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Ablajan Keyume, Zeynepgul Esmayil, Liju Wang, Feng Jun



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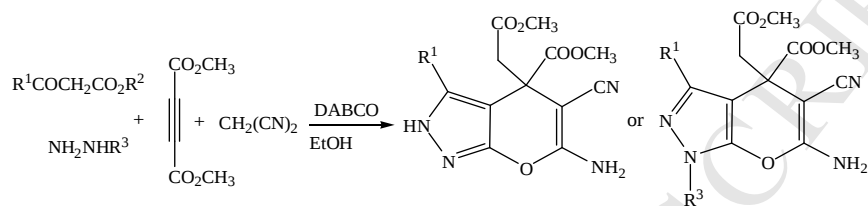
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Graphical Abstract

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Ablajan Keyume*, Zeynepgul Esmayil, Liju Wang, Feng Jun

Key Laboratory of Oil & Gas Fine Chemicals, Ministry of Education & Xinjiang Uyghur Autonomous Region, College of Chemistry and Chemical Engineering, Xinjiang University, Urumqi 830046, P. R. China

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ABSTRACT

We developed an efficient and simple one-pot synthesis of functionalized multi-substituted 2,4-dihydro-pyrano[2,3-c]pyrazole dicarboxylates from β -ketoesters, hydrazine, dimethyl acetylene dicarboxylate and malononitrile in EtOH. This four-component one-pot reaction carried out in the presence of DABCO catalyst showed advantages over a one-pot three-component method in its simple procedure, high yield and low toxicity.

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1. Introduction

Pyrazole and pyran are both cyclic structural units found in many different natural and synthetic products that exhibit a wide range of biological activities.¹ Pyran-containing compounds exhibit analgesic, anticoagulant and anticancer effects.² 4H-Chromenes contain skeletons similar to pyran. They are widespread in nature and are pharmacologically potent, being employed as drugs, pigments and in cosmetics.³ Many types of bioactive drugs with broad medicinal and agrochemical applications contain both pyran and pyrazole rings,⁴ and both structures are present in pyrano[2,3-c]pyrazole and its derivatives, which exhibit various biological activities and pharmacological properties.⁵ Therefore, much research has focused on the synthesis and bioactivities of compounds containing these structural units.

One-pot, multicomponent reactions (MCRs) constitute a powerful class of synthetic methods owing to their short reaction times, high efficiencies and unique selectivities.⁶ They can form several chemical bonds simultaneously.⁷ The carbon-carbon triple bond is one of the most important structural units in the formation of different rings via its reaction with different active substances. Acetylenedicarboxylate has recently been extensively used in the one-pot formation of various heterocyclic systems.⁸ The pharmaceutical and biological importance of multi-substituted dihydropyrano[2,3-c]pyrazoles has led to the investigation of many synthetic approaches.⁹ The triple carbon-

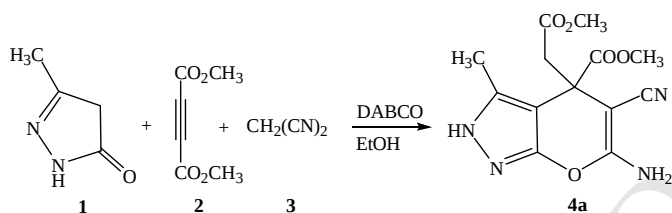
carbon bond can undertake various active roles in organic reactions for the facile and efficient production of various heterocycles.¹⁰ For example, the reaction of alkynes with O-acetyl- or O-benzoyl-substituted anilines yields the corresponding quinolines.¹¹ Highly functionalized 1,4-pyranonaphthoquinone has been prepared from the reaction of aniline, diethylacetoacetylene, hydroxyl-naphthalene-1,4-dione and an aldehyde under microwave irradiation without the use of a solvent or catalyst.¹² An atom-economic method for the one-pot synthesis of substituted 4H-chromene, 4H-pyran and oxepine derivatives has been developed using a reaction involving three components: 1,3-diketones, diethyl acetylenedicarboxylate and malononitrile.¹³ Multicomponent domino reactions in aqueous media catalyzed by L-proline have recently been developed for the synthesis of functionalized 4H-pyrano[2,3-c]pyrazoles.¹⁴ Variations of, or an increase in the number of the various functional groups in the pyran or pyrazole ring may modify the biological properties of the resulting compounds.

Pyrazol-5-one, malononitrile, β -ketoesters and hydrazine are important intermediates that are widely used in the synthesis of functionalized and novel organic compounds.¹⁵ Many recent studies have reported the one-pot synthesis of pyrano[2,3-c]pyrazoles in the presence of different catalysts.¹⁶ Acetylene dicarboxylate has been little used in the synthesis of novel bioactive compounds, but it is considered here in the important task of developing a direct, convenient, economic and efficient

multicomponent one-pot method for the synthesis of novel pyrano[2,3-c]pyrazole compounds.

2. Results and discussion

To establish the optimal reaction conditions for the synthesis of multi-substituted pyrano[2,3-c]pyrazol dicarboxylates, our initial experiments were focused on a one-pot reaction involving pyrazol-5-one, malononitrile and dimethyl acetylenedicarboxylate. For a low risk and eco-friendly process, both EtOH and water are used as the preferred solvents for all of the optimization experiments. All of the reactions were performed via heating at 50 °C for the appropriate time period. Compound **4a** was obtained in low to moderate yields in the presence of DBU, DDC, KOH, CH₃COONa, HOAc, L-proline or piperidine as the catalyst. Interestingly, **4a** was obtained in a high yield using DABCO (20 % mol) as a catalyst in EtOH rather than in water (Table 1). The yield decreased when the reaction proceeded at room temperature or over 50 °C. According to experimental results, the protocol described in Scheme 1 is applicable to a different type of pyrazole-5-ones. The detailed experimental results obtained with this three-component one-pot reaction were listed in Table 2. Based on these results, a reaction stirring in EtOH with 20 % mol DABCO catalyst was selected as the optimal condition for the three-component synthesis of other compounds.



Scheme 1. Three-component one-pot reaction for the synthesis of compound **4a**.

Table 1. Optimization of reaction conditions involving catalyst and solvent screening for the synthesis of **4a**.

Entry	catalyst (20 mol %)	solvent	t (h)	yield of 4a (%)	
				A ^a	B ^b
1	Piperidine	EtOH	2 h	50	60
2	Pyrrolidine	EtOH	2 h	45	30
3	KOH	EtOH	2 h	20	25
4	CH ₃ COONa	EtOH	2 h	25	30
5	L-proline	EtOH	2 h	65	70
6	DBU	EtOH	2 h	35	45
7	DDC	EtOH	2 h	20	25
8	DABCO	CH ₂ Cl ₂	1 h	35	45
9	DABCO	MeCN	1 h	50	65
10	DABCO	MeOH	1 h	65	72
11	DABCO	H ₂ O	1 h	50	55
12	DABCO	EtOH	1 h	80	88

^a Isolated yields for the three-component one-pot reaction.

^b Isolated yields for the four-component one-pot reaction.

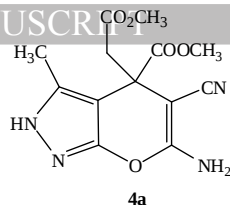
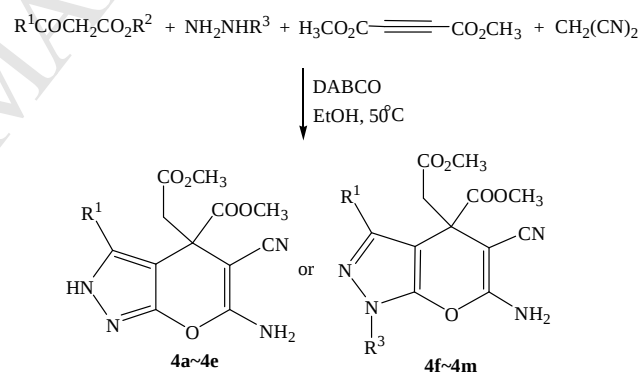


Fig. 1 Structure of compound **4a**.

In comparison to the three-component one-pot reaction, for the four-component reaction, four reactants, including β-ketoesters, hydrazine, dimethyl acetylenedicarboxylate and malononitrile underwent a reaction to yield the target compound without preparing pyrazol-5-one. Similar to the above optimization, the catalysts, solvents and temperature used in the four-component reaction for the synthesis of compound **4a** were also investigated to optimize the reaction conditions (Table 1). It is important to note that when the reaction was carried out in EtOH using DABCO as the catalyst, the yield of product **4a** reached 88 %. However, the yield of compound **4a** decreased when the reaction was attempted in water using the same catalyst. In addition, some by-products mixed with **4a** were formed when water or EtOH was used as the solvent at room temperature or under refluxing conditions in the presence of the DABCO catalyst.



Scheme 2. Four-component one-pot reaction for the synthesis of compounds **4a-m**.

Next, all of the compounds were prepared via the four-component one-pot reaction due to the short reaction time, simple procedure and atom efficiency of this method. Although the four-component reaction appears simple and convenient, the product was not formed when the four reactants were mixed together, despite using a similar method to the three-component one-pot reaction. Therefore, the dropwise addition of a mixture of acetylene diacetate and malononitrile into a mixture of methyl/ethyl acetoacetate and hydrazine in ethanol under stirring and heating at 50 °C to produce the final product in high yield.

After having established the optimal reaction conditions, the effect of the type of β-ketoesters and hydrazine was examined. The reaction yield and reaction time indicated that hydrazine hydrate increased the reaction rate more than phenyl hydrazine or substituted phenyl hydrazine (Table 2). While phenyl hydrazine or substituted phenyl hydrazine participate in the reaction, the completion of the reaction required more time, and the yield of product decreased. The results listed in Table 2 also

indicated that 3-oxo aliphatic acid methyl ester afforded slightly higher yields compared to methyl benzoyl acetate. Methyl β -ketoesters resulted in slightly higher yields than that of ethyl β -ketoesters (not shown). The results obtained in the three- and four-component reactions were summarized in Table 2.

Table 2. One-pot synthesis of dihydropyrano[2,3-c]pyrazoles.

Entry	Product	R ¹	R ²	R ³	% yield ^b	
					A ^a	B ^b
1	4a	Me	Me	H	85	90
2	4b	Et	Me	H	80	85
3	4c	n-Pro	Me	H	83	88
4	4d	iso-Pro	Me	H	85	92
5	4e	Ph	Me	H	75	82
6	4f	Me	Me	Me	82	87
7	4g	Me	Me	Ph	80	85
8	4h	Et	Me	Ph	76	83
9	4i	n-Pro	Me	Ph	80	85
10	4j	iso-Pro	Me	Ph	82	88
11	4k	Ph	Me	Ph	78	82
12	4l	Me	Me	<i>o</i> -ClPh	70	80
13	4m	Me	Me	<i>p</i> -ClPh	75	80

^a Isolated yields for the three-component one-pot reaction.

^b Isolated yields for the four-component one-pot reaction.

The structures of **4a~4m** were confirmed by ¹H NMR, ¹³C NMR, MS and IR spectroscopy. The detailed data and associated spectra were provided in the supporting information. The structure of representative compound **4a** was precisely determined by single crystal X-ray analysis (Fig. 2). (See the supporting information for more details).

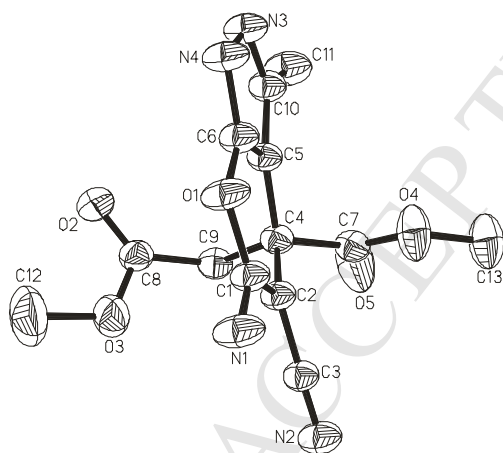
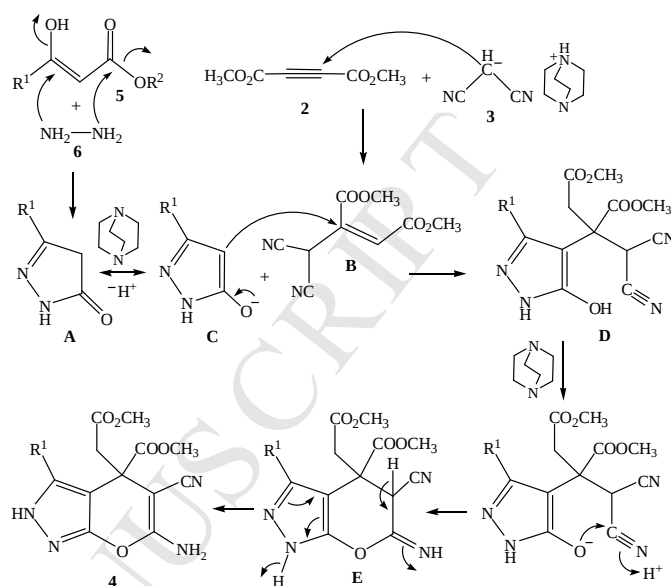


Figure 2. ORTEP diagram for the molecular structure of compound **4a**.

A detailed reaction mechanism catalyzed by DABCO for the formation of pyrano[2,3-c]pyrazole dicarboxylate were shown in Scheme 3. The role of DABCO might just simply be serving as an effective general base. Initially, spontaneous condensation of methyl β -ketoesters with the hydrazine formed pyrazole-5-one, and the Michael addition of malononitrile to dimethyl acetylene dicarboxylate led to the formation of intermediate B,

respectively. Then, the Michael addition of intermediate C to intermediate B resulted in an intermediate D in the presence of DABCO. Next, the intramolecular nucleophilic attack of the oxygen to one of the nitrile group afforded intermediate E. Finally, subsequent isomerization occurred in the pyran and pyrazole rings produced the final compound **4**.



Scheme 3. Proposed reaction mechanism for the DABCO catalyzed one-pot synthesis of multi-substituted pyrano[2,3-c]pyrazole dicarboxylates.

Conclusion

We report a novel and convenient one-pot synthesis of multi-substituted pyrano[2,3-c]pyrazole dicarboxylate derivatives using three-component and four-component reactions. The four-component reactions proceeded smoothly and resulted in good to excellent yields. Our one-pot method offers several advantages, including short reaction time, simple experimental procedure and no toxic byproducts. The products are interesting nitrogen and oxygen heterocyclic molecules containing two carboxyl groups.

4. Experimental

4.1. General information

All of the solvents were purified and dried prior to use. The chemicals used in this study were analytical grade and purchased from TCI, Alfa and Acros. The melting points were measured in open capillary tubes and are uncorrected. The ¹H and ¹³C NMR spectra were recorded on a Bruker INOVA-400 NMR instrument at 400 MHz and 100 MHz using TMS as an internal standard and DMSO-*d*₆ as the solvent. The chemical shifts were given in parts per million (δ scale), and the coupling constants were given in Hertz. Infrared (IR) spectra were recorded on a BRUKER-EQUINOX55 spectrophotometer using KBr pellets. ESI-MS spectra and HRMS data were obtained using a TOF-MS spectrometer.

4.2 General procedure for the three-component one-pot synthesis of multi-substituted pyrano[2,3-c] pyrazole dicarboxylates (**4**)

DABCO (20 % mmol) was added to a mixture of pyrazol-5-one (2 mmol), dimethyl acetylenedicarboxylate (2 mmol) and malononitrile (2 mmol) in 10 mL of EtOH. The reaction mixture was heated at 50 °C for 30–60 min. After the reaction was completed, the mixture was cooled to room temperature. After 5 mL of water was added, the resulting clear solution was maintained at room temperature for one additional day resulting in the precipitation of large quantities of crystals. White or colorless crystals were obtained via filtration with successive washing with water and ethyl acetate. The pure product was obtained by recrystallization in 80 % EtOH.

4.3 General procedure for the four-component one-pot synthesis of multi-substituted pyrano[2,3-*c*] pyrazole dicarboxylates (4)

An EtOH (5 mL) solution containing dimethyl acetylenedicarboxylate (2 mmol), malononitrile (2 mmol) and DABCO (20 % mmol) was added dropwise within 5 min to a stirred mixture of ethyl acetoacetate (2 mmol) and hydrazine (2 mmol) in EtOH (5 mL) at room temperature. Then, the reaction mixture was stirred at 50 °C for 30–60 min. After completion of the reaction, a white or colorless crystal was obtained using a method similar to one mentioned above.

4.3.1. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-methyl-2,4-dihydropyrano[2,3-*c*] pyrazole-4-carboxylate (4a). White crystals, mp: 212–214 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 12.25 (s, 1H, NH), 7.08 (s, 2H, NH₂), 3.65 (s, 3H, OCH₃), 3.44 (s, 3H, OCH₃), 3.06 (d, 1H, *J*=15.2 Hz, CH_{2a}), 2.91 (d, 1H, *J*=15.2 Hz, CH_{2b}), 2.12 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.0, 169.2, 161.7, 154.5, 135.8, 119.0, 94.2, 55.8, 52.8, 51.2, 43.4, 10.3. IR (KBr) ν : 3322, 3173 (brs, NH, NH₂), 2958 (C-H), 2196 (C $\equiv$ N), 1733 (COO), 1662, 1601, 1497, 1409 (C=C, C=N), 1252, 1155, 1073, 1022 (C-O, C-N), 717, 580 (N-H) cm⁻¹; ESI-MS *m/z*: 307 [M+H]⁺, 247, 233; HRMS (ESI) calcd for C₁₃H₁₅N₄O₅ [M+H]⁺ 307.0964, found 307.0995.

