

Conversion of Spirostane and Solanidane into Pregnane (1) Introduction of Oxygen Function into C-23

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A spirostane derivative possessing a hydroxyl group at C-23, esculeogenin A, a saponin of tomato saponin, was found to be easily converted into the corresponding pregnane derivative by refluxing with aqueous pyridine. Therefore, introduction of a hydroxyl group into the C-23 of diosgenin (as representative of spirostane derivatives) and solasodine (as representative of spirostane derivatives) was attempted by the reaction of $\text{NaNO}_2\text{-BF}_3\cdot\text{Et}_2\text{O}$. In diosgenin, the objective compound was obtained by the reaction in AcOH. However, in solasodine, we obtained a 23-nitroso derivative by the reaction in AcOH and 23,24-bisnorcholanolic acid 22-16 lactone, or vespertilin, in AcOH and CHCl_3 .

Key words spirost-23-hydroxylation; diosgenin; solasodine; nitroso derivative; bisnorcholanolic acid

Nohara *et al.* found that the spirostane saponin of tomato saponin, (*5 α ,22*S*,23*S*,25*S**)-*3 β ,23,27*-trihydroxyspirostane, or esculeogenin A (**1**),¹⁾ was converted to *3 β ,16 β* -dihydroxy-*5 α* -pregna-20-one (**2**) by refluxing with diluted aqueous pyridine.^{2,3)} Its reaction was considered to be triggered by the existence of the 23-hydroxyl group on the spirostane skeleton. Preparation of steroidal hormones depends mainly upon the Marker degradation.⁴⁾ When the 23-hydroxyspirostane, spirostane, and tomatidane compounds are prepared, the pregnane compound might be derived from them by a similar simple reaction. Therefore, we first tried to introduce an oxygen group at C-23 in them by the reported method.^{5,6)} In this report, diosgenin (a representative spirostane) and solanidine (a representative spirostane derivative) were used as crude original substances.

First, to a suspension of diosgenin (**3**, 1 g) in acetic acid

(40 mL), NaNO_2 (3.0 g) and $\text{BF}_3\cdot\text{Et}_2\text{O}$ in ether (4.5 mL) were added gradually with stirring for 1 h at rt. After 1.5 h, the reaction mixture was treated in the usual manner to give a product that was chromatographed on silica gel with *n*-hexane:acetone = 4:1 \rightarrow 3:1 to give compounds **4** (349 mg, 33.8%) and **5** (137 mg, 11.9%).

Compound **4** consisted of colorless needles (mp 199–202°C; $[\alpha]_D^{20}$ -84.4° [CHCl_3]). High-resolution fast atom bombardment mass spectrometry (HR-FAB-MS) revealed a molecular ion at m/z 451.2826, indicating a molecular formula of $\text{C}_{27}\text{H}_{40}\text{O}_4 + \text{Na}$ (Calcd: 451.2824), and a carbonyl group appeared at 1732 cm^{-1} in the IR spectrum.

The $^1\text{H-NMR}$ spectrum exhibited two tertiary methyl groups at δ 0.79 (3H, s, H_3 -18) and 1.02 (3H, s, H_3 -19); two secondary methyl groups at δ 0.93 (6H, d, $J=6.7\text{ Hz}$, H_3 -21, H_3 -27); two methylene protons at δ 3.59 and 3.78 (1H, m and

