

One-Pot Four Component Synthesis of 3-Amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine Derivatives Mediated by [DBUH][OAc]¹

G. Jeevan Raghavendra^{a*} and V. Siddaiah^a

^a Department of Organic Chemistry, Foods, Drug and Water, College of Science and Technology, Andhra University, Visakhapatnam, Andhra Pradesh, 530003 India

*e-mail: jeevan.gutala@gmail.com

Received July 5, 2018

Abstract—One-pot, four component, green, and efficient synthesis of 3-amino-1-(5-nitro-1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine derivatives by reaction of phthalic anhydride with hydrazine hydrate, 5-nitro-1*H*-indole-3-carboxaldehyde–indole-3-carboxaldehydes and malononitrile–ethyl cyanoacetate in the presence of [DBUH][OAc] at 60–65°C is developed. The method is characterized by short reaction time, high yields, and purity of products formed.

Keywords: [DBUH][OAc], green synthesis, one-pot reaction

DOI: 10.1134/S1070363218090219

INTRODUCTION

Phthalazines are heterocycles that possess multiple biological activities such as antimicrobial [1], anticonvulsant [2], antifungal [3], anticancer [4], and anti-inflammatory [5]. Therefore, a number of methods of synthesis of phthalazine derivatives [6, 7] has been developed. Recently, synthesis of 1-aryl-1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones by one-pot, three component condensation of phthalhydrazide with malononitrile–ethyl cyanoacetate and benzaldehydes was reported [8–10]. 1-Aryl-1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones were synthesized in a four component process of phthalic anhydride with hydrazine, malononitrile/ethyl cyanoacetate and benzaldehydes using basic ionic liquids, such as 1,8-diazabicyclo[5.4.0]undec-7-en-8-ium acetate [11], pyrrolidinium acetate [11] and TEA as a catalyst under ultrasound-sonication [12].

Ionic liquids (ILs) have attracted close attention due to their distinctive properties such as negligible vapour pressure, high thermal stability, wide liquid temperature range, easy recyclability, excellent chemical stability, and strong solvent power for a wide

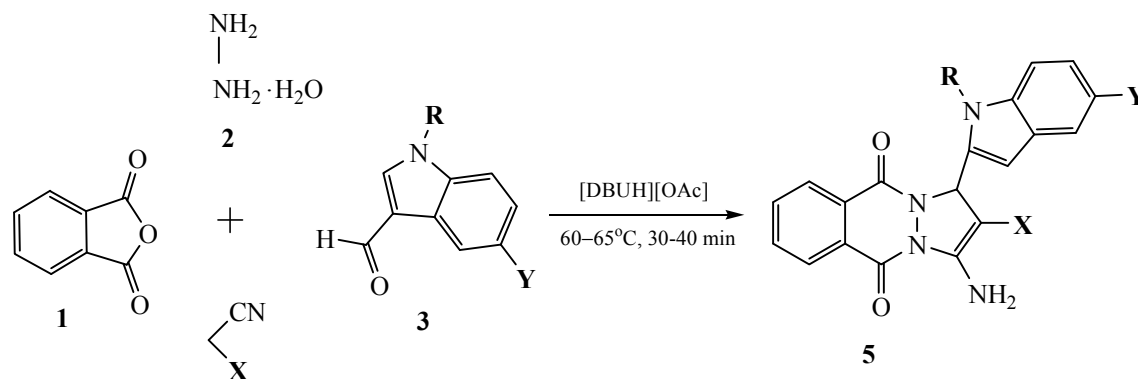
range of organic and inorganic compounds. By modification of cations and/or anions, the properties of ILs can be turned in many ways for various applications in organic synthesis [13–17]. DBU is a strong organic base and has been extensively applied in the base-induced reactions with excellent catalytic activity. However, separation of DBU from a reaction mixture is generally difficult. DBU-based ILs (DBU-ILs) overcome this drawback and exhibit the basicity similar to that of DBU accompanied by the general features of ILs [18].

