Contents lists available at ScienceDirect





CrossMark

# Phytochemistry Letters

journal homepage: www.elsevier.com/locate/phytol

# Pyrazole alkaloids from watermelon (Citrullus lanatus) seeds

Takashi Kikuchi, Aoi Ikedaya, Akiko Toda, Kenji Ikushima, Takahiro Yamakawa, Rina Okada, Takeshi Yamada, Reiko Tanaka\*

Laboratory of Medicinal Chemistry, Osaka University of Pharmaceutical Sciences, 4-20-1 Nasahara, Takatsuki, Osaka 569-1094, Japan

#### ARTICLE INFO

Article history: Received 19 January 2015 Received in revised form 16 February 2015 Accepted 20 February 2015 Available online 7 March 2015

*Keywords: Citrullus lanatus* Alkaloid Glycoside Inhibitory activity on melanogenesis

# 1. Introduction

Citrullus lanatus (Thunb.) Matsum. and Nakai (Citrullus vulgaris Schrad, Japanese name: suika) is cultivated throughout the world for use as a fruit as well as a medicine. Its fruits have been used as a diuretic, its seed coat has been used to treat hematemesis and melena, and its kernel has been used as an antihypertensive and in the treatment of acute cystisis (Shanghai Scientific and Technical Publishers, Shogakukan, Inc., 1985). Previous studies detected polyphenols (Lako et al., 2006), flavonols (Lako et al., 2006), carotenoids (Lako et al., 2006), amino acids (Shanghai Scientific and Technical Publishers, Shogakukan, Inc., 1985), including citrulline and arginine, and triterpenes (Ripperger and Seifert, 1975) in the fruits of C. lanatus. The apoptogenic effects of cucurbitacin L 2-O-glucoside on colon adenocarcinoma, and antiinflammatory activities of cucurbitacin E, both isolated from C. lanatus, have been reported previously (Hassan et al., 2011). We recently isolated two new cucurbitane-type triterpenes, 24hydroperoxycucurbita-5,25-dien-3β-ol and 25-hydroperoxycucurbita-5,23-dien-3 $\beta$ -ol, that had hydroperoxy groups on their side chains from the Et<sub>2</sub>O fraction of a C. lanatus seed MeOH extract, and examined their cytotoxic activities against the HL-60, P388, and L1210 cell lines, and melanogenesis inhibitory activities in  $\alpha$ -MSH-stimulated B16 4A5 cells (Kikuchi et al., 2013a). In the present study, we describe the isolation and structural elucidation

\* Corresponding author. Tel.: +81 72 690 1084; fax: +81 72 690 1084. *E-mail address:* tanakar@gly.oups.ac.jp (R. Tanaka).

#### ABSTRACT

Three new compounds, *i.e.* 1-[2-(5-hydroxymethyl-1*H*-pyrrole-2-carbaldehyde-1-yl)ethyl]-1*H*-pyrazole (**1**), 1-({[5-( $\alpha$ -D-galactopyranosyloxy)methyl]-1*H*-pyrrole-2-carbaldehyde-1-yl]-ethyl)-1*H*-pyrazole (**2**), and (4-hydroxyphenyl)methanol 4-[ $\beta$ -D-apiofuranosyl(1 $\rightarrow$ 2)-O- $\beta$ -D-glucopyranoside] (**3**), were isolated from an extract of watermelon seeds. Compounds **1** and **2** were pyrazole-alkaloids with a pyrrole ring. This is the first study to show compounds with pyrrole and pyrazole rings in a molecule isolated from natural products. In the evaluation for melanogenesis inhibitory, compound **1** exhibited modest inhibitory activity on melanogenesis without cytotoxicity. Meanwhile compound **2** showed some inhibitory activity accompanied by some cytotoxicity.

© 2015 Phytochemical Society of Europe. Published by Elsevier B.V. All rights reserved.

of compounds **1–3** as well as the inhibitory effects of compounds from watermelon on melanogenesis in B16 4A5 cells.

