

# $m$ -C<sub>Ar</sub>-H Bond Alkylations and Difluoromethylation of Tertiary Phosphines Using a Ruthenium Catalyst

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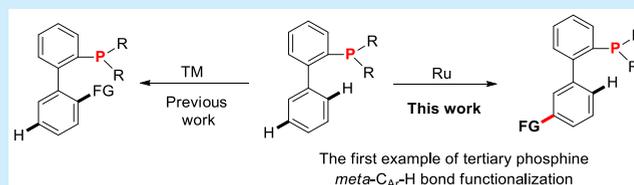


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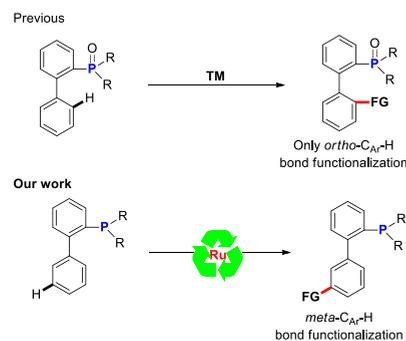
**ABSTRACT:**  $m$ -C<sub>Ar</sub>-H bond functionalization of tertiary phosphines was developed using [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> as a catalyst. Desired product structures were confirmed by single-crystal X-ray diffraction. Mechanistic experiments indicated that  $m$ -C<sub>Ar</sub>-H bond functionalization was a radical reaction and that a hexagonal ruthenacycle complex was a crucial intermediate in the process. Therefore, this study provides a novel method for the late-stage *meta*-position modification of biphenyl monophosphine ligands.



Tertiary phosphines (PAr<sub>3</sub>) are important molecular skeletons widely found in agricultural chemicals, biologically active molecules, and functional materials, and as ligands in transition-metal catalysts.<sup>1</sup> Their characteristics, including physicochemical, steric, and electronic properties, and biological and catalytic activities, are significantly changed by different R group structures.<sup>2</sup> In particular, biphenyl monophosphines that are air-stable and have greater steric hindrance are often employed as highly efficient ligands in the transition-metal catalysis of diverse C-X bond construction (X = C, O, N, and others).<sup>3</sup> Consequently, the straightforward and efficient synthesis of biphenyl monophosphines has attracted considerable research interest. In recent years, transition-metal-catalyzed C-H bond functionalization has seen intensive development and extensive applications in organic synthesis owing to its atom and step efficiency compared with traditional functional group chemistry.<sup>4</sup> Initially, as the strong chelating properties of undisguised monophosphines readily poison the metal catalyst, P(V)=O was often employed as a directing group to realize transition-metal-catalyzed selective C<sub>Ar</sub>-H bond functionalization,<sup>5</sup> with subsequent reduction providing various biphenyl monophosphine derivatives. Recently, undisguised P(III) atom-directed transition-metal-catalyzed selective C<sub>Ar</sub>-H bond functionalization has been reported, providing a step-economical method for the late-stage modification of monophosphine ligands.<sup>6</sup> However, these modification and functionalization methods for biphenyl monophosphines are limited to the *ortho* position relative to the directing group (Scheme 1). Therefore, remote C<sub>Ar</sub>-H bond functionalization of biphenyl monophosphines remains challenging.

In recent years, remote C<sub>Ar</sub>-H bond functionalization has made important progress through several ingenious strategies, including using steric effects,<sup>7</sup> electronic effects,<sup>8</sup> U-shaped templates,<sup>9</sup> a norbornene relay,<sup>10</sup> traceless directing groups,<sup>11</sup> a copper-promoted four-ring intermediate,<sup>12</sup> and the *ortho/para*-

## Scheme 1. P=O or P-Assisted C<sub>Ar</sub>-H Bond Functionalization



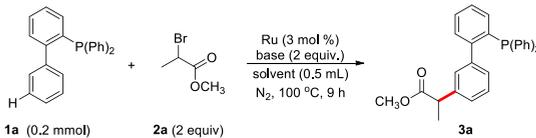
directing effect of the Ru-C<sub>Ar</sub> bond.<sup>13</sup> In these  $m$ -C<sub>Ar</sub>-H bond functionalization strategies, the ruthenium-catalyzed  $m$ -C<sub>Ar</sub>-H bond functionalization reaction type is diverse, the ruthenium catalyst is relatively inexpensive, and the reaction system is relatively simple. Furthermore, special substrate structures, large U-shaped directing groups, directing group removal, and the addition of an extra cocatalyst are not needed. However, ruthenium-catalyzed  $m$ -C<sub>Ar</sub>-H bond functionalizations have been limited to nitrogen-containing groups as directing groups. Using other atoms, such as phosphine and oxygen, to assist ruthenium-catalyzed  $m$ -C<sub>Ar</sub>-H bond functionalization remains challenging. Herein, we report P atom-directed  $m$ -C<sub>Ar</sub>-H bond functionalization catalyzed by ruthenium, which provides a

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new approach to the *meta*-position-modified synthesis of biphenyl monophosphine ligands.

