

Synthesis of Some New Heterocycles of Pharmaceutical Interest: Pyridinyl and Isoxazolyl Quinoxaline Derivatives

Osama S. Moustafa

Chemistry Department, Faculty of Science, Assiut University, Assiut 71516, Egypt

3-(*p*-Acetyl-anilinomethyl)quinoxalin-2(1*H*)-one (**3**) was prepared by the reaction of 3-bromomethylquinoxalin-2(1*H*)-one (**1**) with *p*-aminoacetophenone (**2**) in pyridine. Reaction of *p*-acetyl compound (**3**) with aromatic aldehydes yield the corresponding chalcones (**4a-c**). Condensation of latter chalcones with malononitrile afforded cyanopyridines (**5a-c**). Also, the reaction of chalcones (**4a-c**) with hydroxylamine hydrochloride furnished isoxazoles (**6a-c**). The reaction of bromo compound (**1**) with *p*-aminobenzophenone yield (**8**) which was condensed with hydrazine hydrate to get the corresponding hydrazone derivatives (**9**). Some of the synthesized compounds have been screened for their antimicrobial activity against various strains of bacteria and fungi.

Keywords: Quinoxaline; Isoxazoles; Antimicrobial activity.

INTRODUCTION

Quinoxaline derivatives are known to possess interesting biological properties which show antibacterial, fungicidal, insecticidal, anthelmic and cytotoxic activities.¹ Recently, that some quinoxaline derivatives inhibit selectively the platelet-derived growth factor (PDGF) receptor kinase, PDGF-dependent DNA synthesis in cells² and its unique inhibitory activity against a panel of human cancer cell lines.³ On the other hand, a quinonimine system (naphthooxazinoquinoxalinone derivative)⁴ is of interest because this system constitutes the structure of several antitumor antibiotics such as actionmycin D,⁵ questiomycin A,⁶ and glycosylquestiomycin.⁷ Also, among the wide variety of heterocycles that have been explored for the viral and antifungal activities of developing pharmaceutically important molecules, cyanopyridines,⁸⁻¹⁰ isoxazoles¹¹⁻¹³ have been similarly investigated. The present investigation, and in continuation of our work on quinoxaline moiety;¹⁴⁻¹⁸ is concerned with the use of 3-bromomethylquinoxalin-2(1*H*)-one for the synthesis of many heterocycles systems of a new type with the aim of investigating their antimicrobial properties.

RESULTS AND DISCUSSION

The starting compound 3-bromomethylquinoxalin-2(1*H*)-one **1** was reacted with *p*-aminoacetophenone **2** in the

presence of pyridine as a basic catalyst to get 3-(*p*-acetyl-anilinomethyl)quinoxalin-2(1*H*)-one **3** which was treated with various aromatic aldehydes to give chalcones **4a-c** (the IR spectra of these compounds showed two absorption bands at 970, 730 cm⁻¹ due to the presence of these compounds as a mixture their trans and cis forms, respectively;²¹ the latter compounds on treatment with malononitrile in the presence of ammonium acetate gave cyanopyridinylquinoxalinones **5a-c** through Michael reaction²² with the elimination of water and hydrogen (Scheme I).

Reaction of chalcones **4a-c** with hydroxylamine hydrochloride in the presence of sodium acetate in acetic acid furnished isoxazolylquinoxalinone derivatives **6a-c** (Scheme II).^{18b,23}

