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Design and synthesis of a series of α -benzyl phenylpropanoic acid-type peroxisome proliferator-activated receptor (PPAR) gamma partial agonists with improved aqueous solubility



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ABSTRACT

In the continuing study directed toward the development of peroxisome proliferator-activated receptor gamma (hPPARy) agonist, we attempted to improve the water solubility of our previously developed hPPAR γ -selective agonist **3**, which is insufficiently soluble for practical use, by employing two strategies: introducing substituents to reduce its molecular planarity and decreasing its hydrophobicity via replacement of the adamantyl group with a heteroaromatic ring. The first approach proved ineffective, but the second was productive. Here, we report the design and synthesis of a series of α -benzyl phenylpropanoic acid-type hPPAR γ partial agonists with improved aqueous solubility. Among them, we selected (**R**)-7j, which activates hPPAR γ to the extent of about 65% of the maximum observed with a full agonist, for further evaluation. The ligand-binding mode and the reason for the partial-agonistic activity are discussed based on X-ray-determined structure of the complex of hPPAR γ ligand-binding domain (LBD) and (*R*)-7j with previously reported ligand-LDB structures. Preliminal apoptotic effect of (**R**)-**7** gagainst human scirrhous gastric cancer cell line OCUM-2MD3 is also described.

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1. Introduction

Peroxisome proliferator-activated receptor γ (hPPAR γ) is the most extensively investigated subtype among the three hPPAR subtypes, hPPAR α , hPPAR δ and PPAR γ . It is expressed abundantly in adipose tissue and macrophages,¹ and is up-regulated in various types of cancer cells.^{2,3} hPPAR γ is a master regulator of fatty acid and glucose homeostasis,⁴⁻⁶ but also has a range of other activities.^{7–10} Some hPPAR γ ligands, such as pioglitazone (1) have been

Abbreviations: NR, nuclear receptor; hPPAR, human peroxisome proliferatoractivated receptor; LBD, ligand-binding domain; PDB, protein data bank.

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reported to induce cell differentiation and apoptosis in malignant tumor cells,¹¹ and hPPAR γ ligands may have potential as anticancer agents. We are interested in the potential application of hPPARy ligands to treat scirrhous gastric cancer, which has a poor survival rate of less than 20%.12,13

We have extensively studied the structural design and synthesis of subtype-selective hPPAR ligands based on phenylpropanoic acid as a scaffold. We have successfully created a series of α -substituted phenylpropanoic acid-type hPPAR α -selective agonists,^{14,15} a hPPAR $\alpha\delta$ dual agonist (TIPP-401),^{16,17} a hPPAR δ -selective agonist (TIPP-204),¹⁸ and a hPPAR $\alpha\delta\gamma$ pan agonist (TIPP-703 (2)).¹⁹ Furthermore, we recently reported a hPPARy-selective agonist (MEKT-1 (3)), which was designed on the basis of a comparative X-ray crystallographic analysis of **3** complexed with the hPPARγ ligand binding domain (LBD) (Fig. 1).²⁰

Although **3** exhibited potent, hPPARγ-selective in vitro transactivation activity, its aqueous solubility is very poor. Therefore, for

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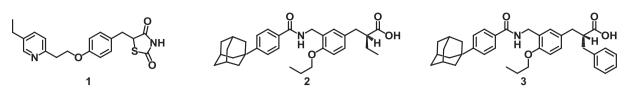


Figure 1. The chemical structures of pioglitazone (1) and our hPPAR γ agonists 2 and 3.

in vivo evaluation studies, it was necessary to improve the solubility of **3**. Here, we report the design, synthesis and in vitro pharmacological evaluation of a series of α -benzyl phenylpropanoic acid-type hPPAR γ -selective partial agonists with improved aqueous solubility, as candidate anti-scirrhous gastric cancer agents. We also discuss the binding mode of these compounds and the reason why they exhibit partial agonistic activity, based a comparison of the X-ray-determined structures of ligand complexes with the hPPAR γ ligand binding site.

2. Experimental design

2.1. Chemistry

The synthesis of the present series of compounds is illustrated in Scheme 1 (routes 1, 2). Compounds (\mathbf{R})- $7\mathbf{a}$ -(\mathbf{R})- $7\mathbf{c}$ and (\mathbf{R})- $7\mathbf{e}$ were prepared according to reported methods, by amide-alkylation²¹ of the formyl derivative **6** with the corresponding benzamides, followed by LiOOH hydrolysis (route 1).

Compounds (R)-7d and (R)-7f-(R)-7l were prepared from 6 by an alternative method (route 2). Compound 6 was treated with hydroxylamine HCl, and subsequent reduction with 10% Pd on carbon afforded aminomethylbenzene derivative 9 as the hydrochloric acid salt. Compound 9 was treated with the appropriate benzoic acid derivatives in the presence of the condensing agent, and subsequent hydrolysis with LiOOH afforded the desired compounds.

2.2. Hydrophobicity assessment with HPLC

Reversed-phase HPLC analyses were performed on an analytical column (PegasilODS SP100 column (4.6 mm \times 150 mm; flow rate

of 1 mL/min; solvent, MeCN/0.1% TFA = 3:1 v/v; detection at 254 nm or 296 nm).

2.3. Thermodynamic aqueous solubility

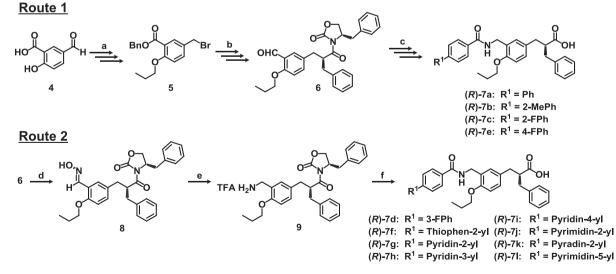
Thermodynamic solubility determination was based on the method of Avdeef and Testa.²² The concentration of sample solution was calculated using a previously determined calibration curve, corrected for the dilution factor of the sample.

2.4. Transient transfection assays

The African green monkey kidney cell line CV-1 was used for the transfection assay. CV-1 cells were seeded in 24-well plates and cultured for 24 h. The transfection mixtures were added to the cells and the plates were incubated for 5 h according to the manufacturer's instructions. After the transfection, incubation was continued for an additional 40 h in the presence of the assay compounds or reference compounds. Cell lysates were prepared with a lysis buffer and used in the luciferase and β GAL assays. Luciferase and β GAL activities were measured according to the methods of Umesono et al.,²³ with slight modifications. A substrate reagent kit (Picagene, Toyo Ink) was used for the luciferase assay.

2.5. Expression, purification, and crystallization

Human PPAR γ LBD (residues 203–477) was prepared as previously described,²⁴ that is, recombinant hPPAR γ LBD was expressed as an N-terminal His-tagged protein in *Escherichia coli* BL21 (DE3), using the pET28a vector (Novagen). The protein was purified by means of several steps of column chromatography (His-Trap HP column (GE Healthcare), HiPrep 26/10 desalting column (GE



Scheme 1. Synthetic route to the present series of compounds. Reagents and conditions: (a) (1) BnBr, KHCO₃, DMF, rt, overnight, quant.; (2) nPrI, K₂CO₃, DMF, 60 °C, overnight, quant.; (3) NaBH₄, EtOH, rt 2 h, 91%; (4) PBr₃, diethylether, 0 °C, 1.5 h, 52%; (b) (1) (\mathbf{R})-3-(3-phenylpropanoyl)-4-benzyloxazolidin-2-one, LiHMDS, dehydrated THF, -50 to 0 °C, 2 h, 64%; (2) H₂, 10% Pd-C, AcOEt, rt, 3 h, 55%; (3) BH₃-THF, dehydrated THF, 0 °C, overnight; (4) activated MnO₂, CH₂Cl₂, rt, overnight, 65% (2 steps); (c) (1) benzamide derivative, triethylsilane, TFA, toluene, reflux, 48 h, 97%; (2) LiOH-H₂O, 30% H₂O₂, THF/H₂O = 4:1 (v/v), 0 °C, 2.5 h then rt 3 h, 86%; (d) hydroxylamine HCl, pyridine, ethanol, reflux, 24 h, 93%; (e) H₂, 10% Pd-C, ethanol-HCl, rt, 3 h, 96%; (f) (1) benzoic acid derivative, diethyl cyanophosphonate, triethylamine, DMF, rt, 12-24 h, 46–82%; (2) LiOH-H₂O, 30% H₂O₂, THF/H₂O = 4:1 (v/v), 0 °C, 2.5 h then rt 3 h, 86%; d) hydroxylamine HCl, pyridine, ethanol, reflux, 24 h, 93%; (e) H₂, 10% Pd-C, ethanol-HCl, rt, 3 h, 96%; (f) (1) benzoic acid derivative, diethyl cyanophosphonate, triethylamine, DMF, rt, 12-24 h, 46–82%; (2) LiOH-H₂O, 30% H₂O₂, THF/H₂O = 4:1 (v/v), 0 °C, 2.5 h then rt 3 h, 86%; (d) hydroxylamine HCl, pyridine, ethanol, reflux, 24 h, 93%; (e) H₂, 10% Pd-C, ethanol-HCl, rt, 3 h, 96%; (f) (1) benzoic acid derivative, diethyl cyanophosphonate, triethylamine, DMF, rt, 12-24 h, 46–82%; (2) LiOH-H₂O, 30% H₂O₂, THF/H₂O = 4:1 (v/v), 0 °C, 2.5 h then rt 3 h, 51-94%.

Healthcare), Benzamidine FF column (GE Healthcare), HisTrap HP column (GE Healthcare), HiTrap Q HP column (GE Healthcare), Hi-Prep 26/10 desalting column (GE Healthcare)). Apo PPARγ LBD crystals were soaked in ligand solution for two or three weeks prior to crystallography.

2.6. Diffraction data collection

The diffraction data for hPPAR γ LBD complex with (*R*)-**7j** were collected on BL38B1 at SPring-8 (Harima, Japan) and were processed using HKL-2000.²⁵

2.7. Structure analysis and refinement

The structure of hPPAR α LBD complex with (**R**)-**7j** was solved by the molecular-replacement method with the program CNS^{26} using the previously reported structures as probes. The correctly positioned molecules were refined with CNS and $O.^{27}$ The initial atomic model of (**R**)-**7j** was built using *MOE* (Ryoka Systems Inc.) and topology and parameter files for the refinement were generated by the *HIC-Up* server.²⁸

The structure of hPPAR γ LBD was solved by the molecular replacement method using Molrep. Apo PPAR γ LBD structure (PDB code 1PRG) was used as a search model and refined with Refmac5. The crystallographic data and data collection statistics of all the crystals are provided in Table 1.