Based on the above, we report herein synthesis of 3-amino-1-(5-nitro/chloro-1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine derivatives by reaction of phthalic anhydride with hydrazine hydrate, indole-3-carboxaldehydes and malononitrile/ethyl cyanoacetate in the presence of [DBUH][OAc] at 60–65°C.

RESULTS AND DISCUSSION

Initial optimization of the one-pot four component reaction started with the reaction of phthalic anhydride **1** with hydrazine hydrate **2** in-situ and formation of phthalhydrazide as an intermediate in the presence of the ionic liquid. Addition of 5-nitroindole-3-carboxaldehyde **3a** and malononitrile **4a** to the above

¹ The text was submitted by the authors in English.

Scheme 1. Synthesis of 3-amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine derivatives **5a–5h**.

Y = NO₂ (**3a**, **5a**, **5e**), H (**3b–3d**, **5b–5d**, **5f–5h**), R = H (**3a**, **3b**, **5a**, **5b**, **5e**, **5f**), CH₃ (**3c**, **5c**, **5g**), C₂H₅ (**3d**, **5d**, **5h**), X = CN (**5a–5d**), COOEt (**5e**, **5f–5h**).

reaction mixture in the presence of different ionic liquids ([DBUH][OAc], [bmim][Br] or [bmim][OH]) at room temperature led to formation of 3-amino-1-(5-nitro-1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile **5a**. This process was used in the following studies as a simple model method (see Table 1).

Following optimization of the process was carried out by using different ratio of the reagents in the range of temperature 20–80°C over various reaction time (Table 1). This approach allowed to work out the most favorable conditions of the synthesis (Scheme 1) that were the one-pot reaction of compound **1** (1 equiv.) with compounds **2** (1 equiv.), **3a** (1 equiv.) and **4a** (1 equiv.) in the presence of [DBUH][OAc] as a medium (5 equiv.), the reaction time 30 min and temperature range 60–65°C. These led to the highest yield (85%) of the product 3-amino-1-(5-nitro-1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile **5a** (see Table 1).

The structure of **5a** has been elucidated from ¹H and ¹³C NMR, IR, and Mass spectroscopic data.

This newly developed synthetic method was successfully extended to malononitrile, indole-3-carboxaldehyde **3b** and its *N*-methyl and *N*-ethyl derivatives **3c**, **3d** (Table 2).

