

## Asymmetric hydrogenation with the use of chiral carborane amidophosphite derivatives in supercritical carbon dioxide and CH<sub>2</sub>Cl<sub>2</sub>

S. E. Lyubimov,\* V. A. Ol'shevskaya, P. V. Petrovskii, E. A. Rastorguev, T. A. Verbitskaya, V. N. Kalinin, and V. A. Davankov

A. N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences, 28 ul. Vavilova, 119991 Moscow, Russian Federation.  
Fax: +7 (499) 135 6471. E-mail: lssp452@mail.ru

New chiral carborane-containing amidophosphites containing the BINOL fragment (BINOL stands for 2,2'-dihydroxy-1,1'-binaphthyl) have been synthesized. The study of efficiency of these compounds as ligands in the Rh-catalyzed asymmetric hydrogenation of enamides in supercritical carbon dioxide (scCO<sub>2</sub>) and CH<sub>2</sub>Cl<sub>2</sub> showed that enantioselectivity of the process is considerably higher in scCO<sub>2</sub>.

**Key words:** carboranes, asymmetric hydrogenation, amidophosphites, supercritical carbon dioxide, rhodium complexes.

At present, significant amount of chemical reactions is carried out in organic solvents, such as chlorinated hydrocarbons, methanol, acetone, which are significantly dangerous for the environment, that stimulates a search for the alternative, environmentally friendly media. Supercritical carbon dioxide (scCO<sub>2</sub>) is one of such media because of availability of this compound and its environmental and fire safety.<sup>1,2</sup> In addition, carbon dioxide can be easily and completely removed from the reaction products within several minutes. Despite of this advantages, as of this moment scCO<sub>2</sub> is mainly used for extraction<sup>3</sup> and considerably less frequently as the medium for organic synthesis.<sup>1</sup> Note that scCO<sub>2</sub> possesses excellent miscibility with gases, for example, with hydrogen, that makes it easy to transport H<sub>2</sub> to the reaction surface of catalysts.<sup>4</sup> In addition, in the case of scCO<sub>2</sub> no separation of gaseous and liquid organic phases occurs, that is observed when organic solvents are used in hydrogenation processes. Therefore, the study of hydrogenation in scCO<sub>2</sub> is an actual problem. At present, low attention is paid to the asymmetric metal-complex hydrogenation in scCO<sub>2</sub>, though promising results have been obtained in this field. For instance, the use of metal-complex catalysts with chiral phosphine ligands demonstrated high enantioselectivity in hydrogenation of a series of prochiral unsaturated substrates. Usually, 12–48 h are required to reach the full conversion, that is comparable with the results obtained in organic solvents.<sup>5,6</sup> Earlier,<sup>7–9</sup> it has been shown that chiral phosphites and amidophosphites are more promising ligands for asymmetric hydrogenation in scCO<sub>2</sub>, which provide the high rate and enantioselectivity of the reac-

tion. One of the new trends in the development of efficient chiral catalysts for hydrogenation processes in scCO<sub>2</sub> is incorporation of a bulky carborane group into the ligand composition.<sup>10–12</sup> In the present work, we demonstrate results of testing of new amidophosphite ligands containing carborane substituents in the asymmetric Rh-catalyzed hydrogenation reaction of enamides in scCO<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub> to yield amino acid derivatives.

### Results and Discussion

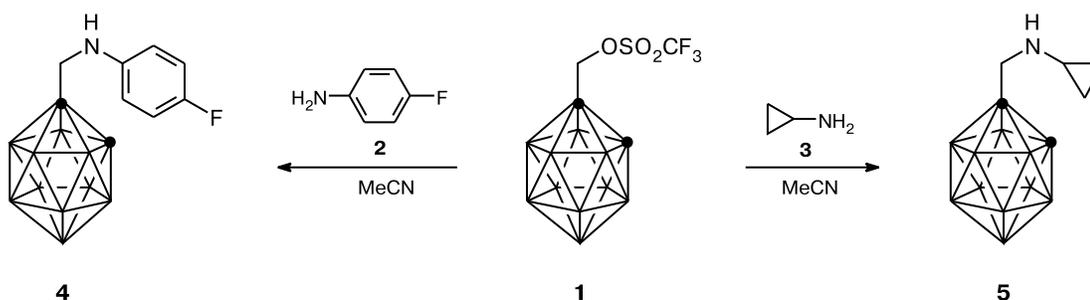
The reaction of *ortho*-carboranylmethyl triflate (**1**) with 4-fluoroaniline (**2**) or cyclopropylamine (**3**) led to carborane-containing amines **4** and **5** (Scheme 1).

