



Subscriber access provided by University of Sussex Library

Article

NHC-promoted asymmetric beta-lactone formation from arylalkylketenes and electron-deficient benzaldehydes or pyridinecarboxaldehydes

James Douglas, James E. Taylor, Gwydion Churchill, Alexandra M. Z. Slawin, and Andrew D. Smith J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/jo4003079 • Publication Date (Web): 26 Feb 2013 Downloaded from http://pubs.acs.org on February 27, 2013

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



NHC-promoted asymmetric β -lactone formation from arylalkylketenes and electron-deficient benzaldehydes or pyridinecarboxaldehydes

James Douglas, ^a James E. Taylor, ^a Gwydion Churchill, ^b Alexandra M. Z. Slawin ^a and Andrew D. Smith ^a*

^a EaStCHEM, School of Chemistry, University of St Andrews, North Haugh, St Andrews, KY16 9ST, UK

^b AstraZeneca, Process Research and Development, Macclesfield, Cheshire, SK10 2NA, UK

e-mail: ads10@st-andrews.ac.uk

A chiral NHC catalyses the asymmetric formal [2+2]-cycloaddition of alkylarylketenes with both electron deficient benzaldehydes and 2- and 4-pyridinecarboxaldehydes to generate stereodefined β -lactones. In the benzaldehyde series, optimal product diastereo- and enantiocontrol is observed using 2-nitrobenzaldehyde (up to 93:7 dr (*syn:anti*) and 93% ee). Substituted 2- and 4-pyridinecarboxaldehydes are also tolerated in this process, generating the corresponding β -lactones in good yield and enantioselectivity, although the diastereocontrol in these processes is highly dependent upon the aldehyde substitution.

These processes are readily scalable, allowing multi-gram quantities of the β -lactone products to be prepared. Derivatisation of these products, either through ring-opening into the corresponding stereodefined β -hydroxy- and β -amino acid derivatives without loss of stereochemical integrity, or *via* cross-coupling, is demonstrated.

Introduction:

The β-lactone motif is of widespread interest in chemistry, serving as versatile starting materials in complex molecule¹ and building block synthesis, ^{1b,2} monomers in biodegradable polymer synthesis,³ as well as being the core structure in a range of natural products with notable pharmacological properties.⁴ A number of synthetic methods have been used to prepare these scaffolds in enantioenriched form,⁵ ranging from substrate^{5b} and chiral auxiliary⁶ controlled processes to catalytic asymmetric methods. Within the area of asymmetric Lewis base catalysis, a variety of strategies have been utilised to promote the catalytic asymmetric formation of β -lactones from ketenes and aldehydes.⁸ Building on the pioneering work of Wynberg using cinchona alkaloid catalysts, the work of the Nelson, Romo and Calter groups has expanded this approach, with a range of elegant intra- and intermolecular strategies developed. Although versatile, typical limitations of the intermolecular process using cinchona derivatives include the requirement for parent or mono-substituted ketene(s) and aliphatic aldehydes for good reactivity. To date, relatively few studies have been reported that use disubstituted ketenes and benzaldehydes in catalytic asymmetric βlactone formation. In this area, Fu has shown that planar chiral 4-(pyrrolidino)pyridine (PPY) derivative 1 promotes asymmetric β -lactone formation from symmetrical dialkylketenes and benzaldehydes with high enantioselectivity (up to 91% ee, Scheme 1). 13 Kerrigan has recently shown that BINAPHANE 2 can be

used to promote asymmetric β -lactone formation from alkylarylketenes and 4-substituted benzaldehydes, giving preferentially *anti*- β -lactones with high diastereo-and enantiocontrol (up to 94:6 dr and 92% ee). ¹⁴

Scheme 1. Previous Lewis base-promoted asymmetric β -lactone formation from disubstituted ketenes and benzaldehydes

Ye has previously utilized NHCs to promote β-lactone formation from alkylarylketenes using trifluoromethylketones¹⁵ (up to >20:1 dr, 99% ee) or activated 2-oxo-aldehydes¹⁶ (up to 20:1 dr, 99% ee, Scheme 2). Notably, the use of ethyl-2chloro-phenylketene is a necessary substrate constraint for high diastereoselectivity in the latter process, with 4-chloro-benzaldehyde proving inactive to β-lactone formation in this study. Building upon these precedents and our interest in NHC-mediated asymmetric processes, ¹⁷ we now report the development of an alternative and scalable NHC-promoted asymmetric β-lactone synthesis from alkylarylketenes and both benzaldehydes bearing electron withdrawing substituents and pyridinecarboxaldehydes. Furthermore, derivatisation of the β -lactone products, either through ring-opening into the corresponding stereodefined β -hydroxy- and β -amino acid derivatives, or *via* cross-coupling, is demonstrated.

Scheme 2. Previous and proposed asymmetric β -lactone formation from disubstituted ketenes and aldehydes using NHCs

This work:

$$\begin{array}{c|c} O & O & NHC & O & NHC & O \\ Ar^{1} & Ar^{2} & Ar^{2} & Ar^{1} & R & N & R^{1} \\ \end{array}$$

Results and Discussion:

Evaluating NHC-promoted β-lactone formation using benzaldehydes

As NHCs are known to promote benzoin reactions of benzaldehydes¹⁸ as well as ketene dimerisation processes,¹⁹ at the outset of our investigations these were recognised as possible competitive reaction manifolds. NHC precatalyst **3** was chosen for our studies given its precedent to participate in enantioselective cycloaddition processes using ketenes, yet its moderate reactivity in benzoin reactions.²⁰ Initial studies employed ethylphenylketene and triazolium precatalyst **3** with a range of substituted benzaldehydes (Table 1). Whilst benzaldehyde gave no β -lactone products (giving only ketene dimer), promising reactivity was observed with electron-deficient benzaldehydes utilizing dropwise ketene addition to minimise dimerisation. The reactions using 4-trifluoromethyl- or 4-nitrobenzaldehyde performed at 0 °C gave

good yields of β-lactone product (5 and 6) with moderate levels of antidiastereoselectivity (entries 2 and 3).²¹ Improved enantiocontrol was achieved at the detriment of product conversion at -78 °C using 4-nitrobenzaldehyde (entry 4).²² Using 2-nitrobenzaldehyde, high syn-diastereoselectivity (89:11 syn:anti) was observed at 0 °C, giving major syn-diastereoisomer 7 in 94% ee (entry 5). Further optimization using 2-nitrobenzaldehyde was achieved through lowering the reaction temperature, with consistently high levels of diastereo- and enantioselectivity observed (entries 6 to 7). Interestingly, lowering the reaction temperature below -50 °C had a detrimental effect on product ee (entry 8), postulated to be due to a competitive KHMDS-catalysed racemic pathway at this temperature.²³ The absolute and relative configuration within 7 was proven through X-ray crystallographic analysis.²⁴ Following the promising reactivity and stereoselectivity observed using 2nitrobenzaldehyde, the ability of 2-halobenzaldehydes to participate in this reaction process was investigated. 2-Fluoro-, 2-chloro- and 2-bromobenzaldehyde all gave the corresponding B-lactones (8-10) with poor dr (entries 9-11).²⁵ consistent with the 2nitro substituent being a necessary constraint for optimal diastereoselectivity. ²⁶

 11^g

2-BrC₆H₄

Table 1. Initial screening of benzaldehydes for NHC-promoted β-lactone synthesis

Unless stated, ketene added dropwise as a solution in toluene. adr determined by H NMR analysis of the crude reaction mixture. ^bIsolated yield of single diastereoisomer. ^cee determined by HPLC analysis. ^dKetene added in a single portion. ^e1.5 equiv of aldehyde. ^f1.2 equiv of aldehyde. ^gIsolated yield and ee of products after ring opening into the corresponding β -hydroxy acids and derivatisation.

95,71

56:44

-50

With 2-nitrobenzaldehyde identified as giving optimum diastereo- and enantiocontrol in this process, its scope and limitations were explored through variation of the ketene component (Table 2). Firstly, a series of alkylphenylketenes were reacted under the optimized conditions. Incorporation of a methyl substituent led to a decrease in dr while maintaining high levels of enantioselectivity for both diastereoisomers of product 11 (entry 1). Ethyl and n-butyl substitution gave β -lactones 7 and 12 in high dr and ee, while i-butyl incorporation leads to high dr but poor ee (entries 2-4). Variation of the aryl unit within a series of ethylarylketenes was also investigated, with the incorporation of both electron-withdrawing and electron-donating substituents providing β -lactones 14-18 in high dr and ee (entries 5-9). Further substrate variation showed that iso-propyl-3-thiopheneketene gave β -lactone 19 with poor dr, although the anti-diastereoisomer was formed in high ee (entry 10). The NHC-catalysed reaction of either ethyl-2-tolyl- or ethyl-1-naphthylketene with 2-nitrobenzaldehyde gave no β -lactone products, returning the aldehyde starting material (entries 11 and 12). Notably, the observed trend in product diastereoselectivity using NHC-mediated catalysis (reduced dr for methylarylketenes) is opposite to that observed by Kerrigan using phosphine catalysis (high dr for methylarylketenes, reduced dr for n-butylarylketenes).

Table 2. Reaction scope and limitations employing 2-nitrobenzaldehyde

Ketene added dropwise as a solution in toluene. ^adr determined by ¹H NMR analysis of the crude reaction mixture. ^bIsolated yield of mixture of diastereoisomers. ^cee determined by HPLC analysis. ^dIsolated yield of single diastereoisomer. ^eee determined by HPLC analysis via ring-oepning with NaN₃. ^f1.2 equiv of ketene and 1.0 equiv of aldehyde used.

Evaluating NHC-promoted β-lactone formation using pyridinecarboxaldehydes

Although the use of 2-nitrobenzaldehyde allows efficient access to a range of β -lactones in good yield and typically excellent diastereo- and enantiocontrol, the limited scope of this methodology due to the requirement for a 2-nitro substituent for reasonable diastereoselectivity reduces its synthetic versatility. To address this, the ability of heteroaromatic aldehydes to participate in this methodology was investigated. Accordingly, furfural and all isomers of pyridinecarboxaldehyde were identified as possible participants in this protocol. The formation of chiral pyridine containing compounds is of significance due to their use in both asymmetric catalysis and appearance in biologically relevant molecules. Although a number of methods have previously been developed to prepare these motifs, enantiomerically enriched pyridyl substituted β -lactones, to the best of our knowledge, are unreported.

While both furfural and 3-pyridinecarboxaldehyde gave the corresponding β-lactone products with unsatisfactory levels of conversion (<10%), 2- and 4pyridinecarboxaldehyde both proved efficient coupling partners with ethylphenylketene (Scheme 3). Notably, 2-pyridinecarboxaldehyde preferentially gave the syn-diastereoisomer 21 (74:26 syn:anti), while 4-pyridinecarboxaldehyde preferentially gave the *anti*-diastereoisomer 20 (17:83 syn:anti), consistent with the differences in syn:anti product distributions previously observed using 2- and 4nitrobenzaldehydes.³² The absolute and relative configuration within β-lactone **21** was unambiguously identified via X-ray crystallographic analysis.³³ consistent with the sense of asymmetric induction previously observed using 2-nitrobenzaldehyde.

Scheme 3. Initial reactivity employing 2- and 4-pyridinecarboxaldehydes

Subsequent studies probed the generality of this process through variation of the alkylarylketene and through the effect of substitution within the 2pyridinecarboxaldehyde (Table 3).33 Using 2-pyridinecarboxaldehyde, variation of either the aryl unit within the ketene or the alkyl chain length had little effect upon reaction stereoselectivity, with syn-β-lactones 22-25 formed in good dr and ee (entries 1-4). The effect of both electronic and steric perturbation within the 2-pyridyl motif investigated. Using methylphenylketene, 6-bromo-2was next pyridinecarboxaldehyde gave syn-β-lactone 26 (70% combined yield, 80:20 dr syn:anti, 86% ee syn), while 3-bromo-2-pyridinecarboxaldehyde displayed a reversal in diastereoselectivity (18:82 syn:anti) and excellent enantioselectivity for the major anti diastereoisomer 27 (91% ee), albeit with moderate conversion into product (entries 5 and 6).

Table 3. NHC-promoted β-lactone formation using 2-pyridinecarboxaldehydes

Entry	Ar	R	dr ^a (syn:anti)	Major product	Yield % ^b	ee % ^c (syn,anti)
1	Ph	Me	77:23	Phi N Me	78	88,86
2^d	Ph	<i>n</i> -Bu	82:18	Phu N n-Bu 23	62	84,37
3^d	4-FC ₆ H ₄	Et	82:18	F Et N	73	79,42
4	4-BrC ₆ H ₄	Et	80:20	Br Et N	45	80,–
5	Ph	Me	80:20	Phi N Br	70	86,67
6	Ph	Me	18:82	Ph. N Me Br 27	14	-,91

Ketene added dropwise as a solution in toluene. ^adr determined by ¹H NMR analysis of the crude reaction mixture. ^bIsolated yield of separable diastereoisomers. ^cee determined by HPLC analysis. ^d1.1 equiv of ketene and 1.0 equiv of aldehyde used.

The compatibility of a range of 4-pyridinecarboxaldehyde derivatives of increasing complexity with the methodology was next investigated (Table 4). The use of 3-fluoro-4-pyridinecarboxaldehyde resulted in a switch in diastereoselectivity compared with the parent 4-pyridinecarboxaldehyde, giving moderate diastereocontrol in favour of *syn*-stereoisomer **28** (*syn:anti* 63:37, entry 1). 4-Quinolinecarboxaldehyde gave *anti* β-lactone **29** with excellent diastereoselectivity (3:97 *syn:anti*) and promising

enantioselectivity (82% ee *anti*) in 57% yield (entry 2). The use of a complex 3,5-disubstituted-4-pyridinecarboxaldehyde gave β -lactone **30** with no diastereocontrol, providing readily separable *syn* and *anti*-lactones (50:50 dr) in good yield and moderate enantioselectivity (70% combined yield, 76% ee *syn*, 74% ee *anti*, entry 3).

Table 4. NHC-promoted β-lactone formation using 4-pyridinecarboxaldehyde derivatives

Ketene added dropwise as a solution in toluene. ^adr determined by ¹H NMR analysis of the crude reaction mixture. ^bIsolated yield of separable diastereoisomers. ^cee determined by HPLC analysis. ^d1.1 equiv of ketene and 1.0 equiv of aldehyde used.

Postulated Reaction Mechanism

While Kerrigan favours Lewis base addition to the aldehyde as the initial step in the phosphine-catalysed β-lactone synthesis from ketenes and benzaldehydes, in this NHC-mediated process, concurrent with the ideas of Ye, we propose initiation through NHC addition to the ketene (Scheme 4). Addition of *in situ* generated NHC 31 to ketene 32 *anti*- to the aryl unit generates azolium enolate intermediate 33. Subsequent concerted but asynchronous formal [2+2] cycloaddition with electron

deficient benzaldehydes or 2- and 4-pyridinecarboxaldehydes yields zwitterionic intermediate **34**, with subsequent catalyst regeneration and β -lactone formation. The consistency of configuration at the C(3) position of all the major diastereoisomeric β -lactone products observed herein is consistent with favourable reaction upon the *Re*-face addition of the azolium enolate **33**. However, given the obvious subtle interplay between steric and electronic factors within the aldehyde component that leads to diastereocontrol in these processes, a full rationale for the differing observed *syn*- or *anti*-selectivities with change in the aldehyde unit is at best speculative.