2.8. Adipogenesis assay

The adipogenesis assay was performed as previously described.²⁹ Mouse adipogenic fibroblast cells, 3T3-L1, were cultured in DMEM with 10% FBS. After the cells became confluent, they were pretreated with 1 μ M dexamethasone (Dex), 0.5 mM 3-isobutyl-1-methylxanthine (IBMX), and 5 μ g/mL insulin to initiate adipogenesis for 2 days. Various concentrations of ligands were added to the cells, and after 6 days, the cells were stained with Oil Red O. The degree of adipogenesis was quantitatively measured in terms of absorbance at OD550. The error bars are the standard deviations (SD).

2.9. Cell lines

A human scirrhous gastric cancer cell line, OCUM-2MD3³⁰ was cultured in DMEM/F12 (Ham) (1:1) (Invitrogen, Carlsbad, CA). A normal human fibroblast cell line, OUMS-24,³¹ was cultured in DMEM (Nissui). All media were supplemented with 10% fetal bovine serum (Invitrogen), 100 μ g/mL kanamycin (Meiji Seika, Tokyo, Japan) and 0.5 μ g/mL Fungizone (Invitrogen).

2.10. Apoptosis assay

The apoptosis assay was performed as previously described.³² Cells were inoculated into flat-bottomed 6-well plates, incubated for 24 h and then treated with test compound. Seventy-two hours later, 1 μ g/mL Hoechst 33342 (Invitrogen) and 5 μ g/mL propidium iodide (Sigma) were added to the medium and the cells were incubated in the dark for 20 min. All of the cells were collected on a glass slide and evaluated with a fluorescence microscope. Cells with fragmented or shrunken nuclei were counted as apoptotic cells.

3. Results and discussion

3.1. Structure-activity and structure-solubility relationships

In order to create phenylpropanoic acid-type PPARγ-selective agonists with improved aqueous solubility, we focused on the

biphenyl derivative (**R**)-**7a** as a lead compound. The poor aqueous solubility of **3** might be attributed primarily to the presence of an adamantyl substituent at the hydrophobic tail part of the molecule, even though this substituent is one of the critical determinants for potent hPPAR γ -selective agonistic activity. Although biphenyl derivative (**R**)-**7a** exhibited decreased activity compared with **3**, it still showed hPPAR γ agonistic activity at sub-micromolar concentration (EC₅₀ = 86 nM in our assay system).

We have previously reported biphenylcarboxylic acid-type hPPAR δ partial agonists with improved aqueous solubility.³³ In that case, the improvement of solubility was attributed to two factors, that is, disruption of molecular planarity by the introduction of a substituent at the *ortho*-position of the attached benzene ring, and decreased hydrophobicity due to the introduction of a heteroaromatic ring. We also applied the same strategies in the present work, and the results are summarized in Table 1.

We evaluated the aqueous solubility of the compounds by calculating Clog P values³⁴ and we also determined the retention times of the compounds in reversed-phase high-performance liquid chromatography (RP-HPLC). In the case of adamantyl derivative **3**, the Clog P value is extremely high, more than 9, and the RP-HPLC retention time is about 30 min. These data clearly indicated that **3** is extremely hydrophobic, being even more hydrophobic than typical steroid drugs. However, in the case of the phenyl derivative (**R**)-**7a**, Clog P is about 7.1, so that this compound is about 100 times more hydrophilic than **3**. In accordance with this, compound (**R**)-**7a** was eluted much faster than **3**, with a retention time of 7.9 min.

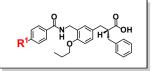
The ortho-substituted derivatives ((R)-**7b**, (R)-**7c**) exhibited comparable ((R)-**7b**) or somewhat increased ((R)-**7c**) hPPAR γ agonistic activity. However, their aqueous solubility was not improved; indeed in the case of (R)-**7b**, it was decreased as compared to the non-substituted derivative (R)-**7a**. As expected, introduction of a small fluorine atom at the *meta* or *para* position ((R)-**7d**, (R)-**7e**) of the distal benzene ring did not significantly affect the in vitro biological activity or the aqueous solubility.

It is interesting to note that thiophen-2-vl phenvl derivative (**R**)-**7f** exhibited somewhat greater aqueous solubility with retention of the hPPAR γ agonistic activity, as compared to the phenyl derivative (**R**)-7a. Although the thiophene ring, a well-known bioisostere of the benzene ring, is less hydrophobic than the benzene ring, the hydrophobicity of (R)-7f was increased. The reason for this is unclear. In any event, unsymmetrization of the biphenyl moiety of (R)-7a by the introduction of a substituent at the ortho-position of the benzene ring or by replacement of one of the benzene rings with an bioisostere was not very effective in this case. Therefore, we next tried to substitute a heterocyclic ring (pyridine, ((R)-7g-(R)-7i), pyrimidine ((R)-7j, (R)-7l), pyrazine ((R)-**7k**)) at the distal benzene ring. The results are included in Table 1. The position of the nitrogen atom of the monoazine is important for potent hPPAR γ agonistic activity, that is, the hPPAR γ agonistic activity decreased in the order of 2-pyridyl ((\mathbf{R})-7g) > 3-pyridyl $((\mathbf{R})-\mathbf{7h}) > 4$ -pyridyl $((\mathbf{R})-\mathbf{7i})$, and pyrimidin-2-yl $((\mathbf{R})-\mathbf{7j}) >$ pyradin-2-yl ((**R**)-7h) > pyrimidin-5-yl ((**R**)-7i). These results indicated that the distal nitrogen atom interacts unfavorably with the hPPAR_Y LBD, causing a considerable decrease of the activity.

As for aqueous solubility, the introduction of a pyridine ring instead of the benzene ring as the distal substituent improved the aqueous solubility by 10-fold, as judged from $\operatorname{Clog} P$ (7.18 to 6.04 and 5.83). In accordance with this, (**R**)-7**g**–(**R**)-7**i** (retention times: 3.6, 2.3 and 1.8 min, respectively) were eluted much faster than the phenyl analogue (**R**)-7**a** (retention time: 7.9 min). These compounds exhibited similar ClogP values, whereas their retention times were somewhat different, that is, the compounds eluted faster in the order of (**R**)-7**i** (4-pyridyl) > (**R**)-7**b** (3-pyridyl) > (**R**)-7**g** (2-pyridyl). Thus, it appears that these two hydrophobic parameters do not necessarily correlate well.

Table 1

hPPARy transactivation activities, ClogP values and retention times (rt) on reversed-phase HPLC of the present series of derivatives



No.	R	EC ₅₀ ^a (nM)	Efficacy ^b	Clog P ^c	rt ^d (min)	Aqueous solubility ^f)	No.	R	EC ₅₀ ª (nM)	Efficacy ^b	Clog P ^c	rt ^d (min)	Aqueous solubility ^f)
		hPPARγ				(µg/mL)			hPPARγ				(µg/mL)
3		3.60	NT ^e	9.58	30.5 ± 0.02	<1.00							
7a	\mathbb{O}^{λ}	85.9	95.7	7.18	7.89 ± 0.01	<1.00	7g		186.1	113	6.04	3.57 ± 0.01	17.0
7b		73.8	72.7	7.38	9.71 ± 0.01	<1.00	7h	$\operatorname{res}_{\mathbb{N}}^{\lambda}$	844	100	5.83	2.31 ± 0.01	69.5
7c	\mathbb{C}_{F}^{λ}	49.5	95.0	7.37	7.66 ± 0.01	<1.00	7i	$\operatorname{res}^{\lambda}$	1688	108	5.83	1.78 ± 0.01	34.1
7d	$\mathbf{P}^{\lambda}_{\mathbf{F}}$	28.7	99.7	7.37	8.10 ± 0.01	<1.00	7j	_N_∕ ⟨N	34.6	65.0	5.12	4.25 ± 0.01	301
7e	F	102.6	104	7.37	7.82 ± 0.02	<1.00	7k	${\rm Arc}_{\rm N}^{\rm N}{\rm Arc}_{\rm N}^{\rm Arc}$	713.9	58.2	5.12	3.845 ± 0.01	32.2
7f	Cs l	65.2	76.0	7.37	7.42 ± 0.01	<1.00	71		3095	82.7	4.91	3.23 ± 0.01	190

 a,b EC₅₀ values and % efficacy are given relative to the positive control, rosiglitazone for PPAR γ .

^c CLogP values were estimated with ChemDraw Ultra version 10.0.

^d PegasilODS SP100 column (4.6 mm × 150 mm; flow rate of 1 mL/min; solvent, MeCN: 0.1% TFA = 3:1 v/v; detection at 254 nm or 296 nm).

e NT means not tested.

^f Aqueous solubility in phosphate buffer (pH = 7.2–7.4).

In the case of (\mathbf{R}) - $7\mathbf{j}$ - (\mathbf{R}) - $7\mathbf{l}$, the position of the nitrogen atom is also important for potency towards hPPAR γ . The pyrimidin-2-yl derivative $((\mathbf{R})$ - $7\mathbf{j})$ exhibited potent activity, comparable to that of the phenyl analog $((\mathbf{R})$ - $7\mathbf{a}$). However, others, that is, the pyrazin-2-yl $((\mathbf{R})$ - $7\mathbf{k})$ and pyrimidin-5-yl $((\mathbf{R})$ - $7\mathbf{l})$ derivatives, exhibited decreased hPPAR γ agonistic activity. These results indicated that *meta* position nitrogen atoms interact unfavorably with hPPAR γ LBD, thereby decreasing the activity.

As for aqueous solubility, (\mathbf{R}) - $7\mathbf{j}$ - (\mathbf{R}) - $7\mathbf{l}$ exhibited about 100fold increased aqueous solubility, as judged from $C\log P$ (from 7.18 for (\mathbf{R}) - $7\mathbf{a}$ to 5.12, 5.12 and 4.91, respectively). In accordance with this, these diazines were eluted more quickly than the phenyl analogue (\mathbf{R}) - $7\mathbf{a}$. However, they were eluted more slowly than the pyridine derivatives (\mathbf{R}) - $7\mathbf{g}$ - (\mathbf{R}) - $7\mathbf{i}$. These results indicate that introduction of hydrophilic aromatics, such as a pyridine ring or pyrimidine ring, is an effective strategy to improve the aqueous solubility of the present series of compounds.

