

# A Novel Strategy for the Enantioselective Synthesis of the Steroidal Framework Using Cascade Ring Expansion Reactions of Small Ring Systems –Asymmetric Total Synthesis of (+)-Equilenin

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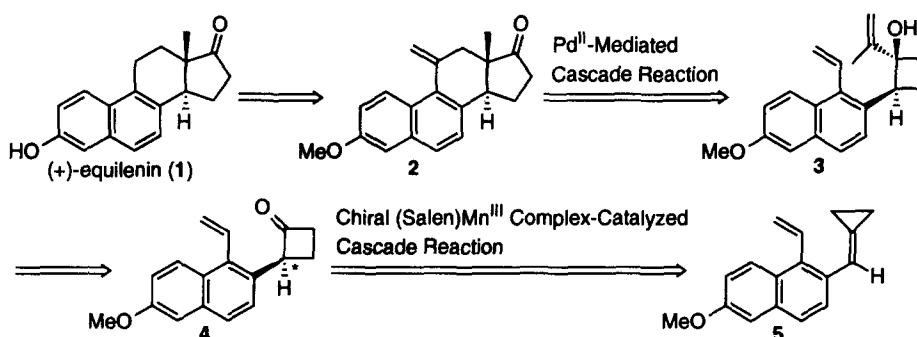
## Abstract

Enantioselective synthesis of (+)-equilenin (**1**) utilizing a novel strategy is described. The key steps are two cascade ring expansion reactions of small ring systems; 1) chiral (salen)Mn<sup>III</sup> complex-catalyzed cascade reaction of cyclopropylidene; 2) Pd<sup>II</sup>-mediated cascade reaction of the cyclobutanone. © 1999 Elsevier Science Ltd. All rights reserved.

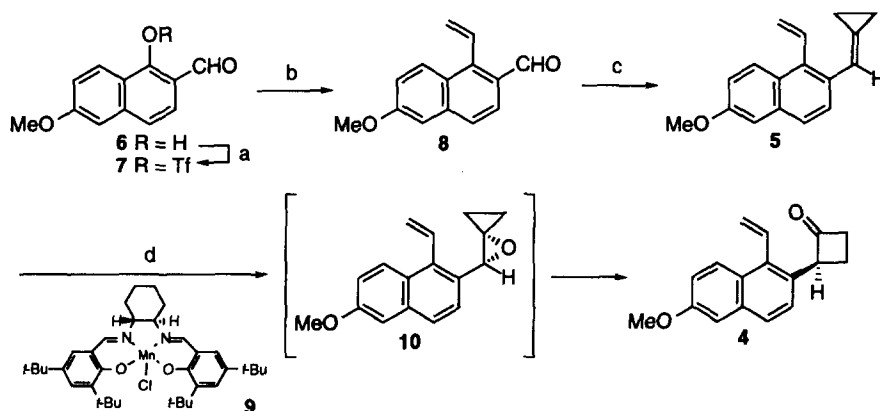
**Keywords:** Asymmetric synthesis; Rearrangement; Steroids and sterols; Stereocontrol

Steroids are one of the most widely distributed groups of natural products displaying a variety of physiologically important features, so that numerous synthetic approaches have been developed.<sup>1</sup> Here, we disclose a novel strategy for the enantioselective synthesis of (+)-equilenin (**1**) based on two cascade ring expansion reactions of small ring systems as outlined in Scheme 1. Our plan for constructing steroidal C,D rings exploits a Pd<sup>II</sup>-mediated cascade ring expansion and insertion process involving a cyclobutanone derivative, methodology of which has previously been reported by us (**3**→**2**, Scheme 1).<sup>2</sup> The chiral cyclobutanone **4**, a precursor of **3**, could be prepared *via* chiral (salen)Mn<sup>III</sup> complex-catalyzed asymmetric epoxidation<sup>3</sup> of the cyclopropylidene **5**, followed by its enantiospecific rearrangement.<sup>4</sup>