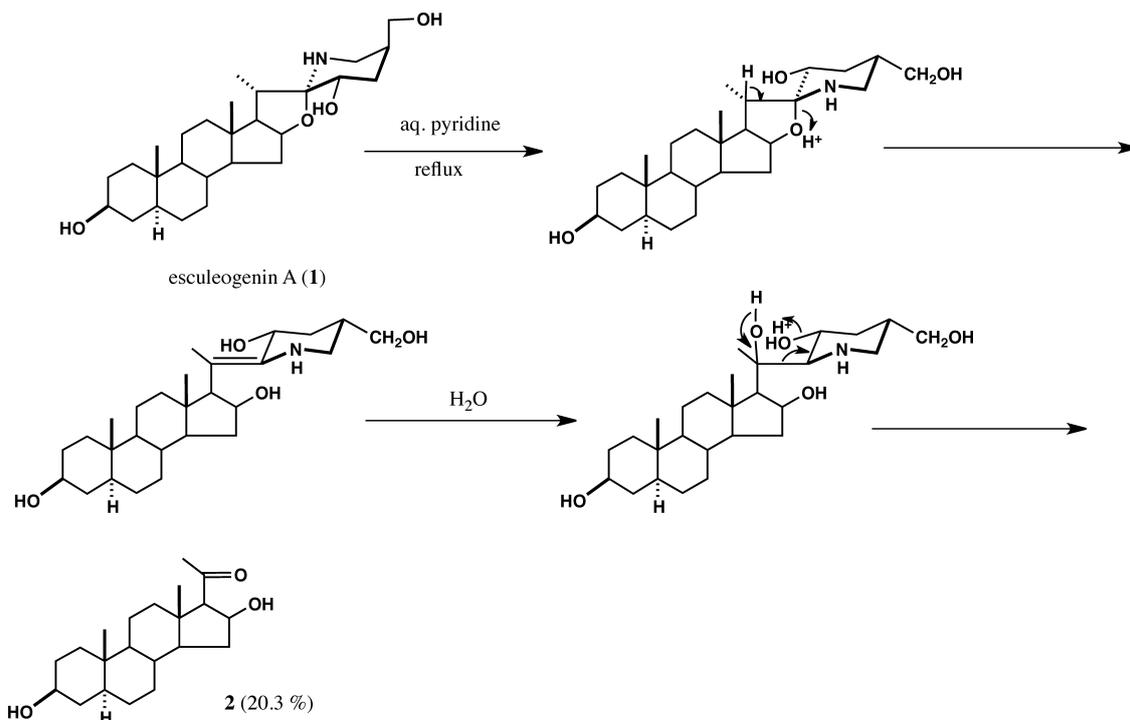


Chart 1. Mechanism for Conversion of Esculeogenin A (**1**) into Pregnane Derivative (**2**)

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1H, t-like, $J=11.3$ Hz, Heq-26, Hax-26, respectively); two oxygen-bearing methine protons at δ 3.51 (1H, m, H-3) and 4.61 (1H, dd, $J=6.8, 15.7$ Hz, H-16); and one olefinic proton at δ 5.34 (1H, m, H-6). In the ^{13}C -NMR spectrum (in CDCl_3), one carbonyl carbon newly occurred at δ 201.8, which correlated with H-20 at δ 2.89 (1H, m). Therefore, the structure of compound **4** was characterized as 23-oxodiosgenin.^{5,6)}

For compound **5**, an amorphous powder having $[\alpha]_{\text{D}} -99.6^\circ$ (CHCl_3), HR-FAB-MS revealed a molecular ion at m/z 540.2935, indicating a molecular formula of $\text{C}_{29}\text{H}_{43}\text{NO}_7+\text{Na}$ (Calcd: 540.2937). A carbonyl group appeared at 1730cm^{-1} in the IR spectrum. The ^1H -NMR spectrum showed signals due to H₃-18 (3H, s) at δ 0.76, H₃-19 (3H, s) at δ 1.02, H₃-27 (3H, d, $J=4.3$ Hz) at δ 0.93, and H₃-21 (3H, d, $J=4.3$ Hz) at δ 0.94; one acetyl signal (3H, s) at δ 2.17; H-3 (1H, m) at δ 3.77; Heq-26 (1H, t-like, $J=11.2$ Hz) at δ 3.77; and H-16 (1H, m) at δ 4.51. The ^{13}C -NMR spectrum revealed the presence of one hydroxyimino group at δ 178.0, together with one carbonyl carbon at δ 201.6 and one acetyl carbonyl carbon at δ 169.5. Furthermore, since heteronuclear multiple bond coherence (HMBC) analysis indicated the existence of a hydroxyl group at C-5, the presence of an imino group at C-6 is suggested. The carbonyl group is located at C-23, as in compound **4**. The acetyl group was estimated to be bound to the hydroxyimino group because the H-3 was not down shifted. Therefore, the structure of compound **5** was characterized as 3 β ,5-dihydroxy-6-acetoxyimino-23-oxo-(25*R*)-spirostane.

