Journal Pre-proofs

Cu₂O-catalyzed C-S coupling of quaternary ammonium salts and sodium alkane-/arene-sulfinates

Youming Huang, Hongyi Chen, Wenting Zheng, Qingle Zeng

PII:	S0040-4039(20)30793-0
DOI:	https://doi.org/10.1016/j.tetlet.2020.152320
Reference:	TETL 152320
To appear in:	Tetrahedron Letters
Received Date:	3 July 2020
Revised Date:	21 July 2020
Accepted Date:	2 August 2020



Please cite this article as: Huang, Y., Chen, H., Zheng, W., Zeng, Q., Cu₂O-catalyzed C-S coupling of quaternary ammonium salts and sodium alkane-/arene-sulfinates, *Tetrahedron Letters* (2020), doi: https://doi.org/10.1016/j.tetlet.2020.152320

This is a PDF file of an article that has undergone enhancements after acceptance, such as the addition of a cover page and metadata, and formatting for readability, but it is not yet the definitive version of record. This version will undergo additional copyediting, typesetting and review before it is published in its final form, but we are providing this version to give early visibility of the article. Please note that, during the production process, errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

© 2020 Elsevier Ltd. All rights reserved.



Tetrahedron Letters

journal homepage: www.elsevier.com

Cu₂O-catalyzed C-S coupling of quaternary ammonium salts and sodium alkane-/arene-sulfinates

Youming Huang, Hongyi Chen, Wenting Zheng, and Qingle Zeng*

State Key Laboratory of Geohazard Prevention and Geoenvironment Protection, Chengdu University of Technology, College of Materials, Chemistry & Chemical Engineering, Chengdu 610059, China. <u>ginglezeng@hotmail.com</u>

ARTICLE INFO

ABSTRACT

Article history: Received Received in revised form Accepted Available online A new protocol for the synthesis of (enantioenriched) benzylic sulfones *via* the Cu₂Ocatalyzed C-S bond cross coupling of alkane-/arene-sulfinates and (enantioenriched) benzylic quaternary ammonium salts has been developed. The product benzylic sulfones were obtained in good to high yields (75-96%). Chiral arylmethyl sulfones with high enantiomeric excess (90-94% ee) were also synthesized in the presence of Cu₂O and 1,1'-bis-(diphenylphosphino)ferrocene (dppf).

2009 Elsevier Ltd. All rights reserved.

Keywords: Copper(I) oxide Quaternary ammonium salts Sodium sulfinates Chiral sulfones C-S coupling

Introduction

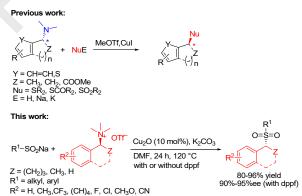
Many therapeutically interesting compounds contain sulfone motifs, such as Dapsone [1], and the nonsteroidal antiandrogen Casodex [2]. Sulfones also find extensive applications in organic and drug synthesis. For example, primary alkyl aryl sulfones can be converted into the corresponding carboxylic acids in good to excellent yields *via* double deprotonation and exposure to atmospheric oxygen [3]; sulfones were used in the preparation of various γ -alkylidene butenolides [4]; and halogenation of the α -position of sulfones afforded α halo sulfones [5].

Sulfones are typically prepared by oxidizing sulfides [6]; however, this is often accompanied by the formation of sulfoxide by-products. In recent years, H_2O_2 [7] and oxygen [8] have been utilized as oxidants to synthesize sulfones from thioethers. Sulfones have also been obtained *via* the cross coupling of halogenated arenes with sodium sulfinate [9].

Recently (enantioenriched) quaternary ammonium salts or tertiary benzylic amines that are activated in situ by methyl triflate were reported as a novel class of electrophiles and coupled with various nucleophiles, such as boron reagents [10], thiols [11], amines [12], and phosphates [13] via C-N cleavage. Inspired by related studies and based on our ongoing studies on carbonheteroatom coupling reactions [14], herein we report the Cu₂O-catalyzed C-S bond cross coupling of (enantioenriched) quaternary ammonium salts and sodium sulfinates via C-N bond cleavage (Scheme 1).

