

Tetrahedron Letters 41 (2000) 525-529

TETRAHEDRON LETTERS

Stereocontrolled introduction of an amino group at C-6 of D-galactose via (3,3)-sigmatropic rearrangements — novel synthesis of lincosamine and 7-*epi*-lincosamine precursors

Jozef Gonda,^{a,*} Eva Zavacká,^a Miloslav Budešínský,^b Ivana Císarová^c and Jaroslav Podlaha^c ^aDepartment of Organic Chemistry, P.J. Šafarik University, Moyzesova 11, 04167 Košice, Slovak Republic

^bInstitute of Organic Chemistry and Biochemistry, Academy of Sciences of the Czech Republic, 16610 Prague 6, Czech Republic

^cDepartment of Inorganic Chemistry, Charles University, Albertov 2030, 12840 Prague 2, Czech Republic

Received 20 September 1999; accepted 10 November 1999

Abstract

A new and stereoselective route to the aminoglycoside components of the antibiotics lincomycin and clindamycin is presented. The key step involves diastereoselective introduction of the amino group at C-6 of D-galactose by (3,3)-signatropic rearrangements of the corresponding allylic (*Z*)-trifluoroacetimidate and (*Z*)- and (*E*)-allylic thiocyanates. Epoxidation of the resulting trifluoroacetamide with *m*-CPBA led to the epoxide with high *threo*selectivity. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: rearrangements; diastereoselection; imidic acid; imidic acid derivatives; thiocyanates.

Lincomycin 1 and the semi-synthetic 7-chloro analogue clindamycin 2 are commercial antibiotics used against bacterial infections in humans and animals.¹ Synthetic work on the preparation of 1 and 2 has mainly been focused on lincosamine (3), i.e. the carbohydrate moiety of lincomycin (Fig. 1). Synthesis of 3^2 entailed the stereocontrolled construction of the pyranose ring using furan-based chemistry³ or hetero Diels–Alder cycloaddition.⁴ Syntheses starting from *myo*-inositol⁵ or D-threonine⁶ have also



* Corresponding author. Fax: +421-95-62 22124; e-mail: jgonda@kosice.upjs.sk (J. Gonda)

^{0040-4039/00/}\$ - see front matter © 2000 Elsevier Science Ltd. All rights reserved. *PII:* S0040-4039(99)02110-3

been reported. An alternative approach⁷ towards the preparation of **3** comprises a chain-extension of D-galactopyranose, illustrating the very hindered nature of C-6 and the difficulty of introducing the amino group at this position stereoselectively. An elegant solution to this problem has been independently presented by Stick⁸ and Knapp⁹ and recently by van Delft.¹⁰

This paper concerns a simple approach to introducing a nitrogen atom at C-6 of galactose, stereoselectively via aza-Claisen rearrangements of trifluoroacetimidates and thiocyanates and subsequent diastereoselective epoxidation of the resulting double-bonds leading to 7,8-anhydro-6-deoxy-6-trifluoroacetamide-1,2:3,4-di-*O*-isopropylidene-D-*threo*-D-*galacto*-octopyranose **18** as a useful lincosamine and 7-*epi*-lincosamine precursor. This methodology was previously used for the stereoselective synthesis of the (+)-polyoxamic acid,¹¹ thymine polyoxin C¹² and 5'-branched 5'-aminoribonucleoside derivatives.¹³

The starting (*Z*)- and (*E*)-allylic imidates **7**, **8**, **9**, and **10** were synthesized from the 1,2:3,4-di-*O*-isopropylidene- α -D-*galacto*-hexaldioaldo-1,5-pyranose **4** by the route shown in Scheme 1. The Wittig olefination of **4** with Ph₃P=CHCOOCH₃ in MeOH and reduction of the resulting ester with DIBAL-H led selectively to the (*Z*)-allylic alcohol **5**.¹⁴ Side-chain elongation of the dialdose **4** with (CH₃O)₂POCH₂COOCH₃/NaH in THF and reaction of the (*E*)-ester using DIBAL-H afforded the (*E*)allylic alcohol **6**, selectively. Subsequently, **5** was converted to the (*Z*)-trichloroacetimidate **8** and (*Z*)trifluoroacetimidate **10** by reaction with CCl₃CN/NaH and CF₃CN/NaH (diethyl ether, 0°C), respectively. The imidates (*E*)-**7** and (*E*)-**9** were prepared from (*E*)-allylic alcohol **6** (Scheme 1). The crude imidates were purified by flash chromatography (SiO₂, diethyl ether:hexane=50:50).