### 2. Results and discussion

Three new compounds (1–3) were isolated from the seeds of *C. lanatus* (Fig. 1).

Compound **1** exhibited an  $[M+H]^+$  ion in the HR-FAB-MS at m/z220.1089, which was compatible with the molecular formula C<sub>11</sub>H<sub>14</sub>N<sub>3</sub>O<sub>2</sub> (calcd 220.1086). Its IR spectrum showed absorption that indicated a hydroxy group  $(\nu_{max}\;3424\,cm^{-1})$  and carbonyl group ( $\nu_{\text{max}}$  1649 cm<sup>-1</sup>). <sup>1</sup>H and <sup>13</sup>C NMR spectra indicated the presence of an oxymethylene [ $\delta_{\rm H}$  4.02 (s);  $\delta_{\rm C}$  55.9 (t)], pyrrole ring  $[\delta_{\rm H} 6.16 \text{ (d, } I = 4.1 \text{ Hz}), 7.03 \text{ (d, } I = 4.1 \text{ Hz}); \delta_{\rm C} 111.3 \text{ (d)}, 126.8 \text{ (d)},$ 133.4 (s), 145.3 (s)], pyrazole ring [ $\delta_{\rm H}$  6.19 (t, J = 2.0 Hz), 7.16 (dd, I = 0.6, 2.0 Hz), 7.49 (dd, I = 0.6, 2.0 Hz);  $\delta_{C}$  106.9 (d), 131.8 (d), 140.9 (d)], and formyl group [ $\delta_{\rm H}$  9.45 (s);  $\delta_{\rm C}$  181.1 (d)]. The following correlations were observed in the HMBC spectrum; H-1'  $[\delta_{\rm H} 4.52 \text{ (t, } I = 5.9 \text{ Hz})]/\text{C-5} [\delta_{\rm C} 131.8 \text{ (d)}]; \text{H-2'} [\delta_{\rm H} 4.71 \text{ (t, } I)]$ I = 5.9 Hz]/C-2" [ $\delta_{\rm C}$  133.4 (s)] and C-5" [ $\delta_{\rm C}$  145.3 (s)]; H-3" [ $\delta_{\rm H}$  7.03  $(d, J = 4.1 \text{ Hz})/C-7'' [\delta_C 181.1 (d)]; H-6'' [\delta_H 4.02 (s)]/C-4'' [\delta_C 111.3]$ (d)] and 5" [ $\delta_{C}$  145.3 (s)] (Fig. 2). In the <sup>1</sup>H–<sup>1</sup>H COSY spectrum, H-3  $[\delta_{\rm H} 7.49 \text{ (dd, } J = 0.6, 2.0 \text{ Hz})]/\text{H}-4 [\delta_{\rm H} 6.19 \text{ (t, } J = 2.0 \text{ Hz})], \text{H}-4/\text{H}-5$  $[\delta_{\rm H} 7.16 \,(\text{dd}, J = 0.6, 2.0 \,\text{Hz})]$ , and H-3"  $[\delta_{\rm H} 7.03 \,(\text{d}, J = 4.1 \,\text{Hz})]/\text{H-4"}$  $[\delta_{\rm H} 6.16 \text{ (d, } J = 4.1 \text{ Hz})]$  were observed (Fig. 2). Therefore, **1** was established as 1-[2-(5-hydroxymethyl-1H-pyrrole-2-carbaldehyde-1-yl)ethyl]-1H-pyrazol (Table 1).

Compound **2** exhibited an  $[M+H]^+$  ion in the HR-FAB-MS at m/z 382.1617, which was compatible with the molecular formula

http://dx.doi.org/10.1016/j.phytol.2015.02.017

1874-3900/© 2015 Phytochemical Society of Europe. Published by Elsevier B.V. All rights reserved.



Fig. 1. Structures of compounds 1-3 from watermelon seeds.

