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Reaction of Polarised α-Ketoketene N,N-Acetals with Dimethyl Acetylenedicarboxylate: Synthesis of Novel 5-Aroyl-1-aryl-6-arylamino-4-methoxycarbonyl-2-oxo-1,2-dihydropyridines ¹

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The reaction of dimethyl acetylenedicarboxylate (2) with enamines^{2,3} and simple ketene S,N- and N,N-acetals^{4,5} derived from secondary amines is well documented. There is also one report³ on the reaction of an enaminoketone derived from pyrrolidine and ethyl 3-anilinocrotonate with 2. The products of these reactions are either the substituted butadienes formed by ring opening of the initial cyclobutene intermediates or those derived by further transformation of butadienes.

In continuation of our studies on the polarised α -ketoketene S.N- and N.N-acetals¹, a new class of vinylogous amides, we now report our results on the reaction of α -ketoketene S.N-

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1,3,4	Ar	R
а	\rightarrow	<u></u>
b	H₃CO- (_)-	<u></u>
c	CI-()	<u></u>
d	H3CO-	CI —
е	H ₃ CO	H ₃ C -
f	H ₃ C-{_}	н₃со-{_}
1,3 g	_	C ₂ H ₅

Scheme A

and N,N-acetals (1) with dimethyl acetylenedicarboxylate (2) to give the Michael adducts 3, as expected^{6,7}. These adducts are subsequently cyclised to the corresponding pyridones 4a-f in the presence of triethylamine in methanol (Scheme A).

Thus, when 1a was reacted with 2 at room temperature in benzene the open-chain Michael adduct 3a was obtained. The ketene N,N-acetals 1b-f similarly yielded the corresponding 3b-f (Table 1). When 3a was heated in methanol in the presence of triethylamine, it underwent smooth cyclisation to give 6-anilino-5-benzoyl-4-methoxycarbonyl-2-oxo-1-phenyl-1,2-dihydropyridine (4a). The structure of 4a was confirmed by spectral and analytical data (Table 2). The adduct 3g ($R = C_2H_5$) was unstable and the corresponding cyclic product 4g could not be obtained.

The S, N-acetals 1h and 1i similarly yielded the corresponding Michael adducts 3h and 3i on reaction with 2 (Table 1). The adducts 3h and 3i, although quite stable, failed to give the desired pyridones 4h, i in refluxing methanol and triethylamine (Scheme B).

