

ORIGINAL ARTICLE

New barbiturates and thiobarbiturates as potential enzyme inhibitors

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Abstract

A series of 27 new barbiturates and thiobarbiturates have been synthesized by a convenient multi-component reaction in overall excellent yields (87–96%). All the synthesized compounds were characterized by ¹H, ¹³C NMR, EIMS and elemental analysis (C, H, N and S). Furthermore, all compounds were screened for *in vitro antioxidant* (DPPH radical scavenging), *lipoxygenase*, *chymotrypsin*, α -*glucosidase* and *anti-urease* activities. Out of the series, 23 in DPPH, 14 in *lipoxygenase*, 2 in *chymotrypsin* have shown appreciable IC₅₀ values.

Keywords

α -Glucosidase and anti-urease, antioxidant (DPPH radical scavenging), barbituric acid, chymotrypsin, lipoxygenase, thiobarbituric acid, thiosemicarbazone

History

Received 28 September 2013

Revised 3 February 2014

Accepted 7 February 2014

Published online 25 March 2014

Introduction

Barbiturates and thiobarbiturates are cheap and readily available synthon used in the manufacturing of diverse pharmaceutical products¹ due to their remarkable biological potentials for various medical conditions. Barbiturates are well-known anesthetic and sleep inducing agents² by acting through central nervous system. These are employed as hypnotic, sedative, anticonvulsant and for anesthesia^{3,4}. Furthermore, some are used for the treatment of psychiatric disorders such as anxiety and epilepsy by posing effects on motor and sensory functions^{5,6} and are also reported as bactericidal agent^{3,4,7}.

Previously we have reported and identified various pharmacophores responsible for potent activity of barbiturate/thiobarbiturate nucleus via molecular modeling studies^{8–11}. It is evident that the imide or β -amino α , β -unsaturated amide functionality is the basic structural moiety in these compounds which has multidimensional array of interaction sites resulting in either potential urease inhibitors (PUI), potential antibacterial (PAB), antifungal Inhibitors (PAF) and potential kinase inhibitors (PKI; Figure 1). In case of PUI, PAB and PAF the mode of action is believed to be mainly involving the scavenging or encapsulation of the metal ion whereas in PKI the presence of various donors/receptors interactions via hydrogen bonding, electronic interactions and metal ion encapsulation seems to be the cause. Further studies are still required to strengthen and verification of these findings.

Emphasizing on these valuable pharmacophores, various new libraries of barbiturates/thiobarbiturates were identified and

are synthetically afforded in the present series of compounds (1–27) via an efficient one pot multi-component reaction in an excellent yield. All the synthesized compounds were extensively characterized and screened for *in vitro antioxidant* (DPPH radical scavenging), *lipoxygenase*, *chymotrypsin*, α -*glucosidase* and *anti-urease* activities.

Material and methods

General

All reagents and solvents were used as supplied and/or recrystallized/redistilled as necessary. Melting points were taken on a Fisher–Johns melting point apparatus and are uncorrected. Elemental analyses were performed on a Leco CHNS-9320 elemental analyzer. The NMR spectra were recorded in DMSO-d6 on Bruker (Rhenistetten-Forchheim, Germany) AM 300 spectrometers operating at 300 MHz, using TMS as an internal standard. ¹H chemical shifts are reported in (ppm) and coupling constants in Hz. The electron impact mass spectra (EIMS) were determined with a Finnigan MAT-312 and a JEOL MS Route mass spectrometer. The progress of the reaction and purity of the products were checked on TLC plates coated with Merck silica gel 60 GF254 and the spots were visualized under ultraviolet light at 254 and 366 nm and/or spraying with iodine vapors. *In vitro* enzyme inhibition and antibacterial assays of the synthesized compounds were carried out at the Department of Biotechnology, The Islamia University of Bahawalpur.

Synthesis

General procedure for the preparation of compounds (1–27)

To a hot stirred solution of barbituric/thiobarbituric acid (2.00 mmoles) and ethylorthoformate (2.02 mmoles) in 2-butanol

