The Journal of Organic Chemistry

#### Article

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Bowen Xu, Yiping Gao, Jianwei Han, Zejing Xing, Sihan Zhao, Ziyang Zhang, Runlin Ren, and Limin Wang J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.9b01323 • Publication Date (Web): 13 Jun 2019 Downloaded from http://pubs.acs.org on June 13, 2019

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# Hypervalent Iodine(III) Mediated Tosyloxylation of 4-Hydroxycoumarins

Bowen Xu,<sup>†</sup> Yiping Gao,<sup>‡</sup> Jianwei Han,<sup>\*†</sup> Zejing Xing,<sup>†</sup> Sihan Zhao,<sup>†</sup> Ziyang Zhang,<sup>†</sup>

Runlin Ren,<sup>†</sup> and Limin Wang\*<sup>†</sup>

<sup>†</sup>Key Laboratory for Advanced Materials and Feringa Nobel Prize Scientist Joint Research Center, Institute of Fine Chemicals, School of Chemistry and Molecular Engineering, East China University of Science & Technology, 130 Meilong Road, Shanghai, 200237, China

<sup>‡</sup> Shanghai Xuhui Central Hospital, Shanghai 200031, P. R. China

Email: Jianwei Han - jianweihan@ecust.edu.cn; Limin Wang -

wanglimin@ecust.edu.cn

\* Corresponding author



up to 95% yield *sp*<sup>2</sup> C-H activation

#### Abstract

An efficient approach was developed for synthesis of 3-tosyloxy-4-hydroxycoumarins under mild conditions by using Koser's reagents. The reaction tolerated various functional groups and the products served as useful aromatic building blocks. Additionally, a plausible mechanism via iodonium ylide was proposed and the oral anticoagulant Warfarin was synthesized in good yield. Key words: Tosyloxylation, HTIB, Synthetic methods, Hypervalent iodine, Ylide.

## Introduction

Recent years have witnessed a rapid growth in the field of hypervalent iodine chemistry with exploring their novel reactivity in a diverse range of chemical transformations.<sup>1</sup> For example, hydroxy(tosyloxy)iodo]benzene (HTIB), which is known as Koser's reagent, is one of these useful reagents to perform  $\alpha$ -functionalization of ketones such as tosyloxylation, hydroxylation, and acyloxylation.<sup>2</sup> Following the seminal report that Koser et al. used HTIB to produce corresponding  $\alpha$ -tosyloxyketones from enolizable ketones and its first asymmetric version by Wirth,<sup>3</sup> much efforts were contributed to explore the activated aliphatic ketones as well as the catalytic and enantioselective version of these reactions (Scheme 1a).<sup>4</sup> Recently, Legault et al. reported an elegant study on the mechanistic insights of the  $\alpha$ -tosyloxylation of ketones with computational method, in which HTIB formed a novel iodonium intermediates as Lewis acids to enhance the electrophilicity of carbonyl compounds.<sup>5</sup> An experimental evidence supported this mechanism in the  $\alpha$ -tosyloxylation of enol esters by the same research group.<sup>6</sup> As such, HTIB involved tosyloxylations for transferring TsO-groups to the target products would be ideal. Considering that arylsulfonates can serve as excellent leaving groups for further coupling reactions, access to the aromatic compounds bearing OTs group is a topic of much interest. The tosyloxylations of aromatic substrates with HTIB is one highly desirable method. However, in sharp contrast, tosyloxylations of aromatics or heterocyclic rings of this type have rarely been Page 3 of 35

reported.<sup>7</sup> In fact, as early as 1978, Kappe et al. reported the reaction of 4-hydroxycoumarin with (diacetoxyiodo)benzenes (DIB) to afford iodonium ylides in good yields. Very recently, Panda et al. reported an efficient protocol for the synthesis of coumestans, in which the iodine(III) reagents generated from iodobenzene in situ by oxidants of *m*CPBA to afford the ylides.<sup>8</sup> The iodonium ylides, also called as zwitterionic iodonium compounds (ZICs),<sup>9</sup> were transformed into 4-aryloxy-3-iodocoumarins via a variation of Smiles rearrangement, which further gave coumestan derivatives (Scheme 1b).<sup>10</sup>

On the other hand, 4-hydroxycoumarins are important compounds in the synthesis of bioactive molecules such as warfarin, acenocoumarol and phenprocoumon, which are commonly prescribed as oral anticoagulants for prevention and treatment of thromboembolic disorders.<sup>11</sup> Additionally, sulfonyl group is a biologically active pharmacophore as a part of several clinically used drugs.<sup>12</sup> As a result, recently many researchers have shown interested in sulfonyl-containing coumarin derivatives. For examples, in 2016, Salar et al. reported a library of coumarin sulfonates by reacting hydroxycoumarin with different substituted sulfonyl chlorides, in which eleven compounds were demonstrated to have potential suppressive effect on production of reactive oxygen species. Later, coumarin derived sulfonyl esters were also found to be effective inhibitors of alkaline phosphatases.<sup>13</sup> However, to the best of our knowledge, there is no report concerning synthesis of 3-sulfonyl-4-hydroxycoumarins which might also be potentially bioactive.

(a) Previous reports on tosyloxylation by activating sp<sup>3</sup> C-H bond with HTIB

(b) Synthesis of iodonium ylides by iodine(III) reagents<sup>8</sup>



(c) Tosyloxylation with HTIB via iodonium ylides (this work)



#### Scheme 1. Tosyloxylation with HTIB.

Given our continued interest in the hypervalent iodine chemistry,<sup>14</sup> we recently synthesized iodonium salts that incorporated vicinal groups for exploring their unique reactivity.<sup>14a</sup> When the reactivity pattern was attempted to prepare heterocyclic iodonium salts bearing vicinal OTf groups, the tosyloxylation of 4-hydroxycoumarins proceeded smoothly without phenyl migration by thermal rearrangement of iodonium ylides. Herein we reported the detailed results in the preparation of 3-sulfonyl substituted 4-hydroxycoumarins (Scheme 1c).

#### **Results and Discussion**

Preliminarily, treatment of 4-hydroxycoumarin (1a) with 0.4 equivalent HTIB (2a) as model reaction in dichloromethane (DCM) at room temperature gave 40% yield of **3aa** after 3 hours (Table 1, entry 1). In order to improve the yield of **3aa**, we found that the conversion of **1a** afforded **3aa** in 70% or 77% yield respectively when 1 or 1.2 equivalent HTIB was used. Further increasing the amount of **2a** did not result in a

better yield (Table 1, entries 2-4). Then, the effect of solvent was also examined and the results were shown in Table 1. Reaction with toluene or MeCN as solvent furnished lower yields at 46% and 52% respectively. Almost no reaction took place in DMF and the reaction mixture changed into dark color at the moment when the solvent was added (Table 1, entries 5-7). To our delight, the elevated temperature of 35 °C increased the yield to 85%, while higher temperature cannot result in a better result (Table 1, entries 8-9). An investigation of reaction time indicated that the reaction can finish in one hour (Table 1, entries 10-11).

Table 1. Reaction optimization<sup>a</sup>



Entry	Eqv. of <b>2a</b>	Solvent	Temp.(°C)	Time(h)	Yield (%)
1	0.4	DCM	r.t.	3	40
2	1	DCM	r.t.	3	70
3	1.2	DCM	r.t.	3	77
4	2	DCM	r.t.	3	77
5	1.2	Tol	r.t.	3	46
6	1.2	MeCN	r.t.	3	52
7	1.2	DMF	r.t.	3	-
8	1.2	DCM	35	3	85
9	1.2	DCM	reflux	3	86
10	1.2	DCM	35	1	85
11	1.2	DCM	35	0.5	36

<sup>*a*</sup> Reaction conditions: **1a** (0.3 mmol), **2a** and solvents (10 mL).