Next, compound **4** (250 mg) was reduced with NaBH_4 (200 mg) in MeOH (6 mL) for 30 min at rt. After a small amount of AcOH was added, its reaction mixture was evaporated to give a residue, which was then chromatographed on silica gel with *n*-hexane–acetone = 4:1 \rightarrow 3:1 to provide compound **6** (160 mg, 64.0%). Compound **6** appeared as colorless needles (mp 212–214°C; $[\alpha]_{\text{D}} -112.4^\circ$ [CHCl_3]). HR-FAB-MS revealed a molecular ion at m/z 494.2877, indicating a molecular formula of $\text{C}_{27}\text{H}_{42}\text{O}_4+\text{Na}$ (Calcd: 494.2880), and a hydroxyl group appeared at 3550cm^{-1} in the IR spectrum. The ^1H -NMR spectrum (in pyridine- d_5) displayed signals due to H₃-21 (d, $J=6.7$ Hz) at δ 0.73, H₃-18 (s) at δ 0.99, H₃-19 (s) at δ 1.02, H₃-27 (d, $J=6.7$ Hz) at δ 1.18, Hax-26 (t-like, $J=11.0$ Hz) at δ 3.46, Heq-26 (m) at δ 3.54, H-3 (m) at δ 3.82, H-16 (dd, $J=7.6, 15.6$ Hz) at δ 4.64, H-6 (m) at δ 5.36, and H-23 (dd, $J=4.5, 11.3$ Hz) at δ 3.82. Their coupling constants indicated that the hydroxyl group was oriented equatorially (S). In the ^{13}C -NMR spectrum (in pyridine- d_5), one hydroxyl

carbon newly occurred at δ 67.4. The structure of compound **6** was characterized as (22*S*,23*S*,25*R*)-3 β ,23-dihydroxy-spirost-5-ene,23*S*-hydroxydiosgenin.

Next, to a suspension of solasodine (**7**, 307 mg) in acetic acid (18 mL), NaNO_2 (924 mg) and $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (4.5 mL) were added gradually with stirring for 1 h at rt. After 1.5 h, the reaction mixture was treated in the usual manner to give a product that was chromatographed on silica gel with *n*-hexane–acetone = 4:1 \rightarrow 3:1 to give compound **8** (184 mg, 56.2%) as a major product.

Compound **8** exhibited $[\alpha]_{\text{D}} -98.2^\circ$ (MeOH) and a molecular formula of $\text{C}_{27}\text{H}_{42}\text{N}_2\text{O}_3+\text{Na}$ at m/z 465.3090 (Calcd: 465.3093) in positive HR-FAB-MS. The ^1H -NMR spectrum showed two tertiary groups at δ 0.82 (3H, s, H₃-18) and 1.00 (3H, s, H₃-19); two secondary methyl groups at δ 0.90 (3H, d, $J=6.8$ Hz, H₃-27) and 1.17 (3H, d, $J=7.5$ Hz, H₃-21); two oxygen-bearing methine protons at δ 3.50 (1H, m, Hax-26) and 3.97 (1H, m, Heq-26); one olefinic proton at δ 5.31 (1H, m, H-6), along with a down-shifted proton at δ 4.66 (1H, dd, $J=3.6, 13.8$ Hz), which was assigned to H-23 because HMBC was observed between H-23 and C-22 at δ 102.0; and H-25 at δ 44.1. The ^{13}C -NMR spectrum of **7** displayed a total of 27 carbon signals, including four methyl carbons at δ 15.0, 16.4, 19.1, and 19.5; two oxygen-bearing carbons at δ 74.8 and 83.1; one hemiacetal carbon at δ 102.0; one double bond at δ 121.4 and 140.9; and one nitrogen-bearing methine carbon at δ 44.1. The chemical shifts of H-23 in the ^1H - and ^{13}C -NMR spectra indicated that an imino group was attached to C-23; its configuration was determined to be equatorial by its coupling constants.

