



ELSEVIER

Contents lists available at ScienceDirect

Inorganic Chemistry Communications

journal homepage: www.elsevier.com/locate/inoche

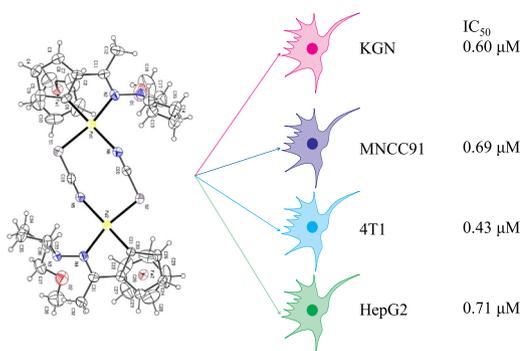
Short communication

Organometallic binuclear Pd(II) complex: Synthesis, crystal structure and in-vitro antitumor activity study

Guidong Gong, Xingxing Gao, Xianhong Yu, Huiya Zhang, Jiaqi Yang, Zhonghui Zhang, Guoyuan Du, Yuan Cao, Gang Zhao*

The school of chemistry engineering, Sichuan University, Chengdu 610064, Sichuan, PR China

GRAPHICAL ABSTRACT



ARTICLE INFO

Keywords:

Planar chiral
 Ferrocene Cyclopalladated compound
 Crystal structure
 Antitumor activity

ABSTRACT

A novel planar chiral ferrocene cyclopalladated compound (**C2**) was synthesized and fully characterized. The absolute configuration of **C2** has been determined by single-crystal X-ray analysis. The cytotoxic activities of **C2** and cisplatin were evaluated against liver (MHCC97 and HepG2), mouse breast (4T1) and ovarian granulosa (KGN) cancer cell lines. **C2** was about 23 times more potent than cisplatin in the suppression of KGN cells.

Cancer seriously threatened human health, the reason of which may be the difficulty in developing new highly active anticancer drugs [1]. Synthesizing new compounds and using them as anticancer drug candidates was an effective way to solve this problem [1,2]. Platinum-based compounds like cisplatin, carboplatin and oxaliplatin play key roles in cancer therapeutics because of their high anticancer activity [3]. However, side effect and drug resistance limited their use in oncotherapy [4]. Under this consideration, palladium-based compounds attracted attention due to the similarly chemical characteristic of platinum and palladium.

Some cyclopalladated compounds were synthesized and their

anticancer activities were evaluated [5–9]. Structure-activity relationship (SAR) study illustrated that the anticancer activity of cyclopalladated compounds was highly related to the C-N cycle, the N-group and the bridge anion [9–11]. Besides, the chirality part also plays a very important role in the anticancer ability of cyclopalladated compounds [8]. It has reported that some cyclopalladated compounds could oxidize the thiol-group of the proteins in organelle membrane and then induce cancer cell apoptosis [5,7]. However, the target of cyclopalladated compounds used as anticancer drug candidates is still unclear. As a result, more cyclopalladated compounds should be synthesized and further mechanization of them inducing cancer cell apoptosis should be studied.

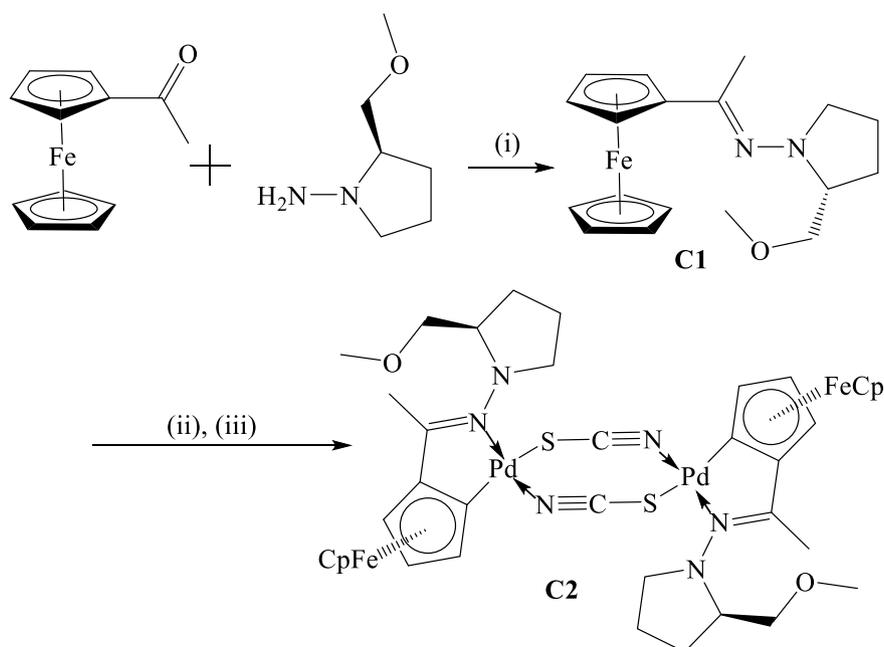
* Corresponding author.

