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# Three new monomeric indole alkaloids from the roots of Catharanthus roseus

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# Three new monomeric indole alkaloids from the roots of *Catharanthus roseus*

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Two new alkaloid glycosides (1 and 2) and a new monomeric indole alkaloid (3), together with 13 known ones (4-16), were isolated from the roots of *Catharanthus roseus*. The structures of new compounds were elucidated on the basis of the spectroscopic methods. The absolute configuration of compound 3 was established by the modified Mosher's method.

Keywords: Catharanthus roseus; Apocynaceae; indole alkaloid glycoside; Mosher's method

#### 1. Introduction

The plant Catharanthus roseus (Apocynaceae) is widely distributed in tropical and semitropical regions. Up to now, over 130 indole alkaloids have been isolated from this plant [1]. Vinblastine and vincristine are the two typical indole alkaloids isolated from this plant in the 1950-1960s [2,3], which were used in clinic for the treatment of various leukemias, Hodgkin's disease, and solid tumors [4]. With the aim of searching for indole alkaloids with high biological activities, we had isolated some new indole alkaloids from the aerial parts of this plant [5,6]. Recently, we have carried out the phytochemical investigations on the roots of this plant. This paper reports the isolation and structural elucidation of three new alkaloids (1-3)(Figure 1) from the roots of C. roseus, together with 13 known ones (4-16).

#### 2. Results and discussion

Compound 1 was obtained as an amorphous powder. The molecular formula of 1 was determined as C<sub>26</sub>H<sub>34</sub>N<sub>2</sub>O<sub>7</sub> on the basis of its HR-ESI-MS at m/z 487.2429  $[M + H]^+$ . The IR spectrum of 1 implied the presence of hydroxyl group  $(3357 \text{ cm}^{-1})$ , ester group  $(1730 \text{ cm}^{-1})$ , and double bond  $(1594 \text{ cm}^{-1})$ . The <sup>1</sup>H NMR spectrum showed three aromatic protons at  $\delta_{\rm H}$  6.82 (1H, dd, J = 8.5, 2.5 Hz), 6.73 (1H, d, J = 2.5 Hz), and 6.56 (1H, d, J = 8.5 Hz; an olefinic proton at  $\delta_{\text{H}} 5.48$ (1H, q, J = 7.0 Hz); three methyl groups at  $\delta_{\rm H}$  3.80 (3H, s), 2.65 (3H, s), and 1.52 (3H, d, J = 7.0 Hz); and an anomeric proton at  $\delta_{\rm H}$  4.61 (1H, d, J = 7.5 Hz). Accordingly, the  ${}^{13}C$  NMR spectrum of **1** displayed the signals for an aromatic ring ( $\delta_{\rm C}$  153.2, 150.1, 142.6, 116.9, 112.4, and 111.4), a pair of olefinic carbons ( $\delta_{\rm C}$  138.5 and

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Figure 1. Chemical structures of compounds 1-3

121.9), an ester carbonyl carbon ( $\delta_{\rm C}$  174.5), three methyls ( $\delta_{\rm C}$  52.4, 35.5, and 13.7), and an arabinopyranose unit ( $\delta_{\rm C}$  104.4, 74.4, 72.5, 69.8, and 67.3).

The aglycone and sugar unit were obtained by acid hydrolysis of 1. The spectral and the optical rotation data of the aglycone were the same as those of 10-hydroxycathafoline [7], confirming the aglycone was 10-hydroxycathafoline. The sugar unit was derived and analyzed by GC using authentic samples as references, which suggested the sugar was L-arabinopyranose. The coupling constant (J = 7.5 Hz) of the anomeric proton signal indicated the presence of  $\alpha$ -L-arabinopyranosyl. The HMBC correlation between H- $1' (\delta_{\rm H} 4.61)$  and C-10 ( $\delta_{\rm C} 153.2$ ) (Figure 2) suggested that the arabinopyranose was attached to C-10 position of 10-hydroxycathafoline. Based on the above data, 1 was determined to be 10-hydroxycathafoline  $10-O-\alpha-L$ -arabinopyranoside.