First of all, the triflate **7**, prepared from the hydroxynaphthaldehyde **6**<sup>5</sup> (61%), was subjected to Stille reaction<sup>6</sup> with *tri-n*-butylvinylstannane to give the vinyl naphthaldehyde **8** (98%), which upon Wittig reaction with cyclopropylidene-triphenylphosphorane under modified McMurry conditions<sup>7</sup> afforded the cyclopropylidene derivative **5** (70%) (Scheme 2). With the cyclopropylidene **5** in hand, the critical cascade asymmetric epoxidation–ring expansion reaction was examined. When a mixture of **5** and a 5 mol% of (*R,R*)-(salen)Mn<sup>III</sup> complex **9** was treated with sodium hypochlorite as an oxidant<sup>3</sup>, the reaction successfully proceeded to provide the desired chiral cyclobutanone **4** (78% e.e., 55% yield) in one step *via* oxaspiropentane intermediate **10**<sup>8</sup>.



**Scheme 1.** Retrosynthesis of (+)-equilenin

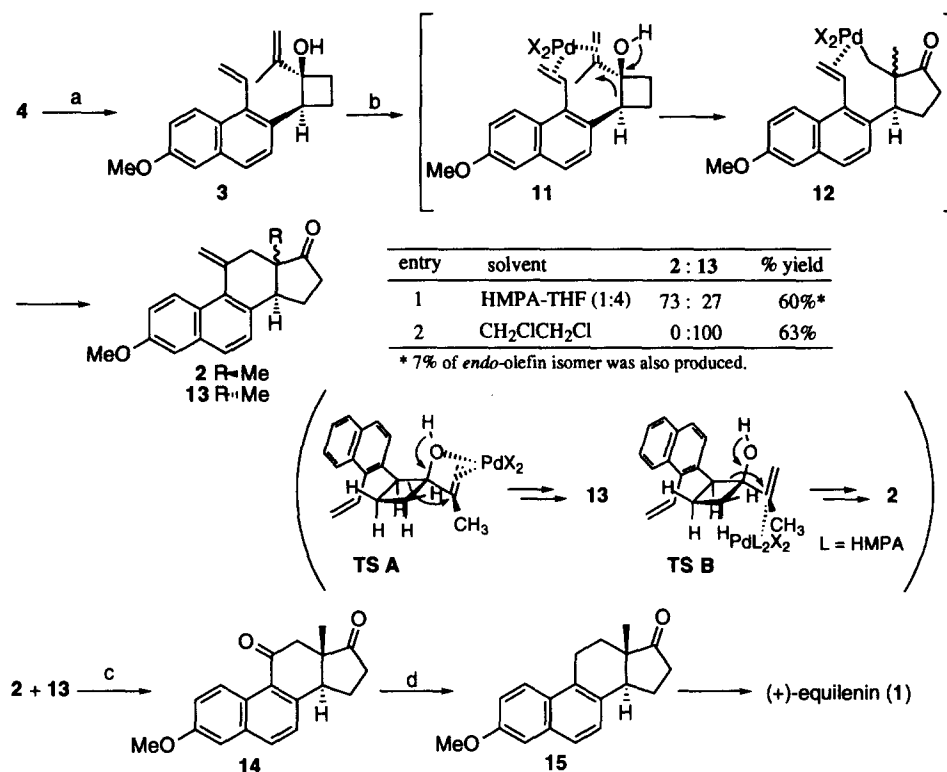


**Scheme 2.** Reagents and conditions. a)  $\text{TiF}_2\text{O}$ , DMAP, pyridine,  $0^\circ\text{C}$ , 61%; b) *tri-n*-butylvinylstannane,  $\text{Pd}(\text{PPh}_3)_4$ , LiCl, THF, reflux, 98%; c) cyclopropylenetriphenylphosphorane, NaH, THF,  $62^\circ\text{C}$ , 70%; d) 5 mol % catalyst 9, NaClO, 4-PPNO,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C}$ , 55%, 78% e.e.

