

ORIGINAL PAPER

Polar constituents of *Ligustrum vulgare* L. and their effect on lipoxygenase activity

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The present work summarizes results of isolation and identification of polar constituents of the methanolic extract of *Ligustrum vulgare* L. leaves and of the evaluation of inhibiting activity of selected isolates on rat lung cytosol fraction lipoxygenase. Six different compounds were isolated from the ethylacetate and butanol portions of the methanolic extract (hydroxytyrosol and its glucoside, ligustroflavon, oleuropein, acteoside, echinacoside). The inhibitory activity of oleuropein, echinacoside and the water infusion of *Ligustrum vulgare* leaves tested on LOX was expressed as IC₅₀. Kinetic parameters ($K_{\rm M}$, $V_{\rm max}$) and type of inhibition were determined. As the most effective in competitive inhibition of LOX, oleuropein was proved. (© 2011 Institute of Chemistry, Slovak Academy of Sciences

Keywords: Ligustrum vulgare L., oleuropein, echinacoside, lipoxigenase activity

Introduction

Different species of the *Ligustrum* genus have been used in traditional Chinese and Japanese medicine due to its liver-protecting (Yim et al., 2001), antiviral (Ma et al., 2001), and anti-mutagenic activities (Shoemaker et al., 2005). In the folk medicine of Azerbaijan, common privet leaves (*Ligustrum vulgare*) are used in hypertension therapy (Hammermann et al., 1971), which has been supported by recent studies of hypotensive and diuretic effects of common privet since different extracts of this plant act as dual angiotensin converting enzymes and neutral endopeptidase inhibitors (Kiss et al., 2008). Common privet leaves are currently used as an oropharyngeal antiinflammatory agent in the ethnomedicine of southern Italy and the plant is regarded as an antirheumatic agent in Cyprus. In Anatolia (Turkey), fresh plant leaves are still chewed to cure aphtae (Pieroni & Pachaly, 2000a). Most of these diseases can be connected with the reactive oxygen species balance in all human tissues. Phenolic antioxidants are recognized as the main active principles (Nagy et al., 2006, 2009); however, other classes of secondary metabolites such as essential oils or polysaccharides also exhibit antioxidant properties (Stojanovic-Radic et al., 2010; Hromádková et al., 2010, Jiménez et al., 2010).

A variety of redox active compounds have been

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identified as inhibitors of lipoxygenases. Lipoxygenases (EC 1.13.11.12) (LOXs) catalyze dioxygenation of polyunsaturated fatty acids with a *cis*, *cis*-1,4-pentadiene configuration to their corresponding hydroperoxide derivatives (Bezáková et al., 1996). LOX is the key enzyme in the biosynthesis of leukotrienes playing an important role in the pathophysiology of several inflammatory diseases since products of lipoxygenase catalyze oxygenation as hydroxyeicosatetraenoic or hydroperoxy eicosatetraenoic acid, leukotrienes, and lipoxins seem to be involved in inflammatory reactions (rheumatoid arthritis and psoriasis). There is good evidence that leukotrienes are mediators of asthmatic responses, glomerular nephritis, myocardial ischemia, and cancer (Young, 1999).

The aim of the present study was the isolation and identification of constituents of *Ligustrum vulgare* L. leaves and the evaluation of their potential antiinflammatory activity. From the anti-inflammatory test systems, the in vitro assay of inhibitory activity on the rat lung cytosol fraction lipoxygenase was chosen.

Experimental

Leaves of *Ligustrum vulgare* L. were collected during September 2006 in Arborétum Mlyňany, Institute of Dendrobiology, Slovak Academy of Sciences, and dried at room temperature. Samples were identified by Dr. Tomaško (Arborétum Mlyňany) and a voucher specimen is deposited there.