4.3.2. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-ethyl-2,4-dihydropyrano[2,3-*c*] pyrazole-4-carboxylate (4b). White crystals; mp: 232–234 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 12.27 (s, 1H, NH), 7.09 (s, 2H, NH₂), 3.64 (s, 3H, OCH₃), 3.43 (s, 3H, OCH₃), 3.05 (d, 1H, *J*=15.2 Hz, CH_{2a}), 2.91 (d, 1H, *J*=15.2 Hz, CH_{2b}), 2.49 (q, 2H, CH₂), 1.09 (t, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.3 (C₁), 169.2, 161.7, 154.4, 141.1, 119.0, 93.5, 55.8, 52.8, 51.1, 43.5, 17.9, 12.5; IR (KBr) ν : 3321, 3280, 3169 (brs, NH, NH₂), 2955 (C-H), 2196 (C $\equiv$ N), 1732 (COO), 1662, 1598, 1498, 1410 (C=C, C=N), 1249, 1155, 1020 (C-O, C-N), 727, 585 (N-H) cm⁻¹; ESI-MS *m/z*: 321 [M+H]⁺; HRMS (ESI) calcd for C₁₄H₁₆N₄O₅ [M+H]⁺ 321.1120, found 321.1146.

4.3.3. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-propyl-2,4-dihydropyrano[2,3-*c*] pyrazole-4-carboxylate (4c). White crystals; mp: 237–239 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 12.25 (s, 1H, NH), 7.09 (s, 2H, NH₂), 3.63 (s, 3H, OCH₃), 3.43 (s, 3H, OCH₃), 3.04 (d, 1H, *J*=15.2 Hz, CH_{2a}), 2.45 (m, 2H, CH₂), 2.91 (d, 1H, *J*=15.2 Hz, CH_{2b}), 1.51 (m, 2H, CH₂), 0.85 (t, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.4 (C₁), 169.2, 161.7, 154.4, 139.6, 119.0, 93.9, 55.8, 52.8, 51.1, 43.4, 26.4, 20.9, 13.4; IR (KBr) ν : 3321, 3173 (brs, NH₂), 2954 (C-H), 2195 (C $\equiv$ N), 1730 (COO), 1720 (COO), 1661, 1595, 1495, 1408 (C=C, C=N), 1246, 1156, 1070, 1016 (C-O, C-N), 722, 580 (N-

H) cm⁻¹; ESI-MS *m/z*: 335 [M+H]⁺, 275, 261; HRMS (ESI) calcd for C₁₅H₁₉N₄O₅ [M+H]⁺ 335.1277, found 335.1353.

4.3.4. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-isopropyl-2,4-dihydro-2,4-dihydro pyrano[2,3-*c*] pyrazole-4-carboxylate (4d). White crystals; mp: 218–220 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 12.30 (s, 1H, NH), 7.09 (s, 2H, NH₂), 3.64 (s, 3H, OCH₃), 3.42 (s, 3H, OCH₃), 3.04 (d, 1H, *J*=15.2 Hz, CH_{2a}), 2.90 (d, 1H, *J*=15.2 Hz, CH_{2b}), 2.87 (m, 1H, *J*=6.8 Hz, CH), 1.19 (d, 3H, *J*=6.8 Hz, CH₃), 1.09 (d, 3H, *J*=6.8 Hz, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.5, 169.2, 161.7, 154.0, 145.4, 119.0, 92.6, 55.9, 52.8, 51.1, 43.6, 24.9, 21.7, 21.3; IR (KBr) ν : 3351, 3172 (brs, NH₂), 2980, 2954 (C-H), 2195 (C $\equiv$ N), 1731 (COO), 1662, 1602, 1495, 1408 (C=C, C=N), 1241, 1154, 1137, 1016 (C-O, C-N), 697, 629 (N-H) cm⁻¹; ESI-MS *m/z*: 335 [M+H]⁺, 275, 261. HRMS (ESI) calcd for C₁₅H₁₉N₄O₅ [M+H]⁺ 335.1277, found 335.1327.

4.3.5. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-phenyl-2,4-dihydropyrano[2,3-*c*]pyrazole-4-carboxylate (4e). White crystals; mp: 212–214 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 12.82 (s, 1H, NH), 7.50–7.25 (m, 5H, C₆H₅), 7.24 (s, 2H, NH₂), 3.64 (s, 3H, OCH₃), 3.36 (s, 3H, OCH₃), 2.88 (d, 1H, *J*=15.2 Hz, CH_{2a}), 2.59 (d, 1H, *J*=15.2 Hz, CH_{2b}); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.6, 169.0, 161.6, 154.9, 139.0, 129.2, 128.7, 127.9, 118.5, 95.3, 55.7, 52.9, 51.1, 43.4; IR (KBr) ν : 3306, 3183 (brs, NH, NH₂), 2955 (C-H), 2193 (C $\equiv$ N), 1731 (COO), 1641, 1599, 1500, 1407 (C=C, C=N), 1241, 1156, 1031 (C-O, C-N), 768, 688, 636 (C₆H₅) cm⁻¹; ESI-MS *m/z*: 369 [M+H]⁺, 309, 295. HRMS (ESI) calcd for C₁₈H₁₇N₄O₄ [M+H]⁺ 369.1121, found 369.1143.

4.3.6. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1,3-dimethyl-4H-pyrano[2,3-*c*] pyrazole-4-carboxylate (4f). White crystals; mp: 152–154 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.31 (s, 2H, NH₂), 3.65 (s, 3H, OCH₃), 3.57 (s, 3H, OCH₃), 3.44 (s, 3H, CH₃), 3.05–2.91 (dd, 2H, *J*=15.2 Hz, CH₂), 2.01 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.8, 169.2, 160.5, 144.3, 142.0, 118.1, 92.8, 56.7, 52.8, 51.2, 44.2, 33.3, 12.9; IR (KBr) ν : 3403, 3325, 3218 (brs, NH₂), 2956 (C-H), 2199 (C $\equiv$ N), 1736 (COO), 1663, 1561, 1437, 1403 (C=C, C=N), 1240, 1010 (C-O, C-N) cm⁻¹; ESI-MS *m/z*: 321 [M+H]⁺; HRMS (ESI) calcd for C₁₄H₁₆N₄O₅ [M+H]⁺ 321.1120, found 321.1155.

4.3.7. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-methyl-4H-pyrano[2,3-*c*] pyrazole-4-carboxylate (4g). White crystal; mp: 192–194 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.75–7.34 (m, 5H, C₆H₅), 7.49 (s, 2H, NH₂), 3.71 (s, 3H, OCH₃), 3.47 (s, 3H, OCH₃), 3.12–2.98 (dd, *J*=15.6 Hz, CH₃), 2.14 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.5, 169.3, 160.4, 144.6, 144.0, 137.1, 129.3, 126.4, 119.9, 118.1, 95.5, 56.5, 55.9, 53.0, 51.3, 43.9, 18.5, 13.1; IR (KBr) ν : 3440, 3337, 3226 (brs, NH₂), 2952 (C-H), 2193 (C $\equiv$ N), 1737 (COO), 1702 (COO), 1646, 1521, 1444, 1395 (C=C, C=N), 1233, 1007 (C-O, C-N), 756, 696 cm⁻¹; ESI-MS *m/z*: 383 [M+H]⁺, 405 [M+Na]⁺; HRMS (ESI) calcd for C₁₉H₁₉N₄O₅ [M+H]⁺ 383.1254, found 383.1277.

4.3.8. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-ethyl-4H-pyrano[2,3-*c*]pyrazole-4-carboxylate (4h). White crystals; mp: 148–150 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.77–7.34 (m, 5H, C₆H₅), 7.49 (s, 2H, NH₂), 3.70 (s, 3H, OCH₃), 3.45 (s, 3H, OCH₃), 3.10 (d, 1H, *J*=15.6 Hz, CH_{2a}), 2.98 (d, 1H, *J*=15.6 Hz, CH_{2b}), 2.50 (m, 2H, CH₂), 1.17 (t, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.8, 169.3, 160.4, 149.2, 144.0, 137.2, 129.3, 126.4, 119.9, 118.1, 94.9, 56.6, 53.1, 51.3, 43.9, 40.0, 38.8, 20.3, 11.7; IR (KBr) ν : 3439, 3341, 3231 (brs,

NH₂), 2978, 2957 (C-H), 2195 (C≡N), 1739 (COO), 1707 (COO), 1648, 1578, 1519, 1440, 1398 (C=C, C=N), 1249, 1135, 1069, 1013 (C-O, C-N), 763, 691 cm⁻¹. ESI-MS *m/z*: 397 [M+H]⁺. HRMS (ESI) calcd for C₂₀H₂₁N₄O₅ [M+H]⁺ 397.1434, found 397.1477.

4.3.9. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-propyl-4H-pyrano[2,3-*c*] pyrazole-4-carboxylate (4i). White crystals; mp: 156–158 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.33–7.78 (m, 5H, C₆H₅), 7.51 (s, 2H, NH₂), 3.69 (s, 3H, OCH₃), 3.46 (s, 3H, OCH₃), 3.10 (d, 1H, *J*=15.6 Hz, CH_{2a}), 2.99 (d, 1H, *J*=15.6 Hz, CH_{2b}), 2.48 (m, 2H, CH₂), 1.61 (m, 2H, CH₂), 0.93 (t, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.8 (C₁), 169.2, 160.4, 148.0, 143.9, 137.5, 129.3, 126.4, 119.9, 118.1, 95.16, 56.6, 53.01, 51.3, 43.9, 28.9, 20.4, 13.7; IR (KBr) *v*: 3490, 3356, 3194 (brs, NH₂), 2957 (C-H), 2194 (C≡N), 1727 (2COO), 1658, 1596, 1520, 1454, 1396 (C=C, C=N), 1246, 1006 (C-O, C-N), 745, 684 cm⁻¹; ESI-MS *m/z*: 411 [M+H]⁺. HRMS (ESI) calcd for C₂₁H₂₃N₄O₅ [M+H]⁺ 411.1590, found 411.1566.

4.3.10. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-isopropyl-4H-pyrano[2,3-*c*]pyrazole-4-carboxylate (4j). White crystals; mp: 182–184 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.51 (s, 2H, NH₂), 7.34–7.76 (m, 5H, C₆H₅), 3.70 (s, 3H, OCH₃), 3.45 (s, 3H, OCH₃), 3.10 (d, *J*=15.2 Hz, CH_{2a}), 2.98 (d, *J*=15.2 Hz, CH_{2b}), 2.80 (m, 1H, *J*=6.8 Hz, CH), 1.24 (d, 3H, *J*=6.8 Hz), 1.12 (d, 3H, *J*=6.8 Hz); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.0 (C₁), 169.2, 160.3, 153.5, 143.5, 137.1, 129.3, 126.4, 120.0, 118.1, 94.2, 56.7, 53.0, 51.3, 44.0, 26.6, 22.2, 21.7; IR (KBr) *v*: 3388, 3318, 3209 (brs, NH₂), 2971 (C-H), 2200 (C≡N), 1748 (COO), 1725 (COO), 1663, 1598, 1520, 1456, 1396 (C=C, C=N), 1228, 1166, 1091, 1000 (C-O, C-N), 754, 691 cm⁻¹; ESI-MS *m/z*: 411 [M+H]⁺. HRMS (ESI) calcd for C₂₁H₂₃N₄O₅ [M+H]⁺ 411.1590, found 411.1559.

4.3.11. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1,3-diphenyl-4H-pyrano[2,3-*c*] pyrazole-4-carboxylate (4k). White crystals; mp: 220–224 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.36–7.87 (m, 10H, 2C₆H₅), 7.64 (s, 2H, NH₂), 3.70 (s, 3H, OCH₃), 3.40 (s, 3H, OCH₃), 2.97 (d, 1H, *J*=15.6 Hz, CH₂); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.0, 169.0, 160.3, 147.4, 144.5, 136.9, 132.5, 129.5, 128.7, 128.5, 127.7, 127.0, 120.4, 117.7, 95.9, 56.6, 53.1, 51.2, 44.0; IR (KBr) *v*: 3401, 3323, 3210 (brs, NH₂), 2954 (C-H), 2200 (C≡N), 1736 (COO), 1657, 1596, 1520, 1454, 1401 (C=C, C=N), 1240, 1074, 997 (C-O, C-N), 755, 688

cm⁻¹; ESI-MS *m/z*: 445 [M+H]⁺. HRMS (ESI) calcd for C₂₄H₂₁N₄O₅ [M+H]⁺ 445.1590, found 445.1528.

4.3.12. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-(*o*-chlorophenyl)-3-propyl-2,4-dihydropyrano[2,3-*c*]pyrazole-4-carboxylate (4l). White crystals, mp: 188–190 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.70–7.52 (m, 4H, *o*-Cl-C₆H₄), 7.30 (s, 2H, NH₂), 3.72 (s, 3H, CH₃), 3.46 (s, 3H, OCH₃), 3.13 (d, 1H, *J*=15.6 Hz, CH_{2a}), 2.96 (d, 1H, *J*=15.6 Hz, CH_{2b}), 2.14 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.6 (C₁), 169.2, 160.4, 145.2, 145.1, 133.7, 131.2, 130.7, 130.2, 129.7, 128.2, 118.2, 93.9, 56.7, 53.0, 51.3, 44.3, 13.1; IR (KBr) *v*: 3449, 3315, 3172 (brs, NH₂), 2198 (C≡N), 1736 (COO), 1664, 1582, 1538, 1436, 1399 (C=C, C=N), 1241, 1138, 1092, 1018 (C-O, C-N) cm⁻¹; ESI-MS *m/z*: 417 [M+H]⁺. HRMS (ESI) calcd for C₁₉H₁₈N₄O₅ [M+H]⁺ 417.0887, found 417.0890.

4.3.13. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-(*p*-chlorophenyl)-3-methyl-2,4-dihydropyrano[2,3-*c*]pyrazole-4-carboxylate (4m). White crystals; mp: 208–210 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.80 (d, 2H, *J*=9.2 Hz, *p*-Cl-C₆H₄), 7.54 (d, 2H, *J*=9.2 Hz, *p*-Cl-C₆H₄), 7.52 (s, 2H, NH₂), 3.70 (s, 3H, CH₃), 3.46 (s, 3H, OCH₃), 3.12 (d, 1H, *J*=15.6 Hz, CH_{2a}), 2.98 (d, 1H, *J*=15.6 Hz, CH_{2b}), 2.14 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.4 (C₁), 169.2, 160.3, 145.1, 144.1, 135.9, 130.5, 129.2, 121.2, 118.0, 95.7, 56.5, 53.1, 51.3, 43.9, 13.1; IR (KBr) *v*: 3411, 3329, 3229 (brs, NH₂), 2955 (C-H), 2196 (C≡N), 1722 (COO), 1659, 1591, 1523, 1393 (C=C, C=N), 1237, 1161, 1092, 1025 (C-O, C-N) cm⁻¹; ESI-MS *m/z*: 417 [M+H]⁺. HRMS (ESI) calcd for C₁₉H₁₈N₄O₅ [M+H]⁺ 417.0887, found 417.0967.

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Supplementary Material

Supplementary data associated with this article can be found in the online version, at <http://dx.doi.org/10.1016/j.tet>.

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Supporting Information

Convenient DABCO-catalyzed one-pot synthesis of multi-substituted pyrano[2,3-c]pyrazole dicarboxylates

Ablajan, Keyume*, Zeynepgul Esmayil, Liju Wang and Feng Jun

Key Laboratory of Oil & Gas Fine Chemicals Ministry of Education & Xinjiang Uyghur Autonomous Region, College of Chemistry and Chemical Engineering, Xinjiang University, Urumqi 830046, P. R. China.

E-mail: ablajan209@hotmail.com

CONTENTS

1. General information.....	S2
2. General procedure for the preparation of compounds 4a-4m	S2
3. Physical and spectroscopic data for compounds 4a-4m	S3-S8
4. ¹ H NMR, ¹³ C NMR, IR and MS spectra.....	S8-S32
5. X-ray crystallographic data for compound 4a	S33-S40

1. General information

All of the solvents were purified and dried prior to use. The chemicals used in this study were analytical grade and purchased from TCI, Alfa and Acros. The melting points were measured in open capillary tubes and are uncorrected. The ^1H and ^{13}C NMR spectra were recorded on a Bruker INOVA-400 NMR instrument at 400 MHz and 100 MHz using TMS as an internal standard and $\text{DMSO-}d_6$ as the solvent. The chemical shifts were given in parts per million (δ scale), and the coupling constants were given in Hertz. Infrared (IR) spectra were recorded on a BRUKER-EQUINOX55 spectrophotometer using KBr pellets. ESI-MS spectra and HRMS data were obtained using a TOF-MS spectrometer.