C<sub>17</sub>H<sub>24</sub>N<sub>3</sub>O<sub>7</sub> (calcd for 382.1609). Its IR spectrum showed absorption that indicated hydroxy groups  $(3408 \text{ cm}^{-1})$  and a carbonyl group (1656 cm<sup>-1</sup>). <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra suggested compound 2 was a glycoside of 1 since the signals of 2 were similar to those of 1, except for the sugar moiety. This was confirmed by HMBC and <sup>1</sup>H-<sup>1</sup>H COSY spectra (Fig. 2). The <sup>1</sup>H and  $^{13}$ C NMR signals of the sugar moiety [ $\delta_H$  4.79 (d);  $\delta_C$  62.8 (t), 70.0 (d), 71.1 (d), 71.3 (d), 72.9 (d), 99.5 (d)] suggested the presence of a terminal p-galactopyranose, and its configuration at C-1<sup>'''</sup> was established as the  $\alpha$ -orientation due to the coupling constant of H-1<sup>'''</sup> (J = 4.1 Hz). This was confirmed by acid hydrolysis, derivatization with L-cysteine methyl ester hydrochloride and o-tolylisothiocyanate, and HPLC analysis. Therefore, compound 2 was established as  $1-(\{[5-(\alpha-D-galactopyranosyloxy)methyl]-1H-pyr$ role-2-carbaldehyde-1-yl}-ethyl)-1H-pyrazol (Table 1). Compounds **1** and **2** are rare pyrazole-alkaloids. Kumar et al. (2013) isolated L- $\alpha$ amino- $\beta$ -(pyrazolyl-*N*)-propanoic acid, a naturally occurring amino acid with a pyrazole ring, from the juice of watermelons. The pyrazole rings of 1 and 2 may be derived from this amino acid. This is the first study to show compounds with pyrrole and pyrazole rings in a molecule isolated from natural products; however, pyrroleimidazole alkaloids have been isolated from the Mediterranean sponge (Fattorusso and Taglialatela-Scafati, 2000).

Compound **3** exhibited an [M+Na]<sup>+</sup> ion in the HR-FAB-MS at *m/z* 441.1375, which had the molecular formula of  $C_{18}H_{26}O_{11}Na$  (calcd 441.1372). Its IR spectrum revealed absorption bands indicating hydroxy groups ( $\nu_{max}$  3392 cm<sup>-1</sup>). The <sup>13</sup>C NMR spectrum of **3** displayed  $\delta_C$  62.5 (t), 71.4 (d), 78.0 (d), 78.6 (d), 78.7 (d), 101.0 (d) attributed to inner glucose, and  $\delta_C$  66.1 (t), 75.5 (t), 78.1 (d), 80.8 (s), 110.8 (d) attributed to terminal apiose. In the <sup>1</sup>H NMR spectrum, the A<sub>2</sub>B<sub>2</sub>-type aromatic proton signals, [ $\delta_H$  7.04 (d, 8.8) and 7.27 (d, 8.8)] and hydroxy methylene signal [ $\delta_H$  4.53 (s);  $\delta_C$  64.8 (t)] suggested the presence of a hydroxy methyl phenyl moiety. The following correlations were observed in the HMBC spectrum: H-1 [ $\delta_H$  4.53 (s)]/C-1' [ $\delta_c$  136.5 (s)]; H-2' [ $\delta_H$  7.27

(d)]/C-1'; H-3' [δ<sub>H</sub> 7.04 (d)]/C-4' [δ<sub>c</sub> 158.4 (s)]; H-1" [δ<sub>H</sub> 4.95 (d)]/C-4'; H-1"' [δ<sub>H</sub> 5.46 (d)]/C-2" [δ<sub>c</sub> 78.6 (d)], C-3"'; H-5"/C-1", C-4"; H-6"/C-4"; H-2"'/C-3"'; H-4"/C-3"'; H-5"/C-3"'(Fig. 2). On acid hydrolysis, **3** afforded D-glucose and D-apiose, which were identified by derivatization with L-cysteine methyl ester hydrochloride and *o*-tolylisothiocyanate, and HPLC analysis. Thus, compound **3** was established as (4-hydroxyphenyl)methanol 4-[β-D-apiofuranosyl(1→2)-O-β-D-glucopyranoside] (Table 1).