To realize the phosphine-assisted modification and decoration of *m*-C<sub>Ar</sub>-H bonds, [1,1'-biphenyl]-2-ylidiphenylphosphane and methyl-2-bromopropanoate were selected as classic reactants, and [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> and K<sub>2</sub>CO<sub>3</sub> were used as the catalyst and base, respectively, to explore and optimize the C-H bond functionalization conditions. In a thick-walled pressure Schlenk tube under a N<sub>2</sub> atmosphere, the reaction was carried out at 100 °C for 9 h, as shown in Table 1. Pleasingly, a

Table 1. Screening of Reaction Conditions



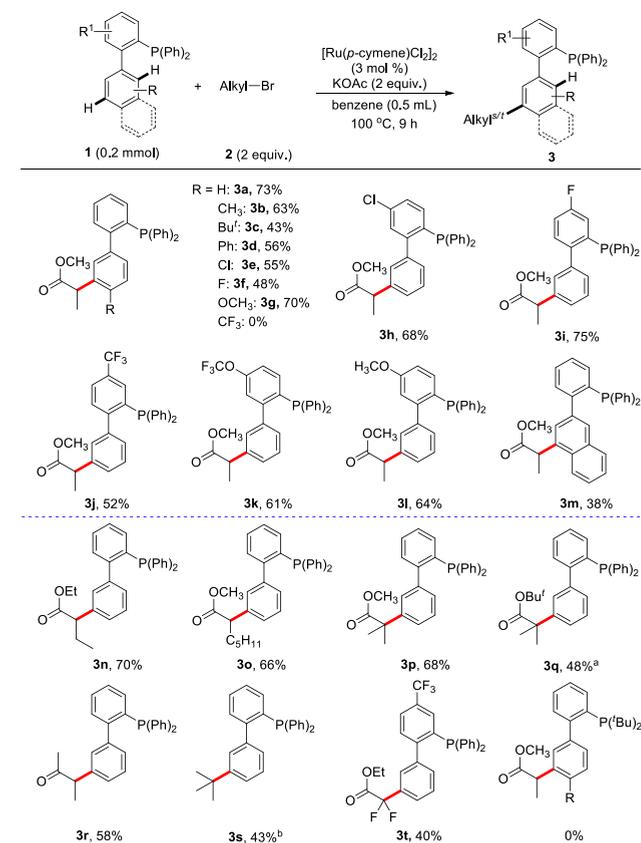
entry	Ru catalyst	base	solvent	yield (%)
1	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ] <sub>2</sub>	K <sub>2</sub> CO <sub>3</sub>	benzene	13
2	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ] <sub>2</sub>	KOAc	benzene	73
3	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	Cs <sub>2</sub> CO <sub>3</sub>	benzene	18
4	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	Na <sub>2</sub> CO <sub>3</sub>	benzene	25
5	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	CsOAc	benzene	56
6	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	NaOAc	benzene	70
7	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	K <sub>3</sub> PO <sub>4</sub>	benzene	30
8	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	KOAc	toluene	62
9	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	KOAc	xylene	45
10	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	KOAc	1,4-dioxane	22
11	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	KOAc	THF	28
12	[Ru( <i>p</i> -cymene)Cl <sub>2</sub> ]	KOAc	CH <sub>3</sub> CN	0
13	Ru( <i>p</i> -cymene)(OAc) <sub>2</sub>	KOAc	DMF	0
14	Ru(PPh <sub>3</sub> ) <sub>3</sub> Cl <sub>2</sub>	KOAc	benzene	42
15	RuCl <sub>3</sub>	KOAc	benzene	0
16	Ru <sub>3</sub> (CO) <sub>12</sub>	KOAc	benzene	0
17		KOAc	benzene	0