Scheme 4. Proposed mechanism

Scale up and derivatisation procedures

To demonstrate the synthetic utility of this methodology, this reaction of ethylphenylkene with 2-nitrobenzaldehyde could be conveniently carried out on a preparative scale with reduced NHC loadings (2.5 mol%), providing >2.5g of β -lactone 7 as a single diastereoisomer after purification and in high ee (83% yield, 95%)

ee), which could be recrystallised to enantiopurity. Using higher precatalyst **3** loading of 10 mol%, 2-pyridyl β -lactone **21** could also be obtained as a single diastereoisomer on a >3.5g scale after purification (74% yield, 82:18 dr *syn:anti*) and in good enantioselectivity for the major *syn* β -lactone (83% ee).

Scheme 5. Scale up of NHC-catalysed reactions with 2-nitrobenzaldehyde and 2-pyridinecarboxaldehyde

^aCombined isolated yield of separable diastereoisomers. ^bee determined by HPLC analysis.

Functionalisation of β -lactones 7 and 21 was subsequently achieved through ring opening with either azide or hydroxide to give the corresponding α , α -disubstituted β -amino- and β -hydroxy acid derivatives 36-39 as single diastereoisomers without loss of enantiopurity (Scheme 6). Alternatively, 21 could be treated with benzylamine to generate β -hydroxy acid amide 40 as single diastereoisomer.

Scheme 6. Derivatisation via ring-opening of β-lactones

^aee determined by HPLC analysis *via* conversion into NBn, OBn derivative. ^bee determined by HPLC analysis *via* conversion into OBn ester. ^cee determined by HPLC analysis *via* conversion into Me ester.

Finally, further complexity within the pyridyl substituted β -lactone series could be obtained *via* palladium cross coupling of β -lactone **26** bearing a bromine substituent (Scheme 7). Both Buchwald-Hartwig³⁵ and Suzuki³⁶ couplings of **26** with morpholine and indole **43** respectively provided highly complex β -lactone frameworks **41** and **42** in good yield (66% and 92% yield respectively) and with no erosion of enantiopurity.

Scheme 7. Functionalisation of β-lactones by cross-coupling

^aee determined by HPLC analysis.

Conclusion: In summary, an efficient and scalable methodology for the stereocontrolled formation of β -lactones catalyzed by chiral NHCs has been developed. The reaction of a range of disubstituted ketenes with either 2-nitrobenzaldehyde or a variety of 2- and 4-pyridinecarboxaldehydes proceeds with generally excellent levels of enantio- and diastero-selectivity. Importantly, this methodology expands the scope of the formal [2+2] cycloaddition between ketenes and aldehydes to heteroaromatic aldehydes for the first time, allowing access to highly functionalised novel structural architectures. Notably, no competing benzoin or significant formation of ketene dimerisation products were observed under the reaction conditions, with the β -lactones readily transformed into useful synthetic building blocks. Further studies focusing upon the generation and reaction of azolium enolates in NHC-mediated catalysis are underway, alongside mechanistic and kinetic investigations to advance our understanding of the reaction dynamics in these systems.

Experimental

General: All reactions were performed in flame dried glassware using anhydrous solvents. The required aldehydes where purified by Kugelrohr distillation under reduced pressure prior to use. All other reagents were obtained from commercial sources and were used without further purification. Room temperature (rt) refers to 20–25 °C, with temperatures between 0 °C and –50 °C obtained using an immersion cooler. ¹H NMR spectra were acquired at either 300, 400, or 500 MHz, ¹³C{¹H} NMR spectra were acquired at either 75, 100, or 125 MHz, and ¹⁹F{¹H} NMR spectra were acquired at 376 MHz. Chemical shifts are quoted in parts per million (ppm) relative to the residual solvent peak, coupling constants, J, are quoted in Hertz (Hz), NMR peak assignments were confirmed using 2D ¹H COSY and 2D ¹H NOESY where necessary. Infra-red spectra were recorded as thin films on either NaCl plates or KBr discs. Mass spectrometry (m/z) data were acquired using either electrospray ionisation (ESI), electron impact (EI), atmostpheric solids analysis probe (ASAP), or nanospray ionisation (NSI) using a TOF mass analyser. Optical rotations were recorded with a path length of 1 dm and concentrations, c, are quoted in g/100 mL. All chiral HPLC traces were compared with an authentic racemic trace prepared using racemic 3.

General procedure for the preparation of ketenes: A flame dried two-neck round bottom flask separated by a sintered adaptor to a second two-neck round bottom flask under an argon atmosphere is charged with anhydrous Et₂O and the appropriate acid chloride (1 equiv) before being cooled to 0 °C. Et₃N (1.1 equiv) is added dropwise over 30 min and the reaction stirred overnight at 0 °C. The solution was warmed to rt and filtered through the sintered adaptor into the second flask and concentrated. The

crude oil was transferred *via* cannula into a flame dried Kugelrohr flask and purified by distillation.

Ethylphenylketene: Prepared according to the general procedure from 2-phenylbutanoyl chloride (3.00 g, 16.4 mmol) and Et₃N (2.52 mL, 18.1 mmol) in Et₂O (45 mL). The crude oil was purified *via* Kugelrohr distillation 80–90 °C (5 mbar) {lit 70 °C (0.5 torr)}³⁷ to give ethylphenylketene (1.40 g, 60%) as light yellow oil that is stable for up to two months in the freezer under an argon atmosphere. ¹H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$ 1.24 (3H, t, *J* 7.4, CH₂CH₃), 2.44 (2H, q, *J* 7.4, CH₂CH₃), 7.02–7.10 (3H, m, Ar*H*), 7.28–7.35 (2H, m, Ar*H*).

Methylphenylketene: Prepared according to the general procedure from 2-phenylpropanoyl chloride (4.05 g, 24.0 mmol) and Et₃N (3.34 mL, 24.0 mmol) in Et₂O (50 mL). The crude oil was purified *via* Kugelrohr distillation 60–80 °C (5 mbar) {lit 50 °C, (4 torr)}³⁷ to give methylphenylketene (1.44 g, 45%) as yellow/orange oil. 1 H NMR (400 MHz, CDCl₃) $\delta_{\rm H}$ 1.92 (3H, s, C*H*₃), 6.93–6.95 (2H, m, Ar*H*), 6.97–7.01 (1H, m, Ar*H*), 7.21–7.25 (2H, m, Ar*H*).

Butylphenylketene: Prepared according to the general procedure from 2-phenylhexanoyl chloride (3.00 g, 14.2 mmol) and Me₂EtN (1.70 mL, 15.7 mmol) in Et₂O (45 mL). The crude oil was purified *via* Kugelrohr distillation 110–120 °C (5 mbar) to give butylphenylketene (1.26 g, 51%) as yellow/orange oil. ¹H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$ 0.87 (3H, t, *J* 7.2, *n*Bu*H*), 1.30–1.53 (4H, m, *n*Bu*H*), 2.32 (2H, t, *J* 7.4, *n*Bu*H*), 6.93–7.01 (3H, m, Ar*H*), 7.19–7.25 (2H, m, Ar*H*).

Isobutylphenylketene: Prepared according to the general procedure from 4-methyl-2-phenylpentanoyl chloride (3.80 g, 18.0 mmol) and Me₂EtN (2.15 mL, 20.0 mmol) in Et₂O (45 mL). The crude oil was purified *via* Kugelrohr distillation 110–117 °C (5 mbar) {lit 37–46 °C, (0.8 torr)}³⁸ to give isobutylphenylketene (1.75 g, 56%) as yellow/orange oil. 1 H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$ 0.92 (6H, d, *J* 6.6, CH(CH₃)₂), 1.66–1.83 (1H, m, CH(CH₃)₂), 2.19 (2H, d, *J* 7.0, CH₂CH(CH₃)₂), 6.95–7.01, (3H, m, Ar*H*), 7.17–7.25 (2H, m, Ar*H*).

Ethyl-4-flourophenylketene: Prepared according to the general procedure from 2-(4-fluorophenyl)butanoyl chloride (2.70 g, 13.5 mmol) and EtMe₂N (1.60 mL, 14.8 mmol) in Et₂O (40 mL). The crude oil was purified *via* Kugelrohr distillation 104–110 °C (7 mbar) to give ethyl-4-flourophenylketene (1.21 g, 55%) as yellow/orange oil. v_{max} (thin film)/cm⁻¹ 2100; ¹H NMR (300 MHz, CDCl₃) δ_H 1.13 (3H, t, *J* 7.4, CH₂CH₃), 2.33 (2H, q, *J* 7.4, CH₂CH₃), 6.87–7.97 (4H, m, Ar*H*); ¹³C{¹H} NMR (75 MHz CDCl₃): δ_C 12.9, 17.5, 41.1, 116.1 (d, *J* 21.7), 125.5 (d, *J* 7.6), 128.5 (d, *J* 3.1), 160.5 (d, *J* 243.4), 205.7.

Ethyl-4-chlorophenylketene: Prepared according to the general procedure from 2-(4-chlorophenyl)butanoyl chloride (763 mg, 3.51 mmol) and Et₃N (0.49 mL, 3.51 mmol) in Et₂O (20 mL). The crude oil was purified *via* Kugelrohr distillation 125–135 °C (7 mbar) to give ethyl-4-chlorophenylketene (0.37 g, 56%) as yellow/orange oil. 1 H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$ 1.34 (3H, t, *J* 7.4, CH₂CH₃), 2.53 (2H, q, *J* 7.4, CH₂CH₃), 7.05–7.09 (2H, m, Ar*H*), 7.37–7.42 (2H, m, Ar*H*).

Ethyl-4-bromophenylketene: Prepared according to the general procedure from 2-(4-

bromophenyl)butanoyl chloride (5.00 g, 19.1 mmol) and Et₂MeN (2.30 mL, 21.0 mmol) in Et₂O (45 mL). The crude oil was purified *via* Kugelrohr distillation 180 °C (7 mbar) to give ethyl-4-bromophenylketene (1.21 g, 28%) as yellow/orange oil. v_{max} (thin film)/cm⁻¹ 2100; ¹H NMR (300 MHz, CDCl₃) δ_{H} 1.13 (3H, t, *J* 7.4, CH₂CH₃), 2.33 (2H, q, *J* 7.4, CH₂CH₃), 6.87–7.97 (4H, m, Ar*H*); ¹³C{¹H} NMR (75 MHz CDCl₃): δ_{C} 11.7, 15.9, 40.7, 116.2, 124.4, 130.9, 131.0, 203.2.

Ethyl-4-tolylketene: Prepared according to the general procedure from 2-(p-tolyl)butanoyl chloride (2.16 g, 11.0 mmol) and Et₃N (1.53 mL, 11.0 mmol) in Et₂O (40 mL). The crude oil was purified *via* Kugelrohr distillation 110–120 °C (7 mbar) {lit 68–72 °C, (0.2 torr)}¹⁹ to give ethyl-4-tolylketene (0.92 g, 52%) as yellow/orange oil. ¹H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$ 1.29 (3H, t, J 7.4, CH₂CH₃), 2.40 (3H, s, ArCH₃), 2.50 (2H, q, J 7.4, CH₂CH₃), 7.02 (2H, d, J 8.1, ArH), 7.21 (2H, d, J 8.1, ArH).

Ethyl-4-methoxyphenylketene: Prepared according to the general procedure from 2-(4-methoxyphenyl)butanoyl chloride (3.00 g, 14.1 mmol) and Et₃N (3.93 mL, 28.2 mmol) in Et₂O (45 mL). The crude oil was purified *via* Kugelrohr distillation 140–150 °C (3 mbar) to give ethyl-4-methoxyphenylketene (1.19 g, 48%) as yellow/orange oil. v_{max} (thin film)/cm⁻¹ 2096; ¹H NMR (300 MHz, CDCl₃) δ_{H} 1.21 (3H, t, J 7.4, CH₂CH₃), 2.41 (2H, q, J 7.4, CH₂CH₃), 3.79 (3H, s, ArCH₃), 6.86–6.91 (2H, m, Ar*H*), 6.95–6.99 (2H, m, Ar*H*).

Isopropyl-3-thionylketene: Prepared according to a literature procedure³⁹ from 3-methyl-2-(thiophen-3-yl)butanoyl chloride (2.27 g, 11.2 mmol, 1 equiv) and Me₂EtN (5.46 mL, 50.4 mmol, 4.5 equiv) in THF (30 mL) at 0 °C for 10 min then rt for 4.5 h. The crude oil was purified *via* Kugelrohr distillation 85–90 °C (3 mbar) {lit 71–73 °C}³⁹ to give isopropyl-3-thionylketene (1.09 g, 59%) as yellow/orange oil. ¹H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$ 1.25 (6H, d, *J* 6.7, CH(CH₃)₂), 2.77 (1H, hep, *J* 6.7, CH(CH₃)₂), 6.79–6.82 (2H, m, Ar*H*), 7.33–7.35 (1H, m, Ar*H*).

General procedure for NHC catalysed [2+2] cycloadditions: To a flame dried Schlenk flask under an argon atmosphere was added NHC precursor 3 (0.1 equiv), KHMDS (0.5 M in toluene, 0.09 equiv) and toluene (to give 0.025 M NHC 3) and the mixture stirred for 15 min. The solution was cooled to -50 °C before the required aldehyde (1.2 equiv) was added followed by dropwise addition of a solution of the required ketene (1.0 equiv) in toluene (0.17 M) over 30 min. The solution was stirred at rt for the time stated before being opened to air and stirred for 30 min. The solution was concentrated *in vacuo* and the crude product purified by silica gel chromatography.