HPLC analysis of samples was performed using a Hewlett-Packard HP 1100 (Agilent Technologies, USA) liquid chromatograph with a diode-array detector (DAD; monitoring wavelengths: 230 nm, 254 nm, 290 nm, and 334 nm). Compounds were separated on a 50 mm \times 2.1 mm, 1.8 μ m particle Eclipse XDB C18 column (Agilent Technologies, USA) using a gradient prepared from acetonitrile (component A) and 0.2 % of aqueous HCOOH (component B). The gradient was: 0-36 min, 10 % A + 90 % B; 36-40 min, 100 % A. The flow rate was 0.3 mL min⁻¹ and the column temperature was 30 °C. Mass spectra were acquired with an Agilent 1100 LC/MSD Ion Trap VL (Agilent Technologies, USA) in the negative-ion mode. Compounds were separated as described for HPLC-DAD. The mass spectrometer settings were: MWD 254, target mass of 300, compound stability of 35 %, trap drive level of 100 %, and MS-MS. UV spectra were recorded on-line using DAD during the HPLC analysis and a Specord UV-VIS (C. Zeiss, Jena, Germany) in methanol after an addition of specific diagnostic reagents in accordance with literature (Mabry et al., 1970). NMR spectra were recorded on a Varian Inova 600 MHz spectrometer (Agilent Technologies, USA); CD_3OD was used as the solvent. Column chromatography (CC) was performed on silica gel (Silpearl, particle size of 0.40–0.063 mm; Kavalier, Votice, Czech Republic), Sephadex LH-20 (Pharmacia Biotech, Uppsala, Sweden), and polyamide S6 (particle size of 0.05–0.16 mm; Riedel–De Haen, Seelze-Hannover, Germany). TLC was performed on silica gel (Silufol UV 254) and polyamide (DC-Alufolien Polyamid 11 F 254; layer thickness of 0.15 mm) plates from Merck (Darmstadt, Germany). TLC detection reagents (Naturstoffreagent A, polyethylene glycol 4000, and aniline phthalate) were prepared according to www.cob.uha/site%20cot/toolbox.php_files/TLC %20bible.pdf (Retrieved 16 December 2010).

For HPLC identification, standards of echinacoside (PhytoLab, Germany), oleuropein, apigenin, glucose, and rhamnose were purchased from C. Roth (Karlsruhe, Germany). Acteoside was isolated from *Paulow-nia tomentosa* fruits (Šmejkal et al., 2008) and kindly provided by the authors.

Powdered dried leaves of Ligustrum vulgare L. (442 g) were extracted using methanol in a Soxhlet apparatus until the extract became colorless. After filtration, the methanol extract was evaporated to dryness by a vacuum rotary evaporator. The residue (216.8 g)was then suspended in distilled water and then partitioned successively with chloroform (30.4 g, 14 %), diethyl ether (2.4 g, 1.11 %), ethyl acetate (28.7 g, 1.11 %)13.24 %), and butan-1-ol (54.3 g, 25 %), leaving the H_2O portion (101 g, 46.6 %). The butan-1-ol portion (54.3 g) was adsorbed on silica gel and subjected to column chromatography on a silica gel column with chloroform-methanol mixtures of increasing polarity as the eluents (0-100 %). A total of 348 fractions (each of approximately 100 mL) were collected. Fractions with similar TLC profiles were pooled to provide combined fractions (I-XV). Fraction VII was further purified on a Sephadex LH-20 column using propan-2-ol as the eluent. According to the TLC assay, twelve combined subfractions were obtained (each of approximately 50 mL). Fraction 6 from this separation was submitted for further CC on polyamide using $CHCl_3/MeOH/butan-2$ -one ($\varphi_r = 12:2:1$) as the eluent. This resulted in the isolation of compound I(10.2)mg). Fraction XIV was purified on a Sephadex LH-20 column using methanol as the eluent. According to the TLC assay, seven combined subfractions were obtained (each of approximately 100 mL). Fraction 6 from this separation was submitted for further CC on polyamide using CHCl₃/MeOH/butan-2-one ($\varphi_{\rm r} = 10$ (2:1) as the eluent. Totally, 409 fractions were collected (approximately 1 mL of each). Fractions 40-100 gave 7.2 mg of compound II. Fractions 119-206 were combined and divided by SPE on column C-18 (Varian Mega Bond Elut, Agilent Technologies, USA) using a LiChrolut (Merck, Germany) equipment and an H₂O/MeOH ($\varphi_r = 9:1$) mixture as the eluent system with an increasing MeOH portion. This separation resulted in the isolation of compound III (4.5 mg). The ethyl acetate portion (28.7 g) was submitted to column chromatography on a Sephadex LH-20