2. General procedure for the preparation of compounds 4a-4m

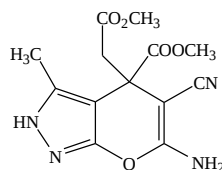
(1) General procedure for the three-component one-pot synthesis of multisubstituted pyrano[2,3-c]pyrazole dicarboxylates (4)

DABCO (20 % mmol) was added to a mixture of pyrazol-5-one (2 mmol), dimethyl acetylenedicarboxylate (2 mmol) and malononitrile (2 mmol) in 10 mL of EtOH. The reaction mixture was heated at 50 °C for 30–60 min. After the reaction was completed, the mixture was cooled to room temperature. After 5 mL of water was added, the resulting clear solution was maintained at room temperature for one additional day resulting in the precipitation of large quantities of crystals. White or colorless crystals were obtained via filtration with successive washing with water and ethyl acetate. The pure product was obtained by recrystallization in 80 % EtOH.

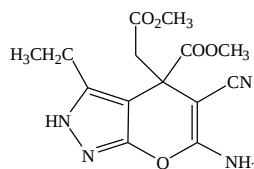
(2) General procedure for the four-component one-pot synthesis of multisubstituted pyrano[2,3-c]pyrazole dicarboxylates (4)

An EtOH (5 mL) solution containing dimethyl acetylenedicarboxylate (2 mmol), malononitrile (2 mmol) and DABCO (20 % mmol) was added dropwise within 5 min to a stirred mixture of ethyl acetoacetate (2 mmol) and hydrazine (2 mmol) in EtOH (5 mL) at room temperature. Then, the reaction mixture was stirred at 50 °C for 30–60 min. After completion of the reaction, a white or colorless crystal was obtained using a method similar to one mentioned above.

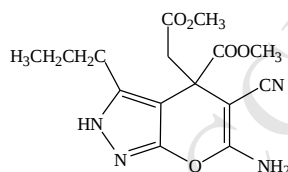
3. Physical and spectroscopic data for compounds 4a-4m



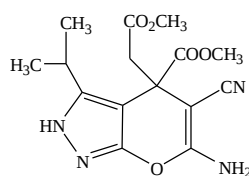
4.3.1. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-methyl-2,4-dihydropyrano[2,3-c]pyrazole-4-carboxylate (4a)*. White crystals, mp: 212–214 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 12.25 (s, 1H, NH), 7.08 (s, 2H, NH_2), 3.65 (s, 3H, OCH_3), 3.44 (s, 3H, OCH_3), 3.06 (d, 1H, $J=15.2$ Hz, CH_{2a}), 2.91 (d, 1H, $J=15.2$ Hz, CH_{2b}), 2.12 (s, 3H, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 172.0, 169.2, 161.7, 154.5, 135.8, 119.0, 94.2, 55.8, 52.8, 51.2, 43.4, 10.3. IR (KBr) ν : 3322, 3173 (brs, NH, NH_2), 2958 (C-H), 2196 ($\text{C}\equiv\text{N}$), 1733 (COO), 1662, 1601, 1497, 1409 ($\text{C}=\text{C}$, $\text{C}=\text{N}$), 1252, 1155, 1073, 1022 (C-O, C-N), 717, 580 (N-H) cm^{-1} ; ESI-MS m/z : 307 $[\text{M}+\text{H}]^+$, 247, 233; HRMS (ESI) calcd for $\text{C}_{13}\text{H}_{15}\text{N}_4\text{O}_5$ $[\text{M}+\text{H}]^+$ 307.0964, found 307.0995.



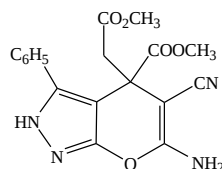
4.3.2. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-ethyl-2,4-dihydropyrano[2,3-c]pyrazole-4-carboxylate (4b)*. White crystals; mp: 232–234 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 12.27 (s, 1H, NH), 7.09 (s, 2H, NH_2), 3.64 (s, 3H, OCH_3), 3.43 (s, 3H, OCH_3), 3.05 (d, 1H, $J=15.2$ Hz, CH_{2a}), 2.91 (d, 1H, $J=15.2$ Hz, CH_{2b}), 2.49 (q, 2H, CH_2), 1.09 (t, 3H, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 172.3 (C_1), 169.2, 161.7, 154.4, 141.1, 119.0, 93.5, 55.8, 52.8, 51.1, 43.5, 17.9, 12.5; IR (KBr) ν : 3321, 3280, 3169 (brs, NH, NH_2), 2955 (C-H), 2196 ($\text{C}\equiv\text{N}$), 1732 (COO), 1662, 1598, 1498, 1410 ($\text{C}=\text{C}$, $\text{C}=\text{N}$), 1249, 1155, 1020 (C-O, C-N), 727, 585 (N-H) cm^{-1} ; ESI-MS m/z : 321 $[\text{M}+\text{H}]^+$; HRMS (ESI) calcd for $\text{C}_{14}\text{H}_{16}\text{N}_4\text{O}_5$ $[\text{M}+\text{H}]^+$ 321.1120, found 321.1146.



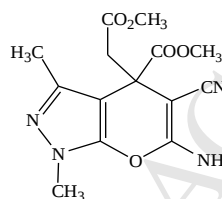
4.3.3. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-propyl-2,4-dihydropyrano[2,3-c]pyrazole-4-carboxylate (4c)*. White crystals; mp: 237–239 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 12.25 (s, 1H, NH), 7.09 (s, 2H, NH_2), 3.63 (s, 3H, OCH_3), 3.43 (s, 3H, OCH_3), 3.04 (d, 1H, $J=15.2$ Hz, CH_{2a}), 2.45 (m, 2H, CH_2), 2.91 (d, 1H, $J=15.2$ Hz, CH_{2b}), 1.51 (m, 2H, CH_2), 0.85 (t, 3H, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 172.4 (C_1), 169.2, 161.7, 154.4, 139.6, 119.0, 93.9, 55.8, 52.8, 51.1, 43.4, 26.4, 20.9, 13.4; IR (KBr) ν : 3321, 3173 (brs, NH_2), 2954 (C-H), 2195 ($\text{C}\equiv\text{N}$), 1730 (COO), 1720 (COO), 1661, 1595, 1495, 1408 ($\text{C}=\text{C}$, $\text{C}=\text{N}$), 1246, 1156, 1070, 1016 (C-O, C-N), 722, 580 (N-H) cm^{-1} ; ESI-MS m/z : 335 $[\text{M}+\text{H}]^+$, 275, 261; HRMS (ESI) calcd for $\text{C}_{15}\text{H}_{19}\text{N}_4\text{O}_5$ $[\text{M}+\text{H}]^+$ 335.1277, found 335.1353.



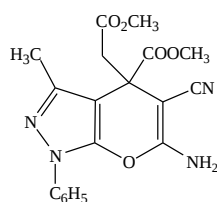
4.3.4. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-isopropyl-2,4-dihydro-2,4-dihydro pyrano [2,3-c]pyrazole-4-carboxylate (4d)*. White crystals; mp: 218–220 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 12.30 (s, 1H, NH), 7.09 (s, 2H, NH_2), 3.64 (s, 3H, OCH_3), 3.42 (s, 3H, OCH_3), 3.04 (d, 1H, $J=15.2$ Hz, CH_{2a}), 2.90 (d, 1H, $J=15.2$ Hz, CH_{2b}), 2.87 (m, 1H, $J=6.8$ Hz, CH), 1.19 (d, 3H, $J=6.8$ Hz, CH_3), 1.09 (d, 3H, $J=6.8$ Hz, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 172.5, 169.2, 161.7, 154.0, 145.4, 119.0, 92.6, 55.9, 52.8, 51.1, 43.6, 24.9, 21.7, 21.3; IR (KBr) ν : 3351, 3172 (brs, NH_2), 2980, 2954 (C-H), 2195 ($\text{C}\equiv\text{N}$), 1731 (COO), 1662, 1602, 1495, 1408 (C=C, C=N), 1241, 1154, 1137, 1016 (C-O, C-N), 697, 629 (N-H) cm^{-1} ; ESI-MS m/z : 335 $[\text{M}+\text{H}]^+$, 275, 261. HRMS (ESI) calcd for $\text{C}_{15}\text{H}_{19}\text{N}_4\text{O}_5$ $[\text{M}+\text{H}]^+$ 335.1277, found 335.1327.



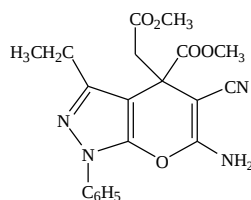
4.3.5. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-3-phenyl-2,4-dihydro-2,4-dihydro pyrano [2,3-c]pyrazole-4-carboxylate (4e)*. White crystals; mp: 212–214 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 12.82 (s, 1H, NH), 7.50–7.25 (m, 5H, C_6H_5), 7.24 (s, 2H, NH_2), 3.64 (s, 3H, OCH_3), 3.36 (s, 3H, OCH_3), 2.88 (d, 1H, $J=15.2$ Hz, CH_{2a}), 2.59 (d, 1H, $J=15.2$ Hz, CH_{2b}); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 172.6, 169.0, 161.6, 154.9, 139.0, 129.2, 128.7, 127.9, 118.5, 95.3, 55.7, 52.9, 51.1, 43.4; IR (KBr) ν : 3306, 3183 (brs, NH, NH_2), 2955 (C-H), 2193 ($\text{C}\equiv\text{N}$), 1731 (COO), 1641, 1599, 1500, 1407 (C=C, C=N), 1241, 1156, 1031 (C-O, C-N), 768, 688, 636 (C_6H_5) cm^{-1} ; ESI-MS m/z : 369 $[\text{M}+\text{H}]^+$, 309, 295. HRMS (ESI) calcd for $\text{C}_{18}\text{H}_{17}\text{N}_4\text{O}_4$ $[\text{M}+\text{H}]^+$ 369.1121, found 369.1143.



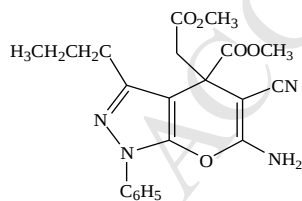
4.3.6. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1,3-dimethyl-4H-pyrano[2,3-c] pyrazole-4-carboxylate (4f)*. White crystals; mp: 152–154 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 7.31 (s, 2H, NH_2), 3.65 (s, 3H, OCH_3), 3.57 (s, 3H, OCH_3), 3.44 (s, 3H, CH_3), 3.05–2.91 (dd, 2H, $J=15.2$ Hz, CH_2), 2.01 (s, 3H, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 171.8, 169.2, 160.5, 144.3, 142.0, 118.1, 92.8, 56.7, 52.8, 51.2, 44.2, 33.3, 12.9; IR (KBr) ν : 3403, 3325, 3218 (brs, NH_2), 2956 (C-H), 2199 ($\text{C}\equiv\text{N}$), 1736 (COO), 1663, 1561, 1437, 1403 (C=C, C=N), 1240, 1010 (C-O, C-N) cm^{-1} ; ESI-MS m/z : 321 $[\text{M}+\text{H}]^+$; HRMS (ESI) calcd for $\text{C}_{14}\text{H}_{16}\text{N}_4\text{O}_5$ $[\text{M}+\text{H}]^+$ 321.1120, found 321.1155.



4.3.7. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-methyl-4H-pyrano[2,3-c]pyrazole-4-carboxylate (4g)*. White crystal; mp: 192–194 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 7.75–7.34 (m, 5H, C_6H_5), 7.49 (s, 2H, NH_2), 3.71 (s, 3H, OCH_3), 3.47 (s, 3H, OCH_3), 3.12–2.98 (dd, $J=15.6$ Hz, CH_3), 2.14 (s, 3H, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 171.5, 169.3, 160.4, 144.6, 144.0, 137.1, 129.3, 126.4, 119.9, 118.1, 95.5, 56.5, 55.9, 53.0, 51.3, 43.9, 18.5, 13.1; IR (KBr) ν : 3440, 3337, 3226 (brs, NH_2), 2952 (C-H), 2193 ($\text{C}\equiv\text{N}$), 1737 (COO), 1702 (COO), 1646, 1521, 1444, 1395 (C=C, C=N), 1233, 1007 (C-O, C-N), 756, 696 cm^{-1} ; ESI-MS m/z : 383 $[\text{M}+\text{H}]^+$, 405 $[\text{M}+\text{Na}]^+$; HRMS (ESI) calcd for $\text{C}_{19}\text{H}_{19}\text{N}_4\text{O}_5$ $[\text{M}+\text{H}]^+$ 383.1254, found 383.1277.

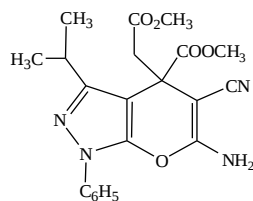


4.3.8. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-ethyl-4H-pyrano[2,3-c]pyrazole-4-carboxylate (4h)*. White crystals; mp: 148–150 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 7.77–7.34 (m, 5H, C_6H_5), 7.49 (s, 2H, NH_2), 3.70 (s, 3H, OCH_3), 3.45 (s, 3H, OCH_3), 3.10 (d, 1H, $J=15.6$ Hz, CH_{2a}), 2.98 (d, 1H, $J=15.6$ Hz, CH_{2b}), 2.50 (m, 2H, CH_2), 1.17 (t, 3H, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 171.8, 169.3, 160.4, 149.2, 144.0, 137.2, 129.3, 126.4, 119.9, 118.1, 94.9, 56.6, 53.1, 51.3, 43.9, 40.0, 38.8, 20.3, 11.7; IR (KBr) ν : 3439, 3341, 3231 (brs, NH_2), 2978, 2957 (C-H), 2195 ($\text{C}\equiv\text{N}$), 1739 (COO), 1707 (COO), 1648, 1578, 1519, 1440, 1398 (C=C, C=N), 1249, 1135, 1069, 1013 (C-O, C-N), 763, 691 cm^{-1} . ESI-MS m/z : 397 $[\text{M}+\text{H}]^+$. HRMS (ESI) calcd for $\text{C}_{20}\text{H}_{21}\text{N}_4\text{O}_5$ $[\text{M}+\text{H}]^+$ 397.1434, found 397.1477.

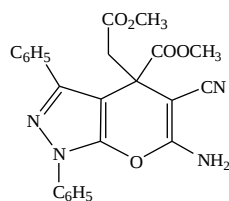


4.3.9. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-propyl-4H-pyrano[2,3-c]pyrazole-4-carboxylate (4i)*. White crystals; mp: 156–158 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): 7.33–7.78 (m, 5H, C_6H_5), 7.51 (s, 2H, NH_2), 3.69 (s, 3H, OCH_3), 3.46 (s, 3H, OCH_3), 3.10 (d, 1H, $J=15.6$ Hz, CH_{2a}), 2.99 (d, 1H, $J=15.6$ Hz, CH_{2b}), 2.48 (m, 2H, CH_2), 1.61 (m, 2H, CH_2), 0.93 (t, 3H, CH_3); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): 171.8 (C1), 169.2, 160.4, 148.0, 143.9, 137.5, 129.3, 126.4, 119.9, 118.1, 95.16, 56.6, 53.01, 51.3, 43.9, 28.9, 20.4, 13.7; IR (KBr) ν : 3490, 3356, 3194 (brs, NH_2),

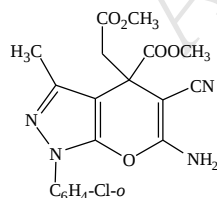
2957 (C-H), 2194 (C≡N), 1727 (2COO), 1658, 1596, 1520, 1454, 1396 (C=C, C=N), 1246, 1006 (C-O, C-N), 745, 684 cm⁻¹; ESI-MS *m/z*: 411 [M+H]⁺. HRMS (ESI) calcd for C₂₁H₂₃N₄O₅ [M+H]⁺ 411.1590, found 411.1566.



4.3.10. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-phenyl-3-isopropyl-4H-pyrano[2,3-c]pyrazole-4-carboxylate (4j). White crystals; mp: 182–184 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.51 (s, 2H, NH₂), 7.34–7.76 (m, 5H, C₆H₅), 3.70 (s, 3H, OCH₃), 3.45 (s, 3H, OCH₃), 3.10 (d, *J*=15.2 Hz, CH_{2a}), 2.98 (d, *J*=15.2 Hz, CH_{2b}), 2.80 (m, 1H, *J*=6.8 Hz, CH), 1.24 (d, 3H, *J*=6.8 Hz), 1.12 (d, 3H, *J*=6.8 Hz); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.0 (C₁), 169.2, 160.3, 153.5, 143.5, 137.1, 129.3, 126.4, 120.0, 118.1, 94.2, 56.7, 53.0, 51.3, 44.0, 26.6, 22.2, 21.7; IR (KBr) *v*: 3388, 3318, 3209 (brs, NH₂), 2971 (C-H), 2200 (C≡N), 1748 (COO), 1725 (COO), 1663, 1598, 1520, 1456, 1396 (C=C, C=N), 1228, 1166, 1091, 1000 (C-O, C-N), 754, 691 cm⁻¹; ESI-MS *m/z*: 411 [M+H]⁺. HRMS (ESI) calcd for C₂₁H₂₃N₄O₅ [M+H]⁺ 411.1590, found 411.1559.