We then assessed the exact thermodynamic solubility of these compounds. As expected, all the biphenyl derivatives ((R)-7a-(R)-7f) exhibited very poor aqueous solubility (<1 µg/mL), but the introduction of a nitrogen atom at the distal benzene ring improved the solubility to 17.0 µg/mL ((R)-7g) and 301 µg/mL ((R)-7j). The number of introduced nitrogen atoms was correlated with the thermodynamic solubility. We selected compound (R)-7j for further evaluation, based on its activity towards hPPAR γ and good aqueous solubility. The intrinsic activity of (R)-7j suggests that it is a partial agonist, with the ability to activate hPPAR γ to the extent

of about 65% of the maximum activity observed with the full-agonist pioglitazone (1).

3.2. Partial-agonistic activity of (R)-7j

To characterize the hPPAR γ partial-agonistic profile of (*R*)-**7j**, we compared the ability of **2** (an hPPAR γ full-agonist), and (**R**)-**7j** to stimulate adipocyte differentiation of murine preadipocyte 3T3-L1 cells (Fig. 2). It is well known that PPAR γ agonists promote the conversion of a variety of preadipocyte and stem cell lines into mature adipocytes.³⁵ Both compounds in the concentration range from 1 to 1000 nM stimulated adipocyte differentiation and induced triglyceride accumulation, as indicated by Oil Red O staining. Compound 2 exhibited dose-dependent adipocyte differentiationpromoting ability, providing the maximum absorption intensity $(OD_{550} = 0.2 \text{ in these experiments})$. On the other hand, (**R**)-7j stimulated adipocyte differentiation at the lowest concentration of 1 nM, but dose-dependency was not maintained at high concentration and the absorption intensity was about 60% of the maximum at every concentration tested. These results indicate that (**R**)-7i is a hPPAR γ partial-agonist.

3.3. Structural features of the complex of (R)-7j with hPPAR γ ligand-binding domain (LBD)

To understand the structural basis for the activity of (R)-7j, we examined the three-dimensional structure of (R)-7j complexed

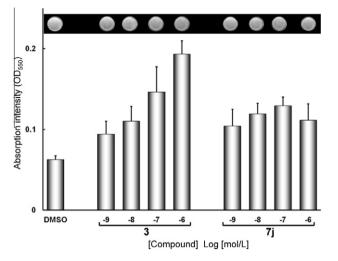


Figure 2. Adipogenesis-inducing activity of **3** and (**R**)-**7j**. After pretreatment with 1 μ M Dex, 0.5 mM IBMX, and 5 μ g/mL insulin, 3T3-L1 cells were incubated with the indicated concentrations of **3** and (**R**)-**7j**, respectively. Oil red O-stained cells are shown at the top and are quantified at the bottom by measurement of OD₅₅₀ (±SD).

with the hPPAR γ LBD by means of X-ray crystallographic analysis. The resulting three-dimensional structures of hPPAR γ LBD–(*R*)-**7j** complex are illustrated in Figure 3A–G. The crystallographic data and refinement statistics are summarized in Table 2.

Like the reported structure of the hPPAR γ LBD–**3** complex, hPPAR γ LBD has a three-layered sandwich folded structure comprising mainly α -helices, and the overall interaction mode with the ligand is basically conserved.

The pyrimidin-2-yl derivative (\mathbf{R})- $7\mathbf{j}$ exhibited somewhat weaker hPPAR γ agonistic activity than the adamantyl derivative $\mathbf{3}$, and the X-ray crystallographic analysis of hPPAR γ LBD–(\mathbf{R})- $7\mathbf{j}$ complex clearly revealed the reason for this difference. The pyrimidin-2-yl ring and adamantyl ring are hosted in the same binding pocket, that is, the large Y3 arm³⁶ cavity composed of the side chains of amino acid residues Leu255, Glu259, Ile262, Phe264, Arg280 and Ile281 (Fig. 3D and E). The planar pyrimidin-2-yl ring would interact less effectively with this large hydrophobic cavity, as compared to the three-dimensionally expanded hydrophobic adamantyl ring, especially as regards the hydrophobic interactions with Ile262, Phe264, and Ile 281, which are positioned on both sides of the pyrimidin-2-yl ring of (\mathbf{R})- $7\mathbf{j}$ (Fig. 3F and G).

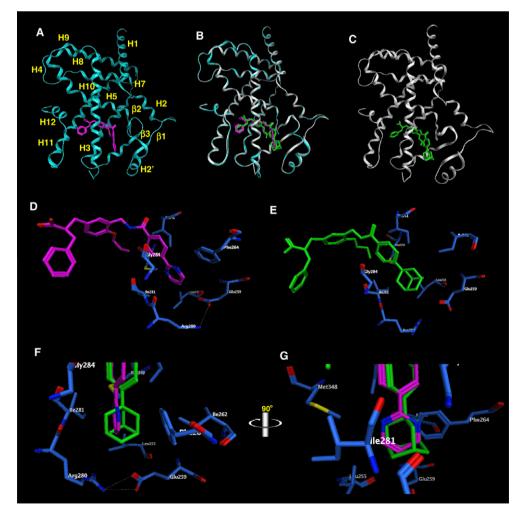


Figure 3. (A–G) Crystal structures of hPPARγ LBD–3 complex (PDB: 2ZNO) and hPPARγ LBD–(*R*)–7j complex (PDB: 3VSO). (A) Whole structure of hPPARγLBD–(*R*)–7j complex. Protein is represented as a blue ribbon model and (*R*)–7j is depicted as a magenta cylinder model. The numbering of the second structure is also depicted. The nomenclature of the helices is based on the RXR-α crystal structure. (B) The superimposed structures of **3** and (*R*)–7j complexed with hPPARγ LBD. (C) Whole structure of hPPARγ LBD–**3** complex. Protein is represented as a white ribbon model and **3** is depicted as a green cylinder model. (D) Zoomed view of the binding mode of the hydrophobic tail part of **3** in the Y3 arm. The side chains of amino acids of the Y3 arm are depicted as blue cylinder models. (E) Zoomed view of the binding mode of the hydrophobic tail part of **3** and (*R*)–7j. (G) image F rotated by 90°.

Table :	2
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Crystallographic data and refinement statistics

Crystal	PPARg-7j
Data collection	
Space group	C2
Unit cell constants	
a (Å)	93.35
b (Å)	61.71
<i>c</i> (Å)	118.98
β (°)	102.79
Wavelength (Å)	1
Resolution ^a (Å)	50.0-1.97
(Highest shell)	2.04-1.97
Measured reflections	172342(16510)
Unique reflections	46474(4586)
Completeness	99.2(99.7)
$I/\sigma(I)$	17.2(3.7)
Redundancy	3.7(3.6)
R _{merge} ^b	4.2(40.1)
Refinement	
Resolution (Å)	30.0-2.0
$R_{\rm work}^{\rm c}/R_{\rm free}^{\rm d}$ (%)	21.9/25.7
Number of atoms	
Protein	4184
Water	117
Ligand	38
Average B-factor (Å ²)	
Protein	41.3
Water	36.7
Ligand	57
r.m.s.d.	
Bond Lengths (Å)	0.007
Angles (°)	1.0
PDB code	3VSO

^a Values in parentheses indicate statistics for the highest resolution shells.

^b $R_{\text{merge}} = \sum_{hkl} \sum_{i} |I_i(hkl) - \langle I(hkl) \rangle | / \sum_{hkl} \sum_{i} |I_i(hkl), \text{ where } \langle I(hkl) \rangle$ is the mean I(hkl) over symmetry-equivalent reflections.

 $c R_{work} = \sum_{hel} F_{obs} - F_{calc} |/ \Sigma_{hkl} | F_{obs} |$, where F_{obs} and F_{calc} are the observed and calculated structure factors, respectively.

 $^{\rm d}$ $R_{\rm free}$ was calculated using 5% of the total reflections, which were chosen randomly and omitted from the refinement.

It is interesting to note Glu259 and Arg280 show a hydrogen bonding interaction to form a 'salt bridge'³⁷ that determines the depth of the Y3 arm. The rank order of the decrease in hPPAR γ agonistic activity of 4-pyridyl ((**R**)-**7i**) > 3-pyridyl ((**R**)-**7h**) > 2-pyridyl ((**R**)-**7g**) derivatives might be attributed to unfavorable interaction between the nitrogen atom of the distal heteroaromatic ring and the anionic side chain of Glu259.

3.4. Structural basis of the partial agonism of (*R*)-7j towards hPPAR γ

The qualitative characteristics of corepressor dissociation and coactivator recruitment are reported to be regulated differentially by the three-dimensional structure around the C-terminal H12 helix and the surrounding helixes of the nuclear receptor.³⁸ In the case of the binding of a full agonist, such as rosiglitazone, the H12 helix forms a part of the coactivator-binding surface, along with the H3 helix and H5 helix.³⁹ In this case, the positions and directions of the side chains of two distinct hydrophilic amino acids, Lys301 (in H3 helix) and Glu471 (in H12 helix), are reported to be critically important, because these amino acids clamp the LXXLL motif (L, leucine; X, any amino acid) of the coactivator, and dock it appropriately to express full-agonistic activity.³⁹ Therefore, we first compared the three-dimensional structures of two PPAR γ full agonists, rosiglitazone and farglitazar,⁴⁰ complexed with hPPAR γ LBD⁴¹ (Fig. 4A–F).

The main frameworks of hPPARγ LBD complexed with rosiglitazone and farglitazar are depicted in Figure 4A, B, and a superposition of these two complexes is shown in Figure 4C. The overall folds are well matched in these two complexes, and the positions and the directions of the side chains of Lys301 and Glu471 overlap well in these structures. The root mean square deviations (RSMDs) of these two amino acid residues are 0.71 and 0.44 Å, respectively (Fig. 4E, F).⁴² These results indicate that the two hPPAR γ full agonists will be able to recruit coactivator complexes with almost the same efficacy, eliciting full agonistic activity.