E-mail address: gzhao@scu.edu.cn (G. Zhao).<https://doi.org/10.1016/j.inoche.2019.05.017>

Received 19 April 2019; Received in revised form 10 May 2019; Accepted 10 May 2019

Available online 11 May 2019

1387-7003/ © 2019 Elsevier B.V. All rights reserved.



Scheme 1. Overview of synthesized compound **C2**. i) $C_6H_5CH_3$, $110\text{ }^\circ\text{C}$, 24 h. ii) $Pd(OAc)_2$, MeOH, r.t., 24 h. iii) KSCN, MeOH, r.t., 2 h.

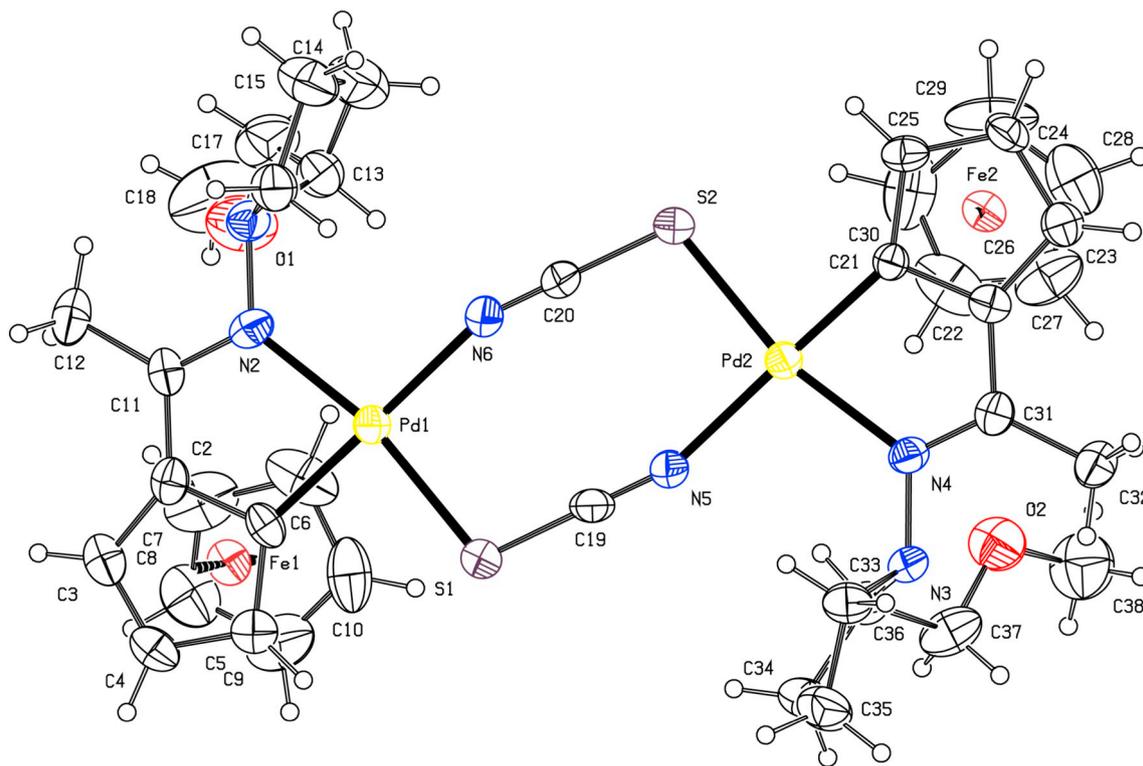


Fig. 1. X-ray crystal structures of **C2** presented as ellipsoid models.

