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Antimycobacterial activities of novel 2-(sub)-3-fluoro/nitro-5, 12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid

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Abstract—Various 2-(sub)-3-fluoro/nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid derivatives were synthesized from 2-aminothiophenol by a five-step reaction, evaluated for in-vitro and in-vivo antimycobacterial activities against *Mycobacterium tuberculosis* (MDR-TB), and *Mycobacterium smegmatis* (MC2), and also tested for the ability to inhibit the supercoiling activity of DNA gyrase from *M. smegmatis*. Among the thirty-four synthesized compounds, 2-(3-(diethylcarbamoyl)piperidin-1-yl)-)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (7I) was found to be the most active compound in vitro with MIC of 0.18 and 0.08 μ M against MTB and MTR-TB, respectively. Compound 7I was found to be 2 and 570 times more potent than isoniazid against MTB and 3.12 – log 10 protections, respectively, at the dose of 50 mg/kg body weight. © 2007 Elsevier Ltd. All rights reserved.

1. Introduction

Tuberculosis (TB) is a worldwide pandemic: about 2 billion people, equal to one-third of the world's total population, are infected with Mycobacterium tuberculosis (MTB), the microbes that cause TB. TB is a leading killer among HIV-infected people with weakened immune systems; about 200,000 people living with HIV/AIDS die from TB every year. Multidrug-resistant TB (MDR-TB) is a form of TB that does not respond to the standard treatments using first-line drugs; MDR-TB is present in virtually all countries recently surveyed by World Health Organization (WHO) and partners. 450,000 new MDR-TB cases are estimated to occur every year; the highest rates of MDR-TB are in countries of the former Soviet Union and in China.¹ The WHO declared TB a global health emergency in 1993, and the Stop TB Partnership proposed a Global Plan to stop TB which aims to save 14 million lives between 2006 and 2015.² In the course of screening to discover

new compounds employed in the chemotherapy of tuberculosis,^{3,4} we also identified benzothiazolo[3,2-*a*]quinoline-6-carboxylic acid derivatives which inhibited in-vitro *Mycobacterium tuberculosis* H_{37} Rv (MTB) and multi-drug resistant *Mycobacterium tuberculosis* (MDR-TB). We report herein the results concerning the synthesis of newer benzothiazolo[3,2-*a*]quinoline-6-carboxylic acid derivatives by modifying 2nd position with various unreported bulky secondary amino functions to study the influence of lipophilic character at 2nd position on activity against MTB and also to study the antimycobacterial effect of fluoro/nitro groups at 3rd position, in-vitro and in-vivo antituberculous activity of first representative compounds of this family together with toxicological results.

2. Results and discussion

2.1. Synthesis

Several of the quinolone antibacterials, such as moxifloxacin and gatifloxacin, have been examined as inhibitors of MTB.⁵ However, no large systematic study has been undertaken to either optimize the fused quinolone antibacterials against MTB or examine

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specific structure-activity relationship of the fused quinolone against MTB. Lipophilic phenyl substituents at N-1 position of simple fluoroquinolone were reported earlier⁶ and also substitution of a sulfur atom at the 2 position of 1.4-dihydro-4-oxoquinoline-3-carboxylic acid derivatives can lead to active compounds.⁷ Hence, a conformationally restricted rigid analog formed by bridging the phenyl and quinolone rings by a sulfur atom would provide a useful indication of the relevance of the planarity of the ring system to antimycobacterial activity. The titled compounds were synthesized from 2aminothiophenol by a five-step synthesis. 2-Aminothiophenol (1), on reaction with ethyl malonyl chloride (2) in presence of triethylamine, yielded ethyl 2-(benzothiazol-2-yl)acetate (3) via two intermediates ethyl 2-(2mercaptophenylcarbamoyl)acetate and its enol tautomer. Compound 3 on reaction with 2,4-dichloro-5-fluoro benzoyl chloride/2,4-dichloro-5-nitro benzoyl chloride in presence of sodium hydride vielded ethyl 2-(benzothiazol-2-vl)-3-(2,4-dichloro-5-fluoro/nitrophenyl)-3-oxopropanoates (4a-b). Compounds 4a-b undergo cyclization in presence of sodium hydride in ethylene glycol dimethyl ether yielding ethyl 2-Chloro-3-fluoro/nitro-5,12-dihydro-5oxobenzothiazolo[3,2-a]quinoline-6-carboxylates (5a **b**), which on hydrolysis with 2 N sodium hydroxide yielded 2-chloro-3-fluoro/nitro-5,12-dihydro-5-oxobenzothiazolo [3,2-a]quinoline-6-carboxylic acids (6a-b). The titled compounds 7-8a-s were prepared by treating 6a-b with appropriate secondary amines in presence of potassium carbonate under microwave irradiation in DMSO. When compared to conventional method⁸ of 2-3 h process, microwave-assisted synthesis was performed with short reaction times (2-3 min) with ease and was environment friendly. The purity of the synthesized compounds was monitored by thin layer chromatography (TLC) and elemental analyses, and the structures were identified by spectral data.

2.2. Antimycobacterial activity

The compounds were screened for their in-vitro antimycobacterial activity against MTB, MDR-TB, and *M. smegmatis* ATCC 14468 (MC²) by agar dilution method for the determination of MIC in duplicate.⁹ The MDR-TB clinical isolate was resistant to isoniazid, rifampicin, ethambutol, and ofloxacin. The minimum inhibitory concentration (MIC) is defined as the minimum concentration of compound required to give complete inhibition of bacterial growth, and MICs of the synthesized compounds along with the standard drugs for comparison are reported in Tables 1 and 2.