Phosphorylation of amines **4** and **5** afforded new chiral carboranyl amidophosphites L<sup>1</sup> and L<sup>2</sup> (Scheme 2).

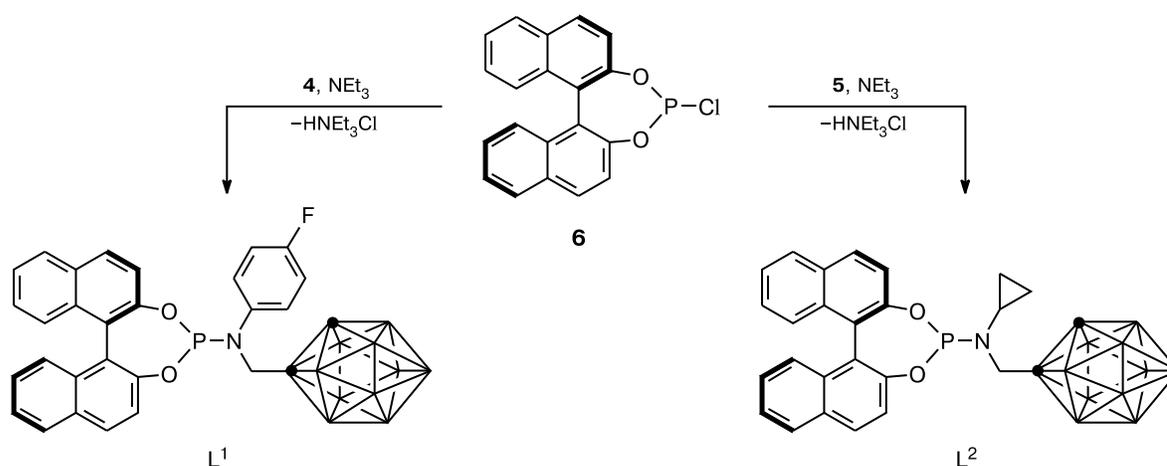
Ligands L<sup>1</sup> and L<sup>2</sup> are well soluble in organic solvents, such as CH<sub>2</sub>Cl<sub>2</sub> and C<sub>6</sub>H<sub>6</sub>. The presence of the fluorine atom in ligand L<sup>1</sup> provides its good solubility also in the nonpolar hexane. Fluorine-containing ligands, as compared to their analogs containing no fluorine atoms, are characterized by better solubility in scCO<sub>2</sub> as well, which is a nonpolar enough medium.<sup>5</sup>

Amidophosphites L<sup>1</sup> and L<sup>2</sup> were tested in the Rh-catalyzed asymmetric hydrogenation of a series of enamides **7a–c** — precursors of phenylalanine and its chlorine- and fluorine-containing analogs (Scheme 3, Table 1). Note that *N*-acetyl derivatives of 4-chlorophenylalanine possess antiinflammatory properties, whereas the drugs based on 4-fluorophenylalanine exhibit antimicrobial properties.<sup>9</sup>

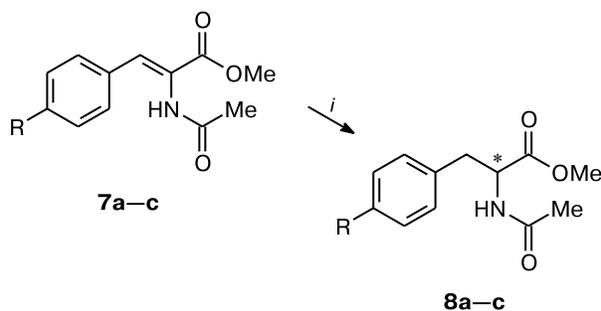
Scheme 1



Scheme 2



Scheme 3



*i.* H<sub>2</sub>, scCO<sub>2</sub> or CH<sub>2</sub>Cl<sub>2</sub>, [Rh(COD)<sub>2</sub>]BF<sub>4</sub>/2L, 0.5 mol.%,  
R = H (**a**), F (**b**), Cl (**c**)

