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Ruthenium(II) Catalysis/Noncovalent Interaction Synergy for Cross-Dehydrogenative Coupling of Arene Carboxylic Acids

Suman Dana,^a Deepan Chowdhury,^a Anup Mandal,^a Francis A. S. Chipem,^b and Mahiuddin Baidya^{*a}

^{*a*}Department of Chemistry, Indian Institute of Technology Madras, Chennai 600036, India ^{*b*}Department of Chemistry, Manipur University, Canchipur 795003, India

ABSTRACT: A ruthenium catalyzed cross-dehydrogenative coupling is developed with the aid of weakly coordinating carboxylic acid group towards the dimerization of arene carboxylic acids. The protocol is operationally simple and suitable to fabricate diverse homo- as well as cross-dimerized products in high yields. Computational insights have also been unveiled to comprehend the plausible reaction mechanism. The critical innovation of the synthetic strategy hinges on the soluble basic additive DBU, which constitutes a synergy of Ru(II)-catalysis with noncovalent interaction and thus, stabilizes pivotal intermediates to promote the challenging dimerization process.

KEYWORDS: C–H activation, Ru(II)-catalysis, noncovalent interaction, weak coordination, cross–dehydrogenative coupling, diphenic acids, DFT computation.

Demystifying powerful and efficient synthetic arsenals for practical accessibility to natural products and versatile synthetic building blocks from structurally simplified progenitors is a continuous enterprise in contemporary organic synthesis. On this ground, owing to their diverse display in organic synthesis, 2,2'-biaryl acid motifs, both in symmetrical and unsymmetrical substitution patterns, have always remained in the focus of general interest.^{1,2} They are the key precursors for the synthesis of numerous



Figure 1. Representative biologically active molecules with 2,2'-biaryl acid scaffold and derivatives thereof

natural products and drug molecules,²⁻⁴ high performance polymers,⁵ diverse C₂-symmetric chiral ligands or auxiliaries,⁶ and their sugar analogues are naturally occurring biologically active molecules that exhibit therapeutic properties (Figure 1). Despite the immense importance of this multi-faceted class of biaryl analogues, modular synthetic routes towards these scaffolds are limited. Traditionally, they have been synthesized using transition-metal promoted Ullmann-coupling strategy,^{7,8} Suzuki cross-coupling method,9 or oxidation of phenanthrene derivatives.^{10,11} However, all these processes require prefunctionalized precursors and offer limited access to these molecules. Thus, a strategic development of direct synthetic route to streamline the construction of these molecular architectures would greatly enrich the synthetic chemist's repertoire. Ideally, a C-H/C-H cross-coupling of commercially

available, cheap, and bench-stable benzoic acids would be the most appropriate disconnection, which can reduce the synthetic overhead for the rapid production of diverse diphenic acids. To address the aforementioned goal, we envisaged to devise a Ru(II)-catalyzed direct C–H/C–H cross-dehydrogenative coupling (CDC) of benzoic acids (Scheme 1).



Scheme 1. Hypothesis for Cross-dehydrogenative Dimerization of Benzoic Acids

Notably, in last two decades, transition-metal catalyzed cross-dehydrogenative coupling reactions of otherwise inert C–H bonds have been realized as a powerful and straightforward transformative tool for the regioselective functionalization of organic molecules since it can evade

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tedious prefunctionalization processes, widen the substrate accessibility, and render easy late-stage functionalization.¹² However, major breakthroughs came, particularly in the arena of weakly coordinating common organic functional group assisted CDC reactions, under expensive iridium, rhodium, and palladium catalysis, and solution for the use of inexpensive and bench-stable ruthenium(II) catalysts is largely unanswered and mostly restricted to Hecktype olefination processes.¹²⁻¹⁴ In fact, even after the tremendous advancement of carboxylate directed ruthenium(II) catalysis for annulation,^{15a-d} olefination,^{15e-i} alkylation,^{15j-k} alkynylation,^{15l} arylation,¹⁶ allylation,¹⁷ and heteroatom functionalization¹⁸ of arenes, carboxylate directed cross-dehydrogenative coupling reaction between two arenes is still unknown under ruthenium catalysis. To the best of our knowledge, till date, the cross-dehydrogenative dimerization of benzoic acids was accomplished only under Rh-catalysis (Scheme 2).131-n