In the first phase of screening against MTB, all the compounds showed excellent in-vitro activity against MTB with MIC of less than 12.82 μ M. Five compounds (7i, 7j, 7l, 7m, and 7s) inhibited MTB with MIC of less than 1 μ M and were more potent than standard fluoroquinolone gatifloxacin (MIC: 1.04 μ M). When compared to isoniazid (MIC: 0.36 μ M), three compounds (7i, 7j, and 7s) were equally active and one compound (7l) was found to be more active against MTB. Compound 2-(3-(diethylcarbamoyl)piperidin-1-yl)-)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (71) was found to be the most active compound in vitro with MIC of 0.18 µM against MTB and was 2 and 5.8 times more potent than isoniazid and gatifloxacin, respectively. Subsequently some of the compounds were evaluated against MDR-TB, and among the twenty-two compounds screened all the compounds inhibited MDR-TB with MIC ranging from 0.08 to 7.55 μ M and were found to be more active than isoniazid (MIC: 45.57 µM) and gatifloxacin (MIC: 8.34 µM). Elevan compounds (7a, 7i, 7j, 7l, 8l, 7m, 8m, 7o, 8o, 7s, and 8s) inhibited MDR-TB with MIC of less than 1 µM. Compound 71 was found to be the most active compound in vitro with MIC of $0.08 \,\mu\text{M}$ against MDR-TB and was 104 and 570 times more potent than gatifloxacin and isoniazid, respectively. The compounds were also evaluated against MC^2 in which all the compounds inhibited MC^2 with MIC ranging from 2.79 to 113.26 µM and sixteen compounds were found to be more active than isoniazid (MIC: 45.57 µM).

With respect to structure-MTB activity relationship, at C₂ position we have studied with various substituted piperazines (7–8a–f), (thio) morpholines (7–8g–h), substituted piperidines (7–8i–l), fused piperazines, piperidines, and pyrrolidine (7–8m–r). A comparison of the substitution pattern at C₇ demonstrated that the order of activity was substituted piperidines > fused piperazines, pyrrolidine, and piperidines > substituted piperazines, (thio) morpholines. With respect to C₃ position among the fluoro (7a–r) and nitro (8a–r) substituents, in most of the cases compounds with fluoro substituents were found to be more active against MTB.

2.3. In-vitro cytotoxicity

Some compounds were further examined for toxicity (IC_{50}) in a mammalian Vero cell line upto 62.5 µg/ml concentrations. After 72 h of exposure, viability was assessed on the basis of cellular conversion of MTT into a formazan product using the Promega Cell Titer 96 non-radioactive cell proliferation assay¹⁰ and the results are reported in Tables 1 and 2. Twenty compounds when tested showed IC₅₀ values ranging from 50.0 to 130.3 μ M. A comparison of the substitution pattern at C₃ demonstrated that nitro substituted analogs were more cytotoxic than the fluoro substituted analogs. These results are important as the nitro group substituted compounds with their increased cytoliability are much less attractive in the development of a quinolone for the treatment of TB. This is primarily due to the fact that the eradication of TB requires a lengthy course of treatment, and the need for an agent with a high margin of safety becomes a primary concern. The IC₅₀ for the compound 7l was found to be 126.1 μ M and showed selectivity index (IC₅₀/MIC) of 1576 against MDR-TB.

2.4. In-vivo antimycobacterial activity

Subsequently, compound 71 was tested for efficacy against MTB at a dose of 50 mg/kg (Table 3) in CD-1 mice. In this model,¹¹ the mice were infected intravenously with *M. tuberculosis* ATCC 35801. Drug treatment by intraperitoneal route began after 10 days of





Compound	R	R ₁	Yield (%)	Mp (°C)	IC50 (µM)	MIC (µM)		
						MTB	MDRTB	MC ²
7a	F		70	253–255	104.5	2.61	0.65	10.45
8a	NO ₂		75	209–210	50.0	5.01	NT	39.99
7b	F		77	267–268	NT	6.37	NT	50.87
8b	NO ₂		76	207–210	NT	6.04	NT	48.22
7c	F		73	181–182	>117.6	1.47	5.89	5.89
8c	NO ₂		72	209–210	111.9	5.60	NT	44.76
7d	F		75	259–261	NT	12.82	NT	102.55
8d	NO ₂	v _e n₅	73	287–290	NT	12.15	NT	97.17
7e	F		77	237–238	111.7	1.39	5.59	2.79
8e	NO ₂	F	75	233–235	NT	10.66	NT	85.24
7f	F		75	>300	NT	2.52	5.06	40.42
8f	NO ₂	F Of Org	72	281-283	NT	4.85	NT	77.45

 Table 1 (continued)

Compound	R	R ₁	Yield (%)	Mp (°C)	IC ₅₀ (µM)	MIC (µM)		
						MTB	MDRTB	MC^2
7g	F	\$N	76	277–278	NT	3.76	7.55	7.55
8g	NO ₂	\	74	268–271	NT	7.09	NT	113.26
7h	F		70	256–260	NT	7.34	NT	29.31
8h	NO ₂	/	79	237–238	NT	6.90	NT	110.26
7i	F		72	260–262	130.3	0.39	0.19	6.53
8i	NO ₂		75	294–296	61.7	3.08	1.54	98.71

NT, Not tested.

inoculation of the animal with microorganism and was continued for 10 days. After 35 days post infection the spleens and right lungs were aseptically removed, the number of viable organisms was determined and compared with the counts from negative (vehicle treated) controls (Mean culture forming units (CFU) in lung: 7.99 ± 0.16 and in spleen: 9.02 ± 0.21). Compound **71** decreased the bacterial load in lung and spleen tissues with 2.78 and $3.12 - \log 10$ protections, respectively and was considered to be promising in reducing bacterial count in lung and spleen tissues. When compared to gatifloxacin at the same dose level 71 decreased the bacterial load with 0.81 and $1.02 - \log 10$ protections in lung and spleen tissues, respectively. Compound 71 was found to be less active than isoniazid in the in-vivo study. The reason for this less in-vivo activity might be the low bioavailability of the compound (Scheme 1).