Scheme B

Table 1. Michael Adducts 3a-i prepared

Prod- uct	Reaction time	Yield [%]	m.p. [°C]	Molecular formula ^a	I.R. (Nujol) v [cm ⁻¹]	$^{\text{t}}$ H-N.M.R. (CDCl ₃) δ [ppm]
3a	5 h	83	102-103°	C ₂₇ H ₂₄ N ₂ O ₅ (456.5)	3420-3340 (NH); 1740, 1720 (ester CO); 1660 (aryl-CO)	3.48 (br. s, 6H, COOCH ₃); 6.2-6.8 (m, 6H, 5H _{arom} + =CH-); 7.0-7.6 (m, 10 H _{arom}); 7.9-8.15 (br s, 2 H, NH)
3b	7 h	82	130°	$C_{28}H_{26}N_2O_6$ (486.5)	3430-3360 (NH); 1735, 1725 (ester CO); 1662 (aryl-CO)	3.50 (s, 3 H, COOCH ₃); 3.58 (s, 3 H, COOCH ₃); 3.80 (s, 3 H, OCH ₃); 6.2-7.7 (m, 15 H, $14 H_{arom} + = CH -)$
3c	15 h	80	104-105°	C ₂₇ H ₂₃ ClN ₂ O ₅ (490.9)	3440-3360 (NH); 1738, 1730 (ester CO); 1665 (aryl-CO)	3.58 (br. s, 6 H, COOCH ₃); 6.2-7.7 (m, 15 H, 14 H _{arom} + =CH-)
3d	10 h	78	155~157°	C ₂₈ H ₂₄ Cl ₂ N ₂ O ₆ (555.4)	3400-3340 (NH); 1745, 1730 (ester CO); 1665 (aryl-CO)	3.48 (s, 3 H, COOCH ₃); 3.60 (s, 3 H, COOCH ₃); 3.85 (s, 3 H, OCH ₃); 6.1-6.5 (m, 3 H, 2 H _{arom} + ==CH=); 6.5-6.9 (m, 4 H _{arom}); 7.1-7.7 (m, 6 H _{arom}); 7.82 (br. s, 2 H, NH)
3e	8 h	88	104~105°	$C_{30}H_{30}N_2O_6$ (514.6)	3380-3330 (NH); 1738, 1730 (ester CO); 1660 (aryl-CO)	2.10 (s, 3 H, CH ₃); 2.35 (s, 3 H, CH ₃); 3.50 (s, 3 H, COOCH ₃); 3.56 (s, 3 H, COOCH ₃); 3.81 (s, 3 H, OCH ₃); 6.4–6.8 (m, 7 H, 6 H _{arom} + =CH—); 7.2–7.4 (m, 6 H _{arom})
3f	10 h	85	48-51°	C ₃₀ H ₃₀ N ₂ O ₇ (530.6)	3440-3360 (NH); 1740, 1730 (ester CO); 1665 (aryl-CO)	2.32 (s, 3 H, CH ₃); 3.43 (s, 3 H, COOCH ₃); 3.50 (s, 3 H,
3g	15 min	90	95-97°	b	3300 (NH); 1718, 1692 (ester CO); 1620 (aryl-CO)	0.90-1.40 (2 br. t, 6 H, CH ₂ CH ₃); 3.40-3.70 (br. s, 6 H
3h	20 h	57	d	$C_{22}H_{21}NO_5S$ (411.5)	3450-3300 (NH); 1735, 1720 (ester CO); 1670 (aryl-CO)	1.85 (br. s, 3 H, SCH ₃); 3.38–3.91 (br. s, 6 H, COOCH ₃); 6.7–8.8 (m, 12 H, 9 H _{arom} + =CH-+ NH)
3i	15 h	59	d	C ₁₈ H ₂₁ NO ₅ S (363.4)	3380 (NH); 1740–1705 (ester CO); 1660 (aryl-CO)	1.02–1.32 (br. t, 3 H, CH ₂ CḤ ₃); 2.30 (br. s, 3 H, SCH ₃) 3.4–4.0 (m, 8 H, CḤ ₂ CH ₃ , COOCḤ ₃); 5.20 (br. s, 1 H NH); 6.9–7.7 (m, 6 H, 5 H _{arom} + =CH—)

^a Satisfactory microanalyses obtained: C ± 0.43 , H ± 0.23 , N ± 0.33 .

^b Unstable, not analysed.

c In trifluoroacetic acid.

d Viscous semi-solid.

Table 2. 5-Aroyl-1-aryl-6-arylamino-4-methoxycarbonyl-2-oxo-1,2-dihydropyridines 4a-f prepared

Prod- uct	Yield [%]	m.p. [°C]	Molecular formula ^a	M.S. m/e (M ⁺)	I.R. (Nujol) v [cm ⁻¹]	¹H-N.M.R. (CF ₃ COOH) δ [ppm]	
4a			$C_{26}H_{20}N_2O_4$ (424.5)	424	3340 (NH); 1740 (ester CO); 1680 (pyridone CO); 1648 (aryl-CO)	3.62 (s, 3 H, COOCH ₃); 6.5–7.2 (m, 6 H, 5 H _{aron} + 3-H); 7.2–7.8 (m, 10 H _{aron})	
4b	64	227°	$C_{27}H_{22}N_2O_5$ (454.5)	454	3330 (NH); 1738 (ester CO); 1680 (pyridone CO); 1633 (aryl-CO)	3.70 (s, 3 H, COOCH ₃); 3.95 (br. s, 3 H, OCH ₃); 6.7–7.4 (m, 7 H, 6 H _{arom} + 3-H); 7.5–8.0 (m. 8 H _{arom})	
4c	61	211-212°	C ₂₆ H ₁₉ ClN ₂ O ₄ (458.9)	460, 458	3320 (NH); 1735 (ester CO); 1685 (pyridone CO); 1632 (aryl-CO)	3.74 (s, 3 H, COOCH ₃); 6.5-7.3 (m, 5 H, 4 H _{arom} + 3-H); 7.4-8.0 (m, 10 H _{arom})	
4d	59	225°	$C_{27}H_{20}Cl_2N_2O_5$ (523.3)	_	3275 (NH); 1740 (ester CO); 1685 (pyridone CO); 1670 (aryl-CO) ^b	3.68 (s, 3 H, COOCH ₃); 3.94 (s, 3 H, OCH ₃); 6.4–7.9 (m, 13 H, 12 H _{arom} + 3-H)	
4e	65	189-191°	$C_{29}H_{26}N_2O_5$ (482.5)	482	3320 (NH); 1732 (ester CO); 1685 (pyridone CO); 1618 (aryl-CO)	1.50 (s, 3 H, CH ₃); 1.72 (s, 3 H, CH ₃); 3.05 (s, 3 H, COOCH ₃); 3.30 (s, 3 H, OCH ₃); 5.8-7.1 (m. 13 H, 12 H _{1170m} + 3-H)	
4f	67	143-144°	$C_{29}H_{26}N_2O_6$ (498.5)	498	3260 (NH); 1730 (ester CO); 1678 (pyridone CO); 1615 (aryl-CO) ^b	2.40 (s, 3 H, CH ₃); 3.31 (s, 3 H, COOCH ₃); 3.68 (s, 3 H, OCH ₃); 3.71 (s, 3 H, OCH ₃); 6.5–7.7 (m, 13 H, 12 H _{arom} + 3-H)	