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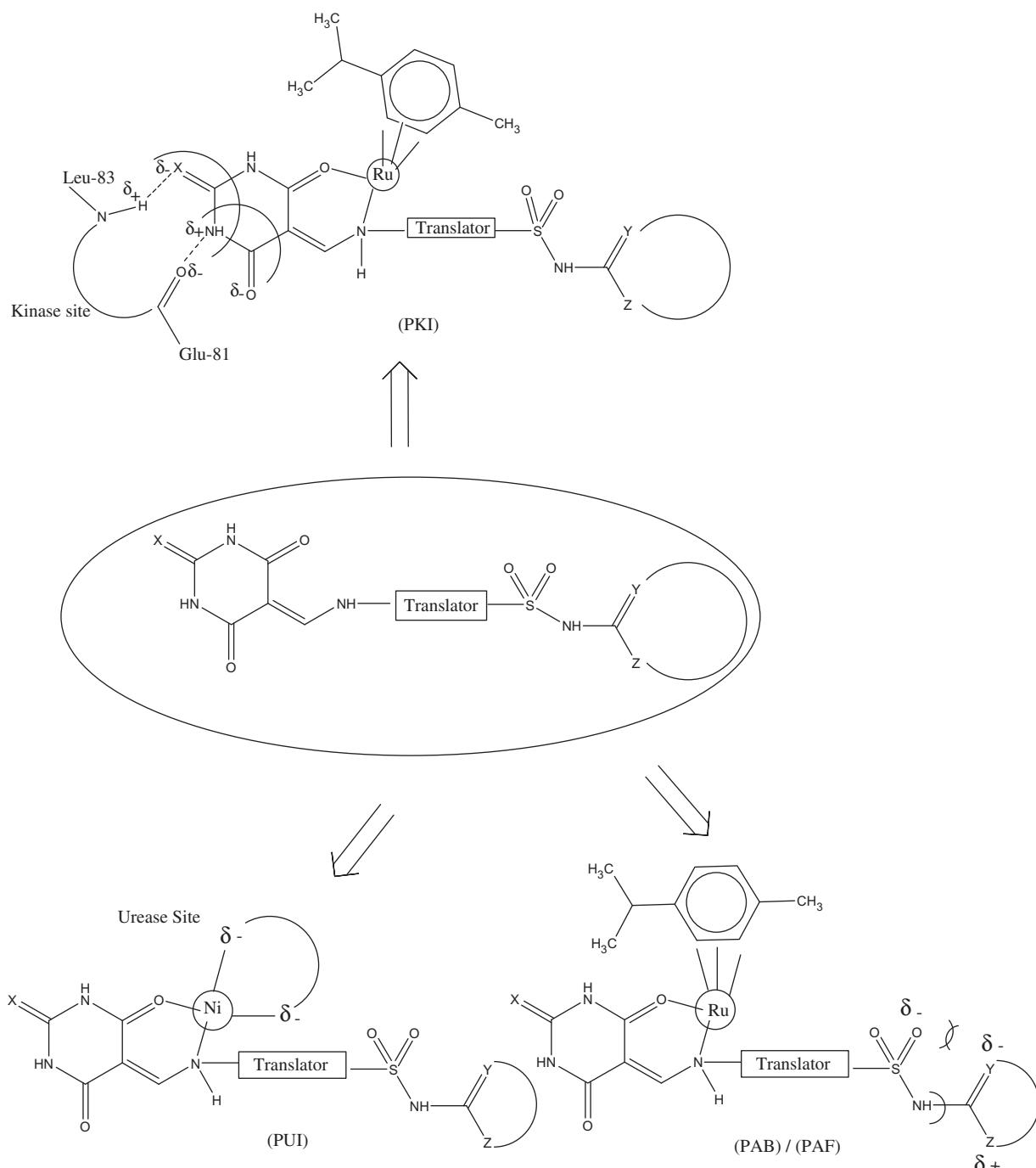


Figure 1. Pharmacophore site combination. ($X = O, S$). PUI: Potential urease inhibitors ($Y, Z = O, N, S$); (PAB)/(PAF): potential antibacterial and antifungal inhibitors ($Y, Z = O, N, S$); (PKI): potential kinase inhibitors⁸.

(10 mL) was added the respective amine (2.00 mmoles). Then the reaction mixture was refluxed till the completion of the reaction (for ≈ 3 h). The precipitates formed during refluxing were collected by suction filtration. Washing with hot ethanol afforded TLC pure products in good to excellent yield⁸ (Scheme 1).

Enzyme inhibition (*in vitro*)

DPPH radical scavenging activity

Stable 1,1-diphenyl-2-picrylhydrazyl radical (DPPH) was used for determining antioxidant activity according to the reported method¹². Different concentrations of compounds in respective solvents were added at an equal volume (10 μ L) to 90 μ L of 100 μ M methanolic DPPH in a total volume of 100 μ L in 96-well plates. After mixing contents, these were incubated at 37 °C for

