One-Step Multicomponent Self-Assembly of a First-Generation Sierpiński Triangle: From Fractal Design to Chemical Reality**

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Abstract: A novel terpyridine-based architecture that mimics a first-generation Sierpiński triangle has been synthesized by multicomponent assembly and features $tpy-Cd^{II}$ -tpy connectivity (tpy = terpyridine). The key terpyridine ligands were synthesized by the Suzuki cross-coupling reaction. Mixing two different terpyridine-based ligands and Cd^{II} in a precise stoichiometric ratio (1:1:3) produced the desired fractal architecture in near-quantitative yield. Characterization was accomplished by NMR spectroscopy, mass spectrometry, and transmission electron microscopy.

N aturally occurring species often self-assemble into fractal shapes and forms.^[1,2] In 1915, the Polish mathematician Waclaw Sierpiński formulated an equilateral triangular fractal, which was constructed by connecting smaller equilateral triangles^[3] (Figure 1). Later, Mandlebrot described these triangular fractals as Sierpiński gaskets (triangles or sieves).^[1,4] The number of triangles in a Sierpiński triangle can be found by using the formula $N_n = 3^n$, where N is the number of triangles and n the number of iterations.^[5]

Simple fractal motifs have been successfully integrated into synthetic organic entities over the past forty years in the form of dendrimers.^[6–9] There is a unique relationship between the 1 \rightarrow 3 branched dendrimers^[6] and the Sierpiński triangle, which is presented in Figure 1.^[10,11] Recent advances in supramolecular chemistry^[12] have paved the way to a better understanding of how to quantitatively assemble various simple architectures in one step,^[13–22] including the desired triangular predecessor constructs.^[13,14]

Coordination-driven self-assembly using tailored 2,2':6',2''-terpyridine (tpy) building blocks has highlighted their ability to form stable linear tpy- M^{II} -tpy complexes as

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Figure 1. Conceptual progression of a $1 \rightarrow 3$ dendritic branching pattern and its geometric relationship to the classical Sierpiński triangle.^[10]

the structural walls of the polygons possessing rigid organic vertices. Therefore, programmed structural features are easily incorporated into the poly(terpyridine) building blocks, generating avenues to novel nanomolecular constructs^[23] in quantitative or near-quantitative yields.^[24] Both homoleptic and heteroleptic connectivity has been used extensively in these assemblies, but heteroleptic assembly with a need for multicomponent building blocks still remains challenging owing to a tendency to produce competitive products.^[25-27] Recently, we have reported various heteroleptic materials, such as a hexagonal spoked wheel,^[28] molecular bowties and butterflies,^[29] and a molecular rhombus,^[30] which were obtained by either multicomponent or one-step assembly. Herein, we report the design and synthesis of a first-generation molecular Sierpiński triangle using two unique ligands with tpy–Cd^{II}–tpy connectivity (Figure 2).

Retrosynthetic analysis of the first-generation Sierpiński triangle (Figure 1) revealed that it would require two easily accessible components for its assembly: a tetrakis(terpyridine) "K" unit for the walls and the core region and a ditopic 60°-directed "V" unit for the vertices. Cadmium was chosen owing to a unique combination of thermodynamic stability and kinetic lability for its complexes. A 1:1 ratio of these "K" and "V" ligands was coupled with precisely three equivalents of Cd^{II}; the overall stability of the highly symmetric polytriangular architecture was projected to be favored over any oligomeric possibilities. In essence, there would be no loose ends or uncoordinated ligands.

Initially, a THF solution of 2,3,5,6-tetrabromocatechol and iodomethane (Scheme 1) was heated at reflux in an inert atmosphere (N₂, K₂CO₃, 12 h) to generate **3**. Boronic acid $\mathbf{1}^{[31]}$ was prepared using commercially available 4-formylphenylboronic acid. The desired "K" ligand **4** was easily synthesized (72%) from **3** by treatment with **1** using standard Suzuki cross-coupling reaction conditions (K₂CO₃, [Pd(PPh₃)₂Cl₂])

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Figure 2. Terpyridine-based Sierpiński triangle 5.



Scheme 1. Synthesis of the key terpyridine building blocks 2 ("V") and 4 ("K").

under an atmosphere of $\operatorname{argon}^{[29]}$ Its ¹H NMR spectrum showed the characteristic peak at 3.76 ppm for the installed OCH₃ marker and the presence of two completely different arms (**a** and **b**, Scheme 1). The ¹³C NMR spectrum showed a peak at 60.76 ppm, which corresponds to the OCH₃ substituents.

