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A NOVEL, STEREOCONTROLLED APPROACH TO RING B ALKYLATED ESTRATETRAENES

G. Bojack,^{1a} H. Künzer,* K. Rölfing,^{1b} and M. Thiel

Research Laboratories, Schering AG-Berlin, Müllerstrasse 170-178, D-13342 Berlin, Germany

Summary. A four-step reaction protocol, culminating in an oxy-Cope rearrangement, has been developed to transform 2 into 7. 9α -Alkylated derivatives of equilin, e.g., 12, as well as C(7)-C(9) propano-bridged 19-norsteroids, like 17, demonstrate synthetic potential for 7 in estrogen receptor ligand synthesis. Copyright © 1996 Elsevier Science Ltd

Soon after their discovery some sixty years ago, equine estrogens have found widespread use in hormone replacement therapy.² Their impact on synthetic steroid chemistry, however, has been modest although the structurally unique equilin skeleton (1) is nicely set up for intriguing backbone modifications of potential pharmacological interest. Based on our earlier observation³ that 2 readily enters into ene-type α -hydroxy-alkylation at C(7), we have devised a novel approach to C(9) α -alkylated equilin derivatives⁴ and certain C(7)-C(9) propano-bridged 19-norsteroids. Both stratagems have now been exploited in estrogen receptor ligand syntheses, two representative examples of which are outlined below.

Attachment of a functionalized three-carbon chain to C(9) was envisioned to occur indirectly by way of an anion-accelerated [3,3]-sigmatropic reorganization⁵ involving an appropriate C(7) α -substituted intermediate. While the most straightforward access, ene reaction between 2 and acrolein, was prohibitively ineffective in our hands, a three-step alternative afforded ample quantities of the substrate(s) required for oxy-Cope rearrangement. Thus, oxidation of the known equilol-paraformaldehyde adduct 3³ (DMSO, Py-SO₃, (C₂H₅)₃N, 22 °C; 77%), $3\rightarrow 4$, followed by vinylmagnesium bromide addition (THF, CH₂CHMgBr, -78 °C; 66% combined yield), gave a separable mixture (silica gel; *t*-butyl methyl ether/hexane, 1:4; acetone/hexane, 1:9) of diastereoisomeric allylic alcohols 5 and 6.



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15 X= H

Both compounds furnished the same rearranged key intermediate 7 upon exposure to potassium hydride/18crown-6 in tetrahydrofuran at 0 °C for two hours. Best results materialized (83% yield) when the reaction was quenched inversely into vigorously stirred ice/water via cannula. Otherwise, rearrangement efficacy was severely eroded as a consequence of aldehyde enolate/aldehyde selfcondensation.

Having established a reliable access to 7, we next focused on estrogen receptor ligands derived thereof. In a first series of experiments, formation of a 9 α -propyl analogue was addressed. The desired side-chain oxidation state was adjusted by formyl group reduction (CH₃OH, NaBH₄, 0 °C; 93%), 7 \rightarrow 8, tosylation (CHCl₃, pyridine, *p*-TsCl, 0 °C, 24 h; 84%), 8 \rightarrow 9, and tosylate displacement (THF, LiB(C₂H₅)₃H, 22 °C, 1 h; 88%), 9 \rightarrow 10. Removal of the *t*-butyldimethylsilyl protecting group (THF, HCl, H₂O, 22 °C, 16 h; 91%) and cleavage of the phenolic methyl ether (toluene, DIBAH, reflux, 2 h; 95%) completed the synthesis of 12. Further investigations into the synthetic potential of 7 revealed an attractive option to bridge positions C(7) and C(9) by a three-carbon unit.⁶ Treatment of 7 with a catalytic amount of *p*-toluenesulfonic acid in dichloromethane (22 °C, 1 h) triggered off a remarkably facile, stereocontrolled intramolecular ene-type reaction. The pentacyclic steroid 13 was obtained in 84% yield. Attempts to rid the bridge of the sterically hindered secondary hydroxyl group via mesylation (CHCl₃, pyridine, CH₃SO₂Cl, 22 °C, 48 h), 13 \rightarrow 14, and substitution/reduction (diglyme, H₂O, NaI, Zn, 100 °C, 2 h),⁷ 14 \rightarrow 15, had to cope with an elimination by-product in an additional hydrogenation step (C₂H₅OH, CH₃CO₂C₂H₅, Pd/C, H₂, 22 °C, 4 h; 80% overall). The deprotection scheme utilized above delivered 17 in 91% overall yield for these last two operations.⁸

Estrogen receptor affinity for both 12 and 17 turned out to be low.

References and Notes

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- Propano-bridging has been achieved across other centers of the steroid skeleton. See, e.g., (a) Pitt, C. G.; Rector, D. H.; Cook, C. E.; Wani, M. C. J. Med. Chem. 1979, 22, 966. (b) Bull, J. R.; Mountford, P. G. Synlett 1994, 711.
- 7. (a) Fujimoto, Y.; Tatsuno, T. Tetrahedron Lett. 1976, 17, 3325. (b) Kocovsky, P.; Cerny, V. Collection Czechoslov. Chem. Commun. 1979, 44, 246.
- 8. Physical data for selected steroids are as follows. 4: mp 111-113 °C (pentane); $[\alpha]_D^{22}$ +268.8° (c 0.51, CH₂Cl₂). 8: amorphous; $[\alpha]_D^{22}$ +40.0° (c 0.51, CHCl₃). 9: mp 163-164 °C (acetone/hexane); $[\alpha]_D^{22}$ +32.0° (c 0.52, CHCl₃). 12: mp 202-204 °C (acetone/hexane); ¹³C NMR (75 MHz, C₅D₅N) δ 156.0, 139.2, 135.4, 134.6, 127.3, 117.0, 114.7, 114.1, 81.2, 45.5, 45.0, 43.3, 42.0, 37.1, 34.8, 30.6, 30.3, 21.8, 17.5, 14.3, 11.3; $[\alpha]_D^{22}$ +64.9° (c 0.51, CH₃OH). 16: amorphous; $[\alpha]_D^{22}$ +65.0° (c 0.51, CHCl₃). 17: mp 212-213 °C (acetone/hexane); ¹³C NMR (75 MHz, CD₃OD) δ 155.5, 140.1, 138.2, 135.9, 134.3, 126.9, 114.3, 114.2, 83.0, 44.4, 44.4, 40.1, 38.0, 35.9, 34.2, 33.8, 33.4, 30.0, 23.5, 21.2, 17.4; $[\alpha]_D^{22}$ +71.0° (c 0.47, CH₃OH).

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