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Letter

Synthesis of Six-Membered Spiro Azacyclic Oxindole Derivatives via a One-Pot Process of Umpolung Allylation/Aza-Prins Cyclization

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ene, H₂O, and trimethylsilyl bromide; this one-pot protocol allows access to six-membered spiro azacyclic oxindole derivatives in good to excellent yields. Notably, while the general aza-Prins cyclization involves amines and aldehydes, the present synthetic strategy

TMSBr È. • One-pot process, mild conditions, short reaction times Rapid access to spiro[indoline-3,2'-piperidin]-2-ones and 5',6'-dihydro-1'H-spiro[indoline-3,2'-pyridin]-2-ones

represents the first aza-Prins cyclization that utilizes the umpolung property of N-2,2,2-trifluoroethylisatin ketimines.

S pirooxindoles, which are prevalent in means a stractive target motifs owing to their diverse biological activities.² Among these molecules, spiro N-heterocyclic oxindoles exhibit important pharmacological activities such as IRAP inhibitory, SIRT1 inhibitory, antibreast cancer, antitumor, and antimalarial activities (Figure 1).³ In addition, the trifluoromethyl group has been known to improve the metabolic stability, bioavailability, and lipophilicity of bioactive molecules.⁴ Despite their importance, only a few synthetic methods that use N-2,2,2-trifluoroethylisatin ketimines have been developed for the construction of spiro N-heterocyclic oxindoles featuring CF₃ groups.⁵

Zhao and co-workers have accomplished the [3 + 3]cycloaddition of N-2,2,2-trifluoroethylisatin ketimines and N_iN' -dialkyloxyureas in the presence of NaH and PhI(OH)-(OTs) for the synthesis of spiro-1,3,5-triazinan-2-one oxindoles.⁶ Azomethine ylides reacted with N,N'-dialkyloxy diaza-



Figure 1. Representative examples of bioactive molecules containing the spirooxindole motif

Scheme 1. Synthetic Strategies for Spirooxindole Derivatives: (a) General Aza-Prins Reaction; (b) One-Pot Umpolung Allylation/Aza-Prins Reaction



(b) This work: Umpolung Allylation/Aza-Prins Reaction



allyl cations generated in situ in moderate yields. To the best of our knowledge, this is the only reported example of [3 + 3]cycloaddition of N-2,2,2-trifluoroethylisatin ketimines thus far. Consequently, a practical and efficient synthetic methodology for six-membered spiro azacyclic oxindole backbones remains.

The well-established aza-Prins cyclization^{7,8} is a stereoselective strategy for the synthesis of piperidines. Generally, the aza-Prins cyclization involves the condensation of homoallylic amines with aldehydes in the presence of a Lewis acid to generate iminium ions, which cyclize spontaneously through intramolecular nucleophilic attack from the olefin (Scheme 1a). Inspired by the aza-Prins cyclization, we

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^{*a*}Reaction conditions: **1a** (0.1 mmol), **2a** (3.0 equiv), and DBU (1.0 equiv) in DCE (1.0 M), 0 °C, 30 min; then H_2O (1.0 equiv), TMSBr (3.0 equiv), rt, 10 min. ^{*b*}Isolated yield. ^{*c*}Temperature of the allylation step. ^{*d*}Temperature and time of the cyclization step. ^{*e*}The ellipsoids are drawn at the 50% probability level.

envisioned that *N*-2,2,2-trifluoroethylisatin ketimine is a suitable substrate for the synthesis of six-membered spiro azacyclic oxindoles via an umpolung allylation/aza-Prins reaction (Scheme 1b). Unlike previous aza-Prins reactions, this synthetic method would take advantage of the umpolung property of *N*-2,2,2-trifluoroethylisatin ketimine.⁹ Thus, we decided to investigate the umpolung allylation/aza-Prins reaction of *N*-2,2,2-trifluoroethylisatin ketimines in a one-pot process for the preparation of spiro[indoline-3,2'-piperidin]-2-ones. Herein, we report the first efficient and facile synthetic methodology that uses the umpolung allylation/aza-Prins reaction of *N*-2,2,2-trifluoroethylisatin ketimines to produce useful spiro[indoline-3,2'-piperidin]-2-ones and 5',6'-dihydro-1'H-spiro[indoline-3,2'-pyridin]-2-ones.