Satisfactory microanalyses obtained: C ± 0.46 , H ± 0.26 , N ± 0.24 .

Michael Adducts 3; General Procedure:

A solution of the α -ketoketene N,N-acetal 1a-g (0.02 mol) or S,N-acetal 1h-i (0.02 mol) and dimethyl acetylenedicarboxylate (2; 0.026 mol) in dry benzene (80 ml) is stirred at room temperature for 0.25-20 h (Table; monitored by T.L.C., silica gel, 5% ethyl acetate/benzene). The solvent is removed and the crude adducts 3a-f are crystallised from ether/hexane as bright yellow solids. The adducts 3h and 3i are obtained as viscous semi-solids (T.L.C., silica gel, 5% ethyl acetate/benzene single spot) by column chromatography over a silica gel column using benzene/hexane (1:1) as eluent.

The adduct 3g is separated as a greenish-yellow solid after stirring for 15 min, which is filtered immediately and washed with hexane. Attempts to further purify 3g are not successful and on longer keeping in benzene solution, it is converted to a greenish-yellow, viscous semisolid, which showed several spots on T.L.C. analysis (silica gel, 35% ethyl acetate/benzene).

5-Aroyl-1-aryl-6-arylamino-4-methoxycarbonyl-2-oxo-1,2-dihydropyridines 4a-f; General Procedure:

A solution of the adduct 3a-f (0.01 mol) and triethylamine (16 ml) in absolute methanol (50 ml) is heated at $80-90\,^{\circ}\mathrm{C}$ with stirring for 6 h (checked by T.L.C., silica gel, 5% ethyl acetate/benzene). The solvent is removed under reduced pressure, the residue diluted with water (75 ml), and extracted with ethyl acetate (3×50 ml). The organic layer is dried with sodium sulphate and evaporated to give crude pyridones (4a-f) which are crystallised from chloroform/hexane (Table 2).

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b In KBr.

¹ Part XXIV of the series; Part XXIII: S. S. Bhattacharjee, C. V. Asokan, H. Ila, H. Junjappa, *Synthesis* **1982**, 1062.

² K. C. Brannock, R. D. Burpitt, V. W. Goodlett, J. G. Thweatt, J. Org. Chem. 28, 1464 (1963).

³ C. F. Huebner, L. Dorfman, M. M. Robison, E. Donoghue, W. G. Pierson, P. Strachan, J. Org. Chem. 28, 3134 (1963).

⁴ R. Gompper, Angew. Chem. 81, 348 (1969); Angew. Chem. Int. Ed. Engl. 8, 312 (1969).

⁵ K. C. Brannock, R. D. Burpitt, J. G. Thweatt, J. Org. Chem. 28, 1697 (1963).

⁶ V. Aggarwal, A. Kumar, H. Ila, H. Junjappa, Synthesis 1981, 157.

⁷ V. Aggarwal, H. Ila, H. Junjappa, Synthesis 1982, 65.