3-Methyl-4-(2-nitrophenyl)-3-phenyloxetan-2-one 11: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), methylphenylketene (132 mg, 1.00 mmol) and 2-nitrobenzaldehyde (181 mg, 1.20 mmol) in toluene (10 mL) at -50 °C for 3.5 h. The crude product (dr 43:57 *syn:anti*) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give *syn-11* (88 mg, 30%) as colourless solid and *anti-11* (103 mg, 35% yield) as colourless solid. **syn-11:** mp 68–70 °C; $[\alpha]_D^{20}$ –61 (c 0.15, CHCl₃); Chiral HPLC analysis Chiralpak

OJ-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (S,R): 32.0 min, t_R minor (R,S): 36.2 min, 90% ee; v_{max} (KBr)/cm⁻¹ 2925, 1834 (C=O), 1527, 1341, 1351 1246, 1133, 945, 911, 866, 790, 745, 699; ¹H NMR (400 MHz, CDCl₃) δ_H 2.20 (3H, s, CH₃), 6.22 (1H, s, CH-lactone), 7.00–7.09 (5H, m, ArH), 7.31–7.35 (1H, m, ArHNO₂) 7.55 (1H, td, J 7.5, 1.1, ArHNO₂), 7.62 (1H, d, J 7.7, ArHNO₂), 7.96 (1H, dd, J 8.2, 1.2, ArHNO₂); 13 C { 1 H} NMR (75 MHz, CDCl₃) δ_c 22.4, 67.8, 82.1, 124.9, 126.6, 126.7, 128.0, 128.2, 128.5, 129.2, 132.8, 134.2, 134.4, 146.3, 173.0; HRMS (ASAP) $C_{16}H_{14}O_4N$ [M+H]⁺ requires 284.0917; found 284.0912 (-1.9 ppm). *anti-11*: mp 128–130 °C; $[\alpha]_D^{20}$ –41 (c 0.3, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (S,S): 22.5 min, t_R minor (R,R): 30.4 min, 90% ee; v_{max} (KBr)/cm⁻¹ 2982, 1836 (C=O), 1613, 1578, 1526, 1443, 1342, 1125, 1085, 861, 790, 737, 725, 630; ${}^{1}H$ NMR (400 MHz, CDCl₃) δ_{H} 1.34 (3H, s, CH₃), 6.13 (1H, s, CH-lactone), 7.37–7.56 (5H, m, ArH), 7.60–7.64 (1H, m, ArHNO₂), 7.84 (1H, td, J 7.6, 0.9, ArHNO₂), 7.95 (1H, d, J 7.8, ArHNO₂), 8.25 (1H, dd, J 8.2, 1.0, ArHNO₂); 13 C{ 1 H} NMR (75 MHz, CDCl₃) δ_c 15.5, 65.9, 82.0, 125.5, 126.1, 128.3, 128.5, 129.1, 129.8, 132.3, 134.9, 137.6, 146.8, 173.0; HRMS (ASAP) $C_{16}H_{14}O_4N [M+H]^+$ requires 284.0904; found 284.0912 (+2.8 ppm).

(3S,4R)-3-Ethyl-4-(2-nitrophenyl)-3-phenyloxetan-2-one 7: Prepared according to the general procedure from NHC precursor **3** (28.5 mg, 0.05 mmol), KHMDS (0.09 mL, 0.045 mmol), ethylphenylketene (73.1 mg, 0.50 mmol) and 2-nitrobenzaldehyde (91 mg, 0.60 mmol) in toluene (10 mL) at -50 °C for 3.5 h. The crude product (dr 93:7 *syn:anti*) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give *syn-7* (157 mg, 83%) as off white solid. mp 58–62 °C [(\pm) *syn-7* mp 86 °C]; [α]_D²⁰ -44 (c

0.45, CHCl₃); Chiral HPLC analysis Chiralpak OJ-H (5% IPA:hexane, flow rate 0.5 mL min⁻¹, 220 nm) t_R major: (S_R) 43.7 min, t_R minor: (R_S) 51.1 min, 93% ee; v_{max} (KBr)/cm⁻¹ 3094, 1768 (C=O), 1646, 1513, 1452, 1377, 1214, 1124, 971, 883, 851, 778, 738; ¹H NMR (400 MHz, CDCl₃) δ_H 1.09 (3H, t, J 7.4, CH₂CH₃), 2.51–2.59 (2H, m, CH₂CH₃), 6.19 (1H, s, CHlactone), 6.94–7.02 (5H, m, ArH), 7.20–7.27 (1H, m, ArHNO₂), 7.38–7.48 (2H, m, ArHNO₂), 7.90 (1H, dd, J 8.2, 1.2, ArHNO₂); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_c 9.5, 29.4, 73.1, 79.4, 124.8, 126.9, 127.9, 128.3, 128.6, 129.1, 134.1, 146.6, 172.2; HRMS (EI) C₁₇H₁₆O₄N [M+H]⁺ requires 298.1074; found 298.1078 (+1.4 ppm).

(3S,4R)-3-Butyl-4-(2-nitrophenyl)-3-phenyloxetan-2-one 12: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), butylphenylketene (174 mg, 1.00 mmol) and 2-nitrobenzaldehyde (181 mg, 1.20 mmol) in toluene (10 mL) at -50 °C for 3.5 h. The crude product (dr 94:6 syn:anti) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give syn-12 (240 mg, 74%) as off white solid. mp 64–66 °C; $[\alpha]_D^{20}$ –20 (c 0.60, CHCl₃); v_{max} (KBr)/cm⁻¹ 2958, 1828 (C=O), 1527, 1346, 1117, 918, 790, 744, 701; ¹H NMR (300 MHz, CDCl₃) δ_H 0.93 (3H, t, J 7.3, nBuCH₃), 1.24–1.50 (3H, m, nBuH), 1.66–1.81 (1H, m, nBuH), 2.56 (2H, dd, J 9.3, 7.3, nBuH), 6.25 (1H, s, CH-lactone), 7.00–7.09 (5H, m, ArH), 7.28–7.33 (1H, m, ArHNO₂), 7.45–7.53 (2H, m, ArHNO₂), 7.97 (1H, dd, J 8.2, 1.0, ArHNO₂); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ_c 13.9, 22.9, 27.0, 36.1, 72.4, 79.7, 124.7, 126.7, 127.8, 128.2, 128.5, 129.0, 132.8, 134.0, 134.1, 134.1, 146.5, 172.2; HRMS (ASAP) C₁₉H₂₀O₄N [M+H]⁺ requires 326.1387; found 326.1381 (–1.8 ppm). syn-12 was ring opened with NaN₃ and converted into its

benzyl ester to allow chiral HPLC analysis: Chiral HPLC analysis Chiralpak AD-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R minor: 4.9 min, t_R major: 6.4 min, 89% ee.

(3S,4R)-3-Isobutyl-4-(2-nitrophenyl)-3-phenyloxetan-2-one 13: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 0.09 mmol), iso-butylphenylketene (167 mg, 1.00 mmol) and 2nitrobenzaldehyde (181 mg, 1.20 mmol) in toluene (10 mL) at -50 °C for 6 h. The crude product (dr 94:6 syn:anti) was purified by silica gel chromatography(95:5 petrol:Et₂O) to give syn-13 (95 mg, 30%) as off white solid. mp 78–80 °C; $[\alpha]_D^{20}$ 0.0 (c 0.3, CHCl₃); Chiral HPLC analysis Chiralpak OJ-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R : 10.6 min, t_R : 12.3 min, <10% ee; v_{max} (KBr)/cm⁻¹ 2958, 1819 (C=O), 1613, 1582, 1521, 1448, 1341, 1243, 1118, 974, 904, 788, 842, 706, 642, 534; ¹H NMR (300 MHz, CDCl₃) δ_H 0.82 (3H, d, J 6.7, i-BuCH₃), 1.06 (3H, d, J 6.7, i-BuCH₃), 1.90 (1H, hep, J 6.6, i-BuH) 2.44 (1H, dd, J 14.6, 6.4, i-BuCH₂), 2.65 (1H, dd, J 14.6, 6.4, i-BuCH₂), 6.20 (1H, s, CH-lactone), 7.00-7.11 (5H, m, ArH), 7.26–7.34 (1H, m, ArHNO₂), 7.41–7.46 (2H, m, ArHNO₂), 7.97 (1H, dt, J 8.1, 0.7, Ar HNO_2); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ_c 23.8, 23.9, 25.4, 45.3, 72.2, 80.1, 124.7, 126.7, 127.7, 128.2, 128.5, 129.0, 132.6, 134.0, 134.3, 146.8, 172.0; HRMS (ASAP) $C_{19}H_{20}O_4N$ [M+H]⁺ requires 326.1387; found 326.1382 (-1.5 ppm).

(3S,4R)-3-Ethyl-3-(4-fluorophenyl)-4-(2-nitrophenyl)oxetan-2-one 14: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethyl-4-flourophenylketene (211 mg, 1.20 mmol) and 2-nitrobenzaldehyde (151 mg, 1.00 mmol) in toluene (10 mL) at -50 °C for 3 h.

The crude product (dr 93:7 *syn:anti*) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give *syn-***14** (216 mg, 69%) as colourless solid. mp 102–104 °C; $[\alpha]_D^{20}$ –19 (c 0.85, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R minor (R,S): 11.1 min, t_R major (S,R): 12.3 min, 85% ee; v_{max} (KBr)/cm⁻¹ 2971, 1831 (C=O), 1614, 1605, 1522, 1319, 1353, 1310, 1299, 1225, 1167, 1144, 1118, 1017, 949, 894, 859, 816, 787, 745, 682, 654, 609, 536, 506; 1 H NMR (400 MHz, CDCl₃) δ_H 1.15 (3H, t, J 7.4, CH₂CH₃), 2.53–2.68 (2H, m, CH₂CH₃), 6.23 (1H, s, C*H*-lactone), 6.74–7.81 (2H, m, 4-FAr*H*), 7.00–7.07 (2H, m, 4-FAr*H*), 7.33–7.38 (1H, m, Ar*H*NO₂), 7.53 (2H, dd, J 1.0, 4.9, Ar*H*NO₂), 8.00 (1H, d, J 8.1, Ar*H*NO₂); $^{19}F\{^1$ H} NMR (376 MHz, CDCl₃) δ_F –114; $^{13}C\{^1$ H} NMR (100 MHz, CDCl₃) δ_c 9.5, 29.6, 72.4, 79.5, 115.6 (d, J 21.4), 125.0 128.2, 128.7 (d, J 8.2), 129.4, 129.7 (d J 3.1) 132.7, 134.3, 146.6, 162.1 (d, J, 248), 172.0; HRMS (ASAP) $C_{17}H_{15}FO_4$ N [M+H]⁺ requires 316.0980; found 316.0974 (–1.8 ppm).

(3S, 4R)-3-(4-Chlorophenyl)-3-ethyl-4-(2-nitrophenyl)oxetan-2-one 15: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethyl-4-chlorophenylketene (181 mg, 1.00 mmol) and 2-nitrobenzaldehyde (181 mg, 1.20 mmol) in toluene (10 mL) at -50 °C for 3 h h. The crude product (dr 93:7 *syn:anti*) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give *syn-*15 (262mg, 79%) as off white solid. mp 78–80 °C; $[\alpha]_D^{20}$ –15 (c 0.5, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (5% IPA:hexane, flow rate 0.5 mL min⁻¹, 220 nm) t_R minor (R,S): 22.5 min, t_R major (S,R): 24.9 min, 86% ee; V_{max} (KBr)/cm⁻¹ 2975, 1822 (C=O), 1614, 1525, 1497, 1343, 1250, 1130, 1014, 939, 900, 860, 840, 792, 745, 685; ¹H NMR (300 MHz, CDCl₃) δ_{H} 1.14 (3H, t, J 7.4, CH₂CH₃),

2.52–2.67 (2H, m, CH_2CH_3), 6.22 (1H, s, CH-lactone), 6.97–7.08 (4H, m, 4-ClArH), 7.33–7.41 (1H, m, Ar HNO_2), 7.50–7.55 (2H, m, Ar HNO_2), 7.98–8.04 (1H, m, Ar HNO_2); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ_c 9.4, 29.6, 72.4, 79.3, 125.0, 128.2, 128.3, 128.9, 129.5, 132.5 (2×C), 134.0, 134.4, 146.7, 171.7; HRMS (ASAP) $C_{17}H_{15}ClO_4N$ [M+H]⁺ requires 332.0684; found 332.0681 (–0.9 ppm).

(3S,4R)-3-(4-Bromophenyl)-3-ethyl-4-(2-nitrophenyl)oxetan-2-one : Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethyl-4-bromophenylketene (225 mg, 1.00 mmol) and 2-nitrobenzaldehyde (181 mg, 1.20 mmol) in toluene (10 mL) at -50 °C for 3 h. The crude product (dr 94:6 syn:anti) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give syn-16 (257mg, 68%) as pale brown solid. mp 102–104 °C; $[\alpha]_{p}^{20}$ -14 (c 0.5, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (5% IPA:hexane, flow rate 0.2 mL min⁻¹, 220 nm) t_R minor (R,S): 56.8 min, t_R major (S,R): 61.7 min, 92% ee; v_{max} (KBr)/cm⁻¹ 2974, 1821 (C=O), 1613, 1526, 1494, 1342, 1249, 1129, 1077, 1011, 957, 938, 899, 860, 839, 791, 744, 721, 652, 517; ¹H NMR (300 MHz, CDCl₃) δ_H 1.14 (3H, t, J 7.4, CH₂CH₃), 2.52–2.66 (2H, m, CH₂CH₃), 6.22 (1H, s, CH-lactone), 6.91–6.96 (2H, m, 4-BrAr*H*), 7.19–7.23 (2H, m, 4-BrAr*H*), 7.34–7.41 (1H, m, $ArHNO_2$), 7.50–7.55 (2H, m, $ArHNO_2$), 8.00–8.04 (1H, m, $ArHNO_2$); $^{13}C\{^1H\}$ NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta_c 9.4, 29.6, 72.5, 79.3, 122.2, 125.0, 128.2, 128.6, 129.5, 131.8,$ 132.5, 133.1, 134.5, 146.7, 171.7; HRMS (ASAP) C₁₇H₁₅BrO₄N [M+H]⁺ requires 376.0184; found 376.0176 (-0.8 ppm).

(3S,4R)-3-Ethyl-4-(2-nitrophenyl)-3-(p-tolyl)oxetan-2-one 17: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethyl-4-tolylketene (167 mg, 1.00 mmol) and 2-nitrobenzaldehyde (181 mg, 1.20 mmol) in toluene (10 mL) at -50 °C for 3 h. The crude product (dr 90:10 syn:anti) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give syn-17 (162 mg, 52%) as off white solid. mp 100–102 °C; $[\alpha]_D^{20}$ –39 (c 0.25, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 mm) t_R minor (R,S): 9.6 min, t_R major (S,R): 10.5 min, 83% ee; v_{max} (KBr)/cm⁻¹ 2975, 1825 (C=O), 1612, 1526, 1341, 1236, 1134, 1104, 951, 937, 890, 858, 837, 742, 722, 697, 520; ¹H NMR (300 MHz, CDCl₃) δ_H 1.14 (3H, t, J 7.4, CH₂CH₃), 2.14 (3H, s, ArCH₃), 2.52–2.66 (2H, m, CH₂CH₃), 6.23 (1H, s, CH-lactone), 6.82–6.93 (4H, m, 4-tolylArH), 7.31 (1H, ddd, J 8.5, 6.8, 1.8, ArHNO₂), 7.47–7.55 (2H, m, ArHNO₂), 7.98 (1H, dd, J 8.3, 1.1, ArHNO₂); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_c 9.5, 21.0, 29.5, 72.7, 79.6, 124.8, 126.7, 128.3, 129.1, 129.3, 130.7, 133.0, 134.2, 137.6, 146.5, 172.4; HRMS (ASAP) C₁₈H₁₈O₄N [M+H]⁺ requires 312.1230; found 312.1226 (–1.4 ppm).