Scheme 2. Previous Reports on Rh-Catalyzed Crossdehydrogenative Dimerization of Benzoic Acids

Considering this synthetic space, we first evaluated the mechanistic feasibility of this approach by calculating the activation energy (E_a) for the formation of alleged intermediate B from the ruthenacycle A, which necessitates a second C-H metalation step (Scheme 1). At this scenario, ruthenacycle A has to mislay its stable 5-membered chelation with high kinetic barrier i.e. $E_a \sim 53.4$ kcal/mol. Recent trends in organic chemistry have rigorously harnessed various noncovalent interactions such as ion-pair, hydrogen-bonding etc., to steer diverse organic transformations including C-H activation.¹⁹ Thus, we hypothesized that this kinetically constrained process could be circumvented through beneficial noncovalent interactions in the transition-state (Scheme 1). It has been shown that protonated organic bases alluringly stabilize the carboxylate intermediates in several organocatalytic processes.²⁰ We were optimistic that a judicial choice of an organic base (Y) would stabilize the TS, illustrated in Scheme 1, favoring the transformation from C to D, which later on reductive elimination would lead to the product diphenic acid P.

Guided by this considerations on the merger of noncovalent interaction with Ru(II) catalysis, at the outset, we examined the oxidative dimerization of 4-toluic acid (1a) as a model substrate under aerial atmosphere (Table 1). Gratifyingly, the treatment of **1a** with [RuCl₂(*p*cymene)]2 (2.5 mol %), *i*Pr2NEt base in dioxane solvent at 110 °C dispensed the desired product 2a in 15% yield after the esterification (entry 1). Other organic bases such as DMAP, DABCO, and *N*-methyl morpholine were unable to promote this transformation (entries 2-4). The hindered amidine base DBU was documented to stabilize the proline derived enamine carboxylate intermediate with prolinate-DBUH⁺ ion pair formation.^{19a,b} Thus, we next performed the reaction using DBU as the base. To our

delight, the transformation occurred smoothly with increased yield (54%, entry 5). Other organic solvents were screened, but they were found less effective than dioxane (entries 6–8). When molecular oxygen was used in lieu of air, yield was improved slightly and further examinations of different oxidants provided CuO as the

Table 1. Optimization of Ru(II)-Catalyzed Dimerization of Aromatic Acids^a

	COOH ^{i)[} Me si) 1a	Ru(<i>p</i> -cymene)Cl ₂] ₂ (2.5 mol %) base (1.0 equiv) base (1.0 equiv) blvent, 110 °C, 24 h Me K ₂ CO ₃ , Mel, rt, 4 h	CO ₂ Me Me MeO ₂ C 2a	
Entry	Base	Oxidant	Solvent	Yield (%)
1	<i>i</i> Pr ₂ NEt	air	Dioxane	15
2	DMAP	air	Dioxane	NR
3	DABCO	air	Dioxane	NR
4	<i>N</i> -methyl morpholine	air	Dioxane	NR
5	DBU	air	Dioxane	54
6	DBU	air	DMF	12
7	DBU	air	Toluene	40
8	DBU	air	MeNO ₂	NR
9	DBU	02	Dioxane	59
10	DBU	CuO	Dioxane	72
11	DBU	MnO ₂	Dioxane	56
12	DBU	Ag ₂ O	Dioxane	30
13	DBU	Cu(OAc) ₂ .H ₂ O	Dioxane	NR
14	DBU	CuO	Dioxane	49^{b}
15	K_2CO_3	CuO	Dioxane	NR
16	NaOAc	CuO	Dioxane	NR
17	DBN	CuO	Dioxane	51

^aUnless otherwise noted, all reactions were conducted on a 0.3 mmol scale with oxidant (1.0 equiv), base (1.0 equiv), solvent (0.4 mL) at 110 °C. Yields are those of isolated products. ^bReaction was performed at 100 °C. DMAP: 4dimethylaminopyridine; DABCO: 1,4diazabicyclo[2.2.2]octane; DBU: 1,8-diazabicyclo[5.4.0] undec-7ene; DBN: 1,5-diazabicyclo[4.3.0]non-5-ene.

best choice, delivering 2a in 72% isolated yield (entries 9-13). The lowering of reaction temperature to 100 °C had a detrimental effect on the reaction outcome (entry 14). As anticipated, common inorganic bases such as K₂CO₃ and NaOAc remained unproductive (entries 15–16). Interestingly, in line with our hypothesis, the use of DBU congener DBN supplied 51% yield of 2a, signifying the important role of amidine bases to promote this CDC reaction (entry 17).