small amount of the desired product was obtained in the benzene solvent (entry 1). Base screening indicated that KOAc was the most suitable base for the transformation, providing desired product 3a in 73% isolated yield (entry 2). Other bases, such as Na<sub>2</sub>CO<sub>3</sub>, Cs<sub>2</sub>CO<sub>3</sub>, CsOAc, NaOAc, and K<sub>3</sub>PO<sub>4</sub>, were also effective in the process (entries 3–7, respectively), although the yields were lower than that obtained using KOAc. The target product was also obtained in other aromatic solvents, such as toluene, xylene, 1,4-dioxane, and THF (entries 8–11, respectively). The reaction did not proceed in CH<sub>3</sub>CN and DMF as solvents (entries 12 and 13, respectively). Further investigation indicated that Ru-(PPh<sub>3</sub>)<sub>3</sub>Cl<sub>2</sub> was also an effective catalyst for *m*-C<sub>Ar</sub>-H bond alkylation, giving the target product in a moderate yield (entry 14). Other ruthenium complexes, such as RuCl<sub>3</sub> and Ru<sub>3</sub>(CO)<sub>12</sub>, were ineffective in the process (entries 15 and 16, respectively). The desired product was not obtained when no catalyst was added to the system (entry 17). Furthermore, additional experiments indicated that the temperature and reaction time were critical factors in the transformation. Both prolonging the reaction time and increasing the reaction temperature resulted in product decomposition and reduced yields.

With optimized conditions in hand, we next investigated the scope and generality of this phosphine-assisted ruthenium-catalyzed *m*-C<sub>Ar</sub>-H bond alkylation of arenes with alkyl

bromides, as shown in Scheme 2. Initially, several [1,1'-biphenyl]-2-ylidiphenylphosphane molecules with various sub-

Scheme 2. Substrate Scope<sup>c</sup>



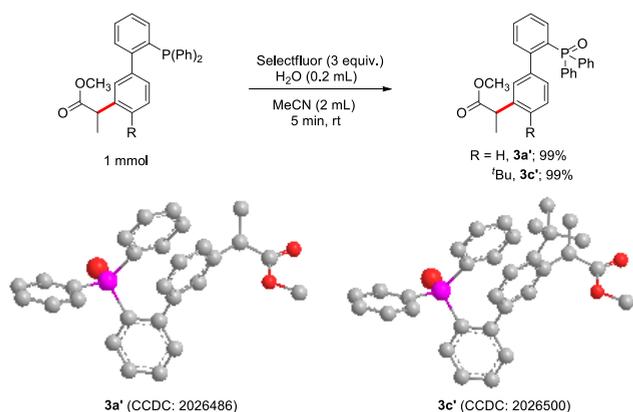
<sup>a</sup>For 3.5 h. <sup>b</sup>For 12 h. <sup>c</sup>Reaction conditions: 1 (0.2 mmol), 2 (2 equiv), [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> (3 mol %), KOAc (2 equiv), benzene (0.5 mL), N<sub>2</sub>, 100 °C, 9 h. Isolated yield.

stituents in different positions on the phenyl ring were selected as substrates. [1,1'-Biphenyl]-2-ylidiphenylphosphanes with different alkyl (3b and 3c) and aryl (3d) groups were suitable substrates for the transformation. Although the bulky *tert*-butyl group usually has a marked steric effect in many organic reactions, the desired product was afforded in a moderate yield under the optimized conditions (3c). Halogen substituents survived in the *m*-C<sub>Ar</sub>-H bond alkylation, offering the potential opportunity to construct complex functional molecules (3e, 3f, 3h, and 3i). Further experiments indicated that the electronic effect of the reactive phenyl ring was obvious in the transformation. An electron-donating substituent (-OCH<sub>3</sub>, 3g) was favorable for transformation, while an electron-withdrawing (-CF<sub>3</sub>) group completely impeded the reaction. The electronic effect of the directing group on the phenyl ring was not obvious in the process. Neither electron-withdrawing groups nor electron-donating groups had a significant effect on the reaction (3h–3l). Fluorine-containing groups with unique chemical, biological, and physical properties were also compatible with the *m*-C<sub>Ar</sub>-H bond functionalization (3i–3k). Notably, [2-(naphthalen-2-yl)phenyl]-diphenylphosphane was also tolerated in the process, providing the product in a moderate isolated yield (3m). Furthermore, various alkyl bromides were employed as alkylating reagents in the transformation. The results showed that both *sec*-alkyl and

*tert*-alkyl bromides were good coupling reagents in the phosphine-assisted  $m$ -C<sub>Ar</sub>-H bond alkylation (**3b**–**3s**). However, no alkylated products were found in the system when primary alkyl bromides, such as ethyl 2-bromoacetate and ethyl 4-bromobutanoate, were employed as alkylation reagents. To our delight, the alkylation conditions were also extended to phosphine-assisted  $m$ -C<sub>Ar</sub>-H bond difluoroalkylation (**3t**), providing a novel method for preparing *meta*-fluoroalkylated biphenyl monophosphine compounds. Unfortunately, when alkyl phosphine was used as the directing group, the desired product was not found in the system.