2.5. DNA gyrase inhibition

The benzthiazoloquinolone derivatives synthesized and studied in this report were tested for their ability to inhibit supercoiling activity of DNA gyrase. The bacterial targets for quinolones and fluoroquinolones are the type II DNA topoisomerases, DNA gyrase, and topoisomerase IV. These ATP-dependent enzymes act by a transient double-stranded DNA break, followed by strand passage and religation reactions to facilitate DNA transaction processes.¹² DNA gyrase is unique in catalyzing the negative supercoiling of DNA and is essential for DNA replication, transcription, and recombination. In all the species of mycobacteria including MTB, DNA gyrase is the sole type II topoisomerase carrying out the reactions of both the type II topoisomerases. Further, our earlier studies have revealed that DNA gyrase from *M. tuberculosis* and that from *M. smegmatis* are highly similar at protein level, antigenic properties, and catalytic activities.¹³ The supercoiling assay results with various compounds using M. smegmatis DNA gyrase are presented in Figure 1. The IC₅₀ values presented in Table 4 show that the compounds **7j**, **7l**, and **7r** inhibit supercoiling activity with a IC₅₀ value of 40 µg/ml. The other two compounds tested, that is, **7e** and **7i**, have IC₅₀ values >50 µg/ml, which is much higher than that of the positive control moxifloxacin though these compounds show comparable cell killing activity against MTB, MDR-MTB and MC².

2.6. Phototoxic evaluation

Quinolones in general have favorable safety profiles; phototoxicity has become a significant factor in the clinical use of some¹⁴ quinolones. Indeed, the first quinolone, nalidixic acid, caused light-induced dermal effects. This type of response has now been demonstrated for almost all Fluoroquinolones,¹⁵ although the relative phototoxic potential varies greatly among compounds. Phototoxicity is considered to be an acute, light-induced irritation response characterized by dermal inflammation, with erythema and edema as primary clinical endpoints. Phototoxicity with the quinolones is generally thought to result from the absorption of light by the parent compound or a metabolite in tissue.¹⁶ This photosensitized chromophore may then transfer its absorbed photo energy to oxygen molecules, creating an environment for the production of reactive oxygen species such as singlet oxygen. These reactive species are then thought to attack cellular lipid membranes, initiating the inflammatory process.

Three (**7**j, **7**l, **7m**) compounds were evaluated for potential phototoxicity in a standardized in-vivo test system that has been used previously to assess quinolone antibiotics.¹⁷ The test compounds (140 mg/kg) and the positive control lomefloxacin hydrochloride (140 mg/kg) were evaluated for phototoxicity and both ears of each mouse were evaluated for changes indicative of a positive response: erythema, edema or a measurable increase in ear thickness. Change from baseline was calculated sep-





Compound	R	R ₁	Yield (%)	Mp (°C)	IC ₅₀ (µM)			
						MTB	MDRTB	MC^2
7j	F		79	229–232	119.5	0.36	0.36	2.98
8j	NO ₂		70	214–215	113.6	2.84	2.84	90.91
7k	F		75	233–235	NT	2.77	5.56	44.41
8k	NO ₂	0	74	224–225	NT	5.31	NT	84.75
71	F	(C ₂ H ₆) ₂ N C N	70	256–257	126.1	0.18	0.08	3.15
81	NO ₂		77	227–230	59.8	2.99	0.75	95.68
7m	F		78	280–283	>137.5	0.86	0.86	6.89
8m	NO ₂		74	198–199	129.8	1.62	0.39	103.85
7n	F		71	271–272	115.0	2.87	1.43	45.99
8n	NO_2	$O = C - NHC(CH_3)_3$	76	202–205	NT	2.73	5.49	43.81
70	F	HOOC	80	263–265	>130.6	3.26	0.39	52.25
80	NO ₂	$\frown \frown \frown$	79	255–258	123.65	1.54	0.77	98.92
7p	F		71	264–266	NT	5.50	NT	43.96
8p	NO_2		79	219–220	NT	10.49	NT	83.94

 Table 2 (continued)

Compound	R	R ₁	Yield (%)	Mp (°C)	IC ₅₀ (µM)			
						MTB	MDRTB	MC^2
7q	F	H ₃ CO	74	>300	>123.9	1.55	3.09	6.20
8q	NO ₂		76	237-240	117.6	1.47	2.93	47.04
7r	F	t BOC NH		258–259	>122.7	0.37	0.18	6.14
8r Gati INH	NO ₂	_		239–240 —	58.24 >155.3 >455.8	2.91 1.04 0.36	0.73 8.34 45.57	93.19 2.08 45.57

 Table 3. In vivo activity data of 7l, gatifloxacin, and isoniazid against

 M. tuberculosis ATCC 35801 in mice

Compound	Lungs (log CFU ± SEM)	Spleen (log CFU ± SEM)
Control	7.99 ± 0.16	9.02 ± 0.21
Gatifloxacin (50 mg/kg)	6.02 ± 0.23	6.92 ± 0.07
Isoniazid (25 mg/kg)	5.86 ± 0.23	4.71 ± 0.10
71 (50 mg/kg)	5.21 ± 0.12	5.9 ± 0.06

arately for each animal and time point and analyzed for statistical significance and are presented in Table 5. The drug and time factors were analyzed by separate univariate methods. Orthogonal contrasts were used to test for both linear and quadratic trends over time in each group by Student's t-tests to test whether the change from baseline ear thickness was significantly different from zero. The results indicated that lomefloxacin showed significant increase in ear thickness from 4 to 96 h and 24 to 96 h when compared within time points and with the control, respectively. No significant difference in ear thickness was observed with the test compounds at various time-points when compared with the pre-drug reading (0 h) and were non-toxic when compared with the negative (vehicle-treated) and positive controls (lomefloxacin). No erythema occurred in mice dosed with 140 mg/kg of 7j, 7l, and 7m throughout the 96 h study.

