

Synthesis of Ring B Unsaturated Estriols. Confirming the Structure of a Diagnostic Analyte for Smith–Lemli–Opitz Syndrome

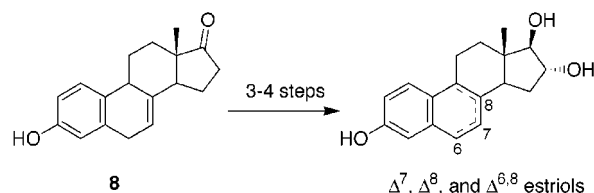
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ABSTRACT



Brief partial syntheses are described for ring B unsaturated estriols, which are candidate metabolites diagnostic for Smith–Lemli–Opitz syndrome prenatally. These steroids are also likely metabolites of the Premarin preparation used in estrogen replacement therapy. Equilin (8) was converted in three steps to 7-dehydroestriol, which was isomerized to 8-dehydroestriol. The simplicity of the transformations belies the lability of these previously inaccessible metabolites and their synthetic precursors.

The Smith–Lemli–Opitz syndrome¹ (SLOS) is an autosomal recessive disorder of cholesterol biosynthesis caused by a defect in the enzymatic conversion of 7-dehydrocholesterol to cholesterol (Scheme 1). This defect results in an abnormal accumulation of 7- and 8-dehydrocholesterol (1 and 2), which can be detected by GC/MS analysis of blood from affected individuals.² Prenatal detection is also important because SLOS is a serious birth defect characterized by mental retardation, multiple developmental malformations, and a high carrier frequency.³ Prenatal diagnosis presently necessitates amniocentesis or chorionic villus sampling.⁴ We are

developing less invasive diagnostic methods that target aberrant metabolites recently found in urine and serum during pregnancy.⁵ Validation of these methods requires authentic standards of estrogen metabolites unsaturated in ring B (e.g., 3 and 4). These dehydroestriols are also of interest as candidate metabolites of equine steroids contained in Premarin, which is widely used in estrogen replacement therapy.⁶

Equilin (8) and estriol (6) have been known since the seminal isolation studies of estrogens in the 1930s,⁷ but definitive syntheses of dehydroestriols have not been reported. Although an early patent describes the Birch reduc-

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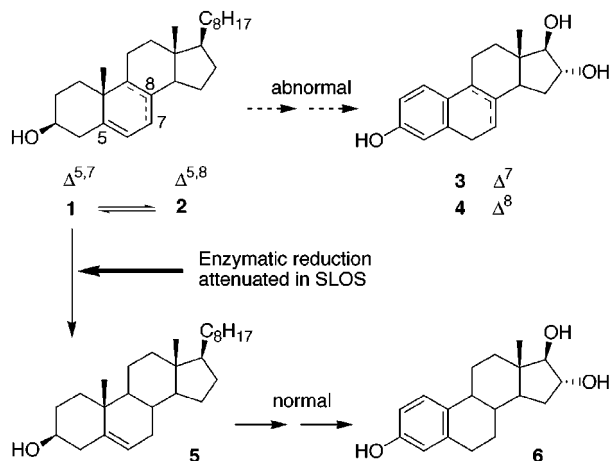
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Scheme 1. Metabolism of Fetal $\Delta^{5,7}$ and $\Delta^{5,8}$ Sterols to Estriols



tion of 6,8-didehydroestriol **15** to **3** (identified based on its UV spectrum),⁸ reductions of similar equilenins generally give mixtures of olefins.⁹ Other synthetic efforts have been thwarted by the facile aromatization of equilins.¹⁰ This Letter describes the first synthesis of homogeneous samples of dehydroestriols.

Our primary target was the preparation of compounds **3** and **4**. Considering the modest amounts of material required for bioanalytical purposes, we focused on partial synthesis from available estrogens, such as equilin, equilenin (**10**), and the Torgov diene.¹¹ In a standard synthetic approach to estriols, the 16-hydroxyl is introduced by acid hydrolysis of a 16 α ,17 α -epoxide formed from the enol acetate of estrone.¹² However, application of this method to equilin resulted in aromatization to 16 α -hydroxyequilenin.¹⁰ Another approach to 16-hydroxylation entails 16-bromination of estrone, followed by hydrolysis in DMF to the ketol.¹³ Although many unsaturated 17-ketosteroids can be selectively brominated at C-16 with CuBr₂ in refluxing methanol,^{13,14} this reaction was reported to give a complex mixture for equilin.¹⁰ Aiming to overcome these problems, we set out to devise conditions for implementing the simple ring D manipulations shown in the retrosynthetic analysis (Scheme 2) without triggering the indicated side reactions, namely, aromatization of ring B,¹⁵ epimerization at C-14,⁷ double-bond isomerization,^{7,16} and ketol rearrangement.^{13a}