In comparison with amine-derived pyridinium salts [11], diazonium compounds [15] or diazotates [16], highly enantiopure benzylic quaternary ammonium salts

are readily available, and the various chiral sulfone products obtained by our method have excellent ee values (90-95%ee).



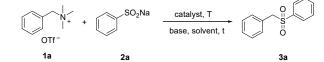
Scheme 1. Synthesis of (enantioenriched) sulfones.

N.N-Initially, the reaction of dimethyl(phenyl)methanamine and methyl triflate afforded its corresponding quaternary ammonium salt (1a). The cross coupling reaction of quaternary ammonium salt (1a) and sodium phenylsulfinate (2a) under various conditions was then investigated (Table 1). In the presence of Cu(MeCN)₄PF₆ (10 mol%) and K₂CO₃ (1.0 eq.), the coupling reaction gave the targeted C-S bond coupling product 3a in good yield (72%, Table 1, Entry 1). Next, the examination of other catalysts and the control reaction without a transition metal catalyst showed that Cu₂O was also a suitable catalyst (Entries 3 vs 1-7). Cu₂O was selected because it is less expensive than Cu(MeCN)₄PF₆ and can be recycled. DMF was found to be the best solvent after the screening of several common solvents (Entries 11 vs 3, 8-10). Increasing the amount of substrate 2a to 2.0 mmol afforded a much

3j: 82%

a stronger base was determined to be unnecessary (Entries 14 vs 12). Increasing the temperature to $120 \,^{\circ}\text{C}$ gave the best result of 95% yield (Entries 15 vs 12 and 16).

Table 1. Optimization of the reaction conditions.^a

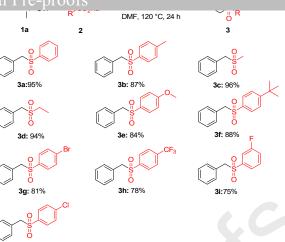


Entry	Catalyst	Base	Solvent	Temp (°C)	Yield 3a (%) ^b	
1	Cu(MeCN) ₄ PF ₆	K ₂ CO ₃	1,4-	110	72	
			dioxane		12	
2	CuI	K_2CO_3	1,4-	110	20	
			dioxane		20	
3	Cu ₂ O	K_2CO_3	1,4-	110	62	
	P C 1		dioxane		02	
4	FeCl ₃	K_2CO_3	1,4-	110	trace	
_			dioxane			
5	NiCl ₂	K_2CO_3	1,4-	110	26	
	a. at		dioxane			
6	CuCl ₂	K_2CO_3	1,4-	110	trace	
7.0		V CO	dioxane	120		
7 °		K ₂ CO ₃	DMF	120	trace	
8	Cu ₂ O	K_2CO_3	MeCN	80	trace	
9	Cu ₂ O	K_2CO_3	THF	60	trace	
10	Cu ₂ O	K_2CO_3	H_2O	100	trace	
11	Cu ₂ O	K_2CO_3	DMF	110	75	
12 °	Cu ₂ O	K_2CO_3	DMF	110	86	
13 ^d	Cu ₂ O	K ₂ CO ₃	DMF	110	84	
14 °	Cu ₂ O	t-	DMF	110	75	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$						
15 °	Cu ₂ O	K ₂ CO ₃	DMF	120	95	
16°	Cu ₂ O	K ₂ CO ₃	DMF	150	92	
^a Reag	ents and condition	ons: Unles	s otherwise	noted.	the	

^a Reagents and conditions: Unless otherwise noted, the reactions were carried out with 1.0 mmol of **1a**, 1.0 mmol of **2a**, 0.1 mmol of catalyst and 1.0 mmol of base, in 5 mL of solvent for 24 h in a sealed test tube with a standard ground mouth. ^b Isolated yield. ^c **2a** (2.0 mmol) was used. ^d **2a** (3.0 mmol) was used.