3. Materials and methods

Melting points were taken on an electrothermal melting point apparatus (Buchi BM530) in open capillary tubes and are uncorrected. Infrared spectra (KBr disk) were run on Jasco IR Report 100 spectrometer. ¹H NMR spectra were scanned on a JEOL Fx 300 MHz NMR spectrometer using DMSO- d_6 as solvent. Chemical shifts are expressed in δ (ppm) relative to tetramethylsilane. Elemental analyses (C, H, and N) were performed on Perkin-Elmer model 240 C analyzer and the data were within $\pm 0.4\%$ of the theoretical values.

3.1. Synthesis of ethyl 2-(benzothiazol-2-yl)acetate (3)

Ethyl malonyl chloride (5.0 g) (2) is added to a solution of 2-aminothiophenol (3.6 ml) (1) and triethylamine (4.6 ml) in anhydrous ether, and the reaction mixture is stirred for 3 h at room temperature. The triethylamine hydrochloride salt is filtered and washed with diethyl ether. The product is purified by distillation yielding 5.4 g of compound 3 (mp: 190–192 °C).

3.2. Synthesis of ethyl 3-(benzothiazol-2-yl)-4-(2,4-dichloro-5-fluoro/nitrophenyl)-2,4-dioxobutanoate (4a-b)

Compound **3** (5.4 g) dissolved in anhydrous tetrahydrofuran (100 ml) is added slowly to a cold solution of sodium hydride (60% in mineral oil) (1.1 g) suspended in anhydrous tetrahydrofuran. After stirring for 1 h, a solution of 2,4-dichloro-5-fluoro benzoyl chloride/2,4dichloro-5-nitro benzoyl chloride in anhydrous tetrahydrofuran is added dropwise over a period of 15 min. The reaction mixture is stirred at 25 °C for 2 h. The solvent is removed under reduced pressure, and the residue is redissolved in ethyl acetate. The organic layer is washed with water and saturated aqueous sodium chloride, and then dried over magnesium sulfate. Removal of solvent in vacuo yields a solid which is recrystallized to give compounds **4a–b.** (**4a**: Yield: 83%; mp: 210–213 °C, **4b**: Yield: 79%; mp: 219–222 °C).

3.3. Synthesis of ethyl 2-chloro-3-fluoro/nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (5a-b)

Compounds **4a–b** (approximately 4.0 g) dissolved in ethylene glycol dimethyl ether (75 ml) are added slowly to a solution of sodium hydride (60% in mineral oil) (0.42 g)



Scheme 1. Synthetic protocol of compounds. Reagents: (a) $(C_2H_5)_3N$; (b) NaH, 2,4-dichloro-5-fluoro benzoyl chloride/2,4-dichloro-5-nitro benzoyl chloride; (c) NaH; (d) 2N NaOH; (e) MWI.

in ethylene glycol dimethy ether (75 ml). The mixture is heated at 160 °C for 30 min. The reaction mixture is allowed to cool to room temperature and then treated with cold water. The precipitate is collected and washed with water yielding compounds **5a–b** (**5a**: Yield: 62%; mp: 220–221 °C, **5b**: Yield: 67%; mp: 230–234 °C).

3.4. Synthesis of 2-chloro-3-fluoro/nitro-5,12-dihydro-5oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (6ab)

Compounds **5a–b** (approximately 0.5 g) in tetrahydrofuran (100 ml) were treated with 2 N sodium hydroxide (15 ml) and the mixture was heated at 85 °C for 6 h and was cooled. The precipitate was filtered and washed with ether followed by cold water. The solid was dissolved in trifluoroacetic acid (15 ml) and water (500 ml) was added, and the precipitate was filtered yielding **6a–b** (**6a**: Yield: 92%; mp: >300 °C, **6b**: Yield: 94%; mp: >300 °C).

3.5. Synthesis of 2-(sub)-3-fluoro/nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (7-8a-s)

Compounds 6a-b (1.0 equiv) in dimethylsulfoxide (2.5 ml) and appropriate secondary amines (4 equiv)

were irradiated in a microwave oven at an intensity of 80% with 30 s/cycle. The number of cycle in turn depended on the completion of the reaction, which was checked by TLC. The reaction timing varied from 1.5 to 3 min. After completion of the reaction, the mixture was poured into ice-cold water and washed with water and isopropanol to give titled products 7-8a-r.

3.5.1. 2-(4-((4-Chlorophenyl)(phenyl)methyl)piperazin-1yl)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (7a). Yield: 70%; mp: 253–255 °C; IR (KBr) cm⁻¹: 2910, 1722, 1710, 1620, 1460–1370; ¹H NMR (DMSO- d_6) δ ppm: 2.59–3.12 (m, 8H, CH₂ of piperazine), 4.22 (s, 1H, CH of diphenylmethyl), 5.9 (s, 1H, C₁-H), 6.3–7.15 (m, 13H, Ar-H), 7.26 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₃₃H₂₅ClFN₃O₃S) C, H, N.

3.5.2. 2-(4-(2-Furoyl)piperazin-1-yl)-3-nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-carboxylic acid (8b**). Yield: 75%; mp: 207–210 °C; IR (KBr) cm⁻¹: 2890, 1710, 1724, 1624, 1460–1360; ¹H NMR (DMSO- d_6) δ ppm: 3.16–3.26 (m, 8H, CH₂ of piperazine), 5.9 (s, 1H, C₁-H), 6.32–7.68 (m, 7H, Ar-H), 7.26 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₅H₁₈N₄O₇S) C, H, N.



