PRODUCTS

Bioactive Triterpenoids from the Twigs of Chaenomeles sinensis

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Supporting Information

ABSTRACT: Chaenomeles sinensis has been consumed traditionally for the treatment of throat diseases, diarrhea, inflammatory diseases, and dry beriberi. Repeated chromatography of the CHCl₃-soluble fraction from the 80% MeOH extract of *C. sinensis* twigs led to the isolation of three new triterpenoids, sinenic acid A (1), 3β -O-cis-feruloyl- 2α , 19α -dihydroxyurs-12-en-28-oic acid (2), and 3β -O-cis-caffeoylbetulin (3), together with 20 analogues. The chemical structures of 1–3 were determined using diverse NMR techniques and HRMS data analysis, chemical methods, and computational approaches supported by advanced statistics (CP3). All the



purified compounds were evaluated not only for their cytotoxicity against four human tumor cell lines (A549, SK-OV-3, SK-MEL-2, and HCT-15) but for their potential neuroprotective effects through induction of nerve growth factor in C6 glioma cells. Their anti-inflammatory effects were also assessed by measuring nitric oxide levels in lipopolysaccharide-insulted murine microglia BV2 cells.

Chaenomeles sinensis (Thouin) Koehne is a deciduous tree belonging to the family Rosaceae and is distributed widely in Korea, Japan, and mainland China. The ripened yellow fruit of C. sinensis, also called Chinese quince, has been used in Korean traditional medicine to treat throat diseases, diarrhea, inflammatory diseases, and dry beriberi.¹⁻³ Owing to this fruit being hard and possessing a strong acid taste, it is usually consumed in a candied form, as a tea, or in liquor in order to act as an effective cough medicine and diuretic.⁴ The extract of C. sinensis fruit has been reported not only to possess antiviral, antidiabetic, antiacetylcholinesterase, antihyperglycemic, antihyperlipidemic, and antioxidant activities⁵⁻⁸ but to produce triterpenoids, flavonoids, and phenolic compounds exhibiting tissue factor inhibitory and antipruritic activities.¹⁻³ In contrast to the relatively large number of studies on C. sinensis fruits due to its ethnopharmacological applications, there have been few investigations of the potentially bioactive constituents of C. sinensis twigs.^{9,10}

In a continuing search for bioactive secondary metabolites from Korean medicinal plants, chemical investigations of *C. sinensis* twigs were carried out, leading to the isolation and characterization of three new (1-3) and 20 known triterpenoids (4-23). The structures of these purified compounds were established employing spectroscopic and spectrometric analysis, chemical methods, and computational analysis aided by statistics. All these phytochemicals (1-23) were also evaluated for their cytotoxic, neuroprotective, and anti-inflammatory activities.

RESULTS AND DISCUSSION

Compound 1 was isolated as a white powder, and its molecular formula was established as C30H46O4 based on the deprotonated HRFABMS ion signal $[M - H]^-$ observed at m/z469.3313 (calcd for $C_{30}H_{45}O_4$, m/z 469.3312). The ¹H NMR spectrum of 1 displayed resonances for seven singlet methyl protons [$\delta_{\rm H}$ 1.24, 1.20, 0.99, 0.98, 0.95, and 0.92 (×2)], an olefinic proton [$\delta_{\rm H}$ 5.45 (1H, brd, J = 3.6 Hz)], an oxygenated methine proton [$\delta_{\rm H}$ 3.90 (1H, d, J = 3.7 Hz)], and two methylene protons [$\delta_{\rm H}$ 3.01 (1H, d, J = 12.3 Hz) and 2.29 (1H, d, J = 12.3 Hz)]. The ¹³C NMR spectrum of 1 revealed 30 resonances including a keto carbonyl ($\delta_{\rm C}$ 213.1), a hydroxy carbonyl ($\delta_{\rm C}$ 179.9), two olefinic carbons ($\delta_{\rm C}$ 144.6 and 121.8), and an oxygenated methine carbon ($\delta_{\rm C}$ 82.9). These spectroscopic data (Table 1) were comparable to those of $2\alpha_{3}\alpha$ -dihydroxyolean-12-en-28-oic acid (7) except for the presence of the keto carbonyl resonance ($\delta_{\rm C}$ 213.1) and the