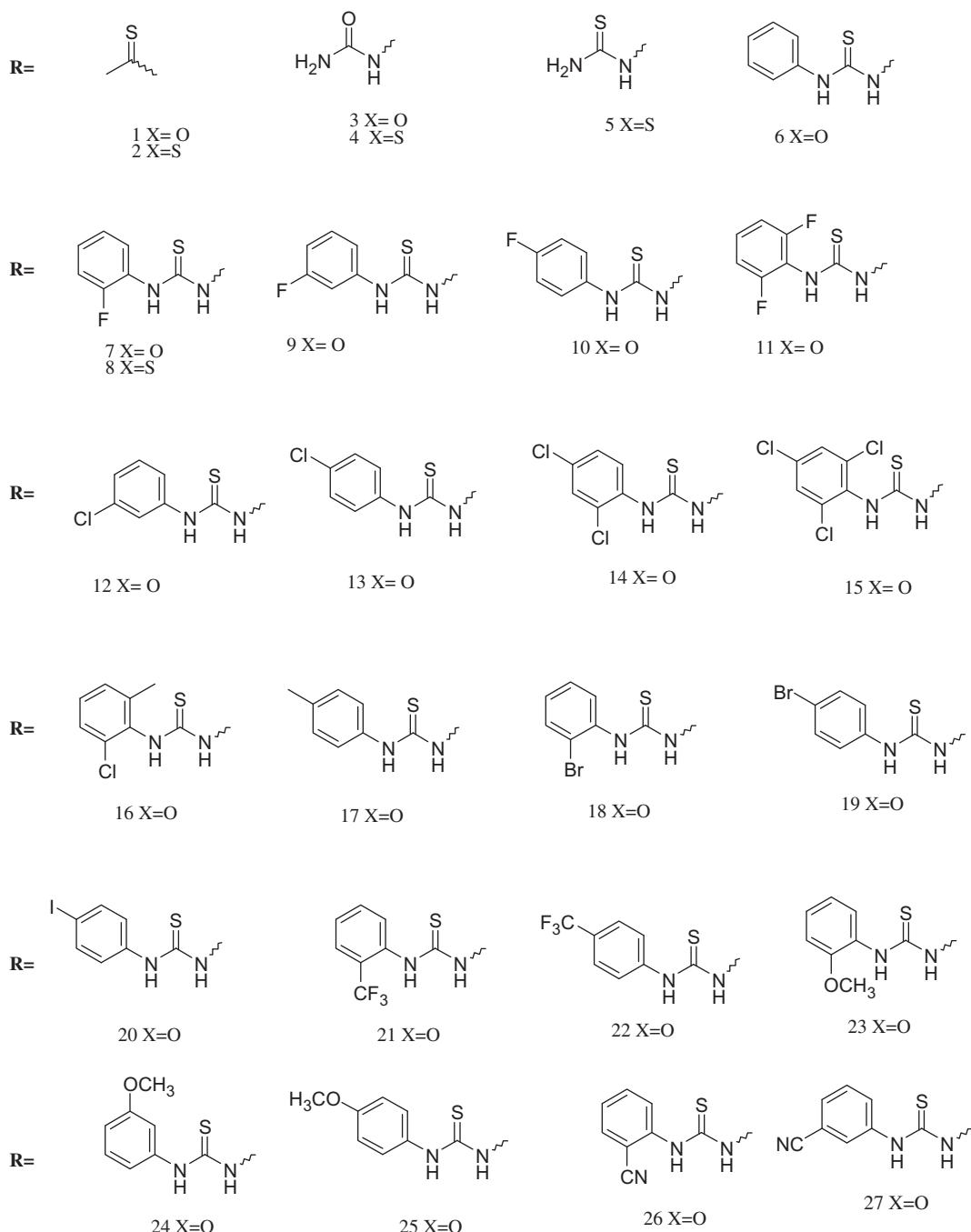
30 min. Absorbance was then measured at 517 nm. Quercetin was used as standard antioxidants. Decrease in absorbance indicated that the increased radical scavenging activity which was determined by the following formula.

$$\text{Percent scavenging activity} = [100$$

$$- \left(\frac{\text{Abs of test compound}}{\text{Abs of control}} \right) \times 100$$

Lipoxygenase

Lipoxygenase (LOX) activity was assayed according to the reported method^{13–15} with slight modifications. A total volume of 200 μ L assay mixture contained 140 μ L sodium phosphate



Scheme 1. Synthesis of **1–27**.

buffer (100 mM, pH 8.0), 20 µL test compound and 15 µL (600 U) purified lipoxygenase enzyme (Sigma, St. Louis, MO). Contents were mixed, pre-read at 234 nm and pre-incubated for 10 min at 25 °C. Reaction was initiated by addition of 25 µL substrate solution. Change in absorbance was observed after 6 min at 234 nm. Baicalein (0.5 mM well⁻¹) was used as a positive control.

Chymotrypsin

α -Chymotrypsin inhibition activity was performed according to slightly modified method of Rehman et al¹⁶. A total volume of 100 μ L assay mixture contained 60 μ L Tris-HCl buffer (50 mM pH 7.6), 10 μ L test compound and 15 μ L (0.9 units) purified α -chymotrypsin enzyme (Sigma, St. Louis, MO). The contents were mixed and incubated for 20 min at 37 °C and pre-read at

410 nm. The reaction was initiated by the addition of 15 mL (1.3 mM) substrate (*N*-succinyl phenyl-alanine-*p*-nitroanilide). The change in absorbance was observed after 30 min at 410 nm. Chymostatin (0.5 mM well⁻¹) was used as a positive control.

