Synthesis of a 3,1-Benzoxazin-4-one, 2,4(1H,3H)-Quinolinediones, and 2,4(1H,3H)-Quinazolinediones from the Reaction of Phosphoryl-Stabilized Anions Containing no α -Hydrogen Atoms with Isatoic Anhydride

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Since isatoic anhydride (1) is a versatile reagent for the syntheses of heterocyclic compounds containing nitrogen atoms, its chemistry has been widely studied¹. From this point of view, we have been interested in the reactivity of 1 toward phosphorus compounds. Although it is known that 1 reacts with a phosphonium ylid² and a phosphonate carbanion³ to give a resonanced-stabilized ylid and a quinclylphosphonate derivative, no Wittig or Wittig-Horner reaction product is obtained. On the other hand, we previously reported that the reaction of 1 with an α -phosphono- γ -butyrolactone carbanion containing no hydrogen atoms on the α -carbon atom led to a benzoxazine derivative⁴. This result prompted us to investigate the reaction of 1 with phosphoryl-stabilized anions similarly bearing no α -hydrogen atoms.

The reaction of 1 with an ethyl 2-diethylphosphonopropanoate carbanion (3a) in refluxing benzene and benzene/dimethylformamide (4:1) for 5 h gave 2-(1-ethoxycarbonylethyl)-4H-3,1-benzoxazin-4-one (4a) and 3-ethoxycarbonyl-3-methyl-2,4(1H,3H)-quinolinedione (5a) in 64% and 4%, and 18% and 43% yields, respectively. In contrast, similar treatment of N-sodioisatoic anhydride (2), prepared in situ from 1 and an equivalent amount of sodium hydride, with 3a in the mixed solvent led exclusively to 5a in 60% yield.

In comparison to 3a, the reaction of 1 with ethyl 2-diethylphosphonobutanoate (3b) and ethyl 2-diethylphosphono-4-pentenoate carbanions (3c) in the mixed solvent gave no products corresponding to 4a or 5a, instead unreacted 1 was recovered. However, 2, on similar treatment with 3b, 3c, or the ethyl 2-diethylphosphono-3-phenylproparaoate carbanion (3d), afforded only the expected products, 3-ethyl- (5b), 3-allyl- (5c), and 3-benzyl-3-ethoxycarbonyl-2,4(1H,3H)-quinolinediones (5d) in 50%, 69%, and 20% yields, respectively (Scheme A).

Thus, the reaction products and their yields were dependent upon both amounts of base and solvent employed. As shown in Scheme A, these results indicate that hydrogen abstraction from the 1-position of isatoic anhydride (1) by sodium hydride (or the phosphonate carbanion 3) would cause cleavage of the anhydride ring to generate a reactive intermediate isocyanate, which could subsequently react with 3 to yield the intermediate 6, which undergoes a Wittig-Horner reaction to give the quinolinediones 5 (path A), and/or the ketenimine derivative, which undergoes cyclization to give the benzoxazine 4a (path B).

On the other hand, the reaction of 1 with a diethyl N-phenyl-phosphoramidate anion (7a) in refluxing benzene for 5 h led to only 3-phenyl-2,4(1H,3H)-quinazolinedione (8a) in low yield, but the reaction using 2 in benzene/dimethylformamide (4:1) gave 8a in 70% yield. Similar treatment of 2 with var-

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ious phosphoramidate anions **7b-d** in benzene/dimethylformamide provided 3-substituted-2,4(1*H*,3*H*)-quinazolinediones **8b-d** as shown in Table 1 (Scheme **B**).

Scheme B

The formation of the quinazolinediones 8 could be accounted for by a mechanism similar to that for the formation of 5.

Thus, this method for heterocyclic compounds such as the 3,1-benzoxazin-4-one, 2,4-quinolinediones, and quinazolinediones is versatile due to its generality and simplicity, and the ease of preparation of various types of phosphoryl-stabilized anions containing no α -hydrogen atoms used as starting reagents.

All melting points were measured in open capillary tubes and are uncorrected. The N.M.R. spectra were recorded with a JEOL JNM-FX-60 or JNM-PMX-60 spectrometer with TMS as an internal standard. The l.R. spectra were obtained with a Shimazu IR-27c spectrometer. The mass spectra were recorded with a Hitachi RMU-6E spectrometer.