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Table 1. ¹H [ppm, mult. (J in Hz)] and ¹³C NMR Data of Compounds 1 and 4 in Pyridine- d_5

| | | 1 | | 4 |
|----------|-----------------|----------------------|-----------------|----------------------|
| position | $\delta_{ m C}$ | $\delta_{ m H}$ | $\delta_{ m C}$ | $\delta_{ m H}$ |
| 1ax | 51.0 | 3.01, d (12.3) | 53.4 | 2.20, d (12.0) |
| 1eq | | 2.29, d (12.3) | | 2.48, d (12.0) |
| 2 | 213.1 | | 210.8 | |
| 3 | 82.9 | 3.90, d (3.7) | 83.4 | 4.19, brs |
| 4 | 41.9 | | 45.5 | |
| 5 | 49.8 | 2.00, overlap | 54.5 | 1.56, brd (13.6) |
| 6ax | 18.9 | 1.39, overlap | 18.7 | 1.37, overlap |
| 6eq | | 1.53, brd (14.0) | | 1.58, overlap |
| 7ax | 32.6 | 1.58, td (12.5, 3.3) | 32.6 | 1.54, overlap |
| 7ex | | 1.33, overlap | | |
| 8 | 40.0 | | 39.8 | |
| 9 | 47.8 | 2.06, overlap | 47.4 | 1.97, overlap |
| 10 | 42.6 | | 43.3 | |
| 11 | 23.4 | 1.96, overlap | 23.5 | 1.83, overlap |
| 12 | 121.8 | 5.45, brd (3.6) | 121.7 | 5.46, t (3.6) |
| 13 | 144.6 | | 144.7 | |
| 14 | 42.1 | | 42.0 | |
| 15ax | 28.0 | 2.15, td (14.4, 4.6) | 28.1 | 2.14, overlap |
| 15eq | | 1.18, overlap | | 1.16, overlap |
| 16a | 23.5 | 2.08, overlap | 23.5 | 2.09, overlap |
| 16b | | 1.96, overlap | | 1.97, overlap |
| 17 | 46.4 | | 46.4 | |
| 18 | 41.7 | 3.29, dd (13.7, 4.0) | 41.7 | 3.30, dd (14.1, 3.2) |
| 19ax | 46.2 | 1.77, t (13.7) | 46.2 | 1.80, t (14.1) |
| 19eq | | 1.28, overlap | | 1.29, overlap |
| 20 | 30.7 | | 30.7 | |
| 21ax | 34.0 | 1.43, td (13.8, 4.0) | 34.0 | 1.43, overlap |
| 21eq | | 1.20, overlap | | 1.19, overlap |
| 22ax | 32.9 | 2.04, overlap | 32.9 | 2.04, td (14.1, 4.4) |
| 22eq | | 1.82, brd (13.4) | | 1.82, overlap |
| 23 | 27.4 | 1.24, s | 29.2 | 1.32, s |
| 24 | 21.4 | 0.92, s | 16.7 | 0.87, s |
| 25 | 16.4 | 0.95, s | 16.1 | 0.85, s |
| 26 | 16.8 | 0.98, s | 17.1 | 0.95, s |
| 27 | 25.8 | 1.20, s | 25.8 | 1.28, s |
| 28 | 179.9 | | 179.9 | |
| 29 | 33.0 | 0.92, s | 33.0 | 0.95, s |
| 30 | 23.5 | 0.99, s | 23.5 | 1.00, s |
| OH-3 | | 7.52, overlap | | n/d |

absence of the oxygenated carbon resonance ($\delta_{\rm C}$ 66.1) observed in $2\alpha_{,}3\alpha_{-}$ dihydroxyolean-12-en-28-oic acid (7).¹¹ The keto carbonyl group was located at C-2 based on the HMBC cross-peak from H-1 [$\delta_{\rm H}$ 3.01 (1H, d, J = 12.3 Hz) and 2.29 (1H, d, J = 12.3 Hz)] to C-2 ($\delta_{\rm C}$ 213.1) (Figure 1). The 2D structure of 1 was established via thorough analysis of the COSY, HSQC, and HMBC NMR spectra (Figure 1). The α orientation of OH-3 was confirmed given the strong and weak NOESY correlations from H-3 to H-24 and H-23, respectively. The configurations of the remaining stereogenic centers were identical to those of $2\alpha_3\alpha$ -dihydroxyolean-12-en-28-oic acid (7) based on the NOESY correlations of 1 (Figure 2) and their ¹³C NMR chemical shift values similar to those of the reported oleanane derivatives (Table 1).^{11,12} Therefore, the structure of 1 was established as 2-oxo- 3α -hydroxyolean-12-en-28-oic acid, and it was assigned the trivial name sinenic acid A.

The HRFABMS of compound 2 (white powder), which exhibited a deprotonated ion at m/z 663.3892 $[M - H]^-$ (calcd for $C_{40}H_{55}O_8$, m/z 663.3891), led to its molecular formula being established as C40H56O8. The comparison of its ¹H and ¹³C NMR data (Table 2) with those of tormentic acid (11) indicated that 2 possesses an additional cis-feruloyl moiety, which was deduced from several characteristic NMR chemical shifts: $\delta_{\rm H}$ 7.77 (1H, d, J = 2.0 Hz), 7.14 (1H, dd, J = 8.2, 2.0 Hz), 6.90 (1H, d, J = 12.9 Hz), 6.78 (1H, d, J = 8.2 Hz), 5.90 (1H, d, J = 12.9 Hz), and 3.88 (3H, s); $\delta_{\rm C}$ 168.7, 149.5, 148.4, 145.2, 128.4, 126.7, 117.5, 115.7, 115.1, and 56.7.13 Moreover an HMBC cross-peak from H-3 to C-9' confirmed the cisferuloyl motif as being substituted at C-3 (Figure 1). Alkaline hydrolysis of 2 afforded tormentic acid (11), which was substantiated by comparing the resultant ¹H NMR spectrum with previously reported data.¹³ The full determination of the 1D and 2D NMR supported the assignment of structure of 2 as 3β -O-cis-feruloyl- 2α , 19α -dihydroxyurs-12-en-28-oic acid.