α -Glucosidase

The α -glucosidase inhibition activity was performed according to the slightly modified method of Pierre et al¹⁷. Total volume of the reaction mixture of 100 μ L contained 70 μ L 50 mM phosphate buffer saline, pH 6.8, 10 μ L (0.5 mM) test compound, followed by the addition of 10 μ L (0.057 units) enzyme. The contents were mixed, pre-incubated for 10 min at 37 °C and pre-read at 400 nm. The reaction was initiated by the addition of 10 μ L of 0.5 mM substrate (*p*-nitrophenyl glucopyranoside). Acarbose was used as positive control. After 30 min of incubation at 37 °C, absorbance

was measured at 400 nm using Synergy HT microplate reader. All experiments were carried out in triplicates.

Urease

The enzyme assay is the modified form of the commonly known Berthelot assay. A total volume of 85 µL assay mixture contained 10 µL of phosphate buffer of pH 7.0 in each well in the 96-well plate followed by the addition of 10 µL of sample solution and 25 µL of enzyme solution (0.1347 units). Contents were pre-incubated at 37 °C for 5 min. Then, 40 µL of urease stock solution (20 mM) was added to each well and incubation continued at 37 °C for further 10 min. After given time, 115 µL phenol hypochlorite was added in each well (freshly prepared by mixing 45 µL phenol reagent with 70 µL of alkali reagent). For color development, incubation was done at 37 °C for another 10 min. Absorbance was measured at 625 nm using the 96-well plate reader Synergy HT.

Statistical analysis

Statistical analysis was performed by Microsoft Excel 2003. Results are presented as mean ± SEM. The percent inhibition was calculated by the help of following equation.

$$\text{Inhibition (\%)} = \frac{\text{Control} - \text{Test}}{\text{Control}} \times 100$$

IC_{50} values (concentration at which there is 50% enzyme inhibition) of compounds were calculated using EZ-Fit Enzyme kinetics software (Perrella Scientific Inc., Amherst, NH). All the measurements were done in triplicate. Synergy HT (BioTek, Winooski, VT) 96-well plate reader was used in all experiments. The positive and negative controls were included in the assay.

Results and discussion

This study comprises of synthesis and *in vitro* evaluation of enzyme inhibitory activities of 27 new barbiturates and thiobarbiturates.

Chemistry

For synthesis, to a refluxing solution of thio/barbituric acid and triethylorthoformate in 2-butanol was added respective amine. The precipitates formed were collected and washed with hot ethanol offering pure products in excellent yields (87–96%). The structures of the synthesized compounds were deduced by analytical and spectroscopic (¹H-NMR, ¹³C-NMR and EIMS) data (supplementary material). Satisfactory elemental analyses (±0.4% of calculated values) were obtained for all the compounds.

N-[(2,4,6-trioxotetrahydropyrimidin-5(2H)-ylidene)methyl]ethanethioamide (1)

¹H-NMR (300 MHZ, DMSO-d₆, δ, ppm): 2.94 (3H, s, CH₃), 5.31 (1H, s, NH), 8.00 (1H, s, =CH), 11.51 (2H, s, NHBA); ¹³C-NMR (300 MHZ, DMSO-d₆, δ, ppm): 208, 178, 150, 100, 90, 23; EIMS (70 eV) m/z (%): ([M+], 213 (77)), 197 (14), 185 (17), 155 (100), 144 (32), 128 (37), 114 (14), 99 (13), 85 (11), 69 (25), 59 (44), 44 (41); Anal. Calcd. for C₇H₇N₃O₃S: C, 39.44; H, 3.29; N, 19.72; S, 15.02; Found: C, 39.43; H, 3.28; N, 19.69; S, 15.05.

N-[(4,6-dioxo-2-thioxotetrahydropyrimidin-5(2H)-ylidene)methyl]ethanethioamide (2)

¹H-NMR (300 MHZ, DMSO-d₆, δ, ppm): 2.81 (3H, s, CH₃), 5.11 (1H, s, NHCS), 8.20 (1H, s, =CH), 11.45 (2H, s, CONH); ¹³C-NMR (300 MHZ, DMSO-d₆, δ, ppm): 208, 178, 158, 91, 23; EIMS (70 eV) m/z (%): ([M+], 229 (35)), 213 (23), 171 (100),

160 (6), 144 (10), 127 (4), 113 (5), 101 (8), 84 (3), 59 (14), 43 (14); Anal. Calcd. for C₇H₇N₂O₂S₂: C, 39.07; H, 3.26; N, 13.02; S, 29.77; Found: C, 39.09; H, 3.28; N, 13.00; S, 29.79.