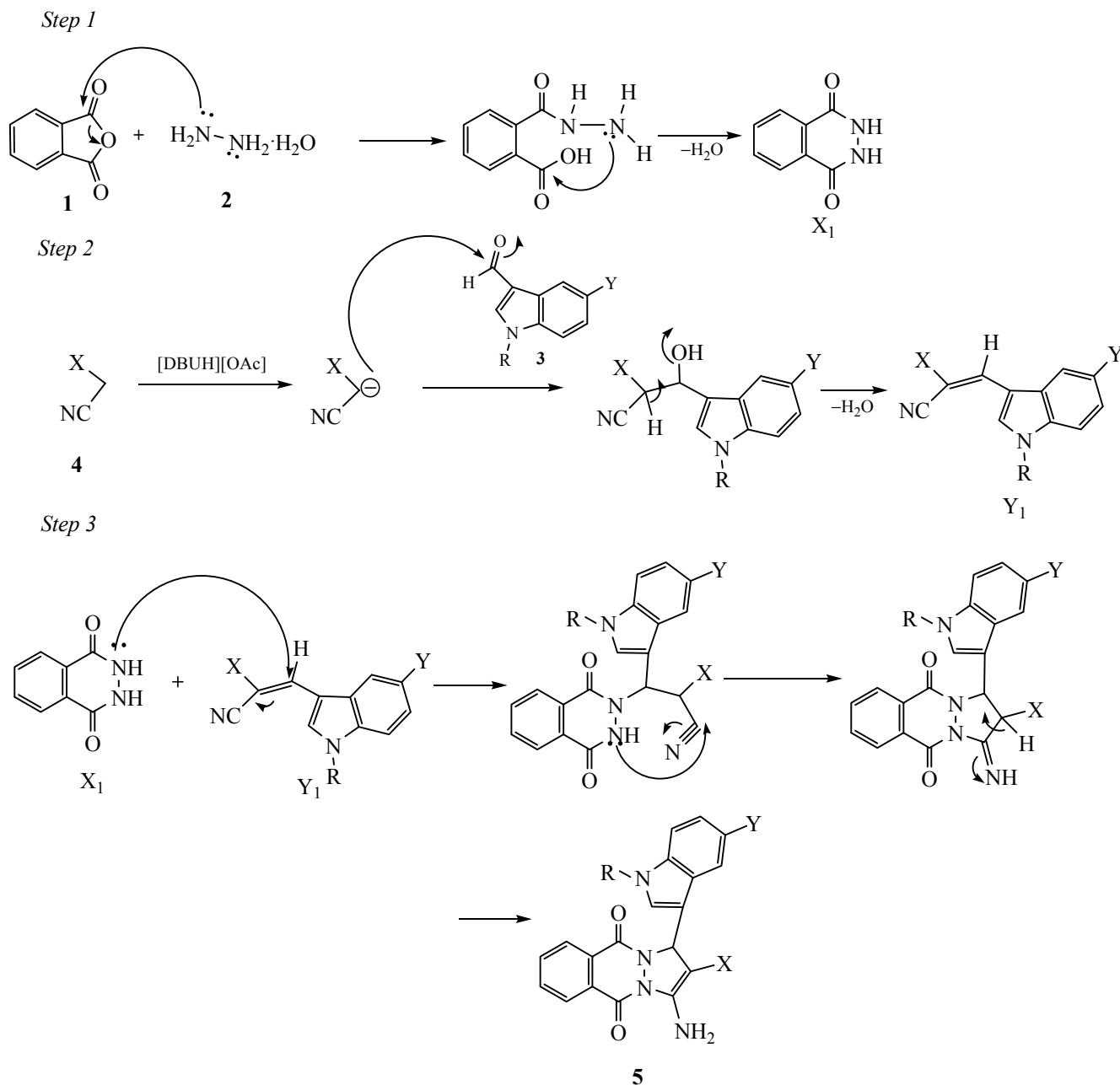
The plausible mechanism of the process can be presented as follows (Scheme 2). In the first step, formation of phthalhydrazide X₁ is achieved by nucleophilic addition of hydrazine hydrate **2** to phthalic anhydride **1**, which is followed by dehydration. The second step involves formation of heterodiene Y₁ by the standard Knoevenagel condensation of indole-3-carboxaldehyde **3** with malononitrile or ethyl cyanoacetate **4**. In the third step, the Michael-type addition of phthalhydrazide X₁ to the heterodiene Y₁ leads to formation of the intermediate iminomethylene derivative which undergoes cyclisation affording **5**.

EXPERIMENTAL

Melting points were determined in open capillary tubes. TLCs were carried out on silica gel G and visualized by iodine vapour or under UV light. IR

Table 1. Effects of ionic liquids and the reaction conditions upon the yield of the compound **5a**

| [Ionic liquid]/equiv. | Temperature, °C | Time, min | Yield of 5a , % | [Ionic liquid]/equiv. | Temperature, °C | Time, min | Yield of 5a , % |
|-----------------------|------------------|-----------|------------------------|-----------------------|-----------------|-----------|------------------------|
| [DBUH][OAc]/5 | Room temperature | 120 | 65 | [DBUH][OAc]/5 | 60 | 30 | 85 |
| [bmim][Br]/5 | Room temperature | 240 | 60 | [DBUH][OAc]/5 | 80 | 30 | 60 |
| [bmim][OH]/5 | Room temperature | 180 | 55 | [DBUH][OAc]/2 | 60 | 90 | 70 |
| [DBUH][OAc]/5 | 40 | 90 | 75 | [DBUH][OAc]/10 | 60 | 30 | 80 |

Scheme 2. Plausible mechanism for 3-amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine derivatives **5**.

spectra were recorded on a Perkin Elmer 1000 spectrophotometer in KBr pellets. ^1H and ^{13}C NMR spectra were recorded in $\text{DMSO}-d_6$ using TMS as an internal standard at 400 MHz operating frequency. Mass spectra were measured on an Agilent-LCMS instrument.

