

Letter

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Cysteine-Targeted Insecticides against *A. gambiae* Acetylcholinesterase are neither Selective nor Reversible Inhibitors

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KEYWORDS: *acetylcholinesterase; inhibitor; insecticide; Anopheles gambiae; human; selectivity.*

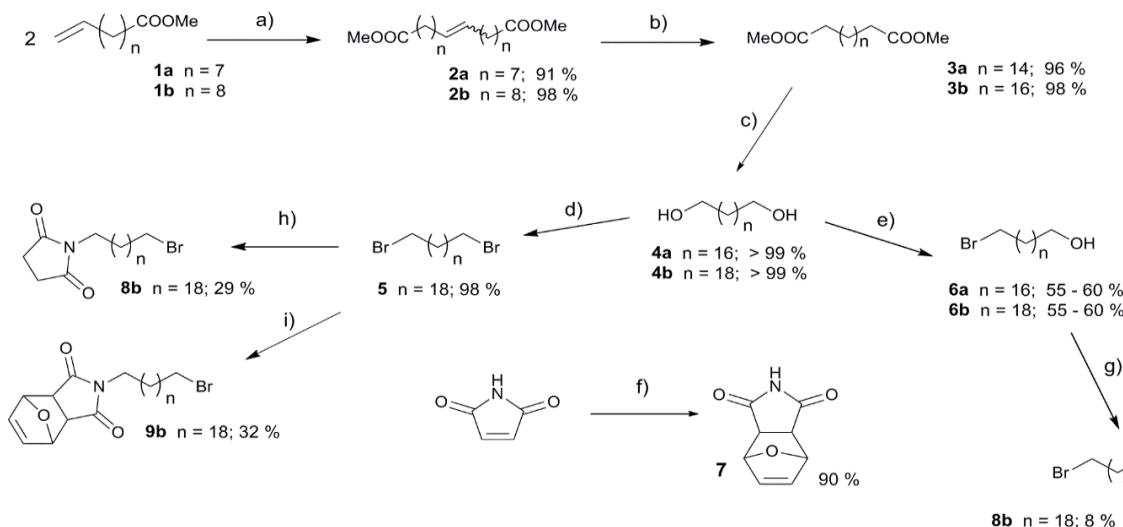
ABSTRACT: Acetylcholinesterase cysteine-targeted insecticides against malaria mosquito *Anopheles gambiae* and other mosquitos have already been introduced. We have applied the olefin metathesis for the preparation of cysteine-targeted insecticides in high yields. The prepared compounds with either a succinimide or maleimide moiety were evaluated on *Anopheles gambiae* and human acetylcholinesterase with relatively high irreversible inhibition of both enzymes, but poor selectivity. The concept of cysteine binding was not proved by several methods, and poor stability was observed of the chosen most potent/selective compounds in a water/buffer environment. Thus, our findings do not support the proposed concept of cysteine-targeted selective insecticides for the prepared series of succinimide or maleimide compounds.

According to the World Malaria Report 2017, there were 216 million cases of malaria with 445,000 deaths in 2016. Almost 2.7 billion people were reported at risk of malaria. Based on these alarming data, malaria remains a huge threat.¹ Chemical insecticides are one of the major tools used for insect elimination and/or prevention of the spread of vector-borne diseases.^{2,3} However, traditional chemical insecticides are often associated with significant toxicity to mammals and beneficial insects such as honey bees or bumble bees.^{4,5}