On the other hand, the overall structures of hPPARy LBD complexed with hPPAR γ partial agonists, including (*R*)-7j, are somewhat different. The frameworks of hPPARy LBD complexed with several representative hPPARγ partial agonists (**(R)-7j**, LT127,⁴³ phenoxyacetic acid derivative,⁴⁴ GW-0072,⁴⁵ pyrazole derivative,⁴⁶ and GQ-16⁴⁷) are depicted in Figure 5A–F, and a superposition of the six complexes is shown in Figure 5G. The folds of the upper half of the hPPARy LBD, composed of the H1, H2, H4, H5, H8 and H9 helixes, are well matched in these complexes. However, the folds of the lower half of the hPPAR_Y LBD, composed of H2', H3, H6, H10–H12 helixes and β 1– β 3 sheets, are somewhat different. The position of Lys301 in the H3 helix varies somewhat (RMSD = 1.10 Å, as compared to 0.71 Å for full agonists) in these structures. However, the side chain of Glu471 in the H12 helix faces in various directions in these complexes, and the RSMD is much larger (1.89 Å as compared to 0.44 Å for full agonists). These substantial differences in structure might be one of the driving forces behind the partial agonistic activity of these compounds, including (**R**)-7j.

To better understand the key differences in the binding of hPPAR γ full agonists and hPPAR γ partial agonists, we focused on the binding mode of these agonists to Tyr473 in the H12 helix. As already reported, Tyr473 is believed to play a critical role in the full agonist-mediated stabilization of the LBD of hPPAR γ (Fig. 6A–C). In the case of the full agonist rosiglitazone, the side chain phenol oxygen of Tyr473 forms a tight hydrogen bond with the nitrogen atom of the thiazolidine-2,4-dione ring. Furthermore, the side chain phenol oxygen of Tyr473 has a hydrogen bonding interaction with the carbonyl oxygen atom of the thiazolidine-2,4-dione ring. These hydrogen bonding interactions tightly fix Tyr473 in the H12 helix, forming a section of the tranccriptional coactivator-binding surface of the PPAR γ LBD Figure 6A.

In the case of the partial agonist N-[1-(4-fluorophenyl)-3-(2-thenyl)-1H-pyrazole-5-yl]-3,5-bis- (trifluoromethyl)benzenesulfonamide (PDB: 2GOH), the 4-fluorophenyl ring is located close to the H12 helix, but it has no hydrophobic or hydrogen bonding interaction with this helix (Fig. 6C). Therefore, there is no significant ligand-mediated interaction to fix Tyr473 in the H12 helix. The absence of interaction (or the presence of weak interaction) with Tyr473 in the H12 helix could provide the structural basis for the partial agonist functionality of not only this compound, but also the other neutral-and carboxylic acid-type hPPAR γ partial agonists shown in Figures 5B–F 6B.

In the case of α -benzylphenylpropanoic acid (**R**)-**7j**, the acidic carboxylic acid moiety hydrogen bonds with the side chain phenol oxygen of Tyr473. However, the distance between these two functionalities is longer than that in the case of hPPAR γ LBD complexed with the thiazolidine-2,4-dione-type hPPAR γ full agonist, rosiglitazone. Furthermore, there is only a single hydrogen bonding interaction between the acidic moiety of (**R**)-**7j** and Tyr473 in the H12 helix. Considering these two factors, Tyr473 in the H12 helix is thought to be fixed less effectively when (**R**)-**7j** is bound to hPPAR γ -LBD, resulting in partial agonistic activity Figure 6B.

3.5. Preliminal apoptotic effect of (*R*)-7j on scirrhous gastric cancer cell line

In order to investigate the potential of hPPAR γ partial agonist (*R*)-**7j** as an anti-scirrhous gastric cancer agent, we preliminally compared the apoptosis-inducing effect of (*R*)-**7j** and **3**, together

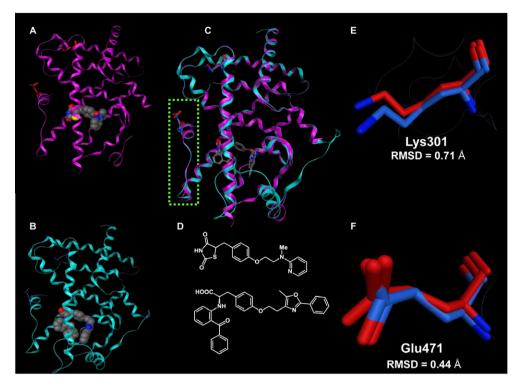


Figure 4. (A, B) Crystal structures of hPPARγ LBD–rosiglitazone (full agonist) complex (PDB: 2PRG) and hPPARγ LBD–farglitazar (full agonist) complex (PDB: 1FM9). (C) The superimposed structures of hPPARγ LBD–rosiglitazone complex and hPPARγ LBD–farglitazar. Both ligands are depicted as cylinder models, and Lys301 and Glu473 are highlighted as cylinder models in magenta (rosiglitazone complex) and cyan (farglitazar complex). The N-terminal AF2 region (H12 helix) is highlighted as a light green dotted box. (D) Chemical structures of rosiglitazone and farglitazar. (E) Zoomed view of the superimposed Lys301 of hPPARγ LBD–rosiglitazone complex and hPPARγ LBD–farglitazar complex. Jr Complex Lys301 is highlighted as a cylinder model in magenta (rosiglitazone complex) or blue (farglitazar). (F) Zoomed view of the superimposed Glu473 of hPPARγ LBD–rosiglitazone complex and hPPARγ LBD–farglitazar complex. Glu473 is highlighted as a cylinder model in magenta (rosiglitazone complex) or blue (farglitazone. (F) Zoomed view of the superimposed Glu473 or hPPARγ LBD–rosiglitazone complex and hPPARγ LBD–farglitazar complex. Glu473 is highlighted as a cylinder model in magenta (rosiglitazone complex) or blue (farglitazone. (F) Zoomed view of the superimposed Glu473 or hPPARγ LBD–rosiglitazone complex) or blue (farglitazar). (F) Zoomed view of the superimposed Glu473 or hPPARγ LBD–rosiglitazone complex and hPPARγ LBD–farglitazar complex. Glu473 is highlighted as a cylinder model in magenta (rosiglitazone complex) or blue (farglitazone comple

with the hPPAR γ full agonist troglitazone, on human scirrhous gastric cancer cell line OCUM-2MD3. Troglitazone was used as a positive control, based on reports that it suppresses the growth of gastric cancer cell lines, such as MKV45 cells,⁴⁸ SNU-216 and SNU-668 cells.⁴⁹

OCUM-2MD3 cells and OUMS-24 (a normal human fibroblast cell line) cells were incubated for 24 h and then treated with the indicated concentrations of test compounds. After 72 h, Hoechst 33342 and propidium iodide were added to the medium, and the cells were incubated in the dark for 20 min. All the cells were then collected on a glass slide, and evaluated under a fluorescence microscope. Cells with fragmented and/or shrunken nuclei were counted as apoptotic cells. The results, expressed as % increase of apoptotic cells versus the DMSO control, are illustrated in Figure 7.

Troglitazone (hPPAR γ full agonist) dose-dependently enhanced apoptosis in the concentration range of 1–100 μ M. Thus, troglitazone induces apoptosis of human scirrhous gastric cancer cell line OCUM-2MD3, as well as the other gastric cancer cell lines, MKV45, SNU-216 and SNU-668,^{48,49} while it exhibited little effect on normal cells. The finding that troglitazone selectively induced apoptosis of human scirrhous gastric cancer cells is important, even though the reason is unknown. One possibility is that different amounts of PPAR γ may be expressed in the cell lines.

The hPPAR γ full agonist **3** also dose-dependently enhanced apoptosis in the concentration range of 1–100 μ M, and it was more potent than troglitazone (10 μ M **3** effectively induced apoptosis of OCUM-2MD3 cells). However, 100 μ M **3** induced apoptosis of both OCUM-2MD3 and OCUM-24 cell lines. Therefore, **3** may show poorer selectivity for human scirrhous gastric cancer cells, as compared to troglitazone.

Finally, we found that hPPAR γ partial agonist (**R**)-**7**j also effectively induced apoptosis of OCUM-2MD3 cells. Treatment of (**R**)-

7j also dose-dependently enhanced apoptosis in the concentration range of 1–100 μ M. The potency was equal to or somewhat greater than that of the hPPAR γ full-agonist troglitazone (based on the activity at 10 μ M). Furthermore, (*R*)-**7j** shows high selectivity for cancer cells, comparable to that of troglitazone. These results indicated that partial activation of PPAR γ might sufficient to induce apoptosis of human scirrhous gastric cancer cells.

In conclusion, we have created the novel α -benzyl phenylpropanoic acid-type hPPAR γ partial agonist (**R**)-**7j**, which shows increased aqueous solubility as a result of replacement of the adamantyl group of **3** with a pyrimidin-2-yl group. It also selectively induced apoptosis of OCUM-2MD3 human scirrhous gastric cancer cells. In vivo pharmacological evaluation of **7j** as anti-scirrhous gastric cancer agent is under way.

4. Experimental section

4.1. Chemistry: general methods

Melting points were determined with a Yanagimoto hot-stage melting point apparatus and are uncorrected. ¹H NMR and ¹³C NMR spectra were recorded on a Varian VNMRS-400 (400 MHz) SC-NMR spectrometer. Proton chemical shifts were referenced to TMS as an internal standard. FAB-MS was carried out with a VG70-SE. Chiral HPLC analyses were performed on an analytical column (CHIRALPAK IA column (4.6 × 150 mm; detection at 254 nm). HPLC analyses were performed on an analytical column (PegasilODS SP100 column (4.6 × 150 mm; flow rate of 1 mL/min; solvent, MeCN/0.1% TFA = 3:1 v/v; detection at 254–296 nm). All optical rotation data was recorded at 25 °C on a Jasco P-2000 polarimeter with a 50 mm cell (concentration reported as g/100 mL).