The molecular formula of compound 3 (white powder) was assigned as $C_{39}H_{56}O_5$ based on the deprotonated HRFABMS ion signal at m/z 603.4044 [M – H]⁻ (calcd for $C_{39}H_{55}O_5$, m/z 603.4044). Inspection of the ¹H and ¹³C NMR data of 3 (Table 2) suggested that this compound shares close structural similarities with betulin (22).¹⁴ When compared to 22, however, there was an additional *cis*-caffeoyl moiety in 3, which was verified by the presence of the NMR resonances at δ_H 7.31 (1H, d, J = 2.1 Hz), 7.02 (1H, dd, J = 8.2, 2.1 Hz), 6.82 (1H, d, J = 12.7 Hz) and δ_C 168.6, 148.2, 145.9, 144.8, 128.5, 124.7, 118.6, 117.6, and 115.8. The location of this additional *cis*-



Figure 1. COSY (blue bold) and HMBC (red arrows) correlations of 1-3.



Figure 2. NOESY correlations (yellow dashed) of 1 and 4. The geometry of the 3D structures was initially minimized at the MMFF force field and subsequently optimized at the B3LYP/6-31+G(d,p) level. Some protons were removed for a clearer presentation.

caffeoyl moiety was confirmed at C-3 via the HMBC cross-peak of H-3/C-9' (Figure 1). The structure of **3** was verified when betulin (**22**) was produced from the alkaline hydrolysis of **3**, and this was confirmed by the comparison of the ¹H NMR spectrum of the hydrolysate with that of **22**.¹⁴ The structure of **3** was consequently established as 3β -O-cis-caffeoylbetulin.

Compound 4 was obtained as a white powder with the identical molecular formula with that of 1 ($C_{30}H_{46}O_4$), from its HRFABMS showing a deprotonated ion at m/z 469.3311 [M – H]⁻ (calcd for $C_{30}H_{45}O_4$, m/z 469.3312). The ¹H and ¹³C NMR data of 4 were found to be closely comparable to those of 1, but there were slight shifts in the resonances of an oxygenated methine [$\delta_{\rm H}$ 4.19 (1H, brs), $\delta_{\rm C}$ 83.4, for 4; $\delta_{\rm H}$ 3.90 (1H, d, J = 3.7 Hz), $\delta_{\rm C}$ 82.9, for 1] and a keto carbonyl ($\delta_{\rm C}$ 210.8, for 4; $\delta_{\rm C}$ 213.1, for 1) group, suggesting that 4 is the C-3 epimer of 1. Indeed, the planar structure of 4 was determined to be identical with that of 1 via NMR data analysis. The β orientation of the C-3 hydroxy group was determined based upon the NOESY correlations from H-3 to H-1_{avt} H-5, and H-23 (Figure 2), confirming that 4 was the C-3 epimer of 1. The structure of 4 was consequently established as 2-oxo-3 β hydroxyolean-12-en-28-oic acid and named sinenic acid B. Compound 4 was purified from a natural source for the first time, although it has been reported as a synthetic product.¹⁵

The epimeric relationship of **1** with **4** was reaffirmed using the CP3 protocol featuring quantum mechanics (QM)-based calculations of NMR shielding tensors coupled with advanced statistics. This strategy has been extensively applied in structural validation and revision when the NMR spectra of a pair of diastereomers are available.¹⁶ Major conformers of **1** and **4** were identified upon conformational searches utilizing Macromodel (Schrodinger LLC), and their GIAO shielding constant calculations were computed at the B3LYP/6-31G(d,p) level employing the Gaussian 09 package (Gaussian Inc.) (see the Experimental Section). These resultant chemical shift values were averaged given their Boltzmann populations calculated based upon the relative MMFF94 potential energies (Table S1, Supporting Information), and these averaged values were subjected to the calculation of CP3 analysis (see the Experimental Section; Tables S2 and S3, Supporting Information). The CP3 analysis supported the NMR-based assignments produced for each epimer (i.e., 1 and 4) with 99.9% probability when both the ¹H and ¹³C NMR chemical shift values were considered (Figure 3; Figure S29, Supporting Information).

