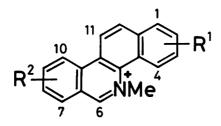
A NOVEL AND BIOMIMETIC SYNTHESIS OF (±)-CHELAMINE, (±)-CHELIDONINE, SANGUINARINE, AND DIHYDROSANGUINARINE FROM COPTISINE VIA A COMMON INTERMEDIATE

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 $(\pm)$ -Chelamine and  $(\pm)$ -chelidonine, B/C cis hexahydrobenzo[c]-phenanthridine alkaloids were stereoselectively synthesized from coptisine via the key intermediate, which was also converted to fully aromatized benzo[c]phenanthridine alkaloids, sanguinarine and dihydrosanguinarine.

Benzo [c] phenanthridine alkaloids can be classified into two groups, fully aromatized benzo [c] phenanthridines and B/C hexahydro ones, and they have been shown to be biosynthesized from protoberberine alkaloids.<sup>1)</sup> Many efforts<sup>2)</sup> have been focused on development of efficient and convenient methods for a synthesis of fully aromatized benzo [c] phenanthridine alkaloids because of their potential pharmacological activities. Several syntheses of hexahydrobenzo [c] phenanthridine alkaloids<sup>3,4,5)</sup> have also been reported. However, no report has so far been made on the synthesis of both types of alkaloids from a common intermediate. This communication deals with a novel and stereoselective synthesis of  $(\pm)$ -chelamine (7),  $(\pm)$ -chelidonine (8), sanguinarine (9), and dihydrosanguinarine (10) from coptisine (1), a protoberberine alkaloid, *via* the common and key intermediate (6) according to a biogenetic route, as a continuation of our synthetic studies on benzo [c] phenanthridine alkaloids.<sup>6</sup>



Fully Aromatized Benzo[C]phenanthridine

B/C Hexahydro Benzo[c]phenanthridine

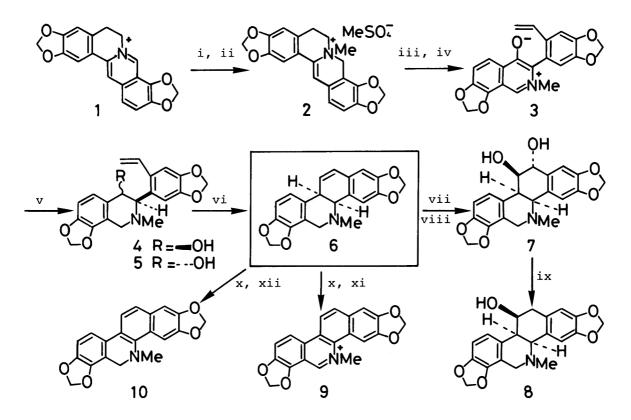
Reduction of coptisine (]) with lithium aluminium hydride in dry tetrahydrofuran, followed by methylation with dimethyl sulfate gave the methosulfate (2) [93%; mp 254-257 °C (dec.)]. The C<sub>6</sub>-N bond fission and introduction of oxygen function at the  $C_{1,3}$  position of 2 were realized by the Hofmann elimination, followed by oxidation with *m*-chloroperoxybenzoic acid to afford the betaine (3) [96%; mp 227-229 °C; δ 8.16, 7.26 (2H, AB-q, J=9), 7.63 (1H, s), 6.32 (1H, dd, J=17.5 and 11), 5.55 (1H, dd, J=17.5 and 1.2), 5.08 (1H, dd, J=11 and 1.2)]. The betaine (3) was reduced with sodium borohydride in refluxing methanol to furnish predominantly the *cis* alcohol (4) [80%; mp 172-175 °C; δ 4.38 (1H, br-s), 3.88 (1H, d, J=2.2)] along with the trans alcohol (5)[15%;  $\delta$  4.78 (1H, d, J=8), 3.61 (1H, d, J=8)]. Treatment of the *cis* alcohol (4) with concentrated sulfuric acid in acetic acid at room temperature effected stereoselective cationic cyclization to provide the benzo [c] phenanthridine (6) [98%; mp 144-145 °C;  $\delta$  6.36 (1H, dd, J=9.5 and 3.1), 5.78 (1H, br-d, J=9.5), 3.71 (1H, m), 3.41 (1H, br-d, J=5.1)]. The *cis*-fused stereochemistry of 6 was unambiguously ascertained from the coupling constant between the  $H_{4b}$  and  $H_{10b}$  (J=5.1)<sup>7</sup> in its <sup>1</sup>H-NMR spectrum. In the same manner, the trans alcohol (5) also gave 6 exclusively in 94% yield.