To further confirm the product structures obtained by phosphine-assisted  $m$ -C<sub>Ar</sub>-H bond alkylation, the liquid desired products were oxidized to solid phosphine oxides in a H<sub>2</sub>O/CH<sub>3</sub>CN solvent [1:10 (v/v)] at room temperature for 5 min using Selectfluor as an oxidant (Scheme 3).<sup>14</sup> The

### Scheme 3. Oxidation of Desired Products and Single-Crystal X-ray Structures

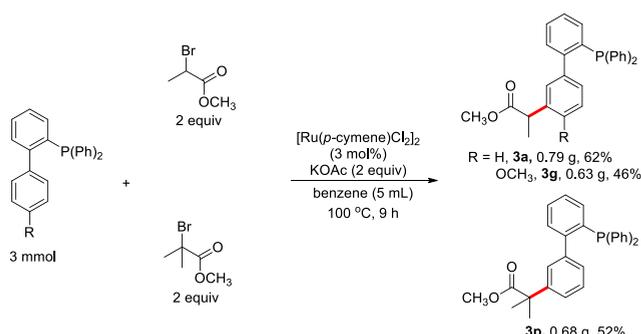


phosphine oxide single crystals were grown in a CH<sub>2</sub>Cl<sub>2</sub>/petroleum ether mixed solvent by slow evaporation at room temperature, and their crystal structures were determined by single-crystal X-ray diffraction.

In a gram-scale experiment, the phosphine-assisted ruthenium-catalyzed  $m$ -C<sub>Ar</sub>-H bond alkylation of arenes with alkyl bromides also proceeded smoothly, giving the target product in a moderate yield (Scheme 4). This development provides a novel and practical strategy for the late-stage *meta*-position modification of biphenyl monophosphine ligands.

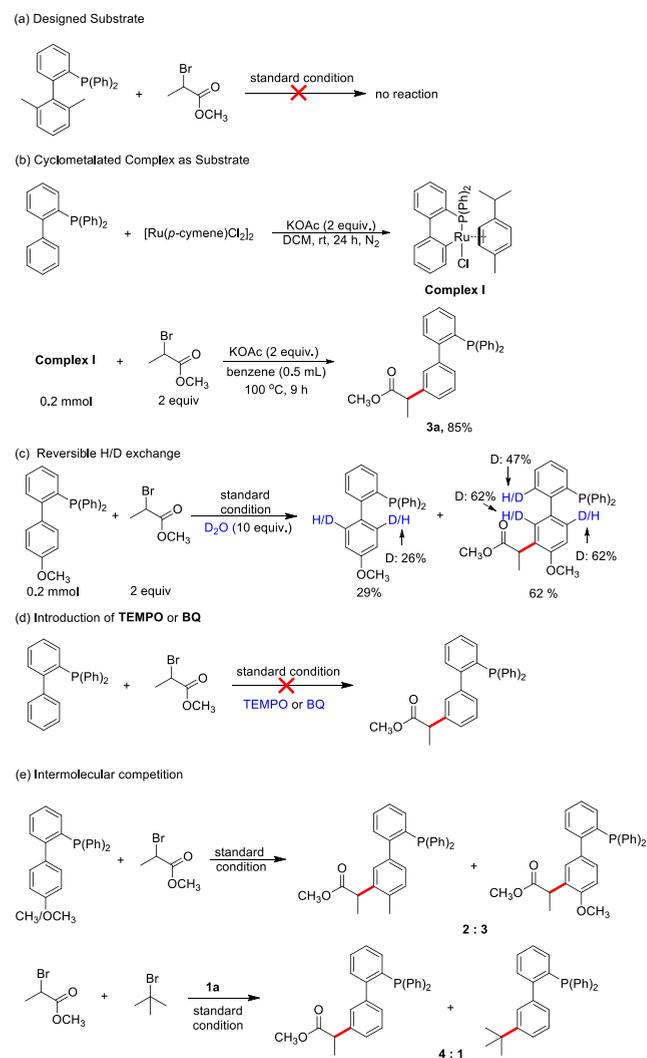
To clarify the mechanism of this ruthenium-catalyzed tertiary phosphine  $m$ -C<sub>Ar</sub>-H bond functionalization, some experiments were designed and conducted, as shown in

### Scheme 4. Gram-Scale Experiment



Scheme 5. First, when (2',6'-dimethyl-[1,1'-biphenyl]-2-yl)-diphenylphosphane, in which the two *ortho* sites of the tertiary