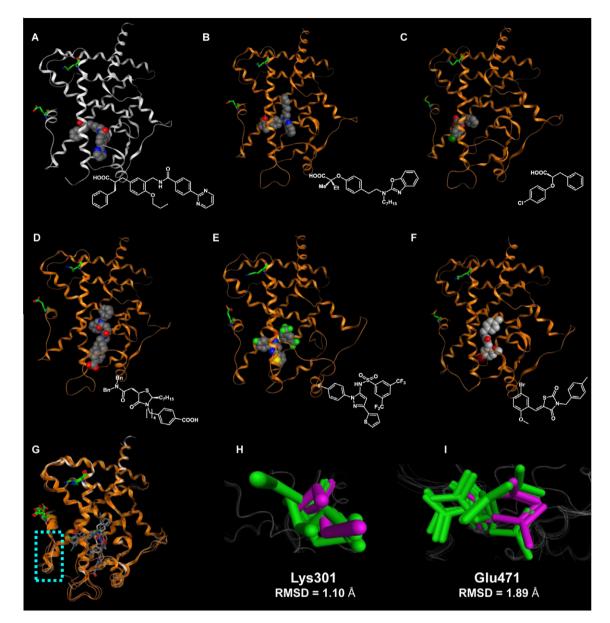


Figure 5. (A–F) Crystal structures of hPPARγ LBD–partial agonist complexes. (A) Crystal structure of hPPARγ LBD–**7** complex (PDB: 3VSO). (B) Crystal structure of hPPARγ LBD–LT127 complex (PDB: 2I42). (C) Crystal structure of hPPARγ LBD–(2S)-2-(4-chlorophenoxy)-3-phenylpropanoic acid complex (PDB: 3CDP). (D) Crystal structure of hPPARγ LBD–N-[1-(4-fluorophenoxy)-3-phenylpropanoic acid complex (PDB: 3CDP). (D) Crystal structure of hPPARγ LBD–N-[1-(4-fluorophenoxy)-3-phenylpropanoic acid complex (PDB: 3CDP). (D) Crystal structure of hPPARγ LBD–N-[1-(4-fluorophenoxy)-3-phenylpropanoic acid complex (PDB: 3CDP). (D) Crystal structure of hPPARγ LBD–N-[1-(4-fluorophenoxy)-3-(2-thenyl)-1*H*-pyrazole-5-yi]-3,5-bis(trifluoromethyl)benzene-sulfonamide complex (PDB: 2GOH). (F) Crystal structure of hPPARγ LBD–Q-16 complex (PDB: 3T03). (G) The superimposed structures of hPPARγ LBD–partial agonist complexes. All ligands are depicted as cylinder models, and Lys301 and Glu473 are highlighted as green cylinder models. The N-terminal AF2 region (H12 heix) is highlighted as a light green dotted box. (H) Zoomed view of the superimposed Lys301 of hPPARγ LBD–partial agonist complexes. All Lys301 residus are highlighted as a green cylinder model. (I) Zoomed view of the superimposed Glu473 of hPPARγ LBD–Partial agonist complexes. All Glu473 residues are also highlighted as a green cylinder model.

4.2. (*R*)-2-(3-([1,1'-Biphenyl]-4-ylcarboxamidomethyl)-4-propoxybenzyl)-3-phenylpropanoic acid ((*R*)-7a)

A mixture of 5-((*R*)-2-benzyl-3-((*S*)-4-benzyl-2-oxo-oxazolidin-3yl)-3-oxopropyl)-2-propoxy-benzaldehyde (**6**) (250 mg, 0.51 mmol), 4-phenylbenzamide (300 mg, 1.54 mmol), triethylsilane (0.25 mL, 1.54 mmol), trifluoroacetic acid (0.12 mL, 1.54 mmol) and 40 mL of dehydrated toluene was refluxed for 48 h under an argon atmosphere. The reaction mixture was poured into water, and the whole was extracted with ethyl acetate. The extract was washed with brine, dried over anhydrous magnesium sulfate and concentrated. The residue was purified by silica gel column chromatography (eluant; *n*-hexane/ethyl acetate = 1:1 v/v) to afford 162 mg (47%) of the intermediate derivative as a pale yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.74 (d, *J* = 8.4 Hz, 2H), 7.58 (t, *J* = 8.6 Hz, 4H), 7.45 (t, *J* = 7.4 Hz, 2H), 7.37 (t, *J* = 7.4 Hz, 1H), 7.29–7.15 (m, 10H), 6.88–6.85 (m, 2H), 6.81 (d, *J* = 8.4 Hz, 1H), 6.61 (t, *J* = 5.2 Hz, 1H), 4.69 (dd, *J* = 14.4, 5.6 Hz, 1H), 4.64–4.55 (m, 2H), 4.41–4.35 (m, 1H), 4.00–3.93 (m, 2H), 3.90 (dd, *J* = 8.8, 2.4 Hz, 1H), 3.74 (t, *J* = 8.2 Hz, 1H), 3.08 (dd, *J* = 13.4, 9.0 Hz, 1H), 2.98 (dd, *J* = 13.4, 9.0 Hz, 1H), 2.91–2.78 (m, 3H), 2.51 (dd, *J* = 13.4, 8.6 Hz, 1H), 1.87–1.78 (m, 2H), 1.04 (t, *J* = 7.4 Hz, 3H); MS (FAB) 667 (M+H)⁺.

This intermediate (157 mg, 0.24 mmol) was dissolved in 16 mL of tetrahydrofuran and 4 mL of water under an argon atmosphere with ice-cooling. To this solution was added 30% aqueous hydrogen peroxide (0.27 mL, 2.40 mmol). Then a solution of lithium hydroxide monohydrate (40 mg, 0.96 mmol) in water (1 mL) was added, and the mixture was stirred for 3.5 h at 0 °C. An aqueous

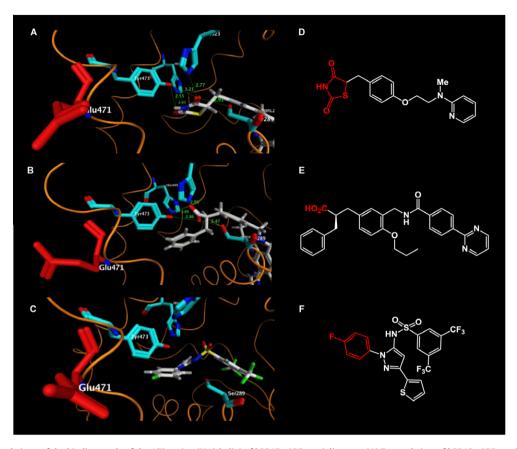


Figure 6. (A–C) Zoomed views of the binding mode of the AF2 region (H12 helix) of hPPARγ LBD–rosiglitazone. (A) Zoomed view of hPPARγ LBD–rosiglitazone (full agonist) complex. (B) Zoomed view of hPPARγ LBD–**7** (partial agonist) complex. (C) Zoomed view of hPPARγ LBD–*N*-[1-(4-fluorophenyl)-3-(2-thenyl)-1*H*-pyrazole-5-yl]-3,5-bis(trifluoromethyl)benzenesulfonamide (partial agonist) complex.

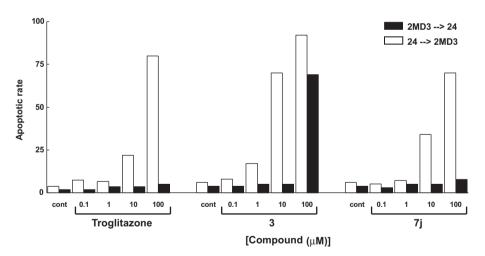


Figure 7. Apoptosis-inducing activity of troglitazone, 3 and (R)-7j on OCUM-2MD3 and OUMS-24 cells.

solution of sodium sulfite (1.00 g/6 mL) was further added and stirring was continued for 15 min. The reaction mixture was acidified with 10% HCl, and then extracted with ethyl acetate. The extract was washed with brine, dried over anhydrous magnesium sulfate and concentrated. The residue was purified by silica gel column chromatography (eluant; *n*-hexane/ethyl acetate = 1:1 v/v) to afford 100 mg (82%) of the title compound as a white amorphous solid. ¹H NMR (400 MHz, CDCl₃) δ 7.80 (d, *J* = 8.4 Hz, 2H), 7.63–7.57 (m, 4H), 7.45 (t, *J* = 7.4 Hz, 2H), 7.37 (t, *J* = 7.4 Hz, 1H), 7.27–7.23 (m, 2H), 7.04 (dd, *J* = 8.4, 2.4 Hz, 1H), 6.81 (t, *J* = 5.8 Hz, 1H), 6.77 (d, *J* = 8.0 Hz, 1H), 4.65–4.55 (m, 2H), 3.95 (t, *J* = 6.4 Hz, 2H), 3.02–2.86 (m, 3H), 2.80–2.73 (m, 2H), 1.88–1.79 (m, 2H), 1.05 (t, J = 7.4 Hz, 3H); ¹³C NMR (300 MHz, CDCl₃) δ 178.85, 167.07, 155.74, 144.11, 140.01, 138.92, 133.25, 130.90, 130.46, 129.18, 128.91, 128.87, 128.42, 127.92, 127.44, 127.16, 126.39, 125.94, 111.12, 69.50, 49.42, 40.31, 37.80, 36.92, 22.74, 10.74; MS (FAB) 508 (M+H)⁺; calcd for C₃₃H₃₄NO₄ 508.2487, found 508.2503; [α]_D –8.7° (*c* 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be >98% by chiral HPLC, using a 4.6 × 250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/*n*-hexane/TFA (1:2:0.03 v/v/v): retention time 17.4 min (the *S* enantio-

mer has a retention time of 15.4 min); HPLC purity was estimated to be 99.8% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with CH₃CN/0.1% TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.3. (*R*)-2-Benzyl-3-(3-((2'-methyl-[1,1'-biphenyl]-4ylcarboxamido)methyl)-4-propoxy-phenyl)propanoic acid ((*R*)-7b)

This compound was prepared by means of a procedure similar to that used for **7a**. ¹H NMR (400 MHz, CDCl₃) δ 7.78 (d, J = 8.4 Hz, 2H), 7.35 (d, J = 8.4 Hz, 2H), 7.28–7.22 (m, 5H), 7.20– 7.17 (m, 5H), 7.04 (dd, J = 8.4, 2.2 Hz, 1H), 6.81–6.76 (m, 2H), 4.65-4.56 (m, 2H), 3.96 (t, *J* = 6.4 Hz, 2H), 3.02-2.87 (m, 3H), 2.82-2.73 (m, 2H), 2.24 (s, 3H), 1.88-1.80 (m, 2H), 1.06 (t, I = 7.4 Hz, 3H); ¹³C NMR (300 MHz, CDCl₃) δ 178.82, 167.16, 155.77, 145.13, 140.85, 138.87, 135.21, 133.01, 130.82, 130.53, 130.45, 129.57, 129.38, 129.20, 128.92, 128.45, 127.75, 126.76, 126.44, 125.99, 125.88, 111.12, 69.51, 49.35, 40.31, 37.75, 36.94, 22.75, 20.40, 10.77; MS (FAB) 522 (M+H)⁺; calcd for C₃₄H₃₆NO₄ 522.2644, found 522.2660; $[\alpha]_{\rm D}$ –9.3° (*c* 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be 97% by chiral HPLC using a 4.6×250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/*n*-hexane/TFA (1:2:0.03 v/v/v): retention time 14.5 min (the S enantiomer has a retention time of 13.2 min); HPLC purity was estimated to be 99.4% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with CH₃CN/ 0.1% TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.4. (*R*)-2-Benzyl-3-(3-((2'-fluoro-[1,1'-biphenyl]-4ylcarboxamido)methyl)-4-propoxy-phenyl)propanoic acid ((*R*)-7c)

This compound was prepared by means of a procedure similar to that used for **7a**.