Compound **2** was obtained as an amorphous powder. The molecular formula  $C_{26}H_{34}N_2O_7$  was established by its HR-ESI-MS at m/z 487.2440 [M + H]<sup>+</sup>. The IR spectrum showed the presence

of the absorption bands ascribable to hydroxyl group  $(3367 \,\mathrm{cm}^{-1})$ , ester group  $(1733 \,\mathrm{cm}^{-1})$ , and double bond  $(1596 \text{ cm}^{-1})$ . The <sup>1</sup>H NMR spectrum gave three protons due to aromatic ring  $[\delta_{\rm H} 7.04 \,(1\text{H}, \text{d}, J = 2.5 \,\text{Hz}), 6.81 \,(1\text{H}, \text{dd},$ J = 8.0, 2.5 Hz, and 6.22 (1H, d,  $J = 8.0 \,\mathrm{Hz}$ ]; one olefinic proton [ $\delta_{\mathrm{H}}$  5.50 (1H, q, J = 7.0 Hz); three methyl signals  $[\delta_{\rm H} 3.78 (3 {\rm H}, {\rm s}), 2.57 (3 {\rm H}, {\rm s}), {\rm and} 1.60$ (3H, d, J = 7.0 Hz)]; and an anomeric proton [ $\delta_{\rm H}$  4.59 (1H, d,  $J = 7.0 \,\rm{Hz}$ )]. The <sup>13</sup>C NMR spectrum showed the presence of an aromatic ring ( $\delta_C$  151.8, 146.6, 139.3, 118.0, 115.6, and 107.0), a pair of olefinic carbons ( $\delta_{\rm C}$  139.9 and 124.5), an ester carbonyl carbon ( $\delta_{\rm C}$  175.3), three methyl groups ( $\delta_C$  51.9, 28.8, and 14.0), and an arabinopyranose unit ( $\delta_{\rm C}$  104.9, 74.4, 72.6, 69.8, and 67.2).

Acid hydrolysis of **2** afforded Larabinopyranose and aglycone, the aglycone was elucidated as 10-hydroxydeformodihydropseudoakuammigine [8,9] by the comparison of its NMR spectral data. The HMBC correlation between H-1' ( $\delta_{\rm H}$ 4.59) and C-10 ( $\delta_{\rm C}$  151.8) (Figure 2) indicated that the L-arabinopyranose unit



Figure 2. Key HMBC and ROESY correlations of 1–3.

was attached to C-10 position of 10hydroxydeformodihydropseudoakuammigine. Hence, **2** was characterized as 10hydroxydeformodihydropseudoakuammigine 10-O- $\alpha$ -L-arabinopyranoside.

Compound 3 was isolated as colorless needles. The molecular formula was determined as C<sub>21</sub>H<sub>26</sub>N<sub>2</sub>O<sub>3</sub> on the basis of its HR-ESI-MS at m/z 355.2019  $[M + H]^+$ . The IR spectrum of 3 showed the characteristic absorptions due to hydroxyl group  $(3374 \text{ cm}^{-1})$  and ester group  $(1707 \text{ cm}^{-1})$ . The <sup>1</sup>H NMR spectrum showed four protons for aromatic ring [ $\delta_{\rm H}$  7.27 (1H, dd, J = 7.0, 1.0 Hz), 6.99 (1H, ddd, J = 7.5, 7.0, 1.5 Hz), 6.74 (1H, ddd, J = 8.0, 7.5, 1.0 Hz), and 6.67 (1H, dd, J = 8.0, 1.5 Hz)], and two methyl groups [ $\delta_{\rm H}$  3.69 (3H, s) and 0.96 (3H, d,  $J = 7.0 \,\mathrm{Hz}$ ]. The <sup>13</sup>C NMR spectrum displayed the signals of an aromatic ring  $(\delta_{\rm C} 151.1, 139.4, 128.6, 124.6, 122.3, and$ 121.2), one ester carbonyl ( $\delta_{\rm C}$  176.5), two methyls ( $\delta_{\rm C}$  52.5 and 7.4), and five methylenes ( $\delta_{\rm C}$  56.3, 44.9, 37.5, 32.2, and 30.4). The above data indicated that **3** was an analog of vindolinine [10].