Compounds **1–3** and the positive control, arbutin were evaluated for there melanogenesis inhibitory effects against  $\alpha$ -MSH-induced melanogenesis in B16 4A5 melanomas (Table 2). To determine safe concentrations, the cytotoxicities of three compounds were examined using the MTT assay. Compound **1** did not exhibit cytotoxicity against B16 4A5 cells at 3–100  $\mu$ M. It also modestly inhibited melanogenesis in B16 4A5 cells (melanin content: 80.3% at 30  $\mu$ M, 64.1% at 100  $\mu$ M). This compound may be valuable in the development of skin-whitening agents. Compound **2** exhibited some melanogenesis inhibitory activity (melanin content: 86.0% at 30  $\mu$ M, 48.6% at 100  $\mu$ M); however, this was attributed to its cytotoxic effects (cell viability: 92.7% at 30  $\mu$ M, 70.4% at 100  $\mu$ M).

#### 3. Experimental

### 3.1. General

Chemicals and reagents were purchased as follows: fetal bovine serum (FBS) from Invitrogen Co. (Carlsbad, CA, USA), 3-(4,5dimethyl-2-thiazolyl)-2,5-diphenyl-2*H*-tetrazolium bromide (MTT) from Sigma–Aldrich Japan Co. (Tokyo, Japan), Dulbecco's modified Eagle's medium (D-MEM), and antibiotics from Nacalai Tesque, Inc. (Kyoto, Japan). All other chemicals and reagents were of analytical grade. Optical rotations were measured with a JASCO DIP-1000 digital polarimeter. IR spectra were recorded on a Perkin-Elmer 1720X FTIR spectrophotometer. The <sup>1</sup>H (600 MHz) and <sup>13</sup>C



Table 1		
<sup>1</sup> H (600 MHz) and	<sup>13</sup> C (150 MHz) NMR spectroscopic data of compounds 1	-3 (CD <sub>3</sub> OD). <sup>a</sup>

No.	1			2			3				
	$\delta_{C}$		$\delta_{\rm H} (J \text{ in Hz})$	δ <sub>c</sub>			$\delta_{\rm H}$ (J in Hz)	$\delta_{C}$			$\delta_{\rm H} (J \text{ in Hz})$
1								64.8	t		4.53 s
3	140.9	d	7.49 dd (0.6, 2.0)	140.9	d		7.50 dd (0.9, 2.1)				
4	106.9	d	6.19 t (2.0)	106.9	d		6.19 t (2.1)				
5	131.8	d	7.16 dd (0.6, 2.0)	131.9	d		7.23 dd (0.9, 2.1)				
1′	53.4	t	4.52 t (5.9)	53.2	t		4.53 t (5.9)	136.5	s		
2′	47.3	t	4.71 t (5.9)	47.4	t		4.75 td (2.0, 5.8)	129.5	d		7.27 d (8.8)
3′								117.4	d		7.04 d (8.8)
4′								158.4	s		
5′								117.4	d		7.04 d (8.8)
6′								129.5	d		7.27 d (8.8)
1″								101.0	d		4.95 d (7.6)
2″	133.4	S		133.8	S			78.6	d		3.63 m
3″	126.8	d	7.03 d (4.1)	126.2	d		7.03 d (4.1)	78.7	d		3.58 t (8.5)
4″	111.3	d	6.16 d (4.1)	113.1	d		6.26 d (4.1)	71.4	d		3.39 dd (8.5, 9.7)
5″	145.3	s		141.0	S			78.0 <sup>b</sup>	d		3.41 m
6″	55.9	t	4.02 s	60.7	t	А	4.03 d (12.9)	62.5	t	А	3.68 dd (5.3, 12.0)
						В	4.23 d (12.9)			В	3.88 dd (2.0, 12.0)
7″	181.1	d	9.45 s	181.3	d		9.47 s				
1‴				99.5	d		4.79 d (4.1)	110.8	d		5.46 d (1.5)
2‴				70.0	d		3.77 dd (4.2, 10.3)	78.1 <sup>b</sup>	d		3.94 d (1.5)
3‴				71.3	d		3.71 brd (10.3)	80.8	S		
4‴				71.1	d		3.86 dd (1.2, 3.5)	75.5	t	А	3.78 d (9.5)
										В	4.05 d (9.5)
5‴				72.9	d		3.76 brd (11.7)	66.1	t	А	3.51 d (11.4)
										В	3.54 d (11.4)
6‴				62.8	t	А	3.68 dd (4.9, 11.1)				
						В	3.72 dd (6.7, 11.1)				