General procedure of synthesis of 3-amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine derivatives (5**).** Phthalic anhydride

(**1**) (10 mM) and hydrazine hydrate (**2**) (10 mM) were added to [DBUH][OAc] (50 mM) and heated for 10 min at 60–65°C to form phthalhydrazide as an intermediate. Then, to this reaction mixture were added 5-nitroindole-3-carboxaldehydes (**3a**) (10 mM) and malanonitrile/ethylcyanoacetate (**4**) (10 mM). The reaction mixture was heated for 20–35 min until no starting materials could be detected by TLC. Upon completion of the process, cold water was added and

Table 2. Synthesis data for compounds **5a–5h**

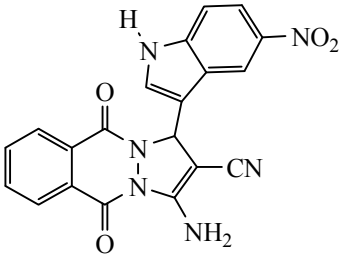
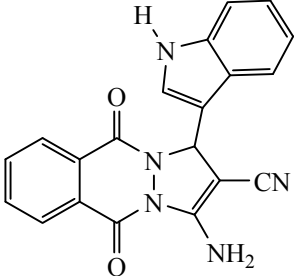
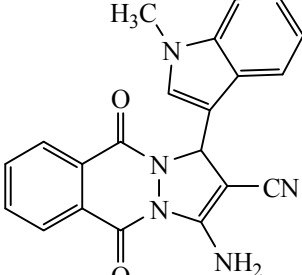
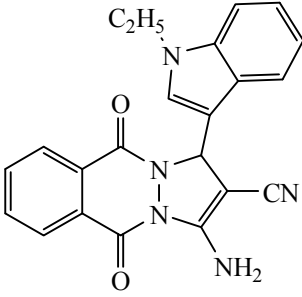
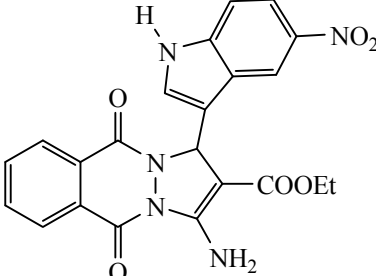
| Comp. no. | Reagents | | | | Product | Time, min | Yield, % |
|-----------|----------|----------|----------------------------------------------------------|--------------------------------------------|--------------------------------------------------------------------------------------|-----------|----------|
| 5a | 1 | 2 | 3a (R = H, Y = NO ₂) | 3a (R = H, Y = NO ₂) |  | 30 30 | 30 |
| 5b | 1 | 2 | 3b (R = H, Y = H) | 4a (X = CN) |  | 40 | 82 |
| 5c | 1 | 2 | 3c (R = CH ₃ , Y = H) | 4a (X = CN) |  | 45 | 78 |
| 5d | 1 | 2 | 3d (R = C ₂ H ₅ , Y = H) | 4a (X = CN) |  | 40 | 81 |
| 5e | 1 | 2 | 3a (R = H, Y = NO ₂) | 4b (X = COOEt) |  | 30 | 83 |

Table 2. (Contd.)

| Comp. no. | Reagents | | | | Product | Time, min | Yield, % |
|-----------|----------|---|---------------------------------------------------|----------------|---------|-----------|----------|
| 5f | 1 | 2 | 3b (R = H, Y = H) | 4b (X = COOEt) | | 45 | 82 |
| 5g | 1 | 2 | 3c (R = CH ₃ , Y = H) | 4b (X = COOEt) | | 45 | 82 |
| 5h | 1 | 2 | 3d (R = C ₂ H ₅ , Y = H) | 4b (X = COOEt) | | 40 | 81 |

the solid residue was filtered off. The corresponding product **5** was recrystallized from ethanol.