2-[(2,4,6-trioxotetrahydropyrimidin-5(2H)-ylidene)methyl]hydrazincarboxamide (3)

¹H-NMR (300 MHZ, DMSO-d₆, δ, ppm): 6.56 (2H, s, NH₂), 8.83 (1H, s, =CH), 9.93 (2H, s, NHNH), 11.12 (2H, s, NHBA); ¹³C-NMR (300 MHZ, DMSO-d₆, δ, ppm): 167, 157, 151, 99; EIMS (70 eV) m/z (%): ([M+], 213 (1)), 199 (3), 128 (100), 100 (10), 85 (49), 69 (9), 42 (85); Anal. Calcd. for C₆H₇N₅O₄: C, 33.80; H, 3.29; N, 32.86; Found: C, 33.82; H, 3.28; N, 32.87.

2[(4,6-dioxo-2-thioxotetrahydropyrimidin-5(2H)-ylidene)methyl]hydrazincarboxamide (4)

¹H-NMR (300 MHZ, DMSO-d₆, δ, ppm): 6.49 (2H, s, NH₂), 8.06 (1H, s, =CH), 9.90 (2H, s, NHNH), 11.38 (1H, s, NHCOBA), 11.92 (1H, s, CSNHBA); ¹³C-NMR (300 MHZ, DMSO-d₆, δ, ppm) 177, 174, 163, 157, 155, 90; EIMS (70 eV) m/z (%): ([M+], 229 (1)), 199 (7), 186 (47), 144 (19), 116 (8), 127 (6), 110 (11), 69 (20), 59 (12), 44 (100); Anal. Calcd. for C₆H₇N₅O₂S₂: C, 29.39; H, 2.86; N, 28.57; S, 26.12; Found: C, 29.37; H, 2.85; N, 28.58; S, 26.11.

2[(4,6-dioxo-2-thioxotetrahydropyrimidin-5(2H)-ylidene)methyl]hydrazincarbothioamide (5)

¹H-NMR (300 MHZ, DMSO-d₆, δ, ppm): 7.05 (2H, s, NH₂), 8.07 (1H, s, =CH), 11.44 (2H, s, NHNH), 11.97 (2H, s, NHBA); ¹³C-NMR (300 MHZ, DMSO-d₆, δ, ppm) 181, 177, 162, 91; EIMS (70 eV) m/z (%): ([M+], 245 (100)), 186 (61), 158 (48), 144 (22), 136 (22), 127 (6), 115 (20), 101 (5), 89 (25), 77 (11), 63 (14), 50 (8), 44 (17); Anal. Calcd. for C₆H₇N₅O₂S₂: C, 29.39; H, 2.86; N, 28.57; S, 26.12; Found: C, 29.37; H, 2.87; N, 28.58; S, 26.11).

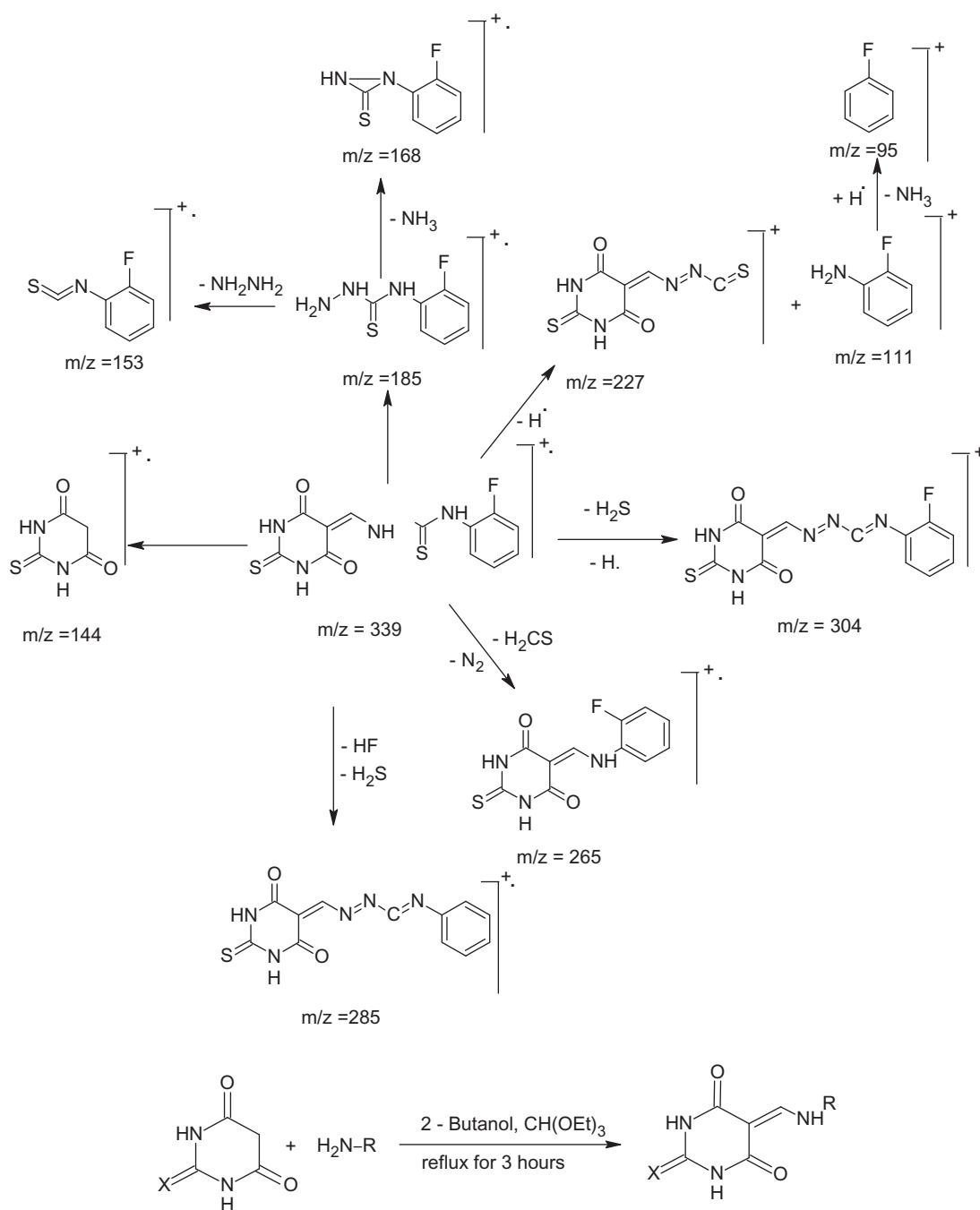
N-(phenyl-2-[(2,4,6-trioxotetrahydropyrimidin-5(2H)-ylidene)methyl]hydrazincarbothio amide (6)