Next, to increase the solubility of the reaction mixture, a small amount of CHCl_3 was added to the reaction. To a solution of solasodine (**7**, 304 mg) in AcOH (14 mL) and CHCl_3 (5 mL), NaNO_2 (890 mg) and BF_3 ethelate (4.5 mL) were added gradually and left standing at rt for 2 h. The usual treatment yielded compound **9** (120.5 mg, 48.2%) having $[\alpha]_{\text{D}} -82.6^\circ$ (MeOH) and a molecular formula of $\text{C}_{22}\text{H}_{32}\text{O}_3+\text{Na}$ at m/z 367.2252 (Calcd: 367.2249), as revealed by positive HR-FAB-MS. The ^1H -NMR spectrum of **9** showed two tertiary methyl groups at δ 0.90 (3H, s, H₃-18) and 0.95 (3H, s, H₃-19); one secondary methyl group at δ 1.23 (3H, d, $J=7.5$ Hz, H₃-21); two oxygen-bearing methine protons at δ 3.46 (1H, m, H-3) and 4.89 (1H, m, H-16); and one olefinic proton at δ 5.31 (1H, m, H-6). The ^{13}C -NMR spectrum displayed a total of 22 carbon signals including two oxygen-bearing carbons at δ 71.7

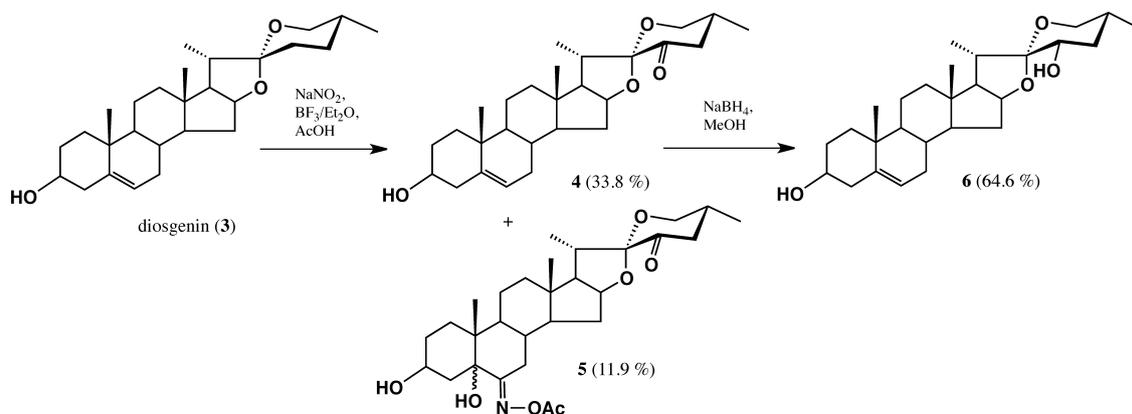


Chart 2. Introduction of Oxygen Function into C-23

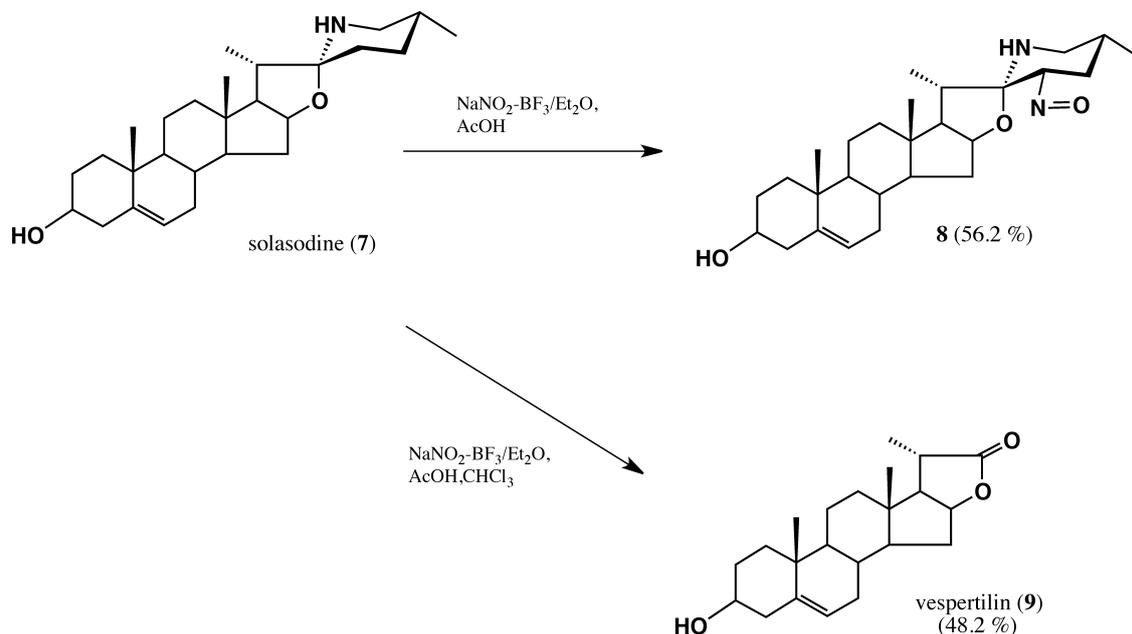


Chart 3. Reaction of Solasodine

and 82.9; one trisubstituted double bond at δ 121.3 and 141.0; and one lactone carbonyl carbon at δ 181.5. This compound was coincident with vespertilin.⁷⁾

Consequently, in the reaction of solasodine with NaNO_2 and BF_3 , the results when only AcOH was used in suspension state as a solvent differed from those obtained when AcOH and CHCl_3 were used in solution state, thus they would depend on the solubility.