Next, the substrate scope of tosyloxylation of 4-hydroxycoumarin (1a) with various Koser's reagents was explored under the optimized conditions. As shown in Table 2,

the 3-position tosyloxylation of 1a with Koser's reagents including meta- or para-halogen substituted aryl motifs afforded the corresponding products 3ab-3ae in 71-72% yields by the transfer of arylsulfonyl groups (Table 2, entries 1-4). A moderate yield of 68% for **3af** was achieved with **2f** bearing phenyl group (Table 2, entry 5). Koser's reagent of 2g with 2,4-dinitroaryl unit resulted in the desired product (3ag) in an acceptable yield of 46%. Regarding Koser's reagents of 2h-2k contained electron-withdrawing groups such as nitro-, trifluoromethoxy, trifluoromethyl and nitrile groups, the reaction afforded **3ah-3ak** in 54-75% yields (Table 2, entries 6-10). Furthermore, when the Koser's reagents of **2l-2m** with electron-donating groups were used in this protocol, the corresponding products of 3al and 3am were isolated in excellent yields of 85% and 90% respectively, which suggested HTIBs bearing electron-donating group favor this reaction (Table 2, entries 11-12). Of note, the methylsulfonyl group can be transferred to give **3an** in 43% yield when **2n** was employed. Additionally, hydroxyl-(phenyl)-iodanyl naphthalene-1-sulfonate (21) was also well tolerated in this reaction. It was worth to mention that **3ap** with a camphor sulfonate motif was afforded in 59% yield with this procedure (Table 2, entry 15).

OH OH OH Ia	0 0 Ph ⁺ R <sup>-S</sup> 0 <sup>-I</sup> OH <b>2b-p</b>	DCM 35 °C, 1 h	OH OSR OOO
Entry	R	Product	Yield (%)
1	$4-FC_6H_4$	3ab	71
2	$4-ClC_6H_4$	3ac	72
3	$4-BrC_6H_4$	3ad	71
4	$3-ClC_6H_4$	3ae	72
5	C <sub>6</sub> H <sub>5</sub>	3af	68
6	2,4-diNO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	3ag	46
7	$4-OCF_3C_6H_4$	3ah	54
8	$4-CF_3C_6H_4$	3ai	72
9	$4-CNC_6H_4$	3aj	75
10	$4-NO_2C_6H_4$	3ak	72
11	$4-^{t}BuC_{6}H_{4}$	3al	85
12	$4-OMeC_6H_4$	3am	90
13	Me	3an	43
14	2-Naphthalene	<b>3</b> ao	59
15	10-Camphor	Зар	55

#### Table 2. Scope of diverse HTIB in tosyloxylation<sup>a</sup>

<sup>a</sup> Reaction conditions: **1a** (0.3 mmol), **2a** (0.36 mmol) and DCM (10 mL) at 35 °C for 1 hour.

We successively examined the structural diversity of a variety of 4-hydroxycoumarin derivatives by changing the substituted groups on the molecular skeleton. As shown in Table 3, both 6- and 7- position substituted 4-hydroxycoumarins with halogens, methyl and methoxy groups were well tolerated in the tosyloxylation reaction, **3ba-3ja** were furnished in 38-82% yields. Of note, the reaction of 7-dimethylamine-4-hydroxycoumarin **1k** with HTIB afforded the product **3ka** in 64% yield. The compound of **3ka** emitted strong fluorescence by irradiation of ultraviolet light because of its D- $\pi$ -A structure. Interestingly, methyl substituted derivatives **1l-1m** accomplished the tosyloxylated products in good yields of 77-80%.

Moreover, the tosyloxylated benzo-4-hydroxycoumarin derivative **3na** was synthesized in 78% yield under the standard conditions with this protocol. Furthermore, 4-hydroxy-1-methylquinolinone and 4-hydroxy-6-methyl-pyranone were adopted in this reaction, the corresponding products of **3oa** and **3pa**<sup>15</sup> were obtained in 72% and 95% yields, respectively.





<sup>*a*</sup> Reaction conditions: **1a** (0.3 mmol), **2a** (0.36 mmol) and DCM (10 mL) at 35 °C for 1 hour. <sup>*b*</sup> DCM (3 mL) was used.

Next, to show the potential utility of target products, an attempt to convert the tosyloxylation products into aromatic building blocks was described in Scheme 2. In consideration of the benzenesulfonyl group as an excellent leaving group (LG), 3aa was tried in the coupling reactions, however, **3aa** did not exhibit any reactivity due to the existence of ortho-carbonyl and hydroxyl groups. An alternative strategy to protect the hydroxyl with trifluoromethanesulfonic anhydride ( $Tf_2O$ ) was employed to give product 4 in 92% yield. Taking advantage of the leaving group of OTf, several synthetic manipulations to transform 4 into functional molecules 5-8 were carried out. Of note, biologically interesting molecule of vitamin E was directly coupled with 4 in the presence of  $K_2CO_3$  as base; the corresponding conjugated compound 5 was formed in 42% yield. Additionally, the transition metal free coupling reaction of 4 under strong basic conditions using LiHMDS was attempted with diphenylphophane, para-toluenethiol and furan, respectively, the corresponding products 6-8 were obtained in 52-63% yields, which suggested 4 was an excellent coumarin-based motif to prepare the related target molecules. For newly-formed benzenesulfonyl group as leaving group (LG), further transformation of 6-8 with general coupling reactions of Suzuki coupling and others, unfortunately, no positive result was achieved so far, which may be due to the steric hindrance in the ortho-positions.



Scheme 2. Versatility of 3aa in further transformations. a)  $Tf_2O$ ,  $Et_3N$ , DCM. b)  $K_2CO_3$ , Vitamin E, DMF, 100 °C. c) Furan, LiHDMs, Tol, r.t.. d) 4-Methylbenzenethiol, LiHDMs, Tol, 110 °C. e) Ph<sub>2</sub>PH, LiHDMs, Tol, 110 °C. DMF = *N*,*N*-dimethylformamide, LiHDMs = Lithium bis(trimethylsilyl)amide, Tol = toluene.

Control experiments were also performed to clarify the mechanism of this transformation. Coumarin (9) and 4-tosyl substituted coumarin (10) were unable to form the corresponding products under the standard conditions (Scheme 3a & 3b). A low temperature reaction afforded the intermediate 11 in 83% isolated yield, which was isolated and determined by NMR and HRMS.<sup>9</sup> Furthermore, the iodonium ylide 11 can afford 3aa in 80% yield in the presence of TsOH (Scheme 3c). Interestingly, the oral anticoagulant Warfarin was synthesized by using the intermediate of 11. Warfarin 12 was synthesized in one step with 76% yield, which provides an alternative way to afford this bioactive molecule without catalyst or additional promotors.

As such, a plausible mechanism was proposed in scheme 4, of 4-hydroxycoumarin in enol-keto tautomerism attacked the positive iodine center of Koser's reagent to give **13**, which formed iodonium ylide **11** by leaving of TsOH. **11** was easily transformed into **3aa** with the anion of OTs in the solvent.



Scheme 3 Control experiments and synthesis of warfarin



Scheme 4 Plausible mechanism

#### Conclusion

In summary, we have developed an alternative method for 3-position tosyloxylation of 4-hydroxycoumarin derivatives via  $sp^2$  C-H bond activation. The current approach can tolerate various substrates under mild reaction condition without using transition-metal catalyst. Furthermore, the oral anticoagulant Warfarin was synthesized with this procedure, which demonstrated the potential applications of this method in organic synthesis.

#### **Experimental Section**

General Information. (a) Methods: <sup>1</sup>H, <sup>13</sup>C spectra were recorded on a Bruker AVANCE 400 spectrometer, operating at 400 MHz for <sup>1</sup>H NMR and 100 MHz for <sup>13</sup>C NMR respectively. Chemical shifts ( $\delta$ ) are reported in ppm, and coupling constants (J) are in Hertz (Hz). The following abbreviations were used to explain the multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad; Mass spectra were in general recorded on a Waters LCT Premier XE spectrometer; Column chromatography was performed with silica gel (200-300 mesh ASTM). All solvents were dried and/or distilled by standard methods. (b) Materials: All solvents were dried and/or distilled by standard methods. All reagents were purchased from commercial sources and used without further purification. The [hydroxy(sulfonyloxy)iodo]arenes and the substituted 4-hydroxy-2H-chromen-2-one were synthesized according to the literature procedures <sup>16-18,4c</sup> or commercially available and were used as received. The preparation of all other materials is described in detail below.