The other 19 reported phytochemicals were identified via comparison of their observed physical and spectroscopic data as maslinic acid (5),¹² arjunic acid (6),¹⁷ 2α , 3α -dihydroxyolean-12-en-28-oic acid (7),¹¹ (2α)-hydroxy-3-oxoolean-12-en-28-oic acid (8),¹⁸ euscaphic acid (9),¹⁹ 2α -hydroxypomolic acid (10),¹⁹ tormentic acid (11),¹³ 2α -hydroxyursolic acid (12),¹⁹ 2α ,19 α -dihydroxy-3-oxours-12-en-28-oic acid (13),²⁰ 3-epi-2-oxopomolic acid (14),²¹ pomolic acid 3-acetate (15),²² 3β -*O*-*trans*-feruloyl- 2α ,19 α -dihydroxyurs-12-en-28-oic acid (16),²³ betulinic acid (17),¹⁴ alphitolic acid (18),²⁴ 3-*O*-*trans*-caffeoylbetulinic acid (19),²⁵ 3-*O*-*cis*-caffeoylbetulinic acid (20),²⁵ ilekudinol C (21),²⁶ betulin (22),¹⁴ and 3β -*O*-*trans*-caffeoylbetulin (23).²⁷

Using a sulforhodamine B (SRB) assay, the cytotoxicities of the purified compounds 1–23 were evaluated against four human cancer cell lines, namely, A549 (non-small-cell lung adenocarcinoma), SK-OV-3 (ovary malignant ascites), SK-MEL-2 (skin melanoma), and HCT15 (colon adenocarcinoma). According to Table 3, the lupane-type triterpenoids 3, 17, 19–21, and 23 showed cytotoxic potencies with IC₅₀ values ranging from 0.9 to 9.9 μ M. Interestingly, betulin (22) displayed no activity (IC₅₀ > 10.0 μ M), whereas its 3-Ocaffeoyl derivatives 3 and 23 showed cytotoxicity toward SK-OV-3 and SK-MEL-2 cancer cell lines, with the former possessing a *cis*-3-O-caffeoyl functionality and being slightly more potent than its geometric isomer (3, 8.4 and 2.0 μ M; 23, 9.1 and 2.9 μ M). In line with this finding, compound 20, the

Table 2. ¹H [ppm, mult. (J in Hz)] and ¹³C NMR Data of Compounds 2 and 3 in Methanol- d_4

| | | 2 | | 3 |
|----------------------|------------------|----------------------|-----------------|----------------------|
| position | $\delta_{\rm C}$ | $\delta_{ m H}$ | $\delta_{ m C}$ | $\delta_{ m H}$ |
| 1ax | 48.9 | 2.05, overlap | 39.8 | 1.76, overlap |
| 1eq | | 1.08, overlap | | 1.05, overlap |
| 2 | 67.7 | 3.84, overlap | 24.8 | 1.72, overlap |
| 3 | 85.5 | 4.64, d (9.9) | 82.4 | 4.52, dd (9.1, 7.3) |
| 4 | 40.6 | | 39.1 | |
| 5 | 56.6 | 1.03, brd (12.0) | 57.0 | 0.87, overlap |
| 6a | 19.7 | 1.61, overlap | 19.5 | 1.56, overlap |
| 6b | | 1.52, overlap | | 1.50, overlap |
| 7a | 34.2 | 1.64, overlap | 35.5 | 1.50, overlap |
| 7b | | 1.38, overlap | | 1.46, overlap |
| 8 | 42.8 | | 42.3 | |
| 9 | 48.7 | 1.83, overlap | 51.9 | 1.42, overlap |
| 10 | 39.4 | | 38.4 | |
| 11a | 24.9 | 2.08, overlap | 22.2 | 1.48, overlap |
| 11b | | | | 1.30, overlap |
| 12a | 129.2 | 5.33, brt (3.4) | 26.7 | 1.73, overlap |
| 12b | | | | 1.13, overlap |
| 13 | 140.4 | | 38.8 | 1.74, overlap |
| 14 | 41.3 | | 44.0 | |
| 15a | 29.8 | 1.86, overlap | 28.3 | 1.82, td (13.3, 4.2) |
| 15b | | 1.04, overlap | | 1.05, overlap |
| 16ax | 26.8 | 2.59, td (12.7, 5.0) | 30.5 | 1.22, td (13.3, 4.0) |
| 16eq | | 1.56 | | 2.00, overlap |
| 17 | 48.7 | | 49.1 | |
| 18 | 55.3 | 2.55, s | 50.2 | 1.64, t (11.7) |
| 19 | 73.8 | | 49.7 | 2.45, td (11.1, 5.7) |
| 20 | 43.3 | 1.40, overlap | 152.0 | |
| 21a | 27.5 | 1.79, overlap | 31.0 | 2.00, overlap |
| 21b | | 1.25, overlap | | 1.40, overlap |
| 22a | 39.4 | 1.77, overlap | 35.2 | 1.93, dd (12.0, 8.9) |
| 22b | | 1.68, overlap | | |
| 23 | 29.4 | 0.93, s | 28.7 | 0.88, s |
| 24 | 18.3 | 0.89, s | 17.1 | 0.81, s |
| 25 | 17.2 | 1.07, s | 16.9 | 0.92, s |
| 26 | 17.7 | 0.85, s | 16.7 | 1.11, s |
| 27 | 25.0 | 1.39, s | 15.4 | 1.06, s |
| 28a | 182.7 | | 60.5 | 3.77, brd (10.6) |
| 28b | | | | 3.31, overlap |
| 29 | 16.8 | 0.96, d (6.7) | 19.4 | 1.72, s |
| 30a | 27.2 | 1.23, s | 110.4 | 4.71, m |
| 30b | | | | 4.60, m |
| 1' | 128.4 | | 128.5 | |
| 2' | 115.1 | 7.77, d (2.0) | 118.6 | 7.31, d (2.1) |
| 3' | 148.4 | | 145.9 | |
| 4′ | 149.5 | | 148.2 | |
| 5' | 115.7 | 6.78, d (8.2) | 115.8 | 6.74, d (8.2) |
| 6' | 126.7 | 7.14, dd (8.2, 2.0) | 124.7 | 7.02, dd (8.2, 2.1) |
| 7′ | 145.2 | 6.90, d (12.9) | 144.8 | 6.82, d (12.7) |
| 8' | 117.5 | 5.90, d (12.9) | 117.6 | 5.76, d (12.7) |
| 9′ | 168.7 | | 168.6 | |
| OCH ₃ -3' | 56.7 | 3.88, s | | |