When the catalyst formed *in situ* starting from ligand L<sup>1</sup> and [Rh(COD)<sub>2</sub>]BF<sub>4</sub> (COD stands for cyclooct-1,5-diene) was used in scCO<sub>2</sub> at high (45 atm) hydrogen pressure and molar ratio of the substrate to the catalyst 200/1, the complete conversion of substrate **7a** was reached in 2 h, and the product had 48% *ee* (see Table 1, entry 1). A decrease in the pressure of H<sub>2</sub> from 40 to 15 atm allowed us to somewhat increase the enantioselectivity (*cf.* en-

tries 1 and 2). Application of the same catalytic system [Rh(COD)<sub>2</sub>]BF<sub>4</sub>/2L<sup>1</sup> in CH<sub>2</sub>Cl<sub>2</sub> provided only 20% *ee*, with the enantioselectivity remaining the same at both 15 and 40 atm of H<sub>2</sub> (see Table 1, entries 3 and 4). When amidophosphite L<sup>2</sup> was used in hydrogenation of **7a** at 15 atm of H<sub>2</sub> in scCO<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub> at 45 °C, the scCO<sub>2</sub> also provided higher enantioselectivity (see Table 1, entries 5 and 6). In the hydrogenation of fluorine-containing substrate **7b** using the catalyst based on ligand L<sup>1</sup> in scCO<sub>2</sub>, 48% *ee* at 40 atm of H<sub>2</sub> and 52% *ee* at 15 atm of H<sub>2</sub> were achieved (see Table 1, entries 7 and 8). In CH<sub>2</sub>Cl<sub>2</sub>, 21% *ee* at both high and low pressure of H<sub>2</sub> was observed (entries 9 and 10). Ligand L<sup>2</sup> in hydrogenation of **7b** provides lower enantioselectivity in both scCO<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub>, with the enantiomeric excess of the reaction product **8b** in scCO<sub>2</sub> being higher (see Table 1, entries 11 and 12). Hydrogenation of chlorine-containing enamide **7c** with the use of ligand L<sup>1</sup> gave analogous results. For instance, application of scCO<sub>2</sub> provides higher enantioselectivity as compared to CH<sub>2</sub>Cl<sub>2</sub> (see Table 1, entries 13–16). Hydrogenation of **7c** using the catalytic system based on amidophosphite L<sup>2</sup> in scCO<sub>2</sub> showed that higher pressure of H<sub>2</sub> assists in the increase in conversion (entries 17–18).

**Table 1.** Hydrogenation of enamides **7a–c** catalyzed by  $[\text{Rh}(\text{COD})_2]\text{BF}_4/2\text{L}$  (45 °C)

Entry	L	R	$P_{\text{H}_2}$ /atm	Medium	$t/h$	$P_{\text{tot}}$ /atm	$\gamma^a$ (%)	$ee^b$ (%)
1	L <sup>1</sup>	H	40	scCO <sub>2</sub>	2	250	100	48
2	L <sup>1</sup>	H	15	scCO <sub>2</sub>	2	250	100	52
3	L <sup>1</sup>	H	40	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	20
4	L <sup>1</sup>	H	15	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	20
5	L <sup>2</sup>	H	15	scCO <sub>2</sub>	2	250	100	37
6	L <sup>2</sup>	H	15	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	22
7	L <sup>1</sup>	F	40	scCO <sub>2</sub>	2	250	100	48
8	L <sup>1</sup>	F	15	scCO <sub>2</sub>	2	250	100	52
9	L <sup>1</sup>	F	40	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	21
10	L <sup>1</sup>	F	15	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	21
11	L <sup>2</sup>	F	15	scCO <sub>2</sub>	2	250	100	42
12	L <sup>2</sup>	F	15	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	30
13	L <sup>1</sup>	Cl	40	scCO <sub>2</sub>	2	250	100	48
14	L <sup>1</sup>	Cl	15	scCO <sub>2</sub>	2	250	100	51
15	L <sup>1</sup>	Cl	40	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	21
16	L <sup>1</sup>	Cl	15	CH <sub>2</sub> Cl <sub>2</sub>	2	—	100	21
17	L <sup>2</sup>	Cl	40	scCO <sub>2</sub>	3	250	35	36
18	L <sup>2</sup>	Cl	15	scCO <sub>2</sub>	3	250	8	36
19	L <sup>2</sup>	Cl	40	CH <sub>2</sub> Cl <sub>2</sub>	3	—	100	32
20	L <sup>2</sup>	Cl	15	CH <sub>2</sub> Cl <sub>2</sub>	3	—	100	32

<sup>a</sup>  $\gamma$  is the conversion.<sup>b</sup> S-Configuration of products in all the cases.