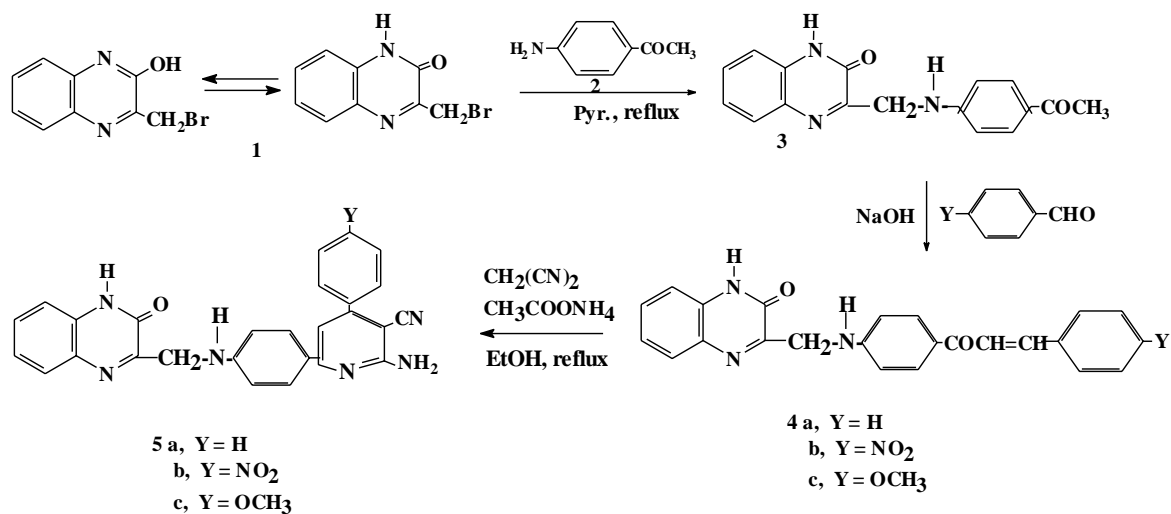
Also the bromo compound **1** was reacted with *p*-aminobenzophenone **7** in basic medium to afford 3-(4-benzoyl-anilinomethyl)quinoxalin-2(1*H*)-one **8**; hydrazinolysis of the latter compound with hydrazine hydrate afforded the corresponding hydrazone **9** (Scheme III).

ANTIMICROBIAL ACTIVITY

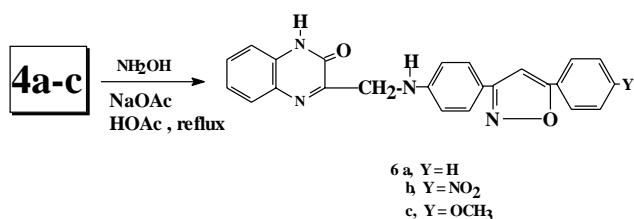
The newly synthesized compounds were screened for their antibacterial activity against three species of different bacteria: Gram positive, *Bacillus subtilis*, *Micrococcus luteus* and Gram negative; *Serratia rhodenil* and three species of fungi: *Aspergillus fumigatus*, *Penicillium chrysogenum*,

* Corresponding author. E-mail: oshmous@acc.aun.eun.eg

Scheme I

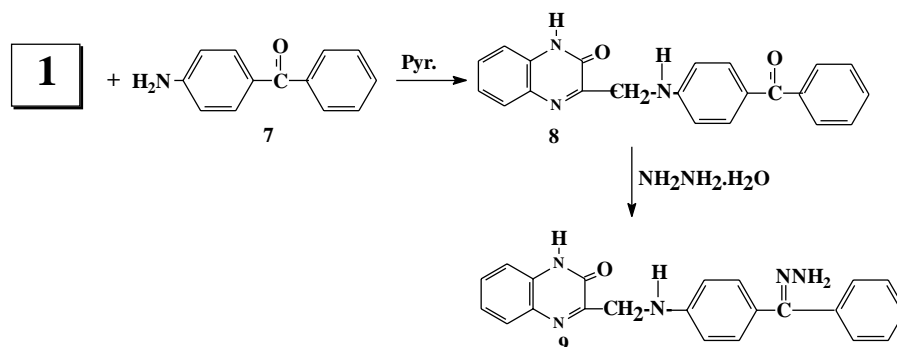


Scheme II



and *Fusarium equiseti* using the filter paper technique^{19,20} measuring the zone of inhibition in mm at 25 µg concentration. The screening results summarized in Table 1 indicated that among of tested compounds, the new compounds showed good growth inhibition against Gram positive bacteria, but only two compounds **3** and **5a** showed good growth inhibition against Gram negative bacteria. However, concerning the antifungal activities only two compounds **3** and **6a** were active against fungi. The results of the antimicrobial activity tests are summarized in Table 1.

Scheme III



EXPERIMENTAL

Melting points were determined on a Gallen Kamp melting point apparatus and are uncorrected. IR spectra were recorded on a Pye-Unicam SP³-100 spectrophotometer using KBr wafer technique. ¹H NMR spectra were measured on a Varian 390-90 MHz NMR spectrometer in the suitable deuterated solvent, using TMS as internal standard. Elemental analyses were performed on a Perkin-Elmer 240 C micro-analyzer and all compounds gave results in an acceptable range. Spectroscopic data are listed in Table 2.

3-Bromomethylquinoxalin-2(1H)-one (1) was prepared according to the literature¹⁴ mp. 245 °C.