cis-3-*O*-caffeoyl derivative of betulinic acid (17), exhibited more potent activity than 17 and its geometric isomer (19) (Table 3).²⁸

The neuroprotective activities of the bioactive entities (1-23) were also evaluated in terms of their inducing potential on nerve growth factor (NGF) secretion in C6 cells (Table 4).



Figure 3. CP3 analysis with experimental and calculated NMR chemical shifts of 1 and 4. Percentiles indicate the probabilities where the experimental 1 H, 13 C, or both NMR chemical shift values of 1 and 4 matched their respective computed values.

 Table 3. Cytotoxicity of Selected Compounds against Four

 Cultured Human Cancer Cell Lines in the SRB Bioassay

| | $IC_{50} (\mu M)^a$ | | | |
|------------------------|---------------------|---------|----------|-------|
| compound | A549 | SK-OV-3 | SK-MEL-2 | HCT15 |
| 3 | >10.0 | 8.4 | 2.0 | >10.0 |
| 17 | 3.1 | >10.0 | >10.0 | 8.7 |
| 19 | 9.9 | >10.0 | >10.0 | >10.0 |
| 20 | 0.9 | 7.5 | 6.7 | 3.8 |
| 21 | 5.1 | >10.0 | >10.0 | >10.0 |
| 23 | >10.0 | 9.0 | 2.9 | >10.0 |
| cisplatin ^b | 3.3 | 3.8 | 2.0 | 4.3 |

 $^a50\%$ inhibitory concentration; the concentration of compound that caused a 50% inhibition in cell growth. bC isplatin was used as a positive control.

Table 4. Effects of Compounds 11, 17, and 19 on NGF Secretion in C6 Cells

| compound | NGF secretion ^{a} (%) | cell viability ^b (%) |
|------------------------|---|---------------------------------|
| 11 | 167.4 ± 9.57 | 100.5 ± 4.07 |
| 17 | 177.6 ± 2.74 | 77.3 ± 4.18 |
| 19 | 355.7 ± 2.94 | 95.0 ± 0.18 |
| 6-shogaol ^c | 168.6 ± 7.16 | 125.8 ± 0.93 |

^{*a*}C6 cells were treated with 20 μ M of each compound. After 24 h, the content of NGF secreted in the C6-conditioned medium was measured by ELISA. The level of secreted NGF is expressed as the percentage of the untreated control (set as 100%). ^{*b*}Cell viability after treatment with 20 μ M of each compound was determined by an MTT assay and is expressed as a percentage (%). Results are the means of three independent experiments, and the data are expressed as mean \pm SD. ^{*c*}Positive control substance.

Compounds 17 and 19 strongly induced NGF release (177.6 \pm 2.74% and 355.7 \pm 2.94%, respectively), when compared with the positive control, 6-shogaol, 168.6 \pm 7.16%, but with no significant cell toxicity toward the C6 cells at the tested dose (20 μ M). These data suggest that the *trans*-oriented caffeoyl moiety substituted at C-3 of betulinic acid (17) might be a key functionality in augmenting NGF secretion. Contrary to this observation, 3-O-cis-caffeoylbetulinic acid (20) showed poor NGF secretion (52.49 \pm 9.53%) presumably because of its potent cytotoxicity (23.56 \pm 1.21%). Compound 11 also exhibited strong NGF secretion in C6 cells (167.37 \pm 9.57%), implying that some functionalized triterpenoids can serve as prototypes for the development of noncytotoxic neuroprotective agents.