The full conversion was successfully reached within 3 h in CH<sub>2</sub>Cl<sub>2</sub>, however, the enantioselectivity has proved somewhat lower than in scCO<sub>2</sub> (entries 19–20).

In conclusion, the asymmetric Rh-catalyzed hydrogenation of enamides involving new carborane-containing chiral amidophosphites in scCO<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub> allows one to reach high conversions at low loading a Rh-catalyst within 2–3 h, with the *ee* values of the products being higher in scCO<sub>2</sub>.

### Experimental

<sup>31</sup>P, <sup>1</sup>H, and <sup>11</sup>B NMR spectra were recorded on an Avance 400 spectrometer (161.98, 400.13, and 128.4 MHz) relatively to 85% H<sub>3</sub>PO<sub>4</sub> in D<sub>2</sub>O, Me<sub>4</sub>Si, and BF<sub>3</sub>·OEt<sub>2</sub>, respectively. Elemental analysis was performed in the Laboratory of Organic Microanalysis of the A. N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences.

All the reactions concerning preparation of catalysts were performed in the atmosphere of dry argon in anhydrous solvents. *o*-Carboranylmethyl triflate (**1**),<sup>13</sup> (*R*<sub>ax</sub>)-2-chlorodinaphtho[2,1-*d*:1',2'-*f*][1,3,2]dioxaphosphepane (**6**),<sup>14</sup> [Rh(COD)<sub>2</sub>]<sub>2</sub>BF<sub>4</sub>,<sup>15</sup> methyl (*Z*)-2-acetamido-3-phenylacrylate (**7a**),<sup>16</sup> methyl (*Z*)-2-acetamido-3-(4-fluorophenyl)acrylate (**7b**),<sup>16</sup> and methyl (*Z*)-2-acetamido-3-(4-chlorophenyl)acrylate (**7c**)<sup>9</sup> were obtained according to the known procedures.

**Synthesis of amines 4 and 5 (general procedure).** A mixture of *o*-carboranylmethyl triflate **1** (1.9 g, 6.2 mmol), the corresponding amine (6.5 mmol), NaOAc (0.74 g, 9 mmol) and MeCN

(10 mL) was refluxed for 4–8 h (TLC monitoring of conversion of **1** using heptane–EtOAc (7 : 3) as an eluent). Water (20 mL) was added to the reaction mixture, followed by extraction with EtOAc (3×7 mL). A combined organic phase was washed with water, dried with Na<sub>2</sub>SO<sub>4</sub>, and filtered, the solvent was evaporated *in vacuo*. The products were purified by column chromatography on silica gel (heptane–EtOAc (10 : 1) in the case of **4** and heptane in the case of **5**).

***N*-(*o*-Carboranylmethyl)-*N*-(4-fluorophenyl)amine (**4**),** white powder. The yield was 1.07 g (65%), m.p. 82–83 °C. Found (%): C, 40.56; H, 6.88; N, 5.16. C<sub>9</sub>H<sub>18</sub>B<sub>10</sub>FN. Calculated (%): C, 40.43; H, 6.79; N, 5.24. <sup>1</sup>H NMR (CDCl<sub>3</sub>),  $\delta$ : 1.38–3.17 (m, 10 H); 3.76 (s, 1 H); 3.86 (s, 2 H); 3.92 (s, 1 H); 6.50–6.60 (m, 2 H); 6.90 (t, 2 H,  $J = 8.4$  Hz). <sup>11</sup>B {H} NMR (CDCl<sub>3</sub>),  $\delta$ : –14.00±–10.82 (m, 4 B); –11.52 (s, 2 B); –9.06 (s, 2 B); –5.14 (s, 1 B); –2.31 (s, 1 B). IR (CHCl<sub>3</sub>):  $\nu_{\text{NH}}$  3440 cm<sup>–1</sup>.