Figure 1. DNA gyrase supercoiling assay. The assays were carried out as described in Section 3. DNA gyrase was pre-incubated with the indicated concentrations of compounds in ice and then rest of the components of the reaction including relaxed DNA was added. (a) Lane 1: relaxed circular DNA, lane 2: supercoiling reaction in presence of 5% DMSO, lane 3: positive control for gyrase inhibition, moxifloxacin at 5 μ g/ml concentration. Lanes 4–8: supercoiling reaction in presence of 50 μ g/ml of compounds 7**j**, 7**i**, 7**i**, 7**e**, and 7**r**, respectively. (b) Lane 1: relaxed circular DNA, lane 2: supercoiling reaction in presence of 50 μ g/ml of compounds 7**j**, 7**i**, 7**i**, 7**e**, and 7**r**, respectively. (b) Lane 1: relaxed circular DNA, lane 2: supercoiling reaction in presence of 50 μ g/ml of compounds 7**j**, 7**i**, 7**i**, 7**e**, and 7**r**, respectively. (b) Lane 1: relaxed circular DNA, lane 2: supercoiling reaction in presence of 5% DMSO, lane 3: moxifloxacin at 5 μ g/ml concentration. Lanes 4 and 5: supercoiling reaction in presence of 40 μ g/ml of compounds 7**j** and 7**e**, respectively. R and S indicate relaxed and supercoiled DNA, respectively.

 Table 4. IC₅₀ values for DNA gyrase inhibition

Compounds	IC ₅₀ (µg/ml)
7e	>50
7i	>50
7j	40
71	40
7r	50

3.5.3. 2-(4-((Benzo[d][1,3]dioxol-6-yl)methyl)piperazin-1yl)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (7c). Yield: 75%; mp: 181–182 °C; IR (KBr) cm⁻¹: 3250, 2890, 1710, 1724, 1624, 1464– 1360; ¹H NMR (DMSO-*d*₆) δ ppm: 2.82–3.12 (m, 8H, CH₂ of piperazine), 3.6 (s, 2H, CH₂ of piperanoyl), 5.86 (s, 2H, -OCH₂O-), 5.9 (s, 1H, C₁-H), 6.40–7.1 (m, 7H, Ar-H), 7.28 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₈H₂₂FN₃O₅S) C, H, N.

3.5.4. 2-(4-Methyl-3-phenylpiperazin-1-yl)-3-nitro-5,12dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-carboxylic acid (8d). Yield: 73%; mp: 287–290 °C; IR (KBr) cm⁻¹: 2890, 1710, 1724, 1620, 1460–1368; ¹H NMR (DMSOd_6) \delta ppm: 2.2 (s, 3H, CH₃), 2.6 (t, 2H, 5-CH₂ of piperazine), 3.15 (t, 2H, 6-CH₂ of piperazine), 3.4 (d, 2H, 2-CH₂ of piperazine), 4.12 (t, 1H, 3-CH of piperazine), 5.9 (s, 1H, C₁-H), 6.3–7.2 (m, 9H, Ar-H), 7.3 (s, 1H, C₄-H), 14.4 (s, 1H, COOH); Anal. (C₂₇H₂₂N₄O₅S) C, H, N.**

3.5.5. 2-(4-(2,3-Dihydrobenzo[b][1,4]dioxin-2-oyl)(piperazin-1-yl)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2*a*]quinoline-6-carboxylic acid (7e). Yield: 77%; mp: 237–

Table 5. Phototoxic evaluation of newer compounds

Group	Ear thickness (mm) ^a							Erythema ^b					
	Time (approximately) after start of irradiaton (h) ^c												
	0	4	24	48	72	96	0	4	24	48	72	96	
Control ^d	0.29 ± 0.02	0.29 ± 0.01	0.32 ± 0.02	0.31 ± 0.01	0.34 ± 0.03	0.34 ± 0.03	0	0	0	0	0	0	
7j	0.29 ± 0.01	0.32 ± 0.03	0.31 ± 0.02	0.29 ± 0.02	0.32 ± 0.02	0.30 ± 0.02	0	0	0	0	0	0	
71	0.27 ± 0.01	0.29 ± 0.02	0.29 ± 0.02	0.29 ± 0.01	0.30 ± 0.01	0.28 ± 0.01	0	2	0	0	0	0	
7m	0.29 ± 0.01	0.30 ± 0.02	0.31 ± 0.01	0.30 ± 0.01	0.28 ± 0.02	0.30 ± 0.02	0	0	0	0	0	0	
Lomefloxacin	0.31 ± 0.01	0.40 ± 0.02	0.48 ± 0.02	0.53 ± 0.02	0.64 ± 0.04	0.60 ± 0.06	0	6	6	6	6	6	

^a Mean ear thickness ±SEM; left and right ears were averaged.

^b Number of mice with erythema.

^c Time zero, pre-dose (mice exposed to UV light immediately after dosing); 4 h, end of irradiation period.

^d Control, 0.5% aqueous solution of sodium carboxymethylcellulose (4 Ns/m²) dosed at 10 ml/kg.

238 °C; IR (KBr) cm⁻¹: 2890, 1710, 1724, 1624, 1460–1360; ¹H NMR (DMSO- d_6) δ ppm: 3.26–3.4 (m, 8H, CH₂ of piperazine), 4.6 (d, 2H, 3-CH₂ of dihydrobenzodioxinyl), 5.14 (t, 1H, 2-CH of dihydrobenzodioxinyl), 5.9 (s, 1H, C₁-H), 6.3–7.12 (m, 8H, Ar-H), 7.3 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₉H₂₂FN₃O₆S) C, H, N.