with propan-2-ol as the eluent. Totally, 112 fractions were collected (about 100 mL of each). Collected fractions were combined to ten fractions according to their TLC profiles. Fraction IV was repeatedly purified on a silica gel column using CHCl₃/MeOH ($\varphi_r = 9 : 1$), which resulted in the separation of compound IV (5.5 mg). Fraction V was purified on a silica gel column CHCl₃/MeOH ($\varphi_r = 9 : 1$) and compound V was isolated (56.8 mg). Fraction IX was purified on a silica gel column with CHCl₃/MeOH ($\varphi_r = 9 : 1$) as the eluent. From this separation, compound VI (3.6 mg) was isolated. The amounts of these substances do not correspond with the real amounts in the plant.

In enzymatic hydrolysis, 2 mg of compound I were dissolved with 2 mg of β -glucosidase in 5 mL of a methanol/H₂O mixture ($\varphi_r = 0.5$) and left standing for 24 h at room temperature. Then, methanol was evaporated on a vacuum rotary evaporator and the water phase was used for TLC of glycoside and agly-con.

In acidic hydrolysis; compound II (2 mg) was dissolved in methanol (2 mL) and heated under reflux with 11 % HCl for 3 h. Aglycone was extracted with chloroform and the aqueous phase was neutralized by the anion-exchange resin Dowex 2×8, (Fluka, Germany) and submitted for TLC comparison with saccharide standards.

The activity of LOX was monitored as an increase in the absorbance at 234 nm, which reflects the formation of the hydroperoxylinoleic acid. Inhibitory effect of the tested compounds was expressed as IC_{50} (Table 1).

The cytosolic fraction from rat lungs (Wistar rat, male 180 g), as a source of LOX, was isolated according to a procedure reported by Kulkarni et al. (1992). Briefly, rat lung homogenate was centrifuged at $1000 \times g$ for 5 min. The pellet obtained contained unbroken cells and the debris was discarded. The resulting supernatant was centrifuged at 10 000 $\times g$ for 15 min to obtain the mitochondrial fraction. The postmitochondrial supernatant was further centrifuged at 100 000 \times g for 60 min to obtain microsomes and cytosol. This fraction was further purified by ammonium sulfate precipitation (60 %), chromatography on Sephadex G-150 (Pharmacia, Sweden), and on Phenyl-Sepharose CL-4B (Pharmacia, Sweden). The purified enzyme was used for LOX activity determination. The protein content in the enzyme preparation was estimated by the method of Bradford (1976).

Linoleic acid (99 %, Sigma, USA) was used as a substrate prepared in solubilized state as described (Kemal et al., 1987) in the concentration of 0.2143×10^{-5} - 0.7143×10^{-5} mol L⁻¹.

For the LOX activity assay, a UV/VIS Spectrometer Perkin–Elmer Lambda 35 (USA), was used. The reaction medium contained a 50 mM Tris-HCl buffer (pH 7.0), 5 μ L of the enzyme protein and solubilized linoleic acid. The compounds were tested in the final

concentration range of 2.5×10^{-5} - 15×10^{-5} mol L⁻¹.

Results and discussion

Compound I was isolated as a white powder, λ_{max} = 280 nm. Mass spectra of compound I show a molecular ion at m/z = 316 with a fragment at m/z = 136 corresponding to aglycone after the elimination of hexose and water (316-162 = 154; 154-18 = 136). ¹³C NMR spectra confirmed the presence of 14 carbonyl atoms. A comparison of the obtained data with data available in literature (Franzyk et al., 2004; Nagao et al., 2001) allows to identify the isolated compound as 2- $(3,4-dihydroxyphenyl)-ethyl-\beta-D-glucopyranoside (hy$ droxytyrosol glucoside). Enzymatic hydrolysis of compound I gave aglycon and glucose, confirmed after the TLC analysis on silica gel with a glucose standard $(EtOAc/HCOOH/CH_3COOH/H_2O, \varphi_r = 100 : 11 :$ 11 : 23, developed four times, aniline-phthalate detection). It is interesting that due to high polarity of aglycone, its extraction with chloroform was not successive and its TLC analysis had to be realized from the water portion.