Ethyl 2-diethylphosphonopropanoate, b.p. 117°C/4 torr (Lit.⁷, b.p. 143-144°C/12 torr), and ethyl 2-diethylphosphonobutanoate, b.p. 95°C/1 torr (Lit.⁸, b.p. 147.5-148°C/10.5 torr) were prepared from triethyl phosphite and ethyl 2-bromopropanoate or ethyl 2-bromobutanoate. Ethyl 2-diethylphosphono-4-pentenoate, b.p. 105-112°C/1.8 torr and ethyl 2-diethylphosphono-3-phenylpropanoate, b.p. 135-145°C/0.1 torr were synthesized by the alkylation reaction of ethyl diethylphosphonoacetate carbanion with allyl bromide or benzyl chloride. Diethyl N-phenyl-, m.p. 91-92°C (Lit.⁹, m.p. 93°C), N-p-chlorophenyl-, m.p. 77-78°C (Lit.¹⁰, m.p. 76°C), and N-cyclohexylphosphoramidates, m.p. 74-75°C (Lit.¹¹, m.p. 71-72°C), and dimethyl N-butylphosphorami

Table. 3-Substituted 2,4(1H,3H)-Quinazolinediones 8a-d (Scheme B)

Prod- uct"	Reaction time ^b at 80°C [h]	Yield [%]	m.p. [°C]		1 H-N.M.R. (DMSO- d_{6}) δ [ppm]
			found	reported	o լիբուդ
8a	10°	6°	278-280°	280-282° 5	7.1-8.15 (m, 9 H _{arom}); 11.63 (br s, 1 H, NH)
8a	5	70			
8b	5	99	290~291°	288°6	7.0-8.2 (m, 8 H _{arom}); 11.50 (br s, 1 H, NH)
8c	5	40	265-266°	270~271°5	0.8-2.8 (m, 10 H); 4.25-5.1 (m, 1 H); 6.8-8.05 (m, 4 H _{arom}); 11.20 (br s, 1 H, NH)
8d	10	31	154-155°	156-157°5	0.7-2.0 (m, 7 H); 4.10 (t, 2 H); 6.9-8.3 (m, 4 H _{arom}); 10.80 (s, 1 H, NH)

 $^{^{\}rm a}$ Satisfactory microanalyses (C $\pm 0.37,$ H $\pm 0.15,$ N $\pm 0.19)$ and mass spectra obtained.

Scheme A

Reaction of 2 (15 mmol; prepared in situ from 1 and sodium hydride) and 7 (15 mmol) in 4:1 benzene/dimethylformamide (50 ml) unless otherwise stated.

^c Reaction using 1 in benzene.

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date, b.p. 100.5 °C/1.0 torr (Lit. 12, b.p. 143 °C/9 torr) were prepared from diethyl or dimethyl phosphite and the corresponding amines in the presence of carbon tetrachloride according to the established procedure 13.

Reaction of Isatoic Anhydride (1) with Ethyl 2-Diethylphosphonopropanoate Carbanion (3a):

Reaction in Benzene as Solvent: To a stirred solution of 3a [15 mmol; generated from ethyl 2-diethylphosphonopropanoate (3.57 g. 15 mmol) and sodium hydride (15 mmol), in dry benzene (50 ml) under nitrogen] is added powdered 1 (2.45 g. 15 mmol). The reaction mixture is refluxed for 5 h and then the solvent is evaporated under reduced pressure. The residue is poured into an aqueous solution of ammonium chloride (60 ml, 0.5 molar) and extracted with ether (3 × 50 ml). The combined organic layer is dried with anhydrous sodium sulfate and concentrated. The resulting solid is chromatographed on silica gel using hexane and benzene/hexane as eluents. The first fraction gives 4a; yield: 2.37 g (64%); m.p. 80-81°C (from hexane).

C₁₃H₁₃NO₄ calc. C 63.15 H 5.30 N 5.67 (247.2) found 63.03 5.26 5.60

M.S.: $m/e = 247 \text{ (M}^+)$.

I.R. (Nujol): v = 1760, 1740, 1645 cm⁻¹.

¹H-N.M.R. (CDCl₃): δ = 1.28 (t, 3 H, CH₃); 1.83 (d, 3 H, >CH—CH₃); 3.83 (q, 1 H, >CH—CH₃); 4.27 (q, 2 H, CH₂); 7.3–7.9 (m, 3 H_{arom}); 8.15–8.40 ppm (br d, 1 H_{arom}).