Ferrocene is an eye-catching modified group because of its unique three-dimensional (3D) structure and chemical property [9,12–14]. Some ferrocene cyclopalladated compounds were synthesized and their catalytic activities or anticancer activities were evaluated [8,9,15,16]. However, the involute structure of ferrocene cyclopalladated compounds limited their use. Consequently, synthesizing new ferrocene cyclopalladated compounds and clearing their 3D structures are needful for expanding their use.

Hence, a novel binuclear plane chirality ferrocene cyclopalladated **C2** was synthesized and fully characterized. The absolute configuration

was tested by X-ray single crystal diffraction. The antiproliferation ability of **C2** on KGN, MHCC91, HepG2 and 4 T1 cancer cell lines was evaluated, taking cisplatin as a positive control.

The structural formulas and the numbering of the compounds under study are presented in Scheme 1. The compound **C2** was synthesized as a reported way [9]. In brief, **C1** was obtained from combing acetylferrocene and (+)-(*R*)-1-amino-2-(methoxymethyl)-pyrrolidine in dry toluene with a quite high yield. **C1** mixed with equivalent $Pd(OAc)_2$ in MeOH, stirred at $25\text{ }^\circ\text{C}$ for 24 h and then excess KSCN was added. **C2** was obtained after separated with a column chromatography. **C2** was

Table 1
Crystal data and structures refinement for **C2**.

| Identification code | | C2 | |
|------------------------------------------------|-----------------------------------------------------------------|-----------|--|
| Empirical formula | $C_{38}H_{46}Fe_2N_6O_2Pd_2S_2$ | | |
| Formula weight | 1007.43 | | |
| Temperature/K | 293.15 | | |
| Crystal system | triclinic | | |
| Space group | P1 | | |
| a/Å | 10.6199(4) | | |
| b/Å | 11.1823(4) | | |
| c/Å | 19.9004(8) | | |
| $\alpha/^\circ$ | 93.488(3) | | |
| $\beta/^\circ$ | 95.138(3) | | |
| $\gamma/^\circ$ | 97.897(3) | | |
| Volume/Å ³ | 2325.04(15) | | |
| Z | 2 | | |
| ρ_{calc}/cm^3 | 1.439 | | |
| μ/mm^{-1} | 1.498 | | |
| F(000) | 1016.0 | | |
| Crystal size/mm ³ | 0.35 × 0.3 × 0.25 | | |
| Radiation | MoK α ($\lambda = 0.71073$) | | |
| 2 θ range for data collection/ $^\circ$ | 5.866 to 52.746 | | |
| Index ranges | −13 ≤ h ≤ 12, −13 ≤ k ≤ 13, −24 ≤ l ≤ 22 | | |
| Reflections collected | 19,193 | | |
| Independent reflections | 13,548 [R _{int} = 0.0243, R _{sigma} = 0.0522] | | |
| Data/restraints/parameters | 13,548/6/927 | | |
| Goodness-of-fit on F ² | 0.997 | | |
| Final R indexes [I ≥ 2 σ (I)] | R ₁ = 0.0406, wR ₂ = 0.0930 | | |
| Final R indexes [all data] | R ₁ = 0.0500, wR ₂ = 0.0990 | | |
| Largest diff. Peak/hole / e Å ^{−3} | 0.63/−0.43 | | |
| Flack parameter | −0.002(15) | | |

fully characterized by ¹H NMR, ¹³C NMR, optical rotation, elemental analysis and high resolution mass spectrum.

Single crystal of **C2** (CCDC 1909320, Fig. 1) was obtained by slowly evaporation of the solvents from dichloromethane/*n*-hexane. The crystal and experimental data of **C2** were summarized in Table 1. Selected bond distances and angles for **C2** were showed in Table 2. X-ray analysis shows that the complex exhibits a triclinic form with the space group P1. The pair of coordinating N atoms bear a *trans* relationship, and each palladium atom in the metallacycle is slightly distorted square-planar coordination environment. The two palladium atoms are bond to two thiocyanato groups. The two thiocyanato groups respectively adopts nearly linear configuration [the angle of N(5)-C(19)-S(1) and N(6)-C(20)-S(2): 179.7(10) and 179.5(9) $^\circ$]. The two palladacycles are approximately co-planar, the relevant dihedral angle being 173.06. All bonds are normal except the Pd–C bond lengths, Pd(1)-C(6) and Pd(2)-C(21) are slightly longer than those of the other dimers, but substantially shorter than the predicted value of 2.05 Å due to the metal-to-ligand back-bonding [17–19].