We are currently preparing tomatidine produced from the aerial part of *Solanum lycopersicum*.

Experimental

General Procedure Optical rotations were measured with a JASCO P-1020 ($l=0.5$) automatic digital polarimeter. FAB-MS were obtained with a glycerol matrix in the positive ion mode using a JEOL JMS-DX300 and a JMS-DX 303 HF spectrometer. The ^1H - and ^{13}C -NMR spectra were measured in pyridine- d_5 with JEOL α -500 spectrometer, and chemical shifts are given on a δ (ppm) scale with tetramethylsilane (TMS) as the internal standard. Column chromatographies were carried out on a Diaion HP-20 (Mitsubishi Chemical Ind.), and silica gel 60 (230–400 mesh, Merck). TLC was performed on silica gel plates (Kieselgel 60 F₂₅₄, Merck) and RP C₁₈ silica gel plates (Merck). The spots on TLC were visualized by UV light (254/366nm) and sprayed with 10% H_2SO_4 , followed by heating.

Reaction of Diosgenin with NaNO_2 and BF_3 -Ethelate in AcOH Firstly, to a solution of diosgenin (**3**, 1g) in acetic acid (40mL), NaNO_2 (3.0g) and $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (4.5 mL) was gradually added with stirring for 1h at rt. After 1.5h, the reaction mixture was treated in an usual manner to give the product which was chromatographed on silica gel with *n*-hexane–acetone = 4:1 \rightarrow 3:1 to give compounds **4** (349mg, 33.8%) and **5** (137mg, 11.9%).