¹³C-N.M.R. (CDCl₃): δ =14.0; 14.2; 46.1; 61.6; 117.2; 127.0; 128.5; 136.3; 146.3; 158.9; 160.2; 169.8 ppm.

The second fraction provides **5a**; yield: 0.15 g (4%); m.p. 147-148 °C (from benzene/hexane).

C₁₃H₁₃NO₄ calc. C 63.15 H 5.30 N 5.67 (247.2) found 63.35 5.28 5.46

M.S.: $m/e = 247 \text{ (M}^+\text{)}$.

I.R. (Nujol): v = 3200, 1760, 1695, 1660 cm⁻¹.

¹H-N.M.R. (CDCl₃): δ = 1.17 (t, 3 H, CH₃); 1.83 [s, 3 H, \geq C(CH₃)(COOC₂H₅)]; 4.27 (q, 2 H, CH₂); 7.1–7.9 (m, 3 H_{arom}); 7.93–8.20 (br d, 1 H_{arom}); 10.37 ppm (br d, 1 H, NH).

¹³C-N.M.R. (CDCl₃): δ =13.8; 20.8; 62.4; 63.9; 116.8; 117.8; 123.9; 128.0; 136.7; 140.9; 166.6; 171.2; 191.6 ppm.

Reaction in Benzene/Dimethylformamide as Solvent: A solution of 1 (2.45 g, 15 mmol) in dry dimethylformamide (10 ml) is added to a solution of 3a (15 mmol) in benzene (40 ml). The reaction mixture is refluxed for 5 h. Similar treatment as described above gives a mixture of 4a; yield: 0.67 g (18%) and 5a; yield: 1.59 g (43%).

Reaction of N-Sodioisatoic Anhydride (2) with 3a in Benzene/Dimethylformamide:

A solution of 1 (2.45 g, 15 mmol) in dry dimethylformamide (10 ml) is added dropwise to a stirred solution of 3a (15 mmol) in dry benzene (40 ml) containing sodium hydride (15 mmol). The reaction is carried out in a similar manner as above and worked up to give 5a; yield: 2.23 g (60%).

3-Ethoxycarbonyl-3-ethyl-2,4(1H,3H)-quinolinedione (5b):

The reaction of 2 (15 mmol) with ethyl 2-diethylphosphonobutanoate carbanion (3b, 15 mmol) in benzene/dimethylformamide (4:1) is carried out in a similar manner as above to give the quinolinedione 5b; yield: 1.95 g (50%); m.p. 262-263 °C (from benzene).

C₁₄H₁₅NO₄ calc. C 64.36 H 5.79 N 5.36 (261.6) found 64.09 5.90 5.09

M.S.: $m/e = 261 \text{ (M}^+)$.

I.R. (KBr): v = 3200, 1750, 1695, 1660 cm⁻¹.

¹H-N.M.R. (CDCl₃): δ = 0.89 (t, 3 H, CH₃); 1.17 (t, 3 H, CH₃); 2.48 (q, 2 H, CH₂); 4.22 (q, 2 H, OCH₂); 7.0-8.1 (m, 4 H_{arom}); 10.31 ppm (br s, 1 H, NH).

¹³C-N.M.R. (CDCl₃): δ =8.9; 14.0; 29.7; 62.4; 69.0; 116.7; 118.9; 123.9; 127.6; 136.6; 140.7; 166.2; 170.3; 190.3 ppm.

3-Allyl-3-ethoxycarbonyl-2,4(1H,3H)-quinolinedione (5c):

The quinolinedione 5c is similarly obtained from the reaction of 2 (20 mmol) with ethyl 2-diethylphosphono-4-pentenoate carbanion (20

mmol) in benzene/dimethylformamide (4:1), yield: 3.77 g (69%); m.p. 121-122 °C (from benzene/hexane).

C₁₅H₁₅NO₄ calc. C 65.92 H 5.53 N 5.13 (273.3) found 65.67 5.74 5.02

M.S.: $m/e = 273 \text{ (M}^+\text{)}.$

I.R. (KBr): v = 3200, 1760, 1700, 1660, 1620 cm⁻¹.