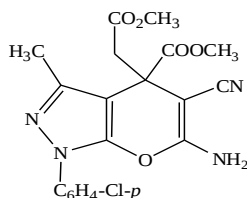


4.3.11. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1,3-diphenyl-4H-pyrano[2,3-c]pyrazole-4-carboxylate (4k). White crystals; mp: 220–224 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.36–7.87 (m, 10H, 2C₆H₅), 7.64 (s, 2H, NH₂), 3.70 (s, 3H, OCH₃), 3.40 (s, 3H, OCH₃), 2.97 (d, 1H, *J*=15.6 Hz, CH₂); ¹³C NMR (100 MHz, DMSO-*d*₆): 172.0, 169.0, 160.3, 147.4, 144.5, 136.9, 132.5, 129.5, 128.7, 128.5, 127.7, 127.0, 120.4, 117.7, 95.9, 56.6, 53.1, 51.2, 44.0; IR (KBr) *v*: 3401, 3323, 3210 (brs, NH₂), 2954 (C-H), 2200 (C≡N), 1736 (COO), 1657, 1596, 1520, 1454, 1401 (C=C, C=N), 1240, 1074, 997 (C-O, C-N), 755, 688 cm⁻¹; ESI-MS *m/z*: 445 [M+H]⁺. HRMS (ESI) calcd for C₂₄H₂₁N₄O₅ [M+H]⁺ 445.1590, found 445.1528.

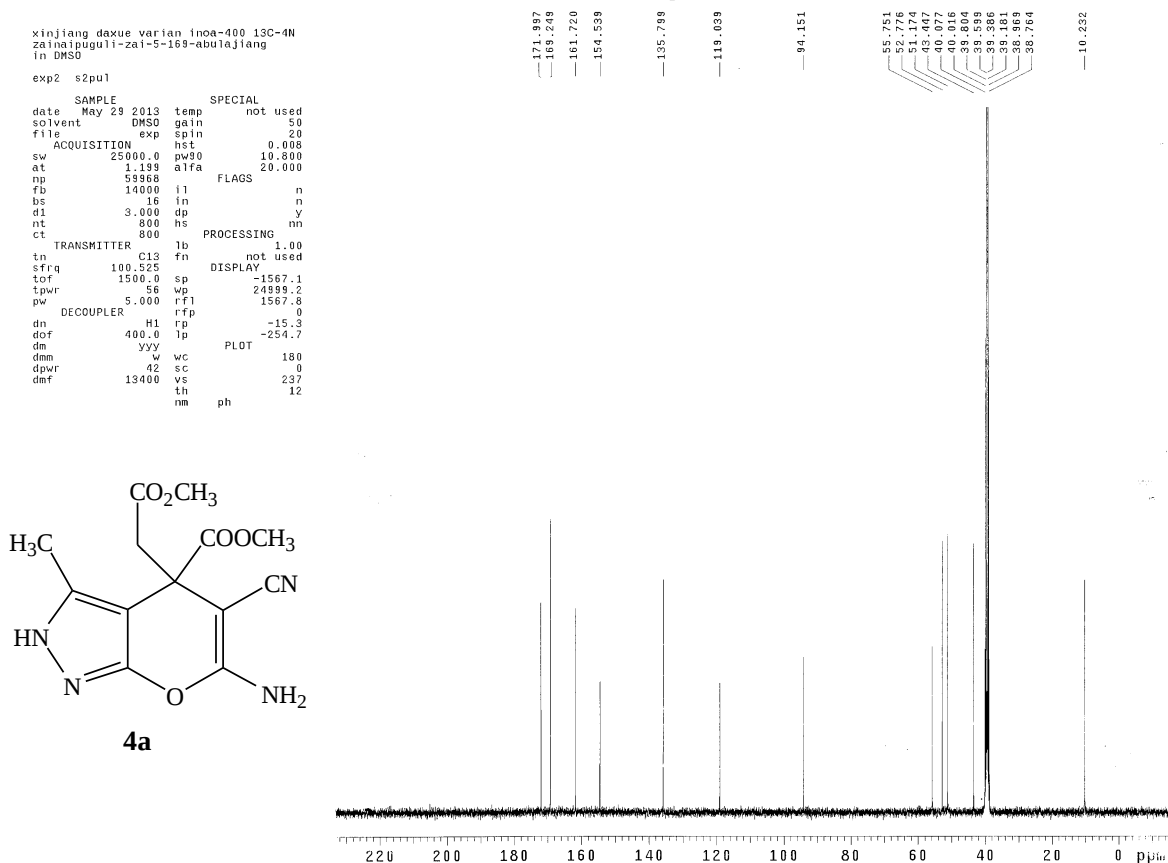
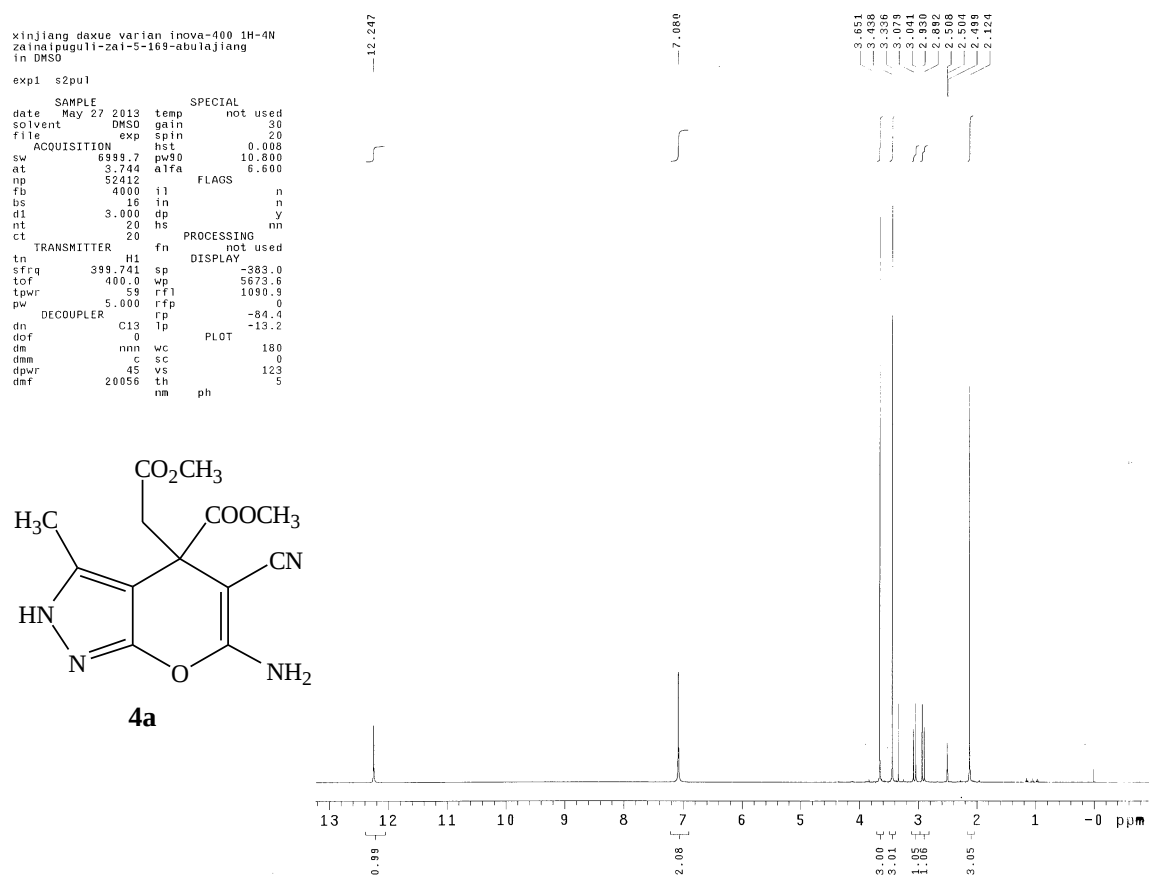


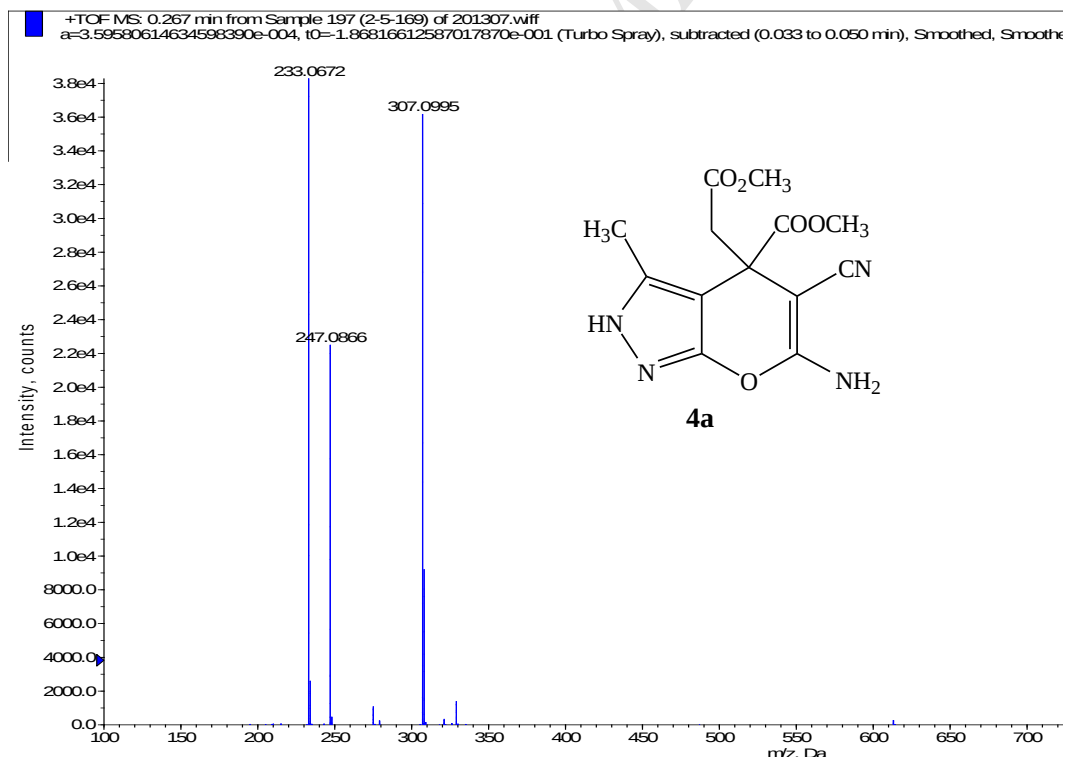
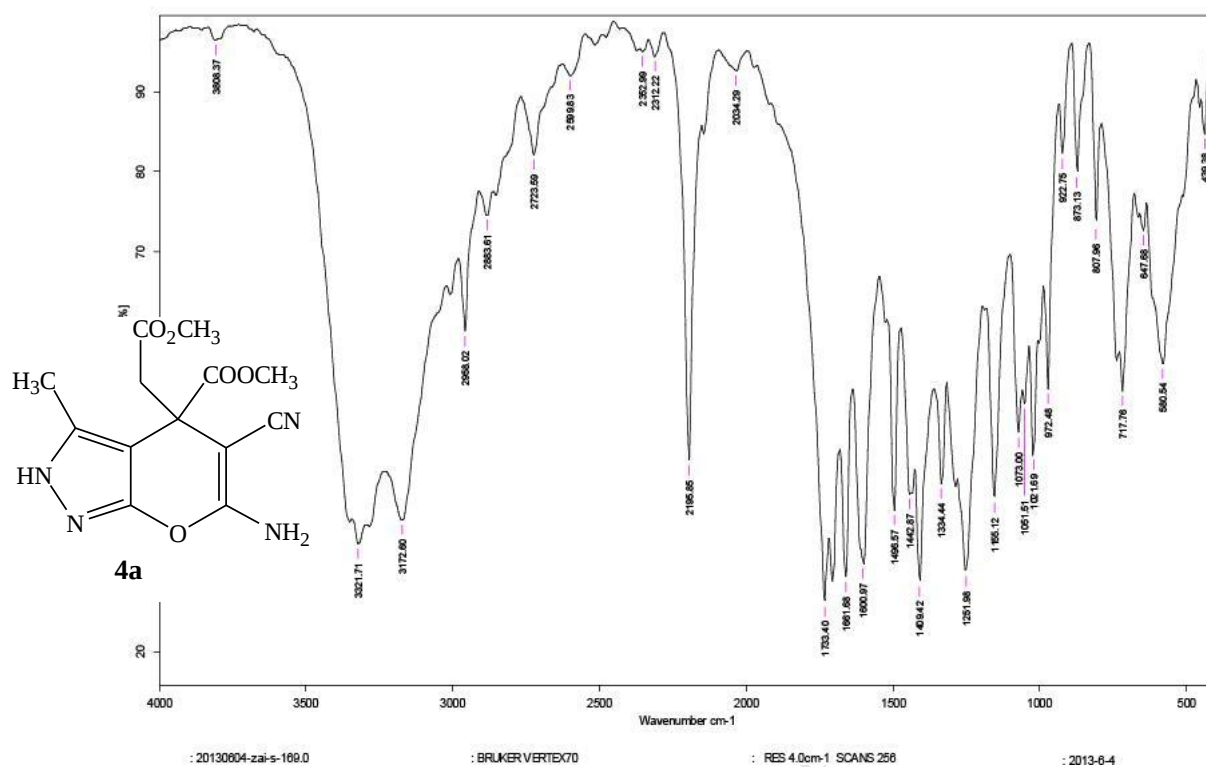
4.3.12. Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-(o-chlorophenyl)-3-propyl-2,4-dihydropyrano[2,3-c]pyrazole-4-carboxylate (4l). White crystals, mp: 188–190 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.70–7.52 (m, 4H, *o*-Cl-C₆H₄), 7.30 (s, 2H, NH₂), 3.72 (s, 3H, CH₃), 3.46 (s, 3H,

OCH₃), 3.13 (d, 1H, $J=15.6$ Hz, CH_{2a}), 2.96 (d, 1H, $J=15.6$ Hz, CH_{2b}), 2.14 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.6 (C₁), 169.2, 160.4, 145.2, 145.1, 133.7, 131.2, 130.7, 130.2, 129.7, 128.2, 118.2, 93.9, 56.7, 53.0, 51.3, 44.3, 13.1; IR (KBr) ν : 3449, 3315, 3172 (brs, NH₂), 2198 (C $\equiv$ N), 1736 (COO), 1664, 1582, 1538, 1436, 1399 (C=C, C=N), 1241, 1138, 1092, 1018 (C-O, C-N) cm⁻¹; ESI-MS m/z : 417 [M+H]⁺. HRMS (ESI) calcd for C₁₉H₁₈N₄O₅ [M+H]⁺ 417.0887, found 417.0890.



4.3.13. *Methyl 6-amino-5-cyano-4-(methoxycarbonylmethyl)-1-(p-chlorophenyl)-3-methyl-2,4-dihydropyrano[2,3-c]pyrazole-4-carboxylate (4m)*. White crystals; mp: 208–210 °C. ¹H NMR (400 MHz, DMSO-*d*₆): 7.80 (d, 2H, $J=9.2$ Hz, *p*-Cl-C₆H₄), 7.54 (d, 2H, $J=9.2$ Hz, *p*-Cl-C₆H₄), 7.52 (s, 2H, NH₂), 3.70 (s, 3H, CH₃), 3.46 (s, 3H, OCH₃), 3.12 (d, 1H, $J=15.6$ Hz, CH_{2a}), 2.98 (d, 1H, $J=15.6$ Hz, CH_{2b}), 2.14 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): 171.4 (C₁), 169.2, 160.3, 145.1, 144.1, 135.9, 130.5, 129.2, 121.2, 118.0, 95.7, 56.5, 53.1, 51.3, 43.9, 13.1; IR (KBr) ν : 3411, 3329, 3229 (brs, NH₂), 2955 (C-H), 2196 (C $\equiv$ N), 1722 (COO), 1659, 1591, 1523, 1393 (C=C, C=N), 1237, 1161, 1092, 1025 (C-O, C-N) cm⁻¹; ESI-MS m/z : 417 [M+H]⁺. HRMS (ESI) calcd for C₁₉H₁₈N₄O₅ [M+H]⁺ 417.0887, found 417.0967.