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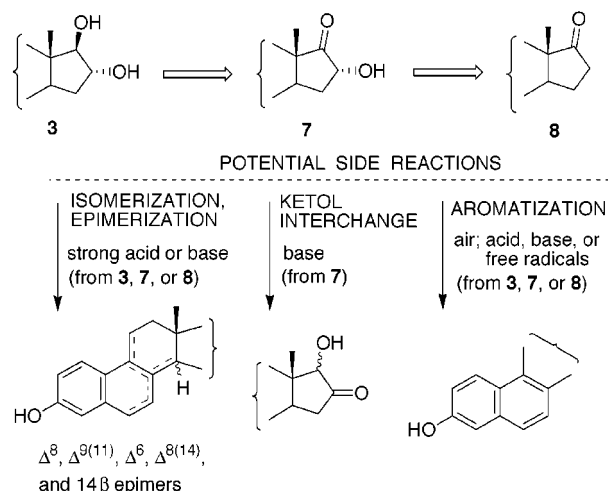
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Scheme 2



In our hands, bromination of equilin with CuBr₂ in MeOH¹⁰ also gave complex mixtures. 1D and 2D NMR analysis of standards and crude reaction mixtures led to identification of 10 bromosteroids, which, together with equilenin and equilin, accounted for >95% of the steroids observed in most reactions. Knowledge of reporter signals for the numerous bromosteroids facilitated byproduct identification and optimization of reaction conditions. Thus, use of 3 equiv of CuBr₂ in methanol¹³ led mainly to bromination in ring A, dehydrogenation to equilenins, and rapid conversion of the desired 16-bromosteroids **9a** and **9b** to dibromides and equilenins (Table 1, entries 1 and 2). Shorter reaction times and a smaller excess of CuBr₂ resulted in large amounts of unbrominated steroids (**8** and **10**) and low conversion to the desired products (Table 1, entries 3–5). In THF, the yield of 16-bromoequilins doubled to 21%, but the product still consisted mainly of equilenins and ring A brominated equilins (Table 1, entry 6). However, reaction in CHCl₃–EtOAc¹⁷ gave >70% conversion to 16-bromoequilins (Table 1, entry 7), and these conditions were sufficiently reproducible to afford gram quantities of the desired products as a 2:1 mixture of 16 α - and 16 β -bromo epimers. Under specific reaction conditions,^{13a} this mixture was cleanly hydrolyzed to 16 α -hydroxyequilin (**7**) without formation of 16-keto byproducts.¹⁸ Reduction of **7** with NaBH₄ led to the target 7-dehydroestriol (**3**).¹⁹

With the intention of synthesizing 8-dehydroestriol (**4**) by a parallel bromination–hydrolysis–reduction scheme, we prepared 8-dehydroestrone (**12**) by isomerizing equilin with LiNHCH₂CH₂NH₂ in ethylenediamine (Scheme 3).²⁰ However, refluxing **12** with CuBr₂ in CHCl₃–EtOAc resulted in virtually no bromination at C-16 or in ring A, the product consisting of a 1:1:2 mixture of **10**, **12**, and 9(11)-dehydroestrone. An alternative attempt to prepare **13** by LiNHCH₂CH₂NH₂ isomerization of **7** gave none of the expected products.²¹ However, isomerization of triol **3**, which lacks the potentially labile¹³ 16,17-ketol functionality of **7**, was

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Table 1. Bromination of Equilin (**8**) with CuBr₂: Effects of Reaction Conditions on Product Distribution^a