***N*-(*o*-Carboranylmethyl)-*N*-cyclopropylamine (**5**),** viscous colorless oil. The yield was 0.66 g (50%). Found (%): C, 33.84; H, 8.94; N, 6.51. C<sub>6</sub>H<sub>19</sub>B<sub>10</sub>N. Calculated (%): C, 33.78; H, 8.98; N, 6.57. <sup>1</sup>H NMR (CDCl<sub>3</sub>),  $\delta$ : 0.24–0.31 (m, 2 H); 0.41–0.49 (m, 2 H); 1.28–3.00 (m, 10 H); 2.14–2.21 (m, 1 H); 3.35 (s, 2 H); 3.94 (s, 1 H). <sup>11</sup>B {H} NMR (CDCl<sub>3</sub>),  $\delta$ : –14.32±–14.30 (m, 4 B); –11.41 (s, 2 B); –9.12 (s, 2 B); –5.55 (s, 1 B); –3.11 (s, 1 B). IR (CHCl<sub>3</sub>):  $\nu_{\text{NH}}$  3336 cm<sup>–1</sup>.

**Synthesis of ligands L<sup>1</sup> and L<sup>2</sup> (general procedure).** Triethylamine (0.14 mL, 1.0 mmol) and the corresponding carborane amine **4** or **5** (1.0 mmol) were added to a solution of (*R*<sub>ax</sub>)-2-chlorodinaphtho[2,1-*d*:1',2'-*f*][1,3,2]dioxaphosphepane<sup>14</sup> (**6**) (0.35 g, 1.0 mmol) in C<sub>6</sub>H<sub>6</sub> (15 mL) and the mixture was stirred for 10 min at 20 °C, heated to the reflux, cooled to room temperatures, a precipitate of HNet<sub>3</sub>Cl was filtered off. The products were purified by column flash-chromatography on silica gel (benzene).

**(*R*<sub>ax</sub>)-2-[*N*-(*o*-Carboranylmethyl)-*N*-(4-fluorophenyl)amino]-dinaphtho[2,1-*d*:1',2'-*f*][1,3,2]dioxaphosphepane (**L<sup>1</sup>**),** viscous colorless oil. The yield was 0.49 g (85%). Found (%): C, 59.95; H, 5.12; N, 2.50. C<sub>29</sub>H<sub>29</sub>B<sub>10</sub>FNO<sub>2</sub>P. Calculated (%): C, 59.89; H, 5.03; N, 2.41. <sup>31</sup>P NMR (CDCl<sub>3</sub>),  $\delta$ : 136.61. <sup>1</sup>H NMR (CDCl<sub>3</sub>),  $\delta$ : 1.36–3.18 (m, 10 H); 3.75 (s, 1 H); 3.84 (s, 2 H); 6.49–6.58 (m, 2 H); 6.90 (t, 2 H,  $J = 8.6$  Hz); 7.21–8.00 (m, 12 H). <sup>11</sup>B {H} NMR (CDCl<sub>3</sub>),  $\delta$ : –14.18±–12.16 (m, 4 B); –11.51 (s, 2 B); –9.06 (s, 2 B); –5.15 (s, 1 B); –2.32 (s, 1 B).

**(*R*<sub>ax</sub>)-2-[*N*-(*o*-Carboranylmethyl)-*N*-cyclopropylamino]-dinaphtho[2,1-*d*:1',2'-*f*][1,3,2]dioxaphosphepane (**L<sup>2</sup>**),** viscous colorless oil. The yield was 0.42 g (80%). Found (%): C, 59.25; H, 5.80; N, 2.59. C<sub>26</sub>H<sub>30</sub>NB<sub>10</sub>O<sub>2</sub>P. Calculated (%): C, 59.19; H, 5.73; N, 2.65. <sup>31</sup>P NMR (CDCl<sub>3</sub>),  $\delta$ : 136.66. <sup>1</sup>H NMR (CDCl<sub>3</sub>),  $\delta$ : 0.24–0.32 (m, 2 H); 0.41–0.49 (m, 2 H); 1.20–3.08 (m, 10 H); 2.14–2.21 (m, 1 H); 3.35 (s, 2 H); 3.95 (s, 1 H); 7.20–8.00 (m, 12 H). <sup>11</sup>B {H} NMR (CDCl<sub>3</sub>),  $\delta$ : –14.32±–12.21 (m, 4 B); –11.40 (s, 2 B); –9.12 (s, 2 B); –5.52 (s, 1 B); –3.08 (s, 1 B).