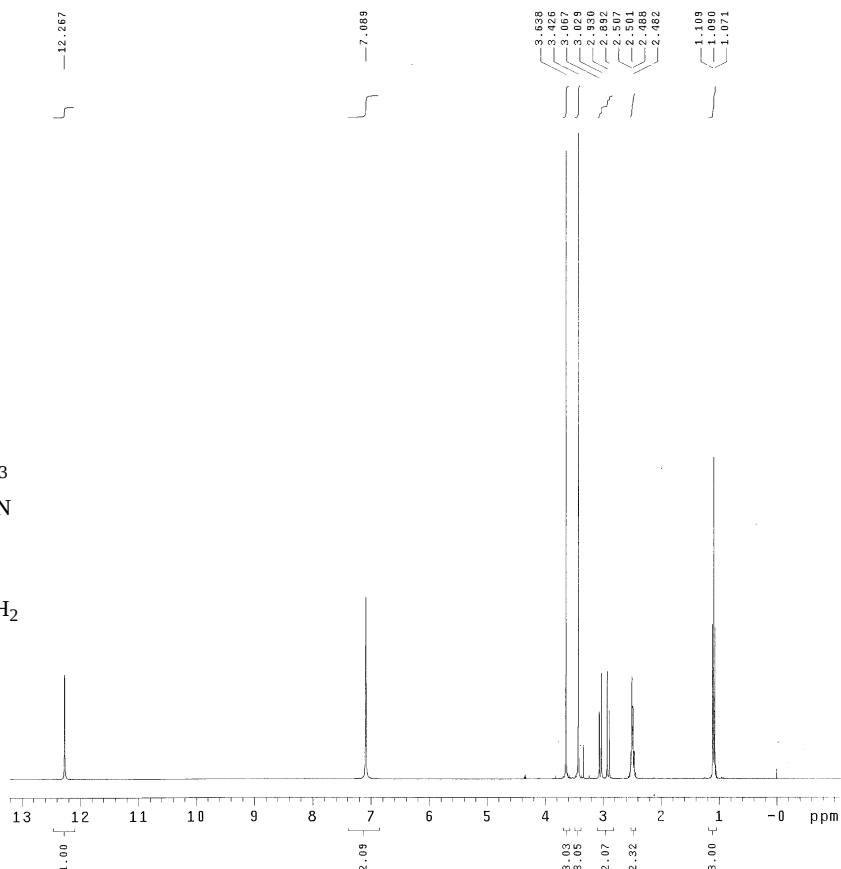
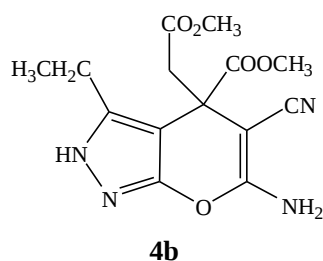
4. ^1H NMR, ^{13}C NMR, IR and MS spectra of compounds 4a-4m



xinjiang daxue varian inova-400 1H-4N
zai-5-166-abulajiang in DMSO

exp1 s2pu1

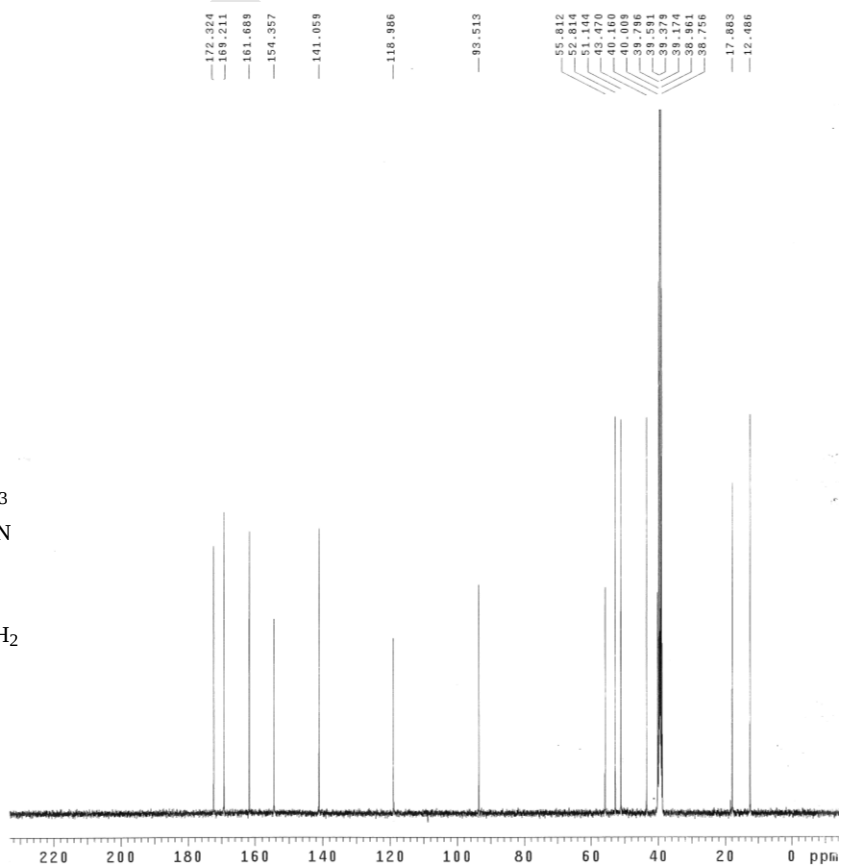
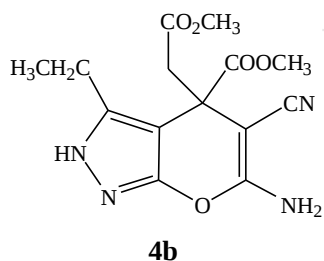
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file		spin	20
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at	3.744	alpha	6.600
np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	20	hs	nn
ct	20	PROCESSING	
TRANSMITTER		fn	not used
tn	H1	sp	DISPLAY
sfrq	399.742	wp	-466.7
tof	800.0	rfl	5752.2
tpwr	59	rfl	691.5
pw	5.000	rfl	0
DECOUPLER		rp	-80.3
dn	C13	lp	-14.1
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dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	137
dmf	20056	th	9
	nm	ph	

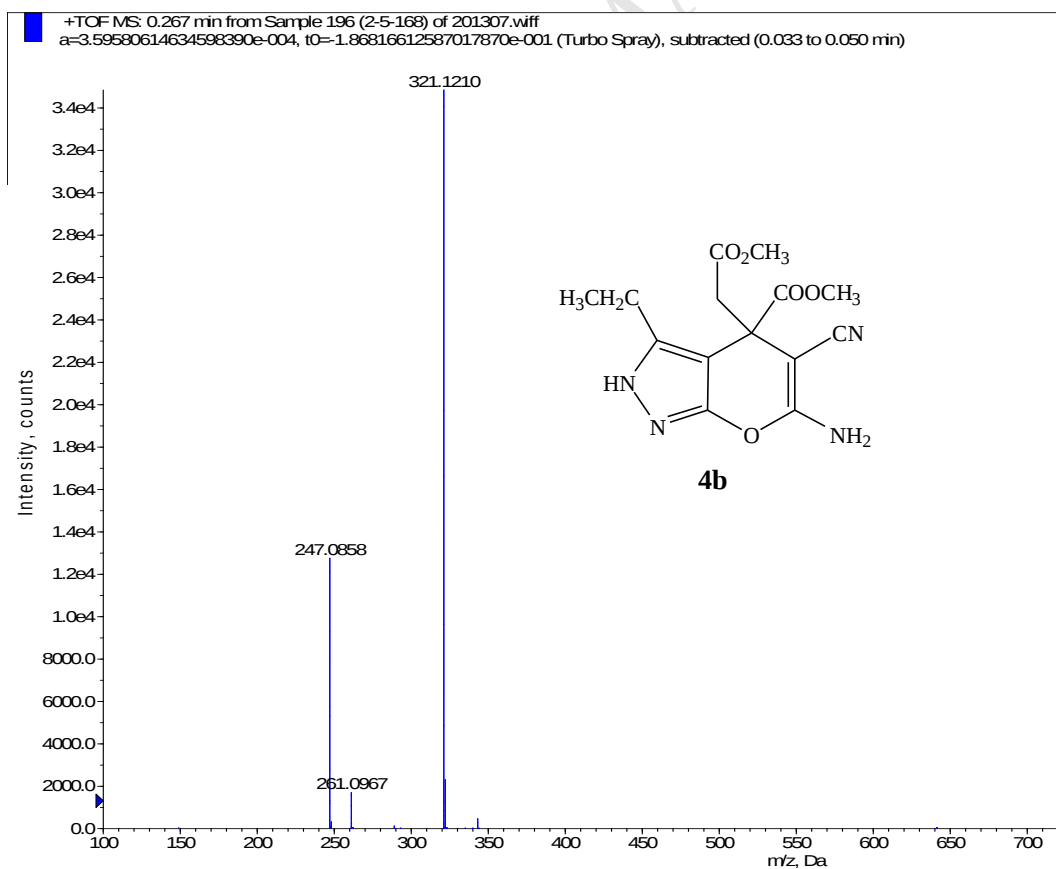
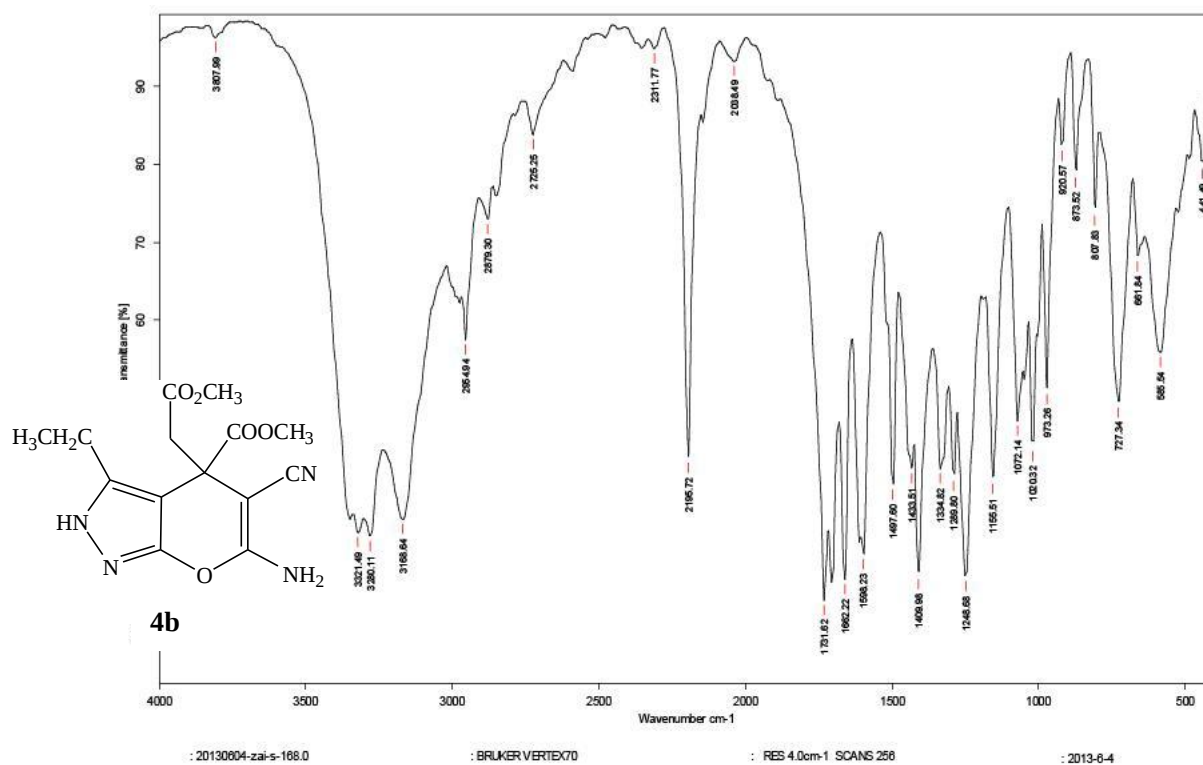


xinjiang daxue varian inova-400 13C-4N
zainaipuguli-zai-5-166-abulajiang
in DMSO

exp2 s2pu1

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np	59968	FLAGS	
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	2000	hs	nn
ct	2000	PROCESSING	
TRANSMITTER		fn	not used
tn	C13	sp	DISPLAY
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tof	1500.0	rfl	2499.2
tpwr	56	rfl	1567.8
pw	5.000	rfl	0
DECOUPLER		rp	-21.6
dn	H1	lp	-261.4
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dpwr	13400	vs	263
dmf		th	12
	nm	ph	

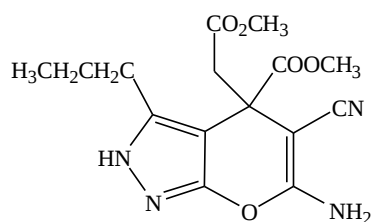




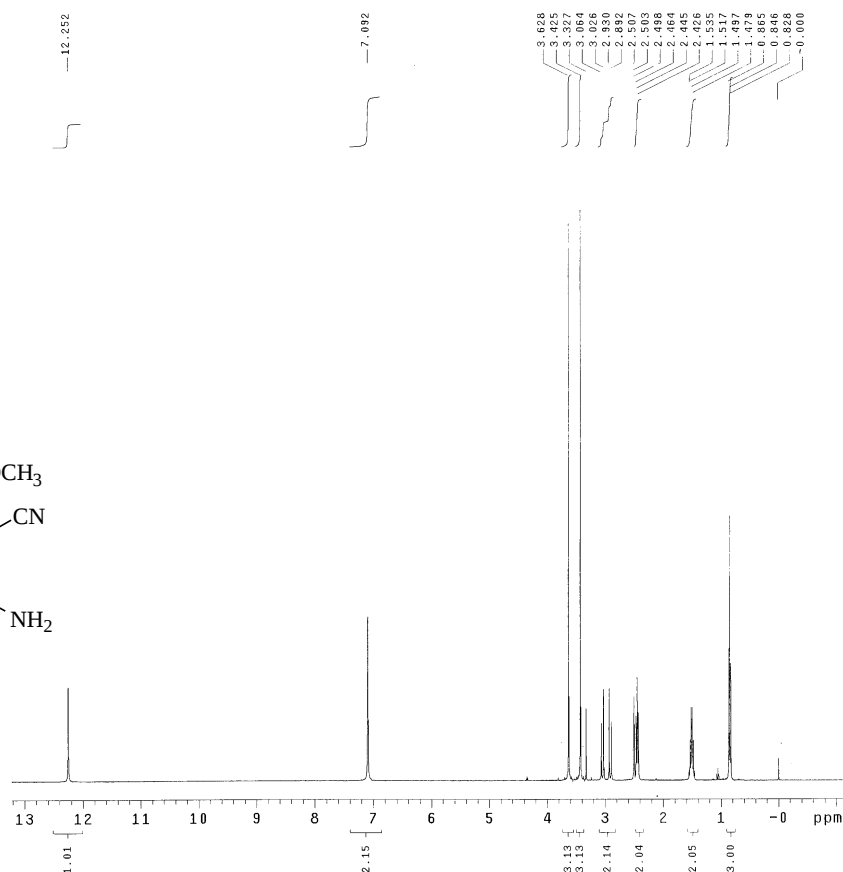
xinjiang daxue varian inova-400 1H-4N
zai-5-167-abulajiang in DMSO

exp1 s2pu1

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file	exp	spin	20
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np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	20	hs	nn
ct	20	PROCESSING	
TRANSMITTER		fn	not used
tn	H1	DISPLAY	
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tof	800.0	wp	5763.5
tpwr	59	rf1	691.5
pw	5.000	rfp	0
DECOUPLER		rp	-83.6
dn	C13	lp	-14.1
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dm	nnn	wc	180
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	nm	ph	



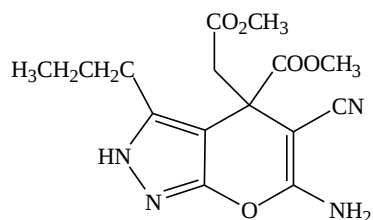
4c



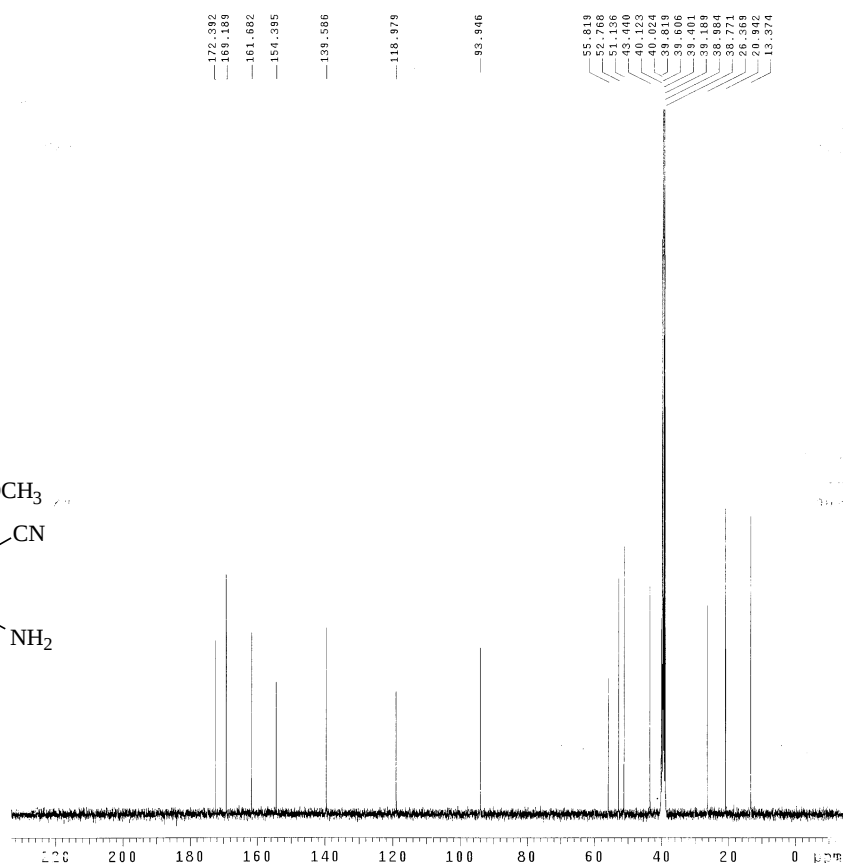
xinjiang daxue varian inova-400 13C-4N
zainatipuguli-zai-5-167-abulajiang
in DMSO