The potential anti-inflammatory effects on the purified molecules (1-23) were also tested via the measurement of produced nitric oxide (NO) levels in lipopolysaccharide (LPS)-stressed murine microglia BV-2 cells (Table 5). Compounds 7

Table 5. Inhibitory Effect of Selected Compounds on NO Production in LPS-Activated BV-2 Cells

| compound | $IC_{50} (\mu M)^{a}$ | cell viability(%) ^b |
|---------------------|-----------------------|--------------------------------|
| 2 | 37.1 | 110.9 ± 7.63 |
| 5 | 17.8 | 30.4 ± 5.22 |
| 7 | 21.6 | 87.0 ± 4.30 |
| 12 | 47.1 | 100.3 ± 3.53 |
| 17 | 4.5 | 59.6 ± 3.18 |
| 18 | 14.5 | 71.0 ± 4.11 |
| 20 | 13.4 | 35.3 ± 2.16 |
| 21 | 25.5 | 88.0 ± 4.94 |
| L-NMMA ^c | 21.8 | 114.1 ± 2.96 |

^{*a*}IC₅₀ value of each compound was defined as the concentration (μ M) that caused 50% inhibition of NO production in LPS-activated BV-2 cells. ^{*b*}Cell viability following treatment with 20 μ M of each compound was determined using the MTT assay and is expressed as a percentage (%). Data are expressed as the mean ± SD of three independent experiments. ^{*c*}Positive control substance.

and **18** exhibited potent inhibition of NO production, with IC₅₀ values of 21.6 and 14.5 μ M, which were comparably potent to the positive control $N^{\rm G}$ -monomethyl-L-arginine (L-NMMA) (21.8 μ M), without significant cell toxicity. Although compounds **5**, **17**, and **20** also exhibited strong activity (17.8, 4.5, and 13.4 μ M), they diminished the cell viability of BV-2 cells with 30.4 ± 5.22%, 59.6 ± 3.18%, and 35.3 ± 2.16% at the tested concentration (20 μ M) as well. This implies that the observed inhibitory activity of **5**, **17**, and **20** on NO production was influenced by their cytotoxicity.

EXPERIMENTAL SECTION

General Experimental Procedures. Optical rotations were obtained on a JASCO P-1020 polarimeter (JASCO, Easton, MD, USA). UV spectra were acquired with a Shimadzu UV-1601 UVvisible spectrophotometer (Shimadzu, Tokyo, Japan). IR spectra were measured on a Bruker IFS-66/S Fourier-transform IR spectrometer (Bruker, Karlsruhe, Germany). NMR spectra were generated using a Bruker AVANCE III 700 NMR spectrometer (Bruker, Karlsruhe, Germany). The HRFABMS chromatograms were recorded on a Waters SYNAPT G2 (Waters, Milford, MA, USA) mass spectrometer. Semipreparative HPLC was conducted utilizing a Gilson 306 pump (Gilson, Middleton, WI, USA) and a Shodex refractive index detector (Shodex, New York, NY, USA) with a Phenomenex Luna C_{18} 10 μm column (10 \times 250 mm, Phenomenex, Torrance, CA, USA) and an Apollo silica 5 μ m column (10 × 250 mm, Apollo, Manchester, UK) at a flow rate 2 mL/min. LPLC was accomplished employing a LiChroprep Lobar-A Si 60 column (240 \times 10 mm, 40–63 μ m, Merck, Darmstadt, Germany) equipped with an FMI QSY-0 pump (Teledyne Isco, Lincoln, NE, USA). Open column chromatography was implemented with silica gel 60 (70-230 and 230-400 mesh; Merck, Darmstadt, Germany) and RP-C₁₈ silica gel (230-400 mesh, Merck, Darmstadt, Germany). TLC was carried out with precoated silica gel F254 plates and RP-18 F254s plates (Merck, Darmstadt, Germany).

Plant Material. Twigs of two-year-old *C. sinensis* were collected in Seoul, Republic of Korea, in January 2012. A voucher specimen for the plant (SKKU-NPL 1206) was authenticated by one of the authors (K.R.L.) and stored at the herbarium of the School of Pharmacy, Sungkyunkwan University, Suwon, Republic of Korea.