The principal mechanism of insecticide action is the formation of a covalent bond to acetylcholinesterase (AChE, EC 3.1.1.7), a vital enzyme that is responsible for rapid hydrolysis of the neurotransmitter acetylcholine (ACh).^{6–8} The standard organophosphate and carbamate insecticides covalently block Ser360 (*Anopheles gambiae* acetylcholinesterase, AgAChE) and thus disable the process of ACh hydrolysis in invertebrates as well as in vertebrates. In recent years, several compounds were introduced having very good and selective inhibition potency against AgAChE over human AChE.^{9–11} Cysteine-targeted insecticides aimed at the Cys447 residue (full-length numbering; in *Torpedo californica* the numbering is Cys286) in the peripheral anionic site (PAS) of some insect species have been proposed to overcome insecticide resistance.¹² The uniqueness of this approach depends on the fact that the Cys residue is missing in the mammalian enzyme. Moreover, the Cys residue is sterically hindered in some insects such as *Drosophila melanogaster* and it is accessible to standard insecticides only with difficulty.^{12,13} Thus, it is believed that Cys-targeted insecticides might possess significant selectivity favouring *Anopheles* AChE over mammalian AChE.^{3,11,14}

In this paper we report the optimization and total synthesis of some of the previously reported Cys-targeted insecticides. Insecticides targeted to free cysteine were first proposed by computational studies in 2012 and some of them have already been investigated.^{12,15} Herein, we describe a novel and more straightforward synthetic strategy for Cys-targeted insecticides starting from commercially available reagents. In other recent work, a similar synthetic approach was applied for development of several novel insecticides. The maleimide-containing compounds were prepared as Cys-targeted molecules that should directly bind to Cys447, whereas succinimide-containing compounds were developed for comparison purposes. All of them were then evaluated *in vitro* for their potency to block the action of recombinant *Anopheles gambiae* acetylcholinesterase (AgAChE1), human acetylcholinesterase (HsAChE) and human butyrylcholinesterase (HsBChE).

For the chemical synthesis, the long polymethylene chains were required as the basic scaffolds for the outlined compounds. The methyl esters with terminal double bond (**1a** and **1b**) were chosen as the starting material. Olefin metathesis was applied (instead of using Grignard reactions) with Grubbs reaction yielding dimeric intermediates **2a** and **2b** in nearly quantitative yields (Scheme 1). There was no need to use a stereoselective reaction, because hydrogenation of the double bond took place in the next step. For this reason, a cheaper 1st generation Grubbs catalyst was preferred and used to obtain esters **3a** and **3b** instead of using one of the 2nd generation.¹⁶ Subsequent reduction of the esters provided alcohols **4a** and **4b** in quantitative yields. The ester reduction can be managed by two synthetic routes. Besides lithium aluminium hydride, diisobutylaluminum hydride (DIBAL-H) was efficiently applied to obtain α,ω -dihydroxy compounds **4a** and **4b** (Scheme 1).