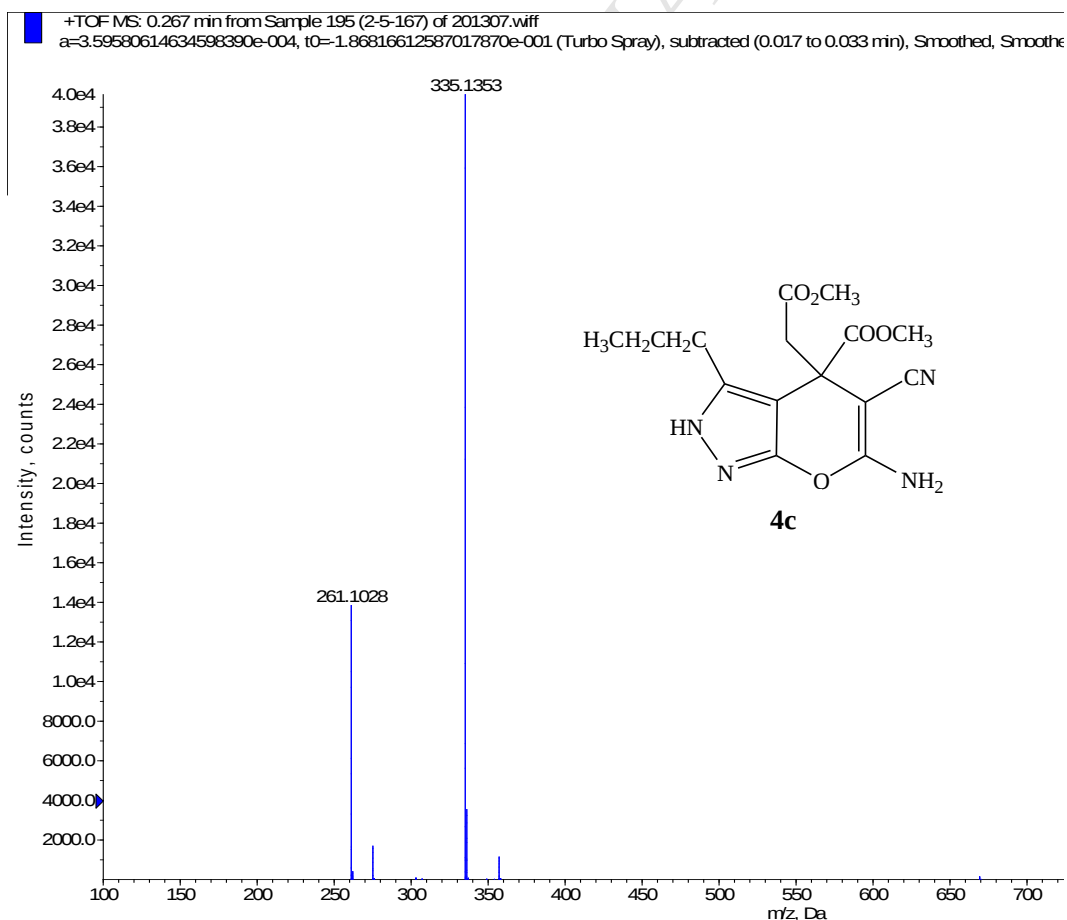
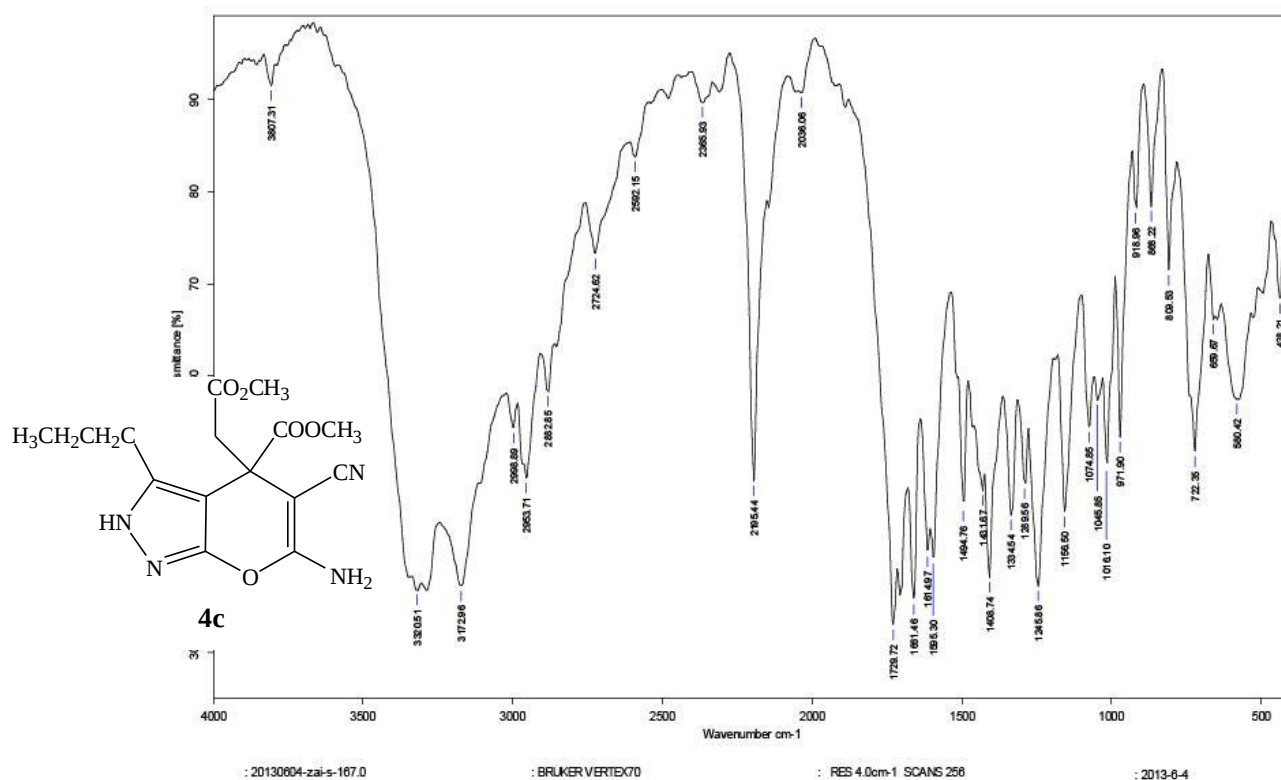
exp2 s2pu1

SAMPLE		SPECIAL	
date	May 28 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	0.008
sw	25000.0	pw90	10.800
at	1.139	ai1a	20.000
np	59968	FLAGS	
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	1000	hs	nn
ct	1000	PROCESSING	
TRANSMITTER		lb	1.00
tn	C13	fn	not used
sfrq	100.525	DISPLAY	
tof	1500.0	sp	-1567.1
tpwr	56	wp	24999.2
pw	5.000	rf1	1567.8
DECOUPLER		rfp	0
dn	H1	rp	-27.9
dof	400.0	lp	-261.4
dm	yyy	PLOT	
dmm	w	wc	180
dpwr	42	sc	0
dmf	13400	vs	284
	nm	ph	12



4c

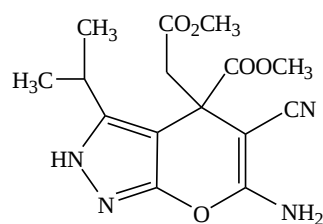




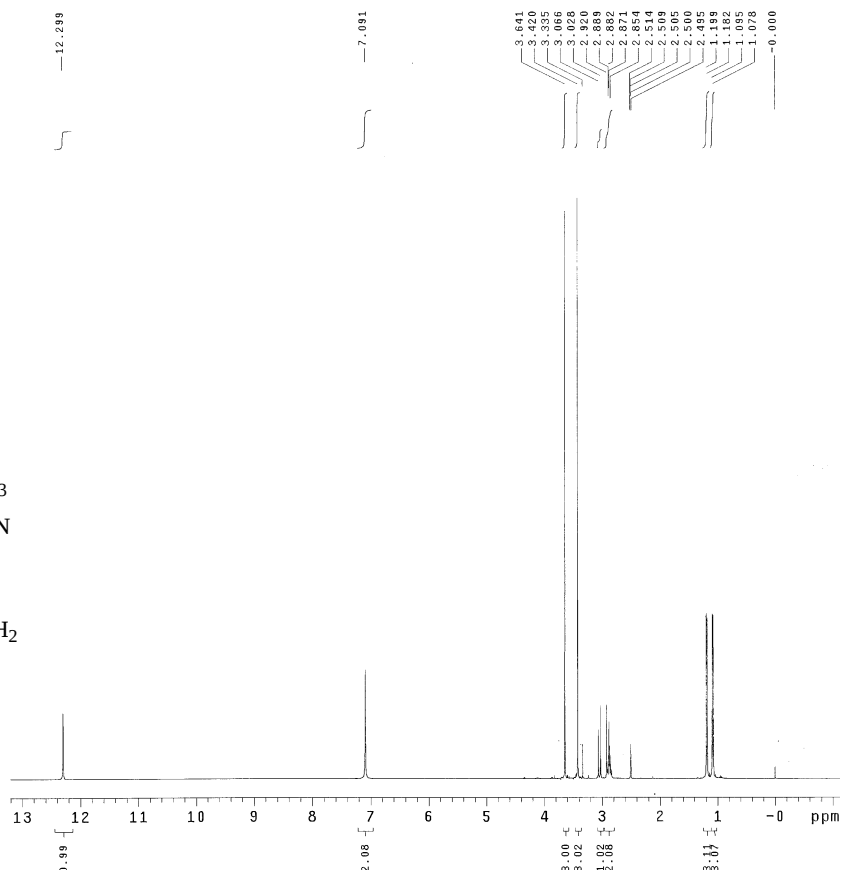
xinjiang daxue varian inova-400 1H-4N
zai-5-166-abulajiang in DMSO

exp1 s2pul

SAMPLE		SPECIAL	
date	May 24 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		SPECIAL	
sw	6999.7	hst	0.008
at	3.744	pw90	10.800
np	52412	alpha	6.600
fb	4000	flags	n
bs	16	in	n
d1	3.000	dp	y
nt	20	hs	nn
ct	20		
TRANSMITTER		PROCESSING	
tn	H1	fn	not used
sfrq	399.741	sp	-483.8
tof	400.0	wp	5763.5
tpwr	5.9	rfl	1090.7
pw	5.000	rff	0
dn	C13	rp	-87.0
dof	0	lp	-9.5
dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	124
dmf	20056	th	2
	nm	ph	



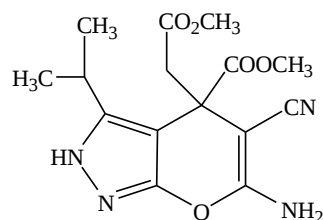
4d



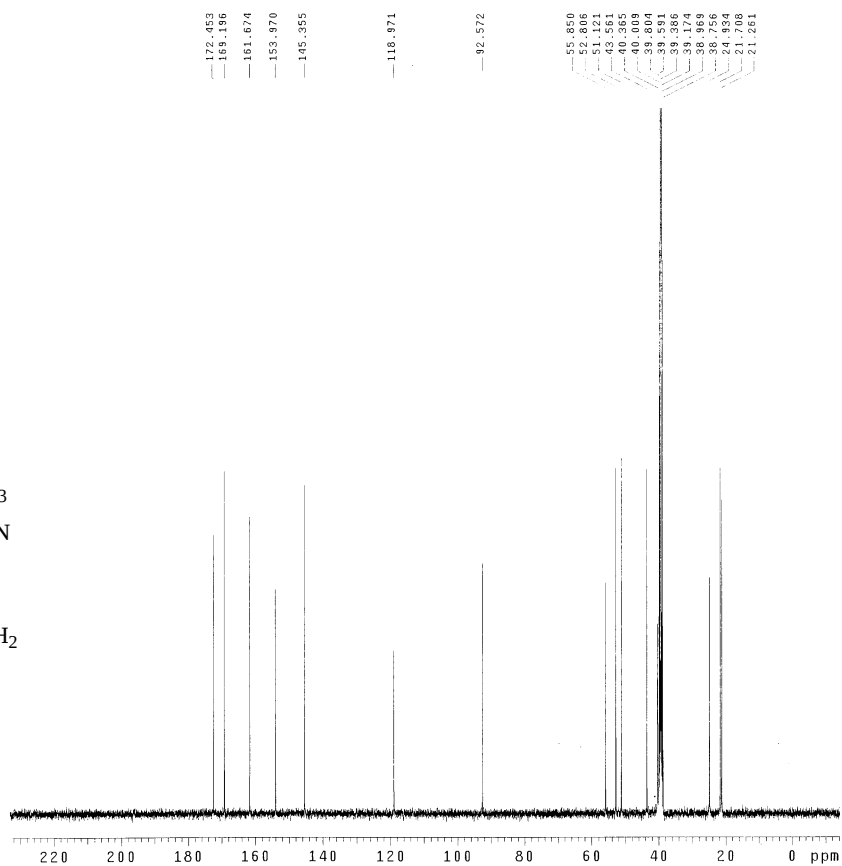
xinjiang daxue varian inova-400 13C-4N
zai-5-166-abulajiang in DMSO

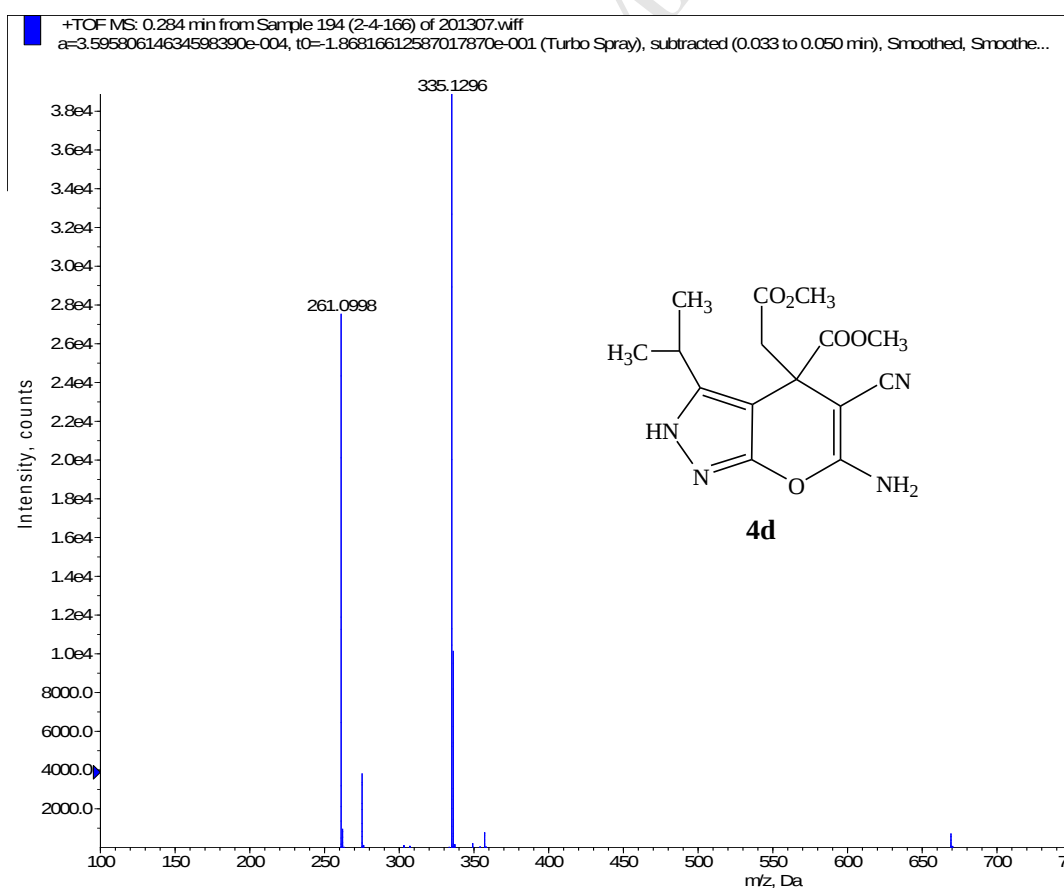
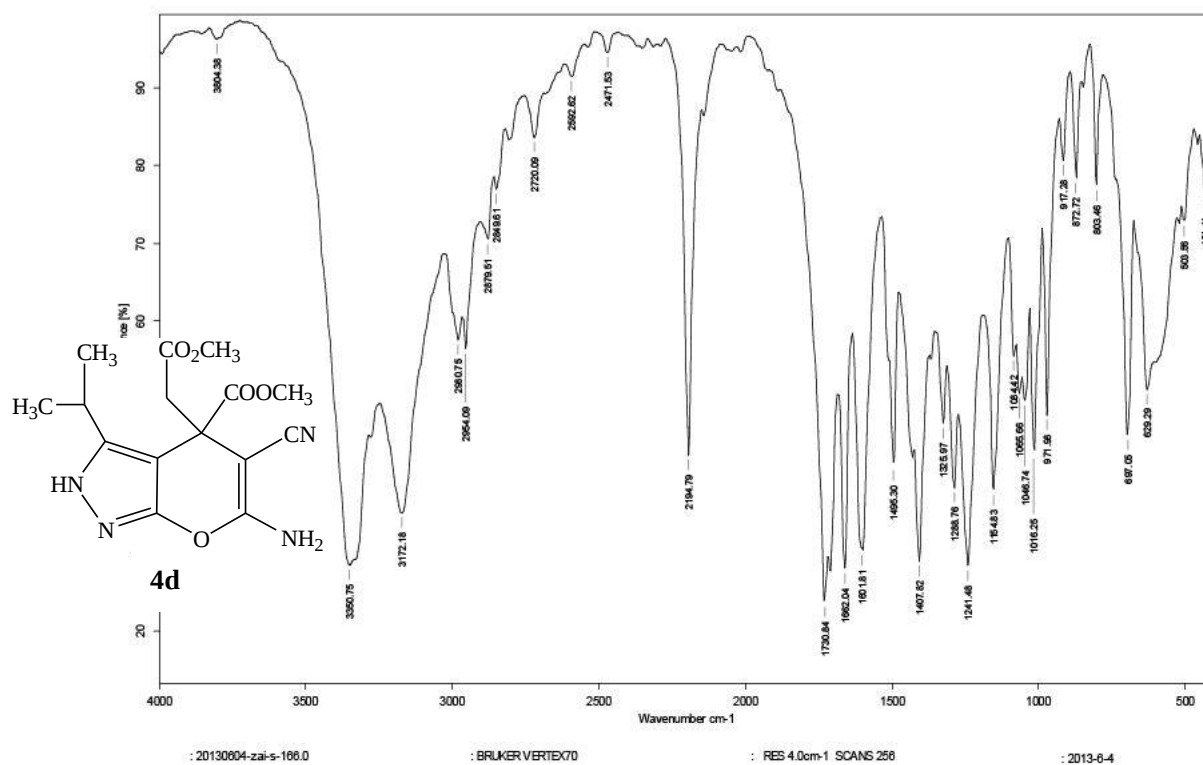
exp2 s2pul