Compound II was isolated as a yellow powder with two absorption maxima at 270 nm and 330 nm indicating a flavonoid structure of the isolated compound. Specific diagnostic reagents shifts (NaOMe, AlCl₃, NaOAc) confirmed the presence of such a structure. According to TLC, acid hydrolysis of compound II yielded apigenin as aglycone (benzene/ethylacetate/HCOOH, $\varphi_{\rm r} = 7:3:1$, Naturstoffreagent A + polyethylene glycol) with glucose and rhamnose as the saccharide moiety (TLC as for compound I). This structure is supported by mass spectra which show a molecular ion peak at m/z = 724and a fragment at m/z = 270 (aglycon). A difference of 454 atomic mass units (AMU) (724–270 =454) indicating the presence of one molecule of hexose and two molecules of deoxyhexose in the structure of the isolated compound (162 + 146 + 146 = 454). ¹H and ¹³C NMR are in good agreement with the spectra of ligustroflavone (Pieroni & Pachaly, 2000b), which allowed the identification of the isolated compound as ligustroflavone (apigenin-7- $O-\beta$ -(2",6"-di- α -D-rhamnopyranosyl)-D-glucopyranoside).

Compound III exhibited maxima at 220 nm, 250 nm, 290 nm, and 330 nm in UV spectrum. MS spectra in the negative mode show a molecular ion at m/z = 785 [M—H]⁻ and fragments at m/z = 623 (785 – 162 = glucose – 18 – 146; rhamnose – 18), m/z = 179 (caffeic acid), m/z = 135 (hydroxytyrosol; 154 – 18). A direct TLC and HPLC comparison with the standard of echinacoside (PhytoLab, Germany) confirmed the structure of compound III as echinacoside.

Compound IV, on TLC after spraying with H₂SO₄ in ether, exhibited a characteristic purple color similar to compound I with $\lambda_{\text{max}} = 280$ nm. Direct

Substrate concentration/ $(10^{-5} \text{ mol } \text{L}^{-1})$	Parameter	Oleuropein	Echinacoside	L. vulgare L.
0.214	$K_{ m M}/({ m mmol}\ { m L}^{-1})$	0.193	0.214	0.212
0.357		0.192	0.196	0.212
0.500		0.191	0.213	0.211
0.643		0.189	0.215	0.207
0.714		0.189	0.186	0.192
	$V_{ m max}/(\mu { m mol \ s^{-1} \ L^{-1}}) \ { m IC}_{50}/(\mu { m mol \ L^{-1}})$	59.9	23.3	286
	$\mathrm{IC}_{50}/(\mu\mathrm{mol}\ \mathrm{L}^{-1})$	82.2	296	0.388^{a}

Table 1. Inhibitory activity of isolated compounds and their kinetic parameters tested on LOX

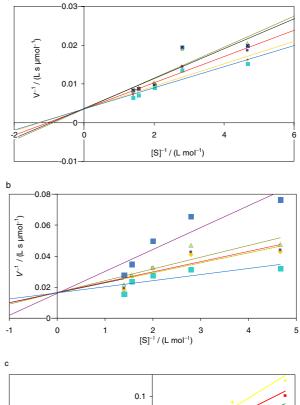
a) mg L^{-1} .

TLC comparison of the isolated compound (silica gel; CHCl₃/MeOH, $\varphi_{\rm r} = 9:1$) with a glycon obtained by enzymic hydrolysis of hydroxytyrosol glucoside (compound *I*) confirmed the structure of hydroxytyrosol for compound *IV*.

Compound V was isolated as a white powder with maxima at 230 nm and 280 nm in the UV spectrum. In mass spectra of compound V, a molecular ion at m/z = 540 is present. A direct TLC and HPLC (Rt, DAD, MS/MS fragmentation) comparison with the standard of oleuropein (C. Roth, Karlsruhe, Germany) allowed the identification of compound V as oleuropein.

