dihydrosiloles: Synthesis and Double-Fold Modification. *Org. Lett.* **2013**, *15*, 2498-2501.

31
32 (25) Ye, C.; Gard, G. L.; Winter, R. W.; Syvret, R. G.; Twamley, B.; Shreeve, J. M. Synthesis
33 of Pentafluorosulfanylpyrazole and Pentafluorosulfanyl-1,2,3-triazole and Their Derivatives as
34 Energetic Materials by Click Chemistry. *Org. Lett.* **2007**, *9*, 3841-3844.
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60