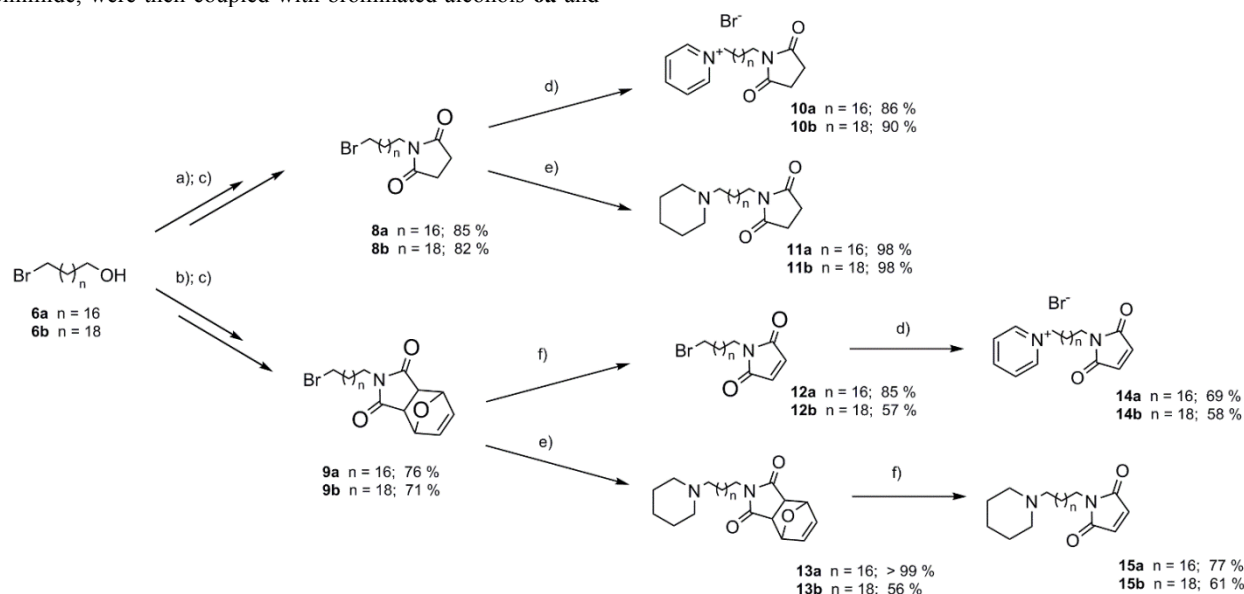


Scheme 1. The application of olefin metathesis to obtain diversely substituted long chain alkanes. Reagents and conditions: a) Grubbs 1st gen. (0.02 eq), reduced pressure up to 2 mbar, RT up to 50 °C; b) Pd(OH)₂ on C (20 %; 0.2 eq), H₂, MeOH/EA (2:1), RT; c) LiAlH₄ 2M solution in THF (2.5 eq), THF, RT; d) NBS (3.0 eq), PPh₃ (3.0 eq), THF, RT; e) HBr 48 % solution in H₂O (3.0 eq), toluene, reflux; f) furan (3.0 eq), dioxane, 90°C; g) DIAD (1.3 eq), PPh₃ (1.3 eq), THF, 0 °C – RT; h) succinimide (1.5 eq), K₂CO₃ (1.5 eq), DMF, 60°C; i) 7 (1.5 eq), K₂CO₃, (1.5 eq), DMF, 60°C.

A challenging step was the attempt at selective protection of one hydroxyl group to successively and selectively proceed towards the formation of alkylating agent. Initial attempts for selective protection were not successful and are described in the Supporting Information (SI chapter 2.4). A convenient approach consisted in the use of HBr as a selective reagent with **4a** and **4b** resulting in mono-bromide intermediates **6a** and **6b**.¹⁷ HBr was also employed for the synthesis in previous work; however, the authors did not fully describe how they obtained mono-bromide intermediates.¹⁵

Further reactions were analogous to those published.¹⁵ Diels-Alder reaction of maleimide with furan provided imide **7**.¹⁸ Imide **7**, and succinimide, were then coupled with brominated alcohols **6a** and

6b. The intermediates **8a/b** and **9a/b** were obtained in two steps via coupling reaction and subsequent bromination with *N*-bromosuccinimide (NBS) in high yields (> 70 %) (Scheme 2). The original procedure was simplified by skipping isolation of the alcohol intermediates (**17a**, **17b**, **18a** and **18b**; described in SI) by direct bromination to get **9a** and **9b**. Succinimide was used for analogous *N*-alkylation to obtain **8a** and **8b**. In this way better yields were achieved and further reduction of maleimide into a succinimide scaffold, as was described in the original work, was avoided.¹⁵



Scheme 2. Synthetic approach for Cys-targeted insecticides. Reagents and conditions: a) succinimide (1.5 eq), K₂CO₃ (1.5 eq), DMF, 60°C; b) 7 (1.5 eq), K₂CO₃, (1.5 eq), DMF, 60°C; c) NBS (1.5 eq); PPh₃ (1.5 eq), THF, RT; d) microwave irradiation (MW), dynamic curve with 100 W and 300 PSI max cap, pyridine (2.0 eq), MeCN, 90°C; e) MW, dynamic curve with 100 W and 300 PSI max cap, piperidine (2.0 eq), K₂CO₃ (3.0 eq), MeCN, 90°C; f) vacuum app. 1 mBar, 130°C.