3-Amino-1-(5-nitro-1H-indol-2-yl)-5,10-dioxo-5,10-dihydro-1H-pyrazolo[1,2-b]phthalazine-2-carbonitrile (5a). mp >220°C. IR spectrum, ν , cm⁻¹: 3116–3440 br.m (NH), 2218 s (CN), 1669 s (CO, amide), 1686 s (CO, amide). ¹H NMR spectrum, δ , ppm: 5.67 s (1H, CH), 7.26–8.68 m (10H, Ar-H, NH₂), δ 11.87 s (1H, NH). ¹³C NMR spectrum, δ , ppm: 61.5, 69.1, 110.6, 111.5, 115.8, 115.9, 119.2, 122.9, 123.6, 127.6, 134.6, 135.6, 138.4, 144.6, 145.8, 161.6, 164.5. MS: m/z : 400 [$M + H$]⁺.

3-Amino-1-(1H-indol-2-yl)-5,10-dioxo-5,10-dihydro-1H-pyrazolo[1,2-b]phthalazine-2-carbonitrile (5b). mp >220°C. IR spectrum, ν , cm⁻¹: 3116–3440 br.m (NH), 2218 s (CN), 1669 s (CO, amide), 1686 s (CO, amide). ¹H NMR spectrum, δ , ppm: 5.67 s (1H, CH), 7.26–8.68 m (11H, Ar-H, NH₂), δ 11.87 s (1H, NH). ¹³C NMR spectrum, δ , ppm: 61.0, 69.0, 110.1, 111.5,

115.9, 119.2, 122.9, 123.9, 127.3, 134.6, 135.8, 138.4, 144.6, 145.8, 161.0, 164.5. MS: m/z : 356 [$M + H$]⁺.

3-Amino-1-(1-methyl-1H-indol-2-yl)-5,10-dioxo-5,10-dihydro-1H-pyrazolo[1,2-b]phthalazine-2-carbonitrile (5c). mp >220°C. IR spectrum, ν , cm⁻¹: 2213 s (CN), 1668 s (CO, amide), 1682 s (CO, amide). ¹H NMR spectrum, δ , ppm: 2.20 s (3H, CH₃), 5.30 s (1H, CH), 7.21–8.68 m (11H, Ar-H, NH₂). ¹³C NMR spectrum, δ , ppm: 23.4, 60.1, 68.1, 111.3, 111.4, 114.8, 118.1, 122.9, 123.6, 127.4, 133.5, 134.7, 138.3, 144.2, 145.9, 161.4, 164.4. MS: m/z : 370 [$M + H$]⁺.

3-Amino-1-(1-ethyl-1H-indol-2-yl)-5,10-dioxo-5,10-dihydro-1H-pyrazolo[1,2-b]phthalazine-2-carbonitrile (5d). mp >220°C. IR spectrum, ν , cm⁻¹: 2216 s (CN), 1663 s (CO, amide), 1678 s (CO, amide). ¹H NMR spectrum, δ , ppm: 1.81 t (3H, CH₃), 2.22 q (2H, CH₂), 5.26 s (1H, CH), 7.21–8.94 m (11H, Ar-H, NH₂). ¹³C NMR spectrum, δ , ppm: 19.2, 23.2, 60.4, 68.5, 111.2, 111.5, 114.2, 118.2, 122.4, 123.2, 127.1,

133.0, 134.3, 138.2, 144.0, 145.2, 161.5, 164.9. MS: m/z : 384 $[M + H]^+$.