SAMPLE		SPECIAL	
date	May 28 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		SPECIAL	
sw	25000.0	hst	0.008
at	1.189	pw90	10.800
np	59968	alpha	20.000
fb	14000	flags	n
bs	16	in	n
d1	3.000	dp	y
nt	1000	hs	nn
ct	1000		
TRANSMITTER		PROCESSING	
tn	C13	lb	1.00
sfrq	100.525	fn	not used
tof	1500.0	sp	-1567.1
tpwr	56	wp	24999.2
pw	5.000	rfl	1567.8
dn	H1	rp	-8.5
dof	400.0	lp	-261.4
dm	yyy	wc	180
dmm	w	sc	0
dpwr	42	vs	260
dmf	13400	th	12
	nm	ph	



4d

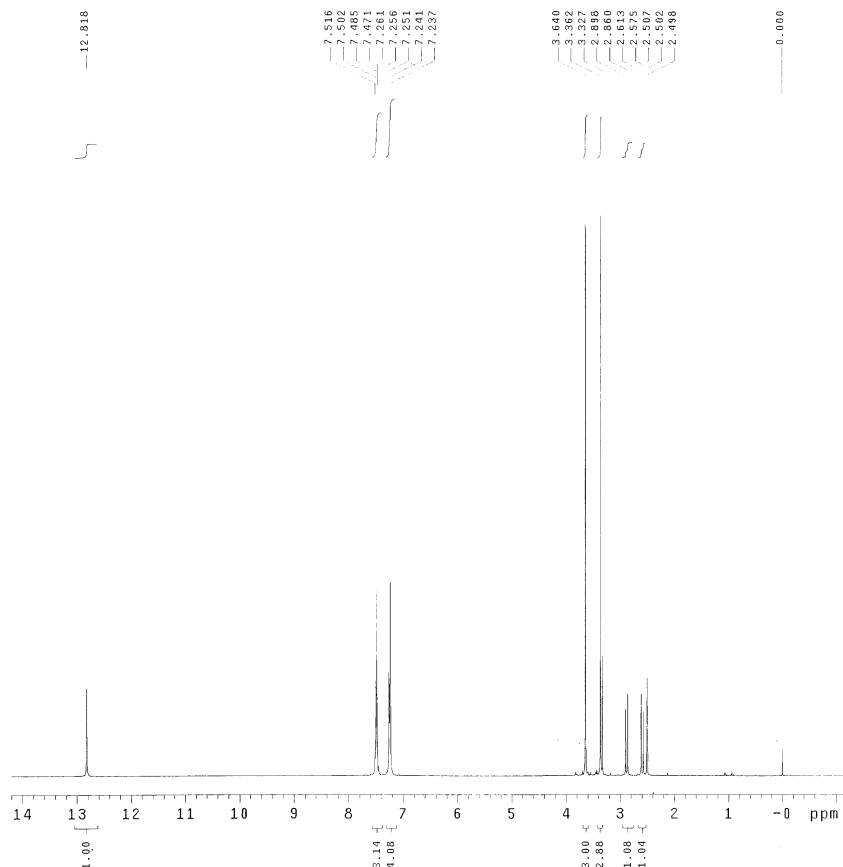
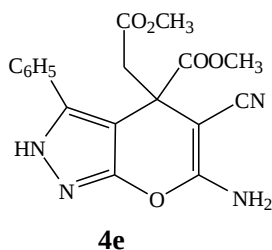




xinjiang daxue varian inova-400 1H-4N
zainaipuguli-zai-6-192-abulajiang
in DMSO

exp1 s2pu1

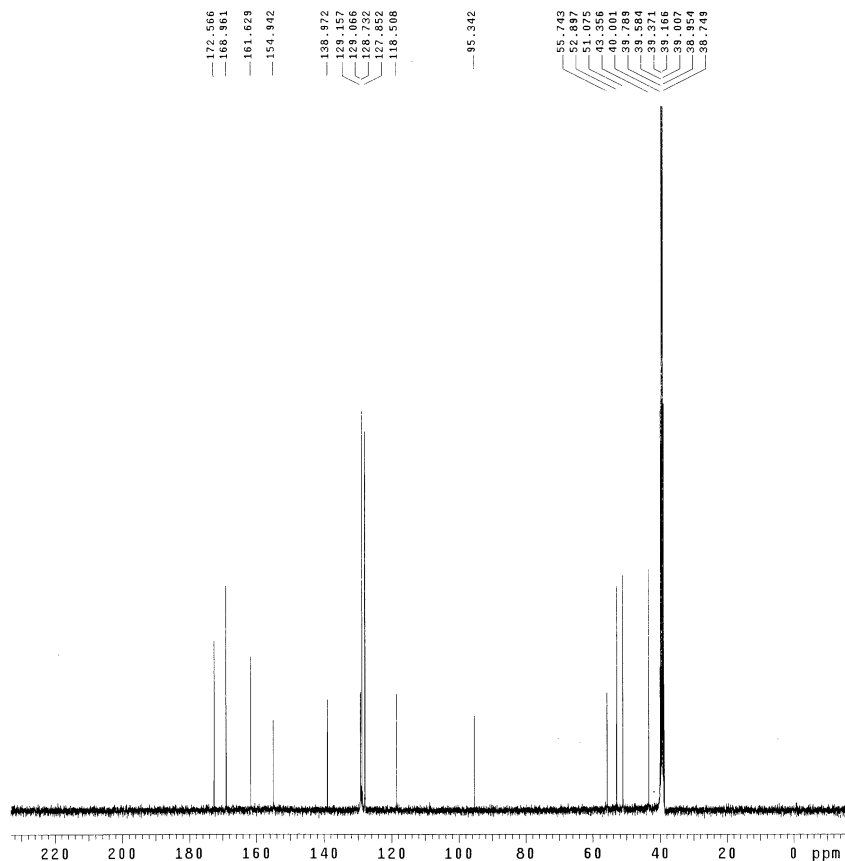
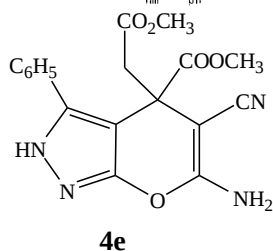
SAMPLE		SPECIAL	
date	Jun 14 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		SPECIAL	
sw	699.7	hst	0.008
at	3.744	pw90	10.800
np	52412	alpha	6.600
fb	4000	FLAGS	
bs	16	il	n
d1	3.000	in	n
nt	40	dp	y
ct	40	hs	nn
TRANSMITTER		PROCESSING	
tn	H1	fn	not used
sfrq	399.742	sp	DISPLAY
tof	800.0	wp	-456.3
tpwr	59	rfl	6132.9
pw	5.000	rfl	692.8
DECOUPLER		rp	0
dn	C13	lp	-83.7
dof	0	PLT	-11.1
dm	nmn	vc	180
dmm	c	sc	0
dpwr	45	vs	120
dmf	20058	th	4
	nm	ph	

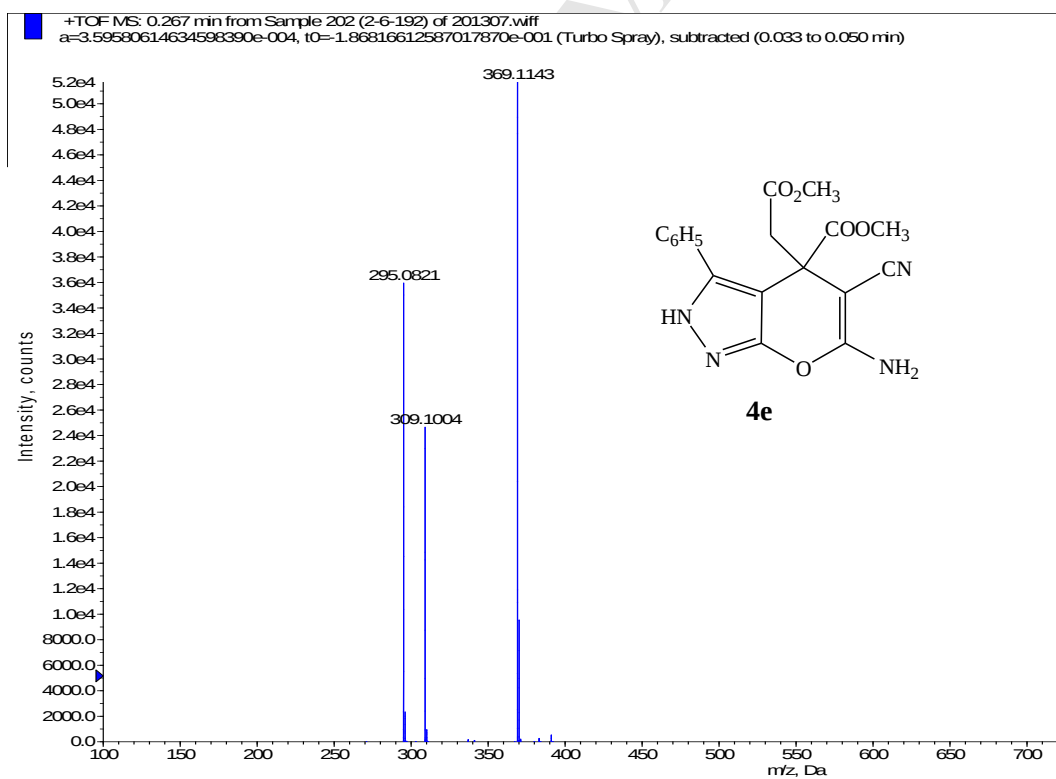
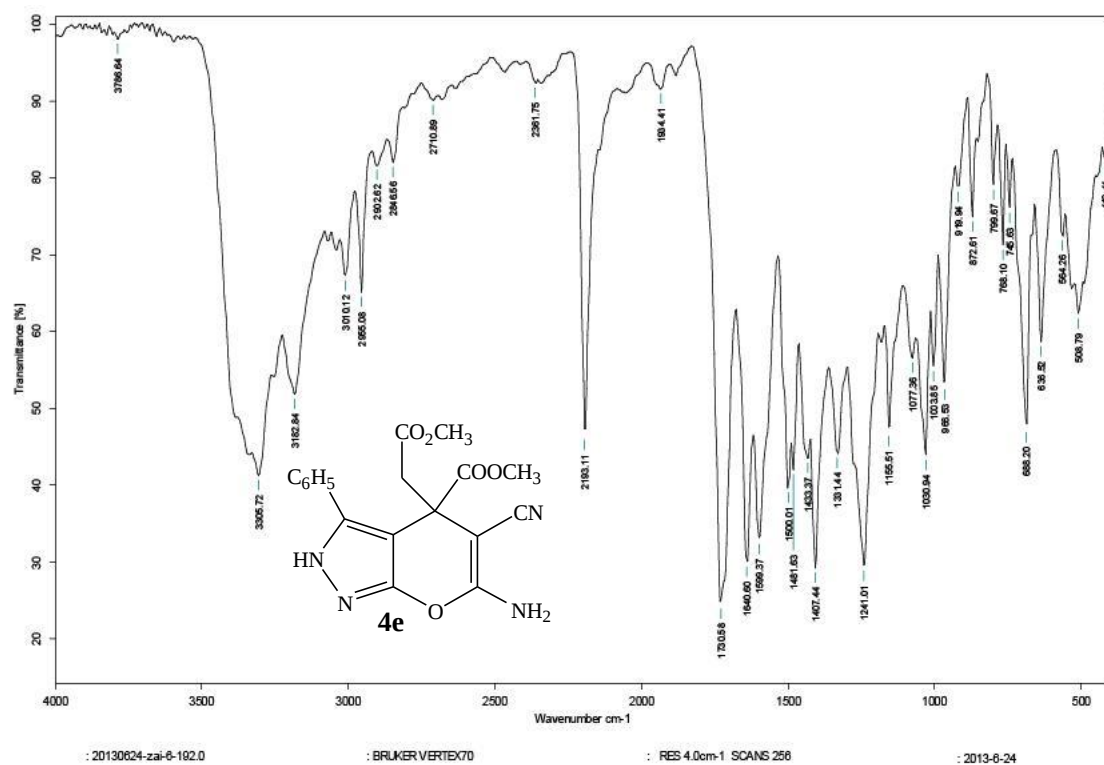


xinjiang daxue varian inova-400 13C-4N
zainaipuguli-zai-6-192-abulajiang
in DMSO

exp2 s2pu1

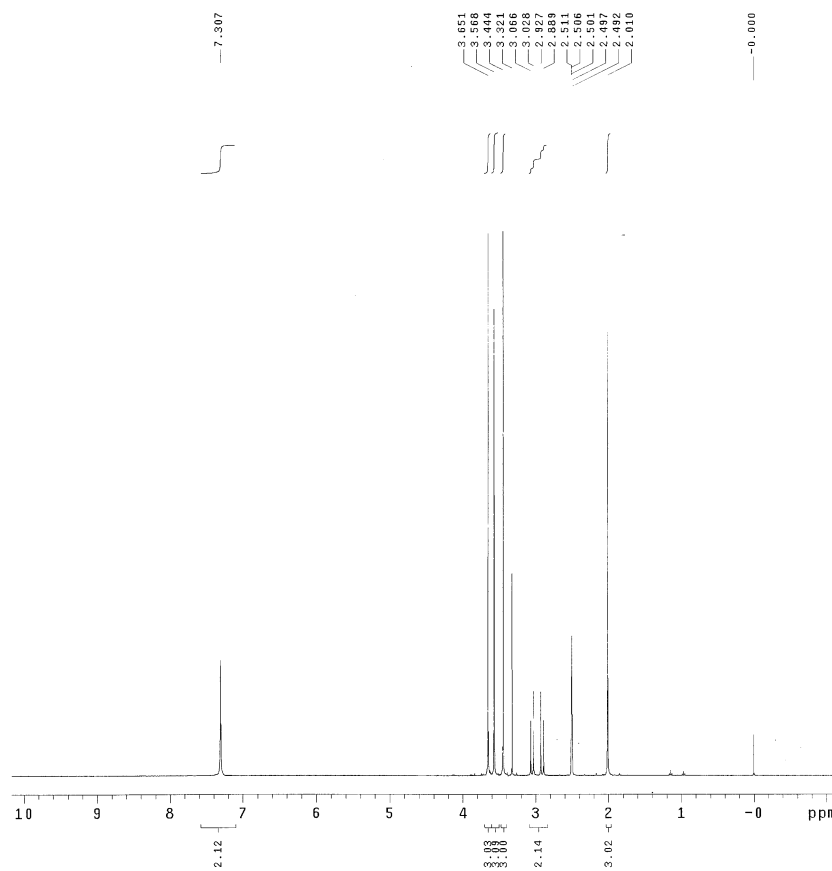
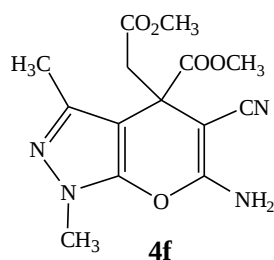
SAMPLE		SPECIAL	
date	Jul 18 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		SPECIAL	
sw	25000.0	hst	0.008
at	1.195	pw90	10.800
np	59368	alpha	20.000
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	600	hs	nn
ct	600	PROCESSING	1.00
TRANSMITTER		PROCESSING	
tn	C13	fn	not used
sfrq	100.525	sp	DISPLAY
tof	1500.0	wp	-1567.1
tpwr	56	rfl	24995.2
pw	5.000	rfl	1567.8
DECOUPLER	H1	rp	0
dn	yyy	lp	-59.2
dof	400.0	PLT	-266.7
dm	w	vc	180
dmm	42	sc	0
dpwr	13400	vs	201
dmf		th	14
	nm	ph	