<sup>a</sup> Assignments were based on <sup>1</sup>H-<sup>1</sup>H COSY, HMQC, HMBC, and NOESY spectroscopic data.

<sup>b</sup> Interchangeable.

(150 MHz) NMR spectra were recorded on an Agilent vnmrs 600 in CP<sub>3</sub>OD with tetramethylsilane as the internal standard. HR-FAB-MS was recorded on a JEOL JMS-7000 mass spectrometer. Silica gel 60 (230–400 mesh, Nacalai Tesque, Inc., Kyoto, Japan) Diaion HP-20 (*Mitsubishi Chemical Co., Tokyo, Japan*), octadecylsilica gel (ODS;100-200 mesh; *Fuji Siliya Chemical, Ltd.*, Aichi, Japan), and sephadex LH-20 *CGE healthcare Bio-Sciences AB*, Uppsala, Sweden). Were used for column chromatography. HPLC [PU-2086 pump and RI-2031 refractive index detector (Jasco, Tokyo, Japan)] was carried out on an ODS column [*Cosmosil 5C<sub>18</sub>-PAQ column* (Nacalai Tesque, Inc., Kyoto, Japan), 25 cm  $\times$  20 mm i.d.] at 35 °C with H<sub>2</sub>O/MeCN [20:1 (HPLC system *I*), 90: 10 (HPLC system *II*), 85: 15 (HPLC system *III*), flow rate 4.0 mL/min].

#### 3.2. Plant material

The seeds of *C. lanatus*, obtained in 2011, in the USA, were purchased from Takada Seeds Co., Ltd. in 2011. A voucher specimen (CL-S-2011-01) was deposited in the Herbarium of the Laboratory of Medicinal Chemistry, Osaka University of Pharmaceutical Sciences.

# Table 2

Melanogenesis inhibitory activities of compounds **1–3** from watermelon seeds.<sup>a,b</sup>

	Melanin content (%)							
	3 μΜ	10 µM	30 µM	100 µM				
1 2 3 Arbutin <sup>c</sup>	$\begin{array}{r} 119.3 \pm 5.1 \\ 103.9 \pm 0.7 \\ 106.6 \pm 2.3 \\ 98.6 \pm 2.5 \end{array}$	$\begin{array}{c} 94.1 \pm 9.8 \\ 99.7 \pm 1.7 \\ 100.9 \pm 1.6 \\ 85.7 \pm 6.7^{*} \end{array}$	$80.3 \pm 9.3^{*}$ $86.0 \pm 2.4^{**}$ $90.5 \pm 1.6^{*}$ $66.6 \pm 6.1^{**}$	$\begin{array}{c} 64.1\pm5.8^{**}\\ 48.6\pm0.6^{**}\\ 65.9\pm2.1^{**}\\ 45.2\pm2.0^{**}\end{array}$				

<sup>a</sup> Melanin content (%) were determined based on absorbances at 405 nm, and comparisons with values for vehicle (DMSO) control (100%). Each value represents the mean ± standard error (S.E.) of three determinations. The concentration of DMSO in the sample solution was 2 mL/mL. Significant differences from the control group were shown as \*p < 0.05, and \*\*p < 0.01.