Compound **4**, colorless needles, mp 199–202°C, $[\alpha]_D$ -84.4° ($c=0.5$, CHCl_3), IR ν_{max} 1732 cm^{-1} (carbonyl carbon). HR-FAB-MS (m/z): 451.2826 [$\text{C}_{27}\text{H}_{40}\text{O}_4+\text{Na}$, Calcd: 451.2824].

^1H -NMR spectrum (in CDCl_3) δ : 0.79 (3H, s, H₃-18), 1.02 (3H, s, H₃-19), 0.93 (6H, d, $J=6.7\text{Hz}$, H₃-21, H₃-27), 3.59, 3.78 (1H, m, and 1H, t-like, $J=11.3\text{Hz}$, Heq-26 and Hax-26, respectively), 3.51 (1H, m, H-3), 4.61 (1H, dd, $J=6.8, 15.7\text{Hz}$, H-16), 5.34 (1H, m, H-6). ^{13}C -NMR spectrum (in CDCl_3): C-1—C-27, δ 37.3, 29.7, 71.8, 43.5, 141.0, 121.2, 31.7, 31.5, 50.1, 36.7, 20.8, 39.6, 40.8, 56.7, 31.7, 83.4, 61.8, 16.0, 19.4, 34.8, 14.4, 109.9, 201.8, 43.9, 35.8, 65.7, 17.1.

Compound **5**, an amorphous powder, $[\alpha]_D$ -99.6° ($c=0.5$, CHCl_3), IR ν_{max} 1730 cm^{-1} (carbonyl carbon). HR-FAB-MS (m/z): 540.2935 ($\text{C}_{29}\text{H}_{43}\text{NO}_7+\text{Na}$, Calcd: 540.2937). ^1H -NMR spectrum (CDCl_3) δ : 0.76 (3H, s, H₃-18), 1.02 (3H, s, H₃-19), 0.93 (3H, d, $J=4.3\text{Hz}$, H₃-27), 0.94 (3H, d, $J=4.3\text{Hz}$, H₃-21), 2.17 (3H, s, acetyl signal), 3.77 (1H, m, H-3), 3.77 (1H, t-like, $J=11.2\text{Hz}$, Heq-26), 4.51 (1H, m, H-16). ^{13}C -NMR spectrum (CDCl_3): C-1—C-27, δ 38.9, 29.8, 68.0, 45.6, 87.1, 178.0, 33.7, 34.5, 55.1, 36.9, 21.2, 39.7, 42.2, 56.7, 32.7, 83.5, 61.9, 15.8, 17.3, 34.6, 14.6, 109.9, 201.6, 45.2, 35.9, 65.6, 17.1, acetyl group, 169.5, 18.2.

Reaction of Compound **4 with NaBH_4** Compound **4** (250mg) was reduced with NaBH_4 (200mg) in MeOH (6mL) for 30min at rt. After addition of a small amount of AcOH , its reaction mixture was evaporated to give a residue, which was then chromatographed on silica gel with *n*-hexane–acetone = 4:1–3:1 to provide compound **6** (160mg, 64.0%).