3-(p-Acetylanilinomethyl)quinoxalin-2(1H)-one (3)

Method A:

To a mixture of *p*-aminoacetophenone **2** (1.35 g, 0.01 mol) in pyridine (2.3 mL, 0.025 mol), 3-bromomethylquino-

Table 1. Antimicrobial Activity of Some Synthesized Compounds [zone of inhibition in mm]

Comp. No.	Antibacterial activity			Antifungal activity		
	Bacillus subtilus	Micrococcus luteus	Serratia rhodenil	Aspergillus fumigatus	Penicillium chrysogen	Fusarium equiseti
3	7	-	16	-	-	12
4b	-	-	-	-	-	-
5a	7	-	19	-	-	-
5b	-	-	-	-	-	-
6a	7	-	-	-	-	13
6b	11	-	-	-	-	-
8	-	-	-	-	-	-
9	7	-	-	-	-	-

Table 2. Spectroscopic Data of Compounds 3-9

Comp. No	IR (ν cm ⁻¹) / ¹ H NMR δ (ppm)
3	3400, 3200 (2NH), 1710, 1670 (2CO), 1610 (C=N); (DMSO-d ₆): δ 2.4 (s, 3H, CH ₃), 4.2 (s, 2H, CH ₂), 7.2-7.9 (m, 8H, Ar-H), 10.5, 11.4 (2s, 2H, 2NH).
4a	3300, 3150 (2NH), 2960 (CH- α), 1700, 1670 (2CO), 1600 (C=N); (CF ₃ COOD): δ 4.3 (s, 2H, CH ₂), 7.4-8.2 (m, 15H, Ar-H and CH=CH).
4b	3350-3100 (2NH), 2950 (CH- α), 1700, 1670 (2CO), 1610 (C=N); (CF ₃ COOD): δ 4.2 (s, 2H, CH ₂), 7.4-8.1 (m, 14H, Ar-H and CH=CH).
4c	3400-3180 (2NH), 2960 (CH- α), 1705, 1670 (2CO), 1620 (C=N); (CDCl ₃): δ 3.58 (s, 3H, OCH ₃), 4.35 (s, 2H, CH ₂), 7.5-8.0 (m, 14H, Ar-H and CH=CH), 10.5, 11.1 (2s, 2H, 2NH).
5a	3450, 3200 (NH, NH ₂), 2220 (CN), 1670 (CO), (DMSO-d ₆): δ 2.6 (s, 2H, NH ₂), 4.35 (s, 2H, CH ₂), 7.5-8.2 (m, 14H, Ar-H and CH, pyr.), 10.4, 11.0 (2s, 2H, 2NH).
5b	3330 (NH, NH ₂), 2220 (CN), 1670 (CO); (DMSO-d ₆): δ 2.45 (s, 2H, NH ₂), δ 4.2 (s, 2H, CH ₂), 7.2-8.1 (m, 13H, Ar-H and CH, pyr.), 10.5, 11.2 (s, 2H, 2NH).
5c	3180-3420 (NH, NH ₂), 2220 (CN); 1670 (CO), (DMSO-d ₆): δ 2.6 (s, 2H, NH ₂), 3.58 (s, 3H, OCH ₃), 4.25 (s, 2H, CH ₂), 7.3-7.9 (m, 13H, Ar-H), 11.3 (s, 1H, NH).
6a	3200 (NH), 1670 (CO), 1620 (C=N); (DMSO-d ₆): δ 4.4 (s, 2H, CH ₂), 7.4-8.2 (m, 14H, Ar-H), 10.5, 11.3 (s, 2H, 2NH).
6b	3200-3300 (NH), 1670 (CO), (DMSO-d ₆): δ 4.45 (s, 2H, CH ₂), 7.3-8.1 (m, 13H, Ar-H), 10.7 (s, 1H, NH).
6c	3240 (NH), 1670 (CO), 1620 (C=N); (DMSO-d ₆): δ 4.50 (s, 2H, CH ₂), 7.4-8.2 (m, 13H, Ar-H), 10.85 (s, 1H, NH).
8	3220, 3500 (NH), 1720, 1670 (2CO); (CDCl ₃): δ 4.3 (s, 2H, CH ₂), 7.4-8.0 (m, 13H, Ar-H), 11.2 (s, 1H, NH).
9	3250-3420 (NH, NH ₂), 1670 (CO); 1600 (C=N); (CF ₃ COOD): δ 4.4 (s, 2H, CH ₂), 7.6-8.2 (m, 13H, Ar-H).

xalin-2(1*H*)-one **1** (2.39 g, 0.01 mol) was added portionwise with constant stirring. The mixture was refluxed for 3 h in 20 mL of ethanol; the contents were poured into a crushed ice (60 g) and conc. HCl (5 mL) mixture. The product thus separated was filtered and crystallized from ethanol, 67% yield, as yellow crystals mp: 261-262 °C, [C₁₇H₁₅N₃O₂; 293].