The final compounds containing the succinimide moiety (**10a/b** and **11a/b**) were obtained using microwave irradiation. The *N*-alkylation of **8a/b** achieved almost quantitative yields in the case of the piperidine analogues (**11a** and **11b**) and around 90 % for the pyridine analogues (**10a** and **10b**; Scheme 2). The final preparation of compounds **14a/b** and **15a/b** containing a maleimide residue required a slightly different approach. In the case of the pyridinium compounds **14a** and **14b**, retro Diels-Alder reaction was used prior to the *N*-alkylation. For the piperidine analogues **15a** and **15b**, the *N*-alkylation of **9a/b** had to precede the retro Diels-Alder reaction to achieve the desired products. Both the *N*-alkylation and the retro Diels-Alder reactions were carried out with good yields over 58 % (Scheme 2). These final steps were slightly distinct from those reported in the original work, by using different conditions and by implementation of microwave (MW) irradiation.¹⁵ Note that the compounds **14a/b** correspond to previously published compounds under codenames **PM18** and **PM20**, and compounds **10b**, **15a** and **11a** are analogous to **PMS20**, **PY18** and **PYS18**, respectively (only the salts are different).¹⁵ The rest of the compounds **10a**, **11b** and **15b** are reported herein for the first time.

Table 1. Inhibitory activity towards *AgAChE1* and *HssAChE*.

Compound		IC ₅₀ ± SEM (μM) ^a		Selectivity index ^b
		<i>AgAChE1</i>	<i>HssAChE</i>	
10a	---	3.60 ± 0.095	3.38 ± 0.143	0.94
10b	PMS20	2.06 ± 0.040	3.39 ± 0.338	1.65
11a	PYS18	9.97 ± 0.213	21.48 ± 0.843	2.15
11b	---	25.87 ± 1.704	29.30 ± 2.536	1.13
14a	PM18	0.465 ± 0.020	3.26 ± 0.194	7.00
14b	PM20	0.810 ± 0.060	2.03 ± 0.138	2.51
15a	PY18	0.545 ± 0.028	3.52 ± 0.523	6.45
15b	---	2.65 ± 0.344	39.50 ± 4.598	14.89
	paraoxon	0.0072 ± 0.0004	0.0084 ± 0.0004	1.16
	bendiocarb	0.0024 ± 0.0003	0.0309 ± 0.0008	12.70
	carbofuran	0.0062 ± 0.0008	0.0221 ± 0.0012	3.59

^a IC₅₀ values measured by modified Ellman's assay 15 min after introduction of inhibitor; ^b Selectivity for *AgAChE1* is determined as ratio IC₅₀ (*HssAChE*) / IC₅₀ (*AgAChE1*)

The crucial mechanism of action should be based on the formation of a covalent bond between the maleimide moiety and Cys447 in *AgAChE1*, resulting in irreversible inhibition. A detailed *in vitro* evaluation was made by Dou *et al.* 2013.¹⁵ For *in vitro* purposes, recombinant *AgAChE1* was prepared and purified in our labs (see SI Figure S1). The kinetic parameters of the enzyme were evaluated (see SI Figure S2). The *K_M* constant of *AgAChE1* was found to be 54.63 μM for acetylthiocholine iodide (ATCh) as a substrate, which is in very good correlation with the previously published data.^{19,20} To verify the enzyme and assay results, the standard inhibitors including paraoxon and carbamates (bendiocarb, carbofuran) were evaluated.

The inhibitory ability of the prepared insecticides towards *HssAChE* and *AgAChE1* was determined using the spectrophotometric method described by Ellman *et al.* (Table 1).²¹ *HssBChE* was used as a common off-target for AChE inhibitors. As shown in the Supporting Information, the *HssBChE* inhibition levels observed for the prepared compounds were over 60 μM (see SI Table S1). Maleimides **14a**, **14b** and **15a** were found to be the most potent inhibitors of *AgAChE1*, and several-fold more potent than corresponding succinimides. For *HssAChE*, maleimides **14a**,

14b and **15b** are similar in potency to the related succinimides **10a**, **10b** and **11b**. However, maleimide **15a** was found to be several-fold more potent than its corresponding succinimide **11a** for *HssAChE*. As *HssAChE* does not contain a free cysteine residue which could react with maleimide **15a**, the inhibitory ability of this compound was found unexpected and cannot be rationalized by interaction with Cys447.