¹H NMR (400 MHz, CDCl₃) δ 7.80 (d, *J* = 8.4 Hz, 2H), 7.59 (dd, I = 8.4, 1.6 Hz, 2H), 7.43 (dt, I = 7.6, 1.8 Hz, 1H), 7.37-7.32 (m, 1H), 7.28-7.14 (m, 8H), 6.80-6.76 (m, 2H), 4.65-4.56 (m, 2H), 3.95 (t, J = 6.6 Hz, 2H), 3.01-2.87 (m, 3H), 2.82-2.73 (m, 2H), 1.88–1.79 (m, 2H), 1.06 (t, I = 7.4 Hz, 3H); ¹³C NMR (300 MHz, CDCl₃) δ 178.97, 166.96, 161.33, 158.04, 155.76, 138.86, 138.82, 138.81, 133.67, 130.82, 130.65, 130.61, 130.46, 129.67, 129.57, 129.21, 129.16, 129.12, 128.92, 128.45, 128.09, 127.91, 127.07, 126.43, 125.92, 124.52, 124.47, 116.37, 116.07, 111.12, 69.50, 49.37, 40.32, 37.77, 36.92, 22.75, 10.76; MS (FAB) 526 (M+H)+; calcd for $C_{33}H_{33}FNO_4$ 526.2394, found 526.2399; $[\alpha]_D - 8.5^\circ$ (c 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be 98% by chiral HPLC using a $4.6 \times 250 \text{ mm}$ CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (1:2:0.03 v/v/v): retention time 15.7 min (the S enantiomer has a retention time of 14.6 min); HPLC purity was estimated to be 99.6% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with $CH_3CN/0.1\%$ TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.5. (*R*)-2-Benzyl-3-(3-((4'-fluoro-[1,1'-biphenyl]-4-ylcarboxamido) methyl)-4-propoxy-phenyl)propanoic acid ((*R*)-7e)

This compound was prepared by means of a procedure similar to that used for **7a**.

¹H NMR (400 MHz, CDCl₃) δ 7.80 (d, *J* = 8.0 Hz, 2H), 7.57–7.52 (m, 4H), 7.27–7.23 (m, 2H), 7.20–7.11 (m, 6H), 7.04 (dd, *J* = 8.2, 2.2 Hz, 1H), 6.80-6.76 (m, 2H), 4.65-4.56 (m, 2H), 3.95 (t, *J* = 6.4 Hz, 2H), 3.01–2.87 (m, 3H), 2.81–2.73 (m, 2H), 1.88–1.79 (m, 2H), 1.05 (t, J = 7.4 Hz, 3H); ¹³C NMR (300 MHz, CDCl₃) δ 179.13, 166.92, 164.42, 161.15, 155.57, 143.07, 138.82, 136.15, 133.27, 130.81, 130.45, 129.20, 128.90, 128.84, 128.74, 128.45, 127.52, 127.01, 126.44, 125.91, 115.95, 115.67, 111.12, 69.50, 49.35, 40.31, 37.73, 36.89, 22.74, 10.77; ¹³C NMR (300 MHz, CDCl₃) δ 179.13, 166.92, 164.42, 161.15, 155.57, 143.07, 138.82, 136.15, 133.27, 130.81, 130.45, 129.20, 128.90, 128.84, 128.74, 128.45, 127.52, 127.01, 126.44, 125.91, 115.95, 115.67, 111.12, 69.50, 49.35, 40.31, 37.73, 36.89, 22.74, 10.77; MS (FAB) 526 (M+H)+; calcd for $C_{33}H_{33}FNO_4$ 526.2394, found 526.2399; $[\alpha]_D - 9.4^\circ$ (c 0.25. CH₃CN). The enantiomeric excess (ee) of this compound was determined to be 97% by chiral HPLC using a 4.6×250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (1:2:0.03 v/v/v): retention time 19.8 min (the S enantiomer has a retention time of 17.4 min); HPLC purity was estimated to be 99.9% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with $CH_3CN/0.1\%$ TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.6. 5-((*R*)-2-Benzyl-3-((S)-4-benzyl-2-oxooxazolidin-3-yl)-3-oxopropyl)-2-propoxybenz-aldehyde oxime (8)

A mixture of 6 (3.31 g, 6.82 mmol), hydroxylamine hydrochloride (568 mg, 8.18 mmol), dehydrated pyridine (0.82 mL, 10.23 mmol) and 40 mL of ethanol was stirred overnight at room temperature. Excess ethanol was evaporated, and the residue was diluted with ethyl acetate, then washed with 1 mol/L HCl, a saturated aqueous solution of sodium hydrocarbonate and brine. The organic layer was separated, dried over anhydrous magnesium sulfate and concentrated to afford 3.18 g of the title compound as a colorless oil, which was used in the next step without further purification. ¹H NMR (400 MHz, CDCl₃) δ 8.50 (s, 1H), 7.65 (d, *I* = 1.6 Hz, 1H), 7.28-7.15 (m, 9H), 7.00 (dd, J=7.6, 2.0 Hz, 2H), 6.81 (d, *J* = 8.4 Hz, 1H), 4.59–4.52 (m, 1H), 4.45–4.41 (m, 1H), 3.95–3.86 (m, 3H), 3.79 (t, *J* = 8.4 Hz, 1H), 3.08 (dd, *J* = 13.4. 8.2 Hz, 1H), 3.05-2.96 (m, 2H), 2.82 (dd, *J* = 13.4, 6.2 Hz, 1H), 2.78 (dd, *J* = 13.6, 6.4 Hz, 1H), 2.49 (dd, *J* = 13.2, 9.2 Hz, 1H), 1.84–1.75 (m, 2H), 1.02 (t, *J* = 7.4 Hz, 3H); MS (FAB) 501 (M+H)⁺.

4.7. (*S*)-3-((*R*)-2-(3-(Aminomethyl)-4-propoxybenzyl)-3phenylpropanoyl)-4-benzyl-oxazolidin-2-one hydrochloride (9)

A mixture of **8** (3.18 g, 6.35 mmol), 10% palladium on carbon (1.60 g), concentrated HCl (8 mL) and 50 mL of ethanol was subjected to catalytic hydrogenation at an initial hydrogen pressure of 400–500 kPa. After completion of the reaction, the catalyst was removed by filtration and the filtrate was concentrated to afford 3.30 g of the title compound as a pale yellow solid, which was used in the next step without further purification. MS (FAB) 487 (M+H)⁺.

4.8. (*R*)-2-Benzyl-3-(4-propoxy-3-((4-(pyridin-2-yl)benzamido)methyl)phenyl)propanoic acid ((*R*)-7g)

Under an argon atmosphere, diethyl cyanophosphonate (DEPC) (0.09 mL, 0.61 mmol) and triethylamine (0.18 mL, 1.30 mmol) were added to a solution of compound **9** (246 mg, 0.47 mmol) and 4-(2-pyridyl)benzoic acid (94 mg, 0.47 mmol) in dehydrated *N*,*N*-dimethylformamide (10 mL) at 0 °C. The reaction mixture

was left to stand overnight, then poured into a saturated aqueous solution of sodium hydrocarbonate, and the whole was extracted with ethyl acetate. The extract was washed with brine, dried over anhydrous magnesium sulfate and concentrated. The residue was purified by silica gel column chromatography (eluant; *n*-hexane/ ethyl acetate = 1:2 v/v) to afford 170 mg (54%) of the intermediate compound as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 8.71 (d, *J* = 5.6 Hz, 1H), 8.00 (d, *J* = 8.4 Hz, 2H), 7.80–7.73 (m, 4H), 7.30–7.15 (m, 11H), 6.90–6.87 (m, 2H), 6.82 (d, *J* = 8.4 Hz, 1H), 6.66 (t, *J* = 5.4 Hz, 1H), 4.69 (dd, *J* = 14.6, 5.8 Hz, 1H), 4.64–4.55 (m, 2H), 4.41–4.35 (m, 1H), 4.00–3.92 (m, 2H), 3.90 (dd, *J* = 8.8, 2.4 Hz, 1H), 3.74 (t, *J* = 8.4 Hz, 1H), 3.08 (dd, *J* = 13.6, 8.8 Hz, 1H), 2.98 (dd, *J* = 13.2, 8.8 Hz, 1H), 2.92–2.79 (m, 3H), 2.51 (dd, *J* = 13.6, 8.8 Hz, 1H), 1.87–1.78 (m, 2H), 1.04 (t, *J* = 7.4 Hz, 3H); MS (FAB) 668 (M+H)⁺.

This compound was treated with 30% hydrogen peroxide, as described for **7a**.