With the optimized conditions in hand, various sodium arene/alkanesulfinates (2) were reacted with benzyl quaternary ammonium salt 1a (Table 2). Both alkane- and arene-sulfinates gave the desired sulfones in good to excellent yields. Sodium alkanesulfinates, including methane- and ethane-sulfinates afforded similar yields as that of sodium benzenesulfinate (3c and 3d vs 3a). Both electron-donating and electron-withdrawing groups on the sodium benzenesulfinates were not beneficial to this coupling reaction. Additionally, electron-withdrawing groups on the sodium benzenesulfinates decreased the yields more than electron-donating groups (3b, 3e, 3f vs 3g-3j).

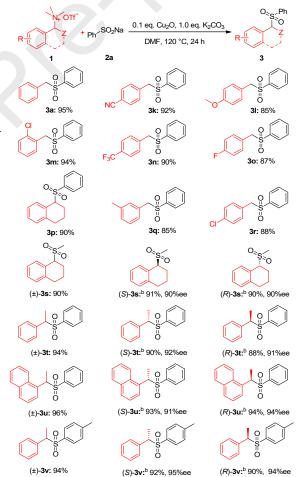
Table 2. Scope of the sodium sulfinates.^a



^a Reagents and conditions: 1.0 mmol **1a**, 2.0 mmol **2**, 0.1 mmol Cu₂O and 1.0 mmol K₂CO₃, in 5 mL DMF, 120 °C, 24 h. Isolated yield.

Next, the reactions of various quaternary ammonium salts (1) with sodium benzenesulfinate (2a) were evaluated (Table 3). Overall, the C-S coupling reactions proceeded smoothly and provided the desired products in high to excellent yields (3a and 3k-3v).

Table 3. Scope of the quaternary ammonium salts.^a



^a Reagents and conditions: 1.0 mmol 1, 2.0 mmol 2a, 0.1 mmol Cu₂O and 1.0 mmol K₂CO₃, in 5 mL DMF, 120 °C, 24 h. Isolated yield. ^b DPPF (0.1 equiv) as ligand.

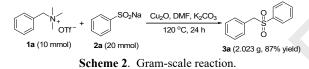
Both electron-withdrawing and electron-donating groups on the benzene ring of the quaternary ammonium salts afforded high yields, and the position of the functional group has no obvious effect (3m, 3r).

Furthermore, these reaction conditions tolerated a wide vari

methoxy (31), cyano (3 κ), Huoro (30), trihuoromethyi (3n), and chloro (3m, 3r).

Disappointingly, enantioenriched quaternary ammonium salts only provided racemic products under the optimized conditions. Fortunately, racemization was inhibited when 1,1'-ferrocenediyl-bis(diphenylphosphine) (dppf) was added as a ligand. For example, (R)- and (S)-1-phenylethanamines from Aladdin reagent company have enantiopurities of 94.1% and 94.6%, respectively (for HPLC traces, see the ESI), and their corresponding (S)- and (R)-benzylic sulfone products 3t have enantiopurities of 92% and 91%, respectively. As a result, all of the enantioenriched benzylic quaternary ammonium salts which were synthesized from commercially available enantioenriched benzylic amines possessed high enantiopurities of 90-95% ((S)- and (R)-3s, 3t, 3u, 3v).

This reaction shows an interesting phenomenon: R forms of the enantioenriched benzylic amine-derived chiral quaternary ammonium salts gave the *S* form of the sulfones and vice versa. For example, the (*R*)-1-phenylethylamine-derived quaternary ammonium salt afforded (*S*)-**3v**, which was verified *via* comparison with the chiral HPLC trace and specific rotation of previously reported (*S*)-**3v** [11]. It appears that the copper-catalyzed C-S bond coupling reaction occurs *via* a S_N2 type substitution, and thus the absolute configurations of the chiral quaternary ammonium salts reverse during the C-S bond coupling reaction.