The chiral cyclobutanone 4 was then converted stereoselectively to the isopropenylcyclobutanol 3 by Grignard reaction with isopropenylmagnesium bromide in the presence of cerium trichloride<sup>9</sup> (82%) (Scheme 3). Next, the second crucial stage in the synthesis,  $\text{Pd}^{\text{II}}$ -mediated cascade ring expansion–insertion reaction, had to be investigated.<sup>2</sup> On the basis of our previous results, we further examined various reaction conditions to construct diastereoselectively the *trans*-naphthohydrindan from 3. Consequently, the *trans*-fused product 2 was selectively produced *via* ring expansion–insertion reaction (11→12) utilizing  $\text{Pd}(\text{OAc})_2$ <sup>10</sup> (1 eq.) in HMPA-THF (1:4) (entry 1) (2:13 = 73:27, 60%, and 7% of *endo*-olefin isomer). Interestingly, when the solvent was changed to 1,2-dichloroethane, the *cis*-fused product 13 was obtained as a sole product (63%) (entry 2). These remarkable effects indicate that solvent polarity is an important factor to control the diastereoselectivity of products. Thus, in non-polar solvent such as 1,2-dichloroethane, the ring expansion reaction has been suggested to proceed *via* intermediate TS A to give 13, in which palladium was associated with olefin and internal alcohol.

In contrast, in the case of polar solvent such as HMPA, the reaction seems to proceed *via* TS B to give **2** in which palladium was associated with only olefin because solvent itself associated to palladium as a ligand.

To complete the synthesis of equilenin, the mixture (73:27) of **2** and **13** was treated with osmium tetroxide and sodium periodate to furnish diketone **14**<sup>11</sup> after the separation of its diastereomer (59% from **2** prepared by entry 1). Finally, the selective reduction of the benzylic ketone of **14** was carried out by hydrogenolysis on Pt-C in the presence of PdCl<sub>2</sub><sup>12</sup> to afford equilenin methyl ether **15**<sup>13</sup> (82%), which could be optically pure form after recrystallization. Since **15** has been converted to **1** with boron tribromide,<sup>13</sup> our asymmetric synthesis of (+)-equilenin (**1**) was achieved.



**Scheme 3.** Reagents and conditions. a) isopropenylmagnesium bromide, CeCl<sub>3</sub>, THF, -78°C, 82%; b) Pd(OAc)<sub>2</sub>, solvent, RT (see above); c) OsO<sub>4</sub>, NaIO<sub>4</sub>, acetone-H<sub>2</sub>O, 59% from **2** (entry 1); d) H<sub>2</sub>, Pt-C, PdCl<sub>2</sub>, EtOH, RT, 82%.

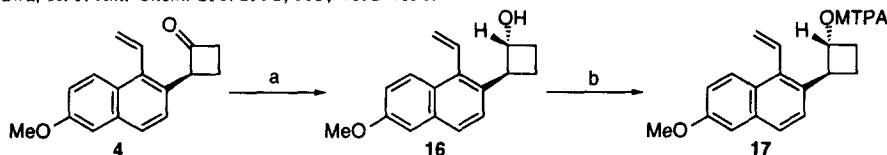
In summary, a new type of cascade ring expansion reactions of small ring systems has been successfully applied to an asymmetric synthesis of (+)-equilenin (**1**).