(3S, 4R)-3-Ethyl-3-(4-methoxyphenyl)-4-(2-nitrophenyl)oxetan-2-one 18: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethyl-4-methoxyphenylketene (211 mg, 1.20 mmol) and 2-nitrobenzaldehyde (151 mg, 1.00 mmol) in toluene (10 mL) at -50 °C for 3 h. The crude product (dr 86:14 *syn:anti*) was purified by silica gel chromatography (90:10 petrol:Et₂O) to give *syn-*18 (247 mg, 75%) as off white solid. mp 58–64 °C; $[\alpha]_D^{20}$ -57 (c 0.25, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (10% IPA:hexane, flow rate 0.8 mL min⁻¹, 220 nm) t_R major (S, S): 18.5 min, t_R minor (S, S): 20.4 min,

86% ee; v_{max} (KBr)/cm⁻¹ 2974, 1828 (C=O), 1612, 1578, 1527, 1465, 1300, 1254, 1185, 1109, 945, 895, 837, 791, 742, 699, 686 ¹H NMR (300 MHz, CDCl₃) δ_H 1.13 (3H, t, *J* 7.4, CH₂C*H*₃), 2.51–2.64 (2H, m, C*H*₂CH₃), 3.63 (3H, s, ArOC*H*₃), 6.21 (1H, s, C*H*-lactone), 6.57 (2H, d, *J* 8.9, 4-MeOAr*H*), 6.93 (2H, d, *J* 8.9, 4-MeOAr*H*), 7.31 (1H, ddd, *J* 8.6, 6.7, 2.1, Ar*H*NO₂), 7.47–7.55 (2H, m, Ar*H*NO₂), 7.96 (1H, dd, *J* 8.3, 0.9, Ar*H*NO₂); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_c 9.4, 29.4, 55.1, 72.3, 79.7, 113.9, 124.8, 125.7, 128.0, 128.2, 129.0, 133.0, 134.2, 146.4, 158.9, 172.4; HRMS (ASAP) C₁₈H₁₈O₅N [M+H]⁺ requires 328.1179; found 328.1175 (–1.8 ppm).

3-Isopropyl-4-(2-nitrophenyl)-3-(thiophen-3-yl)oxetan-2-one 19: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), isopropyl-3-thionylketene (180 mg, 1.00 mmol) and 2-nitrobenzaldehyde (181 mg, 1.20 mmol) in toluene (10 mL) at 0 °C to rt over 12 h. The crude product (dr 45:55 syn:anti) was purified by silica gel chromatography (97:3 petrol:Et₂O) to give syn-19 (73 mg, 23%) as pale brown solid and anti-19 (38 mg, 12%) as off white solid. syn-19: mp 114 °C; $[\alpha]_D^{20}$ –24 (c 0.15, CHCl₃); Chiral HPLC analysis Chiralpak OJ-H (2% IPA:hexane, flow rate 0.5 mL min⁻¹, 220 nm) t_R major (S,R): 44.0 min, t_R minor (R,S): 49.8 min, 58% ee; v_{max} (KBr)/cm⁻¹ 2978, 1825 (C=O), 1613, 1577, 1522, 1342, 1244, 953, 896, 857, 784, 744, 705, 688, 622; ¹H NMR (300 MHz, CDCl₃) δ_H 0.96 (3H, d, J 6.8, i-PrCH₃), 1.45 (3H, d, J 6.8, i-PrCH₃), 2.87 (1H, hep, J 6.8, i-PrH), 6.37 (1H, s, CH-lactone), 6.56 (1H, dd, J 4.9, 1.5, thiopheneArH), 6.97–7.01 (2H, m, thiopheneArH), 7.30–7.38 (1H, m, ArHNO₂), 7.50 (2H, d, J 3.9, ArHNO₂), 8.01 (1H, dt, J 8.1, 0.8, ArHNO₂); 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ_c 18.4, 19.0, 32.8, 74.5, 76.9, 123.0, 124.7, 125.2, 126.7, 128.3, 129.1, 133.2, 134.2,

135.0, 146.6, 171.2; HRMS (ASAP) $C_{16}H_{16}O_4NS$ [M+H]⁺ requires 318.0795; found 318.0791 (-1.1 ppm); *anti*-19: mp 138–142 °C; $[\alpha]_D^{20}$ +40 (*c* 0.1, CHCl₃); Chiral HPLC analysis Chiralpak OJ-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (*S*,*S*): 28.8 min, t_R minor (*R*,*R*): 43.2 min, 92% ee; v_{max} (KBr)/cm⁻¹ 2974, 1818 (C=O), 1613, 1535, 1469, 1389, 1344, 1261, 1227, 1192, 1143, 1121, 995, 955, 902, 883, 863, 831, 790, 744, 710, 686, 638; ¹H NMR (300 MHz, CDCl₃) δ_H 0.41 (3H, d, *J* 6.7, *i*-PrCH₃), 0.93 (3H, d, *J* 6.8, *i*-PrCH₃), 2.19 (1H, hep, *J* 6.7, *i*-PrH) 6.32 (1H, s, CH-lactone), 7.40 (1H, dd, *J* 5.1, 3.0, thiopheneArH), 7.46 (1H, dd, *J* 5.1, 1.3, thiopheneArH), 7.61-7.68 (2H, m, thiopheneArH and ArHNO₂), 7.82 (1H, tdd, *J* 7.9, 1.5, 0.4, ArHNO₂), 7.93 (1H, dd, *J* 8.0, 1.3, ArHNO₂), 8.22 (1H, dd, *J* 8.2, 1.3, ArHNO₂); ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ_c 16.8, 18.2, 29.3, 71.9, 79.0, 124.4, 125.5, 125.6, 128.5, 129.8, 130.1, 131.8, 133.8, 134.2, 148.3, 171.8; HRMS (ASAP) $C_{16}H_{16}O_4NS$ [M+H]⁺ requires 318.0795; found 318.0788 (-2.1 ppm).

3-Ethyl-3-phenyl-4-(pyridin-4-yl)oxetan-2-one **20**: Prepared according to the general procedure from NHC precursor **3** (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethylphenylketene (175 mg, 1.2 mmol) and 4-pyridinecarboxaldehyde (94 μL, 1.0 mmol) in toluene (20 mL) at -50 °C for 2 h. The crude product (dr 17:83, *syn:anti*) was purified by silica gel chromatography (50:50 petrol:Et₂O) to give *anti*-**20** (174 mg, 67%) as colourless solid and *syn-***20** (21 mg, 8% yield) as colourless solid. *anti-***20**: mp 74–82 °C; $[\alpha]_D^{20}$ +18.0 (*c* 1.00, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ_H 1.08 (3H, t, *J* 7.4, CH₂CH₃), 2.30–2.37 (1H, m, CH₄H_BCH₃), 2.38–2.45 (1H, m, CH₄H_BCH₃), 5.51 (1H, s, CH(Ar)), 6.97–6.98 (2H, m, PhH-4, PyH-5), 7.02–7.03 (2H, d, *J* 6.0, PhH), 7.08-7.15 (3H, m, PhH), 8.41 (1H, d, *J* 4.9, PyH-6); ¹³C{¹H} NMR

(125 MHz, CDCl₃) δ_c 9.4, 31.3, 72.1, 81.2, 121.5, 126.9, 127.9, 128.8, 133.7, 144.6, 149.4, 171.3; Chiral HPLC analysis Chiralpak AS-H (10% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm, 30 °C) t_R major (*S*,*S*): 8.61 min, t_R minor (*R*,*R*): 12.0 min, 78% ee; v_{max} (ATR)/cm⁻¹ 2974, 1825 (C=O), 1599, 1414, 1107, 1086, 957, 907, 827; HRMS (ESI) $C_{16}H_{16}O_2N$ [M+H]⁺ requires 254.1176; found 254.1178; *syn-20*: mp 66–70 °C; $[\alpha]_D^{20}$ 0.0 (*c* 0.01, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm) t_R major (*S*,*R*): 15.5 min, t_R minor (*R*,*S*): 18.1 min, 91% ee; v_{max} (ATR)/cm⁻¹ 2974, 1829 (C=O), 1603, 1418, 1242, 1098, 947, 901; ¹H NMR (500 MHz, CDCl₃) δ_H 0.69 (3H, t, *J* 7.4, CH₂CH₃), 1.45–1.53 (1H, m, CH₄H_BCH₃), 1.64–1.71 (1H, m, CH_AH_BCH₃), 5.63 (1H, s, CH(Ar)), 7.37–7.48 (7H, m, PhH, PyH-3,4,5), 8.74 (1H, d, *J* 5.8, PyH-6); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ_C 8.5, 27.5, 69.1, 81.2, 120.7, 126.3, 128.3, 129.3, 136.8, 144.1, 150.2, 171.0; HRMS (ESI) $C_{16}H_{16}O_2N$ [M+H]⁺ requires 254.1176; found 254.1178.

3-Ethyl-3-phenyl-4-(pyridin-2-yl)oxetan-2-one 21: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethylphenylketene (146 mg, 1.00 mmol) and 2-pyridinecarboxaldehyde (114 μL, 1.20 mmol) in toluene (20 mL) at –50 °C for 3 h. The crude product (dr 74:26, syn:anti) was purified by silica gel chromatography (95:5 to 90:10 petrol:EtOAc) to give syn-21 (140 mg, 55%) as pale brown solid and anti-21 (38 mg, 15% yield) as pale brown oil. syn-21: mp 72–76 °C; $[\alpha]_D^{20}$ –63.7 (c 0.51, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R minor (R,R): 8.6 min, t_R major (S,S): 9.3 min, 82% ee; ν_{max} (ATR)/cm⁻¹1819 (C=O), 1591, 1453, 1105, 1094, 955, 893; ¹H NMR (500 MHz, CDCl₃) δ_H 1.07 (3H, t, J 7.4,

CH₂CH₃), 2.28–2.35 (1H, m, CH_AH_BCH₃), 2.37–2.44 (1H, m, CH_AH_BCH₃), 5.71 (1H, s, CH(Ar)), 7.00–7.09 (7H, m, PhH, PyH-3,5), 7.41 (1H, td, J 7.7, 1.5, PyH-4), 8.46 (1H, dd, J 4.8, 0.6, PyH-6); 13 C{ 1 H} NMR (125 MHz, CDCl₃) δ_{c} 9.3, 31.7, 71.8, 82.5, 121.4, 123.3, 127.0, 127.3, 128.3, 134.6, 136.6, 148.8, 155.5, 172.1; HRMS (ESI) C₁₆H₁₆O₂N [M+H]⁺ requires 254.1176; found 254.1176; *anti-21*: $[\alpha]_{D}^{20}$ 0.0 (c 0.26, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major ($S_{c}R$): 8.82 min, t_R minor ($S_{c}R$): 10.4 min, 47% ee; V_{max} (ATR)/cm⁻¹ 2974, 1822 (C=O), 1591, 1437, 1110, 1088, 959, 914, 764; $S_{c}R$ H NMR (500 MHz, CDCl₃) $S_{c}R$ H 0.64 (3H, t, J 7.4, CH₂CH₃), 1.52–1.59 (1H, m, CH_AH_BCH₃), 1.61-1.68 (1H, m, CH_AH_BCH₃), 5.77 (1H, s, CH(Ar)), 7.31–7.36 (2H, m, PhH-4, PyH-5), 7.45 (2H, t, J 7.7, PhH), 7.59 (1H, d, J 7.8, PyH-3), 7.64 (1H, d, J 7.3, PhH), 7.83 (1H, td, J 7.7, 1.4, PyH-4), 8.70 (1H, d, J 4.8, PyH-6); $S_{c}R$ H NMR (125 MHz, CDCl₃) $S_{c}R$ H RMS (ESI) C₁₆H₁₆O₂N [M+H]⁺ requires 254.1176; found 254.1179.

3-Methyl-3-phenyl-4-(pyridin-2-yl)oxetan-2-one 22: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), methylphenylketene (132 mg, 1.00 mmol) and 2-pyridinecarboxaldehyde (114 μ L, 1.20 mmol) in toluene (20 mL) at -50 °C for 3 h. The crude product (dr 77:23 syn:anti) was purified by silica gel chromatography (95:5 to 90:10 petrol:EtOAc) to give syn-22 (144 mg, 60%) as colourless solid and anti-22 (42 mg, 18%) as colourless oil. syn-22: mp 94–98 °C; $[\alpha]_D^{20}$ -37.1 (c 0.49, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R minor (R,R): 15.0 min, t_R major (S,S): 17.2 min, 88% ee; V_{max} (ATR)/cm⁻¹1821 (C=O), 1591, 1453,

1257, 1099, 952, 754; ¹H NMR (400 MHz, CDCl₃) $\delta_{\rm H}$ 2.00 (3H, s, C H_3), 5.68 (1H, s, CH(Ar)), 7.02–7.10 (7H, m, PhH, PyH-3,5), 7.43 (1H, td, J 7.7, 1.4, PyH-4), 8.44 (1H, dd, J 5.1, 1.3, PyH-6); ¹³C { ¹H } NMR (125 MHz, CDCl₃) $\delta_{\rm c}$ 24.7, 66.9, 84.3, 121.0, 123.2, 126.6, 127.4, 128.3, 135.8, 136.5, 148.9, 155.3, 172.7; HRMS C₁₅H₁₄O₂N (ESI) [M+H]⁺ requires 240.1019; found 240.1022; *anti*-22: $[\alpha]_{\rm D}^{20}$ –21.6 (c 0.26, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) $t_{\rm R}$ major ($S_{\rm R}$): 11.6 min, $t_{\rm R}$ minor ($R_{\rm R}$): 13.3 min, 86% ee; $v_{\rm max}$ (ATR)/cm⁻¹ 1828 (C=O), 1591, 1437, 1140, 1086, 995, 961, 768; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 1.24 (3H, s, C H_3), 5.80 (1H, s, C H_3), 7.31–7.36 (2H, m, Ph H_3 -4, Py H_3 -5), 7.45 (2H, t, H_3 -7, Ph H_3), 7.58 (1H, d, H_3 -7, Py H_3 -6); 13C (¹H } NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 21.4, 65.1, 81.9, 120.4, 123.3, 126.0, 127.9, 129.2, 137.1, 139.7, 149.8, 155.5, 172.6; HRMS (ESI) C₁₅H₁₄O₂N [M+H]⁺ requires 240.1019; found 240.1022.

