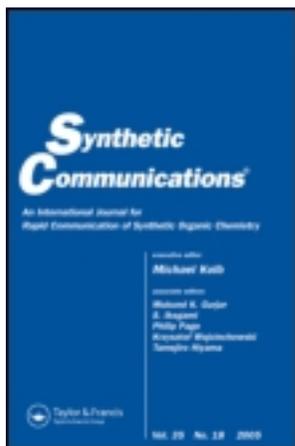


This article was downloaded by: [Moskow State Univ Bibliote]
On: 12 November 2013, At: 00:48
Publisher: Taylor & Francis
Informa Ltd Registered in England and Wales Registered Number: 1072954
Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH,
UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lcyc20>

First Total Synthesis of Desmosdumotin C

Kyoko Nakagawa-Goto ^a, Jiu-Hong Wu ^b & Kuo-Hsiung Lee ^a

^a Natural Products Laboratory, School of Pharmacy, University of North Carolina, Chapel Hill, North Carolina, USA

^b Department of Pharmacy, Beijing, China
Published online: 15 Aug 2006.

To cite this article: Kyoko Nakagawa-Goto, Jiu-Hong Wu & Kuo-Hsiung Lee (2005) First Total Synthesis of Desmosdumotin C, *Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry*, 35:13, 1735-1739, DOI: [10.1081/SCC-200063907](https://doi.org/10.1081/SCC-200063907)

To link to this article: <http://dx.doi.org/10.1081/SCC-200063907>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with

primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

First Total Synthesis of Desmosdumotin C[†]

Kyoko Nakagawa-Goto

Natural Products Laboratory, School of Pharmacy, University of North Carolina, Chapel Hill, North Carolina, USA

Jiu-Hong Wu

Department of Pharmacy, Beijing, China

Kuo-Hsiung Lee

Natural Products Laboratory, School of Pharmacy, University of North Carolina, Chapel Hill, North Carolina, USA

Abstract: Desmosdumotin C (**1**), a novel compound isolated from the roots of *Desmos dumosus*, was synthesized from 2,4,6-trihydroxyacetophenone (**2**), confirming the assigned structure of the natural product.

Keywords: Antitumor, cytotoxicity, desmosdumotin C, total synthesis

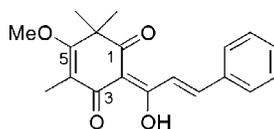
INTRODUCTION

Desmosdumotin C, a novel compound,^[1] was isolated recently by our group from the roots of *Desmos dumosus*. The structure of **1**, which was established mainly by NMR and X-ray analyses, has a cyclohexe-4-ene-1,3-dione skeleton with a conjugated cinnamoyl moiety on the C-2 position and a *gem*-dimethyl group on the C-6 position. Similar compounds, safflomin C,^[2] syzygiol,^[3] and uliginosin A analog,^[4] are known. Although these latter compounds exist as tautomeric mixtures, the isolated desmosdumotin C exists as one enolic tautomer.

Received in the USA June 27, 2004

[†]Antitumor agent 242

Address correspondence to Kuo-Hsiung Lee, Natural Products Laboratory, School of Pharmacy, University of North Carolina, Chapel Hill, NC 27599-7360, USA. Tel.: 919-962-0066; Fax: 919-966-3893; E-mail: khlee@unc.edu



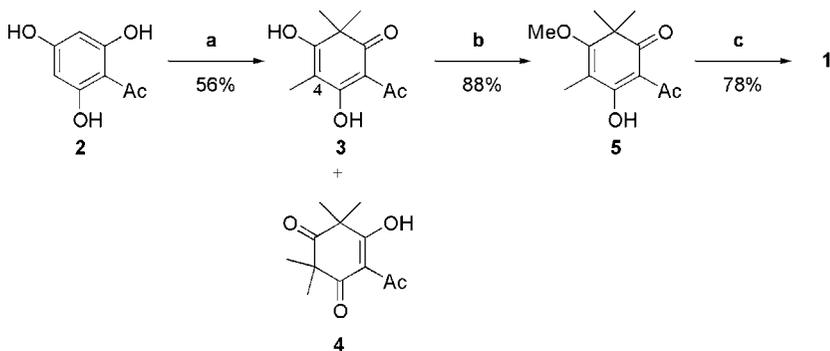
Desmosdumotin C (1)

Figure 1. The structure of Desmosdumotin C.

Desmosdumotin C (1) represents a promising new anticancer lead structure for further new analog development based on its *in vitro* evaluation against a panel of six cancer cell lines.^[1] Thus, to obtain a larger quantity than available from the plant source and to confirm the structure of the natural product, we herein report the first simple total synthesis of 1.

EXPERIMENTAL

Reaction of 2,4,6-trihydroxyacetophenone (2) with three equivalents of methyl iodide in the presence of sodium methoxide produced 2-acetyl-3-hydroxy-4,6,6-trimethylcyclohexa-2,4-dienone (3)^[5] in 56% yield along with tetramethyl 4 in 9% yield. The selective methoxylation of 3 to give 5 was achieved in 88% yield by treatment with TMSCHN₂ at a low temperature. Employing a higher temperature gave a low yield of the desired compound, because of the production of unknown side product(s). The use of other conditions, such as K₂CO₃ and MeI or Li₂CO₃ and MeI,^[6] yielded only tetramethyl derivative 4, because of the presence of a strongly acidic proton on the C-4 position (Scheme 1).