The selectivity for *AgAChE1* was rather poor and most compounds were found relatively potent inhibitors of *HssAChE*. Only in the case of **15b** was the selectivity towards the insect enzyme found to be over 10-fold. However, the prepared compounds possess only poor inhibitory activity towards *AgAChE1* compared to paraoxon or bendiocarb.

Further, time-dependent inhibition was used to determine differences between both enzymes. A rapid decrease in enzyme activity was observed within 3 min, and the activity remained almost steady for 15 minutes for both *HssAChE* and *AgAChE1* (Figure 1 and 2). Surprisingly, increased potency of maleimides vs succinimides for *AgAChE1* is not necessarily evidence of covalent binding, since the same phenomenon is seen for *HssAChE*. Additionally, the rate of enzyme-inhibitor complex formation could be explained by noncovalent interaction between pyridine/piperidine parts of the tested compounds with both enzymes. Such interactions are in general very rapid and there is probably no covalent bond formation involved in this short time period.^{22,23} The previously-published time-dependency¹⁵ did not prove covalent binding of the compounds, but depicted only a slower or faster rate of binding to the enzyme.

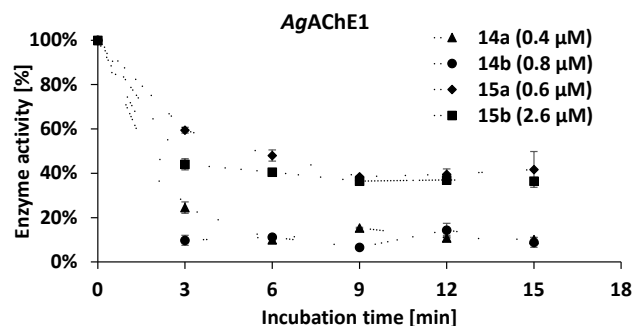


Figure 1. Time-dependent inhibition of maleimide derivatives on *AgAChE1*.

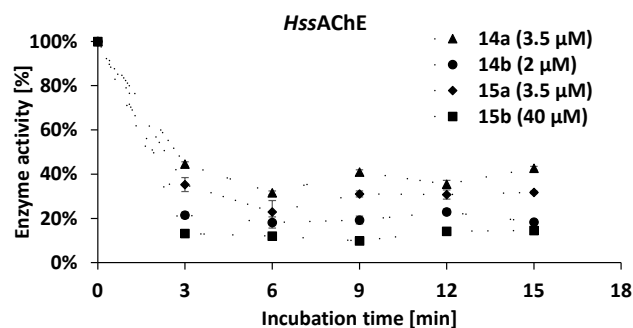


Figure 2. Time-dependent inhibition of maleimide derivatives on *HssAChE1*.

