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8-METHOXYISOQUINOLINE DERIVATIVES THROUGH ortho-SELECTIVE METALATION OF 2-(3-METHOXYPHENYL)ETHYLAMINE

Manfred SCHLOSSER^{*} and Gyula SIMIG

Institut de Chimie organique de l'Université Rue de la Barre 2, CH-1005 Lausanne, Switzerland

<u>Summary</u>: Butyllithium in diethyl ether smoothly metalates 2-*m*-anisyl-*N*-pivaloylethylamine at the aromatic position flanked by the two substituents. Subsequent reaction with N,N-dimethylformamide followed by acid catalyzed cyclization and reduction gives 8-methoxy-1,2,3,4-tetrahydroisoquinoline.

As reported in the preceding communication ^[1], 2-*m*-anisyl-*N*-pivaloylethylamine reacts with *tert*-butyllithium at the benzylic position and, after carboxylation, affords 2-*m*-anisyl-3-*N*-pivaloylaminopropanoic acid (77%, mp 138 - 139 °C). With our "superbasic" mixed metal reagent ^[2], however, the hydrogen/metal exchange occurs at the aromatic position which is adjacent to the methoxy group but distant from the alkyl sidechain (58% of the acid after carboxylation, mp 166 - 167 °C, recryst. from ethyl acetate). Finally, ordinary butyllithium preferentially abstracts a proton from the aromatic position which is *ortho* with respect to both substituents (61% of the acid after carboxylation, mp 128 - 129 °C, recryst. from ethyl acetate).



This reagent controlled site selectivity of the organometallic attack can be advantageously exploited in a simple synthesis of isoquinoline derivatives. If the intermediate resulting from metalation with butyllithium is treated with N,N-dimethylformamide, the aldehyde 1 is obtained which in acidic medium undergoes rapid cyclization. Simultaneous loss of the transient hydroxy function and the pivaloyl moiety led to 8-methoxy-3,4-dihydroiso-quinoline hydrochloride (2·HCl)^[3].



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The heterocyclic compound 2 is a versatile intermediate for further transformations in the heterocyclic field. For example, it may be alkylated with methyl iodide affording 8-methoxy-2-methyl-3,4-dihydroisoquinolinium iodide (3, 71%, mp 185 - 186 °C, recryst. from ethanol) or reduced with sodium borohydride to give 8-methoxy-1,2,3,4-tetrahydroisoquinoline (4).



Until now, the preparation of 8-methoxy-1,2,3,4-tetrahydroisoquinoline (4) has required lengthy and inefficient routes ^[4]. The Bischler-Napieralski type cyclization of N-formyl-2-m-anisylethylamine leads to 6- rather than 8-methoxy-3,4-dihydroisoquinoline ^[5].

Working procedure : 2-m-Anisyl-N-pivaloylethylamine ^[6] (10 mmol) was added to a solution of butyllithium (30 mmol) in hexane (20 mL) and diethyl ether (50 mL). After stirring the suspension 2 h at 25 °C, it was treated with dimethylformamide (60 mmol). After 1 h at 25 °C, a concentrated aqueous solution of ammonium chloride (20 mL) was added. The aldehyde 1 was extracted and purified by elution from silica gel with mixtures of ethyl acetate and hexane; 55%; mp 91 - 92 °C (recryst. from heptane). ^[7] - A solution of aldehyde 1 (5 mmol) in dichloromethane (10 mL) and 10% aqueous hydrochloric acid (20 mL) were vigorously stirred during 24 h at 25 °C. After evaporation of the aqueous phase to dryness, the 8-methoxy-3,4-dihydroisoquinoline hydrochloride demihydrate (2·HCl+ $\frac{1}{2}$ H_O) remained as a colorless residue; 79%; mp 183 - 184 °C (dec.; recryst. from ethanol/diethyl ether)^[7]. - The latter substance (3 mmol) was dissolved in icc-cold methanol (10 mL) which contained sodium borohydride (3 mmol). After 30 min at 0 °C, the solvent was evaporated, a 10% aqueous solution (10 mL) of sodium hydroxide was added and the 8-methoxy-1,2,3,4-tetrahydroisoquinoline (4) was extracted with dichloromethane (3 × 10 mL). This solvent was subsequently replaced by dicthyl ether (10 mL). Anhydrous hydrochloric acid caused the precipitation of the 4 · HCl salt; 69%; mp 261 - 262 °C (dec.). ^[7, 8]

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