¹H NMR (400 MHz, DMSO- d_6) δ 8.91 (t, J = 5.8 Hz, 1H), 8.74(d, I = 4.4 Hz, 1H), 8.19-8.13 (m, 3H), 8.09-8.03 (m, 3H), 7.53 (t, J = 6.0 Hz, 1H), 7.19–7.16 (m, 2H), 7.12–7.09 (m, 3H), 7.03–7.01 (m, 2H), 6.87 (d, J = 8.8 Hz, 1H), 4.51-4.40 (m, 2H), 3.93 (t, *I* = 6.4 Hz, 2H), 2.80–2.70 (m, 3H), 2.67–2.57 (m, 2H), 1.77–1.69 (m, 2H), 0.98 (t, I = 7.2 Hz, 3H); ¹³C NMR (300 MHz, CDCl₃) δ 178.72, 166.91, 156.23, 155.71, 149.54, 141.71, 139.04, 137.22, 134.84, 131.02, 130.36, 129.21, 128.93, 128.40, 127.47, 127.11, 126.35, 125.79, 122.82, 121.19, 111.11, 69.48, 49.44, 40.32, 37.86, 36.94, 22.66, 10.75; MS (FAB) 509 (M+H)+; calcd for $C_{32}H_{33}N_2O_4$ 509.2440, found 509.2421; $[\alpha]_D$ -7.3° (c 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be 97% by chiral HPLC using a $4.6 \times 250 \text{ mm}$ CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (3:2:0.03 v/v/v): retention time 11.8 min (the S enantiomer has a retention time of 10.8 min); HPLC purity was estimated to be 97.2% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with $CH_3CN/0.1\%$ TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.9. (*R*)-2-Benzyl-3-(3-((3'-fluoro-[1,1'-biphenyl]-4-ylcarboxamido) methyl)-4-propoxy-phenyl)propanoic acid ((*R*)-7d)

This compound was prepared by means of a procedure similar to that used for **7g** and **7a**.

¹H NMR (400 MHz, DMSO- d_6) δ 8.87 (t, J = 5.6 Hz, 1H), 8.00 (d, *I* = 8.4, 2H), 7.82 (d, *I* = 8.8 Hz, 2H), 7.61–7.58 (m, 2H), 7.55–7.49 (m, 1H), 7.23 (dt, *J* = 8.8, 1.6 Hz, 1H), 7.19–7.15 (m, 2H), 7.12– 7.09 (m, 3H), 7.03-7.00 (m, 2H), 6.86 (d, J = 8.4 Hz, 1H), 4.50-4.40 (m, 2H), 3.92 (t, J = 6.4 Hz, 2H), 2.79–2.71 (m, 3H), 2.69–2.56 (m, 2H), 1.77–1.69 (m, 2H), 0.99 (t, J = 7.4 Hz, 3H); ¹³C NMR (300 MHz, CDCl₃) δ 179.11, 166.86, 164.78, 161.52, 155.75, 142.74, 142.71, 142.29, 142.19, 138.85, 138.83, 130.84, 130.46, 130.45, 130.44, 130.34, 129.21, 128.90, 128.44, 127.57, 127.15, 126.43, 125.86, 122.83, 122.79, 114.88, 114.60, 114.21, 113.92, 111.13, 69.50, 49.40, 40.33, 37.77, 36.89, 22.74, 10.76; MS (FAB) 526 (M+H)⁺; calcd for C₃₃H₃₃FNO₄ 526.2394, found 526.2399; $[\alpha]_{\rm D}$ –7.6° (*c* 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be 96% by chiral HPLC using a 4.6×250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (1:2:0.03 v/v/v): retention time 18.7 min (the S enantiomer has a retention time of 16.2 min). The purity was estimated to be 99.7% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with $CH_3CN/0.1\%$ TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.10. (*R*)-2-Benzyl-3-(4-propoxy-3-((4-(thiophen-2-yl)benzamido)methyl)phenyl)propanoic acid ((*R*)-7f)

This compound was prepared by means of a procedure similar to that used for **7g** and **7a**.

¹H NMR (400 MHz, DMSO- d_6) δ 12.05 (s, 1H), 8.81 (t, J = 5.6 Hz, 1H), 7.94 (t, J = 8.8 Hz, 2H), 7.76 (d, J = 8.4 Hz, 2H), 7.64 (dd, J = 3.6, 1.2 Hz, 1H), 7.62 (dd, J = 5.2, 1.2 Hz, 1H), 7.19-7.15 (m, 3H), 7.12-7.09 (m, 3H), 7.02-6.99 (m, 2H), 6.86 (d, J = 8.4 Hz, 1H), 4.49–4.38 (m, 2H), 3.92 (t, J = 6.2 Hz, 2H), 2.81– 2.71 (m, 3H), 2.68-2.56 (m, 2H), 1.77-1.68 (m, 2H), 0.98 (t, I = 7.4 Hz, 3H); ¹³C NMR (300 MHz, DMSO- d_6) δ 175.64, 165.75, 154.51, 142.39, 139.42, 136.31, 133.18, 130.55, 128.79, 128.24, 128.23, 127.81, 126.83, 126.75, 126.13, 125.13, 124.92, 111.18, 69.06, 48.85, 37.74, 37.26, 36.79, 22.24, 10.64; MS (FAB) 514 $(M+H)^+$; calcd for C₃₁H₃₂NO₄ S 514.2051, found 514.2070; $[\alpha]_D$ -8.6° (c 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be 96% by chiral HPLC using a 4.6×250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (1:2:0.03 v/v/v): retention time 17.6 min (the S enantiomer has a retention time of 15.6 min); HPLC purity was estimated to be 99.2% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with CH₃CN/0.1% TFA = 3:1 v/v as the eluant and detection at 285 nm.

4.11. (*R*)-2-Benzyl-3-(4-propoxy-3-((4-(pyridin-3-yl)benzamido) methyl)phenyl)propanoic acid ((*R*)-7h)

This compound was prepared by means of a procedure similar to that used for **7g** and **7a**.

¹H NMR (400 MHz, DMSO- d_6) δ 12.05 (s, 1H), 8.96 (d, J = 1.6 Hz, 1H), 8.88(t, J = 5.6 Hz, 1H), 8.60 (dd, J = 4.8, 1.6 Hz, 1H), 8.15 (dt, *J* = 8.8, 1.6 Hz, 1H), 8.03 (d, *J* = 8.4 Hz, 2H), 7.86 (d, *J* = 8.4 Hz, 2H), 7.51 (dd, / = 8.0, 4.8 Hz, 1H), 7.19-7.09 (m, 5H), 7.03-7.00 (m, 2H), 6.87 (d, / = 8.0 Hz, 1H), 4.51-4.40 (m, 2H), 3.93 (t, / = 6.4 Hz, 2H), 2.80-2.70 (m, 3H), 2.67-2.57 (m, 2H), 1.78-1.68 (m, 2H), 0.99 (t, I = 7.4 Hz, 3H); ¹³C NMR (300 MHz, DMSO- d_6) δ 175.65, 165.84, 154.50, 149.60, 147.84, 139.69, 139.45, 134.38, 133.95, 130.58, 128.77, 128.21, 128.14, 128.04, 127.77, 126.85, 126.71, 126.11, 124.01, 111.18, 111.11, 69.06, 48.87, 37.76, 37.24, 36.80, 22.23, 10.63; MS (FAB) 509 $(M+H)^+$; calcd for $C_{32}H_{33}N_2O_4$ 509.2440, found 509.2421; $[\alpha]_D = -8.3^\circ$ (*c* 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be >98% by chiral HPLC using a 4.6×250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/*n*-hexane/TFA (1:2:0.03 v/v/v): retention time 30.7 min (the S enantiomer was not detectable); HPLC purity was estimated to be 99.6% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6 \times 250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with CH₃CN/0.1% TFA = 3:1 v/v as the eluant and detection at 290 nm.

4.12. (*R*)-2-Benzyl-3-(4-propoxy-3-((4-(pyridin-4-yl)benzamido)methyl)phenyl)propanoic acid ((*R*)-7i)

This compound was prepared by means of a procedure similar to that used for **7g** and **7a**.

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.99 (t, *J* = 5.2 Hz, 1H), 8.88 (d, *J* = 4.8 Hz, 2H), 8.23 (d, *J* = 5.2 Hz, 2H), 8.11–8.05 (m, 4H), 7.19–7.09 (m, 5H), 7.03–7.00 (m, 2H), 6.87 (d, *J* = 8.0 Hz, 1H), 4.51–

4.41 (m, 2H), 3.93 (t, J = 6.2 Hz, 2H), 2.80–2.73 (m, 3H), 2.67–2.58 (m, 2H), 1.77–1.69 (m, 2H), 0.98 (t, J = 7.4 Hz, 3H); ¹³C NMR $(300 \text{ MHz}, \text{DMSO-}d_6) \delta$ 175.61, 165.51, 154.56, 139.40, 137.78, 136.25, 130.54, 128.78, 128.37, 128.24, 128.14, 127.88, 127.75, 126.54, 126.16, 111.24, 69.09, 48.82, 37.89, 37.26, 36.80, 22.24, 10.65; MS (FAB) 509 $(M+H)^+$; calcd for $C_{32}H_{33}N_2O_4$ 509.2440, found 509.2426; $[\alpha]_{D}$ –4.3° (*c* 0.25, DMSO). The enantiomeric excess (ee) of this compound was determined to be >98% by chiral HPLC using a 4.6×250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/*n*-hexane/TFA (3:2:0.03 v/v/v): retention time 15.9 min (the S enantiomer was not visible); HPLC purity was estimated to be 99.6% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column $(4.6 \times 250 \text{ mm}, \text{ Senshu Chemical, Japan})$ fitted on a Shimadzu HPLC system, with $CH_3CN/0.1\%$ TFA = 3:1 v/v as the eluant and detection at 290 nm.

4.13. (*R*)-2-Benzyl-3-(4-propoxy-3-((4-(pyrimidin-2-yl)benzamido)methyl)phenyl)propanoic acid ((*R*)-7j)

This compound was prepared by means of a procedure similar to that used for **7g** and **7a**.

¹H NMR (400 MHz, DMSO- d_6) δ 12.06 (s, 1H), 8.94 (d, J = 4.8 Hz, 2H), 8.91 (t, J = 5.8 Hz, 1H), 8.47 (d, J = 8.8 Hz, 2H), 8.04 (d, J = 8.4 Hz, 2H), 7.49 (t, J = 4.8 Hz, 1H), 7.19–7.16 (m, 2H), 7.12– 7.09 (m, 3H), 7.02–7.01 (m, 2H), 6.87 (d, J = 9.2 Hz, 1H), 4.50– 4.40 (m, 2H), 3.93 (t, J = 6.4 Hz, 2H), 2.82–2.71 (m, 3H), 2.70–2.58 (m, 2H), 1.77–1.69 (m, 2H), 0.98 (t, J = 7.4 Hz, 3H); ¹³C NMR (300 MHz, DMSO-d₆) & 175.64, 165.93, 162.63, 157.92, 154.52, 139.55, 139.43, 136.45, 130.59, 128.78, 128.21, 128.07, 127.89, 127.82, 127.57, 126.62, 126.11, 120.41, 111.18, 69.07, 48.86, 37.84, 37.28, 36.78, 22.22, 10.62; MS (FAB) 510 (M+H)⁺; calcd for $C_{31}H_{32}N_3O_4$ 510.2393, found 510.2409; $[\alpha]_D - 3.9^{\circ}$ (*c* 0.25, CHCl₃). The enantiomeric excess (ee) of this compound was determined to be >98% by chiral HPLC using a 4.6 \times 250 mm CHIRALPAK IA column (Diacel Chemical Industries, Ltd.), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (1:2:0.03 v/v/v): retention time 17.5 min (the S enantiomer was not visible): HPLC purity was estimated to be 98.4% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with CH₃CN/0.1% TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.14. (*R*)-2-Benzyl-3-(4-propoxy-3-((4-(pyrazin-2-yl)benzamido)methyl)phenyl)propanoic acid ((*R*)-7k)

This compound was prepared by means of a procedure similar to that used for **7g** and **7a**.