Ethyl-3-amino-1-(5-nitro-1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carboxylate (5e). mp >220°C. IR spectrum, ν , cm^{-1} : 3076–3360 m (NH), 2297 s (CN), 1660 s (CO, amide), 1666 s (CO, amide). ^1H NMR spectrum, δ , ppm: 1.32 t (3H, CH_3), 4.46 q (2H, CH_2), 6.02 s (1H, CH), 7.20–8.69 m (10H, Ar-H, NH_2), 11.79 s (1H, NH). ^{13}C NMR spectrum, δ , ppm: 14.0, 61.9, 69.1, 74.0, 110.6, 111.1, 115.8, 115.9, 119.1, 122.9, 123.6, 127.1, 134.6, 135.7, 140.6, 143.7, 151.6, 155.6. MS: m/z : 402 $[M + H]^+$.

Ethyl-3-amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carboxylate (5f). mp >220°C. IR spectrum, ν , cm^{-1} : 3112–3453 m (NH), 2204 s (CN), 1668 s (CO, amide), 1672 s (CO, amide). ^1H NMR spectrum, δ , ppm: 1.24 t (3H, CH_3), 4.18 q (2H, CH_2), 5.42 s (1H, CH), 7.22–8.68 m (11H, Ar-H, NH_2), 11.78 s (1H, NH). ^{13}C NMR spectrum, δ , ppm: 15.3, 55.2, 60.5, 68.1, 110.3, 111.4, 115.0, 117.2, 122.0, 123.5, 127.2, 133.0, 134.9, 137.3, 142.5, 144.5, 150.1, 156.4. MS, m/z : 403 $[M + H]^+$.

Ethyl-3-amino-1-(1-methyl-1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carboxylate (5g). mp >220°C. IR spectrum, ν , cm^{-1} : 2213 s (CN), 1664 s (CO, amide), 1683 s (CO, amide). ^1H NMR spectrum, δ , ppm: 1.12 t (3H, CH_3), 2.24 s (3H, CH_3), 4.00 q (2H, CH_2), 5.42 s (1H, CH), 7.22–8.64 m (11H, Ar-H, NH_2). ^{13}C NMR spectrum, δ , ppm: 15.3, 22.5, 56.3, 61.4, 67.0, 111.4, 113.5, 114.7, 118.0, 122.0, 122.7, 125.0, 132.4, 134.6, 138.0, 144.1, 143.5, 152.2, 153.5. MS: m/z : 417 $[M + H]^+$.

Ethyl-3-amino-1-(1-ethyl-1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carboxylate (5h). mp >220°C. IR spectrum, ν , cm^{-1} : 2215 s (CN), 1666 s (CO, amide), 1673 s (CO, amide). ^1H NMR spectrum, δ , ppm: 1.18 t (3H, CH_3), 1.69 t (3H, CH_3), 2.39 q (2H, CH_2), 4.15 q (2H, CH_2), 5.26 s (1H, CH), 7.22–8.93 m (11H, Ar-H, NH_2). ^{13}C NMR spectrum, δ , ppm: 15.2, 19.4, 23.6, 54.3, 60.2, 68.4, 111.5, 111.7, 114.0, 118.1, 122.3, 124.1, 126.8, 133.4, 134.9, 138.1, 144.2, 145.5, 151.6, 154.6. MS: m/z : 431 $[M + H]^+$.

CONCLUSIONS

In summary, a novel method of synthesis of 3-amino-1-(5-nitro-1*H*-indol-2-yl)-5,10-dioxo-5,10-di-

hydro-1*H*-pyrazolo[1,2-*b*]phthalazine derivatives has been developed by the reaction of phthalic anhydride with hydrazine hydrate, indole-3-carboxaldehydes and malononitrile/ethyl cyanoacetate in the presence of [DBUH][OAc] at 60–65°C. This one-pot process is characterized by short reaction time, high yields and purity of products isolated.

ACKNOWLEDGMENTS

Authors are very thankful to the authorities of the Department of Organic Chemistry, Foods, Drug and Water, College of Science and Technology, Andhra University, Visakhapatnam 530003, Andhra Pradesh, INDIA for constant support.

CONFLICT OF INTERESTS

No conflict of interests was declared by the authors.

