

# Au(I)/(R)-BINOL—Ti(IV) Concerted Catalyzed Asymmetric Cascade Cycloaddition Reaction of Arylalkynols

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A symmetric synthesis has been widely used in many fields of chemistry. Among them, asymmetric catalysis is the most fashionable method, and it is devoted to the development of chiral catalysts to convert prochiral and racemic substrates into valuable chiral synthesis blocks. Thus, plenty of chiral catalysts have been exploited and synthesized to date. For example, chiral diols are among the most privileged scaffolds in asymmetric catalysis, especially for the derivatives of the chiral 1,1'-bi-2-naphthol (BINOL) skeleton.<sup>1</sup> Of course, it is highly desirable to directly use commercially available or easily synthesized chiral catalysts, as they will provide potential application in industry.

On the other hand, oxo-bridged heterocyclic skeletons are ubiquitous in natural products and bioactive molecules.<sup>2</sup> Therefore, the development of new and efficient methods for the construction of oxo-bridged heterocyclic compounds has attracted increasing attention, especially the development of intermolecular reactions, which are more appealing because of their flexible substrate candidates.<sup>3</sup> However, to the best of our knowledge, there are scarce reports on examples of the asymmetric synthesis of oxo-bridged bicyclic compounds. Recently, bimetallic relay catalysis has shown unique potential in modern asymmetric synthesis.<sup>4</sup> With the combination of a chiral catalytic system and a metal catalyst to activate two different substrates, highly efficient asymmetric chemical transformations can be achieved to afford complex chiral molecules in high yields with high enantio- and diastereoselectivities from simple starting materials. For example, Feng's research group developed an efficient asymmetric synthesis of oxo-bridged oxazocines through a Rh<sup>II</sup>/Zn<sup>II</sup> relay catalytic [4 + 3] cycloaddition reaction of diazoimides and  $\beta$ , $\gamma$ -unsaturated  $\alpha$ -keto esters (Scheme 1a).<sup>5a</sup> In their work, a relatively



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complicated chiral N-O ligand was used. Schneider and coworkers described a rhodium- and chiral phosphoric acidcocatalyzed [4 + 3] cycloaddition of *o*-quinone methides and carbonyl ylides to furnish functionalized oxo-bridged dibenzooxacines in excellent yields and stereoselectivities in a single synthetic step (Scheme 1b).<sup>5b</sup> The authors employed a highly substituted chiral phosphoric acid as a chiral catalyst. These two reactions utilized diazoimides as substrates and proceeded through a [4 + 3] cycloaddition mode. In view of this and as a continuation of work by our research group,<sup>6</sup> here we developed a simple and highly efficient bimetallic relay catalytic asymmetric cascade cycloaddition reaction of arylalkynols  $\mathbf{1}^7$  and dioxopyrrolidines  $\mathbf{2}^8$  via a formal [4 + 4]cycloaddition reaction (Scheme 1c). The chiral oxo-bridged bicyclic compounds 3 were exclusively obtained with excellent diastereoselectivity and high enantioselectivity, and no dominant competitive spirocyclic compounds 4 were observed. More importantly, the commercially available chiral (R)-BINOL-Ti(O<sup>i</sup>Pr)<sub>4</sub> complex and XPhosAuCl are directly used as bimetallic catalysts, which provides a great potential application in future synthesis of natural products and pharmaceutical intermediates. To the best of our knowledge, the use of arylalkynols as 4C synthons to carry out a formal asymmetric [4 + 4] cycloaddition reaction is unprecedented.