 $^b$  Compound 2, 3, and arbutin exhibited some cytotoxicities at 100  $\mu M$  (cell viability 2: 70.4  $\pm$  1.9%; 3; 88.4  $\pm$  0.5%; arbutin: 86.2  $\pm$  1.2%).

<sup>c</sup> Positive control.

#### 3.3. Extraction and isolation

The seeds of *C. lanatus* (10 kg) were pulverized in juice mixer, and subjected to extraction with MeOH under reflux (10 L, 1 week  $4\times$ ). The MeOH extract (250 g) was then partitioned between Et<sub>2</sub>O and  $H_2O(1:1, v/v)$ , followed by *n*-BuOH and  $H_2O(1:1, v/v)$ . The *n*-BuOH fraction (17 g) was subjected to Diaion HP-20 CC [Diaion HP-20 (330 g)] resulting in H<sub>2</sub>O Fr. (4.0 g), MeOH Fr. (12.1 g), and acetone Fr. (0.81 g) were eluted with H<sub>2</sub>O, MeOH, and acetone, respectively]. A portion of MeOH Fr. (11.7 g) was subjected to silica-gel CC [silica-gel (200 g)] to yield 5 fractions, A-E [CHCl<sub>3</sub>/ MeOH  $(1:1) \rightarrow (7:3) \rightarrow (9:1) \rightarrow (1:0)$ ]. Among them, Fr. C (5.71 g)was subjected to ODS CC [ODS (50 g)] to yield 6 fractions, C1-C6  $[MeOH/H_2O (1:1) \rightarrow (7:3) \rightarrow (9:1) \rightarrow (1:0)]$ . C-1, eluted with MeOH/H<sub>2</sub>O (1:1), was carried out ODS CC [ODS (50 g)] to give sixteen fractions: C1-1-C1-16. C1-5 (312.02 mg) was fractionated with preparative HPLC (HPLC system III) to nineteen fractions, C1-5-1–C1-5-19. Purification of C1-5-13 gave **2** (1.47 mg; *t*<sub>R</sub> 232.0 min, HPLC system II). Preparative HPLC of C1-9 (62.6 mg), which was eluted with MeCN/H<sub>2</sub>O (1:9), gave **1** (2.1 mg;  $t_{\rm R}$  66.4 min)(HPLC system III). Fr. D (4.51 g) was subjected to ODS CC [ODS (50 g)] to yield 6 fractions, D1–D7 [MeOH/H<sub>2</sub>O (1:1)  $\rightarrow$  (9:1)  $\rightarrow$  (1:0)], followed by D1 (3.12 g) was carried out re-column chromatography with ODS to give D1-1-D1-9. Fr. D1-1 (1.67 g) was subjected to shephadex LH-20 CC [sephadex LH-20 (40 g)] to yield 6 fractions, D1-1-1-D1-1-7 [MeOH/H<sub>2</sub>O (0:1)  $\rightarrow$  (1:0)]. Preparative HPLC of D1-1-2 (466.6 mg), which was eluted with H<sub>2</sub>O, **3** (3.3 mg;  $t_R$ 104.6 min) (HPLC system I).