¹H-N.M.R. (CDCl₃): δ = 1.16 (t, 3 H, CH₃); 3.17 (d, 2 H, —CH₂—CH=CH₂); 4.21 (q, 2 H, CH₂); 4.8-6.0 (m, 3 H, —CH₂—CH=CH₂); 7.0-8.1 (m, 4 H_{arom}); 10.51 ppm (s, 1 H, NH). ¹³C-N.M.R. (CDCl₃): δ = 13.8; 40.2; 62.5; 68.0; 116.9; 118.8; 120.3; 124.0; 127.7; 130.4; 136.8; 141.1; 166.0; 170.0; 191.0 ppm.

The reaction in benzene gave no product; starting isatoic anhydride was recovered in 88% yield.

3-Benzyl-3-ethoxycarbonyl-2,4(1H,3H)-quinolinedione (5d):

The quinolinedione 5d is similarly produced from the reaction using 2 (15 mmol) and 3d (15 mmol) in 4:1 benzene/dimethylformamide (50 ml); yield: 0.85 g (20%); m.p. 184-184.5°C (from benzene).

C₁₉H₁₇NO₄ calc. C 70.57 H 5.30 N 4.33 (323.3) found 70.52 5.32 4.41

M.S.: $m/e = 323 \text{ (M}^+)$.

I.R. (KBr): v = 3200, 1750, 1695, 1660, 1610 cm⁻¹.

¹H-N.M.R. (CDCl₃); δ = 1.14 (t, 3 H, CH₃); 3.72 (s, 2 H, CH₂Ph); 4.20 (q, 2 H, CH₂); 6.7-7.9 (m, 10 H_{aron}); 9.84 ppm (s, 1 H, NH).

¹³C-N.M.R. (CDCI₃): δ = 13.8; 42.0; 62.6; 69.0; 116.5; 119.0; 123.9; 125.4; 126.0; 127.2; 127.4; 128.0; 129.7; 133.7; 136.6; 140.7; 166.1; 170.1; 191.6 ppm.

3-Phenyl-2,4(1H,3H)-quinazolinedione (8a); Typical Procedure:

To a solution containing sodium hydride (15 mmol) and the phosphoramidate anion 7a, prepared from diethyl N-phenylphosphoramidate (3.44 g, 15 mmol) and sodium hydride (15 mmol), in dry benzene (40 ml) under nitrogen is added a solution of 1 (2.45 g, 15 mmol) in dry dimethylformamide (10 ml). The mixture is refluxed for 5 h. After similar work-up, the ether extract gives 3-phenyl-2,4(1H,3H)-quinazoline-dione (8a); yield: 2.50 g (70%).

I.R. (KBr): v = 3200, 1725, 1660 cm⁻¹.

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See for example: G. M. Coppola, Synthesis 1980, 505 and references cited therein.

² D. T. Connor, M. v. Strandtmann, J. Org. Chem. 38, 1047 (1973).

³ G. M. Coppola, G. E. Hardtmann, O. R. Pfister, J. Org. Chem. 41, 825 (1976).

⁴ T. Minami, M. Matsumoto, H. Suganuma, T. Agawa, J. Org. Chem. 43, 2149 (1978).

⁵ B. Taub, J. B. Hino, J. Org. Chem. 26, 5238 (1961).

⁶ G. R. Dave, G. S. Mewada, G. C. Amin, Acta Chim. Acad. Sci. Hung. 34, 101 (1962); C. A. 59, 627 (1963).

A. E. Arbuzov, A. A. Dunin, Ber. Disch. Chem. Ges. 60, 291 (1927).

⁸ A. E. Arbuzov, A. A. Dunin, Zh. Fiz. Khim. 46, 295 (1914); C. A. 8, 2551 (1914).

⁹ A. Michaelis, G. Schulze, Ber. Dtsch. Chem. Ges. 27, 2572 (1894).

V. Ettel, J. Myska, J. Bestova-Zavorkova, Sb. Vysoke Skoly. Chem. Technol. Praze. Technol. 5, 211 (1961); C. A. 61, 14561 (1964).

T. Wagner-Jauregg, J. J. O'Neill, W. H. Summerson, J. Am. Chem. Soc. 73, 5202 (1951).

K. V. Nikoronov, E. A. Guryler, A. V. Chernova, *Izv. Akad. Nauk SSSR*, Ser. Khim. 1968, 587; C. A. 69, 95864 (1968).

E. Fluck, W. Haubold, Organic Phosphorus Compounds, Vol. 6, Chap. 16, G. M. Kosolapoff, L. Maier, Eds., Wiley-Interscience, New York, London, 1973, p. 614.