Initially, we used ketimine $1a^{6,10-14}$ as the standard substrate to screen various sets of reaction conditions with allyl bromide 2a, and the results are summarized in Table 1. In the presence of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU, 1.0 equiv), 1a and 2a (3.0 equiv) were reacted in CHCl₃ at -10 °C for 30 min; then H₂O (1.0 equiv) and trimethylsilyl bromide (TMSBr, 3.0 equiv) were added, and the reaction continued at 40 °C for 10 min. Gratifyingly, desired product 3a was isolated in 89% yield; it was also obtained in 88%, 82%, and 92% yields when dichloromethane (DCM), CH₃CN, and dichloroethane (DCE), respectively, were used (entries 1 and 2). The structure of 3a



^{*a*}Reaction conditions: **1a** (0.1 mmol), **2a** (3.0 equiv), and DBU (1.0 equiv) in DCE (1.0 M), 0 °C, 30 min; then H_2O (1.0 equiv), TMSBr (3.0 equiv), rt, 10 min. ^{*b*}30 min for the cyclization step. ^{*c*}-10 °C, 30 min for the allylation step and 40 °C, 12 h for the cyclization step.

was confirmed by single-crystal X-ray diffraction analysis (CCDC 2053124). When the temperature of the cyclization step was lowered from 40 °C to room temperature or that of the allylation step was raised from -10 to 0 °C, the reaction afforded 3a in very high yields (entries 3 and 4). In contrast, when the reaction was attempted with an allylation step at room temperature or without H₂O in the cyclization step, lower yields of 3a were observed (entries 5 and 6). Interestingly, 1.0 equiv of H₂O critically affected the yield of the product. Furthermore, when 2.0 equiv of 2a or TMSBr was utilized, 3a was generated in yields of 88% and 70%, respectively (entries 7 and 8). Next, the use of various bases was examined; however, 3a was not produced (entries 9 and 10). Additionally, various reagents that trigger aza-Prins cyclization other than TMSBr were investigated (entries 11-18). In the presence of BF₃·OEt₂, 3a was provided in yield of 75%, while no desired products were formed in the presence of TMSI, TMSOAc, or InCl₃. The expected allylation/aza-Prins reaction occurred when TMSCl,

Letter





^{*a*}Reaction conditions: **1a** (0.1 mmol), **2a** (1.5 equiv), and DBU (1.0 equiv) in DCE (1.0 M), 0 °C, 30 min; then H₂O (1.0 equiv), TMSBr (3.0 equiv), 100 °C, 1 h. ^{*b*}CH₃CN was used instead of DCE. ^{*c*}**2a** (3.0 equiv) was used; 12 h for the cyclization step. ^{*d*}Allyl tosylate was used instead of allyl bromide; rt for the cyclization step. ^{*e*}rt for the cyclization step.



Scheme 4. Proposed Reaction Mechanism

TiCl₄, or FeCl₃ was used, affording mixtures of **3a** and **3a**', which were separated completely, in combined yields of 18%, 46%, and

9%, respectively. Similarly, when 3.0 equiv of TMSOTf or triethylsilyl triflate (TESOTf) was used, mixtures of **3a** and **3a**", which were isolated by column chromatography, were obtained in combined yields of 47% and 46%, respectively. Finally, the reaction was attempted using $Cu(OTf)_2$, and a trace amount of **3a**" was observed.

With the optimal reaction conditions in hand, we explored the ketimine substrate scope, and the results are shown in Scheme 2. The reactions with N-protected ketimines featuring a benzyl, allyl, or trityl group gave excellent yields of 99%, 98%, and 99%, respectively (3aa-3ac). Unprotected ketimine 1ad^{6,10,12} reacted smoothly to generate 3ad in a yield of 51%, while a mixture of 3ae and 3ad was obtained from N-tert-butyldimethylsilyl (TBS)-protected ketimine 1ae¹² in a combined yield of 61%. 5-Halogen-substituted ketimines produced the corresponding products 3b, 3c, 3d, and 3e in 91%, 73%, 83%, and 77% yields, respectively. Similarly, the expected products 3f, 3g, and 3h were furnished in good yields of 86%, 89%, and 82%, respectively, from 6-bromo-, 6-chloro-, and 7-fluoro-N-2,2,2trifluoroethylisatin ketimines. The reactions of 5-methyl and 5trifluoromethoxy ketimines $1i^{6,10-13}$ and $1j^{12}$ gave products 3iand 3i in 98% and 80% yields, respectively. In the case of N-2,2,2-trifluoroethylisatin ketimines with a methoxy group at the C5 or C6 position, the targeted products 3k and 3l were produced in good yields of 76% and 66%, respectively. The

reaction with 5-electron-withdrawing group substituted ketimines 1m, $^{10-12}$ 1p, and 1q led to the formation of the desired products 3m, 3p, and 3q in 63%, 57%, and 83% yields, respectively. With disubstituted ketimines $1n^{12}$ and 1o, 12 the corresponding products 3n and 3o were synthesized in excellent yields of 85% and 94%, respectively.