3-Butyl-3-phenyl-4-(pyridin-2-yl)oxetan-2-one **23**: Prepared according to the general procedure from NHC precursor **3** (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol) butylphenylketene (174 mg, 1.00 mmol) and 2-pyridinecarboxaldehyde (114 μL, 1.20 mmol) in toluene (20 mL) at -50 °C for 3 h. The crude product (dr 82:18, syn:anti) was purified by silica gel chromatography (95:5 to 90:10 petrol:EtOAc) to give syn-**23** (128 mg, 45%) as colourless solid and anti-**23** (47 mg, 17%) as colourless oil. syn-**23**: $[\alpha]_D^{20}$ -41.7 (c 0.36, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (S,S): 8.5 min, t_R minor (R,R): 12.0 min, 84% ee; v_{max} (ATR)/cm⁻¹ 2959, 1825 (C=O), 1591, 1439, 1259, 1109, 930, 761; ¹H NMR (300 MHz, CDCl₃) δ_H 0.87 (3H, t, J 7.2, CH₃), 1.15–1.43 (3H, m, n-

Bu*H*), 1.57–1.71 (1H, m, *n*-Bu*H*), 2.21–2.38 (2H, m, *n*-Bu*H*), 5.70 (1H, s, *CH*(Ar)), 6.98–7.11 (7H, m, Ph*H*, Py*H*-3,4,5), 7.40 (1H, td, *J* 7.8, 1.7, Py*H*-4), 8.46 (1H, ddd, *J* 4.8, 1.6, 0.9, Py*H*-6); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_c 13.9, 22.8, 26.9, 38.4, 71.2, 82.7, 121.3, 123.3, 127.0, 127.3, 128.3, 134.9, 136.5, 148.9, 155.5, 172.2; HRMS (ESI) C₁₈H₂₀NO₂ [M+H⁺ requires 282.1494; found 282.1492; *anti*-23: [α]₀²⁰ +2.4 (*c* 0.21, CHCl₃); Chiral HPLC analysis Chiralpak OD-H (1% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (*S*,*R*): 6.8 min, t_R minor (*R*,*S*): 9.5 min, 37% ee; ν_{max} (ATR)/cm⁻¹ 2957, 1825 (C=O), 1589, 1437, 1259, 1109, 1090, 997, 934, 775; ¹H NMR (500 MHz, CDCl₃) δ_H 0.61 (3H, s, *CH*₃), 0.75–1.03 (3H, m, *n*-Bu*H*), 1.14–1.23 (1H, m, *n*-Bu*H*), 1.48–1.57 (2H, m, *n*-Bu*H*), 5.76 (1H, s, *CH*(Ar)), 7.31–7.36 (2H, m, Ph*H*-4, Py*H*-5), 7.44 (2H, t, *J* 7.7, Ph*H*), 7.58 (1H, d, *J* 7.8, Py*H*-3), 7.63 (2H, d, *J* 7.9, Ph*H*), 7.83 (1H, td, *J* 7.7, 1.5, Py*H*-4), 8.71 (1H, d, *J* 4.7, Py*H*-6); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_c 13.7, 22.6, 26.0, 34.3, 68.7, 82.2, 120.8, 123.4, 126.8, 127.8, 128.9, 137.0, 138.0, 149.7, 155.5, 172.0; HRMS (ESI) C₁₈H₂₀NO₂ [M+H]⁺ requires 282.1494; found 282.1493.

3-Ethyl-3-(4-fluorophenyl)-4-(pyridin-2-yl)oxetan-2-one **24**: Prepared according to the general procedure from NHC precursor **3** (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethylphenylketene (164 mg, 1.00 mmol) and 2-pyridinecarboxaldehyde (114 μ L, 1.20 mmol) in toluene (20 mL) at -50 °C for 3 h. The crude product (dr 82:18, *syn:anti*) was purified by silica gel chromatography (90:10 to 80:20 petrol:EtOAc) to give *syn-24* (159 mg, 59%) as colourless solid and *anti-24* (38 mg, 14%) as colourless solid. *syn-24*: mp 82–84 °C; $[\alpha]_D^{20}$ –57.8 (*c* 0.71, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL

 \min^{-1} , 220 nm) t_R major (S,S): 9.9 min, t_R minor (R,R): 13.5 min, 79% ee; v_{max} (ATR)/cm⁻¹1819 (C=O), 1512, 1438, 1229, 1113, 897, 840, 795, 767; ¹H NMR (300 MHz, CDCl₃) δ_H 1.05 (3H, t, J 7.4, CH₃), 2.21–2.33 (1H, m, CH₄H_BCH₃), 2.33-2.45 (1H, m, $CH_AH_BCH_3$), 5.68 (1H, s, CH(Ar)), 6.71–6.79 (2H, m, PhH-3.5), 6.97–7.09 (4H, m, Ph*H*-2,6, Py*H*-3,5), 7.45 (1H, td, *J* 7.8, 1.7, Py*H*-4), 8.46 (1H, ddd, *J* 4.8, 1.6, 0.6, PyH-6); ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃) δ_c 9.3, 31.8, 71.1, 82.3, 115.2 (d, J 21.4), 121.2, 123.4, 128.8 (d, J 8.1), 130.4 (d, J 3.0), 136.7, 148.9, 155.4, 161.8 (d, J 246), 171.9; HRMS (ESI) $C_{16}H_{15}O_2NF$ [M+H]⁺ requires 272.1081; found 272.1086; anti-24: mp 82–84 °C; $[\alpha]_D^{20}$ 0.0 (c 0.26, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (S,R): 9.51 min, t_R minor (R,S): 10.4 min, 42% ee; v_{max} (ATR)/cm⁻¹ 1827 (C=O), 1591, 1508, 1437, 1219, 1113, 1088, 961, 917, 838; ¹H NMR (300 MHz, CDCl₃) δ_H 0.63 (3H, t, J 7.4, CH₃), 1.45-1.56 (1H, m, $CH_4H_BCH_3$), 1.57-1.68 (1H, m, $CH_AH_BCH_3$), 5.72 (1H, s, CH(Ar)), 7.09–7.17 (2H, m, PhH-3,5), 7.30–7.34 (1H, m, PyH-5), 7.55–7.65 (3H, m, PyH-3, PhH-2,6), 7.82 (1H, td, J 7.7, 1.7, PyH-4), 8.67–8.70 (1H, m, PyH-6); 13 C{ 1 H} NMR (75 MHz, CDCl₃) δ_{c} 8.4, 28.1, 68.4, 82.1, 115.8 (d, J 21.5), 120.6, 123.4, 128.7 (d, J 8.1), 133.2 (d, J 3.1), 137.1, 149.7, 155.3, 162.4 (d, J 246), 171.6; HRMS (ESI) $C_{16}H_{15}O_2NF [M+H]^+$ requires 272.1081; found 272.1086.

(3S,4S)-3-(4-Bromophenyl)-3-ethyl-4-(pyridin-2-yl)oxetan-2-one **25**: Prepared according to the general procedure from NHC precursor **3** (15.0 mg, 0.026 mmol), KHMDS (47 μ L, 0.024 mmol), ethyl-4-bromophenylketene (59 mg, 0.26 mmol) and 2-pyridinecarboxaldehyde (30 μ L, 0.32 mmol) in toluene (5 mL) at -50 °C for 3 h. The crude product (dr 80:20, *syn:anti*) was purified by silica gel chromatography

(95:5 to 90:10 petrol:EtOAc) to give syn-25 (39 mg, 45%) as colourless solid. [α]_D²⁰ -40 (c 0.09, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R minor (R,R): 14.1 min, t_R major (S,S): 15.8 min, 80% ee; v_{max} (ATR)/cm⁻¹ 2974, 1819 (C=O), 1589, 1489, 1115, 1099, 1076, 1009, 943, 901, 843, 824; ¹H NMR (300 MHz, CDCl₃) δ_H 1.06 (3H, t, J 7.4, CH_3), 2.21–2.32 (1H, m, $CH_AH_BCH_3$), 2.33–2.44 (1H, m, $CH_AH_BCH_3$), 5.69 (1H, s, CH(Ar)), 6.89–6.94 (2H, m, ArH), 7.05 (1H, d, J 7.9, PyH-3), 7.10 (1H, ddd, J 7.9, 4.9, 0.9, PyH-5), 7.18–7.23 (2H, m, ArH), 7.48 (1H, td, J 7.9, 1.7, PyH-4), 8.47–8.49 (1H, m, PyH-6); $^{13}C\{^1H\}$ NMR (75 MHz, CDCl₃) δ_c 9.3, 31.8, 71.2, 82.2, 121.3, 121.6, 123.5, 128.8, 131.4, 133.8, 136.8, 149.0, 155.3, 171.6; HRMS (ESI) $C_{16}H_{15}O_2NBr$ [M+H]⁺ requires 332.0281; found 332.0287.

4-(6-Bromopyridin-2-yl)-3-methyl-3-phenyloxetan-2-one **26**: Prepared according to the general procedure from NHC precursor **3** (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), methylphenylketene (145 mg, 1.10 mmol) and 6-bromo-2-pyridinecarboxaldehyde (186 mg, 1.00 mmol) in toluene (20 mL) at –50 °C for 3 h. The crude product (dr 80:20, *syn:anti*) that was purified by silica gel chromatography (5% Et₂O:petrol) to give *syn-***26** (175 mg, 55%) as colourless solid and *anti-***26** (47 mg, 15%) as pale yellow oil. *syn-***26**: mp 112–118 °C; $[\alpha]_D^{20}$ –24.9 (*c* 1.00, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (*S,S*): 13.2 min, t_R minor (*R,R*): 15.9 min, 86% ee; v_{max} (ATR)/cm⁻¹1811 (C=O), 1557, 1437, 1416, 1260, 1119, 962, 878, 791; ¹H NMR (500 MHz, CDCl₃) δ_H 2.00 (3H, s, CH₃), 5.64 (1H, s, CH(Ar)), 7.00 (1H, d, *J* 7.5, PyH-3), 7.05–7.09 (3H, m, PhH), 7.11–7.15 (2H, m, *J* 7.2, PhH), 7.22 (1H, d, *J* 7.8, PyH-5), 7.28 (1H, t, *J* 7.8,

Py*H*-4); 13 C{ 1 H} NMR (125 MHz, CDCl₃) δ_{c} 24.6, 67.5, 83.3, 119.7, 126.6, 127.6, 127.7, 128.5, 135.4, 138.8, 141.1, 156.9, 172.3; HRMS (ESI) C₁₅H₁₃O₂NBr [M+H]⁺ requires 318.0124; found 318.0132; *anti*-26: [α]_D²⁰ +24.0 (*c* 0.20, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (*S*,*R*): 12.8 min, t_R minor (*R*,*S*): 16.0 min, 67% ee; v_{max} (ATR)/cm⁻¹ 2974, 1832 (C=O), 1582, 1558, 1437, 1406, 1123, 1082, 986, 972, 781; 1 H NMR (500 MHz, CDCl₃) δ_{H} 1.28 (3H, s, C*H*₃), 5.73 (1H, s, C*H*(Ar)), 7.35 (1H, t, *J* 7.4, Ph*H*-4), 7.45 (2H, t, *J* 7.7, Ph*H*), 7.51 (1H, d, *J* 7.9, Py*H*-5), 7.54 (1H, d, *J* 7.6, Py*H*-3), 7.61 (2H, d, *J* 7.5, Ph*H*), 7.69 (1H, t, *J* 7.8, Py*H*-4); 13 C{ 1 H} NMR (125 MHz, CDCl₃) δ_{c} 21.5, 65.5, 81.2, 119.3, 126.0, 127.9, 128.1, 129.2, 139.2, 139.4, 142.3, 156.9, 172.1; HRMS (ESI) C₁₅H₁₃O₂NBr [M+H]⁺ requires 318.0124; found 318.0129.

(3S,4R)-4-(3-Bromopyridin-2-yl)-3-methyl-3-phenyloxetan-2-one 27: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), methylphenylketene (145 mg, 1.10 mmol) and 3-bromo-2-pyridinecarboxaldehyde (186 mg, 1.00 mmol) in toluene (20 mL) at -50 °C for 3 h. The crude product (dr 18:82 *syn:anti*) was purified by silica gel chromatography (80:20 petrol:EtOAc) to give *anti-27* (45 mg, 14% yield) as pale brown solid. mp 118–122 °C; $[\alpha]_D^{20}$ +26.5 (*c* 0.50, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm, 30 °C) t_R major (*S,R*): 19.3 min, t_R minor (*R,S*): 21.0 min, 91% ee; ν_{max} (ATR)/cm⁻¹ 1819 (C=O), 1427, 1377, 1132, 1090, 959, 700; ¹H NMR (400 MHz, CDCl₃) δ_H 1.38 (3H, s, CH₃), 6.05 (1H, s, CH(Ar)), 7.25 (1H, dd, *J* 8.1, 4.6, PyH-5), 7.34–7.40 (1H, m, PhH-4), 7.42–7.47 (2H, m, PhH-3,5), 7.58–7.62 (2H, m, PhH-2,6), 7.95 (1H, dd, *J* 8.1, 1.5,

Py*H*-4), 8.75 (1H, dd, *J* 4.6, 1.5, Py*H*-6); 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ_c 17.7, 66.0, 81.0, 120.2, 124.9, 126.1, 128.3, 129.2, 138.8, 140.8, 148.6, 152.6, 172.3; HRMS (ESI) $C_{15}H_{13}O_{2}$ NBr [M+H]⁺ requires 318.0124; found 318.0131.

3-Ethyl-4-(3-fluoropyridin-4-yl)-3-phenyloxetan-2-one 28: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethylphenylketene (161 mg, 1.10 mmol) and 3-fluoro-4pyridinecarboxaldehyde (102 μL, 1.00 mmol) in toluene (20 mL) at -50 °C for 3 h. The crude product (dr 63:37 syn:anti) was purified by silica gel chromatography (90:10 petrol:Et₂O and then 90:10 petrol:EtOAc) to give syn-28 and anti-28 (85 mg, 31%) as colourless oil. syn-28: $[\alpha]_D^{20}$ +24.0 (c 0.03, CHCl₃); Chiral HPLC analysis Chiralpak IC (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm) t_R minor (R,R): 15.2 min, t_R major (S,S): 16.1 min, 37% ee; v_{max} (ATR)/cm⁻¹ (syn and anti); 2974, 1830 (C=O), 1609, 1568, 1493, 1449, 1416, 1250, 1103, 1053, 959, 905, 914, 841, 756; ¹H NMR (400 MHz, CDCl₃) δ_H 1.09 (3H, t, J 7.4, CH₃), 2.35–2.51 (2H, m, CH₂CH₃), 5.80 (1H, s, CH(Ar)), 7.03–7.17 (6H, m, PhH, PyH-5), 8.19 (1H, d, J 4.7, PyH-6), 8.29 (1H, s, PvH-2); ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ_c 9.4, 30.3, 72.6, 76.0 (d, J 3.5), 121.3, 126.3, 128.3, 128.9, 132.5 (d, J 11.1), 133.5, 137.2 (d, J 22.9), 145.8 (d, J 5.0), 151.5 (d, J 235), 170.9; HRMS (ESI) $C_{16}H_{15}O_2NF [M+H]^+$ requires 272.1081; found 272.1086; *anti-28*: $[\alpha]_D^{20}$ +20.0 (c 0.03, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R major (S,R): 8.82 min, t_R minor (R,S): 10.4 min, 64% ee; ¹H NMR (400 MHz, CDCl₃) δ_H 0.70 (3H, t, J 7.4, CH_3), 1.56–1.65 (1H, m, $CH_4H_BCH_3$), 1.68–1.77 (1H, m, $CH_4H_BCH_3$), 5.87 (1H, s, CH(Ar)), 7.39 (1H, tt, J 7.2, 1.3, PhH-4), 7.44–7.48 (2H, m, PhH), 7.51–7.54

(2H, m, Ph*H*), 7.57 (1H, t, *J* 5.5, Py*H*-5), 8.58–8.60 (2H, m, Py*H*-2,6); 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ_c 8.4, 26.9, 69.9, 77.00 (d, *J* 2.7), 122.0, 126.5 (d, *J* 3.5), 128.4, 129.2, 131.8 (d, *J* 11.4), 136.4, 138.1 (d, *J* 22.8), 146.7 (d, *J* 4.9), 156.6 (d, *J* 256), 170.7; HRMS (ESI) $C_{16}H_{15}O_2NF$ [M+H]⁺ requires 272.1081; found 272.1085.