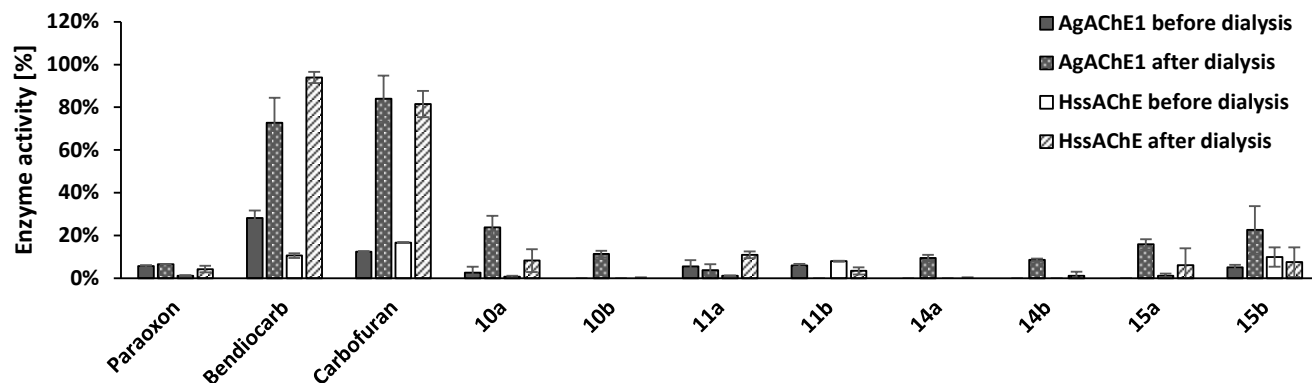


Figure 3. Reversibility of inhibition on *AgAChE1* and *HssAChE* using the dialysis method.

Thus, the mechanism of interaction is truly crucial for this series of compounds. We have tested the reversibility of the reaction using the removal of the selected insecticides by dialysis (Figure 3). Apparently, none of the tested compounds showed reversible action of inhibition for either *AgAChE1* or *HssAChE*. Based on the previous work,¹⁵ maleimide-harboring compounds were presumed to act as irreversible inhibitors and succinimide compounds were prepared only as the controls to prove insect-specific cysteine involvement in covalent bond formation. The presented results are showing different findings, with all succinimide- or maleimide-containing compounds found to be irreversible-acting inhibitors when incubated 30 minutes with enzymes and further dialysed overnight (Figure 3). The irreversible inhibition was shown towards *HssAChE* as well, although in this enzyme the Cys is substituted by Phe (Figure 4). It should also be noted that maleimide is able to form a covalent bond with other amino acid residues (except free cysteine) as was formerly reported.²⁴ However, the most surprising fact is that all succinimides (**10a/b**, **11a/b**) were found to be irreversible inhibitors of both enzymes with recovery less than 20%. Such finding could be plausibly explained by e.g. precipitation or denaturation of the enzymes by the detergent-like succinimides.

Additionally, a molecular docking study was conducted. Compound **15a** was selected from all the tested compounds for its valuable and higher inhibition activity on *AgAChE1* and moderate effect on *HssAChE/BChE*. Two protein structures were used (PDB IDs: 6ARX in Figure 4 and SI Figure S3; 5YDH only in SI Figure S3) to reveal and better understand the binding interactions between ligand **15a** and the *AgAChE1* active site.²⁵ The top docking-scored pose of **15a** was found different in each of the used proteins. For 6ARX with G119S mutation (Figure 4), ligand **15a** resulted in binding of the maleimide moiety close to the catalytic triad and interaction with tyrosine residues (Tyr489 and Tyr493), with the piperidine ring attached at the rim of the gorge (Phe449, Phe 490 and Tyr494). Differently in the 5YDH protein (SI Figure S3), the maleimide moiety was found interacting with the rim of the *AgAChE1* gorge (Cys447, Phe449, Tyr494), with the piperidine scaffold attached close to the active site (Trp245, Tyr291, Glu359). The literature was checked to elucidate these contradictory results for the involvement of tyrosine residues in maleimide binding, but there were no literature data supporting the hypothesis of covalent bond formation between a hydroxyl from either tyrosine or serine residues, and thus different binding of ligand **15a** in 6ARX (Figure 4). Apart from sulfhydryl cysteine residues, only primary amines can be expected to possibly

conjugate with maleimide at pH values higher than 7.5; however, for **15a** this is not the case.²⁶