Upon treatment with peroxyformic acid in formic acid,<sup>8)</sup> the benzo[c]phenanthridine (6) underwent the *trans*-hydroxylation stereoselectively to yield ( $\pm$ )chelamine (7)[91%; mp 246-247 °C;  $\delta$  4.82 (1H, d, J=2.2), 4.08 (1H, m), 3.55 (1H, m), 3.30 (1H, t, J=2.2)]. The stereochemistry of 7<sup>9)</sup> was confirmed as depicted by spectral data and mechanistic consideration.<sup>8a)</sup> The structure of chelamine, isolated from *Corydalis majus*,<sup>10)</sup> was proposed to be 12-hydroxychelidonine<sup>10,11)</sup> and its stereochemistry has recently been clarified from spectral data.<sup>12)</sup> The above synthetic chelamine (7) was shown to be identical with natural chelamine by spectral comparison and thin-layer chromatographic behavior, therefore, the stereochemistry of chelamine is unambiguously established.

A hydroxy group at the  $C_{12}$  position in 7 was regioselectively removed with triethylsilane<sup>13)</sup> in the presence of boron trifluoride etherate in chloroform to produce (±)-chelidonine (8) [82%; mp 214-215 °C (lit.<sup>10)</sup> mp 215-216 °C)], which was identified with natural chelidonine by spectral comparison and thin-layer chromatographic behavior.

On the other hand, the benzo[c]phenanthridine (6) was dehydrogenated with 10% Pd-C in aqueous acetic acid in the presence of maleic acid to afford sanguinarine (9) [47%; mp 279-281 °C (lit.<sup>14a)</sup> mp 286-288 °C)] after treatment with concentrated hydrochloric acid. Dehydrogenation of 6 followed by sodium boro-hydride reduction provided dihydrosanguinarine (]0) [65%; mp 187-188 °C (lit.<sup>14b)</sup> mp 188-189 °C)]. The synthetic sanguinarine and dihydrosanguinarine were proved to be identical with the corresponding alkaloids.

Thus, we have succeeded in not only a highly stereoselective synthesis of  $(\pm)$ -chelamine and  $(\pm)$ -chelidonine, but also an alternative synthesis of sanguinarine and dihydrosanguinarine from coptisine *via* a common intermediate (6) according to a biogenetic route. Therefore, this method provides a general method for a synthesis of B/C *cis* hexahydrobenzo[*c*]phenanthridine alkaloids as well as fully aromatized benzo[*c*]phenanthridine alkaloids.



i: LiAlH<sub>4</sub>/THF; ii: Me<sub>2</sub>SO<sub>4</sub>/benzene; iii: 25%KOH/MeOH; iv: m-CPBA/CH<sub>2</sub>Cl<sub>2</sub>; v: NaBH<sub>4</sub>/MeOH; vi: c.H<sub>2</sub>SO<sub>4</sub>/AcOH; vii: HCO<sub>3</sub>H/HCO<sub>2</sub>H; viii: 20%aq.KOH/EtOH; ix: Et<sub>3</sub>SiH, BF<sub>3</sub>·OEt<sub>2</sub>/CHCl<sub>3</sub>; x: 10% Pd-C/aq.AcOH, maleic acid; xi: c.HCl; xii: NaBH<sub>4</sub>/MeOH

We are very grateful to Professor J. Slavík, J. E. Purkyne University, Czechoslovakia, for a generous supply of natural chelamine and chelidonine, and to Dr. S. Naruto, Dainippon Pharmaceutical Co., Ltd., for a generous gift of natural sanguinarine and dihydrosanguinarine.