		Δ^7		$\Delta^{6,8}$	
8	R ₁ = H	R ₂ = H	10	R ₁ = H	R ₂ = H
9a	R ₁ = α Br	R ₂ = H	11a	R ₁ = α Br	R ₂ = H
9b	R ₁ = β Br	R ₂ = H	11b	R ₁ = β Br	R ₂ = H
9c	R ₁ = α Br	R ₂ = Br	11c	R ₁ = α Br	R ₂ = Br
9d	R ₁ = β Br	R ₂ = Br	11d	R ₁ = β Br	R ₂ = Br
9e	R ₁ = H	R ₂ = Br	11e	R ₁ = H	R ₂ = Br

entry	1	2	3	4	5	6	7
solvent	MeOH			THF		C-EA ^b	
CuBr ₂ ^c	3.0	3.0	1.6	1.6	1.6	1.0	1.2
t (min)	30	90	5	15	30	80	90
% 9a	10	0	6	9	10	16	48
% 9b	2	0	2	2	1	5	24
% 9c	9	4	8	5	4	3	0
% 9d	11	24	1	2	2	1	2
% 9e	2	4	0	0	0	0	1
% 11a	0	0	2	4	5	1	7
% 11b	0	0	0	0	1	1	3
% 11c	35	32	19	18	15	16	1
% 11d	21	30	1	2	2	5	4
% 11e	5	6	0	1	0	2	2
% 8	7	0	34	20	15	42	6
% 10	0	0	17	22	28	1	1

^a Product distributions were determined by ¹H NMR. Desired products (**9a** and **9b**) are highlighted. ^b Chloroform-ethyl acetate 1:1. ^c Molar ratio of CuBr₂ to **8**.

more promising. Despite the poor solubility of **3** and its sluggish rate of isomerization, reaction conditions were found to give **4** as the major product.^{22,23} Semipreparative reverse-

(18) Equilin (101 mg) and freshly ground CuBr₂ (166 mg) were heated in CHCl₃-EtOAc (25 mL each) for 2 h under vigorous reflux (to remove HBr). The crude product (140 mg; 48% **9a**, 24% **9b**) was stirred for 1.5 h at room temperature in DMF-water (3:1, total 10 mL) containing 2 equiv of NaOH. MPLC on silica gel (EtOAc-hexane 3:7) gave **7** (78 mg, containing some $\Delta^{6,8}$ material). Attempts to purify **7** by reverse-phase HPLC (MeOH-H₂O 35:65) gave a 19:1 mixture of **7** and its $\Delta^{6,8}$ analogue. Consequently, the equilenins formed during bromination and hydrolysis were removed after reduction of **7** to **3**. NMR (CDCl₃, 25 °C): **9a** δ 0.814 (s), 4.589 (d, 7.3 Hz), 5.449 (m); **9b** δ 0.995 (s), 4.263 (t, 8 Hz), 5.500 (m); **7** δ 0.865 (s), 4.420 (d, 8.3 Hz), 5.509 (m).

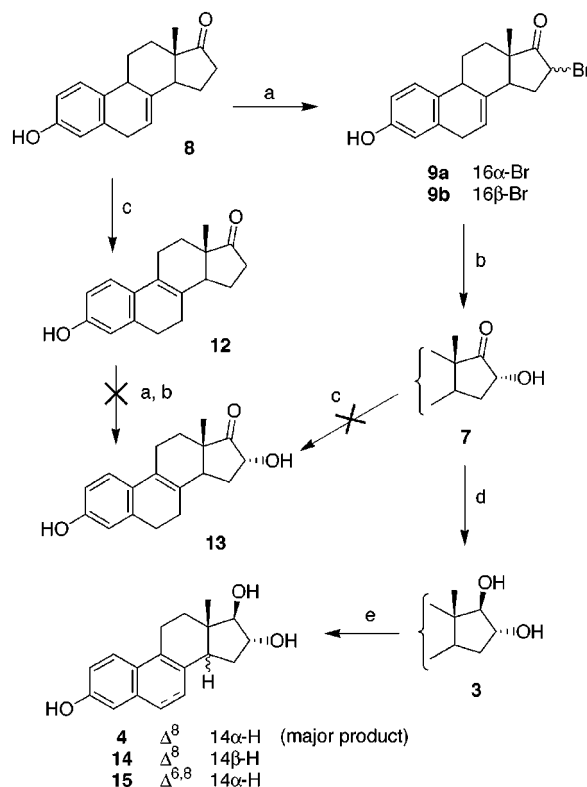
(19) Ketol **7** (446 mg) was reduced with NaBH₄ (47 mg) in MeOH (25 mL) for 2 h at 0 °C. Methanol was removed at <20 °C in a stream of N₂ (higher temperatures resulted in formation of **15**). Addition of cold saturated NH₄Cl (10 mL) followed by extraction with EtOAc gave **3** (452 mg, 81% purity). HPLC purification (250 × 21.2 mm C₁₈ column, MeOH-H₂O 45:55) of a 50-mg sample gave **3** (35 mg, 99% purity).