REFERENCES

1. El-Sakka, S.S., Soliman, A.H., and Imam, A.M., *Afinidad*, 2009, vol. 66, p. 167. <https://core.ac.uk/download/pdf/39152404.pdf>
2. Zhang, L., Guan, L.P., Sun, X.Y., Wei, C.X., Chai, K.Y., and Quan, Z.S., *Chem. Bio. Drug. Design*, 2009, vol. 73, p. 313. doi 10.1111/j.1747-0285.2009.00776.x
3. Ryu, C.-K., Park, R.-E., Ma, M.-Y., and Nho, J.-H., *Bioorg. Med. Chem. Lett.*, 2007, vol. 17, p. 2577. doi 10.1016/j.bmcl.2007.02.003
4. Li, J., Zhao, Y.F., Yuan, X.Y., Xu, J.X., and Gong, P., *Molecules*, 2006, vol. 11, p. 574. doi 10.3390/11070574
5. (a) Sinkkonen, J., Ovcharenko, V., Zelenin, K.N., Bezhan, I.P., Chakchir, B.A., Al-Assar, F., and Pihlaja, K., *Eur. J. Org. Chem.*, 2002, p. 2046. doi 10.1002/1099-0690(200207)2002:13<2046::AID-EJOC2046>3.0.CO;2-C (b) Reddy, Y.D., Narayana, B.S., Reddy, Ch.V.R., and Dubey, P.K., *Synth. Commun.*, 2014, vol. 44, p. 3037. doi 10.1080/00397911.2014.928326
6. Ghahremanzadeh, R., Ahadi, S., Sayyafi, M., and Bazgir, A., *Tetrahedron Lett.*, 2008, vol. 49, p. 4479. doi 10.1016/j.tetlet.2008.05.063
7. Li, J., Zhao, Y.F., Yuan, X.Y., Xu, J.X., and Gong, P., *Molecules*, 2006, vol. 11, p. 574. doi 10.3390/11070574
8. Ghahremanzadeh, R., Shakibaei, G.I., and Bazgir, A., *Synlet*, 2008, vol. 8, p.1129. doi 10.1055/s-2008-1072716
9. Nabid, M.R., Tabatabaie, S.J., Ghahremanzadeh, R., and Bazgir, A., *Ultrason. Sonochem.*, 2010, vol. 17, p. 159. doi 10.1016/j.ultsonch.2009.06.012

10. Raghuvanshi, D.S. and Singh, K.N. *Tetrahedron Lett.*, 2011, vol. 52, p. 5702. doi 10.1016/j.tetlet.2011.08.111
11. Shaterian, H.R. and Mohammadnia, M., *J. Mol Liquids*, 2012, vol. 173, p. 55. doi 10.1016/j.molliq.2012.06.007
12. Liu, L.-P., Lu, J.-M., and Shi, M., *Org. Lett.* 2007, vol. 9, p. 1303. doi 10.1021/ol070178r
13. Zhao, D.B., Fei, Z.F., Geldbach, T.J., Scopelliti, R., and Dyson, P.J., *J. Am. Chem. Soc.*, 2004, vol. 126, p. 15876. doi 10.1021/ja0463482
14. Herrmann, W.A., *Angew. Chem., Int. Ed.*, 2002, vol. 41, p. 1290. doi 10.1002/1521-3773(20020415)41:8<1290::AID-ANIE1290>3.0.CO;2-Y
15. Dupont, J., de Souza, R.F., and Suarez, P.A.Z., *Chem. Rev.*, 2002, vol. 102, p. 3667. doi 10.1021/cr010338r
16. Song, C.E., Oh, C.R., Roh, E.J., and Choo, D.J., *Chem. Commun*, 2000, p. 1743. doi 10.1039/B004645K
17. Yan, N., Yuan, Y.A., Dykeman, R., Kou, Y.A., and Dyson, P.J., *Angew. Chem. Int. Ed.*, 2010, vol. 49, p. 5549. doi 10.1002/ange.201001531
18. Bo, Y., Hongye, Z., Yanfei, Z., Sha, C., Jilei, X., Leiduan, H., and Zhimin, L., *ACS Catal.*, 2013, vol. 3, p. 2076. doi 10.1021/cs400256j