Next, encouraged by these results, we focused our attention on the scope of the allyl bromides as coupling partners (Scheme 3). When the reaction of methyl-substituted allyl bromide 2b was conducted, a mixture of at least four compounds was detected. It was attributed to the formation of a mixture of diastereomers due to a quaternary center at the C4 and C2 positions of the piperidine backbone and to a trace amount of tetrahydropyridine triggered by DBU. To avoid the difficult separation of the mixture and complicacy, reaction temperatures ranging from 60 to 100 °C were examined for the aza-Prins cyclization step. At 100 °C, spiro N-heterocyclic oxindoles 4b and 4b', which were isolated using column chromatography, with carbon-carbon double bonds in the six-membered rings were generated in yields of 56% and 26%, respectively. Using the optimal reaction conditions, the reaction of ketimine 1a with 2a was attempted, affording 4a and 3a in yields of 12%, and 69%, respectively. Cyclopentyl-substituted allyl tosylate $2c^{15}$ was converted to the products 4c and 4c' in 14% and 15% yields, respectively. The expected products 4d and 4d' were obtained in a combined yield of 93% from phenyl-substituted allyl bromide. In contrast to the alkyl groups, the phenyl group enhanced the reaction, and this was attributed to the construction of a fully conjugated styrene moiety in the structure of the product. When **1a** was treated with $2e^{16,17}$ or $2f_{r}^{16-19}$ mixtures of regioisomers 4e and 4e' and regioisomers 4f and 4f' were formed in combined yields of 92% and 81%, respectively. p-Chlorophenyl- and pfluorophenyl-substituted allyl bromides $2g^{16-19}$ and $2h^{17-19}$ were tolerated, and they yielded the adducts 4g and 4g' and adducts 4h and 4h' in combined yields of 89% and 82%, respectively. In the case of polyaromatic substituted allyl bromide 2i,^{16,19} the corresponding products 4i and 4i' were prepared in yields of 64% and 15%, respectively. The reaction with $2j^{20}$ led to give the products 4j and 4j' in a combined yield of 53%. Allyl bromides with a variety of substituents at the C1 position could not be converted to the anticipated products owing to steric hindrance.

Based on the experimental results, a possible reaction mechanism, depicted in Scheme 4, was proposed for the synthesis of 3a from ketimine 1a via umpolung allylation/aza-Prins cyclization. Hydrogen on the carbon atom adjacent to the CF_3 group in ketimine 1a is abstracted by DBU to generate an 2azaallyl anion, which reacts with allyl bromide (Scheme 4a). Interestingly, the 2-azaallyl anion furnishes intermediates I and I' in a ratio of 1:0.2, according to ¹H NMR spectroscopy. This result showed well-known umpolung reactivity^{9a,b,21} of ketimine 1a, which acts as a nucleophile through the imine carbon on either the Re- or Si-face. In addition, some 2-azaallyl anions spontaneously undergo aza-Cope rearrangement to produce I' or allylation occurs at the nucleophilic carbon adjacent to the CF₃ group. Finally, the intermediates I and I' transform into the desired products II and II' through aza-Prins cyclization in the presence of H₂O and TMSBr. Remarkably, water does not react with intermediates I and I' as a nucleophile; thus, we assume that water either solubilizes the iminium bromide generated from DBU to prevent whatever hampers the reaction²² or activates TMSBr to accelerate aza-Prins cyclization.²³ Furthermore, preliminary studies have shown that chiral phosphine ligands,

such as (R)-BINAP, affect the desymmetrization of 1a with promising levels of enantioselectivity (Scheme 4b, 19% enantiomeric excess (ee)). This result suggests that umpolung allylation/aza-Prins cyclization is amenable to an asymmetric catalysis system, and this research is underway.

In conclusion, we have demonstrated the first umpolung allylation/aza-Prins cyclization of N-2,2,2-trifluoroethylisatin ketimines for the syntheses of spiro[indoline-3,2'-piperidin]-2ones and 5',6'-dihydro-1'H-spiro[indoline-3,2'-pyridin]-2-ones in a one-pot process. The six-membered spiro azacyclic oxindole framework, which is found often in natural products and pharmaceuticals, is considered as an important structure that requires facile synthetic routes toward the development of new drug candidates because of the bioactivity of compounds possessing this motif. The developed protocol, which grants access to this crucial scaffold, features a single-step process with broad functional group tolerance under transition-metal free conditions. Moreover, a new [3 + 3] cycloaddition strategy involving aza-Prins cyclization and utilizing the umpolung reactivity of ketimines was established, enabling the construction of six-membered spiro azacyclic oxindoles. This aza-Prins cyclization differs clearly from the general version, which involves an amine, aldehyde, and acid. Further investigations for controlling the enantioselectivity are ongoing.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.1c00292.

Experimental procedures, characterization data, and copies of all ¹H and ¹³C NMR spectra for all isolated compounds (PDF)

Accession Codes

CCDC 2053124 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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