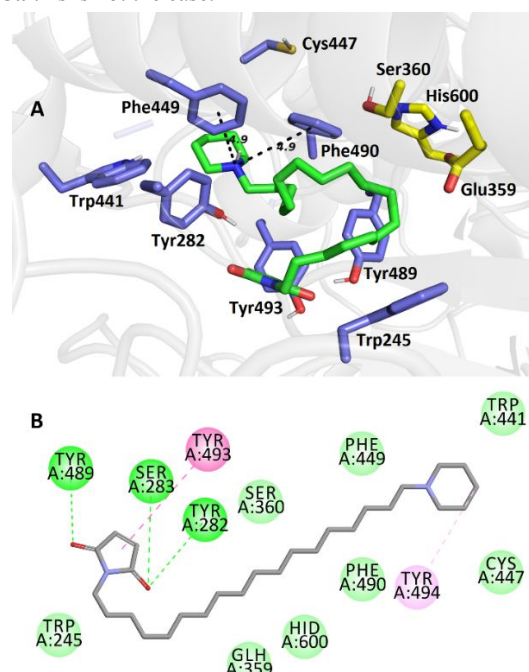


Figure 4. Superimposition of compound **15a** in the *AgAChE1* active site (PDB ID: 6ARX, G119S mutation). The close-up view for ligand is presented as three-dimensional (4A; **15a** is presented in green, important amino acid residues in blue, and catalytic triad residues in yellow) and two-dimensional (4B) diagrams.

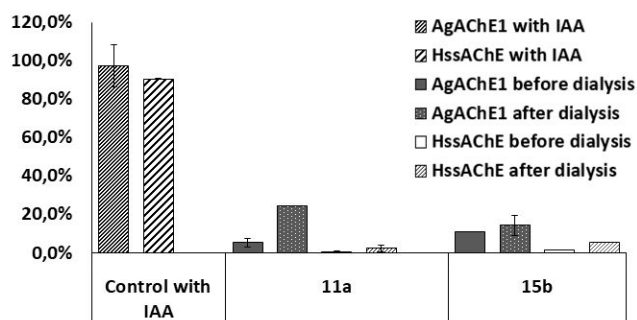


Figure 5. Reversibility of inhibition on *AgAChE1* and *HssAChE* using dialysis method after iodoacetamide (IAA) pretreatment.

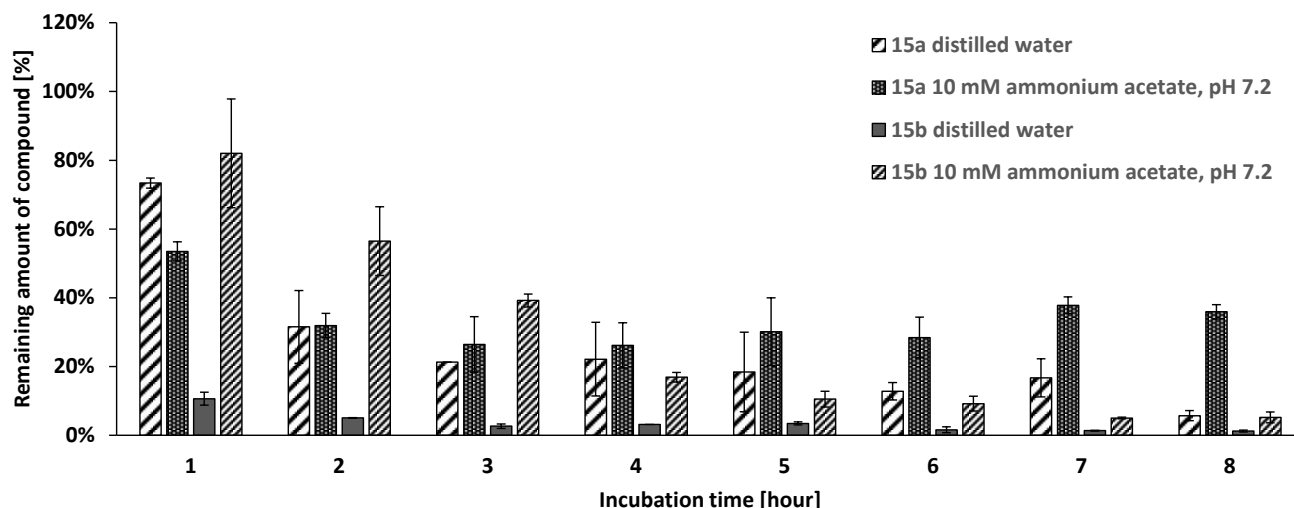


Figure 6. Stability determination of **15a** and **15b** over time in distilled water or buffer.