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Having acquired the optimal conditions, we next sought to explore the scope of the reaction (Table 2). The present catalytic conditions were quite general to diverse carboxylic acids. Aromatic carboxylic acids having functional groups such as methyl, methoxy, halides, and benzoyl at the *ortho*-position were tolerated under the

Table 2. Scope of Ru(II)-Catalyzed Homo-dimerization of Benzoic Acids^a



^{*a*}All reactions were conducted on a 0.4 mmol scale with Cu0 (1.0 equiv), DBU (1.0 equiv), dioxane (0.4 mL) at 110 °C; esterification was performed using MeI (3.0 equiv) and K₂CO₃ (3.0 equiv). Yields of isolated ester products were provided. ^{*b*}Reaction was conducted on 1.0 mmol scale with CuO (1.0 equiv), DBU (1.0 equiv), dioxane (1.0 mL) at 110 °C, 36 h; esterification was performed using MeI (3.0 equiv) and K₂CO₃ (3.0 equiv).

reaction conditions to deliver high yields of the desired products **2b-g**. Biphenyl carboxylic acids were also effectively participated in this transformation to give **2h-i** with extended π -system. Benzoic acid bearing strongly electron withdrawing acetyl functionality in the *meta*-position was endured under the current reaction conditions to provide 56% yield of **2j**. Electron donating methoxy substituent at the *para*-position offered 66% yield of the anticipated product **2k**. Benzoic acids with common protecting groups like benzyl (Bn) and methoxymethyl (MOM) were viable substrates for this strategy (**2l-m**). Importantly, the catalytic conditions were able to accommodate sensitive halogen functionalities at the *para*-position, rendering good yields of the desired products (**2n-o**). Dimerizations of hindered dimethylbenzoic acids and 1-naphthoic acid were also achieved to prepare **2p**, **2q**, and **2r** in 53%, 47%, and 84% yields, respectively.

To adorn the synthetic versatility of the protocol further, we attempted the more challenging cross-dimerization of benzoic acids (Table 3). Employment of electron donating group substituted benzoic acids in combination with electron withdrawing group substituted benzoic acids under the optimal conditions with a change in molar ratio of the reaction partners provided synthetically useful yields of the desired cross-dimerized products (3a-c). Exposure of 1-naphthoic acid to substituted benzoic acids was also fruitful to give desired products **3d-f** in good yields (56-77%). It is worth noting that a suitable choice of benzoic acids is important in the cross-dimerization reaction as electronic bias plays a decisive role in product formation. It has been observed that a combination of two electronrich benzoic acids failed to provide the cross-dimerized product under the standard reaction conditions and mostly homo-dimerized products were observed as major products.

 Table 3. Scope of Ru(II)-Catalyzed Cross-dimerization

 of Benzoic Acids^a



 aAll reactions were conducted on a 0.2 mmol scale with CuO (1.5 equiv), DBU (5.0 equiv), dioxane (0.5 mL) at 110 °C; esterification were performed using MeI (8.0) and K_2CO_3 (4.0 equiv). Yields of isolated ester products were provided.