2-(Ethynylphenyl)methanol (1a) and (E)-1-benzyl-4-benzylidenepyrrolidine-2,3-dione (2a) were initially chosen as the model substrates (Table 1). In the presence of achiral Ph<sub>3</sub>PAuCl, which aimed at catalyzing the intramolecular 5exo-dig hydroalkoxylative cyclization-isomerization of 1a to give the reactive 4C synthon intermediate, various chiral catalytic systems were used to activate 2a and thus enantioselectively induce the subsequent formal [4 + 4]cycloaddition. As shown in Table 1, the two combinations of  $Zn(OTf)_2/{}^tBu-Box$  (L1) and  $Y(OTf)_3/Ph-PyBox$  (L2) gave only undesired cascade [2 + 4] product 4aa in excellent yields with 60% ee (entry 1) and 23% ee (entry 2), respectively. Using the (*R*)-BINOL $-Ti(O^{i}Pr)_{4}$  complex, the oxo-bridged bicyclic benzooxacine 3aa could be obtained in 74% yield with 60% ee and >20:1 dr, and no 4aa was observed by <sup>1</sup>H NMR analysis of the crude product (Table 1, entry 3). Intrigued by this result, we conducted further screening of complexes of chiral BINOL derivatives and  $Ti(O^{i}Pr)_{4}$ . To our surprise, using 3,3'disubstituted (R)-BINOLS, regardless of the substituent (Ph,  $3_{5}$ -(CF<sub>3</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, or 2-naphthyl), only a trace amount of the desired product 3aa was generated, with a great amount of the starting materials remaining untouched (entries 4-6). Next, we changed the ligand species from (R)-BINOL to (R)octahydro-binaphthol (L7), (R)-SPINOL (L8), or diol ligand L9 derived from L-tartaric acid, and no reaction occurred (entries 7-9). The (R)-phosphoric acid (CPA) catalyst was also investigated to activate dicarbonyl substrate 2a through hydrogen-bonding interactions. Consequently, this reaction afforded a 75% yield of the corresponding product 3aa as a racemate (entry 10). Therefore, (R)-BINOL-Ti(O<sup>i</sup>Pr)<sub>4</sub> was determined to be the best chiral metal complex candidate. A variety of Au(I) species were then examined in the presence of 20 mol % (R)-BINOL-Ti(O'Pr)<sub>4</sub> (entries 11-21). It was found that 5 mol % XPhosAuCl gave the best result, affording 3aa in 83% yield with 81% ee (entry 20). Other additives, such as molecular sieve (MS) types, were also examined with the XPhosAuCl/(R)-BINOL-Ti(O'Pr)<sub>4</sub> bimetallic catalyst system, and a surprising phenomenon was observed (entries 22-25). In the absence of 4 Å MS, the reaction gave as the

Table 1. Screening of Chiral Catalytic Systems in the Cascade Reaction  $^{a}$ 

la	OH + Ph 2a Au(1) catalyst (5 mol%) Cat* (20 mol%) toluene, additive, 25°C	Me O J J J J J J J J J J J J J J J J J J	N-Bn or	
	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \\ \end{array} \\ \\ \end{array} \\ \\ \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$		COH OH L3	
$\begin{array}{c c} & Ar & & \\ & OH & L4: Ar = Ph \\ & OH & L5: Ar = 3,5.CF_3C_6H_3 \\ & L6: Ar = 2-naphthyl \end{array} \qquad $				
entry	Cat*	additive	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
$1^d$	$Zn(OTf)_2-L1$	4 Å MS	96	60
$2^d$	$Y(OTf)_3-L2$	4 Å MS	98	23
3	Ti(O <sup>i</sup> Pr) <sub>4</sub> -L3	4 Å MS	78	65
4	$Ti(O^{i}Pr)_{4}-L4$	4 Å MS	trace	_
5	$Ti(O^{i}Pr)_{4}-L5$	4 Å MS	trace	_
6	$Ti(O^{i}Pr)_{4}-L6$	4 Å MS	trace	_
7	$Ti(O^{i}Pr)_{4}-L7$	4 Å MS	_	—
8	$Ti(O^{i}Pr)_{4}$ -L8	4 Å MS	-	_
9	$Ti(O^{i}Pr)_{4}-L9$	4 Å MS	-	_
10	(R)-CPA	4 Å MS	75	0
11	IPrAuCl	4 Å MS	64	81
12	$(2,4-(t-Bu)_2C_6H_3O)_3PAuCl$	4 Å MS	58	80
13	(CF <sub>3</sub> Ph) <sub>3</sub> PAuCl	4 Å MS	68	70
14	(2-MePh) <sub>3</sub> PAuCl	4 Å MS	76	67
15	Bu <sub>3</sub> PAuCl	4 Å MS	70	40
16	(PhO) <sub>3</sub> AuCl	4 Å MS	75	42
17	IPrAuNTf <sub>2</sub>	4 Å MS	18	-7
18	(Me <sub>2</sub> S)AuCl	4 Å MS	67	18
19	JohnPhosAuCl	4 Å MS	78	33
20	XPhosAuCl	4 Å MS	83	81
21 <sup>e</sup>	XPhosAuCl	4 Å MS	75	79
22	XPhosAuCl	_	48	-29
23	XPhosAuCl	3 Å MS	-	_
24	XPhosAuCl	5 Å MS	70	0
25	XPhosAuCl	$MgSO_4$	68	0
26	XPhosAuCl	4 Å MS	78	84