In order to verify the scale-up possibility of this reaction, a gram-scale reaction was performed. A tenmmol-scale reaction provided the corresponding product

In summary, an efficient method was developed for the synthesis of (enantioenriched) benzylic sulfones viathe Cu₂O-catalyzed C-S bond cross coupling reaction of (enantioenriched) benzylic quaternary ammonium salts and sodium alkane-/arene-sulfinates. Enantioenriched sulfones were obtained in excellent enantiopurities using dppf as a ligand.

Acknowledgments

in 87% yield (Scheme 2).

This work was financially supported by the State Key Laboratory of Geohazard Prevention and Geoenvironment Protection (No. SKLGP2018Z002) for financial support.

REFERENCES

[1] D.C. Lopez, R.A. Pearlstein, A.J. Hopfinger, J.K. Seydel, J. Med. Chem. 30 (1987) 900.

[2] T. Gilligan, W. Oh, Prostate 3 (2001) 42.

[3] Y. Chen, Y.L. Lam, Y.H. Lai, Org. Lett. 5 (2003) 1067.

[4] (a) C.B. Amy, P.B. Matthew, M.P. Bettina, J.M. Daniel, K.M. Ryan., S.S. Murphre, Org. Lett. 13 (2011) 1447 (b) I.T. Phan, G.J. Gilbert, G.W. Neil, Synlett 26 (2015) 1867.

[6] (a) C.G. venier, I.G. Squires, Y.Y. Cnen, G.P. Hussmann, J.C. Shei, B.F.J. Smith, Org. Chem. 47 (1982) 3773; (b) D. Villemin, B.A. Alloum, Synth. Commun. 20 (1990) 925.

[7] K. Masayuki, Y. Junya, N. Takuya, H. Yoshiro, Tetrahedron Lett. 65 (2009) 10477.

[8] T. Abeda, H. Siddiki, O. Wataru, K. Kenichi, S. Ken, Green Chem. 18 (2016) 2554.

[9] (a) B. Zheng, T.Z. Jia, P.J. Walsh, Org. Lett. 15 (2013)
1690; (b) Y. Hu, Z.C. Chen, Z.G. Le, Q.G. Zheng, J. Chem. Res. 4 (2004) 267.

[10] (a) P. Maity, D.M. Shacklady, G. P. A. Yap, E.R. Sirianni, M.P. Watson, J. Am. Chem. Soc. 135 (2013) 280;
(b) D.M. Shacklady, K.M Roberts, C.H. Basch, Y.G. Song, M.P. Watson, Tetrahedron 70 (2014) 4257; (c) J. Hu, H. Sun, W. Cai, X. Pu, Y. Zhang, Z. Shi, J. Org. Chem. 81 (2016) 14.

[11] (a) Y. Gui, S. Tian, Org. Lett. 19 (2017) 1554; (b) W.L. Jiang, N.T. Li, L.H. Zhou, Q.L. Zeng, ACS Catal. 8 (2018) 9899.

[12] H.Y. Chen, H. Yang, N. Li, X. Xue, Z. He, Q.L. Zeng, Org. Proc. Res. Dev. 23 (2019) 1679.

[13] N.T. Li, F. Chen, G.H. Wang, Q.L. Zeng, Monatsch. Chem. 151 (2019) 99.

[14] Q.L. Zeng, L. Zhang, Y. Zhou, Chem. Rec. 18 (2018) 1278.

[15] N. Ileby, M. Kuzma, L.R. Heggvik, S.K. Fiksdahl, Tetrahedron: Asymm. 8 (1997) 2193.

[16] R.A. Moss, J. Banger, Tetrahedron Lett. 15 (1974) 3549.