Detailed comparison of NMR spectral data of 3 with those of 19(R)-vindolinine [10] revealed that the NMR data of two compounds were similar, except the absence of a double bond at C-14 and C-15 in 3, suggesting that C-14 or C-15 in 3 may be substituted by a hydroxyl group. The HMBC correlations between H-15 ( $\delta_{\rm H}$  3.91) and C-14 ( $\delta_{\rm C}$  32.2)/C-3 ( $\delta_{\rm C}$ 56.3)/C-19 ( $\delta_{\rm C}$  49.7) (Figure 2) indicated that the hydroxyl group was attached at C-15 position. The relative configuration of **3** was assigned by the ROESY correlations (Figure 2). The absolute configuration of C-15 could be confirmed by the modified Mosher's method. Differences of proton chemical shift ( $\Delta \delta$  values,  $\delta_S - \delta_R$ ) between (S)- $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl ester (3a) and (R)- $\alpha$ methoxy-α-(trifluoromethyl)phenylacetyl ester (3b) (Figure 3) indicated the presence of S configuration at C-15 in 3.



Figure 3.  $\triangle \delta$  values  $(\delta_S - \delta_R)$  of the MTPA esters **3a** and **3b**.

Consequently, **3** was elucidated as 15(S)-hydroxy-14,15-dihydroxindolinine.

In addition, 13 known compounds were identified by comparing their spectral data with those of the literatures as tetrahydroalstonine (4) [11], ajmalicine (5) [12], vindorosine (6) [13], vindoline (7) [14], 19(*S*)-vindolinine (8) [10], 19(*R*)vindolinine (9) [10], (-)-vincapusine (10) [15], yohimbine (11) [12], serpentine (12) [16], 18 $\beta$ -hydroxy-3-epi- $\alpha$ -yohimbine (13) [17], sitsirikine (14) [18], perivine (15) [19], and dihydroantirhine (16) [20], respectively.

#### 3. Experimental

#### 3.1 General experimental procedures

Melting points were recorded on an X-5 micro melting point apparatus without correction. Optical rotations were determined on a JASCO P-1020 polarimeter. UV spectra were obtained on a JASCO V-550 UV-vis spectrophotometer. IR spectra were recorded on a JASCO FT/IR-480 plus spectrometer. <sup>1</sup>H, <sup>13</sup>C, and 2D NMR spectra were determined on a Bruker AV-500 spectrometer in CD<sub>3</sub>OD. ESI-MS were run on a HP-1100 HPLC/EST spectrometer. HR-ESI-MS data were detected on an Agilent 6210 ESI/TOF mass spectrometer. For column chromatography (CC), silica gel (200-300 mesh, Qingdao Marine Chemical Factory, Qingdao, China), Sephadex LH-20 (Pharmacia Biotec AB, Uppsala, Sweden), and Octadecylsilyl (ODS) (YMC, Kyoto, Japan) were used. TLC analyses were carried out using precoated silica gel GF254 plates (Qingdao Marine Chemical Factory). HPLC separations were carried out on a COSMOSIL  $C_{18}$  preparative column (5 µm, 20 × 250 mm, Nacalai Tesque, Kyoto, Japan). L-Arabinopyranose was purchased from Sigma (St Louis, MO, USA).

#### 3.2 Plant material

The roots of *C. roseus* were collected in Haikou City, Hainan Province of China, in April 2009. The plant was authenticated by Prof. Guang-Xiong Zhou of Jinan University. A voucher specimen (No. 0904011) has been deposited in the Institute of Traditional Chinese Medicine and Natural Products, Jinan University, Guangzhou, China.