To demonstrate the synthetic potential, products were utilized in the synthesis of various biologically important biphenyl-tethered heterocycles (Scheme 3). The dimerized products **2a-d** and **2k** were separately reduced using LiAlH₄ and individual crude products were then treated with [RuCl₂(*p*-cymene)]₂ catalyst under oxygen atmosphere to construct 7-membered cyclic lactones (**4a-e**), an abundant core found in various natural products such as isokotanin, ulocladol, and graphislactones (Scheme 3a).^{2d-e,i-o} Lactone **4e** can be easily manipulated to its amide congeners following literature procedure.²¹ Reduction followed by the treatment of concentrated H₂SO₄ on **2c**

generated the 7-membered cyclic ether **6** in 64% yield and such molecular frameworks are potential pharmaceutical target for anticancer drugs (Scheme 3b).²ⁱ Notably, in all these transformations, the compounds were purified only in the final steps. Further, we were also successful to achieve a silver(1)-mediated decarboxylative cyclization to construct biologically potent benzo[c]chromen-6-one **7** in 90% yield (Scheme 3c).^{2p-s}



Scheme 3. Synthetic Applications of Crossdehydrogenative Dimerization Products

To comprehend the plausible reaction mechanism, we performed the deuterium scrambling study, which revealed the reversible nature of C–H metalation (Scheme 4a). Also, significant amount of product **2a** was formed in the presence of excess amounts of radical scavengers such as TEMPO, BHT, and 1,1-diphenylethene, refuting the involvement of radical pathway in the reaction mechanism (Scheme 4b). The use of oxygen (air) was also crucial as the yield dropped significantly when the reaction was executed in inert atmosphere (Scheme 4c). To gain insights on the possibility of Ru(IV)/Ru(II)-mechanistic route, we used strong oxidants in the absence of DBU. Astonishingly, we did not observe any C-C dimerized product, while C-O dimerized product 8a-d were formed when stoichiometric amount of AgNO₃ was used as oxidant along with potassium persulfate (Scheme 4d). The use of sodium benzoate (9) instead of benzoic acid was ineffective to construct the C–C dimerized product even in the presence of DBU (Scheme 4e). These experiments bolstered the unique role of DBU in this transformation and hint of the possible involvement of DBUH+ ion.

(a) H/D Scrambling study:





Radical scavengers: TEMPO = 45%; BHT = 54%; 1,1-diphenylethene = 41%



Scheme 4. Mechanistic Studies of the Crossdehydrogenative Dimerization of Benzoic Acids

With the objective of verifying the decisive role of DBU in this transformation, a computational study using density functional theory (DFT) has been performed to compare the energies of key intermediates and transition-states of the second C–H metalation process. As depicted in Scheme 5, we observed a stabilizing electrostatic interaction between DBUH⁺ and intermediate ruthenium-carboxylate complexes which effectually decrease the activation energy for carbometallation by ~23.84 kCal/mol (Scheme 5).²² The inner-sphere coordination of DBU was ruled out considering the rapid protonation in presence of equimolar benzoic acid. Further, the intermediate I₅ was detected in ESI-HRMS analysis of the crude reaction mixture of **1a**, corroborating the computational findings and our working hypothesis (Scheme 5).

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Scheme 5. Computation Insights to the Role of DBU and Detection of a Key Intermediate with ESI-HRMS Analysis of the Dimerization Reaction of 1a

In conclusion, we have developed a weak coordination assisted Ru(II)-catalyzed cross-dehydrogenative coupling of aromatic carboxylic acids. The protocol tolerated various common organic functional groups to produce good yields of the desired 2,2'-biaryl acid scaffolds. The crossdimerization of benzoic acids has also been achieved through judicious modification of the reaction conditions. The synthetic applicability of this protocol has been showcased through the formation of biologically potent biphenyl tethered lactones, ether, and benzocoumarine. Computational studies disclosed that the basic additive DBU plays a pivotal role to promote this transformation through an electrostatic interaction with Ru(II)carboxylate intermediates. Further applications of this Ru(II)-DBU synergy for carboxylate assisted C-H bond functionalization reaction is ongoing in our laboratory.

ASSOCIATED CONTENT

Supporting Information

This material is available free of charge via the Internet at http://pubs.acs.org.

Synthetic procedures, characterization data for all compounds, copies of ¹H and ¹³C NMR spectra, and computational details (PDF)

AUTHOR INFORMATION

Corresponding Author

* E-mail: mbaidya@iitm.ac.in.

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✓Ru-DBU synergy ✓noncovalent interaction
 ✓bioactive motifs ✓DFT study and mechanistic insights