Compound VI exhibited similar TLC properties as compound III (blue fluorescence under UV light with higher $R_{\rm f}$ compared to compound III, silica gel EtOAc/HCOOH/CH₃COOH/H₂O, $\varphi_{\rm r} = 100 : 11 :$ 11 : 23). UV maxima of compound VI are identical with those of compound III. MS spectra in the negative mode show a molecular ion at m/z = 623 [M— H]⁻. According to the obtained results and the direct TLC (silica gel as mentioned above, polyamide, CHCl₃/MeOH/butan-2-one, $\varphi_{\rm r} = 12 : 2 : 1$) and HPLC comparison with a standard, compound III was identified as acteoside.

It is known from literature (Tattini et al., 2004) that Ligustrum vulgare L. is an abundant source of iridoids and phenolics. HPLC-DAD profiles show the complex polyphenol composition of L. vulgare leaves which include secoiridoids and tyrosol derivatives, and both hydroxycinnamates and flavonoid glycosides. Oleuropein, ligustaloside A and ligustaloside B, and ligstroside constitute the relevant class of secoiridoids. The flavonoid composition of L. vulgare leaves comprises quercetin 3-O-rutinoside, luteolin glucosides, namely luteolin 7-O-glucoside and luteolin 4'-O-glucoside, and both apigenin 7-O-glucoside and apigenin 7-O-rutinoside. Hydroxycinnamates are represented by p-coumaric acid and echinacoside. Appreciable amounts of hydroxytyrosol and hydroxytyrosol glucoside were also detected in L. vulgare leaves. The distribution of phenylpropanoid glycoside acteoside (Scogin, 1992) is restricted to a constellation of families which may constitute a natural ordinal grouping. This distribution includes many families traditionally clustered in a Scrophulariales-Lamiales com-



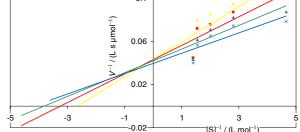


Fig. 1. Kinetic types of LOX inhibition by the tested extracted and isolated compounds: a) echinacoside, b) echinacoside, c) L. vulgare L.

plex containing Retziaceae, Oleaceae, Stilbaceae, and Hippuridaceae, excluding Hydrostachyaceae; the presence of this complex was confirmed in different *Ligustrum* species (Wong et al., 2001; Pan et al., 2002).

The aim of this study was also the evaluation of the

inhibitory effect of the isolated compounds and water infusion from the leaves of L. vulgare on the activity of lipoxygenase isolated from the cytosolic fraction of rat lungs. Data are reported as IC_{50} values (Table 1). Water infusion and two isolated compounds, oleuropein and echinacoside, exhibited remarkable inhibitory effect on LOX. An iridoid derivative, oleuropein, was proved to be the most effective. This is in good correlation with literature data. Oleuropein is one of the most active substances in virgin olive oil (de la Puerta et al., 1999). In intact rat peritoneal leukocytes stimulated with calcium ionophore, oleuropein, tyrosol, hydroxytyrosol, and caffeic acid from virgin olive oil, leukotriene B4 generation was inhibited at the 5-lipoxygenase level.

Fig. 1 shows the determination of the character of inhibition in the presence of inhibitors. Kinetic parameters $K_{\rm M}$ and $V_{\rm max}$ of Michaelis–Menten equation were expressed from Lineweaver–Burk equation: $V^{-1} = K_{\rm M} V_{\rm max}^{-1} [S]^{-1} + V_{\rm max}^{-1};$ where $[S]/({\rm mol}\ {\rm L}^{-1})$ is a substrate concentration; they demonstrate competitive inhibition of all tested inhibitors. This activity can be potentiated with the activity of flavonoids, exhibiting remarkable activity against LOX (Bezáková et al., 2007), isolated from this plant previously (Šeršeň et al., 2006; Mučaji et al., 2006) or on the classical pathway of the complement system (Pieroni & Pachaly, 2000a). These results clearly indicate that the tested samples contribute to the inhibitory activity of L. vulgare water infusion used for anti-inflammatory purposes in traditional medicine. However, the potential therapeutic value and the real clinical efficacy of this plant should be proved by further in vitro and in vivo experiments.

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