## References

- E. Leete and S. J. B. Murrill, Phytochemistry, <u>6</u>, 231 (1967); A. Yagi, G. Nonaka, S. Nakayama, and I. Nishioka, ibid., <u>16</u>, 1197 (1977); A. R. Battersby, J. Staunton, H. C. Summers, and R. Southgate, J. Chem. Soc., Perkin Trans. 1, <u>1979</u>, 45; N. Takao, M. Kamigauchi, and M. Okada, Helv. Chim. Acta, <u>66</u>, 473 (1983); and references cited therein.
- 2) V. Šimánek, "The Alkaloids," ed by A. Brossi, Academic Press, New York, (1985), Vol. 26, p. 185; M. J. Hearn and S. L. Swanson, J. Heterocycl. Chem., <u>18</u>, 207 (1981); S. D. Phillilps and R. N. Castle, ibid., <u>18</u>, 223 (1981); and references cited therein.

- 3) (<u>+</u>)-Chelidonine: W. Oppolzer and C. Robbiani, Helv. Chim. Acta, <u>66</u>, 1119 (1983); M. Cushman, T.-C. Choong, J. T. Valko, and M. P. Koleck, J. Org. Chem., <u>45</u>, 5067 (1980).
- (<u>+</u>)-Homochelidonine: I. Ninomiya, O. Yamamoto, and T. Naito, J. Chem. Soc., Perkin Trans. 1, <u>1983</u>, 2171.
- 5) (<u>+</u>)-Corynoline and (<u>+</u>)-epicorynoline: I. Ninomiya, O. Yamamoto, and T. Naito, J. Chem. Soc., Perkin Trans. 1, <u>1980</u>, 212; M. Cushman, A. Abbaspour, and Y. P. Gupta, J. Am. Chem. Soc., <u>105</u>, 2873 (1983); J. R. Falck and S. Manna, ibid., 105, 631 (1983).
- 6) M. Hanaoka, T. Motonishi, and C. Mukai, J. Chem. Soc., Chem. Commun., <u>1984</u>, 718; M. Hanaoka, H. Yamagishi, M. Marutani, and C. Mukai, Tetrahedron Lett., <u>25</u>, 5169 (1984); M. Hanaoka, H. Yamagishi, and C. Mukai, Chem. Pharm. Bull., <u>33</u>, 1763 (1985); M. Hanaoka, S. Yoshida, and C. Mukai, Tetrahedron Lett., <u>26</u>, 5163 (1985).
- 7) I. Ninomiya, T. Naito, T. Kiguchi, and T. Mori, J. Chem. Soc., Perkin Trans. 1, 1973, 1696.
- 8) a) I. Ninomiya, O. Yamamoto, and T. Naito, J. Chem. Soc., Perkin Trans. 1, <u>1980</u>, 212; <u>1983</u>, 2165; b) M. Onda, H. Yamaguchi, and Y. Harigaya, Chem. Pharm. Bull., <u>28</u>, 866 (1980); c) G. Nonaka and I. Nishioka, ibid., <u>23</u>, 521 (1975).
- 9) The stereochemistry of 7 was further supported from <sup>1</sup>H-NMR spectrum of its diacetyl derivative [δ 6.23 (lH, d, J=8.8), 5.41 (lH, dd, J=8.8 and 4.9), 4.03 (lH, d, J=4.9), 3.65 (lH, t, J=4.9)].
- J. Slavík, L. Slavíková, and J. Brabenec, Coll. Czech. Chem. Commun., <u>30</u>, 3697 (1965).
- 11) J. Slavík and L. Slavíková, Coll. Czech. Chem. Commun., <u>42</u>, 2686 (1977).
- 12) The private communication form Prof. J. Slavík.
- 13) D. N. Kursanov, Z. N. Parnes, and N. M. Loim, Synthesis, 1974, 633.
- 14) a) M. Onda, K. Yonezawa, and K. Abe, Chem. Pharm. Bull., <u>19</u>, 31 (1971);
  b) C. Tani and N. Takao, Yakugaku Zasshi, <u>82</u>, 755 (1962).

(Received February 15, 1986)