Thermal rearrangement of the trichloroacetimidates (*E*)-7 and (*Z*)-8 (in xylene at 200°C, 30 h) takes place with decomposition and gave the trichloroacetamides **11** and **12** obtained as a mixture (50:50) in low yields (22 and 25%) (Scheme 1). In contrast, thermolysis (2 h, 200°C) of the (*E*)-trifluoroacetimidate **9** afforded a 42:58 mixture of the trifluoroacetamides **13** and **14** in 93% yield. Finally, rearrangement of the (*Z*)-trifluoroacetimidate **10** gave the trifluoroacetamides **13** and **14** with good stereoselectivity (**13**:**14**=9:91) and yield (95%) (Scheme 1). The (3,3)-sigmatropic rearrangement of allylic imidates can often be accelerated by the use of metal catalysis, particularly by the Hg(II) and Pd(II) species.¹⁵ However, despite repeated attempts, no rearrangement was observed when **7**–**10** were exposed to such catalysts. These negative results can be compared with literature data.¹⁶ The resulting product 14^{\dagger} was then transformed into the α -aminoaldehyde 15 (Scheme 2) which is a precursor of lincosamine (3).² We found that epoxidation of 14 with *m*-CPBA in CH₂Cl₂ (2 days at RT) led to the epoxide 18^{\ddagger} with high *threo*-selectivity¹⁷ (18:19=97:3) (Scheme 2).



The configurations of the newly introduced stereocenters at C-6 and C-7 in **18** were established by X-ray analysis (Fig. 2), thus confirming the configurations at C-6 and C-7 as (R).



Fig. 2. ORTEP diagram of epoxide 18 showing crystallographic numbering

[†] All compounds showed ¹H, ¹³C, IR, and HRMS spectra consistent with the reported structures. Spectroscopic data of **14**: ¹H NMR data (500 MHz, CDCl₃): 1.33 (3H, s, CH₃), 1.35 (3H, s, CH₃), 1.50 (3H, s, CH₃), 1.52 (3H, s, CH₃), 3.86 (1H, dd, H5, $J_{5,4}=1.8$ Hz, $J_{5,6}=4.7$ Hz), 4.32 (1H, dd, H2, $J_{2,1}=5.1$ Hz, $J_{2,3}=2.3$ Hz), 4.42 (1H, dd, H4, $J_{4,3}=8.1$ Hz, $J_{4,5}=1.8$ Hz), 4.61 (1H, dd, H3, $J_{3,2}=2.3$ Hz, $J_{3,4}=8.1$ Hz), 4.79 (1H, m, H6), 5.32 (1H, ddd, H8a, $J_{8a,8b}=1.7$ Hz, $J_{8a,7}=10.4$ Hz, $J_{7,8a}=10.4$ Hz, $J_{7,8b}=17.4$ Hz, $J_{7,6}=5.9$ Hz), 8.06 (1H, bd, N–H, J=0.8); ¹³C NMR data (125.7 MHz, CDCl₃): 23.80 (CH₃), 24.77 (CH₃), 25.45 (CH₃), 26.03 (CH₃), 54.71 (C-6), 65.88 (C-2), 70.28 (C-4), 70.94 (C-4), 71.69 (C-5), 96.64 (C-1), 108.80 (*C*(CH₃)₂), 109.73 (*C*(CH₃)₂), 116.02 (*C*F₃), 118.39 (C-8), 132.15 (C-7), 156.72 (*C*=O).