#### Procedure for Synthesis of 3-tosylated 4-hydroxy-2H-chromen-2-one:

To a 50 ml round-bottomed flask was added the substituted 4-hydroxy-2*H*-chromen-2-one **1** (0.3 mmol) and [hydroxy(sulfonyloxy)iodo]arenes **2** (0.36 mmol). DCM (10 ml) was added and the mixture was stirred at 35 °C under oil bath for 1 hour. Then the solvent was evaporated under vacuum. The crude products were purified using flash column chromatography on silica gel to afford the desired product.

**4-Hydroxy-2-oxo-***2H***-chromen-3-yl 4-methylbenzenesulfonate** (**3aa**). Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 85 mg (85%) of **3aa** as white solid. For a gram-scale experiment, following the general procedure, the reaction of **1a** (1.62 g, 10 mmol), **2a** (4.7 g, 12 mmol) and DCM (100 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 2.72 g (82%) of **3aa** as white solid. M.P.: 155-156 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.99 (d, *J* = 8.4 Hz, 1H), 7.92 (d, *J* = 8.2 Hz, 2H), 7.72 (t, *J* = 7.8 Hz, 1H), 7.51 (d, *J* = 8.2 Hz, 2H), 7.49 – 7.42 (m, 2H), 2.49 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  157.8, 157.6, 151.0, 145.5, 132.8, 129.8, 128.4, 124.5, 124.0, 116.6, 116.5, 116.4, 21.2. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>12</sub>NaO<sub>6</sub>S [M+Na]<sup>+</sup> 355.0252, found 355.0253.

**4-Hydroxy-2-oxo-***2H***-chromen-3-yl 4-fluorobenzenesulfonate (3ab).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2b** (143 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 72 mg (71%) of **3ab** as white solid. M.P.: 95-96 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.18 – 8.08 (m, 2H), 7.93 (dd, *J* = 7.7, 1.2 Hz, 1H), 7.64 – 7.58 (m, 1H), 7.55 – 7.44 (m, 2H), 7.35 (t, *J* = 8.0 Hz, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  165.3(d, *J*<sub>C-F</sub> = 252 Hz), 160.5, 158.3, 151.4, 132.9, 132.1, 131.6(d, *J*<sub>C-F</sub> = 10 Hz), 124.4, 123.8, 118.6, 116.6, 116.4(d, *J*<sub>C-F</sub> = 23 Hz), 116.2. HRMS m/z (ESI-TOF): calculated for C<sub>15</sub>H<sub>8</sub>O<sub>6</sub>SF [M-H]<sup>-</sup> 335.0026, found 335.0024.

general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2c** (148 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 76 mg (72%) of **3ac** as white solid. M.P.: 143-145 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.04 – 7.97 (m, 2H), 7.94 (dd, J = 8.2, 1.3 Hz, 1H), 7.77 – 7.70 (m, 2H), 7.70 – 7.62 (m, 1H), 7.39 (t, J = 7.6 Hz, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  158.8, 157.8, 151.1, 139.6, 134.9, 132.6, 130.3, 129.5, 124.3, 124.2, 117.1, 116.7, 116.4. HRMS m/z (ESI-TOF): calculated for C<sub>15</sub>H<sub>8</sub>O<sub>6</sub>SC1 [M-H]<sup>-</sup> 350.9730, found 350.9736.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl 4-bromobenzenesulfonate** (**3ad**). Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2d** (164 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 84 mg (71%) of **3ad** as white solid. M.P.: 154-155 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.92 (t, *J* = 8.3 Hz, 3H), 7.85 (d, *J* = 8.3 Hz, 2H), 7.60 (t, *J* = 7.7 Hz, 1H), 7.39 – 7.28 (m, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  160.3, 158.2, 151.3, 135.9, 132.3, 130.2, 128.5, 126.7, 124.3, 123.9, 118.4, 116.6, 116.3. HRMS m/z (ESI-TOF): calculated for C<sub>15</sub>H<sub>8</sub>O<sub>6</sub>SBr [M-H]<sup>-</sup> 394.9225, found 394.9233.

**4-Hydroxy-2-oxo-***2H***-chromen-3-yl 3-chlorobenzenesulfonate (3ae).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2e** (148 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 76 mg (72%) of **3ae** as white solid. M.P.: 147-148 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.05 (s, 1H), 7.95 (dd, *J* = 6.3, 2.1 Hz, 2H), 7.90 (dd, *J* = 8.1, 1.1 Hz, 1H), 7.68 (dd, *J* 

= 15.4, 7.5 Hz, 2H), 7.41 (t, J = 7.2 Hz, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  158.6, 157.8, 151.1, 137.8, 134.6, 133.8, 132.7, 131.3, 127.8, 127.1, 124.3, 124.2, 116.7, 116.5. HRMS m/z (ESI-TOF): calculated for C<sub>15</sub>H<sub>8</sub>O<sub>6</sub>SCl [M-H]<sup>-</sup> 350.9730, found 350.9736.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl benzenesulfonate (3af).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2f** (136 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 65 mg (68%) of **3af** as white solid. M.P.: 139-140 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.03 – 7.97 (m, 2H), 7.96 – 7.91 (m, 1H), 7.80 (t, *J* = 7.5 Hz, 1H), 7.66 (t, *J* = 7.8 Hz, 3H), 7.39 (t, *J* = 8.1 Hz, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  158.1, 157.7, 151.1, 136.0, 134.6, 132.7, 129.3, 128.3, 124.3, 124.1, 116.7, 116.4. HRMS m/z (ESI-TOF): calculated for C<sub>15</sub>H<sub>9</sub>O<sub>6</sub>S [M-H]<sup>-</sup> 317.0120, found 317.0120.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl 2,4-dinitrobenzenesulfonate (3ag).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2g** (140 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 56 mg (46%) of **3ag** as colourless oil. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.77 (d, *J* = 2.8 Hz, 1H), 8.34 (dd, *J* = 9.4, 2.8 Hz, 1H), 7.83 (dd, *J* = 7.7, 1.3 Hz, 1H), 7.53 – 7.45 (m, 1H), 7.22 (dd, *J* = 8.3, 6.7 Hz, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  159.1, 157.3, 152.3, 139.6, 137.7, 130.4, 129.0, 124.6, 122.5, 121.3, 117.5, 117.4, 115.9. HRMS m/z (ESI-TOF): calculated for C<sub>15</sub>H<sub>7</sub>N<sub>2</sub>O<sub>10</sub>S [M-H]<sup>-</sup> 406.9821, found 406.9821.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl 4-(trifluoromethoxy)benzenesulfonate (3ah).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2h** (167 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 65 mg (54%) of **3ah** as white solid. M.P.: 171-172 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.19 – 8.10 (m, 2H), 7.95 (dd, *J* = 8.2, 1.5 Hz, 1H), 7.73 – 7.60 (m, 3H), 7.44 – 7.38 (m, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  158.4, 157.7, 152.4, 151.1, 134.7, 132.8, 131.3, 124.4, 124.1, 121.2, 119.8(q, *J*<sub>C-F</sub> = 257 Hz), 116.7, 116.7, 116.5. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>9</sub>O<sub>7</sub>SF<sub>3</sub>K [M+K]<sup>+</sup> 440.9659, found 440.9658.

**4-Hydroxy-2-oxo-***2H***-chromen-3-yl 4-(trifluoromethyl)benzenesulfonate** (3ai). Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2h** (161 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 83 mg (72%) of **3ai** as white solid. M.P.: 186-187 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.24 (d, *J* = 8.3 Hz, 2H), 8.04 (d, *J* = 8.4 Hz, 2H), 7.91 (dd, *J* = 7.8, 1.3 Hz, 1H), 7.66 – 7.55 (m, 1H), 7.34 (t, *J* = 8.3 Hz, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  160.6, 158.2, 151.4, 140.8, 133.58(d, *J*<sub>C-F</sub> = 32 Hz), 132.1, 129.3, 126.3(q, *J*<sub>C-F</sub> = 4 Hz), 124.4, 123.8, 123.4(d, *J*<sub>C-F</sub> = 272 Hz), 118.7, 116.7, 116.3. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>9</sub>O<sub>6</sub>SF<sub>3</sub>Na [M+Na]<sup>+</sup> 408.9970, found 408.9980.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl 4-cyanobenzenesulfonate (3aj).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2j** (145 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 77

mg (75%) of **3aj** as white solid. M.P.: 167-168 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$ 8.23 – 8.09 (m, 4H), 8.00 – 7.89 (m, 1H), 7.71 – 7.59 (m, 1H), 7.37 (t, J = 8.1 Hz, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  159.2, 157.9, 151.2, 140.3, 133.3, 132.6, 129.1, 124.2, 124.2, 117.5, 117.3, 116.8, 116.7, 116.4. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>8</sub>NO<sub>6</sub>S [M-H]<sup>-</sup> 342.0072, found 342.0077.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl 4-nitrobenzenesulfonate (3ak).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2k** (152 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 78 mg (72%) of **3ak** as white solid. M.P.: 132-133 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.44 (d, *J* = 8.9 Hz, 2H), 8.30 (d, *J* = 8.8 Hz, 2H), 7.91 (d, *J* = 7.6 Hz, 1H), 7.60 (t, *J* = 7.2 Hz, 1H), 7.33 (t, *J* = 8.8 Hz, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  161.1, 158.3, 151.5, 150.5, 142.4, 132.0, 130.0, 124.4, 124.3, 123.7, 119.0, 116.7, 116.2. HRMS m/z (ESI-TOF): calculated for C<sub>15</sub>H<sub>8</sub>NO<sub>8</sub>S [M-H]<sup>-</sup> 361.9971, found 361.9978.