#### 3.3 Extraction and isolation

The dried roots (12.5 kg) of C. roseus were extracted with 95% EtOH. The solution was evaporated in vacuo to get residue (890 g). The residue was dissolved in  $H_2O$ to form a suspension, and then adjusted to pH 3 with 6% HCl. The acidic suspension was partitioned with CHCl<sub>3</sub> to remove the neutral components. The aqueous phase was basified with 2% NH<sub>3</sub>·H<sub>2</sub>O to pH 10 and then extracted with CHCl<sub>3</sub> to obtain a total alkaloid part (130 g), 120 g of which was subjected to silica gel CC (CHCl<sub>3</sub>-CH<sub>3</sub>OH,  $100:0 \rightarrow 0:100$ ) to give nine fractions. Fraction 1 (19.0 g) was subjected to silica gel [petroleum ether (PE)-EtOAc,  $9:1 \rightarrow 1:1$ ] and Sephadex LH-20 (MeOH) columns to afford 4 (22 mg), 5 (25 mg), 6 (20 mg), and 7 (36 mg). Fraction 2 (14.0 g) was subjected to silica gel column (PE-EtOAc,  $9:1 \rightarrow 7:3$ ) and purified by preparative HPLC (MeOH-H<sub>2</sub>O, 88:12) to yield 8 (25 mg) and 9 (18 mg). Fraction 3 (10.5 g) was subjected to ODS column (MeOH-H<sub>2</sub>O,  $20:80 \rightarrow 100:0$ ) and preparative HPLC (MeOH-H<sub>2</sub>O, 45:55) to obtain 1 (14 mg), 2 (21 mg), 3 (18 mg), 10 (16 mg), 11 (12 mg), 12 (16 mg), and 13 (10 mg). The same methods were used to isolate 15 (11 mg) and 16 (8 mg) from fraction 4 (11.0 g), as well as 14 (12 mg) from fraction 5 (8.5 g), respectively.

#### 3.3.1 10-Hydroxycathafoline 10-O- $\alpha$ -Larabinopyranoside (1)

An amorphous powder, mp 204–206°C.  $[\alpha]_{D}^{22} + 29.3$  (c = 0.14, MeOH). UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 210 (1.47), 243 (1.97), 302 (1.55) nm. IR (KBr)  $\nu_{max}$ : 3357, 1730, 1594, 1475, 1072, 762 cm<sup>-1</sup>. For <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1. HR-ESI-MS m/z: 487.2429 [M + H]<sup>+</sup> (calcd for C<sub>26</sub>H<sub>35</sub>N<sub>2</sub>O<sub>7</sub>, 487.2439).

#### 3.3.2 10-Hydroxydeformodihydropseudoakuammigine 10-O- $\alpha$ -L-arabinopyranoside (**2**)

An amorphous powder, mp 196–198°C.  $[\alpha]_{D}^{22}$  – 20.9 (c = 0.11, MeOH). UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 214 (2.13), 230 (2.25), 292 (1.96) nm. IR (KBr)  $\nu_{max}$ : 3367, 1733, 1596, 1085, 794 cm<sup>-1</sup>. For <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1. HR-ESI-MS m/z: 487.2440 [M + H]<sup>+</sup> (calcd for C<sub>26</sub>H<sub>35</sub>N<sub>2</sub>O<sub>7</sub>, 487.2439).

## *3.3.3 15(S)-Hydroxy-14,15- dihydrovindolinine* (*3*)

Colorless needles, mp 182–184°C.  $[\alpha]_D^{22}$ -42.0 (*c* = 0.15, MeOH). UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 211 (0.68), 239 (1.15), 295 (0.69) nm. IR (KBr)  $\nu_{max}$ : 3374, 1707, 1607, 1466, 1214, 753 cm<sup>-1</sup>. For <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1. HR-ESI-MS *m/z*: 355.2019 [M + H]<sup>+</sup> (calcd for C<sub>21</sub>H<sub>27</sub>N<sub>2</sub>O<sub>3</sub>, 355.2016).

## 3.4 Acid hydrolysis and GC analysis of 1 and 2

Compounds 1 and 2 (each 3.0 mg) were dissolved in 2 mol/l HCl (1.0 ml) and heated at 80°C for 2 h, respectively. The