(20) Rajmakers, P. H. U.S. Patent 5,739,363, 1998.

(21) Reaction of **7** (50 mg) with 0.3 M Li in ethylenediamine (3 mL) gave mainly unreacted **7**, whereas 1 M Li led to a complex mixture.

(22) Dehydroestriol **3** (100 mg) was heated at 40 °C for 48 h in ethylenediamine (3.4 mL) containing LiNHCH₂CH₂NH₂ (prepared by adding 13.4 mmol of MeLi-LiBr in ether to ethylenediamine, followed by evaporation of the ether at 55 °C). NMR of the crude product (104 mg)

Scheme 3^a



^a (a) CuBr₂, CHCl₃-EtOAc 1:1, reflux; (b) NaOH, DMF-H₂O 3:1, rt, 1.5 h; (c) LiNHCH₂CH₂NH₂, NH₂CH₂CH₂NH₂, 30 °C, 1 h; (d) NaBH₄, MeOH, 0 °C, 2 h; (e) LiNHCH₂CH₂NH₂, NH₂CH₂-CH₂NH₂, 40 °C, 48 h.

phase HPLC afforded **4**, **14**, and **15**, which were characterized by 2D NMR and NOE difference spectroscopy to confirm the regio- and stereochemical structure assignments.

With availability of authentic samples of the dehydroestriols, we compared their GC mobilities and mass spectral fragmentation with those of the SLOS urinary metabolites²⁴

indicated a 9:5:3:3 mixture of **4**, **3**, **14**, and **15**. Preparative HPLC (250 × 21.2 mm C₁₈ column, MeOH-H₂O 35:65, 7 mL/min) gave **14** (*t_R* 146 min), **15** (*t_R* 159 min), **3** (*t_R* 165 min), and **4** (*t_R* 172 min). NMR (CD₃OD, 25 °C): **3** δ 0.636 (s), 3.578 (d, 5.4 Hz), 4.043 (ddd, 9.1, 5.4, 2.0 Hz), 5.358 (br d, 3.4 Hz); **4** δ 0.757 (s), 3.549 (d, 5.3 Hz), 4.119 (ddd, 9.0, 5.3, 1.8 Hz); **14** δ 0.946 (s), 3.560 (d, 6.9 Hz), 3.999 (ddd, 8.8, 7.9, 6.9 Hz); **15** δ 0.667 (s), 3.668 (d, 5.6 Hz), 4.228 (ddd, 9.1, 5.6, 2.1 Hz).

(23) The forcing isomerization conditions resulted in partial epimerization of **4** to **14** and prompted a thorough structure of determination of all products by NMR. In contrast, byproducts were negligible in the preparation of **12** (3% **3**, 1% 6-dehydroestrone, and 1–2% 14 β steroids) and are frequently absent in base-catalyzed olefin isomerizations: Pines, H.; Stalick, W. M. *Base-catalyzed reactions of hydrocarbons and related compounds*; Academic Press: New York, 1977; Chapter 2.

(24) **Isolation of dehydroestriols:** Urine from a pregnant woman carrying an SLOS fetus was processed by our standard methods for analyzing urinary steroids: Shackleton, C. H. L. *J. Steroid Biochem. Mol. Biol.* **1993**, *45*, 127–140. Briefly, steroid sulfates and glucuronides from a C₁₈ solid-phase extraction (SPE) of the urine sample were hydrolyzed with *Helix pomatia* (Roman snail) digestive juice (Sigma-Aldrich). The resulting unconjugated steroids were reextracted by SPE and fractionated on Sephadex LH-20 (100 × 10 mm column; cyclohexane-ethanol 4:1) as described: Setchell, K. D.; Shackleton, C. H. L. *Clin. Chim. Acta* **1973**, *47*, 381–388. GC/MS analysis of individual 5-mL fractions (as TMS ethers) revealed that dehydroestriol was eluted between 140 and 165 mL. The only two steroids found in this fraction were dehydroestriol and didehydroestriol **15**.