### Scheme 5. Designed Experiment

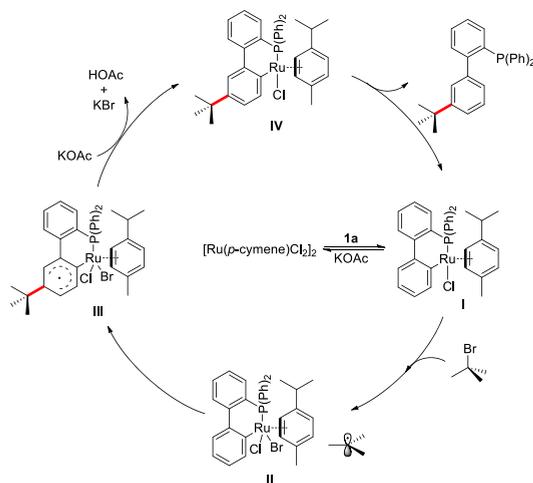


phosphine directing group were blocked by two methyl groups, was employed as the substrate, the desired product was not found in the system under the optimized conditions (Scheme 5a). The results showed that *o*-C<sub>Ar</sub>-H bond ruthenation was essential for  $m$ -C<sub>Ar</sub>-H bond functionalization. Second, hexagonal ruthenacycle complex I was synthesized by the reaction of [1,1'-biphenyl]-2-yl-diphenylphosphane with [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> in CH<sub>2</sub>Cl<sub>2</sub> at rt for 24 h (for details, see the Supporting Information).<sup>6c</sup> The desired product was obtained in excellent yield when complex I was used as the substrate, indicating that the hexagonal ruthenacycle was a crucial intermediate in the  $m$ -C<sub>Ar</sub>-H bond functionalization process (Scheme 5b). Third, obvious H/D exchange was observed in the residual substrate and desired product when 10 equiv of D<sub>2</sub>O was injected into the  $m$ -C<sub>Ar</sub>-H bond functionalization process. This phenomenon showed not only that *o*-C-H ruthenation was a reversible process but also that water was tolerated in the reaction system (Scheme 5c). Fourth, the desired product was not observed when free radical quenchers, such as TEMPO and BQ, were added to the reaction tube (Scheme 5d), showing that  $m$ -C<sub>Ar</sub>-H bond

functionalization might be a single-electron-transfer (SET) radical process. Finally, intermolecular competition experiments of two reactant analogues were conducted to evaluate the impact of electrons and structure on the transformation (Scheme 5e). The results showed that tertiary phosphines with higher electron density showed higher reactivity, and that  $\alpha$ -bromocarboxylates showed better reactivity than alkyl bromides.

On the basis of our experimental results and literature reports relating to ruthenium-catalyzed  $C_{Ar}$ -H bond functionalization,<sup>13</sup> a plausible mechanism for the  $m$ - $C_{Ar}$ -H bond alkylation process was proposed, as shown in Scheme 6. First,

### Scheme 6. Plausible Mechanism



reversible  $o$ - $C_{Ar}$ -H bond ruthenation of substrate **1a** by  $[Ru(p\text{-cymene})Cl_2]_2$  assisted by the phosphine directing group gives key hexagonal ruthenacycle(II) intermediate I. SET from ruthenacycle(II) intermediate I to the alkyl bromide generates an alkyl radical and ruthenium(III) complex II after bromide transfer. Then, the radical reacts with intermediate II at the directing group *meta* position, aided by the  $C_{Ar}$ -Ru *ortho/para*-directing effect, to provide active intermediate III. Deprotonation of intermediate III generates stable intermediate IV. Finally, ligand exchange of intermediate IV with **1a** gives the desired product and regenerates intermediate I for the next catalytic cycle.

In summary, we have developed a  $m$ - $C_{Ar}$ -H bond alkylation and fluoroalkylation of tertiary phosphines with alkyl bromides using  $[Ru(p\text{-cymene})Cl_2]_2$  as the catalyst. Mechanistic experiments indicated that the  $m$ - $C_{Ar}$ -H bond functionalization of tertiary phosphines was a radical process, and that a hexagonal ruthenacycle complex was a crucial reaction intermediate. This study provides a novel approach to accessing *meta*-position-modified biphenyl monophosphine ligands. Further investigations to develop new  $m$ - $C_{Ar}$ -H bond functionalization reactions of tertiary phosphines and apply this chemistry in synthesis are underway.

### ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c03377>.

Detailed experimental procedures and compound characterization data (PDF)

### Accession Codes

CCDC 2026486 and 2026500 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif), or by emailing [data\\_request@ccdc.cam.ac.uk](mailto:data_request@ccdc.cam.ac.uk), or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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### Notes

The authors declare no competing financial interest.

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