4-Hydroxy-2-oxo-2*H*-chromen-3-yl 4-(tert-butyl)benzenesulfonate (3al). Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2l** (156 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 95 mg (85%) of **3al** as white solid. M.P.: 75-77 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 7.88 (dd, J = 12.4, 5.1 Hz, 3H), 7.60 (dd, J = 10.9, 5.0 Hz, 3H), 7.39 – 7.20 (m, 2H), 1.25 (s, 9H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 157.9, 157.6, 157.6, 151.0, 133.0, 132.8, 128.3, 126.2, 124.5, 124.0, 116.8, 116.5, 116.2, 35.1, 30.6. HRMS m/z (ESI-TOF): calculated for C<sub>19</sub>H<sub>18</sub>O<sub>6</sub>SNa [M+Na]<sup>+</sup> 397.0722, found 397.0732.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl 4-methoxybenzenesulfonate (3am).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2m** (147 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 94 mg (90%) of **3am** as white solid. M.P.: 153-155 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.98 – 7.81 (m, 3H), 7.72 – 7.61 (m, 1H), 7.41 (t, *J* = 7.5 Hz, 2H), 7.25 – 7.10 (m, 2H), 3.88 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  163.9, 157.7, 157.6, 151.0, 132.7, 130.9, 126.9, 124.4, 124.0, 116.6, 116.4, 114.5, 55.9. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>11</sub>O<sub>7</sub>S [M-H]<sup>-</sup> 347.0225, found 347.0229.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl methanesulfonate (3an).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2n** (114 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 33 mg (43%) of **3an** as white solid. M.P.: 153-155 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.98 (dd, *J* = 7.9, 1.2 Hz, 1H), 7.72 – 7.65 (m, 1H), 7.42 (dd, *J* = 12.0, 4.3 Hz, 2H), 3.48 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  158.5, 158.1, 151.0, 132.6, 124.3, 124.1, 117.2, 117.0, 116.4, 39.7. HRMS m/z (ESI-TOF): calculated for C<sub>10</sub>H<sub>7</sub>O<sub>6</sub>S [M-H]<sup>-</sup> 254.9963, found 254.9963.

**4-Hydroxy-2-oxo-2***H***-chromen-3-yl naphthalene-2-sulfonate (3ao).** Following the general procedure, the reaction of **1a** (49 mg, 0.3 mmol), **2o** (154 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 65

mg (59%) of **3ao** as white solid. M.P.: 116-117 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$ 8.68 (s, 1H), 8.24 – 8.14 (m, 3H), 8.12 – 8.04 (m, 1H), 7.82 (dd, J = 7.8, 1.6 Hz, 1H), 7.78 – 7.72 (m, 1H), 7.71 – 7.64 (m, 1H), 7.46 – 7.37 (m, 1H), 7.20 – 7.12 (m, 2H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  165.7, 159.6, 152.1, 135.7, 134.6, 131.4, 130.4, 129.4, 129.0, 128.9, 128.6, 127.8, 127.3, 124.8, 123.7, 123.4, 122.3, 116.4, 115.7. HRMS m/z (ESI-TOF): calculated for C<sub>19</sub>H<sub>11</sub>O<sub>6</sub>S [M-H]<sup>-</sup> 367.0276, found 367.0276.

4-Hydroxy-2-oxo-2H-chromen-3-yl((1R,2S,4R)-7,7-dimethyl-6-oxobicyclo[2.2.1]h eptan-2-yl) methanesulfonate (3ap). Following the general procedure, the reaction of 1a (49 mg, 0.3 mmol), 2p (163 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 65 mg (55%) of **3ap** as white solid. M.P.: 156-157 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.01 (dd, *J* = 8.1, 1.6 Hz, 1H), 7.71 - 7.54 (m, 1H), 7.45 - 7.28 (m, 2H), 3.89 (q, J = 15.1 Hz, 2H), 2.47 - 2.28(m, 2H), 2.10 (t, J = 4.4 Hz, 1H), 1.98 (d, J = 18.4 Hz, 2H), 1.61 (ddd, J = 13.6, 9.4, 4.6 Hz, 1H), 1.51 – 1.38 (m, 1H), 1.08 (s, 3H), 0.87 (s, 3H). A signal for OH-proton is not visible.  ${}^{13}C{}^{1}H$  NMR (100 MHz, DMSO- $d_6$ )  $\delta$  213.6, 160.5, 159.0, 151.3, 131.9, 124.4, 123.8, 118.9, 117.3, 116.2, 57.8, 54.9, 49.8, 47.7, 42.2, 41.9, 26.2, 24.9, 19.2. HRMS m/z (ESI-TOF): calculated for C<sub>19</sub>H<sub>19</sub>O<sub>7</sub>S [M-H]<sup>-</sup> 391.0860, found 391.0852. 6-Fluoro-4-hydroxy-2-oxo-2*H*-chromen-3-yl 4-methylbenzenesulfonate (**3ba**). Following the general procedure, the reaction of 1b (54 mg, 0.3 mmol), 2a (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 60 mg (57%) of **3ba** as white solid. M.P.: 89-90 °C. <sup>1</sup>H NMR (400

MHz, DMSO- $d_6$ )  $\delta$  7.87 (d, J = 8.3 Hz, 2H), 7.45 (dd, J = 9.0, 3.1 Hz, 1H), 7.33 (d, J = 8.0 Hz, 2H), 7.29 – 7.20 (m, 1H), 7.16 (dd, J = 8.9, 4.4 Hz, 1H), 2.34 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  163.7, 159.2, 157.7(d,  $J_{C-F} = 237$  Hz), 148.2, 143.8, 135.3, 129.2, 128.1, 117.8(d,  $J_{C-F} = 6$  Hz), 116.5, 109.8(d,  $J_{C-F} = 24$  Hz), 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>10</sub>O<sub>6</sub>SF [M-H]<sup>-</sup> 349.0182, found 349.0178.

**6-Chloro-4-hydroxy-2-oxo-2***H***-chromen-3-yl <b>4-methylbenzenesulfonate** (**3ca**). Following the general procedure, the reaction of **1c** (59 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 75 mg (68%) of **3ca** as white solid. M.P.: 97-98 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.87 (d, *J* = 8.3 Hz, 2H), 7.68 (d, *J* = 2.7 Hz, 1H), 7.39 (dd, *J* = 8.7, 2.7 Hz, 1H), 7.32 (d, *J* = 8.0 Hz, 2H), 7.13 (d, *J* = 8.7 Hz, 1H), 2.34 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  164.2, 159.2, 150.7, 143.7, 135.5, 130.0, 129.1, 128.0, 126.7, 125.1, 123.9, 117.9, 116.3, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>10</sub>O<sub>6</sub>SCI [M-H]<sup>-</sup> 364.9887, found 364.9895.