Taking cisplatin as a positive control, Mouse breast cancer cell line 4 T1, human liver cancer cell lines HepG2 and MHCC97, human ovarian granulosa cell line KGN were used to evaluate the antiproliferation

Table 2
Selected bond distances and angles for **C2**.

| Bond distance (Å) | | | |
|-------------------|----------|------------|----------|
| Pd1-S1 | 2.314(3) | Pd2-S2 | 2.318(2) |
| Pd1-N2 | 2.081(7) | Pd2-N4 | 2.088(7) |
| Pd1-N6 | 2.097(8) | Pd2-N5 | 2.106(8) |
| Pd1-C6 | 1.986(8) | Pd1-C21 | 1.947(8) |
| Bond angle (°) | | | |
| N2-Pd1-S1 | 167.4(2) | N4-Pd2-S2 | 166.4(2) |
| N2-Pd1-N6 | 99.0(3) | N4-Pd2-N5 | 99.7(3) |
| N6-Pd1-S1 | 93.5(2) | N5-Pd2-S2 | 93.6(2) |
| C1-Pd1-S1 | 87.3(3) | C21-Pd2-S2 | 87.3(2) |
| C1-Pd1-N2 | 80.1(4) | C21-Pd2-N4 | 79.4(3) |
| C1-Pd1-N6 | 177.8(3) | C21-Pd2-N5 | 178.2(3) |

Table 3

IC₅₀ values of **C2** and cisplatin against HepG2, MHCC97, 4T1 and KGN cell lines.

| Cell lines | IC ₅₀ [μM] value ± SD ^a | |
|------------|-----------------------------------------------|-------------|
| | Cisplatin ^b | C2 |
| HepG2 | 4.6 ± 1.8 | 0.71 ± 0.2 |
| MHCC97 | 5.6 ± 0.2 | 0.69 ± 0.03 |
| 4T1 | 4.85 ± 0.5 | 0.43 ± 0.07 |
| KGN | 14 ± 0.3 | 0.60 ± 0.07 |

^a The values are shown as the mean values of more than two experiments in triplicate; the cells were incubated with **C2** or cisplatin for 72 h.

^b Cisplatin was used as a positive control.

activity of **C2** [20,21]. The CCK8 assay was used to measure the cell viability after treated by **C2** or cisplatin respectively. As shown in Table 3, **C2** illustrated different antiproliferation to different cancer cell lines, and the IC₅₀ to these 4 cancer cell lines were lower than 1 μM. Besides, the IC₅₀ of **C2** to these 4 cancer cell lines was much lower than the positive control cisplatin, which meant that **C2** shows high potential as an anticancer drug candidate. The reason for the high antiproliferation ability of **C2** showed in CCK8 assay may be the co-working of the palladium part and the ferrocene part with different antiproliferation mechanisms [6,7,22,23].

In conclusion, a novel planar chiral ferrocene cyclopalladated compound **C2** was synthesized and fully characterized. The absolute configuration of **C2** was confirmed by X-ray single crystal diffraction. The X-ray analysis showed that **C2** exhibits a triclinic form with the space group P1. The antiproliferative activity assays showed that **C2** were antiproliferative to four cancer cell lines with different potencies. The mechanism of **C2** against tumor cells is proceeding.

Acknowledgements

GZ is grateful for the support from Sichuan Province Science and Technology Support Program (No. 19YYJC2612). GDG is grateful for the support from Sichuan Province Science and Technology Support Program (No. 2019JDRC0106) and the invaluable support received from Shelly Chen over these years. We are grateful for the Analytical & Testing Center of Sichuan University for X-ray diffraction work and we are also grateful to Dr. Luo for his help of single crystal analysis. We are grateful for ceshigo (www.ceshigo.com) for NMR test.

Appendix A. Supplementary material

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.inoche.2019.05.017>.