### 3.4. Acid hydrolysis and determination of sugar configuration

Determination of the absolute sugar configuration was carried out according to a method reported previously with a little modification (Tanaka et al., 2007). A solution of 2 (1 mg) and 3(1 mg), respectively, in 2 M CF<sub>3</sub>COOH (1 mL) was heated under reflux for 3 h. The mixture was extracted with EtOAc (3 times), and the H<sub>2</sub>O layer was concentrated to dryness. The residue was then dissolved in pyridine (0.1 mL) and stirring with L-cystein methyl ester hydrochloride (0.5 mg) at 60 °C for 1 h. The reaction mixture was treated with *o*-tolylisothiocyanate, and heated at 60 °C for 1 h. The reaction mixture was analyzed by reversed-phase HPLC [column: *Cosmosil 5 C*<sub>18</sub>-*PAQ column* (Nacalai Tesque), 250 mm × 4.6 mm i.d. (5 µm); mobile phase: MeCN-H<sub>2</sub>O in 1% AcOH (2:8, v/v); detection: refractive index; flow rate: 1.0 mL/min; column temperature: 35 °C] to identify the derivatives of D-galactose in **2**, and D-glucose and D-apiose in **3** by comparison of their retention times with those of authentic samples ( $t_R$ : D-glucose 28.4 min, L-glucose 26.0 min, D-apiose 45.4 min, D-galactose 22.4 min, L-galactose 23.3 min).

#### 3.5. Cell culture

The cell line B16 4A5 cells were grown in Dulbecco's modified Eagle medium (D-MEM). The medium was supplemented with 10% FBS and antibiotics (100 units/mL penicillin and 100  $\mu$ g/mL streptomycin). The cells were incubated at 37 °C in a 5% CO<sub>2</sub> humidified incubator.

#### 3.6. Viability of melanoma cells

Melanoma cells proliferation was examined according to a method reported previously (Yamada et al., 2011), with a little modification. Briefly, B16 4A5 cells [obtained from Riken Cell Bank (Tsukuba, Ibaraki, Japan)] ( $3 \times 10^3$  cells in 100 µL), preincubated for 24 h, were treated for 72 h with test samples dissolved in dimethyl sulfoxide (DMSO) at a final concentration of 100, 30, 10, and 3 µM, and MTT solution was added. After 3 h of incubation, 20% NDS containing 0.1 M HCl was added to dissolve the formazan produced in the cells. The absorbance of each well was read at 570 nm using a microplate reader.

#### 3.7. Assay of melanin content

The assay of melanin content was performed as described previously (Kikuchi et al., 2013b), with a little modification. B16 4A5 cells ( $3 \times 10^4$  cells in 1000 µL) were pre-incubated in  $\alpha$ -MSH (100 nM) containing medium. Test samples dissolved in DMSO were added to the medium and the cells were cultured for 72 h. The medium was removed and the cells were dissolved in 2 M NaOH containing 10% DMSO. The amount of melanin was determined spectrophotometrically by measuring absorbance at 405 nm using a microplate reader. The optical density of control cells was assumed to be 100%.

# 3.8. 1-[2-(5-Hydroxymethyl-1H-pyrrole-2-carbaldehyde-1-yl)ethyl]-1H-pyrazole (1)

Amorphous solid,  $[\alpha]_D^{27}$  –2.4 (*c* = 0.055, EtOH); UV  $\lambda_{max}^{EtOH}$  nm (log  $\varepsilon$ ): 219.0 (3.33), 255.0 (3.26), 292.5 (3.58); IR  $\nu_{max}^{KBr}$  cm<sup>-1</sup>: 3480, 3424, 3354, 1649, 1585, 1377, 1157, 1089; FAB-MS *m/z*: 220

 $[M+H]^+$ ; HR-FAB-MS m/z: 220.1089 (calcd for 220.1086:  $C_{11}H_{14}N_3O_2$ ).

# 3.9. 1-({[5-( $\alpha$ -D-Galactopyranosyloxy)methyl]-1H-pyrrole-2-carbaldehyde-1-yl}-ethyl)-1H-pyrazole (**2**)

Amorphous solid,  $[\alpha]_D^{27}$ ; 62.4 (*c* = 0.062, EtOH); UV  $\lambda_{max}^{EtOH}$  nm (log  $\varepsilon$ ): 206.5 (3.66), 256.5 (3.84), 282.0 (3.84), 294.0 (3.70); IR  $\nu_{max}^{KBr}$  cm<sup>-1</sup>: 3408, 2922, 1743, 1723, 1656, 1640, 1400, 1353, 1400, 1281, 1029; FAB-MS *m/z*: 382 [M+H]<sup>+</sup>; HR-FAB-MS *m/z*: 382.1617 (calcd for 382.1609: C<sub>17</sub>H<sub>24</sub>N<sub>3</sub>O<sub>7</sub>).