**6-Bromo-4-hydroxy-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate (3da).** Following the general procedure, the reaction of **1d** (72 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 80 mg (65%) of **3da** as white solid. M.P.: 105-106 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.94 (d, J = 2.5 Hz, 1H), 7.92 (d, J = 8.2 Hz, 2H), 7.65 (dd, J = 8.7, 2.5 Hz, 1H), 7.41 (d, J = 8.2 Hz, 2H), 7.21 (d, J = 8.7 Hz, 1H), 2.42 (s, 3H). A

signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  158.7, 150.8, 144.2, 134.8, 133.5, 129.3, 128.1, 126.8, 118.4, 116.5, 115.0, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>10</sub>O<sub>6</sub>SBr [M-H]<sup>-</sup> 408.9381, found 408.9385.

**4-Hydroxy-6-methyl-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate (3ea).** Following the general procedure, the reaction of **1e** (53 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 85 mg (82%) of **3ea** as white solid. M.P.: 145-146 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.91 (d, *J* = 8.3 Hz, 2H), 7.77 (s, 1H), 7.50 (d, *J* = 8.1 Hz, 3H), 7.33 (d, *J* = 8.4 Hz, 1H), 2.48 (s, 3H), 2.43 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  158.1, 157.8, 149.2, 145.3, 133.7, 133.4, 133.1, 129.7, 128.4, 123.7, 116.7, 116.5, 116.2, 21.2, 20.3. HRMS m/z (ESI-TOF): calculated for C<sub>17</sub>H<sub>13</sub>O<sub>6</sub>S [M-H]<sup>-</sup> 345.0433, found 345.0428.

**4-Hydroxy-6-methoxy-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate** (**3fa**). Following the general procedure, the reaction of **1f** (58 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 54 mg (50%) of **3fa** as white solid. M.P.: 144-145 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.89 (d, *J* = 8.3 Hz, 2H), 7.43 (d, *J* = 8.0 Hz, 2H), 7.37 (d, *J* = 3.0 Hz, 1H), 7.26 (d, *J* = 9.0 Hz, 1H), 7.17 (dd, *J* = 9.0, 3.0 Hz, 1H), 3.80 (s, 3H), 2.42 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  158.4, 155.3, 145.6, 144.7, 134.0, 129.5, 128.3, 119.6, 117.5, 116.9, 106.2, 55.6, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>17</sub>H<sub>13</sub>O<sub>7</sub>S [M-H]<sup>-</sup> 361.0382, found 361.0383.

**7-Fluoro-4-hydroxy-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate (3ga).** Following the general procedure, the reaction of **1g** (54 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 40 mg (38%) of **3ga** as white solid. M.P.: 90-91 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.96 (dd, *J* = 8.9, 6.2 Hz, 1H), 7.87 (d, *J* = 8.3 Hz, 2H), 7.45 (d, *J* = 8.1 Hz, 2H), 7.37 (dd, *J* = 9.6, 2.4 Hz, 1H), 7.28 (td, *J* = 8.7, 2.5 Hz, 1H), 2.43 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  164.1(d, *J*<sub>C-F</sub> = 249 Hz), 158.4, 157.7, 152.3(d, *J*<sub>C-F</sub> = 13 Hz), 145.2, 133.1, 129.7, 128.4, 126.4(d, *J*<sub>C-F</sub> = 11 Hz), 116.0, 114.14, 112.2(d, *J*<sub>C-F</sub> = 23 Hz), 104.0(d, *J*<sub>C-F</sub> = 26 Hz), 21.2. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>10</sub>O<sub>6</sub>SF [M-H]<sup>-</sup> 349.0182, found 349.0175.

# **7-Chloro-4-hydroxy-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate (3ha).** Following the general procedure, the reaction of **1h** (59 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 58 mg (53%) of **3ha** as white solid. M.P.: 85-86 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) $\delta$ 7.88 (t, *J* = 9.3 Hz, 3H), 7.52 (s, 1H), 7.42 (t, *J* = 9.1 Hz, 3H), 2.41 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>) $\delta$ 158.9, 157.6, 151.6, 145.1, 136.5, 133.4, 129.6, 128.3, 125.8, 124.3, 117.0, 116.5, 116.4, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>10</sub>O<sub>6</sub>SCI [M-H]<sup>-</sup> 364.9887, found 364.9887.

**7-Bromo-4-hydroxy-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate** (**3ia**). Following the general procedure, the reaction of **1i** (72 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 75 mg (61%) of **3ia** as white solid. M.P.: 95-96 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.94 (d, J = 2.5 Hz, 1H), 7.92 (d, J = 8.2 Hz, 2H), 7.65 (dd, J = 8.7, 2.5 Hz, 1H), 7.41 (d, J = 8.2 Hz, 2H), 7.21 (d, J = 8.7 Hz, 1H), 2.42 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  159.6, 157.8, 151.6, 145.0, 133.5, 129.6, 128.3, 126.9, 126.0, 124.8, 119.1, 117.8, 116.6, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>16</sub>H<sub>10</sub>O<sub>6</sub>SBr [M-H]<sup>-</sup> 408.9381, found 408.9384.

**4-Hydroxy-7-methoxy-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate** (**3ja**). Following the general procedure, the reaction of **1j** (58 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 76 mg (70%) of **3ja** as white solid. M.P.: 134-135 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.83 (d, *J* = 8.3 Hz, 2H), 7.72 (d, *J* = 8.8 Hz, 1H), 7.35 (d, *J* = 8.0 Hz, 2H), 6.83 (dd, *J* = 8.8, 2.4 Hz, 1H), 6.78 (d, *J* = 2.4 Hz, 1H), 3.76 (s, 3H), 2.35 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  162.2, 158.7, 153.1, 144.5, 134.2, 129.4, 128.2, 125.5, 115.2, 111.4, 100.3, 55.8, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>17</sub>H<sub>13</sub>O<sub>7</sub>S [M-H]<sup>-</sup> 361.0382, found 361.0376.

# 7-(Dimethylamino)-4-hydroxy-2-oxo-2*H*-chromen-3-yl 4-methylbenzenesulfonate (3ka). Following the general procedure, the reaction of 1k (62 mg, 0.3 mmol), 2a (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 72 mg (64%) of 3ka as grey solid. M.P.:

168-169 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.84 (d, J = 8.3 Hz, 2H), 7.70 (d, J = 9.0 Hz, 1H), 7.45 (d, J = 8.1 Hz, 2H), 6.77 (dd, J = 9.1, 2.4 Hz, 1H), 6.52 (d, J = 2.3 Hz, 1H), 3.02 (s, 6H), 2.43 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  158.4, 158.1, 153.3, 153.2, 145.2, 133.1, 129.7, 128.4, 124.6, 113.6, 109.3, 103.4, 97.2, 38.9, 21.2. HRMS m/z (ESI-TOF): calculated for C<sub>18</sub>H<sub>16</sub>NO<sub>6</sub>S [M-H]<sup>-</sup> 374.0698, found 374.0698.

**4-Hydroxy-8-methyl-2-oxo-2***H***-chromen-3-yl <b>4-methylbenzenesulfonate** (**3la**). Following the general procedure, the reaction of **1l** (53 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 80 mg (77%) of **3la** as white solid. M.P.: 146-147 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.88 (d, J = 8.3 Hz, 2H), 7.77 (d, J = 7.0 Hz, 1H), 7.52 (d, J = 7.3 Hz, 1H), 7.46 (d, J = 8.1 Hz, 2H), 7.28 (t, J = 7.7 Hz, 1H), 2.44 (s, 3H), 2.34 (s,3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  158.5, 157.7, 149.4, 145.2, 133.6, 133.2, 129.7, 128.4, 125.3, 123.9, 121.8, 116.7, 116.6, 21.2, 15.2. HRMS m/z (ESI-TOF): calculated for C<sub>17</sub>H<sub>13</sub>O<sub>6</sub>S [M-H]<sup>-</sup> 345.0433, found 345.0440.

**4-Hydroxy-5-methyl-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate** (**3ma**). Following the general procedure, the reaction of **1m** (53 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 83 mg (80%) of **3ma** as white solid. M.P.: 96-97 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.86 (d, J = 8.3 Hz, 2H), 7.81 (d, J = 8.4 Hz, 1H), 7.45 (d, J = 8.0 Hz, 2H), 7.26 – 7.18 (m, 2H), 2.43 (s, 3H), 2.41 (s, 3H). A signal for OH-proton is not

 visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  158.3, 157.8, 151.1, 145.3, 143.5, 133.0, 129.7, 128.4, 125.4, 123.8, 116.4, 116.1, 114.1, 21.2, 21.0. HRMS m/z (ESI-TOF): calculated for C<sub>17</sub>H<sub>13</sub>O<sub>6</sub>S [M-H]<sup>-</sup> 345.0433, found 345.0433.