All of the proton peaks were assigned using COSY and NOESY NMR spectroscopy. The ¹³C NMR assignments and the mass-spectrometric data are in full agreement with the ligand structures. The colorless "V" ligand **2** was also synthesized (80%) by a slightly modified procedure^[29] from commercially available 4,5-dibromo-1,2-dimethoxybenzene under Suzuki cross-coupling conditions. It was easily characterized by ¹H NMR spectroscopy and mass spectrometry. The aromatic region of ligand **2** showed one set of characteristic terpyridine peaks and one set of aromatic phenyl-spacer

peaks; a sharp singlet was observed at 4.02 ppm, which corresponds to the OCH₃ markers. The ¹³C NMR spectrum displayed a peak at 56.40 ppm for the OCH₃ group also.

The facile one-step assembly of Sierpiński triangle **5** (Scheme 1) utilized an exactly 1:1 solution of **2** and **4** in CHCl₃ to which a methanolic solution of Cd(NO₃)₂·4H₂O (3 equiv) was added. The solution was stirred for 30 minutes at 25 °C; then, excess NH₄PF₆ was added to effect counterion exchange to PF₆⁻. The desired PF₆ complex precipitated and was filtered and repeatedly washed with MeOH to remove excess NH₄PF₆. Complex **5** was obtained without any further purification as a light yellow solid in >95 % yield.

The Sierpiński triangle 5 was completely characterized by ¹H, COSY, and NOESY NMR spectroscopy and ESI and traveling-wave ion mobility (TWIM) mass spectrometry, along with transmission electron microscopy (TEM). The ¹H NMR spectrum of **5** showed the characteristic features of a tpy-Cd^{II}-tpy complex, with a sharp and simple pattern being indicative of a discrete structure with a high degree of inherent structural symmetry. The structural simplicity and symmetry were reflected by the ¹H NMR spectrum. The imbedded methoxy markers within each ligand appeared in the product as two distinct singlets at 3.98 ppm (from "V") and 3.87 ppm (from "K") with a precise 1:1 ratio, initially supporting the formation of a product that possesses D_{3h} symmetry. Notably, peaks that would be expected for polymeric structures and impurities were not observed. The aromatic region exhibited the peak ratio expected for three different sets of 3',5'-tpy protons for arms **a**, **b**, and **c** (8.92, 8.91, and 8.74 ppm, respectively; Figure 2 and Figure 3); the downfield shifts resulted from deshielding upon complexation.^[28] The 6,6"-tpy protons were noticeably shifted upfield (7.92, 7.68, and 7,80 ppm, respectively, for **a**, **b**, and **c**) when compared with the free ligands,^[23] as expected. All of the peaks in the ¹H NMR spectrum were assigned and the assignments verified with the aid of 2D-COSY and 2D-NOESY experiments; they are in complete agreement with the proposed structure.

The ¹³C NMR spectrum of complex **5** exhibits three signals for the different and readily identifiable 3',5'-tpy



Figure 3. Stacked ¹H NMR spectra (500 MHz) of ligands 2 (bottom) and 4 (top) in CDCl₃ and of complex 5 in CD₃CN (0.25 mgmL⁻¹; center). Arrows indicate assigned resonance shifts that occur upon complex formation.

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carbon atoms at 158.39, 155.32, and 154.93 ppm, respectively, along with two very close, yet distinct peaks for the OCH_3 substituents at 56.02 and 56.00 ppm, which further supports the proposed structure (Supporting Information, Figure S12).

Sierpiński triangle **5** was further characterized by ESI–MS coupled with TWIM mass spectrometry.^[22–24, 32–36] In ESI–MS (Figure 4), a series of dominant peaks were generated at



Figure 4. A) ESI-MS and B) 2D ESI-TWIM MS plot of 5. The charge states of intact assemblies are marked.

m/z values of 763.4, 854.2, 965.2, 1103.9, 1282.4, and 1520.3, which correspond to charge states from 11 + to 6 + from the loss of various numbers of PF₆⁻ anions. The isotope pattern of each peak is in agreement with the corresponding simulated isotope pattern. Additional evidence for formation of the Sierpiński triangle **5** was provided by ESI–TWIM MS (Figure 4); the corresponding spectrum showed a set of single and narrow bands for charge states from 11 + to 6+, which is in agreement with the presence of one single structure **5**.

The structure of complex **5** was further confirmed by comparison of the experimental collision cross sections (CCSs) of the charge states 11 + to 6 +, which were deduced from their drift times measured by ESI–TWIM MS, with the theoretical CCSs of the complex without any counterions (100 energy-minimized structures obtained by molecular modeling). Experimental and theoretical CCSs of triangle **5** are listed in Table 1.