References

- J. Drost, H. Clevers, Nat. Rev. Cancer 18 (2018) 407–418.
- J.A. Beaver, L.J. Howie, L. Pelosof, T. Kim, J. Liu, K.B. Goldberg, R. Sridhara, G.M. Blumenthal, A.T. Farrell, P. Keegan, R. Pazdur, P.G. Kluetz, JAMA Oncol 4 (2018) 849–856.
- S. Dilruba, G.V. Kalayda, Cancer. Chemoth. Pharm. 77 (2016) 1103–1124.
- L. Kelland, Nat. Rev. Cancer 7 (2007) 573.
- C.M.V. Barbosa, C.R. Oliveira, F.D. Nascimento, M.C.M. Smith, D.M. Fausto, M.A. Soufen, E. Sena, R.C. Araújo, I.L.S. Tersariol, C. Bincoletto, Eur. J. Pharmacol. 542 (2006) 37–47.
- D.P. Santana, E.J. Faria PAParedes-Gamero, A.C. Caires, I.L. Nantes, T. Rodrigues, Biochem. J. 417 (2009) 247–256.
- F.A. Serrano, A.L. Matsuo, P.T. Monteforte, A. Bechara, S.S. Smaili, D.P. Santana, T. Rodrigues, F.V. Pereira, L.S. Silva, M.J. Jr, BMC Cancer 11 (296) (2011) 1–8.
- D. Ning, Y. Cao, Y. Zhang, L. Xia, G. Zhao, Inorg. Chem. Commun. 58 (2015) 57–59.
- G. Gong, Y. Cao, F. Wang, G. Zhao, Organometallics 37 (2018) 1103–1113.
- J. Albert, S. García, J. Granell, A. Llorca, M.V. Lovelle, V. Moreno, A. Presa, L. Rodríguez, J. Quirante, C. Calvis, R. Messeguer, J. Badía, L. Baldomà, J. Organomet. Chem. 724 (2013) 289–296.
- E.G. Rodrigues, L.S. Silva, D.M. Fausto, M.S. Hayashi, S. Dreher, E.L. Santos, J.B. Pesquero, L.R. Travassos, A.C.F. Caires, Int. J. Cancer 107 (2003) 498–504.

- [12] G. Gong, Y. Cao, H. Qian, Y. Zhou, H. Zhao, L. Li, F. Wang, G. Zhao, *Chem. Commun.* 54 (2018) 8312–8315.
- [13] R. Chopra, C. de Kock, P. Smith, K. Chibale, K. Singh, *Eur. J. Med. Chem.* 100 (2015) 1–9.
- [14] P. Marzenell, H. Hagen, L. Sellner, T. Zenz, R. Grinyte, V. Pavlov, S. Daum, A. Mokhir, *J. Med. Chem.* 56 (2013) 6935–6944.
- [15] H. Qian, Z. Yin, T. Zhang, S. Yan, Q. Wang, C. Zhang, *Organometallics* 33 (2014) 6241–6246.
- [16] B. Mu, T. Li, W. Xu, G. Zeng, P. Liu, Y. Wu, *Tetrahedron* 63 (2007) 11475–11488.
- [17] G. Zhao, Q.-G. Wang, T.C.W. Mak, *Tetrahedron Asymmetry* 9 (1998) 1557–1561.
- [18] G. Zhao, A. Qingchuan Yang, T.C.W. Mak, *Organometallics* 18 (1999) 3623–3636.
- [19] C. Navarro-Ranninger, I. Lopez-Solera, A. Alvarez-Valdes, J.H. Rodriguez-Ramos, J.R. Masaguer, J.L. Garcia-Ruano, X. Solans, *Organometallics* 12 (1993) 4104–4111.
- [20] J. Albert, R. Bosque, M. Cadena, L.D. Andrea, J. Granell, A. González, J. Quirante, C. Calvis, R. Messeguer, J. Badía, *Organometallics* 33 (2014) 2862–2873.
- [21] J. Albert, R. Bosque, M. Crespo, G. García, J. Granell, C. López, M.V. Lovelle, R. Qadir, A. González, A. Jayaraman, *Eur. J. Med. Chem.* 84 (2014) 530–536.
- [22] G. Jaouen, A. Vessières, S. Top, *Chem. Soc. Rev.* 44 (2015) 8802–8817.
- [23] R. Chopra, C. de Kock, P. Smith, K. Chibale, K. Singh, *Eur. J. Med. Chem.* 100 (2015) 1–9.