**1-Hydroxy-3-oxo-3H-benzo**[**f**]**chromen-2-yl 4-methylbenzenesulfonate** (**3na**). Following the general procedure, the reaction of **1n** (64 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 89 mg (78%) of **3na** as white solid. M.P.: 208-209 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  9.42 (d, *J* = 8.7 Hz, 1H), 8.23 (d, *J* = 9.0 Hz, 1H), 8.09 (d, *J* = 7.3 Hz, 1H), 7.95 (d, *J* = 8.3 Hz, 2H), 7.78 – 7.72 (m, 1H), 7.68 – 7.62 (m, 1H), 7.55 (d, *J* = 9.0 Hz, 1H), 7.50 (d, *J* = 8.1 Hz, 2H), 2.48 (s, 3H). A signal for OH-proton is not visible. <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  157.1, 152.2, 145.2, 134.0, 132.9, 130.3, 129.6, 129.4, 128.9, 128.6, 128.3, 125.7, 117.6, 117.0, 21.2. HRMS m/z (ESI-TOF): calculated for C<sub>20</sub>H<sub>13</sub>O<sub>6</sub>S [M-H]<sup>-</sup> 381.0433, found 381.0442.

**4-Hydroxy-1-methyl-2-oxo-1,2-dihydroquinolin-3-yl 4-methylbenzenesulfonate** (**3oa**). Following the general procedure, the reaction of **1o** (53 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (10 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 75 mg (72%) of **3oa** as white solid. M.P.: 111-112 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  11.63 (s, 1H), 8.02 (dd, *J* = 8.0, 1.3 Hz, 1H), 7.93 (d, *J* = 8.3 Hz, 2H), 7.72 – 7.63 (m, 1H), 7.52 (d, *J* = 8.4 Hz, 1H), 7.47 (d, *J* = 8.0 Hz, 2H), 7.37 – 7.29 (m, 1H), 3.55 (s, 3H), 2.44 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  157.9, 152.9, 144.9, 137.7, 133.8, 131.5, 129.5, 128.3, 123.7, 122.3, 122.0, 115.4, 114.8, 29.2, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>17</sub>H<sub>14</sub>NO<sub>5</sub>S [M-H]<sup>-</sup>

344.0593, found 344.0586.

**4-Hydroxy-6-methyl-2-oxo-***2H***-pyran-3-yl 4-methylbenzenesulfonate** (**3pa**). Following the general procedure, the reaction of **1p** (38 mg, 0.3 mmol), **2a** (141 mg, 0.36 mmol) and DCM (3 mL) at 35 °C for 1 h, after column chromatography on silica gel afforded 79 mg (95%) of **3pa** as white solid. M.P.: 117-118 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  12.37 (s, 1H), 7.84 (d, *J* = 8.3 Hz, 2H), 7.45 (d, *J* = 8.1 Hz, 2H), 6.04 (s, 1H), 2.43 (s, 3H), 2.18 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  161.9, 161.2, 159.4, 145.0, 133.6, 129.7, 128.1, 115.2, 99.8, 21.1, 19.2. HRMS m/z (ESI-TOF): calculated for C<sub>13</sub>H<sub>12</sub>O<sub>6</sub>SNa [M+Na]<sup>+</sup> 319.0252, found 319.0243.

#### 2-Oxo-4-(((trifluoromethyl)sulfonyl)oxy)-2H-chromen-3-yl 4-methylbenzenesulfo

(4). То -nate а ml round-bottomed flask was added 4-hydroxy-2-oxo-2*H*-chromen-3-yl 4-methylbenzenesulfonate **3aa** (5 mmol, 1.66 g) and Tf<sub>2</sub>O (10 mmol, 2.82 g). DCM (20 ml) was added and Et<sub>3</sub>N (0.5 mL) was slowly dropped to the solution. Then the mixture was stirred at room temperature for 3 hours. After TLC, the solvent was evaporated under vacuum. The crude products were purified using flash column chromatography on silica gel to afford 2.13 g (92%) of the desired product **4** as white solid. M.P.: 156-157 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.98 (d, J = 8.4 Hz, 2H), 7.78 - 7.66 (m, 2H), 7.55 - 7.37 (m, 4H), 2.50 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>) δ 151.1, 145.5, 143.0, 141.4, 128.6, 127.2, 124.7, 123.7, 119.4(q, *J*<sub>C-F</sub> = 220 Hz), 112.1, 109.5, 16.6. HRMS m/z (ESI-TOF): calculated for C<sub>17</sub>H<sub>12</sub>O<sub>8</sub>S<sub>2</sub>F<sub>3</sub> [M+H]<sup>+</sup> 464.9926, found 464.9927.

#### 2-Oxo-4-((2,5,7,8-tetramethyl-2-(4,8,12-trimethyltridecyl)chroman-6-yl)oxy)-2H-

chromen-3-vl 4-methylbenzenesulfonate (5). To a solution of 4 (0.2 mmol, 97 mg) in DMF (2 ml), Vitamin E (0.24 mmol, 103 mg) and K<sub>2</sub>CO<sub>3</sub> (0.4 mmol, 55 mg) was added. Then the mixture was stirred at 100 °C under oil bath for 12 hours. After TLC, water (20 mL) was added, and the mixture was firstly extracted with DCM and then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The crude compound was purified by column chromatography on silica gel to afford 63 mg (42%) of desired product 5 as colourless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.97 (t, J = 7.0 Hz, 1H), 7.67 (d, J = 8.3 Hz, 2H), 7.59 (t, J = 7.7 Hz, 1H), 7.34 (t, J = 6.9 Hz, 2H), 7.29 - 7.22 (m, 2H), 2.63 (s, 2H), 2.42 (s, 3H), 2.14 (s, 3H), 2.08 (s, 6H), 1.92 - 1.78 (m, 2H), 1.60 (s, 2H), 1.57 - 1.47 (m, 2H), 1.40 (d, J = 20.1 Hz, 4H), 1.34 - 1.19 (m, 10H), 1.13 (dd, J = 11.7, 7.1 Hz, 3H), 1.06 (dd, J = 7.6, 4.7 Hz, 3H), 0.90 – 0.80 (m, 12H).  $^{13}C{^{1}H}$  NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  158.7, 154.7, 150.9, 149.5, 144.8, 134.5, 132.4, 129.3, 128.4 126.0, 124.7, 124.5, 124.3, 123.3, 118.0, 116.8, 116.4, 75.2, 39.4, 37.4, 32.8, 28.0, 27.0, 24.9, 24.5, 22. 8, 22.7, 20. 8, 13.3, 12.0. HRMS m/z (ESI-TOF): calculated for C<sub>45</sub>H<sub>60</sub>NaO<sub>7</sub>S [M+Na]<sup>+</sup> 767.3957, found 767.3956.

**4-(Furan-3-yl)-2-oxo-2H-chromen-3-yl 4-methylbenzenesulfonate** (6). To a solution of **4** (0.2 mmol, 97 mg ) in toluene (2 ml), furan (1 mmol, 68 mg) was added under argon atmosphere at 0 °C. Then LiHDMs (0.2 mmol) was also slowly added at the same temperature and the mixture was stirred for additional 12 h. After TLC, water (20 mL) was added, and the mixture was firstly extracted with DCM and then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The crude compound was purified by column chromatography on silica gel to afford 40 mg (52%) of desired

product **6** as white solid. M.P.: 191-192 °C. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.99 – 7.88 (m, 3H), 7.62 (s, 1H), 7.56 (d, J = 8.1 Hz, 2H), 7.29 – 7.14 (m, 2H), 6.94 (td, J = 7.6, 1.0 Hz, 1H), 6.86 – 6.81 (m, 1H), 6.60 (d, J = 2.6 Hz, 1H), 2.47 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO- $d_6$ )  $\delta$  162.8, 154.1, 151.3, 146.3, 138.5, 130.8, 130.4, 130.3, 128.5, 128.4, 122.6, 121.9, 120.2, 118.5, 102.4, 79.9, 76.7, 21.2. HRMS m/z (EI-TOF): calculated for C<sub>20</sub>H<sub>14</sub>O<sub>6</sub>S [M]<sup>+</sup> 382.0511, found 382.0513.