Compound **6** exhibited colorless needles, mp 212–214°C, $[\alpha]_D$ -112.4° ($c=0.5$, CHCl_3), IR ν_{max} 3550 cm^{-1} (hydroxyl). HR-FAB-MS (m/z): 467.3017 ($\text{C}_{27}\text{H}_{42}\text{NO}_4+\text{Na}$, Calcd: 467.3012). ^1H -NMR spectrum (in pyridine- d_5) δ : 0.73 (3H, d, $J=6.7\text{Hz}$, H₃-21), 0.99 (3H, s, H₃-18), 1.02 (3H, s, H₃-19), 1.18 (3H, d, $J=6.7\text{Hz}$, H₃-27), 3.46 (1H, t-like, $J=11.0\text{Hz}$, Hax-26), 3.54 (1H, m, Heq-26), 3.82 (1H, m, H-3), 3.82 (1H, dd, $J=4.5, 11.3\text{Hz}$, H-23), 4.64 (1H, dd, $J=7.6, 15.6\text{Hz}$, H-16), 5.36 (1H, m, H-6). ^{13}C -NMR spectrum (in pyridine- d_5): C-1—C-27, δ 37.8, 29.8, 71.2, 43.4, 142.0, 120.9, 33.7, 31.7, 50.5, 37.0, 21.2, 40.3, 41.0, 56.8, 32.7, 81.6, 62.5, 35.8, 14.7, 111.7, 67.4, 38.8, 31.7, 65.2, 16.9.

Reaction of Solasodine (7) with NaNO₂ and BF₃·Et₂O in AcOH To a solution of solasodine (7, 307 mg) in acetic acid (18 mL), NaNO₂ (924 mg) and BF₃·Et₂O (4.5 mL) was gradually added with stirring for 1 h at rt. After 1.5 h, the reaction mixture was treated in the usual manner to give the product which was chromatographed on silica gel with *n*-hexane–acetone = 4:1 → 3:1 to give compound **8** (184 mg, 56.2%) as a major product.

Compound **8**, an amorphous powder, $[\alpha]_D -98.2^\circ$ ($c=1.0$, MeOH). HR-positive FAB-MS (m/z): 465.3090 (Calcd for C₂₇H₄₂N₂O₃+Na: 465.3093). ¹H-NMR spectrum (CDCl₃) δ : 0.82 (3H, s, H₃-18), 0.90 (3H, d, $J=6.8$ Hz, H₃-27), 1.00 (3H, s, H₃-19), 1.17 (3H, d, $J=7.5$ Hz, H₃-21), 3.50 (1H, m, Hax-26), 3.50 (1H, m, H-3), 3.97 (1H, m, Heq-26), 4.66 (1H, dd, $J=3.6, 13.8$ Hz, H-23), 5.31 (1H, m, H-6). ¹³C-NMR spectrum (CDCl₃) C-1—C-27, δ 37.2, 29.7, 71.8, 42.3, 140.9, 121.4, 31.7, 31.5, 50.1, 36.7, 21.0, 39.8, 40.4, 56.5, 32.1, 83.1, 61.6, 16.4, 19.3, 44.1, 15.0, 102.1, 44.1, 42.3, 44.1, 42.3, 19.0.

Reaction of Solasodine (7) with NaNO₂ and BF₃·Et₂O in AcOH and CHCl₃ To a solution of solasodine (7, 304 mg) in AcOH (14 mL) and CHCl₃ (5 mL), NaNO₂ (890 mg) and BF₃·Et₂O (4.5 mL) was added gradually and left stand at rt for 2 h. Usual work up gave compound **9** (120.5 mg, 48.2%), $[\alpha]_D$

-82.6° ($c=1.0$, MeOH). HR-positive FAB-MS (m/z): 361.2252 (Calcd for, C₂₂H₃₂O₃+Na: 361.2249). ¹H-NMR spectrum (CDCl₃) δ : 0.70 (3H, s, H₃-18), 0.95 (3H, s, H₃-19), 1.23 (3H, d, $J=7.5$ Hz, H₃-21), 3.46 (1H, m, H-3), 4.89 (1H, m, H-16), 5.27 (1H, m, H-6). ¹³C-NMR spectrum (CDCl₃): C-1—C-22, δ 37.3, 32.0, 71.7, 42.3, 141.0, 121.0, 31.6, 31.3, 50.2, 36.7, 20.4, 38.3, 41.5, 59.0, 36.5, 82.9, 61.6, 18.1, 19.5, 54.9, 13.8, 181.5.

References and Notes

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