Table 1.	<sup>1</sup> H and <sup>13</sup> C NMR spectral data of <b>1</b>	3 (measured in	500  MHz for <sup>1</sup> H, 125 MHz for	<sup>13</sup> C, in CD <sub>3</sub> OD, δ	in ppm, $J$ in Hz).	
	1		2		3	
Position	δ <sub>H</sub>	$\delta_{\mathrm{C}}$	$\delta_{\rm H}$	δ <sub>C</sub>	$\delta_{\rm H}$	$\delta_{\rm C}$
5	2.57 br s	80.3	1	99.4	1	81.6
3	3.35 m	50.0	2.57	41.9	3.41	56.3
			1.90		3.26	
5	3.86	51.4	3.36	55.8	3.41	44.9
	2.71 t (7.0)		2.79 t (10.5)		2.96 m	
9	3.14 m	31.6	2.32 m	21.5	2.25 dd (15.0, 6.0)	37.5
	1.36 dd (15.5, 5.0)		1.67		1.79	
L	I	44.2	1	58.7	1	60.8
8	Ι	142.6	I	139.3	1	139.4
6	6.73 d (2.5)	112.4	7.04 d (2.5)	115.6	7.27 dd (7.0, 1.0)	124.6
10	I	153.2	I	151.8	6.99 ddd (7.5, 7.0, 1.5)	121.2
11	6.82 dd (8.5, 2.5)	116.9	6.81 dd (8.0, 2.5)	118.0	6.74 ddd (8.0, 7.5, 1.0)	128.6
12	6.56 d (8.5)	111.4	6.22 d (8.0)	107.0	6.67 dd (8.0, 1.5)	122.3
13	Ι	150.1	Ι	146.6	1	151.1
14	2.38 d (18.0)	33.9	1.83	27.3	2.62 m	32.2
	1.79 d (18.0)		1.77		1.73	
15	3.64 s	35.3	3.63	36.3	3.91 t (4.5)	69.4
16	3.01 d (3.5)	53.8	2.84  br s	51.9	3.18 t (9.5)	41.5
17	1	I	1	I	2.09 m	30.4
					1.58 m	
18	1.52 d (7.0)	13.7	1.60 d (7.0)	14.0	0.96 d (7.0)	7.4
19	5.48 q (7.0)	121.9	5.50 q (7.0)	124.5	1.92 q (7.0)	49.7
20	I	138.5	I	139.9	I	50.2
21	3.96	55.5	3.82	59.1	3.74 s	72.7
	3.06 d (16.0)		3.05 d (15.0)			
COOCH <sub>3</sub>	I	174.5	1	175.3	I	176.5
COOCH <sub>3</sub>	3.80 s	52.4	3.78 s	51.9	3.69 s	52.5
N-CH <sub>3</sub>	2.65 s	35.5	2.57 s	28.8		

munu surred in 500 MHz for  $^{1}$ H 125 MHz for  $^{13}$ C in CD<sub>2</sub>OD  $^{3}$  in <sup>1</sup>H and <sup>13</sup>C NMP sneetral data of 1-3 (m Table 1

253

254

continued	
1	
e	
q.	
$\Gamma_a$	

	$\delta_{\rm C}$	
3	$\delta_{ m H}$	
7	$\delta_{\rm C}$	104.9 72.6 74.4 69.8 67.2
	$\delta_{ m H}$	4.59 d (7.0) 3.74 dd (9.0, 7.0) 3.59 dd (9.0, 3.5) 3.85 m 3.90 dd (12.5, 2.5) 3.61 dd (12.5, 1.0)
1	$\delta_{\rm C}$	104.4 72.5 74.4 69.8 67.3
	$\delta_{\rm H}$	4.61 d (7.0) 3.73 dd (9.0, 7.0) 3.59 dd (9.0, 3.5) 3.85 m 3.90 dd (12.5, 2.5) 3.62 dd (12.5, 1.0)
	Position	v 4 % 7 –