# 3.10. (4-Hydroxyphenyl)methanol 4-[ $\beta$ -D-apiofuranosyl (1 $\rightarrow$ 2)-O- $\beta$ -D-glucopyranoside] (3)

Amorphous solid;  $[α]_D^{27}$  –85.6 (*c* = 0.097, EtOH); UV  $\lambda_{max}^{EtOH}$  nm (log ε): 221.5 (3.95), 270.5 (3.11); IR  $\nu_{max}^{KBr}$  cm<sup>-1</sup>: 3392, 1612, 1509, 1225, 1076; FAB-MS *m/z*: 441 [M+Na]<sup>+</sup>; HR-FAB-MS *m/z*: 441.1375 (calcd for 441.1372: C<sub>18</sub>H<sub>26</sub>O<sub>11</sub>Na).

# Acknowledgments

We thank Dr. Katsuhiko Minoura and Dr. Mihoyo Fujitake (Osaka University of Pharmaceutical Sciences) for the NMR and MS measurements.

# Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.phytol.2015.02.017.

#### References

- Fattorusso, E., Taglialatela-Scafati, O., 2000. Two novel pyrrole-imidazole alkaloids from the Mediterranean sponge Agelas oroides. Tetrahedron Lett. 41, 9917–9922.
- Hassan, L.E.A., Koko, W.S., Osman, E.B.E., Dahab, M.M., Sirat, H.M., 2011. In vitro antigiardial activity of *Citrullus lanatus* var. citroides extracts and cucurbitacins isolated compounds. J. Med. Plants Res. 5, 3338–3346.
- Kikuchi, T., Okada, R., Harada, Y., Ikushima, K., Yamakawa, T., Yamada, T., Tanaka, R., 2013a. Cucurbitane-type triterpenes from *Citrullus lanatus* (watermelon) seeds. Nat. Prod. Commun. 8, 1367–1369.
- Kikuchi, T., Takebayashi, M., Shinto, M., Yamada, T., Tanaka, R., 2013b. Three new multiflorane-type triterpenes from pumpkin (*Cucurbita maxima*) seeds. Molecules 18, 5568–5579.
- Kumar, V., Kaur, K., Gupta, G.K., Sharma, A.K., 2013. Pyrazole containing natural products: synthetic preview and biological significance. Eur. J. Med. Chem. 69, 735–753.
- Lako, J., Trenerry, V.C., Wahlqvist, M., Wattanapenpaiboon, N., Sotheeswaran, S., Premier, R., 2006. Phytochemical flavonols, carotenoids and the antioxidant properties of a wide selection of Fijian fruit, vegetables and other readily available foods. Food Chem. 101, 1727–1741.
- Ripperger, H., Seifert, K., 1975. Cucurbitacins from Citrullus lanatus. Tetrahedron 31, 1561–1563.
- Shanghai Scientific and Technical Publishers, Shogakukan, Inc. (Eds.), Chuyakudaijiten, vol. II. Shogakukan, Inc., Tokyo, pp. 1374–1375.
- Tanaka, T., Nakashima, T., Ueda, T., Tomii, K., Kouno, I., 2007. Facile discrimination of aldose enantiomers by reversed-phase HPLC. Chem. Pharm. Bull. 55, 899–901.
- Yamada, T., Muroga, Y., Jinno, M., Kajimoto, T., Usami, Y., Numata, A., Tanaka, R., 2011. New class azaphilone produced by a marine fish-derived *Chaetomium globosum*. The stereochemistry and biological activities. Bioorg. Med. Chem. 19, 4106–4113.