<sup>*a*</sup>Unless otherwise noted, all of the reactions were carried out with **1a** (40 mg, 0.3 mmol), **2a** (55 mg, 0.2 mmol), 4 Å MS (20 mg), Au(I) catalyst (0.01 mmol, 5 mol %), and catalyst (20 mol %) in toluene (2 mL) under N<sub>2</sub>. All of the reactions gave the product **3aa** except for entries 1 and 2. Abbreviations: (*R*)-CPA = (*R*)-(-)-1,1'-binaphthyl-2,2'-diyl hydrogen phosphate; JohnPhosAuCl = 2-(di-*tert*-butylphosphino)-1,1'-biphenyl gold chloride; IPrAuNTf<sub>2</sub> = [1,3-bis(2,6-diisopropylphenyl)-1,3-dihydro-2*H*-imidazol-2-ylidene][bis-(trifluoromethylsulfonyl)azanylidene]gold; XPhosAuCl = 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl gold chloride. In entries 1–10, Ph<sub>3</sub>PAuCl was used as a catalyst. In entries 11–26, (*R*)-BINOL–Ti(O<sup>i</sup>Pr)<sub>4</sub> was used as a chiral catalyst. <sup>b</sup>Yields of the isolated products. In all cases, the *dr* was >20:1. <sup>c</sup>Determined by HPLC analysis. <sup>d</sup>**4aa** was obtained. <sup>e</sup>2.5 mol % Au catalyst was used.

major product the opposite enantiomer as with 4 Å MS (entry 22). Also, racemic **3aa** was obtained with 5 Å MS or MgSO<sub>4</sub> (entries 24 and 25), and no reaction occurred in the presence of 3 Å MS (entry 23). More details were investigated about the

MS (Table S1). All of these results indicated that 4 Å MS may have the appropriate aperture size and water absorption as well as the proper interactions with the substrate and BINOL– Ti(IV). The detailed reason remains unclear at present. Notably, when (2-ethynyl-4-methoxyphenyl)methanol (1b) was used as the substrate, **3ab** was generated with slightly enhanced enantioselectivity under the standard reaction conditions (entry 26).

The reaction system concentration, the amount of (*R*)-BINOL-Ti(O<sup>i</sup>Pr)<sub>4</sub>, the solvent, and temperature as well as the ratio of the two substrates were then examined in detail (Tables S2-S5). Overall, the optimized conditions were finally established as follows: arylalkynol 1 (0.3 mmol, 1.5 equiv) and dioxopyrrolidine 2 (0.2 mmol, 1.0 equiv) along with XPhosAuCl (5 mol %) and the (*R*)-BINOL-Ti(O<sup>i</sup>Pr)<sub>4</sub> complex (20 mol %) as dual catalysts in the presence of 4 Å MS in toluene (2 mL) at 25 °C under N<sub>2</sub>.

With the optimal reaction conditions in hand, the substrate scope with respect to arylalkynols 1 and dioxopyrrolidines 2 was then investigated. As shown in Scheme 2, dioxopyrroli-





"Yields refer to the isolated products, and *ee* values were determined by HPLC analysis. N. R. = no reaction.

dines either with an electron-withdrawing substituent (F, Cl, Br, or CF<sub>3</sub>; **2b–e**) or an electron-donating substituent (OMe or Me; **2f** and **2g**) at the *para* position of the aromatic ring delivered the corresponding chiral products **3ba–bg** in up to 79% yield with up to 92% *ee.* Interestingly, when  $\mathbb{R}^2$  on the dioxopyrrolidine was 3-cyanophenyl (**2h**), the reaction afforded the oxo-bridged benzooxacine **3bh** with the highest dioxopyrrolidine 2i also gave the corresponding product 3bi with high enantioselectivity (92% ee) and diastereoselectivity. Unfortunately, cyclohexyl-substituted dioxopyrrolidine 2k failed to give the target compound 3bk, which might be ascribed to the low reactivity of 2k. In the case of 1a, the reactions with 2a, 2h, and 2i also afforded 3aa, 3ah, and 3ai, respectively, in good yields (up to 83%, 3aa) with high enantioselectivities (up to 92% ee, 3ah). However, the reaction of 1a and 2j generated the product 3aj in moderate yield with low ee with part of the starting material left over, even with a prolonged reaction time. Similarly, the 2-thienyl-substituted heteroaromatic 2l also rendered a moderate yield and ee of the desired product 3al with some starting material remaining. Also, it was observed that all of the arylalkynols 1, regardless of whether they had an electron-withdrawing or electrondonating substituent at the para or meta position of the aromatic ring of the arylalkynol, reacted with 2d and 2e to deliver the corresponding products 3ad-fd and 3ae-fe in good yields (66-85%) with high enantioselectivities (up to 92% ee). The absolute configuration of the major enantiomer of 3ed was unambiguously determined to be (5S,10R,11S) by single-crystal X-ray diffraction analysis (Figure S1).<sup>9</sup> However, no reaction occurred in the case of internal arylalkynol 1g. When the nucleophilic O in substrate 1 (X = O) was changed to N (X = NTs (1h); X = NBn (1i)), no desired aza-bridged bicyclic compound (3hd or 3id) was generated under the standard reaction conditions. These results implied that alkynyl alcohols have high reactivity in this formal [4 + 4]cycloaddition while alkynyl amides or alkynyl amines have low reactivity under the current reaction conditions. Notably, a scaled-up reaction of 1c (3.0 mmol) and 2d (2.0 mmol) was carried out and afforded 725.1 mg of 3cd in 72% yield with 88% ee.