## Acknowledgments

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- The spectral data for **4**:  $[\alpha]_D^{25} +37.5$  (c 0.2 in  $\text{CHCl}_3$ ); IR (neat) 1780  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  2.21-2.22 (1H, m), 2.24-2.55 (1H, m), 3.07-3.13 (1H, m), 3.18-3.27 (1H, m), 3.91 (3H, s), 4.95-5.05 (1H, m), 5.44 (1H, dd,  $J = 2.2$  and 18.0 Hz), 5.76 (1H, dd,  $J = 2.2$  and 12.0 Hz), 7.08 (1H, dd,  $J = 12.0$  and 18.0 Hz), 7.11-7.15 (2H, m), 7.35 (1H, d,  $J = 8.8$  Hz), 7.65 (1H, d,  $J = 8.8$  Hz), 7.96 (1H, d,  $J = 8.8$  Hz);  $^{13}\text{C-NMR}$  (75 MHz,  $\text{CDCl}_3$ )  $\delta$  19.6, 44.9, 55.3, 63.3, 106.1, 118.8, 122.2, 125.3, 126.8, 127.3, 127.4, 129.4, 133.9, 134.1, 135.6, 157.6, 209.5; MS  $m/z$  252 ( $M^+$ ); HRMS calcd for  $\text{C}_{17}\text{H}_{16}\text{O}_2$  252.1149 ( $M^+$ ), found 252.1150. Enantiomeric excess was determined by HPLC analysis (Chiralcel OA column, 10% isopropanol-hexane, 0.5 mL/min,  $\lambda = 254$  nm, 23°C, retention times 21.1min (*R*), 26.2min(*S*)). Absolute configuration was determined by Kusumi's procedure using MTPA esters **17**, which were prepared by reduction of the cyclobutanone **4**, followed by esterification of the major isomer **16** with (*R*) or (*S*)-MTPA (see below): Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, 113, 4092-4096.



Reagents and conditions. a)  $\text{NaBH}_4$ , MeOH, RT, 57% (and 35% of its diastereomer); b) (*R*)-, and (*S*)-MTPA acid, DCC, DMAP,  $\text{CH}_2\text{Cl}_2$ , RT, quant.

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- When  $\text{PdCl}_2(\text{CH}_3\text{CN})_2$  was used, *endo*-olefin isomers were obtained exclusively. In this case, it is presumed that H-Pd<sup>+</sup> complex which was produced *in situ* caused olefin isomerization. In case of using  $\text{Pd}(\text{OAc})_2$ , it seems that H-Pd<sup>+</sup> complex wasn't effectively produced for less acidity of AcOH than that of HCl generated *in situ* resulted to preserve the isomerization.
- The spectral data for **14**: m.p. 182°C (decomp.),  $[\alpha]_D^{25} -31.7$  (c 0.1 dioxane); IR ( $\text{CHCl}_3$ ) 1660, 1740  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.83 (3H, s), 2.08-2.22 (1H, m), 2.35-2.52 (1H, m), 2.58-2.77 (2H, m), 2.70 (1H, d,  $J = 18.9$  Hz), 2.94 (1H, d,  $J = 18.9$  Hz), 3.34 (1H, dd,  $J = 6.3$  and 12.5 Hz), 3.94 (3H, s), 7.15 (1H, d,  $J = 2.8$  Hz), 7.29-7.38 (2H, m), 7.98 (1H, d,  $J = 8.7$  Hz), 9.22 (1H, d,  $J = 8.7$  Hz);  $^{13}\text{C-NMR}$  (75 MHz,  $\text{CDCl}_3$ )  $\delta$  15.0, 21.9, 35.8, 46.0, 48.8, 48.9, 55.3, 106.6, 121.8, 123.7, 126.7, 127.0, 128.0, 134.1, 140.9, 157.6, 199.0, 217.8; MS  $m/z$  292 ( $M^+$ ); HRMS calcd for  $\text{C}_{15}\text{H}_{18}\text{O}_3$  294.1256 ( $M^+$ ), found 294.1246.
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- Takano, S.; Inomata, K.; Ogasawara, K. *J. Chem. Soc. Chem. Commun.* **1990**, 1544. The spectral data for **15**: m.p. 198°C (decomp.), lit., m.p. 199°C (decomp.).  $[\alpha]_D^{23} +78.7^\circ$  (c 0.1 dioxane), lit.,  $[\alpha]_D^{29} +81.9^\circ$  (c 0.4 dioxane). The  $^1\text{H-NMR}$  spectrum was in complete agreement with that of an authentic sample.