3.5.6. 2-(3-(2,6-Diffuorophenyl)-5-methlisoxazol-4-oyl)(piperazin-1-yl)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]-quinoline-6-carboxylic acid (7f). Yield: 75%; mp: >300 °C; IR (KBr) cm⁻¹: 2890, 1710, 1724, 1624, 1464–1360, 1208; ¹H NMR (DMSO-*d*₆) δ ppm: 2.3 (s, 3H, 5-CH₃ of isoxazolyl), 3.12–3.28 (m, 8H, CH₂ of piperazine), 5.9 (s, 1H, C₁-H), 6.32–7.08 (m, 7H, Ar-H), 7.3 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₃₁H₂₁F₃N₄O₅s) C, H, N.

3.5.7. 2-Thiomorpholino-3-nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-carboxylic acid (8g). Yield: 74%; mp: 268–271 °C; IR (KBr) cm⁻¹: 2890, 1710, 1724, 1620, 1460–1368; ¹H NMR (DMSO-***d***₆) \delta ppm: 2.64 (t, 4H, 3,5-CH₂ of thiomorpholine), 3.38 (t, 4H, 2,6-CH₂ of thiomorpholine), 5.9 (s, 1H, C₁-H), 6.32– 6.88 (m, 4H, Ar-H), 7.3 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₀H₁₅N₃O₅S₂) C, H, N.**

3.5.8. 2-(2,6-Dimethylmorpholino)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-carboxylic acid (7h). Yield: 70%; mp: 258–260 °C; IR (KBr) cm⁻¹: 2892, 1710, 1724, 1628, 1460–1360; ¹H NMR (DMSO-***d***₆) \delta ppm: 1.2 (d, 6H, 2,6 CH₃ of morpholino), 3.0 (d, 4H, 2,6-CH₂ of morpholine), 3.9 (m, 2H, 3,5-CH of morpholine), 5.9 (s, 1H, C₁-H), 6.4–6.9 (m, 4H, Ar-H), 7.3 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₂H₁₉FN₂O₄S) C, H, N.**

3.5.9. 2-(4-(Piperidin-1-yl)piperidin-1-yl)-3-nitro-5,12dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-carboxylic acid (8i). Yield: 75%; mp: 294–296 °C; IR (KBr) cm⁻¹: 2890, 1710, 1724, 1624, 1464–1360; ¹H NMR (DMSO***d***₆) \delta ppm: 1.5–1.6 (m, 10H, 5 CH₂), 2.2 (t, 4H, 2 CH₂), 2.7 (m, 1H, CH), 2.8 (t, 4H, 2 CH₂), 5.9 (s, 1H, C₁-H), 6.4–6.88 (m, 4H, Ar-H), 7.3 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₆H₂₆N₄O₅S) C, H, N.**

3.5.10. 2-(4-(4-Chlorophenyl)-4-hydroxypiperidin-1-yl)-3fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-**6-carboxylic acid (7j).** Yield: 79%; mp: 229–232 °C; IR (KBr) cm⁻¹: 2900, 1708, 1724, 1620, 1460–1370; ¹H NMR (DMSO-*d*₆) δ ppm: 2.0 (t, 4H, 3,5-CH₂ of piperidine), 2.7 (t, 4H, 2,6-CH₂ of piperidine), 5.9 (s, 1H, C₁-H), 6.28–7.18 (m, 8H, Ar-H), 7.3 (s, 1H, C₄-H), 10.0 (bs, 1H, OH), 14.62 (s, 1H, COOH); Anal. (C₂₇H₂₀-ClFN₂O₄S) C, H, N.

3.5.11. 2-(4-(6-Chloro-1,2-dihydro-2-oxobenzo]d]imidazol-3-yl)piperidin-1-yl)-3-nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-carboxylic acid (8k). Yield: 74%; mp: 224–225 °C; IR (KBr) cm⁻¹: 3110, 2896, 1712, 1728, 1710, 1624, 1460–1360; ¹H NMR (DMSOd_6) \delta ppm: 1.6–2.4 (m, 8H, 4 CH₂ of piperidine), 4.1 (bm, 1H, CH of piperidine), 5.9 (s, 1H, C₁-H), 6.2–6.9** (m, 7H, Ar-H), 7.3 (s, 1H, C₄-H), 10.8 (s, 1H, NH), 14.2 (s, 1H, COOH); Anal. $(C_{28}H_{20}ClN_5O_6S)$ C, H, N.

3.5.12. 2-(3-(Diethylcarbamoyl)piperidin-1-yl)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-car-boxylic acid (7l).** Yield: 70%; mp: 256–257 °C; IR (KBr) cm⁻¹: 2890, 1710, 1724, 1624, 1464–1360, 1208; ¹H NMR (DMSO- d_6) δ ppm: 1.2 (t, 6H, 2-CH₃ of ethyl), 1.78–2.7 (m, 9H, H of piperidine), 3.24 (q, 4H, 2-CH₂ of ethyl), 5.9 (s, 1H, C₁-H), 6.2–6.8 (m, 4H, Ar-H), 7.3 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₆H₂₆FN₃O₄S) C, H, N.