Scheme 1. Reagents and conditions: a) MeI (3.3 mol eq.), 30% NaOMe in MeOH (3.7 mol eq.), MeOH, reflux, 7 h; b) 2.0 M TMSCHN₂ in Et₂O (excess), EtOAc/MeOH, -78 °C, 3 h; c) 50% KOH aq., EtOH, PhCHO, rt, 23 h.

Finally, desmosdumotin C (**1**) was obtained in 78% yield by the aldol condensation of **5** with benzaldehyde in 50% KOH aqueous solution. Recrystallization from CHCl_3 -MeOH gave **1** as yellow needles. The synthetic product has identical spectral data with those of the natural product.

All melting points were taken on Fisher-Johns and Mel-Temp II melting-point instruments and are uncorrected. IR spectra were recorded on a Perkin-Elmer 1320 spectrophotometer. ^1H NMR spectra were obtained using a Varian Gemini 2000 (300-MHz) NMR spectrometer with TMS as the internal standard. All chemical shifts are reported in ppm. FABMS and HRFABMS spectral analyses were determined on a JOEL HX-110 instrument. Analytical thin-layer chromatography (TLC) was carried out on Merck precoated aluminum silica-gel sheets (Kieselgel 60 F-254). All target compounds were characterized by ^1H , IR, and MS analyses.

2-Acetyl-3,5-dihydroxy-4,6,6-trimethylcyclohexa-2,4-dione (**3**)

A solution of 2,4,5-trihydroxyacetophenone (**2**, 487 mg, 2.9 mmol) and sodium methoxide (1.95 mL, 10.8 mmol, 30% MeOH solution) in anhydrous MeOH (3 mL) containing methyl iodide (0.6 mL, 9.6 mmol) was refluxed for 7 h. The reaction mixture was cooled to 0°C and acidified with 1 N of aqueous HCl, then extracted three times with EtOAc. The combined organic layers were dried over Na_2SO_4 and concentrated in vacuo. The residue was chromatographed on silica gel with EtOAc-hexane (1:9 to 1:4, v/v) as an eluent to provide **3** (343 mg, 56%). Pale yellow prisms, mp: 164 – 165°C (EtOAc-hexane). IR (KBr): 3099 (br), 1656, 1590, 1525, 1474, 1387, 1331, 1278, 1193, 1153 cm^{-1} . ^1H NMR (300 MHz, DMSO-d_6): $\delta = 18.94$ (br s, 1H, chelated-OH), 2.42 [s, 3H, $\text{C}(\text{O})\text{CH}_3$], 1.73 (s, 3H, CH_3), 1.23 (s, 6H, $\text{CH}_3 \times 2$). ^{13}C NMR (300 MHz, DMSO-d_6): $\delta = 199.4$ ($\text{C}=\text{O}$), 196.0 ($\text{C}=\text{O}$), 188.7 ($\text{C}-\text{OH}$), 176.2 ($\text{C}-\text{OCH}_3$), 105.0 ($\text{C}-2$), 101.7 ($\text{C}-6$), 48.2 [$\text{C}(\text{CH}_3)_2$], 27.8 [$\text{C}(\text{O})\text{CH}_3$], 24.3 ($\text{CH}_3 \times 2$), 7.3 (CH_3). MS m/z 209 ($\text{M}^+ - 1$).

2-Acetyl-3-hydroxy-5-methoxy-4,6,6-trimethylcyclohexa-2,4-dione (**5**)

To a solution of **3** (447 mg, 2.13 mmol) in anhydrous EtOAc-MeOH (5:1, 6 mL), a solution of TMSCHN_2 in diethyl ether (5 mL, 10 mmol, 2 M) was added slowly at -78°C under argon atmosphere, and the mixture was stirred for 2 h. At this time, a second aliquot of TMSCHN_2 (4 mL, 8 mmol, 2 M) was added and stirring was continued at -78°C . Acetic acid was then added to destroy the excess TMSCHN_2 . The mixture was concentrated in vacuo, and the residue was purified by silica-gel column chromatography with EtOAc-hexane (1:2) as an eluent to obtain **5** (418 mg, 88%). Yellow oil: IR (KBr): 2979, 2937, 2874, 1659, 1636, 1536, 1469, 1374, 1202, 1131,

978, 915 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ = 18.93 (s) and 18.14 (s) (2:1, 1H, chelated-OH), 3.93 (s) and 3.85 (s) (2:1, 3H, OCH_3), 2.69 (s) and 2.59 (s) [1:2, 3H, $\text{C}(\text{O})\text{CH}_3$], 1.95 (s) and 1.90 (s) (2:1, 3H, CH_3), 1.43 (s) and 1.31 (s) (1:2, 6H, $\text{CH}_3 \times 2$). ^{13}C NMR (300 MHz, CDCl_3) δ C-H COSY (*gem*-dimethyl), 7.3. MS m/z 223 ($\text{M}^+ - 1$).