**Asymmetric hydrogenation of enamides in scCO<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub>.** The compound [Rh(COD)<sub>2</sub>]<sub>2</sub>BF<sub>4</sub> (2 mg, 0.005 mmol), the ligand (0.01 mmol), and CH<sub>2</sub>Cl<sub>2</sub> (0.3 mL) were placed in a 10-mL autoclave. The mixture was stirred for 2 min, the solvent was evaporated *in vacuo*, followed by addition of the corresponding enamide (**7a–c**) (1 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (4 mL), if the latter was used as the solvent. The sealed autoclave was purged with CO<sub>2</sub>, filled with hydrogen to the required pressure (see Table 1), then carbon dioxide was added using a High Pressure Equipment to 250 atm. The reactor was heated to the corresponding tempera-

ture (45 °C) for 5 min and the process was performed with stirring using a magnetic stirrer. After the reaction was completed, CO<sub>2</sub> and H<sub>2</sub> were slowly released, conversion was monitored by <sup>1</sup>H NMR spectroscopy. Enantiomeric ratio of products **8a–c** was determined by HPLC on an Agilent HP-1100 chromatograph using a Kromasil 5-AmyCoat column according to the procedure given in the literature.<sup>9</sup>

This work was financially supported by the Russian Foundation for Basic Research (Project Nos 09-03-12104-ofi\_m and 09-03-91345-NNIO\_a).

### References

1. W. Leitner, P. G. Jessop, *Chemical Synthesis Using Supercritical Fluids*. Wiley-VCH, New York, 1999.
2. P. G. Jessop, T. Ikariya, R. Noyori, *Chem. Rev.*, 1999, **99**, 475.
3. M. A. McHugh, V. Krukoniš, *Supercritical Fluid Extraction*, 2nd ed., Elsevier, 1994.
4. Y. Arai, T. Sako, Y. Takebayashi, *Supercritical Fluids. Molecular Interactions, Physical Properties and New Applications*, Springer, Berlin, 2002.
5. D. J. Cole-Hamilton, *Adv. Synth. Catal.*, 2006, **348**, 1341.
6. R. Skouta, *Green Chem. Lett. Rev.*, 2009, **2**, 121.
7. S. E. Lyubimov, E. E. Said-Galiev, A. R. Khokhlov, N. M. Loim, L. N. Popova, P. V. Petrovskii, V. A. Davankov, *J. Supercritical Fluids*, 2008, **45**, 70.
8. S. E. Lyubimov, V. A. Davankov, E. E. Said-Galiev, A. R. Khokhlov, *Catalysis Commun.*, 2008, **9**, 1851.
9. S. E. Lyubimov, I. V. Kuchurov, V. A. Davankov, S. G. Zlotin, *J. Supercritical Fluids*, 2009, **50**, 118.
10. S. E. Lyubimov, A. A. Tyutyunov, V. N. Kalinin, E. E. Said-Galiev, A. R. Khokhlov, P. V. Petrovskii, V. A. Davankov, *Tetrahedron Lett.*, 2007, **48**, 8217.
11. S. E. Lyubimov, I. V. Kuchurov, A. A. Tyutyunov, P. V. Petrovskii, V. N. Kalinin, S. G. Zlotin, V. A. Davankov, E. Hey-Hawkins, *Catalysis Commun.*, 2010, **11**, 419.
12. S. E. Lyubimov, A. S. Safronov, A. A. Tyutyunov, V. N. Kalinin, E. E. Said-Galiev, A. R. Khokhlov, P. V. Petrovskii, P. M. Valetskii, V. A. Davankov. *Izv. Akad. Nauk, Ser. Khim.*, 2008, **57**, 337 [*Russ. Chem. Bull., Int. Ed.*, 2008, **57**, 345].
13. V. N. Kalinin, E. G. Rys, A. A. Tyutyunov, Z. A. Starikova, A. A. Korlyukov, V. A. Ol'shevskaya, D. D. Sung, A. B. Ponomaryov, P. V. Petrovskii, E. Hey-Hawkins *J. Chem. Soc., Dalton Trans.*, 2005, 903.
14. G. Francio, C. G. Arena, F. Faraone, C. Graiff, M. Lanfranchi, A. Tiripicchio, *Eur. J. Inorg. Chem.*, 1999, 1219.
15. T. V. RajanBabu, T. A. Ayers, G. A. Halliday, K. K. You, J. C. Calabrese, *J. Org. Chem.*, 1997, **62**, 6012.
16. B. S. Jursic, S. Sagiraju, D. K. Ancalade, T. Clark, E. D. Stevens, *Synthetic Commun.*, 2007, **37**, 1709.

Received July 14, 2010