3.5.13. 2-(1,4-Dioxa-8-azaspiro[4.5]dec-8-yl)-3-nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a***]quinoline-6-carboxylic acid (8m). Yield: 74%; mp: 198–199 °C; IR (KBr) cm⁻¹: 2890, 1712, 1724, 1618, 1466–1368; ¹H NMR (DMSO-d_6) \delta ppm: 1.78-2.4 (m, 8H, 4-CH₂ of azaspirodecane), 3.96 (m, 4H, 2-CH₂ of azaspirodecane), 5.9 (s, 1H, C₁-H), 6.2–6.82 (m, 4H, Ar-H), 7.32 (s, 1H, C₄-H), 14.4 (s, 1H, COOH); Anal. (C₂₃H₁₉N₃O₇S) C, H, N.**

3.5.14. 2-(1-(*tert***-Butylcarbamoyl)-3,4-dihydroisoquino-lin-2(1H)-yl)-3-fluoro-5,12-dihydro-5-oxobenzothiazol-o[3,2-a]quinoline-6-carboxylic acid (7n).** Yield: 71%; mp: 271–272 °C; IR (KBr) cm⁻¹: 2890, 1712, 1724, 1626, 1468–1360; ¹H NMR (DMSO-*d*₆) δ ppm: 1.3 (s, 9H, 3 CH₃), 2.66–2.9 (m, 4H, 2 CH₂ of isoquinoline), 4.85 (s, 1H, CH of isoquinoline), 5.88 (s, 1H, C₁-H), 6.28–7.1 (m, 8H, Ar-H), 7.32 (s, 1H, C₄-H), 10.2 (s, 1H, NH), 14.2 (s, 1H, COOH); Anal. (C₃₀H₂₆FN₃O₄S) C, H, N.

3.5.15. 2-(2-Carboxy-5,6-dihydroimidazo[1,2-*a***]pyrazin-7(8H)-yl)-3-nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-***a***]quinoline-6-carboxylic acid (80). Yield: 79%; mp: 255– 258 °C; IR (KBr) cm⁻¹: 3200, 2890, 1710, 1724, 1624, 1464–1360, 1208; ¹H NMR (DMSO-***d***₆) \delta ppm: 3.1–3.8 (m, 6H, 3-CH₂), 5.88 (s, 1H, C₁-H), 6.28–7.0 (m, 4H, Ar-H), 7.6 (s, 1H, CH), 7.32 (s, 1H, C₄-H), 12.12 (s, 1H, 2-COOH), 14.6 (s, 1H, 3-COOH); Anal. (C₂₃H₁₅N₅O₇S) C, H, N.**

3.5.16. 2-(8-(4-Methoxybenzyl)-3,4,5,6,7,8-hexahydroisoquinolin-2(1H)-yl)-3-fluoro-5,12-dihydro-5-oxobenzothiazolo[3,2-a]quinoline-6-carboxylic acid (7p). Yield: 79%; mp: 219–220 °C; IR (KBr) cm⁻¹: 3210, 2890, 1710, 1724, 1620, 1460–1368; ¹H NMR (DMSO-d_6) \delta ppm: 1.6–1.95 (m, 8H, 4-CH₂ of isoquinolinyl), 2.2-3.4 (m, 7H, 2-CH₂ and 1-CH of isoquinolinyl, and CH₂), 3.73 (s, 3H, –OCH₃), 5.88 (s, 1H, C₁-H), 6.28–7.1 (m, 8H, Ar-H), 7.32 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₃₃H₂₉FN₂O₄S) C, H, N.

3.5.17. 2-(3,4-Dihydro-6,7-dimethoxyisoquinolin-2(1H)-yl)-3-nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a***]quino-line-6-carboxylic acid (8q). Yield: 76%; mp: 237–240 °C; IR (KBr) cm⁻¹: 3210, 2890, 1710, 1724, 1620, 1460–1368; ¹H NMR (DMSO-***d***₆) \delta ppm: 2.72–3.6 (m, 6H, CH₂ of isoquinolinyl), 3.7 (s, 6H, –OCH₃), 5.88 (s, 1H, C₁–H), 6.3–6.93 (m, 6H, Ar-H), 7.32 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₇H₂₁N₃O₇S) C, H, N.**

3.5.18. 2-(6-tert-Butoxycarbonylamino-3-azabicyclo]3.1.0]hex-3-yl)-3-nitro-5,12-dihydro-5-oxobenzothiazolo[3,2-*a*]quinoline-6-carboxylic acid (8r). Yield: 76%; mp: 239– 240 °C; IR (KBr) cm⁻¹: 3210, 2890, 1710, 1724, 1620, 1460–1368; ¹H NMR (DMSO-*d*₆) δ ppm: 1.43 (s, 9H, *tert*-butyl), 1.78–3.8 (m, 7H, azabicyclohexane ring H), 4.76 (bs, 1H, NH-*t*-Boc), 5.88 (s, 1H, C₁-H), 6.3–6.8 (m, 4H, Ar-H), 7.32 (s, 1H, C₄-H), 14.2 (s, 1H, COOH); Anal. (C₂₆H₂₄N₄O₇S) C, H, N.

3.6. In-vitro antimycobacterial activity

All compounds were screened for their in-vitro antimycobacterial activity against MTB, MDR-TB, and MC² in Middlebrook 7H11agar medium supplemented with OADC by agar dilution method similar to that recommended by the National Committee for Clinical Laboratory Standards for the determination of MIC in duplicate. The MDR-TB clinical isolate was obtained from Tuberculosis Research Center, Chennai, India, and was resistant to isoniazid, rifampicin, ethambutol, and ofloxacin. The minimum inhibitory concentration (MIC) is defined as the minimum concentration of compound required to give complete inhibition of bacterial growth.

3.7. Cytotoxicity

Some compounds were further examined for toxicity (IC_{50}) in a mammalian Vero cell line at concentrations of 62.5 µg/ml. After 72 h of exposure, viability was assessed on the basis of cellular conversion of MTT into a formazan product using the Promega Cell Titer 96 non-radioactive cell proliferation assay.