**2-Oxo-4-(p-tolylthio)-2***H***-chromen-3-yl <b>4-methylbenzenesulfonate** (**7**). To a solution of **4** (0.2 mmol, 97 mg ) in toluene (2 ml), 4-methylbenzenethiol (0.24 mmol, 30 mg) was added under argon atmosphere at 0 °C. Then LiHDMs (0.2 mmol) was also slowly added at the same temperature and the mixture was stirred at 110 °C under oil bath for additional 12 h. After TLC, water (20 mL) was added, and the mixture was firstly extracted with DCM and then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated in vaccum. The crude compound was purified by column chromatography on silica gel to afford 55 mg (63%) of desired product **7** as white solid. M.P.: 142-143 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.85 (d, *J* = 8.3 Hz, 2H), 7.76 (dd, *J* = 8.1, 1.2 Hz, 1H), 7.47 – 7.37 (m, 1H), 7.31 – 7.22 (m, 3H), 7.13 (td, *J* = 8.3, 1.5 Hz, 3H), 6.99 (d, *J* = 8.0 Hz, 2H), 2.39 (s, 3H), 2.23 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  155.6, 151.1, 145.7, 143.7, 138.2, 136.0, 133.7, 131.9, 130.9, 130.3, 129.7, 128.8, 128.7, 128.0, 124.8, 119.0, 117.0, 21.8, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>23</sub>H<sub>19</sub>O<sub>5</sub>S<sub>2</sub> [M+H]<sup>+</sup> 439.0674, found 439.0675.

**4-(Diphenylphosphanyl)-2-oxo-2***H***-chromen-3-yl 4-methylbenzenesulfonate (8).** To a solution of **4** (0.2 mmol, 97 mg ) in toluene (2 ml), diphenylphosphane (0.24 mmol, 45 mg) was added under argon atmosphere at 0 °C. Then LiHDMs (0.2 mmol) was also slowly added at the same temperature and the mixture was stirred at 110 °C under oil bath for additional 12 h. After TLC, water (20 mL) was added, and the mixture was firstly extracted with DCM and then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The crude compound was purified by column chromatography on silica gel to afford 52 mg (52%) of desired product **8** as colourless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.85 (d, *J* = 8.3 Hz, 2H), 7.76 (dd, *J* = 8.1, 1.2 Hz, 1H), 7.47 – 7.37 (m, 1H), 7.31 – 7.22 (m, 3H), 7.13 (td, *J* = 8.3, 1.5 Hz, 3H), 6.99 (d, *J* = 8.0 Hz, 2H), 2.39 (s, 3H), 2.23 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  155.6, 151.1, 145.7, 143.7, 138.2, 136.0, 133.7, 131.9, 130.9, 130.3, 129.7, 128.8, 128.7, 128.0, 124.8, 119.0, 117.0, 21.8, 21.1. HRMS m/z (ESI-TOF): calculated for C<sub>28</sub>H<sub>22</sub>O<sub>5</sub>PS [M+H]<sup>+</sup> 501.0926, found 501.0927.

**2-Oxo-3-(phenyliodonio)-2***H***-chromen-4-olate (11).** To a 50 ml round-bottomed flask was added 4-hydroxy-2*H*-chromen-2-one **1a** (1 mmol) and DCM (20 ml). Then the temperature was cooled to 0 °C and [hydroxy(sulfonyloxy)iodo]arenes **2** (1 mmol) was added slowly. After stirring for 15 minutes, the solvent was evaporated under vacuum. The crude products were purified using flash column chromatography on silica gel to afford 302 mg (83%) of the desired product **11** as pale yellow solid. M.P.: 136-137 °C. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.91 (d, *J* = 7.7 Hz, 1H), 7.86 (d, *J* = 8.3 Hz, 2H), 7.61 – 7.50 (m, 2H), 7.43 (t, *J* = 7.6 Hz, 2H), 7.32 – 7.21 (m, 2H). <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  172.5, 161.0, 153.9, 133.1, 132.8, 131.1, 130.7, 125.6, 123.4, 119.8, 116.3, 115.0, 82.0. HRMS m/z (EI-TOF): calculated for

C<sub>15</sub>H<sub>9</sub>IO<sub>3</sub> [M]<sup>+</sup> 363.9596, found 363.9599.

**Warfarin** (12). To a 50 ml round-bottomed flask was added ZICs 11 (0.5 mmol) and (*E*)-4-phenylbut-3-en-2-one (0.5 mmol). MeCN (20 ml) was added and the mixture was stirred at 80 °C under oil bath for 12 hours. Then the solvent was evaporated under vacuum. The crude products were purified using flash column chromatography on silica gel to afford 117 mg (76%) of the desired product 12 as white solid. M.P.: 161-162 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.89 (d, J = 7.8 Hz, 0.50H), 7.80 (d, J = 7.7 Hz, 0.53H), 7.56 (t, J = 7.8 Hz, 0.44H), 7.48 (t, J = 7.8 Hz, 0.71H), 7.34 – 7.15 (m, 7.42H), 4.27 (d, J = 3.0 Hz, 0.44H), 4.16 (dd, J = 11.0, 6.8 Hz, 0.53H), 2.62 – 2.33 (m, 1.44H), 1.98 (t, J = 12.7 Hz, 0.56H), 1.69 (s, 1.59H), 1.66 (s, 1.40H). Analytical data are in agreement with the reported ones <sup>19</sup>.

#### Acknowlegements

The work was supported by the National Nature Science Foundation of China (NSFC 21772039), National Key Program (2016YFA0200302, Study on application and preparation of aroma nanocomposites), the Fundamental Research Funds for the Central Universities and Key Laboratory of Organofluorine Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences.

## **Supporting Information**

The characterization data and copies of <sup>1</sup>H, <sup>13</sup>C NMR spectra for all products. The Supporting Information is available free of charge via the internet at <u>http://pubs.acs.org.</u>

# **Conflict of interest**

The author declare no conflict of interest.

## References

 For recent reviews, see: (a) Zhdankin, V. V.; Stang, P. J. Chemistry of polyvalent iodine. *Chem. Rev.* 2008, *108*, 5299-5358. (b) Aradi, K.; Tóth, B. L.; Tolnai, G. L.; Nov &, Z. Diaryliodonium salts in organic syntheses: a useful compound class for novel arylation strategies. *Synlett* 2016, *27*, 1456-1485. (c) Yoshimura, A.; Zhdankin, V. V. Advances in synthetic applications of hypervalent iodine compounds. *Chem. Rev.* 2016, *116*, 3328-3435.

Dong, D-Q.; Hao, S-H.; Wang, Z-L.; Chen, C. Hypervalent iodine: a powerful electrophile for asymmetric α-functionalization of carbonyl compounds. *Org. Biomol. Chem.* 2014, *12*, 4278-4289.

3. (a) Koser, G. F.; Relenyi, A. G.; Kalos, A. N.; Rebrovic, L.; Wettach, R. H.
One-step. alpha-tosyloxylation of ketones with [hydroxy (tosyloxy) iodo] benzene. J.
Org. Chem. 1982, 47, 2487-2489. (b) Koser, G. F., Telua, S., Laali, K. K.
Oxidative-substitution reactions of polycyclic aromatic hydrocarbons with iodine(III)
sulfonate reagents. *Tetrahedron Lett.* 2006, 47: 7011–7015. (c) Hirt, U. H.; Spingler,
B.; Wirth, T. New chiral hypervalent iodine compounds in asymmetric synthesis. J.
Org. Chem. 1998, 63, 7674-7679.