¹H-NMR (300 MHZ, DMSO-d₆, δ, ppm): 5.67 (1H, s, NHAr), 7.01 (1H, d, J 9, ArH), 7.28 (2H, t, J 6,15, ArH), 7.61 (2H, d, J 6, ArH), 8.00 (1H, s, =CH), 10.51 (2H, s, NHNH), 10.94 (2H, s, NHBA); ¹³C-NMR (300 MHZ, DMSO-d₆, δ, ppm) 179, 167, 154, 151, 150, 128, 128, 86; EIMS (70 eV) m/z (%): ([M+], 305 (4)), 231 (100), 187 (3), 170 (67), 154 (58), 128 (90), 118 (22), 77 (49), 44 (31); Anal. Calcd. for C₁₂H₁₁N₅O₃S: C, 47.21; H, 3.61; N, 22.95; S, 10.49; Found: C, 47.22; H, 3.60; N, 22.94; S, 10.50.

The ¹H-NMR spectra of these compounds exhibited three separate singlets at δ = 4.30–6.56, 9.21–11.45 and 10.00–12.81 for the thiosemicarbazones N₄–H, thiosemicarbazone N₂–H and N₃–H and barbiturate NH, respectively. Whereas =CH appeared as a singlet at δ 7.84–9.84. Peaks in APT ¹³C-NMR appeared at δ 175–200 for C=S, 155–170 for C=O, 140–150 for C=CH–NH, 118–140 for aromatic C, 110–70 for C=CH.

In EI mass spectra, all the compounds showed molecular ions of different intensity, which confirmed their molecular weights. The major fragmentation pathway involved the cleavage of the exocyclic N–N, NH–CS and endocyclic NH–CO bonds. Few of the compounds did not show the molecular ion peaks in their spectra. However, the fragments corresponding to thiosemicarbazone moiety, formed by the cleavage of N–N and NH–CS bonds confirmed their structures.

The proposed fragmentation pattern of 8 is presented in Scheme 2.



Scheme 2. Proposed fragmentation pattern of 8.

Bioactivities

Enzyme assays

Results for enzyme inhibition are presented in Table 1. All the compounds are antioxidants except **1**, **5**, **13** and **19**. Compound **10** showed maximum inhibition with IC₅₀ value 63.11 at 0.5 mM and % inhibition of 92.23. 17 Compounds **9**, **17**, **15**, **20**, **14**, **8**, **23**, **25**, **27**, **21**, **12**, **24**, **7** and **11** showed IC₅₀ value of 138.71 at 0.5 mM, 145.25 at 0.5 mM, 196.41 at 0.5 mM, 198.21 at 0.5 mM, 200.31 at 0.5 mM, 204.8 at 0.5 mM, 205.91 at 0.25 mM, 229.1 at 0.25 mM, 303 at 0.25 mM, 329.71 at 0.5 mM, 376 at 0.25 mM, 378.11 at 0.5 mM, 407.11 at 0.5 mM, 407.11 at 0.5 mM, <500 at 0.5 mM, <500 at 0.25 mM and <500 at 0.5 mM against *lipoxygenase*. Against Chymotrypsin almost all the compounds are active but only **18** and **13** showed reportable IC₅₀ value of 78.4 at 0.25 mM, 12.87 at 0.25 mM and 159.3 at 0.5 mM, respectively. Hence

compound **18** found to be most active inhibitor with % inhibition of 78.62.