¹H NMR (400 MHz, DMSO- d_6) δ 12.08 (s, 1H), 9.34 (d, J = 1.2 Hz, 1H), 8.92 (t, J = 5.8 Hz, 1H), 8.75 (dd, J = 4.0, 1.6 Hz, 1H), 8.65 (d, J = 2.4 Hz, 1H), 8.25 (d, J = 8.8 Hz, 2H), 8.05 (d, J = 8.4 Hz, 2H), 7.19–7.16 (m, 2H), 7.12–7.09 (m, 3H), 7.03–7.01 (m, 2H), 6.87 (d, J = 8.8 Hz, 1H), 4.51-4.41 (m, 2H), 3.93 (t, J = 6.4 Hz, 2H), 2.80–2.71 (m, 3H), 2.69–2.57 (m, 2H), 1.77–1.69 (m, 2H), 0.98 (t, J = 7.4 Hz, 3H); ¹³C NMR (300 MHz, CDCl₃) δ 178.61, 166.61, 155.78, 151.75, 144.36, 143.34, 142.18, 138.91, 138.86, 135.81, 130.97, 130.49, 129.28, 128.92, 128.46, 127.70, 127.04, 126.44, 125.79, 111.19, 69.55, 49.37, 40.42, 37.87, 36.95, 22.76, 10.77; MS (FAB) 510 (M+H)⁺; calcd for $C_{31}H_{32}N_3O_4$ 510.2393, found 510.2409; $[\alpha]_D$ -8.3° (c 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be >98% by chiral HPLC using a $4.6 \times 250 \text{ mm}$ CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (1:2:0.03 v/v/v): retention time 25.6 min (the S enantiomer has a retention time of 20.4 min); HPLC purity was estimated to be

98.1% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with CH₃CN/0.1% TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.15. (*R*)-2-Benzyl-3-(4-propoxy-3-((4-(pyrimidin-5-yl)benzamido)methyl)phenyl)propanoic acid ((*R*)-7l)

This compound was prepared by means of a procedure similar to that used for **7g** and **7a**.

¹H NMR (400 MHz, DMSO- d_6) δ 12.06 (s, 1H), 9.21 (s, 3H), 8.92 (t, J = 5.8 Hz, 1H), 8.06 (d, J = 8.8 Hz, 2H), 7.95 (d, J = 8.4 Hz, 2H), 7.20–7.16 (m, 2H), 7.13–7.09 (m, 3H), 7.03–7.00 (m, 2H), 6.87 (d, J = 8.0 Hz, 1H), 4.51–4.41 (m, 2H), 3.93 (t, *J* = 6.4 Hz, 2H), 2.80–2.71 (m, 3H), 2.67–2.57 (m, 2H), 1.78–1.69 (m, 2H), 0.99 (t, I = 7.2 Hz, 3H); ¹³C NMR (300 MHz, DMSO- d_6) δ 175.61, 165.69, 157.71, 154.98, 154.52, 139.41, 136.41, 134.61, 132.44, 130.52, 128.76, 128.35, 128.22, 128.06, 127.82, 126.92, 126.64, 126.13, 111.19, 69.09, 48.85, 37.80, 37.25, 36.80, 22.22, 10.62; MS (FAB) 510 (M+H)⁺; calcd for $C_{31}H_{32}N_3O_4$ 510.2393, found 510.2409; $[\alpha]_D$ -7.6° (*c* 0.25, CH₃CN). The enantiomeric excess (ee) of this compound was determined to be 97% by chiral HPLC using a $4.6 \times 250 \text{ mm}$ CHIRALPAK IA column (Diacel Chemical Industries, Ltd), eluting at a flow rate of 0.5 mL/min with 2-propanol/n-hexane/TFA (1:2:0.03 v/v/v): retention time 32.4 min (the S enantiomer has a retention time of 29.9 min); HPLC purity was estimated to be 99.3% by means of reversed-phase HPLC, using a Pegasil ODS sp100 column (4.6×250 mm, Senshu Chemical, Japan) fitted on a Shimadzu HPLC system, with $CH_3CN/0.1\%$ TFA = 3:1 v/v as the eluant and detection at 254 nm.

4.16. Thermodynamic aqueous solubility

Thermodynamic solubility determination was based on the method of Avdeef and Testa. Briefly, about 1 mg of compound was ground in an agate mortar and taken up in 1.0 mL of an equal volume of a mixture of phosphate buffer (pH 7.2–7.4). The suspension was shaken for 48 h at 25 °C. An aliquot was filtered through a Minisart RC 15 (0.45 μ m). The filtrate was diluted in CH₃CN and injected into an HPLC with UV detection at 254 nm. The concentration of the sample solution was calculated using a previously determined calibration curve, corrected for the dilution factor of the sample.

4.17. Assays with GAL4-PPAR chimera receptors

Receptor expression plasmid: An established chimeric receptor system was utilized to allow comparison of the relative transcriptional activity of the receptor subtypes. The mammalian expression vectors pSG5-GAL4-hPPAR α , pSG5-GAL4-hPPAR γ and pSG5-GAL4-hPPAR δ , which express the ligand binding domains (LBDs) of human PPAR α (amino acids 167–468), PPAR γ 1 (amino acids 176–477) and PPAR δ (amino acids 139–441) each fused to the yeast transcription factor GAL4 DNA binding domain (amino acid 1–147), were used. Reporter plasmid: MH100x4-tk-luc⁴⁵ was used as the reporter plasmid.

4.18. Transient transfection assays

The African green monkey kidney cell line CV-1 was used for the transfection assay. CV-1 cells were seeded in 24-well plates at 0.5×10^5 cells per well and cultured for 24 h. Transfection mixtures for chimeric receptors contained 30 ng of receptor expression plasmid, 250 ng of the reporter plasmid, 350 ng of pCMX- β -galactosidase (β GAL) expression plasmid as a control for transfection efficiency, 120 ng of pGEM4 carrier plasmid and 2 μ L of a lipofection reagent (Lipofectamine 2000, Invitrogen). The mixtures were added to cells and incubated for 5 h according to the manufacturer's instructions. After the transfection, cells were incubated for an additional 40 h in the presence of new compounds or reference compounds. Cell lysates were prepared with a lysis buffer (Passive Lysis Buffer, Promega) and used in the luciferase and β GAL assays. The luciferase and β -GAL activity were measured according to the methods of Umesono et al. with slight modifications. A substrate reagent kit (Picagene, Toyo Ink) was used for the luciferase assay.

4.19. Calculation of relative PPAR transactivation activities

Relative PPAR transcriptional activity with respect to maximal activity was calculated based on the luciferase activity of cells treated with a positive control [10^{-6} M GW-590735 (PPAR α selective agonist) for hPPAR α , 10^{-5} M rosiglitazone maleate (PPAR γ selective agonist) for hPPAR γ and 10^{-7} M GW-501516 (PPAR δ selective agonist) for hPPAR δ], as the maximal activity, and the luciferase activity of cells treated with 0.1% DMSO, taken as the minimum activity.

4.20. Calculation of EC₅₀ values

 EC_{50} values, defined as the concentration of test compounds and positive controls (GW-590735, rosiglitazone maleate and GW-501516 required to produce 50% of maximal reporter activity, were calculated with Prism software (Graphpad Software).

For comparison, GW-590735 activated human GAL4 PPAR α with an EC₅₀ = 0.010 μ M, rosiglitazone activated human GAL4 PPAR γ with an EC₅₀ = 0.1 μ M and GW-501516 activated human GAL4 PPAR δ with an EC₅₀ = 0.0013 μ M.

4.21. Adipogenesis assay

Mouse adipogenic fibroblast cells, 3T3-L1, were cultured in DMEM with 10% FBS. After the cells had become confluent, they were pretreated with 1 μ M dexamethasone (Dex), 0.5 mM 3-isobu-tyl-1-methylxanthine (IBMX), and 5 μ g/mL insulin to initiate adipogenesis for 2 days. Various ligands were added to the cells. After 6 days, the cells were stained with Oil Red O, and the degree of adipogenesis was quantitatively measured at OD550. The error bars represent the standard deviations (SD).

4.22. Protein expression, purification, and crystallization

The recombinant ligand-binding domain (LBD) of human PPAR γ (residues 203–477) was expressed as an *N*-terminal His-tagged protein in *E. coli* BL21 (DE3), using the pET28a vector (Novagen). The protein was purified by means of several steps of column chromatography. Apo PPAR γ LBD crystals were soaked in ligand solution for two or three weeks.

4.23. Diffraction data collection

The diffraction data for hPPAR α LBD complexed with (*R*)-**7**j were collected on BL38B1 at SPring-8 (Harima, Japan) and were processed using HKL-2000. Diffraction data were processed with HKL2000.

4.24. Structure analysis and refinement

The structure of hPPAR α LBD complex with (**R**)-**7j** was solved by the molecular-replacement method with the program *CNS* using

previously reported structures as probes. The correctly positioned molecules were refined with *CNS* and *O*. The initial atomic model of (*R*)-7j was built using *MOE* (Ryoka Systems Inc.) and topology and parameter files for the refinement were generated by the *HIC-Up* server.

The structure of hPPAR γ LBD was solved by the molecular replacement method using Molrep. Apo PPAR γ LBD structure (PDB code 1PRG) was used as a search model and refined with Refmac5. The crystallographic data and data collection statistics of all the crystals are provided in Table 1.

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