4. (a) Yu, J.; Cui, J.; Hou, X.-S.; Liu, S.-S.; Gao, W.-C.; Jiang, S.; Tian, J.; Zhang, C. Enantioselective  $\alpha$ -tosyloxylation of ketones catalyzed by spirobiindane scaffold-based chiral iodoarenes. *Tetrahedron: Asymmetry.* **2011**, *22*, 2039-2055. (b) Kumar, D.; Sundaree, M. S.; Patel, G.; Rao, V. S.; Varma, R. S. Solvent-free facile synthesis of novel  $\alpha$ -tosyloxy  $\beta$ -keto sulfones using [hydroxy (tosyloxy) iodo] benzene. Tetrahedron Lett. 2006, 47, 8239-8241. (c) Yamamoto, Y.; Togo H. PhI-catalyzed a-tosyloxylation of ketones with m-chloroperbenzoic acid and p-toluenesulfonic acid. Synlett 2006, 5, 798-800. (d) Yusubov, M. S.; Wirth, T. Solvent-free reactions with hypervalent iodine reagents. Org. Lett. 2005, 7, 519-521. (e) Zhdankin, V. V.; Kuehl, C. J.; J. Simonsen, A. 1-[Hydroxy(sulfonyloxy)iodo]-1H,1H-perfluoroalkanes: Stable, Fluoroalkyl Analogs of Koser's Reagent. J. Org. Chem. 1996, 61, 8272-8276.

5. (a) Beaulieu, S.; Legault, C. Y. Mechanistic insights on the iodine (III)-mediated  $\alpha$ -oxidation of ketones. *Chem. Eur. J.* **2015**, *21*, 11206-11211. (b) Jobin-Des Lauriers, A.; Legault, C. Y. Iodine (III)-mediated oxidative hydrolysis of haloalkenes: access to  $\alpha$ -halo ketones by a release-and-catch mechanism. *Org. Lett.* **2016**, *18*, 108-111.

6. Basdevant, B.; Legault, C. Y. Study of the reactivity of [hydroxy (tosyloxy) iodo] benzene toward enol esters to access  $\alpha$ -tosyloxy ketones. *J. Org. Chem.* **2015**, *80*, 6897-6902.

7. (a) Yusubov, M. S.; Yoshimura, A.; Zhdankin, V. V. Iodonium ylides in organic synthesis. *ARKIVOC*. **2016**, 342-374. (b) Sreenithya, A.; Surya, K.; Sunoj, R. B. Hypercoordinate iodine (III) promoted reactions and catalysis: an update on current mechanistic understanding. *WIREs Comput. Mol. Sci.* **2017**, *7*, e1299.

8. (a) Kappe, T.; Korbuly, G.; Stadlbauer, W. Ylide von heterocyclen, II: iodoniumund pyridinium- ylide von malonylheterocyclen. *Chem. Ber.* **1978**, *111*, 3857-3866. (b) Panda, N., Mattan, I. One-pot two-step synthesis of 3-iodo-4-aryloxy coumarins and their Pd/C-catalyzed annulation to coumestans. *RSC Adv.* **2018**, *8*, 7716-7725.

 9. (a) Zhu, S.-Z.; Chen, Q.-Y. Phenyliodonium bis(perfluoroalkane sulphonyl) methide; synthesis and reactions as a precursor of bis(perfluoroalkanesulphonyl) carbene. J. Chem. Soc., Chem. Commun. 1990, 1459-1460. (b) Yang, R.-Y.; Dai, L.-X.; Chen, C.-G. Synthesis of cyclic iodonium ylides— $3H-1\lambda^3$ -benziodol-1-ylium ylides via transylidation of iodonium ylides to iodides. J. Chem. Soc., Chem. Commun. 1992, 1487-1488. (c) Malamidou-Xenikaki, E.; Spyroudis, S. Zwitterionic iodonium compounds: Useful tools in organic synthesis. Synlett 2008, 18, 2725-2740. (d) Papoutsis, I.; Spyroudis, S.; Varvoglis, The chemistry A. of 2-oxido-3-phenyliodonio-1, 4-benzoquinones: transformation to 2-cyclopentene-1, 4-diones and cycloadditions. Tetrahedron Lett. 1994, 35, 8449-8452. (e) Spyroudis, S.; Tarantili P. Phenyliodoniophenolates from 1, 3-dihydroxybenzene derivatives. Tetrahedron. 1994, 50, 11541-11552. (f) Yu, J.; Liu, S.-S.; Cui, J.; Hou, X.-S.; Zhang, C. A Mild and Efficient Direct  $\alpha$ -amination of  $\beta$ -dicarbonyl compounds using iodosobenzene and *p*-toluenesulfonamide catalyzed by perchlorate zinc hexahydrate. Org. Lett. 2012, 14, 832-835. (g) Cui, J.; Duan, Y.-N.; Yu, J.; Zhang, C. Iodosobenzene-mediated direct and efficient oxidation of  $\beta$ -dicarbonyls to vicinal tricarbonyls catalyzed by iron (iii) salts. Org. Chem. Front. 2016, 3, 1686-1690.

10. Laschober, R.; Kappe, T. A new and efficient synthesis of coumestan and coumestrol. *Synthesis* **1990**, 387-388.

11. (a) Chen, X.; Jin, D.-Y.; Stafford, D. W.; Tie, J.-K. Evaluation of oral anticoagulants with vitamin K epoxide reductase in its native milieu. *Blood* 2018, *132*, 1974-1984. (b) Serra, S.; Chicca, A.; Delogu, G.; V ázquez-Rodr guez, S.; Santana, L.;

Uriarte, E.; Casu, L.; Gertsch, J. Synthesis and cytotoxic activity of non-naturally substituted 4-oxycoumarin derivatives. *Bioorg. Med. Chem. Lett.* **2012**, *22*, 5791-5794.

12. Salar, U.; Khan, K. M.; Iqbal, J.; Ejaz, S. A.; Hameed, A.; Ai-Rashida, M.; Perveen, S.; Tahir, M. N. Coumarin sulfonates: New alkaline phosphatase inhibitors; in vitro and in silico studies. *Eur. J. Med. Chem.* **2017**, *131*, 29-47.

13. Salar, U.; Khan, K. M.; Jabeen, A.; Faheem, A.; Fakhri, M. I.; Saad, S. M., Perveen S.; Taha, M.; Hameed, A. Coumarin sulfonates: as potential leads for ROS inhibition. *Bioorg. Chem.* **2016**, *69*, 37-47.

14. (a) Chen, H.; Han, J.; Wang, L. Intramolecular aryl migration of diaryliodonium salts: access to *ortho*-iodo diaryl ethers. *Angew. Chem. Int. Ed.* 2018, *57*, 12313
-12317. (b) Qian, X.; Han, J.; Wang, L. *tert*-Butoxide- mediated arylation of 2-substituted cyanoacetates with diaryliodonium salts. *Adv. Synth. Catal.* 2016, *358*, 940-946. (c) Han, J.; Qian, X.; Xu, B.; Wang, L. Potassium *tert*-butoxide mediated arylation of 2-substituted malononitriles using diaryliodonium salts. *Synlett* 2017, *28*, 2139-2142. (d) Xu, B.; Han, J.; Wang, L. Metal- and base- free direct *N*-arylation of pyridazinones by using diaryliodonium salts: an anion effect. *Asian J. Org. Chem.* 2018, *7*, 1674-1680.

15. Hatzigrigoriou, E., Varvoglis A., Bakola-Christianopoulou, M. Preparation of [hydroxy(((+)-10-camphorsulfonyl)oxy)iodo]benzene and its reactivity toward carbonyl compounds. *J. Org. Chem.* **1990**, *55*, 315-318.

16. Chen, Z.; Tong, L.; Du, Z.; Mao, Z.; Zhang, X.; Zou, Y.; Yan, M. Annulation of

β-naphthols and 4-hydroxycoumarins with vinylsulfonium salts: synthesis of dihydrofuran derivatives. *Org. Biomol. Chem.* **2018**, *16*, 2634-2638.

17. Azelmata, J.; Fioritob, S.; Taddeob, V. A.; Genoveseb, S.; Epifanob, F.; Grenier, D. Synthesis and evaluation of antibacterial and anti-inflammatory properties of naturally occurring coumarins. *Phytochem. Lett.* **2015**, *13*, 399–405.

18. Bartlett, M. J.; Turner, C. A.; Harvey, J. E. Pd-catalyzed allylic alkylation cascade with dihydropyrans: regioselective synthesis of furo[3,2-*c*]pyrans. *Org. Lett.* **2013**, *15*, 2430-2433.

19. Massolo, E.; Palmieri, S.; Benaglia, M.; Capriati, V.; Perna, F. Stereoselective organocatalysed reactions in deep eutectic solvents: Highly tunable and biorenewable reaction media for sustainable organic synthesis. *Green Chem.* **2016**, *18*, 792-797.