Extraction and Isolation. Twigs of C. sinensis (7.0 kg) were extracted three times using 80% aqueous MeOH for 1 day under reflux, and the combined extract was filtered. The filtrate was concentrated in vacuo to obtain a crude MeOH extract (320 g). The slurry was suspended in H₂O and sequentially extracted with hexanes, CHCl₃, EtOAc, and *n*-BuOH to produce 3, 15, 6, and 30 g of residues, respectively. The CHCl₃-soluble phase (15 g) was subjected to passage over a silica gel open column (CHCl₃–MeOH, 50:1 \rightarrow 1:1) to furnish nine pooled fractions (C1-C9). Fraction C1 (1.8 g) was chromatographed on an RP-C₁₈ silica gel column eluting with 60% aqueous MeOH to give 10 subfractions (C1-1-C1-10). Fraction C1-3 (50 mg) was purified using semipreparative HPLC, eluting with an isocratic mixture of 55% aqueous CH₃CN, to garner compounds 13 ($t_{\rm R}$ 15.5 min, 2 mg) and 14 ($t_{\rm R}$ 17.3 min, 3 mg). Fraction C1-5 (50 mg) was separated by semipreparative HPLC (80% aqueous MeCN) to obtain compounds 1 ($t_{\rm R}$ 16.6 min, 2 mg), 4 ($t_{\rm R}$ 12.2 min, 2 mg), 8 ($t_{\rm R}$ 15.8 min, 2 mg), and 15 (t_R 11.1 min, 3 mg). Compound 17 (t_R 20.3 min, 6 mg) was acquired upon the purification of fraction C1-6 (50 mg) with semipreparative HPLC (90% aqueous MeCN). Fraction C3 (1.6 g) was subjected to separation on an RP-C₁₈ silica gel open column, eluting with 60% aqueous MeOH, to give 10 subfractions (C3-1-C3-10). Among those subfractions, compounds 2 ($t_{\rm R}$ 16.2 min, 2 mg), 7 $(t_{\rm R}$ 12.9 min, 11 mg), and 16 $(t_{\rm R}$ 19.2 min, 2 mg) were purified from fraction C3-7 (50 mg) by a semipreparative HPLC (90% aqueous MeOH). Fraction C3-8 (40 mg) was purified employing semipreparative HPLC, with an isocratic solvent system (90% aqueous CH₃CN), to furnish compounds 21 (t_R 19.9 min, 11 mg) and 22 (t_R 21.2 min, 2 mg). Fraction C3-9 (60 mg) was purified with semipreparative HPLC, eluting with 80% aqueous CH₃CN, to afford compound 20 ($t_{\rm R}$ 13.3 min, 5 mg). Compounds 3 ($t_{\rm R}$ 20.3 min, 10 mg) and 23 ($t_{\rm R}$ 22.9 min, 11 mg) were isolated upon purification of fraction C3-10 (100 mg) by an isolation strategy using semipreparative HPLC (85% aqueous CH₃CN). Fraction C4 (1.5 g) was fractionated into 12 subfractions (C4-1-C4-12) using an RP-C₁₈ silica gel open column eluting with 70% aqueous MeOH. Fraction C4-6 (20 mg) was purified by semipreparative HPLC (75% aqueous MeOH) to acquire compound 10 ($t_{\rm R}$ 21.2 min, 2 mg). Fraction C4-7 (120 mg) was separated using semipreparative HPLC (CHCl₃-MeOH, 30:1) to yield compounds 6 (t_R 15.6 min, 4 mg) and 9 (t_R 11.3 min, 7 mg). Fraction C4-9 (50 mg) was purified by semipreparative HPLC (CHCl₃-MeOH, 30:1) to yield compounds 11 (t_R 17.7 min, 10 mg) and 18 ($t_{\rm R}$ 18.8 min, 20 mg). Compound 5 ($t_{\rm R}$ 10.3 min, 20 mg) was obtained from fraction C4-10 (110 mg) employing semipreparative HPLC with a solvent mixture of CHCl₃-MeOH (30:1). Compounds 12 ($t_{\rm R}$ 19.3 min, 3 mg) and 19 ($t_{\rm R}$ 17.6 min, 2 mg) were afforded from fraction C4-11 (200 mg) using a LiChroprep Lobar-A Si 60 column followed by semipreparative HPLC (70% aqueous CH₃CN).

Sinenic Acid Ā (1): white powder; $[\alpha]_{D}^{25}$ –11.0 (c 0.1, MeOH); IR (KBr) ν_{max} 3395, 2944, 2870, 1640 cm⁻¹; ¹H (700 MHz) and ¹³C (175 MHz) NMR data in pyridine- d_5 , see Table 1; HRFABMS (negative-ion mode) m/z 469.3313 [M – H]⁻ (calcd for C₃₀H₄₅O₄, m/z 469.3312).

3β-O-cis-Feruloyl-2α,19α-dihydroxyurs-12-en-28-oic acid (2): white powder; $[α]_D^{25}$ +35.0 (c 0.1, MeOH); UV (MeOH) λ_{max} (log ε) 285 (2.80), 229 (2.75) nm; IR (KBr) ν_{max} 3430, 2950, 2880, 1716, 1700, 1625, 1588, 1512 cm⁻¹; ¹H (700 MHz) and ¹³C (175 MHz) NMR data in methanol- d_4 , see Table 2; HRFABMS (negative-ion mode) m/z 663.3892 [M – H]⁻ (calcd for C₄₀H₅₅O₈ m/z 663.3891).