(3S,4S)-3-Ethyl-3-phenyl-4-(quinolin-4-yl)oxetan-2-one 29: Prepared according to the general procedure from NHC precursor 3 (57.0 mg, 0.10 mmol), KHMDS (0.18 mL, 0.09 mmol), ethylphenylketene (175 mg, 1.20 mmol) and 4-quinolinecarboxaldehyde (157 mg, 1.00 mmol, 1.0 equiv) in toluene (20 mL) at -50 °C for 3 h. The crude product (dr 3:97 syn:anti) was purified by silica gel chromatography (85:15 petrol:EtOAc) to give anti-29 (172 mg, 57% yield) as pale yellow solid. mp 121–126 °C; $[\alpha]_{\rm p}^{20}$ +157.2 (c 0.46, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm) t_R major (S,S): 12.3 min, t_R minor (R,R): 17.5 min, 82% ee; v_{max} (ATR)/cm⁻¹ 2976, 1827 (C=O), 1595, 1508, 1246, 1107, 1042, 899, 760; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 0.73 (3H, t, J 7.4, CH₃), 1.51–1.58 (1H, m, $CH_4H_BCH_3$), 1.64–1.71 (1H, m, $CH_4H_BCH_3$), 6.05 (1H, s, CH(Ar)), 7.43–7.50 (3H, m, ArH), 7.51–7.57 (4H, m, ArH), 7.65 (1H, d, J 4.4, QuH-7), 7.73–7.78 (1H, m, Qu*H*-8), 8.20 (1H, d, *J* 8.4, Qu*H*-5), 9.01 (1H, d, *J* 4.3, Qu*H*-6); 13 C{ 1 H} NMR (75 MHz, CDCl₃) δ_{c} 8.6, 23.6, 69.8, 81.2, 118.2, 122.7, 124.9, 126.6, 127.3, 128.6, 129.6, 129.8, 130.8, 135.6, 140.6, 148.0, 150.4, 171.5; HRMS (ESI) $C_{20}H_{18}O_2N [M+H]^+$ requires 304.1332; found 304.1337.

4-(2-Chloro-6-(4-fluorophenyl)pyridin-4-yl)-3-ethyl-3-phenyloxetan-2-one 30:

Prepared according to the general procedure from NHC precursor 3 (28.5 mg, 0.05 mmol), KHMDS (90 μL, 0.045 mmol), ethylphenylketene (80.0 mg, 0.55 mmol) and

2-chloro-6-(4-fluorophenyl)isonicotinaldehyde (118 mg, 0.50 mmol) in toluene (10 mL) at -50 °C for 3 h. The crude product (dr 50:50 syn:anti) was purified by silica gel chromatography (95:5 petrol:Et₂O) to give syn-30 (83 mg, 44%) as colourless solid and anti-30 (50 mg, 26%) as colourless oil. syn-30: mp 114-118 °C; $[\alpha]_D^{20}$ +15.9 (c 0.15, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm) t_R major (S,R): 9.56 min, t_R minor (R,S): 12.0 min, 76% ee; v_{max} (ATR)/cm⁻¹ 1823 (C=O), 1603, 1549, 1514, 1412, 1323, 1159, 1109, 827, 760; ¹H NMR (300 MHz, CDCl₃) δ_H 0.75 (3H, t, J 7.4, CH₃), 1.52–1.64 (1H, m, CH₄H_BCH₃), 1.70–1.82 (1H, m, CH_AH_BCH₃), 5.63 (1H, s, CH(Ar)), 7.15–7.23 (2H, m, p-FPhH-3,5), 7.32 (1H, br. s, PyH-3), 7.38-7.53 (5H, m, PhH), 7.68 (1H, br. s, PyH-5), 8.02-8.09 (2H, m, p-FPhH-2,6); ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃) δ_c 8.6, 27.5, 69.5, 80.9, 115.2, 116.1 (d, J 22), 119.1, 126.2, 128.5, 129.2 (d, J 8.6), 129.4, 133.3 (d, J 3.1), 136.5, 148.1, 152.2, 157.9, 164.3 (d, J 251), 170.5; HRMS (ESI) C₂₂H₁₈O₂NFCl [M+H]⁺ requires 382.1005; found 382.1009; *anti-30*: $[\alpha]_D^{20}$ -11.2 (*c* 0.30, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (2% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm) t_R major (S,S): 11.3 min, t_R minor (R,R): 14.9 min, 74% ee; v_{max} (ATR)/cm⁻¹ 2974, 1825 (C=O), 1603, 1459, 1512, 1450, 1231, 1206, 1099, 902, 832; ¹H NMR (300 MHz, CDCl₃) δ_H 1.10 (3H, t, J 7.4, CH₃), 2.31–2.51 (2H, m, CH₂CH₃), 5.51 (1H, s, CH(Ar)), 6.99 (1H, d, J 0.7, PyH-3), 7.02–7.21 (8H, s, ArH), 7.71–7.78 (2H, m, p-FPhH-2,6); ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃) δ_{c} 9.4, 31.0, 72.5, 80.5, 115.8, 116.1 (d, J 9.1), 120.2, 126.8, 128.3, 128.9 (d, J 8.7), 129.1, 133.4, 148.2, 151. 5, 157.1, 166.7 (d, J 250), 171.0; HRMS (ESI) $C_{22}H_{18}O_2NFC1 [M+H]^+$ requires 382.1005; found 382.1009.

(S)-2-((S)-Amino(2-nitrophenyl)methyl)-2-phenylbutanoic acid 36: According to a literature procedure, ¹³ sodium azide (219 mg, 3.36 mmol, 2.0 equiv) was added to a solution of syn-7 (500 mg, 1.68 mmol, 1.0 equiv) in DMSO (5 mL) in a screw top vial and the mixture heated at 65 °C for 42 h. The solution was cooled to rt, diluted with NaHCO₃ (20mL) and extracted with EtOAc (20 mL). The aqueous phase was acidified with 2 M HCl (50mL) and extracted with EtOAc (3×20 mL). The combined organics were washed with brine (20 mL) and water (20 mL) before being dried over Na₂SO₄ and concentrated to give (S)-2-((S)-azido(2-nitrophenyl)methyl)-2phenylbutanoic acid as light brown solid; (564 mg, 99%); mp 156–160 °C; $[\alpha]_{\rm p}^{20}$ +386 (c 0.25, CHCl₃); v_{max} (KBr)/cm⁻¹ 2975, 2104 (N₃), 1707 (C=O), 1613, 1528, 1360, 1253, 1227, 1141, 1143, 1121, 857, 783, 831, 698, 744, 677; ¹H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$ 1.10 (3H, t, J 7.4, CH₃), 2.10–2.22 (1H, m, CH₄H_BCH₃), 2.55–2.67 (1H, m, CH_AH_BCH₃), 6.32 (1H, s, CHN₃), 6.42 (1H, dd, J 7.9, 1.2, ArH), 6.92–6.98 (2H, m, Ar*H*), 7.18–7.35 (5H, m, Ar*H*), 7.67 (1H, dd, *J* 8.1, 1.4, Ar*H*); ¹³C{¹H} NMR $(125 \text{ MHz}, \text{CDCl}_3) \delta_c 9.5, 27.6, 60.6, 60.6, 123.9, 127.9, 128.2, 128.8, 129.1, 130.5,$ 131.1, 131.6, 136.5, 151.2, 178.9; HRMS (EI) $C_{17}H_{15}O_4N_4$ [M–H]⁻ requires 339.1099; found 339.1093 (-1.7 ppm).

Next, PPh₃ (84 mg, 0.31 mmol, 1.08 equiv) was added to a solution of(*S*)-2-((*S*)-azido(2-nitrophenyl)methyl)-2-phenylbutanoic acid (100 mg, 0.29 mmol, 1.00 equiv) in THF (4 mL) and water (1 mL) in a screw top vial and stirred vigorously for 6 h. The mixture was concentrated before being diluted with methanol and cooled to 0 °C in a screw cap vial. SOCl₂ (0.06 mL, 0.30 mmol, 1.0 equiv) was added dropwise and the vial heated at 65 °C for 12 h. The reaction was concentration and the resulting solid was triturated with cold CHCl₃ to give (*S*,*S*)-36 as off white solid; (75 mg, 81%)

yield); mp 188–194 °C dec.; $[\alpha]_D^{20}$ +213 (c 0.15 MeOH); v_{max} (KBr)/cm⁻¹ 3417, 2800, 1684 (C=O), 1602, 1531, 1496, 1352, 1220, 1195, 857, 713, 603; ¹H NMR (300 MHz, MeOD₃) δ_H 0.73 (3H, t, J 7.1, CH_3), 1.80–1.99 (2H, m, CH_2CH_3), 5.88 (1H, s, $CHNH_2$), 7.12–7.19 (2H, m, ArH), 7.32–7.41 (3H, m, ArH), 7.50–7.55 (1H, m, ArH), 7.59–7.74 (2H, m, ArH), 7.94 (1H, dd, J 7.9, 1.5, ArH); ¹³C{¹H} NMR (75 MHz, MeOD) δ_c 9.9, 28.2, 54.9, 59.9, 126.5, 129.0, 129.1, 129.6, 130.1, 130.3, 131.8, 134.4, 137.4, 151.7, 176.2; HRMS (EI) $C_{17}H_{17}O_4N_2$ [M–H]⁻ requires 313.1194; found 313.1186 (–2.5 ppm).

(S,S)-36 was derivatised into its dibenzyl compound to allow chiral HPLC analysis. Benzylbromide (25 μL, 0.21mmol, 2.2 equiv) and DIEPA (0.37 μL, 0.19 mmol, 2.2 equiv) were added to a solution of (S,S)-36 (30 mg, 0.10 mmol, 1.0 equiv) in DMF (2) mL) and stirred at rt for 12 h. The reaction was diluted with Et₂O, washed with 2 M HCl and water (×2) before being dried over Na₂SO₄ and concentrated. The crude product was purified by silica gel chromatography (90:10 petrol:Et₂O) to give (S)benzyl 2-((S)-(benzylamino)(2-nitrophenyl)methyl)-2-phenylbutanoate (16 mg, 34%) as pale yellow oil; $[\alpha]_{D}^{20}$ +269 (c 0.1, CHCl₃); Chiral HPLC Chiralpak OD-H (1% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R minor (R,R): 15.7 min, t_R major (S,S): 19.2 min, 99% ee; v_{max} (KBr)/cm⁻¹3031, 1730 (C=O), 1646, 1603, 1529, 1496, 1454, 1357, 1218, 1108, 1029, 856, 739, 698; 1 H NMR (400 MHz, CDCl₃) δ_{H} 0.64 (3H, t, J7.3, CH₃), 1.78–1.87 (1H, m, CH₄H_BCH₃), 2.33–2.41 (1H, m, CH_AH_BCH₃), 3.45 (1H, d, J 13.4, OC H_AH_BPh), 3.60 (1H, d, J 13.4, OC H_AH_BPh), 5.00 (1H, d, J 12.3, NCH_AH_BPh), 5.08 (1H, d, J, 12.3, NCH_AH_BPh), 5.30 (1H, s, CHNHBn), 6.75 (1H, d, J 8.0, ArH), 6.91 (2H, dd, J 8.1, 1.4, ArH), 7.11–7.24 (15H, m, ArH), 7.64 (1H, dd, J 8.1, 1.4, Ar*H*); ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ_c 9.7, 28.7, 51.8, 58.6, 61.6, 66.7,

124.0, 127.0, 127.3, 127.8, 128.0, 128.2, 128.3, 128.5, 128.6, 128.7, 130.1, 131.3, 135.5, 135.6, 139.0, 140.5, 152.6, 173.8; HRMS (EI) C₃₁H₃₁O₄N₂ [M+H]⁺ requires 495.2278; found 495.2271 (-1.5 ppm).

(S)-2-((R)-Hydroxy(2-nitrophenyl)methyl)-2-phenylbutanoic acid 37: According to a literature procedure. 13 1 M KOH (1.35 mL, 1.35 mmol, 2.0 equiv) was added to a solution of syn-7 (200 mg, 0.67 mmol, 1.0 equiv) in THF (4 mL) in a screw cap vial and heated at 65 °C for 5 h. The solution was cooled to rt, diluted with NaHCO₃ and extracted with Et₂O. The aqueous layer was slowly acidified with 2 M HCl and extracted with EtOAc (×3). The combined organics were dried over Na₂SO₄ and concentrated to give (S,R)-37 as off white solid; (116 mg, 55%); mp 140–148 °C; $[\alpha]_{p}^{20}$ -406 (c 0.05 MeOH); v_{max} (KBr)/cm⁻¹ 3438, 1705 (C=O), 1528, 1355, 1223, 1037, 943, 857; ¹H NMR (400 MHz, MeOD) $\delta_{\rm H}$ 0.84 (3H, t, J 7.4, CH₃), 1.79 (2H, q, J 7.5, CH₂CH₃), 6.14 (1H, s, CHOH), 6.37 (1H, dd, J 8.0, 1.2, ArH), 7.05–7.14 (3H, m, ArH), 7.16–7.26 (3H, m, ArH), 7.28–7.33 (1H, m, ArH), 7.65 (1H, dd, J 8.1, 1.1, ArH): ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, MeOD) δ_{c} 9.5, 26.7, 49.0, 63.1, 72.0, 123.7, 127.8, 128.0, 129.2, 131.8, 131.9, 132.2, 135.5, 136.8, 151.4, 177.7; HRMS (EI) $C_{17}H_{16}O_5N$ $[M-H]^-$ requires 314.1034; found 314.1029 (-1.6 ppm). (S,R)-37 was derivatised into its benzyl ester to allow chiral HPLC analysis.