In DPPH activity p-F is most effective inhibitor. Comparing DPPH IC₅₀ values of 4 and 6, it appeared that thiobarbiturate with NH₂ is more effective than barbiturate with unsubstituted phenyl ring but less effective than the same with a para Fluro substitution **10**. Whereas same Fluro at meta **9**, is less inhibitory than **10**. Also, from IC₅₀ values of **14** and **15**, it is apparent that electron withdrawing substitution at ortho also effects negatively on inhibition activity of such compounds. Order of *lipoxygenase* activity with respect to substitution position on phenyl ring is m-F > p-CH₃ > diortho-p-triCl > p-I > o,p-diCl > o-F > o-OCH₃ > p-OCH₃ > m-CN > m-OCH₃ > m-Cl > o-F. From IC₅₀ values of **9** and **7** (i.e. m-F and o-F) and **9** and **8** (i.e. X = O and X = S), it can be inferred that electron withdrawing fluoro-substituent at meta position inhibits more than at ortho position and also

Table 1. Enzyme assays-Lipoxygenase; 28 quercetin (standard drug), DPPH radical scavenging; 29 baicalein (standard drug), Chymotripsin; 30 chymostatin (standard drug), Alpha-glucosidase; 31 acarbose (standard drug), Antiurease; 32 thiourea (standard drug).