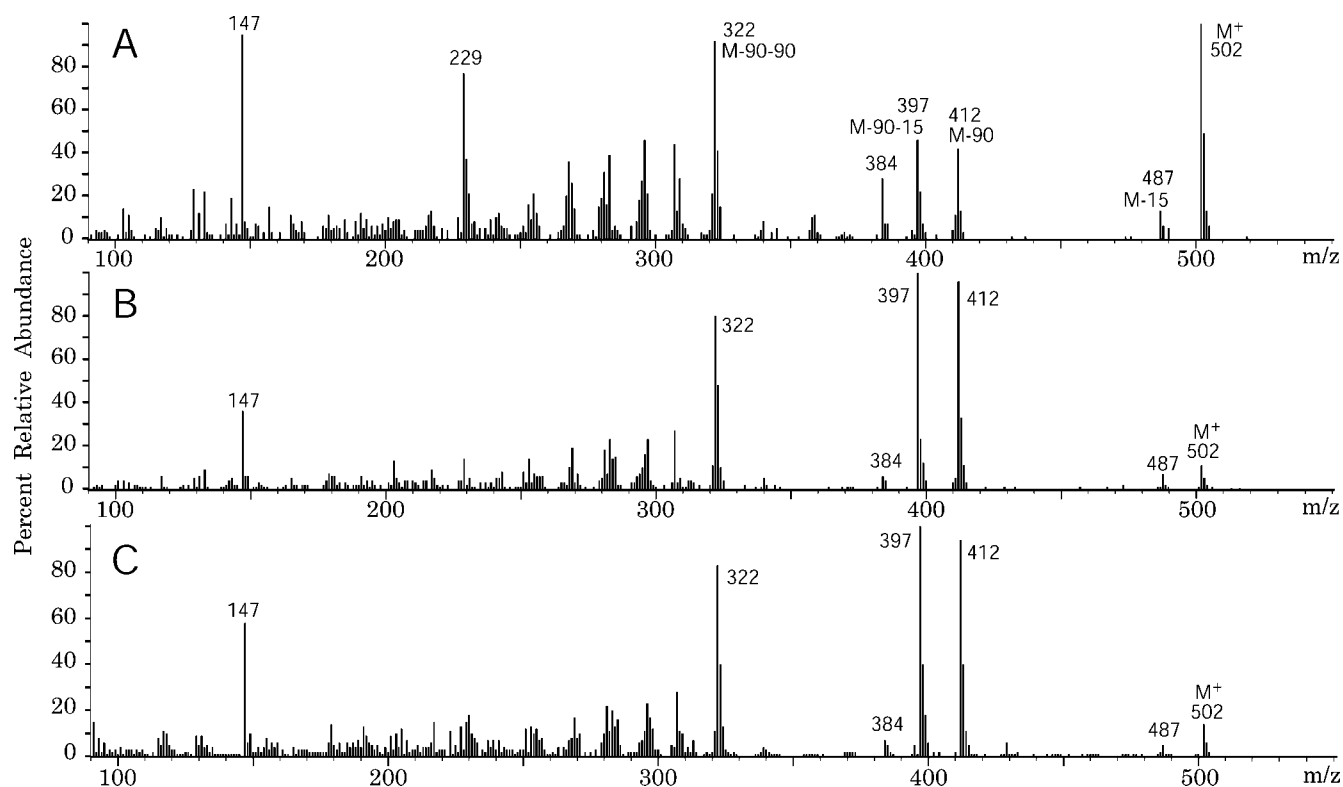


Figure 1. Comparison of GC/MS spectra of tris-TMS derivatives of authentic 7-dehydroestriol (A), 8-dehydroestriol (B), and the dehydroestriol isolated from urine (C). The molecular ions are at m/z 502, and major fragments are formed by losses of trimethylsilanol (-90) and methyl groups (-15). GC retention times for A, B, and C were 18.77 ± 0.03 min.

(Figure 1). Although the TMS ethers of 7- and 8-dehydroestriols coeluted on the nonpolar column used and shared the same parent and fragment ions, the isomers could be distinguished by the relative abundance of these ions. The steroid isolated from urine had the abundance profile of 8-dehydroestriol. This finding does not completely exclude production and excretion of 7-dehydroestriol by SLOS patients since only a few affected individuals have so far been studied. In addition, 7-dehydroestriol is less stable and may undergo aromatization to the didehydroestriol (**15**) found in urine.

In conclusion, we have developed simple and efficient methods for preparing estrogen metabolites unsaturated in ring B. The availability of these reference dehydroestriols

will facilitate the establishment of routine noninvasive prenatal diagnosis for SLOS.

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Supporting Information Available: Tables of NMR signal assignments, NOE difference results, HR-MS data, conformational analysis of **4** and **14**, and ^1H NMR spectra of **3**, **4**, **14**, and **15**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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