TEM afforded the visualization of the triangle 5, which revealed a direct correlation of both the size and shape of

Table 1: Drift times and collision cross sections for the Sierpiński triangle **5**.

	D (6 () 1 1	CCC (¹ / ₂)	CCC (Å2)
Charge	Drift time [ms]	CCS _{exp} [A ²]	CCS _{theory} [A ²]
6+	8.93	1309.2	1388.3
7+	6.49	1295.5	
8+	5.01	1288.7	
9+	4.11	1297.2	
10+	3.40	1288.5	
11+	2.76	1242.3	
average of all charge states	1286.9		

single molecules upon deposition of a dilute solution (ca. 10^{-5} M) of complex **5** in MeCN with PF₆⁻ counterions on carbon-coated copper grids (300 mesh). The molecular framework was observed as a uniform dispersion of individual molecules with triangular shape, clear edges, and discernable vertices. The average distance (5.6 nm) between the vertices correlates well with the size obtained from the optimized molecular model (Figure 5).



Figure 5. Low-magnification TEM image of the Sierpiński triangle **5** showing a uniform field of particles. The high-magnification TEM image clearly exhibits triangular motifs and a slightly larger and rounded picture of a proposed aggregate (where the two triangles illustrate a potential arrangement). Computer-generated models illustrate the different OCH₃ (singlet markers) environments that were observed in ¹H NMR dilution experiments to ascertain the presence of individual versus stacked species.

The TEM image (Figure 5) also suggests that stacking or aggregation occur at higher concentration, where two Sierpiński triangles lie atop each other to generate a "Star of David"-like motif. This phenomenon was supported by ¹H NMR experiments, where at higher concentrations, a small extra peak was observed as a shoulder to the peak assigned to the OCH₃ markers of ligand "K"; thus, the NMR spectrum is a combination of stacked (two or more) and individual species. At lower concentrations, a single peak was observed for these markers. The "V" ligand marker signals are unaffected by the concentration (see the Supporting Information for the details of the NMR dilution experiment, Figure S17).

In conclusion, we have achieved the self-assembly of a first-generation Sierpiński triangle with D_{3h} symmetry using

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tpy–Cd^{II}–tpy connectivities in near quantitative yield through the use of structurally directed, multi-topic 2,2':6',2''-terpyridine ligands possessing programmed structural features. NMR spectroscopy, ESI and TWIM mass spectrometry, and TEM provided great insight into the size, shape, symmetry, and molecular structure of the assembled product. The application of ligands with complementary directivity to the assembly of higher-generation structures is currently underway.

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- B. B. Mandlebrot, *The Fractal Geometry of Nature*, Freeman, San Francisco, CA, **1982**.
- [2] L. F. Richardson, Proc. R. Soc. London Ser. A 1926, 110, 709– 737.
- [3] W. Sierpiński, C. R. Hebd. Seances Acad. Sci. 1915, 160, 302– 305.
- [4] A. Fraenkel, A. Kontorovich, Proc. Integers Conference 2007, 209–227.
- [5] J. M. Borwein, D. H. Bailey, Mathematics by Experiment: Plausible Reasoning in the 21st Century, A. K. Peters, Ltd, Welsley, MA, 2003.
- [6] a) G. R. Newkome, Z. Yao, G. R. Baker, V. K. Gupta, J. Org. Chem. 1985, 50, 2003–2004; b) G. R. Newkome, C. D. Shreiner, Chem. Rev. 2010, 110, 6338–6442.
- [7] C. Ornelas, E. Boisselier, D. Astruc, Chem. Rev. 2010, 110, 1857– 1959.
- [8] K.-I. Sugiura, H. Tanaka, T. Matsumoto, T. Kawai, Y. Sakata, Chem. Lett. 1999, 1193.
- [9] G. R. Newkome, P. Wang, C. N. Moorefield, T. J. Cho, P. P. Mohapatra, S. Li, S.-H. Hwang, O. Lukoyanova, L. Echegoyen, J. A. Palagallo, V. Iancu, S.-W. Hla, *Science* 2006, *312*, 1782– 1785.
- [10] Figure 1 derived from: H.-O. Peitgen, H. Jürgens, D. Saupe, *Chaos and Fractals, New Frontiers of Science*, Springer, New York, **1992**.
- [11] H.-O. Peitgen, personal assistance, 2011.
- [12] J.-M. Lehn, Supramolecular Chemistry: Concepts and Perspectives, VCH, Weinheim, 1995.
- [13] S.-H. Hwang, C. N. Moorefield, F. R. Fronczek, O. Lukoyanova, L. Echegoyen, G. R. Newkome, *Chem. Commun.* 2005, 713– 715.