3.8. In-vivo antimycobacterial activity

One compound was tested for efficacy against MTB at a dose of 25 mg/kg in six-week-old female CD-1 mice six per group. In this model, the mice were infected intravenously through caudal vein with approximately 10^7 viable M. tuberculosis ATCC 35801. Drug treatment by intraperitoneal route began after 10 days of inoculation of the animal with microorganism and was continued for 10 days. After 35 days post infection the spleens and right lungs were aseptically removed and ground in a tissue homogenizer, the number of viable organisms was determined by serial 10-fold dilutions and subsequent inoculation onto 7H10 agar plates. Cultures were incubated at 37 °C in ambient air for 4 weeks prior to counting. Bacterial counts were measured and compared with the counts from negative controls (vehicle treated) in lung and in spleen.

3.9. DNA gyrase supercoiling assay

DNA gyrase was purified from *M. smegmatis* cells using the method described by Manjunatha et al.¹⁸ The supercoiling assays were carried out by the procedure already described.¹¹ Briefly, supercoiling assays were carried out by incubating 400 ng of relaxed circular pUC18 in supercoiling buffer (35 mM Tris–HCl, pH 7.5, 5 mM MgCl₂, 25 mM potassium glutamate, 2 mM spermidine, 2 mM ATP, 50 µg/ml bovine serum albumin, and 90 µg/ml yeast *t*-RNA in 5% (v/v) glycerol) for 30 min. The compounds tested were dissolved in DMSO and preincubated with gyrase in reaction buffer prior to the addition of DNA. Moxifloxacin at final concentration of 5 µg/ml was used as a positive control and another control reaction having 5% DMSO in absence of compounds was also performed. The reaction samples were heat-inactivated at 65 °C for 15 min and applied onto 1% agarose gel for electrophoresis in Tris–acetate–EDTA buffer for 12 h. The gels were stained with ethi-dium bromide.

3.10. Phototoxicity evaluation

Female swiss albino mice, approximately 2 months old and weighing 20–25 g, were used in this study. Before oral dosing, they were fasted overnight for at least 18 h. Food was returned at the end of the 4 h photo-irradiation period. Eighteen mice were randomly distributed into three dosing groups. First group received a single dose of screened compound at 140 mg/kg by oral gavage. A second group received a single dose of 140 mg of lomefloxacin HCl/kg. This lomefloxacin dose is one that, in preliminary experiments in this test system, produced a consistent erythema and ear thickening response. The final group served as a vehicle control and received 10 ml/ kg of the methylcellulose vehicle only. Test animals were exposed to UV light in a manner adapted from that described previously. Animals were irradiated for 4 h, equal to a total UV light irradiation of approximately 18 J/cm². Before dosing, at the end of the irradiation period, and at approximately 24, 48, 72, and 96 h after dosing, both ears of each mouse were evaluated for changes indicative of a positive response: erythema, edema or a measurable increase in ear thickness.

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References and notes

- 1. World Health Organization, Tuberculosis fact sheet 2007.
- Zignol, M.; Hosseini, M. S.; Wright, A.; Weezenbeek, C. L.; Nunn, P.; Watt, C. J.; Williams, C. G.; Dye, C. J. Infect. Dis. 2006, 194, 479.
- Sriram, D.; Yogeeswari, P.; Thirumurugan, R.; Pavana, R. K. J. Med. Chem. 2006, 49, 3448.
- Sriram, D.; Yogeeswari, P.; Dinakaran, M.; Thirumurugan, R. J. Antimicrob. Chemother. 2007, 59, 1194.
- 5. Kubendiran, G.; Paramasivan, C. N.; Sulochana, S.; Mitchison, D. A. J. Chemother. 2006, 18, 617.
- Renau, T. E.; Sanchez, J. P.; Shapiro, M. A.; Dever, J. A.; Gracheck, S. J.; Domagala, J. M. J. Med. Chem. 1995, 38, 2974.
- Kondo, H.; Taguchi, H.; Inoue, Y.; Sakamoto, F.; Tsukamoto, G. J. Med. Chem. 1990, 33, 2012.

- 8. Tsuzuki, Y.; Tomita, K.; Shibamori, K.; Sato, Y.; Kashimoto, S.; Chiba, K. J. Med. Chem. 2004, 47, 2097.
- 9. National Committee for Clinical Laboratory Standards. Antimycobacterial susceptibility testing for Mycobacterium tuberculosis. Proposed standard M24-T. National Committee for Clinical Laboratory Standards, Villanova, Pa., 1995.
- 10. Gundersen, L. L.; Nissen-Meyer, J.; Spilsberg, B. J. Med. Chem. 2002, 45, 1383.
- 11. Sriram, D.; Yogeeswari, P.; Basha, J. S.; Radha, D. R.; Nagaraja, V. *Bioorg. Med. Chem.* **2005**, *13*, 5774. 12. Levine, C. H.; Hiasa, A.; Marians, K. J. *Biochim. Biophys.*
- Acta 1998, 1400, 29.

- 13. Manjunatha, U. H.; Madhusudan, K.; Sandhya, S. Visweswariah.; Nagaraja, V. Curr. Sci. 2000, 79, 968. 14. Ferguson, J.; Walker, E. M.; Johnson, B. E. British
- J. Dermat. 1989, 120, 291.
- 15. Domagala, J. M. J. Antimicrob. Chemother. 1994, 33, 685-706.
- 16. Takayama, S.; Hirohashi, M.; Kato, M.; Shimada, H. J. Toxicol. Environ. Health 1995, 45, 1.
- 17. Mayne, T. N.; Johnson, N. J.; Kluwe, W. M.; Lencoski, D. L.; Polzer, R. J. J. Antimicrob. Chemother. 1997, 39, 67.
- Manjunatha, U. H.; Dalal, M.; Chatterji, M.; Radha, D. 18. R.; Visweswariah, S. S.; Nagaraja, V. Nucleic Acids Res. 2002, 30, 2144.