Sr. no.	Inhibition %				IC ₅₀ (μmol.)					
	DPPH	LOX	Chymotrypsin	α-Glucosidase	Anti-urease	DPPH	LOX	Chymotrypsin	α-Glucosidase	Anti-urease
1	39.85 ± 0.72	26.64 ± 0.31	50.38 ± 0.77	17.92 ± 0.36	12.65 ± 0.36	<500	<500	<500	<500	128.21 ± 0.14
2	53.86 ± 0.25	28.96 ± 0.11	46.82 ± 0.69	44.95 ± 0.55	-3.51 ± 0.55	<500	<500	<500	<500	<500
3	65.36 ± 0.34	28.18 ± 0.14	47.07 ± 1.10	1.56 ± 0.53	-2.59 ± 0.53	362.2 ± 0.18	<500	<500	<500	<500
4	85.31 ± 0.69	10.88 ± 0.31	38.42 ± 0.93	18.78 ± 0.16	-1.44 ± 0.16	94.41 ± 0.05	<500	<500	<500	<500
*5	44.38 ± 0.04	28.70 ± 0.11	43.00 ± 1.81	7.97 ± 0.52	-0.18 ± 0.52	<500	<500	<500	<500	<500
6	73.54 ± 0.34	45.87 ± 0.11	23.47 ± 1.31	29.62 ± 0.52	-6.76 ± 0.52	158.41 ± 0.14	<500	<500	<500	<500
7	90.57 ± 0.41	50.59 ± 0.22	7.00 ± 0.71	19.35 ± 0.85	-3.27 ± 0.85	153.51 ± 0.34	<500	<500	<500	<500
8	73.99 ± 0.72	71.04 ± 0.21	25.82 ± 0.91	5.05 ± 0.73	-8.67 ± 0.73	172.21 ± 0.06	229.11 ± 0.03	<500	<500	<500
9	93.08 ± 0.41	83.36 ± 0.41	25.35 ± 0.59	32.15 ± 0.46	-7.45 ± 0.46	72.11 ± 0.63	138.71 ± 0.11	<500	<500	<500
10	92.23 ± 0.45	49.41 ± 0.24	31.94 ± 1.19	2.56 ± 0.12	-2.93 ± 0.12	63.11 ± 0.15	<500	<500	<500	<500
11	77.36 ± 0.44	52.95 ± 0.41	4.17 ± 1.17	15.22 ± 0.71	-0.03 ± 0.71	159.61 ± 0.82	<500	<500	<500	<500
12	92.54 ± 0.64	60.81 ± 0.72	65.63 ± 0.91	9.53 ± 0.76	3.95 ± 0.76	162.34 ± 0.34	407.11 ± 0.34	<500	<500	<500
13	46.23 ± 0.47	24.25 ± 0.22	64.13 ± 1.39	12.38 ± 0.88	-7.93 ± 0.88	<500	159.3	<500	<500	<500
14	46.81 ± 0.17	81.52 ± 0.34	27.73 ± 0.49	-2.56 ± 0.78	-1.23 ± 0.78	74.41 ± 0.22	200.31 ± 0.60	<500	<500	<500
15	84.81 ± 0.78	85.19 ± 0.55	12.89 ± 1.66	26.17 ± 0.66	9.12 ± 0.66	79.91 ± 0.06	196.41 ± 0.44	<500	<500	<500
16	71.34 ± 0.22	18.87 ± 0.64	30.99 ± 0.61	-0.52 ± 0.84	0.19 ± 0.84	78.11 ± 0.05	<500	<500	<500	<500
17	91.82 ± 0.42	28.18 ± 0.14	17.71 ± 0.93	16.93 ± 0.55	-5.36 ± 0.55	82.33 ± 0.11	145.25 ± 0.35	<500	<500	<500
*18	51.98 ± 0.25	35.12 ± 0.52	78.62 ± 0.73	1.85 ± 0.92	-3.16 ± 0.92	<300	12.8	<500	<500	<500
19	40.21 ± 0.26	15.86 ± 0.77	78.62 ± 1.45	-3.041 ± 1.45	3.89 ± 1.45	<500	<500	<500	<500	<500
20	92.54 ± 0.64	81.26 ± 0.41	22.22 ± 1.29	9.96 ± 0.39	-7.40 ± 0.39	81.21 ± 0.07	198.21 ± 0.01	<500	<500	<500
21	83.77 ± 0.11	63.56 ± 0.71	-3.64 ± 0.23	-4.89 ± 0.27	3.88 ± 0.27	97.91 ± 0.31	378.11 ± 0.64	<500	<500	<500
22	82.08 ± 0.59	48.23 ± 0.21	-3.04 ± 1.65	-3.85 ± 1.65	6.84 ± 1.65	93.71 ± 0.04	<500	<500	<500	<500
23	90.38 ± 0.31	70.25 ± 0.77	21.49 ± 0.99	12.38 ± 1.23	6.84 ± 1.23	70.91 ± 0.66	303.91 ± 0.18	<500	<500	<500
24	88.40 ± 0.31	62.39 ± 0.44	31.25 ± 0.49	20.91 ± 1.64	7.21 ± 1.64	99.51 ± 0.08	407.11 ± 0.67	<500	<500	<500
25	90.25 ± 0.14	61.34 ± 0.55	30.46 ± 0.63	6.12 ± 1.67	-2.25 ± 1.67	95.71 ± 0.05	329.71 ± 0.28	<500	<500	<500
*26	55.08 ± 0.24	9.83 ± 0.22	35.42 ± 1.69	42.53 ± 2.03	-5.96 ± 2.03	211.71 ± 0.18	<500	376.81 ± 0.14	<500	<500
27	86.51 ± 0.25	67.63 ± 0.41	1.14 ± 0.49	36.70 ± 0.56	3.58 ± 0.56	145.51 ± 0.31	16.96 ± 0.14	<500	<500	<500
28	92.68 ± 0.25	93.26 ± 0.96	92.60 ± 0.96	92.60 ± 0.51	8.24 ± 1.3	22.4 ± 1.3	38.60 ± 0.15	38.60 ± 0.15	21.25 ± 0.15	
29										
30										
31										
32										

Methanol was used as solvent and concentration of all the samples was 0.5 mM, except in * and *** where concentration is 0.25 mM and 0.025 mM, respectively.

thiobarbiturate is more effective in this typical activity than barbiturate. In *chymotrypsin* inhibition only o-Br and p-Cl compounds have shown some IC₅₀ values. Therefore, this typical activity can be attributed to the halogens at ortho, para position of such compounds. Comparing IC₅₀ values, it is speculated that electrostatic effects of substituents (translator, Figure 1) play an important role in the enzyme inhibitory potential of the synthesized compounds.

Conclusions

This study successfully afforded the synthesis of targeted 27 new barbiturates/thiobarbiturates derivatives in good to excellent yield via an efficient one pot three component reaction. All the synthesized compounds were extensively characterized and screened for biological activities. All the compounds exhibited appreciable % inhibition at each instance strengthening the idea that structural motifs such as imide/β-amino α,β-unsaturated amide play a vital role as pharmacophore–enzyme interaction site and presence of suitable substituents (translator) eventually magnify inhibition of enzymes in general (Figure 1). *In vitro* evaluation for *antioxidant* (DPPH radical scavenging), *lipoxygenase*, *chymotrypsin*, *α-glucosidase* and *anti-urease* activities yielded 23 compounds in DPPH, 14 in *lipoxygenase* and two in *chymotrypsin* exhibiting appreciable IC₅₀ values.

Declaration of interest

There is no declaration of interest. One of the authors Saira Mumtaz likes to acknowledge Higher Education Commission of Pakistan for providing funding and support throughout this study.

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Supplementary material available online

Supplementary materials