Note: Overlapped signals are reported without designating multiplicity

aglycone was extracted with CH<sub>2</sub>Cl<sub>2</sub>, and the aqueous layer was dried by  $N_2$ . Dry pyridine (1.0 ml) and L-cysteine methyl ester hydrochloride (3.0 mg) were added to the residue. The mixture was heated at  $80^{\circ}$ C for 2 h and then dried by N<sub>2</sub>. N-(Trimethylsilyl)imidazole (0.2 ml) was added, and the mixture was heated at 80°C for 1 h. Finally, H<sub>2</sub>O (1.0 ml) was added to stop the reaction and the aqueous layer was extracted with cyclohexane (1.0 ml). The organic layer was analyzed by GC analysis [column, AT-SE-30  $(0.32 \text{ mm} \times 30 \text{ m}, 0.5 \mu\text{m})$ ; detector, FID; column temperature, 220°C; detector temperature, 270°C; injector temperature, 270°C; and carrier gas, N<sub>2</sub>]. The standard monosaccharides were subjected to the same reaction and GC analysis under the above conditions [ $t_{\rm R}$  (min): 22.142 (L-Ara) and 23.278 (D-Ara)]. As a result, L-Ara [ $t_R$ (min): 22.139 and 22.141] was detected from the hydrolyzates of 1 and 2, respectively.

#### 3.5 Preparation of $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl esters of 3

Compound 3 (4.0 mg) was dissolved in 0.5 ml pyridine, and treated with (R)- $\alpha$ methoxy-α-(trifluoromethyl)phenylacetyl Cl  $(10 \,\mu l)$  for 24 h in an anhydrous circumstance at room temperature. The reaction mixture was poured into water (5.0 ml) and extracted with EtOAc (5.0 ml). The EtOAc extract was purified by silica gel CC [n-hexane-EtOAc (70:30)] to yield (S)- $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl ester (3a, (*R*)- $\alpha$ -methoxy- $\alpha$ -(trifluoro-1.8 mg). methyl)phenylacetyl ester (3b, 1.6 mg) was obtained using the same method by the treatment of (S)- $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl Cl  $(10 \,\mu l)$ .

(S)- $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl ester (**3a**): <sup>1</sup>H NMR (CD<sub>3</sub>OD, 500 MHz)  $\delta_{\rm H}$ : 7.48 and 7.23 (phenyl protons of  $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl), 7.31 (1H, dd, J = 7.0, 1.0 Hz, H-9), 7.09 (1H, ddd, J = 7.5, 7.0, 1.5 Hz, H-10), 6.86 (1H, ddd, J = 8.0, 7.5, 1.0 Hz, H-11), 6.78 (1H, dd, J = 8.0, 1.5 Hz, H-12), 5.48 (1H, t, J = 4.5 Hz, H-15), 3.67 (1H, overlapped, H-21), 3.65 (3H, s, H-23), 3.60 (3H, s, OMe of  $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl), 3.47 (1H, m, H-3a), 3.35 (1H, m, H-3b), 2.98 (1H, t, J = 9.0 Hz, H-16), 2.59 (1H, m, H-14a), 2.32 (1H, overlapped, H-17a), 2.24 (1H, overlapped, H-19), 2.18 (1H, m, H-14b), 1.72 (1H, m, H-17b), 1.12 (3H, d, J = 7.0 Hz, H-18); ESI-MS m/z: 571 [M + H]<sup>+</sup>.

(*R*)- $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl ester (3b): <sup>1</sup>H NMR (CD<sub>3</sub>OD, 500 MHz)  $\delta_{\rm H}$ : 7.47 and 7.23 (phenyl protons of  $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl), 7.26 (1H, dd, J = 7.0, 1.5 Hz, H-9, 7.06 (1H, ddd,  $J = 7.5, 7.0, 1.5 \,\text{Hz}, \text{H-10}, 6.84$  (1H, ddd, J = 8.0, 7.5, 1.5 Hz, H-11), 6.75 (1H, dd, J = 8.0, 1.5 Hz, H-12), 5.42 (1H, t, J = 4.5 Hz, H-15), 3.67 (1H, overlapped, H-21), 3.64 (3H, s, H-23), 3.60 (3H, s, OMe of  $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl), 3.52 (1H, m, H-3a), 3.35 (1H, m, H-3b), 2.96 (1H, t, J = 9.0 Hz, H-16), 2.64 (1H, m, H-14a), 2.29 (1H, overlapped, H-14b), 2.26 (1H, overlapped, H-17a), 2.09 (1H, q, J = 7.0 Hz, H-19), 1.68 (1H, m, H-17b), 0.95 (3H, d, J = 7.0 Hz, H-18; ESI-MS m/z: 571  $[M + H]^{+}$ .

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