(S,R)-37 was derivatised into its benzyl ester to allow chiral HPLC analysis. Benzylbromide (50 μL, 0.37 mmol, 1.2 equiv) and DIEPA (80 μL, 0.37 mmol, 1.2 equiv) were added to a solution of (S,R)-37 (98 mg, 0.31 mmol, 1.0 equiv) in DMF (2 mL) and stirred at rt for 12 h. The reaction was diluted with Et₂O, washed with 2 M HCl and water (×2) before being dried over Na₂SO₄ and concentrated. The crude product was purified by silica gel chromatography (80:20 petrol:Et₂O) to give (S)-

benzyl 2-((*R*)-hydroxy(2-nitrophenyl)methyl)-2-phenylbutanoate (79 mg, 63%) as orange oil. [α]_D²⁰ -807 (c 0.3, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 220 nm) t_R minor (S_R): 22.8 min, t_R major (S_R): 34.1 min, 96% ee; v_{max} (KBr)/cm⁻¹ 3527, 2975, 1714 (C=O), 1607, 1527, 1447, 1355, 1216, 1125, 1039, 990, 912, 856, 785, 738, 699; ¹H NMR (300 MHz, CDCl₃) δ_H 0.83 (3H, t, J 7.3, C H_3), 1.63–1.75 (1H, m, C H_A H_BCH₃), 1.89–2.01 (1H, m, CH_AH_BCH₃), 3.87 (1H, d, J 4.0, CHOH), 5.23 (2H, s, OC H_2 Ph), 6.20 (1H, d, J 4.0, CHOH), 6.37 (1H, dd, J 8.3, 1.1, ArH), 6.83–6.90 (2H, m, ArH), 7.04–7.09 (1H, m, ArH), 7.14–7.21 (2H, m, ArH), 7.23–7.34 (7H, m, ArH), 7.63 (1H, dd, J 8.3, 1.1, ArH); 13 C{¹H} NMR (100 MHz, CDCl₃) δ_c ; 9.2, 26.3, 61.3, 67.2, 69.8, 123.0, 127.3, 127.5, 128.3, 128.5, 128.6, 128.6, 129.4, 131.1, 131.1, 132.5, 135.3, 136.6, 149.8, 175.7; HRMS (EI) C₂₄H₂₄O₅N [M+H]⁺ requires 406.1649; found 406.1650 (+0.2 ppm).

(S)-2-((R)-Amino(pyridin-2-yl)methyl)-2-phenylbutanoic acid 38: Based upon a literature procedure, 13 syn-21 (100 mg, 0.39 mmol, 1 equiv) and NaN₃ (51 mg, 0.78 mmol, 2 equiv) in DMSO (1.25 mL) were heated at 65 °C in a sealed screw cap vial for 48 h. The reaction mixture was cooled to rt, diluted with NaHCO₃ (5 mL) and washed with EtOAc (5 mL). The aqueous phase was acidified to pH 3 with 1 M HCl before being extracted with EtOAc (3×10 mL). The combined organics were dried over MgSO₄ and concentrated before the resulting solid was triturated with Et₂O to give (S)-2-((R)-azido(pyridin-2-yl)methyl)-2-phenylbutanoic acid (103 mg, 88%) as white solid. mp 138 °C dec. (Et₂O); $[\alpha]_D^{20}$ +167.1 (c 0.70 in CHCl₃); ν_{max} (film) 2104 (CN₃), 1699 (C=O), 1599; ¹H NMR (300 MHz, CDCl₃) δ_H : 0.99 (3H, t, J 7.4, CH₃), 2.22–2.34 (1H, m, CH₄H_BCH₃), 2.37–2.53 (1H, m, CH₄H_BCH₃), 5.44 (1H, s, CHN₃),

6.84 (1H, d, J 7.92, C(5)-pyH), 7.09–7.24 (5H, m, ArH), 7.29–7.36 (1H, m, ArH), 7.63 (1H, td, J 7.8, 1.7, C(4)-pyH), 8.75 (1H, d, J 5.2, C(2)-pyH); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_C : 9.6, 28.9, 61.7, 71.2, 124.1, 124.5, 127.4, 127.7, 129.2, 137.2, 138.3, 147.6, 156.1, 173.5; HRMS (NSI⁺) C₁₆H₁₇O₂N₄ [M+H]⁺ requires 297.1346; found 297.1346 (+0.0 ppm).

Based upon a literature procedure, ³⁴ (*S*)-2-((*R*)-azido(pyridin-2-yl)methyl)-2-phenylbutanoic acid (103 mg, 0.35 mmol, 1 equiv) and Pd/C (10% w/w, 37 mg, 35 μmol, 10 mol%) were placed under an atmosphere of N₂ before MeOH (6 mL) was added. The solution was placed under an atmosphere of H₂ using a balloon and stirred at rt for 3 h. The reaction mixture was filtered through a pad of Celite®, washing with MeOH, before being concentrated to give (*S*,*R*)-38 (95 mg, 100%) as white solid. mp 112–115 °C dec.; $[\alpha]_D^{20}$ –12.6 (*c* 0.54 in CHCl₃); v_{max} (film) 3364 (br. N-H), 1591 (br. C=O), 1572; ¹H NMR (300 MHz, CDCl₃) δ_H: 0.78 (3H, t, *J* 7.1, C*H*₃), 1.57–1.85 (2H, m, C*H*₂CH₃), 4.84 (1H, s, C*H*NH₂), 6.13 (1H, d, *J* 7.6, C(5)-py*H*), 7.05–7.35 (8H,m, Ar*H*), 8.38 (2H, br. s, N*H*₂), 8.49 (1H, d, *J* 4.2, C(2)-py*H*); ¹³C (¹H) NMR (75 MHz, CDCl₃) δ_C: 9.8, 24.0, 58.1, 60.9, 123.2, 124.1, 127.0, 128.2, 129.1, 135.9, 140.6, 148.2, 154.5, 178.0; HRMS (NSI⁺) C₁₆H₁₉O₂N₂ [M+H]⁺ requires 271.1442; found 271.1441 (+0.4 ppm).

(S,R)-38 was derivatised into its methyl ester to allow chiral HPLC analysis. (S,R)-38 (95 mg, 0.35 mmol, 1 equiv) was dissolved in MeOH (1.75 mL) under an inert atmosphere of nitrogen. (Trimethylsilyl)diazomethane (0.6 M in hexanes, 0.65 mL, 0.39 mmol, 1.1 equiv) was added dropwise and the resulting solution was stirred at rt for 3 h in the absence of light. The reaction was quenched with NaHCO₃, diluted with H₂O and extracted with Et₂O (×3). The combined organics were washed with brine, dried over MgSO₄, and concentrated. The crude product was purified by silica gel

chromatography (95:5 CH₂Cl₂:MeOH) to give (*S*)-methyl 2-((*R*)-amino(pyridin-2-yl)methyl)-2-phenylbutanoate (47 mg, 47%) as orange oil. $[\alpha]_D^{20}$ –25.0 (*c* 0.58 in CHCl₃); Chiral HPLC analysis, Chiralpak AD-H (5% IPA:hexane, flow rate 1 mL min⁻¹, 254 nm, 30 °C) t_R (minor): 30.4 min, t_R (major): 34.3 min, 95% ee; v_{max} (film) 1724 (C=O), 1589, 1570; ¹H NMR (300 MHz, CDCl₃) δ_H : 0.65 (3H, t, *J* 7.4, CH_AH_BCH₃), 1.65–1.77 (1H, m, CH_AH_BCH₃), 1.91–2.03 (1H, m, CH_AH_BCH₃), 2.50 (2H, br. s, NH₂), 3.69 (3H, s, OCH₃), 4.59 (1H, s, CHNH₂), 6.96 (1H, br. d, *J* 8.0, C(5)-py*H*), 7.01–7.29 (6H, m, Ar*H*), 7.46 (1H, td, *J* 7.6, 1.9, C(4)-py*H*), 8.42–8.46 (1H, m, C(2)-py*H*); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_C : 9.9, 29.0, 51.9, 61.7, 63.0, 122.6, 123.9, 127.0, 128.2, 128.4, 135.9, 139.7, 148.9, 159.7, 174.6; HRMS (NSI⁺) C₁₇H₂₁N₂O₂ [M+H]⁺ requires 285.1598; found 285.1601 (+1.2 ppm).

(S)-2-((S)-Hydroxy(pyridin-2-yl)methyl)-2-phenylbutanoic acid 39: Based upon a literature procedure, ¹³ syn-21 (100 mg, 0.39 mmol, 1 equiv), 1 m KOH (0.79 mL, 0.79 mmol, 2 equiv) and THF (1.3 mL) were heated at 60 °C in a screw cap vial for 16 h. The solution was cooled to rt before being concentrated. The resulting oil was dissolved in EtOAc (5 mL) and acidified to pH 3 with 1 m HCl. The layers were separated and the aqueous extracted with EtOAc (2×5 mL) before the combined organics were washed with brine, dried over MgSO₄, and concentrated to give (S,S)-39 (102 mg, 96%) as pale yellow solid. mp 48–50 °C; $[\alpha]_D^{20}$ –99.6 (c 1.12 in CHCl₃); v_{max} (film) 3439 (br. O-H), 1699 (C=O), 1601; ¹H NMR (300 MHz, CDCl₃) δ_H : 1.00 (3H, t, J 7.4, CH₃), 1.99–2.16 (1H, m, CH₄H_BCH₃), 2.21–2.39 (1H, m. CH_AH_BCH₃), 5.51 (1H, s, CHOH), 6.71 (1H, d, J 7.8, C(5)-pyH), 7.01–7.09 (2H, m, ArH), 7.11–7.20 (4H, m, ArH), 7.48 (1H, td, J 7.8, 1.6, C(4)-pyH), 8.45 (1H, br. d, J 4.9,

C(2)-pyH), 9.06 (1H, br. s, COOH); ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃) δ_{C} : 9.7, 28.4, 61.8, 77.2, 123.2, 124.2, 127.0, 127.6, 129.1, 137.5, 137.6, 146.0, 158.5, 177.4; HRMS (NSI⁺) $C_{16}H_{18}NO_3$ [M+H]⁺ requires 272.1281; found 272.1284 (+1.0 ppm). (S,S)-39 was derivatised into its methyl ester to allow chiral HPLC analysis. (S,S)-39 (108 mg, 0.39 mmol, 1 equiv) was dissolved in MeOH (2 mL) in under an inert atmosphere of nitrogen and cooled to 0 °C. (Trimethylsilyl)diazomethane (0.6 M in hexanes, 0.83 mL, 0.5 mmol, 1.3 equiv) was added dropwise and the resulting solution was stirred at rt for 3 h in the absence of light. The reaction was quenched with NaHCO₃, diluted with H_2O and extracted with Et_2O (×3). The combined organics were washed with brine, dried over MgSO₄, and concentrated to give (S)methyl 2-((S)-hydroxy(pyridin-2-yl)methyl)-2-phenylbutanoate (52 mg, 47%) as pale yellow solid. mp 96–98 °C (Et₂O); $[\alpha]_D^{20}$ –161.9 (c 0.80 in CHCl₃); Chiral HPLC analysis, Chiralpak AD-H (5% IPA:hexane, flow rate 1 mL min⁻¹, 254 nm, 30 °C) t_R (major): 14.3 min, t_R (minor): 20.2 min, 83% ee; v_{max} (film) 1728 (C=O), 1591, 1570, 1499; 1 H NMR (300 MHz, CDCl₃) δ_{H} : 1.02 (3H, t, J 7.4, CH_AH_BCH₃), 2.00–2.12 (1H, m, $CH_4H_BCH_3$), 2.17-2.29 (1H, m, $CH_4H_BCH_3$), 3.78 (3H, s, OCH_3), 5.45 (1H, s, CHOH), 6.64 (1H, d, J 7.9, C(5)-pyH), 6.94–6.99 (2H, m, ArH), 7.08 (1H, ddd, J 7.5, 4.9, 1.2, ArH), 7.14–7.23 (3H, m, ArH), 7.40 (1H, td, J 7.8, 1.8, C(4)-pyH), 8.33–8.38 (1H, m, C(2)-pyH); ${}^{13}C\{{}^{1}H\}$ NMR (75 MHz, CDCl₃) δ_{C} : 9.8, 26.7, 52.3, 61.5, 76.3, 122.6, 122.7, 127.1, 127.4, 128.8, 135.5, 137.3, 147.5, 158.5, 176.1; HRMS (NSI^+) $C_{17}H_{20}NO_3$ $[M+H]^+$ requires 286.1438; found 286.1439 (+0.5 ppm).

(S)-N-Benzyl-2-((S)-hydroxy(pyridin-2-yl)methyl)-2-phenylbutanamide 40: syn-21 (100 mg, 0.039 mmol, 1 equiv) was dissolved in anhydrous CH₂Cl₂ (1.7 mL) in a screw cap vial under a N₂ atmosphere. Benzylamine (0.22 mL, 1.97 mmol, 5 equiv)

and triethylamine (60 μ L, 0.43 mmol, 1.1 equiv) were added and the solution was heated at 40 °C for 16 h. The reaction mixture was cooled to rt before being diluted with CH₂Cl₂ and washed with NH₄Cl (×2) and then brine. The organic layer was dried over MgSO₄ and concentrated. The crude product was purified by silica gel chromatography (99:1 CH₂Cl₂:MeOH) to give (*S*,*S*)-40 (72 mg, 54%) as white solid. mp 157–158 °C (CH₂Cl₂); $[\alpha]_D^{20}$ –152.4 (*c* 0.97 in CHCl₃); Chiral HPLC analysis, Chiralpak AD-H (5% IPA:hexane, flow rate 1 mL min⁻¹, 254 nm, 30 °C) t_R (minor): 37.2 min, t_R (major): 48.7 min, 83% ee; v_{max} (film) 3302 (N-H), 1616 (C=O), 1589, 1545; ¹H NMR (300 MHz, CDCl₃) δ_H : 1.20 (3H, t, *J* 7.4, CH₃), 1.88–2.00 (1H, m, CH₄H_BCH₃), 2.32–2.44 (1H, m, CH_AH_BCH₃), 4.56 (2H, d, *J* 5.8, CH₂Ph), 4.76 (1H, br. s, OH), 5.59 (1H, s, CHOH), 6.00 (1H, br. t, *J* 5.7, NH), 6.60 (1H, d, *J* 8.0, C(5)-pyH), 7.01–7.12 (3H, m, ArH), 7.14–7.42 (9H, m, ArH), 8.38–8.42 (1H, m, C(2)-pyH); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_C : 9.6, 27.6, 44.0, 60.7, 76.1, 122.3, 122.7, 127.4, 127.6, 127.7, 128.0, 128.8, 129.3, 135.4, 138.1, 138.2, 147.3, 159.3, 176.2; HRMS (NSI⁺) C₂₃H₂₅N₂O₂ [M+H]⁺ requires 361.1911; found 361.1914 (+1.0 ppm).