Desmosdumotin C (1)

A solution of **4** (62 mg, 0.28 mmol) in EtOH (1 mL) and 50% KOH in water (1 mL) containing benzaldehyde (0.1 mL, 0.99 mmol) was stirred at rt for 23 h. The reaction mixture was poured into 1 N of ice-cold HCl, then extracted with CH_2Cl_2 . The extract was washed with brine, dried over Na_2SO_4 , and concentrated in vacuo. The residue was chromatographed on silica gel with EtOAc-hexane (1:9, v/v) as an eluent to afford a crystalline solid (80 mg). Recrystallization from CH_2Cl_2 -MeOH gave desmosdumotin C (39 mg without its tautomer) as yellow needles (Figure 1). A second recrystallization of the mother liquor using CH_2Cl_2 -hexane gave 29 mg of **1**, as a mixture with its tautomer. In sum, 68 mg (78%) of **1** were obtained. Yellow needles, mp 98–99 °C (CHCl_3 -MeOH, lit.: 96–97 °C). IR (KBr): 3101, 2978, 2936, 1656, 1623, 1514, 1448, 1425, 1203, 1152, 1119, 977, 944, 759, 698 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ = 19.17 (s, 1H, chelated-OH), 8.33 (d, 1H, $J = 15.5$ Hz), 7.92 (d, 1H, $J = 15.5$ Hz), 7.71–7.62 (m, 2H, Ar-2'', 6''-H), 7.42–7.34 (m, 3H, Ar-3'', 4'', 5''-H), 3.95 (s, 3H, OCH_3), 1.99 (s, 3H, Ar- CH_3), 1.37 (s, 6H, $\text{CH}_3 \times 2$). ^{13}C -NMR (300 MHz, CDCl_3): δ = 198.0 (C-1), 192.4 (C-3), 187.2 (C-1'), 176.6 (C-5), 144.9 (C-2', C-3'), 135.2 (C-1''), 130.6 (C-3'', C-5''), 128.8 (C-4''), 123.2 (C-2'', C-6''), 113.6 (C-2), 106.6 (C-4), 62.1 (OCH_3), 50.4 (C-6), 24.3 ($\text{CH}_3 \times 2$), 9.8 (Ar- CH_3). MS m/z 313 ($\text{M}^+ + 1$).

ACKNOWLEDGMENTS

This study was supported by NIH Grant CA17625 awarded to K. H. L. Thanks are also due to Grant No. 30271533 awarded to J. H. W. from the National Natural Science Foundation of China.

REFERENCES

1. Wu, J. H.; McPhail, A. T.; Bastow, K. F.; Shiraki, H.; Ito, J.; Lee, K. H. Desmosdumotin C, a novel cytotoxic principle from *Desmos dumosus*. *Tetrahedron Lett.* **2002**, *43* (8), 1391–1393.
2. (a) Onodera, J.; Obara, H.; Hirose, R.; Matsuba, S.; Sato, N.; Sato, S.; Suzuki, M. The structure of safflomin C, a constituent of safflower. *Chem. Lett.* **1989** (9), 1571–1574; (b) Sato, S.; Obara, H.; Onodera, J.; Endo, A.; Matsuba, S. Synthesis of model compounds of safflomin C. *Bull. Chem. Soc. Jpn.* **1992**, *65* (2), 452–457.

3. (a) Nishizawa, M.; Yamada, H.; Sano, J.; Ito, S.; Hayashi, Y.; Ikeda, H.; Shiro, M.; Tokuda, H. Structure of syzygiol: A skin-tumor promotion inhibitor. *Tetrahedron Lett.* **1991**, 32 (2), 211–212; (b) Sato, S.; Obara, H.; Endo, A.; Onodera, J.; Matsuba, S. Synthesis of syzygiol; A skin-tumor promotion inhibitor. *Bull. Chem. Soc. Jpn.* **1992**, 65 (9), 2552–2554.
4. Meikle, T.; Stevens, R. β -Tricarbonyl compounds. Part 1. Synthesis of the antibiotics uliginosin A, dihydrouliginosin B, and analogs thereof. *J. Chem. Soc., Perkin Trans. 1* **1978** (11), 1303–1312.
5. (a) Riedl, W.; Nickl, J.; Risse, K. H.; Mitteldorf, R. Bitter principles of hops. X. Constituents of *Filix mas*. 4. Nuclear alkylation of phloroacylophenones. *Chem. Ber.* **1956**, 89, 1849–1863; (b) Nowy, G.; Riedl, W.; Simon, H. Methylation products of tetra-C-methylphloroacetophenone and 2-acetyldihydroresorcinol. New synthesis of benzofuran derivatives. *Chem. Ber.* **1966**, 99 (7), 2075–2082.
6. Wymann, W. E.; Davis, R.; Patterson, J. W., Jr.; Pfister, J. R. Selective alkylations of certain phenolic and enolic functions with lithium carbonate-alkyl halide. *Synth. Commun.* **1988**, 18 (12), 1379–1384.