- [14] M. K. Saha, S. Neogi, M. Schmittel, *Dalton Trans.* **2014**, *43*, 3815–3834.
- [15] R. D. Sommer, A. L. Rheingold, A. J. Ghose, B. Bosnich, J. Am. Chem. Soc. 2001, 123, 3940–3952.
- [16] E. C. Constable, B. A. Hermann, C. E. Housecroft, M. Neuburger, S. Schaffner, L. J. Scherer, New J. Chem. 2005, 29, 1475– 1481.
- [17] S.-H. Hwang, P. Wang, C. N. Moorefield, L. A. Godínez, J. Manríquez, E. Bustos, G. R. Newkome, *Chem. Commun.* 2005, 4672–4674.
- [18] G. R. Newkome, T. J. Cho, C. N. Moorefield, G. R. Baker, R. Cush, P. S. Russo, *Angew. Chem. Int. Ed.* **1999**, *38*, 3717–3721; *Angew. Chem.* **1999**, *111*, 3899–3903.
- [19] V. Stepanenko, F. Würthner, Small 2008, 4, 2158–2161.
- [20] S. Perera, X. Li, M. Guo, C. Wesdemiotis, C. N. Moorefield, G. R. Newkome, *Chem. Commun.* **2011**, 47, 4658–4660.
- [21] Y.-T. Chan, X. Li, J. Yu, G. A. Carri, C. N. Moorefield, C. Wesdemiotis, G. R. Newkome, J. Am. Chem. Soc. 2011, 133, 11967–11976.
- [22] Y.-T. Chan, X. Li, C. N. Moorefield, C. Wesdemiotis, G. R. Newkome, *Chem. Eur. J.* 2011, 17, 7750–7754.
- [23] U. S. Schubert, A. Winter, G. R. Newkome, *Terpyridine-based Materials*, Wiley-VCH, Weinheim, **2011**.
- [24] Y.-T. Chan, X. Li, M. Soler, J.-L. Wang, C. Wesdemiotis, G. R. Newkome, J. Am. Chem. Soc. 2009, 131, 16395–16397.
- [25] P. R. Andres, U. S. Schubert, Macromol. Rapid Commun. 2004, 25, 1371–1375.
- [26] E. J. Constable, K. Harris, C. E. Housecroft, M. Neuburger, *Dalton Trans.* 2011, 40, 1524–1534.
- [27] D. J. Cárdenas, J.-P. Sauvage, Inorg. Chem. 1997, 36, 2777-2783.
- [28] J.-L. Wang, X. Li, X. Lu, I. F. Hsieh, Y. Cao, C. N. Moorefield, C. Wesdemiotis, S. Z. D. Cheng, G. R. Newkome, *J. Am. Chem. Soc.* 2011, *133*, 11450–11453.
- [29] A. Schultz, X. Li, B. Barkakaty, C. N. Moorefield, C. Wesdemiotis, G. R. Newkome, J. Am. Chem. Soc. 2012, 134, 7672– 7675.
- [30] X. Lu, X. Li, J.-L. Wang, C. N. Moorefield, C. Wesdemiotis, G. R. Newkome, *Chem. Commun.* 2012, 48, 9873–9875.
- [31] P. Jarozs, K. Lotito, J. Schneider, D. Kumaresan, R. Schmehl, R. Eisenberg, *Inorg. Chem.* 2009, 48, 2420–2428.
- [32] G. von Helden, M.-T. Hsu, P. R. Kemper, M. T. Bowers, J. Chem. Phys. 1991, 95, 3835–3837.
- [33] M. T. Bowers, P. R. Kemper, G. von Helden, P. A. M. van Koppen, *Science* **1993**, 260, 1446–1451.
- [34] C. S. Hoaglund-Hyzer, A. E. Counterman, D. E. Clemmer, *Chem. Rev.* **1999**, *99*, 3037–3080.
- [35] S. Trimpin, D. I. Plasencia, D. E. Clemmer, Anal. Chem. 2007, 79, 7965–7974.
- [36] G. R. Hilton, A. T. Jackson, K. Thalassions, J. H. Scrivens, Anal. Chem. 2008, 80, 9720–9725.

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Sierpiński Triangles

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One-Step Multicomponent Self-Assembly of a First-Generation Sierpiński Triangle: From Fractal Design to Chemical Reality



Mathematical mimicry leads to the synthesis of a Sierpiński triangle, which was obtained by multicomponent assembly and features terpyridine–Cd^{II}–terpyridine connectivity. Complementary ligand architecture and metal complex lability act in synergy to achieve the desired outcome.

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