The contradictory finding within *in vitro* evaluation and molecular docking raised the question of binding of the presented maleimide/succinimide molecules to *AgAChE1* or *HssAChE*. The enzymes were treated with 100 molar excess of iodoacetamide (IAA) to determine the affinity of the presented inhibitors towards Cys447. IAA is a commonly used blocking agent of free thiol groups.²⁷ Although IAA is not strictly thiol selective (as well as maleimide), the reaction should result in *AgAChE1* enzyme that is resistant to irreversible inhibition by maleimide-based inhibitors. However, both enzymes treated with IAA were found irreversibly inhibited by selected compounds **11a** (succinimide scaffold) and **15b** (maleimide scaffold; Figure 5). Apparently, these results indicate that irreversible inhibition is most probably not mediated by the free cysteine residue Cys447 in *AgAChE1* as was formerly reported.¹⁵ The maleimide or succinimide moiety can also interact with other amino acid residues, e.g. with histidine from the catalytic triad of *AgAChE1* or *HssAChE* and thus may irreversibly inhibit both enzymes this way.

From the point of practical use for insecticidal purposes, stability of the most potent and most selective compounds for *AgAChE1* (**15a** and **15b**) was inspected by incubation at 40°C (Figure 6). Fast degradation of both compounds was observed in a water or buffer environment, with the water degradation found to be apparently more rapid. In particular, compound **15b** (the most selective for *AgAChE1*) was almost completely degraded within one hour in water. Both compounds were more stable in ammonium acetate buffer, but a degradation ratio of over 50% was observed in 3 hours.

In summary, eight compounds with a succinimide or maleimide moiety were prepared and evaluated as cysteine-targeted insecticides. Some of the compounds were shown as potent *AgAChE1* inhibitors, but they were found to be potent inhibitors of *HssAChE* with a limited selectivity ratio as well. The time-dependent inhibition of the compounds was very fast, resulting in inhibition of both enzymes within minutes. The presented compounds acted in an irreversible manner to both *AgAChE1* and *HssAChE*. Molecular docking on two distinct *AgAChE1* proteins indicated contradictory results showing the possibility of different interactions than the proposed Cys-targeted binding. This phenomenon was further confirmed by cysteine blocking treatment of both enzymes, after which they were still found irreversibly inhibited by the tested compounds, suggesting a different mode of irreversible binding. The potential use of the tested compounds might be limited by the relatively low stability depicted within the degradation assay. Thus, our findings do not support the proposed concept of cysteine-targeted selective insecticides. We presume

that this class of molecules will not be effective as insecticides due to the afore-mentioned findings.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: *(to be added by editorial office)*.

General synthetic methods; General procedure and spectral data; HPLC analysis of final compounds; Recombinant *AgAChE1* production and characterization; Enzymatic assays; Molecular modelling studies; Time-dependent stability measurement (PDF).

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Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript. ‡These authors contributed equally.

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ABBREVIATIONS

ACh acetylcholine; AChE acetylcholinesterase; ATCh acetylthiocholine iodide; *AgAChE1* *Anopheles gambiae* acetylcholinesterase; DIBAL-H diisobutylaluminium hydride; *HssAChE* human acetylcholinesterase; *HssBChE* human butyrylcholinesterase; IAA iodoacetamide; IC₅₀ median inhibitory concentration; MW microwave irradiation; NBS *N*-bromosuccinimide; PAS peripheral active site; TBAF tetrabutylammonium fluoride; TBDPSCI *tert*-butyl(chloro)diphenylsilane.

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