(3S,4S)-3-Methyl-4-(6-morpholinopyridin-2-yl)-3-phenyloxetan-2-one 41: Pd(OAc)₂ (7.0 mg, 0.031 mmol, 0.10 equiv), (\pm)-BINAP (39.1 mg, 0.063 mmol, 0.20 equiv) and Cs₂CO₃ (307 mg, 0.942 mmol, 3.0 equiv) were added to an oven dried vial and suspended in anhydrous toluene (4 mL) under an atmosphere of argon. The suspension was stirred for 20 min before *syn*-26 (100mg, 0.314 mmol, 1.00 equiv, 86% ee, >95:5 dr) and morpholine (54 μ L, 0.629 mmol, 2.00 equiv) were added. The reaction was heated at 50 °C for 8.5 h then 80 °C for 4 h. The suspension was cooled to rt, filtered through Celite® and concentrated. The crude product was purified by silica gel chromatography (80:20 petrol:Et₂O) to give (*S*,*S*)-41 (67 mg, 66%) as

colourless solid. mp 80–82 °C; $[\alpha]_{D}^{20}$ –177.5 (*c* 0.48, CHCl₃); Chiral HPLC analysis Chiralpak AS-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm) t_R major (*S,S*): 25.4 min, t_R minor (*R,R*): 31.0 min, 87% *ee*; v_{max} (ATR)/cm⁻¹ 2961, 2365, 1817 (C=O), 1593, 1558, 1473, 1437, 1260, 1244, 1113, 1103, 976, 966, 912, 878, 764; ¹H NMR (300 MHz, CDCl₃) δ_H 1.97 (3H, s, C*H*₃), 3.34 (4H, t, *J* 4.8, morphC*H*₂-3,5), 3.74 (4H, t, *J* 4.8, morphC*H*₂-2,6), 5.43 (1H, s, C*H*(Ar)), 6.35 (1H, d, *J* 8.5, Py*H*-5), 6.53 (1H, d, *J* 7.3, Py*H*-3), 7.05–7.12 (5H, m, Ph*H*), 7.26–7.31 (1H, m, Py*H*-4); ¹³C{¹H} NMR (75 MHz, CDCl₃) δ_c 25.3, 45.5, 66.8, 84.2, 106.5, 111.4, 126.5, 127.2, 128.1, 136.7, 137.9, 153.0, 158.6, 173.3; HRMS (ESI) C₁₉H₂₁O₃N₂ [M+H]⁺ requires 325.1547; found 325.1552.

Tert-butyl 5-methoxy-2-(6-((2S,3S)-3-methyl-4-oxo-3-phenyloxetan-2-yl)pyridin-2-yl)-1H-indole-1-carboxylate 42: Pd(PPh₃)₄ (36.3 mg, 0.031 mmol, 0.10 equiv), syn-26 (100 mg, 0.314 mmol, 1.00 equiv, 86% ee, >95:5 dr) and (1-(tert-butoxycarbonyl)-5-methoxy-1H-indol-2-yl)boronic acid (110 mg, 0.377 mmol, 1.20 equiv) were added to an oven dried vial and suspended in DME (5 mL) under an atmosphere of argon. 2 M Na₂CO₃ (0.47 mL, 0.377 mmol, 3.0 equiv) was added and the reaction heated at 85 °C for 4 h. The solution was cooled to rt before being diluted with EtOAc and washed with water. The organic phase was dried over MgSO₄ and concentrated. The crude product was purified by silica gel chromatography (90:10 petrol:EtOAc) to give (S,S)-42 (140 mg, 92%) as pale yellow oil. [α]_D²⁰ –263.6 (c 0.39, CHCl₃); Chiral HPLC analysis Chiralpak AD-H (5% IPA:hexane, flow rate 1.0 mL min⁻¹, 254 nm) t_R major (S,S): 13.7 min, t_R minor (R,R): 20.6 min, 89% ee; ν_{max} (ATR)/cm⁻¹ 2976, 1830 (C=O), 1732 (C=O), 1593, 1574, 1454, 1369, 1319, 1219, 1159, 1121, 1099, 1059, 1032, 984, 905, 849, 804, 760; ¹H NMR (400 MHz, CDCl₃) δ_H 1.39 (9H, s, OC(CH₃)₃), 1.99 (3H,

s, CC H_3), 3.88 (3H, s, OC H_3), 5.72 (1H, s, CH(Ar)), 6.60 (1H, s, IndH-3), 6.97–7.01 (2H, m, HetArH), 7.06 (1H, d, J 2.6, HetArH), 7.08–7.16 (5H, m, PhH), 7.23 (1H, dd, J 7.8, 0.8, HetArH), 7.47 (1H, t, J 7.8, HetArH), 8.05 (1H, d, J 9.1, PyH-3); 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ_c 25.1, 27.9, 55.8, 67.0, 83.7, 84.5, 103.4, 111.5, 114.2, 116.2, 119.5, 123.2, 126.8, 127.5, 128.4, 129.7, 132.4, 135.9, 136.3, 139.4, 150.0, 152.5, 154.8, 156.2, 172.8; HRMS (ESI) $C_{29}H_{29}O_5N_2$ [M+H] $^+$ requires 485.2071; found 485.2068.

Acknowledgements: The authors would like to thank the Royal Society for a University Research Fellowship (ADS) and the EPSRC and AstraZeneca (Case award to JD) for funding. JET and ADS group research has received funding from the European Research Council under the European Union's Seventh Framework Programme (FP7/2007-2013) / ERC grant agreement n° 279850.

Supporting Information

¹H and ¹³C{¹H} NMR spectra and HPLC trances of all β-lactones and derivatisation products. X-ray crystal structures and CIF files for *syn-7* and *syn-21*. This material is available free of charge *via* the Internet at http://pubs.acs.org.

References

(1) (a) Taunton, J.; Collins, J. L.; Schreiber, S. L. J. Am. Chem. Soc. 1996, 118, 10412-10422; (b) Wan, Z.; Nelson, S. G. J. Am. Chem. Soc. 2000, 122, 10470-10471; (c) Nelson, S. G.; Cheung, W. S.; Kassick, A. J.; Hilfiker, M. A. J. Am. Chem. Soc. 2002, 124, 13654-13655; (d) Wang, Y. C.; Tennyson, R. L.; Romo, D.

- Heterocycles **2004**, *64*, 605-658; (e) Shen, X.; Wasmuth, A. S.; Zhao, J.; Zhu, C.; Nelson, S. G. J. Am. Chem. Soc. **2006**, *128*, 7438-7439.
- (2) (a) Nelson, S. G.; Spencer, K. L. Angew. Chem., Int. Ed. 2000, 39, 1323-1325;
 (b) Nelson, S. G.; Wan, Z.; Stan, M. A. J. Org. Chem. 2002, 67, 4680-4683.
- (3) (a) Abe, H.; Matsubara, I.; Doi, Y. *Macromolecules* **1995**, *28*, 844-853; (b) Leboucher-Durand, M.-A.; Langlois, V.; Guerin, P. *Polym. Bull. (Berlin)* **1996**, *36*, 35-41.
- (4) Pommier, A.; Pons, J. M. Synthesis 1995, 729-744.
- (5) (a) Pommier, A.; Pons, J. M. Synthesis 1993, 441-459; (b) Yang, H. W.; Romo, D. Tetrahedron 1999, 55, 6403-6434; (c) Schneider, C. Angew. Chem., Int. Ed. 2002, 41, 744-746; (d) Orr, R. K.; Calter, M. A. Tetrahedron 2003, 59, 3545-3565.
- (6) (a) Case-Green, S. C.; Davies, S. G.; Hedgecock, C. J. R. Synlett 1991, 781-782;
 (b) Case-Green, S. C.; Davies, S. G.; Hedgecock, C. J. R. Synlett 1991, 779-780;
 (c) Buttero, P.; Montrasio, D. Molecules 2001, 6, 13-20.
- (7) (a) Tamai, Y.; Yoshiwara, H.; Someya, M.; Fukumoto, J.; Miyano, S. *J. Chem. Soc., Chem. Commun.* 1994, 2281-2282; (b) Dymock, B. W.; Kocienski, P. J.; Pons, J.-M. *Chem. Commun.* 1996, 1053-1054; (c) Forslund, R. E.; Cain, J.; Colyer, J.; Doyle, M. P. *Adv. Synth. Catal.* 2005, 347, 87-92; (d) Gnanadesikan, V.; Corey, E. J. *Org. Lett.* 2006, 8, 4943-4945; (e) Kull, T.; Peters, R. *Adv. Synth. Catal.* 2007, 349, 1647-1652.
- (8) Staudinger, H.; Bereza, S. *Liebigs Ann. Chem.*, **1911**, 243-247.
- (9) (a) Wynberg, H.; Staring, E. G. J. J. Am. Chem. Soc. 1982, 104, 166-168; (b)
 Wynberg, H.; Staring, E. G. J. J. Org. Chem. 1985, 50, 1977-1979.

- (10)(a) Nelson, S. G.; Peelen, T. J.; Wan, Z. J. Am. Chem. Soc. 1999, 121, 9742-9743; (b) Zhu, C.; Shen, X.; Nelson, S. G. J. Am. Chem. Soc. 2004, 126, 5352-5353.
- (11)(a) Tennyson, R.; Romo, D. J. Org. Chem. 2000, 65, 7248-7252; (b) Cortez, G. S.; Tennyson, R. L.; Romo, D. J. Am. Chem. Soc. 2001, 123, 7945-7946; (c) Oh, S. H.; Cortez, G. S.; Romo, D. J. Org. Chem. 2005, 70, 2835-2838; (d) Henry-Riyad, H.; Lee, C.; Purohit, V. C.; Romo, D. Org. Lett. 2006, 8, 4363-4366; (e) Purohit, V. C.; Richardson, R. D.; Smith, J. W.; Romo, D. J. Org. Chem. 2006, 71, 4549-4558; (f) Leverett, C. A.; Purohit, V. C.; Romo, D. Angew. Chem., Int. Ed 2010, 49, 9479-9483; (g) Morris, K. A.; Arendt, K. M.; Oh, S. H.; Romo, D. Org. Lett. 2010, 12, 3764-3767.
- (12) Calter, M. A.; Tretyak, O. A.; Flaschenriem, C. Org. Lett. 2005, 7, 1809-1812.
- (13) Wilson, J. E.; Fu, G. C. Angew. Chem., Int. Ed. 2004, 43, 6358-6360.
- (14) Mondal, M.; Ibrahim, A. A.; Wheeler, K. A.; Kerrigan, N. J. *Org. Lett.* **2010**, *12*, 1664-1667.
- (15) Wang, X.-N.; Shao, P.-L.; Lv, H.; Ye, S. Org. Lett. 2009, 11, 4029-4031.
- (16) He, L.; Lv, H.; Zhang, Y.-R.; Ye, S. J. Org. Chem. 2008, 73, 8101-8103.
- (17)(a) Duguet, N.; Campbell, C. D.; Slawin, A. M. Z.; Smith, A. D. Org. Biomol.
 Chem. 2008, 6, 1108-1113; (b) Douglas, J.; Ling, K. B.; Concellón, C.; Churchill,
 G.; Slawin, A. M. Z.; Smith, A. D. Eur. J. Org. Chem. 2010, 2010, 5863-5869.
- (18)(a) Enders, D.; Niemeier, O.; Henseler, A. *Chem. Rev.* 2007, 107, 5606-5655; (b)
 Moore, J. L.; Rovis, T. 2009, 291, 77-144; (c) Campbell, C. D.; Ling, K. B.;
 Smith, A. D.; Cazin, C. S. J., Ed.; Springer Netherlands: 2011, p 263-297.
- (19) Lv, H.; Zhang, Y.-R.; Huang, X.-L.; Ye, S. Adv. Synth. Catal. 2008, 350, 2715-2718.

- (20) Enders, D.; Han, J. Tetrahedron: Asymmetry 2008, 19, 1367-1371.
- (21) The relative configuration of the *syn:anti* β-lactone diastereoisomers was assigned using nOe spectroscopic analysis.
- (22) The relative and absolute configuration within **6** was proven by chemical correlation to Kerrigan's work (ref 14). The configuration within **5** was assigned by analogy.
- (23)(a) Sereda, O.; Wilhelm, R. Synlett 2007, 2007, 3032,3036; (b) Tabassum, S.; Sereda, O.; Reddy, P. V. G.; Wilhelm, R. Org. Biomol. Chem. 2009, 7, 4009-4016.
- (24) The relative and absolute configuration within 7 was determined by X-ray crystallography. Crystallographic data for 7 has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 854302.
- (25)β-Lactones **8-10** derived from 2-halobenzaldehydes proved difficult to purify to homogeneity and were therefore isolated after ring-opening with KOH followed by esterification to give isolable β-hydroxyacids in 25-36% yield over three steps.
- (26) Employing 2-tolylbenzaldehyde under the optimised conditions gave only ketene dimerisation and return of the aldehyde, consistent with the requirement for an electron-withdrawing 2-substituent within the aldehyde to facilitate β -lactone formation.
- (27) DiRocco, D. A.; Oberg, K. M.; Dalton, D. M.; Rovis, T. J. Am. Chem. Soc. 2009, 131, 10872-10874.
- (28)(a) Desimoni, G.; Faita, G.; Quadrelli, P. Chem. Rev. 2003, 103, 3119-3154; (b)
 Shaw, S. A.; Aleman, P.; Vedejs, E. J. Am. Chem. Soc. 2003, 125, 13368-13369;
 (c) Chelucci, G.; Murineddu, G.; Pinna, G. A. Tetrahedron: Asymmetry 2004, 15,

- 1373-1389; (d) Fu, G. C. *Acc. Chem. Res.* **2004**, *37*, 542-547; (e) Kwong, H. L.; Yeung, H. L.; Yeung, C. T.; Lee, W. S.; Lee, C. S.; Wong, W. L. *Coord. Chem. Rev.* **2007**, *251*, 2188-2222.
- (29) O'Hagen, D. Nat. Prod. Rep. 1997, 14, 637-651
- (30) Brody, F.; Ruby, P. R. In *Chemistry of Heterocyclic Compounds: Pyridine and its Derivatives*; John Wiley & Sons, Hoboken, 2008; Vol 14, p 99-589.
- (31) Mulzer, J.; Zippel, M. Tetrahedron Lett. 1980, 21, 751-754.
- (32)A control experiment re-subjecting isolated *syn-21* (>95:5 dr) to the reaction conditions for 24 h resulted in no loss of diastereoselectivity.
- (33) The relative and absolute configuration within **21**, **25** and **27** were determined by X-ray crystallography. Crystallographic data for **21**, **25** and **27** has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers CCDC 923189, 923190 and 923191.
- (34) Armstrong, A.; Geldart, S. P.; Jenner, C. R.; Scutt, J. N. J. Org. Chem. 2007, 72, 8091-8094.
- (35) Schlummer, B.; Scholz, U. Adv. Synth. Catal. 2004, 346, 1599-1626.
- (36) Miyaura, N.; Suzuki, A. Chem. Rev. 1995, 95, 2457-2483.
- (37) Baigrie, L. M.; Seiklay, H. R.; Tidwell, T. T. J. Am. Chem. Soc. 1985, 107, 5391-5396.
- (38) Hodous, B. L.; Fu, G. C. J. Am. Chem. Soc. 2002, 124, 1578-1579.
- (39) Wiskur, S. L.; Fu, G. C. J. Am. Chem. Soc. 2005, 127, 6176-6177.