Method B:

To a solution of *p*-aminoacetophenone **2** (1.35 g, 0.01 mol) in 10 mL of pyridine, 3-bromomethylquinoxalin-2(1*H*)-one **1** (2.39 g, 0.01 mol) was added portionwise with constant stirring. The mixture was stirred for 30 min. at 0 °C; the con-

tents were poured into a crushed ice (60 g) and conc. HCl (5 mL) was added. The product thus separated was filtered and crystallized from ethanol, 71% yield.

Condensation of (3) with aromatic aldehydes

A mixture of acetyl compound **3** (3.25 g, 0.01 mol) and aromatic aldehydes (0.01 mol) in ethanol (20 mL), and NaOH (40%, 5 mL) was stirred for 5 h. The contents were poured into a crushed ice; the product was isolated by acidification (HCl, 5 mL) and crystallized from ethanol.

4a: 72% yield, as reddish crystals mp: 251-252 °C

[C₂₄H₁₉N₃O₂; 381].

4b: 65% yield; as red crystals mp: 280-281 °C [C₂₄H₁₈N₄O₄; 426].

4c: 69% yield, as pale brown crystals mp: 272-274 °C [C₂₅H₂₁N₃O₃; 411].

Reactions of chalcones (4a-c) with malononitrile

A mixture of chalcone **4** (0.1 mol), malononitrile (0.66 g, 0.1 mol) and ammonium acetate (6.16 g, 0.8 mol) was refluxed in ethanol (30 mL) for 6 h on a water-bath; the cooled contents were then poured into crushed ice with constant stirring, and the resulting solid was washed with water and the residue was crystallized from ethanol to give **5a-c**.

5a: 72% yield, as yellowish crystals mp: 222-224 °C [C₂₇H₂₀N₆O; 444].

5b: 72% yield, as pale orange crystals mp: 259-260 °C [C₂₇H₁₉N₇O₃; 489].

5c: 72% yield, as yellow crystals mp: 227-238 °C [C₂₈H₂₂N₆O₂; 474].

Reactions of chalcones (4a-c) with hydroxylamine hydrochloride

Anhydrous sodium acetate (0.73 g, 0.01 mol) dissolved in a minimum amount of hot acetic acid was added to the solution of hydroxylamine hydrochloride (0.7 g, 0.01 mol) in ethanol (20 mL). This solution was added to a solution of chalcones **4a-c** (0.01 mol) in ethanol (20 mL). The mixture was refluxed for 6 h, concentrated and neutralized with NaOH. The product thus isolated was filtered and crystallized from ethanol to yield **6a-c**.

6a: 78% yield, as pale yellow crystals mp: 219-220 °C [C₂₄H₁₈N₄O₂; 394].

6b: 70% yield, as yellowish crystals mp: 230-232 °C [C₂₄H₁₇N₅O₄; 439].

6c: 69% yield, as yellow crystals mp: 227-228 °C [C₂₅H₂₀N₄O₃; 424].

3-(4-Benzoylanilinomethyl)quinoxalin-2(1H)-one (8)

was prepared from **1** (2.39 g, 0.01 mol) and *p*-aminobenzophenone (1.97 g, 0.01 mol) to give **8**, 77% yield, as white crystals mp: 120-122 °C [C₂₂H₁₇N₃O₂; 355].

3-(4-Benzoylhydrazonelanilinomethyl)quinoxalin-2(1H)-one (9)

A solution of ketone compound **8** (3.55 g, 0.01 mol) and hydrazine hydrate (0.01 mol) in ethanol (30 mL) was heated under reflux for one h; the reaction mixture was cooled, the solid separated was filtered and crystallized from ethanol to give 65% yield of **9**, as yellowish crystals mp: 154-155 °C

[C₂₂H₁₉N₅O; 369].

Received December 3, 2002.

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