[‡] Spectroscopic data of **18**: ¹H NMR data (500 MHz, CDCl₃): 1.34 (3H, s, CH₃), 1.37 (3H, s, CH₃), 1.47 (3H, s, CH₃), 1.52 (3H, s, CH₃), 2.54 (1H, dd, H8b, $J_{8b,8a}$ =4.5 Hz, $J_{8b,7}$ =2.6 Hz), 2.80 (1H, dd, H8a, $J_{8a,8b}$ =4.5 Hz, $J_{8a,7}$ =4.0 Hz), 3.28 (1H, dt, H7, $J_{7,8a}$ =4.0 Hz, $J_{7,8b}$ =2.6 Hz, $J_{7,6}$ =2.4 Hz), 4.04 (1H, dd, H5, $J_{5,6}$ =4.7 Hz, $J_{5,4}$ =1.7 Hz), 4.34 (1H, dd, H2, $J_{2,1}$ =4.9 Hz, $J_{2,3}$ =2.3 Hz), 4.52 (1H, dd, H4, $J_{4,3}$ =8.1 Hz, $J_{4,5}$ =1.7 Hz), 4.55 (1H, ddd, H6, $J_{6,5}$ =4.7 Hz, $J_{6,7}$ =2.4 Hz, $J_{6,NH}$ =8.6 Hz), 4.65 (1H, dd, H3, $J_{3,2}$ =2.3 Hz, $J_{3,4}$ =8.1 Hz), 5.55 (1H, d, H1, $J_{1,2}$ =4.9 Hz), 7.49 (1H, bd, NH, J=8.6 Hz); ¹³C NMR data (125.7 MHz, CDCl₃): 23.98 (CH₃), 24.80 (CH₃), 25.53 (CH₃), 26.05 (CH₃), 43.80 (C-8), 49.92 (C-7), 51.11 (C-6), 66.42 (C-2), 70.44 (C-3), 70.90 (C-4), 72.23 (C-5), 96.57 (C-1), 108.90 (*C*(CH₃)₂), 109.85 (*C*(CH₃)₂), 116.0 (*C*F₃), 157.43 (*C*=O).

With the aim of examining similar methodology for the introduction of nitrogen functionality into the 6-position of galactose, we investigated the rearrangement of the allylic thiocyanates 20 and 21 (Scheme 3). The starting thiocyanates 20 and 21 were prepared by S_N^2 displacement of the O-mesyl group in the corresponding mesylates derived from the (E)-allylic alcohol 6 and the (Z)-allylic alcohol 5 by the thiocyanate group (KSCN, CH_3CN , rt, 0.5 h). Thermal rearrangement of the thiocyanates 20 and 21 was carried out at 80°C in xylene under N₂ for 2 h to give a high yield of isothiocyanates 22 and 23 (95% for 20, 90% for 21) and good stereoselectivity (16:17=96:4, in both cases) (Scheme 3). The rearrangement thiocyanate → isothiocyanate is reversible and ratios of isothiocyanates 22 and 23 are time and temperature dependent. We found that heating the thiocyanates 20 and 21 in xylene at 140°C for 2 h gave mixtures of the isothiocyanates 22 and 23 in 50:50 ratio; heating for 16 h at the same temperature led to a mixture of 22 and 23 in a 25:75 ratio. Reaction of 22 with thioacetic acid¹⁸ (1.3 mol) and t-BuOK (0.2 mol) in the presence 18-crown-6-ether in THF at 60°C (4 h) gave 6-acetamido-6,7,8-trideoxy-1,2:3,4di-O-isopropylidene-D-glycero-D-galacto-oct-6-enopyranose 25. The absolute configuration at C-6 in 22 was unambiguously determined by chemical transformations (Scheme 3). Thus, the reaction of the isothiocyanate 22 with toluene-3,4-dithiol in MeOH/H₂O led to the amine 24 in 69% yield which, after reaction with (CF₃CO)₂O/NEt₃ in CH₂Cl₂, afforded trifluoroacetamide 14 in 89% yield. This compound is in all respects identical with 14 prepared from 10 (Scheme 3).



Acknowledgements

The present work was supported by a Grant Agency (No. 1/3229/96) from the Ministry of Education, Slovak Republic and Tempus Structural Joint European Project S-JEP-09101-95.