3β-O-*cis*-Caffeoylbetulin (**3**): white powder; $[\alpha]_D^{25}$ +44.0 (*c* 0.1, MeOH); UV (MeOH) λ_{max} (log ε) 288 (3.05), 231 (2.90) nm; IR (KBr) ν_{max} 3500, 2930, 2888, 1700, 1632, 1600, 1515 cm⁻¹; ¹H (700 MHz) and ¹³C (175 MHz) NMR data in methanol-*d*₄, see Table 2; HRFABMS (negative-ion mode) *m*/*z* 603.4044 [M – H]⁻ (calcd for C₃₉H₅₅O₅, *m*/*z* 603.4044).

Sinenic acid B (4): white powder; $[\alpha]_{25}^{25}$ +76.0 (*c* 0.1, MeOH); IR (KBr) ν_{max} 3401, 2940, 2866, 1639 cm⁻¹; ¹H (700 MHz) and ¹³C (175 MHz) NMR data in pyridine-*d*₅, see Table 1; HRFABMS (negative-ion mode) *m*/*z* 469.3311 [M - H]⁻ (calcd for C₃₀H₄₅O₄, *m*/*z* 469.3312).

Alkaline Hydrolysis of 2 and 3. Compounds 2 and 3 (each 1 mg) were individually hydrolyzed using 0.1 N KOH (2 mL) and stirred at room temperature for 12 h. The reaction mixture was neutralized with Dowex HCR W2 (H⁺ form) resin, and the resin was removed by filtration. A portion of the reaction product was partitioned between $CHCl_3$ -H₂O (1 mL each). The $CHCl_3$ -soluble phase was purified by semipreparative HPLC (90% aqueous MeCN) to afford the aglycones 2a (=11) and 3a (=22), which were identified utilizing ¹H NMR and MS data as tormentic acid and betulin, respectively.^{13,14}

Cytotoxicity Assessment. The cytotoxicity of the compounds addressed in this study against the cultured human tumor cell lines A549, SK-OV-3, SK-MEL-2, and HCT15 was evaluated using the SRB assay.²⁹ Cells addressed in the current study were purchased from the American Type Culture Collection (Manassa, VA, USA) and maintained at the Korea Research Institute of Chemical Technology.

NGF and Cell Viability Assays. The C6 glioma cells (the Korean Cell Line Bank, Seoul, Republic of Korea) were used to assess the release of NGF into the culture medium. The test cells were seeded onto 24-well plates at a density of 1×10^5 cells/well. After 24 h, the cells were treated with serum-free DMEM and incubated with various concentrations of compounds for an additional 24 h. The medium supernatant was collected from the culture plates, and NGF levels were evaluated using an ELISA development kit. Cell viability was also assessed with a 3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide (MTT) assay, in which the results were expressed as a percentage of the control group (untreated cells).

NO Production and Viability in LPS-Stressed BV-2 Cells. The inhibitory effect of the test compounds on LPS-stimulated NO production was scrutinized using BV2 cells, as developed by Dr. V. Bocchini at the University of Perugia (Perugia, Italy).^{30,31} The target cells were seeded on a 96-well plate at 4×10^4 cells/well and treated with/without different concentrations of the purified molecules. LPS (100 ng/mL) was added to BV2 cells, and they were incubated for 24 h. Nitrite (NO₂, soluble oxidation product of NO) concentrations present in the culture medium were measured using the Gries reagent (0.1% N-1-naphthylethylenediamine dihydrochloride and 1% sulfanilamide in 5% phosphoric acid). The supernatant (50 μ L) was mixed with an identical volume of the Gries reagent. After 10 min, absorbance at 570 nm was measured utilizing a microplate reader (Emax, Molecular Devices, Sunnyvale, CA, USA). Generated nitrite concentrations were gauged using graded sodium nitrite solution as a standard. Cell viability was evaluated using the MTT assay.

Computational Analysis.^{16,32} All conformers addressed in the study were found using the Macromodel (version 2015-2, Schrodinger LLC) module with "mixed torsional/low-mode sampling" in the MMFF94 force field. The searches were initially implemented in the gas phase with a 50 kJ/mol energy window limit and 10 000 maximum number of steps to exhaustively explore all potential conformers. The Polak-Ribiere conjugate gradient protocol was utilized to minimize conformers with 10 000 maximum iterations and a 0.001 kJ (mol Å)⁻¹ convergence threshold on the rms gradient. Conformers within 10 kJ/ mol of each global minimum were not subjected to geometry optimization steps and proceeded to gauge-invariant atomic orbital (GIAO) shielding constant calculations using the Gaussian 09 package (Gaussian Inc.) at the B3LYP/6-31G(d,p) level in the gas phase. The calculated NMR properties of compounds 1 and 4 were Boltzmannaveraged given their respective Boltzmann populations (Tables S1-S3, Supporting Information) and used for calculations of CP3 probability analysis facilitated by the applets available at http://www-jmg.ch.cam. ac.uk/tools/nmr/. The stereo images in Figure 2 were rendered using Pymol 1.6.x (Schrödinger LLC).

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jnat-prod.7b00111.

Additional information (PDF)

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Notes

The authors declare no competing financial interest.

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