enantioselectivity (ee 96%), and 3-bromophenyl-substituted

The corresponding oxo-bridged bicyclic benzooxacines 3 could be further transformed (Scheme S1). For example, an elegant Suzuki coupling reaction between 3cd and ptolylboronic acid was conducted and gave the desired product 5cd in 76% yield with a nearly unchanged ee value (87% ee) with the oxo-bridged bicyclic benzooxacine structure unbroken (Scheme S1, eq 1). This result indicated that this structure was relatively stable and tolerant to the base, Pd catalyst, and high temperature. Interestingly, an unexpected ring-opened product 6cd was obtained in 58% yield with 50:1 dr and 89% ee, in which the C–O bond was cleaved but the C=C double bond was unaffected when H<sub>2</sub>/Pd/C reduction was used (Scheme S1, eq 2). Moreover, the Br atom on the aromatic ring was also reduced under these reaction conditions. This reaction demonstrated that the ketal structure of the oxo-bridged benzooxacine could be cleaved under certain reaction conditions. In addition, in the presence of the Lewis acid BF<sub>3</sub>·OEt<sub>2</sub> and the nucleophile TMSCN, a similar ring-opened product 7ae was generated in 75% yield with retention of the ee (Scheme S1, eq 3).

Several control experiments were carried out to investigate the synergism of the Au(I) and (R)-BINOL–Ti(IV) bimetallic catalysts, and it was concluded that both Au(I) and BINOL/ Ti(IV) were necessary to achieve this asymmetric cascade cycloaddition reaction—neither of them was dispensable (Scheme S2). Besides, a good linear relationship was observed between the *ee* of the chiral (R)-BINOL ligand and the *ee* of the product **3ad**, which indicated that one Ti(IV) center was coordinated by one BINOL (Scheme S3). On the basis of these results and the absolute configuration of **3ed**, we propose a possible transition state model (Figure 1). The dioxopyrro-



Figure 1. Proposed transition state model.

lidine is activated by coordination with the (R)-BINOL-Ti complex, and then the **4C** intermediate formed in situ prefers to approach the *Re* face of the dioxopyrrolidine but not the sterically hindered *Si* face. The final product is formed with the (*SR*,10*S*,11*R*) configuration.

In summary, a highly efficient asymmetric catalytic cascade reaction of arylalkynols and dioxopyrrolidines was developed. In this bimetallic cooperative catalysis, a simple and commercially available Au(I) catalyst and a chiral BINOL– Ti(IV) complex gave the corresponding fused oxo-bridged bicyclic benzooxacines in moderate to good yields with high enantioselectivities (up to 96% *ee*) under mild reaction conditions with the formation of three new  $\sigma$  bonds and three new chiral stereogenic centers in a one-pot process. Control experiments indicated that XPhosAuCl, (*R*)-BINOL, and Ti(O<sup>i</sup>Pr)<sub>4</sub> are all mutually reinforcing and indispensable. Furthermore, a plausible transition state model was proposed to explain the origin of the high stereoselectivity. Further applications of this catalytic strategy to other reactions are in progress in our group.

# ASSOCIATED CONTENT

### **3** Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.1c00976.

Screenings of reaction conditions, reaction mechanism, product transformation, crystallographic data, synthesis of substrates, experimental procedures, characterization and analytical data of products, NMR spectra, and HPLC traces (PDF)

# **Accession Codes**

CCDC 1974108 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data\_request/cif, or by emailing data\_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, U.K.; fax: +44 1223 336033.

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### Notes

The authors declare no competing financial interest.

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# DEDICATION

This work is dedicated to the 100th anniversary of chemistry at Nankai University.

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(9) Crystal data for **3ed**:  $C_{27}H_{21}BrClNO_3$ , MW = 522.81; monoclinic,  $P2_1$ ; final R indices  $[I > 2\sigma(I)]$ ,  $R_1 = 0.0481$ ,  $wR_2 = 0.0858$ ; R indices (all data),  $R_1 = 0.0842$ ,  $wR_2 = 0.0774$ ; a = 10.951(2) Å, b = 6.5509(13) Å, c = 15.865(3) Å, V = 1128.6(4) Å<sup>3</sup>, Z = 2; T = 113 K; 13739/5377 collected/unique reflections ( $R_{int} = 0.079$ ); 3232 observations  $[I > 2\sigma(I)]$ ; 299 parameters; goodness of fit on  $F^2 = 0.97$ . Supplementary crystallographic data have been deposited at the Cambridge Crystallographic Data Center (CCDC 1974108).