References

- 1. Mason, D. J.; DeBoer, C. Antimicrob. Agents Chemother. 1962, 554. Herr, R. R.; Bergy, M. A. Antimicrob. Agents Chemother. 1962, 560.
- Golebiowsky, A.; Jurczak, J. In *Total Synthesis of Lincomycin and Related Chemistry*; Lukacs, G.; Ohno, M., Eds. Recent progress in the chemical synthesis of antibiotics. Springer Verlag: Berlin-Heidelberg, 1990; p. 365.
- 3. Golebiowsky, A.; Jurczak, J. Tetrahedron 1991, 47, 1045. Szechner, B.; Achmatowicz, O. Pol. J. Chem. 1994, 68, 1149.
- 4. Danishefsky, S. J.; Larson, E.; Springer, J. P. J. Amer. Chem. Soc. 1985, 107, 1274.
- 5. Chida, N.; Nakazawa, K.; Ninomiya, S.; Amano, S.; Koizumi, K.; Inaba, J.; Ogawa, S. Carbohydrate Lett. 1995, 1, 335.
- 6. Marshall, J. A.; Beaudoin, S. J. Org. Chem. 1996, 61, 581.
- Magerlein, B. J. Tetrahedron Lett. 1970, 33. Howarth, G. B.; Lancze, D. G.; Szarek, W. A.; Jones, J. K. N. Can. J. Chem. 1969, 47, 75. Howarth, G. B.; Szarek, W. A.; Jones, J. K. N. J. Chem. Soc. C 1970, 2218. Saeki, H.; Ohki, E. Chem. Pharm. Bull. 1970, 18, 789. Hems, R.; Horton, D.; Nakadate, M. Carbohydr. Res. 1972, 25, 205. Atsumi, T.; Fukumaru, T.; Matsui, M. Agr. Biol. Chem. 1973, 37, 2627. Woolard, G. R.; Rathbone, E. B.; Szarek, W. A.; Jones, J. K. N. J. Chem. Soc., Perkin Trans. 1 1976, 950. Gateau-Olesker, A.; Sepulchre, A. M.; Vass, G.; Géro, S. D. Tetrahedron 1977, 1474. Czernecki, S.; Valéry, J.-M. J. Carbohydr. Chem. 1990, 9, 767. Bose, A. K.; Mathur, C.; Wagle, D. R.; Naqvi, R.; Manhas, M. S. Heterocycles 1994, 39, 491.
- 8. Engelhardt, L. M.; Skelton, B. W.; Stick, R. V.; Tilbrook, D. M. G.; White, A. H. Aust. J. Chem. 1990, 43, 1657.
- 9. Knapp, S.; Kukkola, P. J. J. Org. Chem. 1990, 55, 1632.
- 10. van Delft, F. L.; de Kort, M.; van der Marel, G. A.; van Boom, J. H. J. Org. Chem. 1996, 61, 1883.
- 11. Savage, I.; Thomas, E. J. J. Chem. Soc., Chem. Commun. 1989, 11, 717.
- 12. Chen, A.; Savage, I.; Thomas, E. J.; Wilson, P. D. Tetrahedron Lett. 1993, 34, 6769.
- 13. Ammenn, J.; Altmann, K.-H.; Belluš, D. Helv. Chim. Acta 1997, 80, 1989.
- 14. Brimacombe, J. S.; Hanna, R.; Kabir, A. K. M. S.; Bennet, F.; Taylor, I. D. J. Chem. Soc., Perkin Trans. 1, 1986, 815.
- 15. Overman, L. E. Angew. Chem., Int. Ed. Engl. 1984, 23, 579.
- Metz, P.; Mues, C.; Schoop, A. *Tetrahedron* 1992, 48, 1071. Mehmandoust, M.; Petit, Y.; Larchevêque, M.; *Tetrahedron Lett.* 1992, 33, 4313. Tanner, D.; He, H. M. Acta Chem. Scand. 1993, 47, 592.
- 17. Luly, J. R.; Dellaria, J. F.; Plattner, J. J.; Soderquist, J. L.; Li, N. J. Org. Chem. **1987**, 52, 1487. Romeo, S.; Rich, D. H. *Tetrahedron Lett.* **1993**, 34, 7187. Albeck, A.; Persky, R. J. Org. Chem. **1994**, 59, 653.
- 18. Drobnica, L.; Kristian, P.; Augustin, J. In *The Chemistry of Cyanates and their Thio Derivatives*; Patai, S., Ed.; John Wiley: New York, 1977; pp. 1003–1221.