exp1 s2pu1

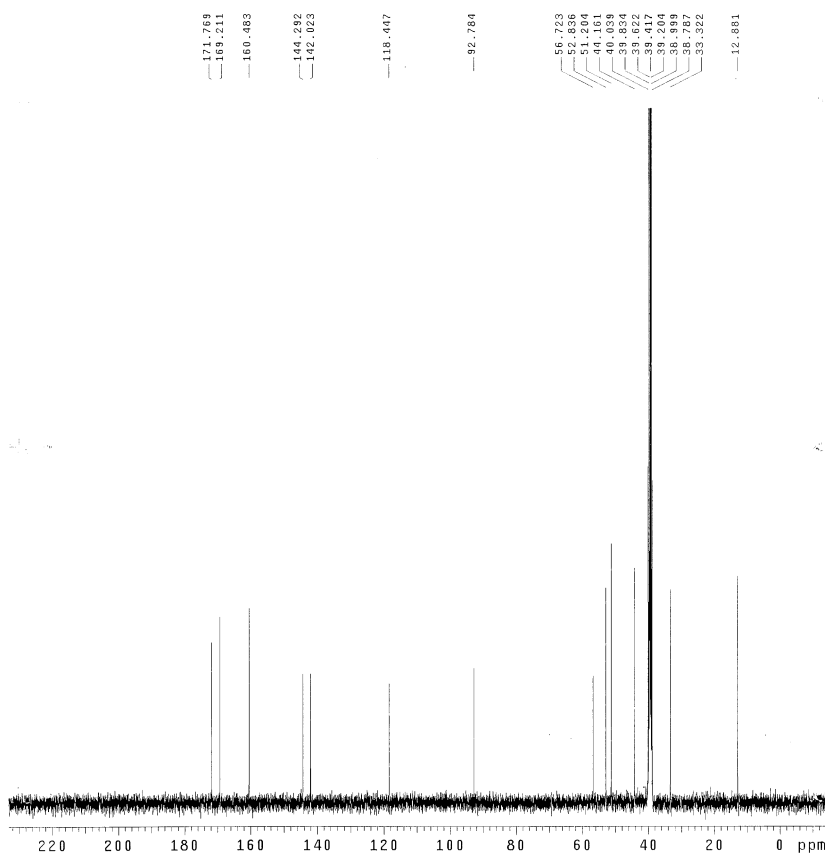
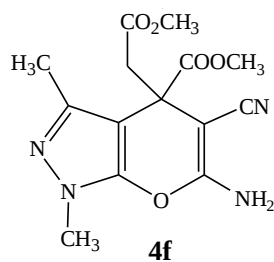
SAMPLE				SPECTRAL		
date	May	24	2013	temp	not	used
solvent			DMSO	gain		30
file			exp	spin	hst	0.008
ACQUISITION						6.600
sw			6999.7	pw90		10.800
at			3.744	alpha		
fb			5240.12		FLAGS	
bs			6	in		
ct			3.000	dp		y
dl			20	hs		
TRANSMITTER				PROCESSING		
sn			H1	fn	not	used
tfreq			393.741	spn	DISPLAY	-484.9
tpwr			400.0	wp		4550.2
tpwr			59	rfp		1091.8
pn				rfp		0
deCOUPLER				rp		-88.8
dm			C13	lp		-6.5
dof						
dm			nmh	wc	PLOT	180
dm				c		
dmf			45	vs		116
dmf			20056	th		
			nm	ph		



xinjiang daxue varian inova-400 1H-4N
zainaipuguli-zai-5-189-abulajiang
in DMSO

exp2 s2pu1

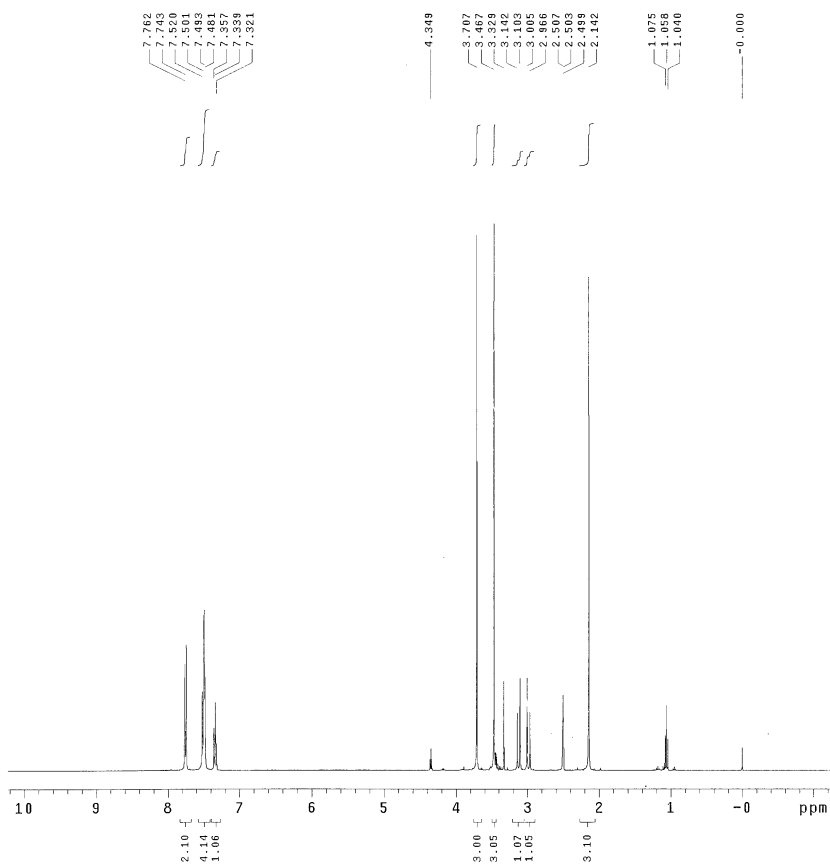
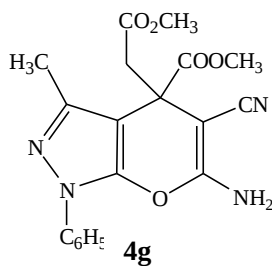
SAMPLE				SPECIAL	
date	May 29	2013	temp	not	used
solvent	DMSO		gain		50
file	exp		file		20
ACQUISITION				rest	0.008
sw	25000.0		pw80		10.800
at	1.199		alpha		20.000
fb	59868			FLAGS	
bs	14000	il			n
dt	16	in	in		n
nt	1000	dp	dp		y
ct	1000	hs			n
TRANSMITTER				PROCESSING	1.00
tn	C13	fn			not used
tsrfq	100.525	sp		DISPLAY	
tpwr	1500.0				-1567.1
pw	5.000	rf1			24999.2
DECOUPLER					1567.8
dn	H1	rf			0
df	400.0	lp			-6.2
dm	yyf			PLOT	-241.1
dmm	w	wc			180
dpr	42	sc			180
dmf	13400				498
	th				
	nm	ph			



xinjiang daxue varian inova-400 1H-4N
zainalipuguli-1-zai-4-160-abulajiang
in DMSO

exp1 s2pu1

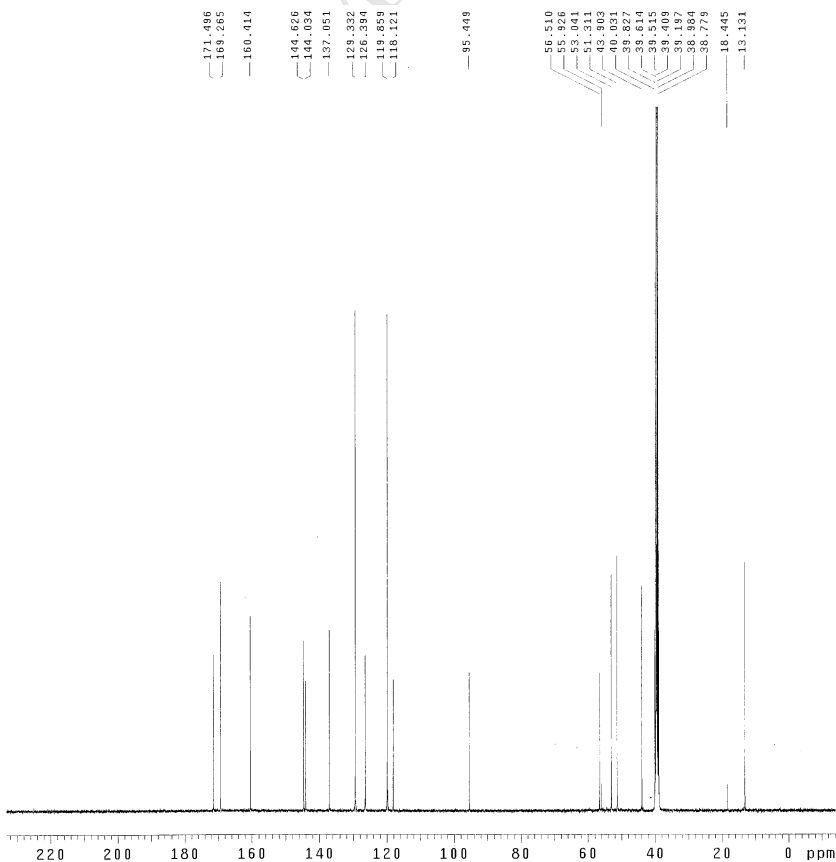
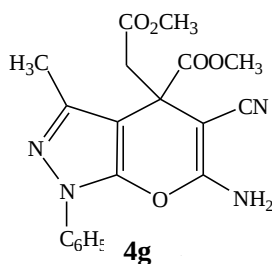
SAMPLE		SPECIAL	
date	May 2 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		hst	0.008
sw	6999.7	pw90	10.800
at	3.744	alfa	6.600
np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
dl	3.000	dp	y
nt	40	hs	nn
ct	40	PROCESSING	
TRANSMITTER		fn	not used
tn	H1	DISPLAY	
sfrq	399.741	sp	-506.9
tof	400.0	wp	4595.1
tpwr	59	rfl	1091.1
pw	5.000	rfl	0
DECOUPLER		rp	-116.4
dn	C13	lp	-14.5
dof	0	PLOT	
dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	118
dmf	20056	th	4
	nm	ph	

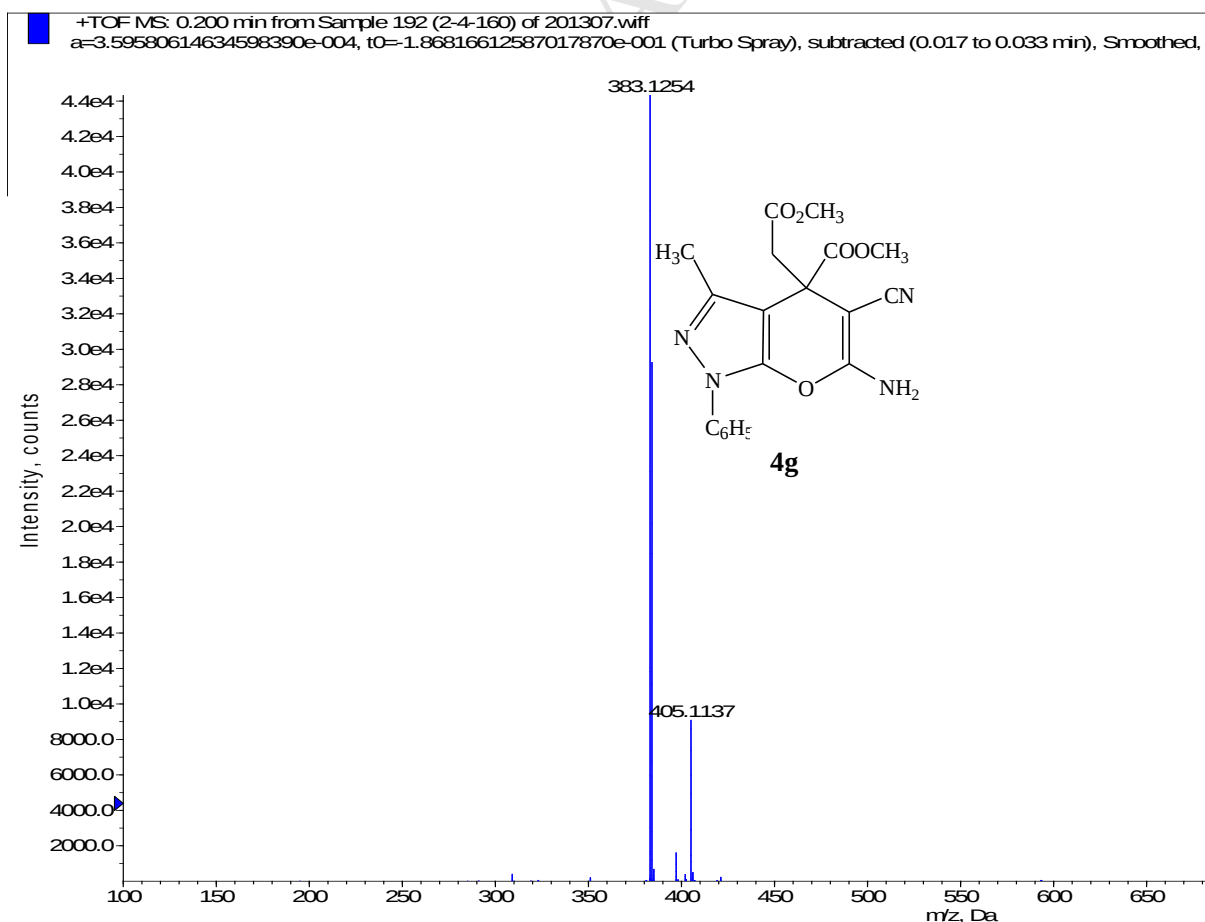
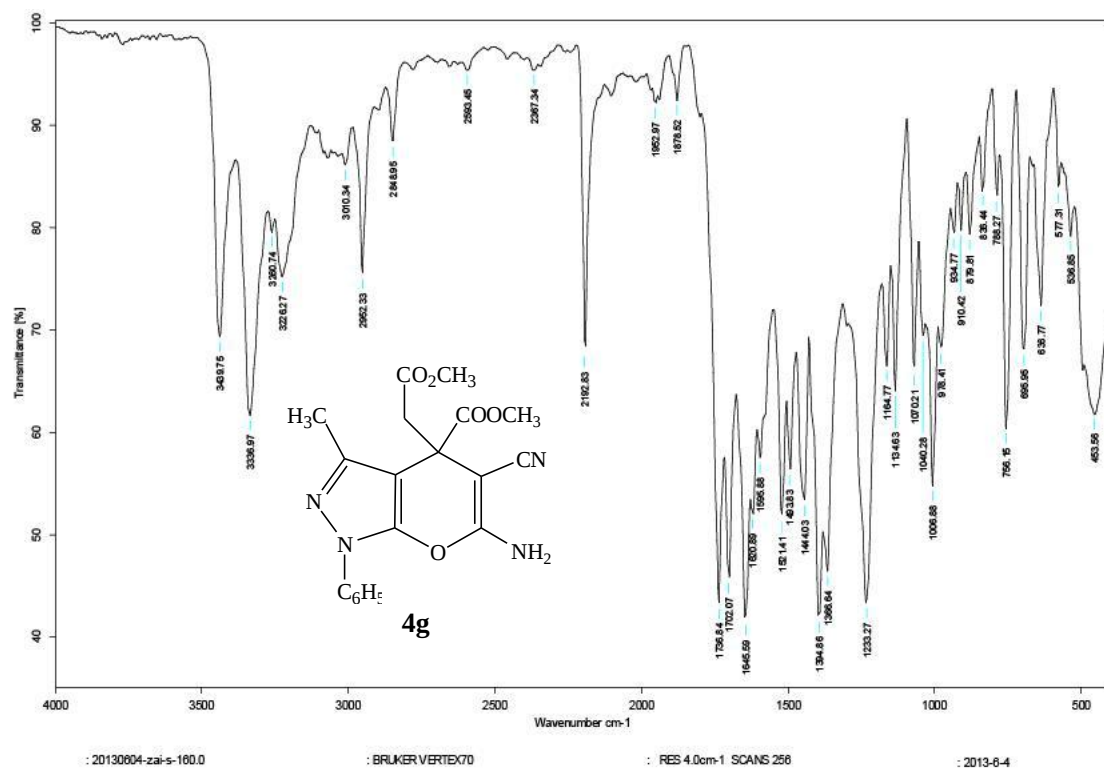


xinjiang daxue varian inova-400 13C-4N
zainalipuguli-1-zai-5-160-abulajiang
in DMSO

exp2 s2pu1

SAMPLE		SPECIAL	
date	May 28 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	0.008
sw	25000.0	pw90	10.800
at	1.139	alfa	20.000
np	59968	FLAGS	
fb	14000	il	n
bs	16	in	n
dl	3.000	dp	y
nt	8800	hs	nn
ct	8800	PROCESSING	
TRANSMITTER		lb	1.00
tn	C13	fn	not used
sfrq	100.525	DISPLAY	
tof	1500.0	sp	-1567.1
tpwr	56	wp	24999.2
pw	5.000	rfl	1567.8
DECOUPLER		rfl	0
dn	H1	rp	-40.4
dof	400.0	lp	-261.4
dm	yry	wc	180
dmm	42	sc	0
dpwr	13400	vs	269
dmf		th	4
	nm	ph	

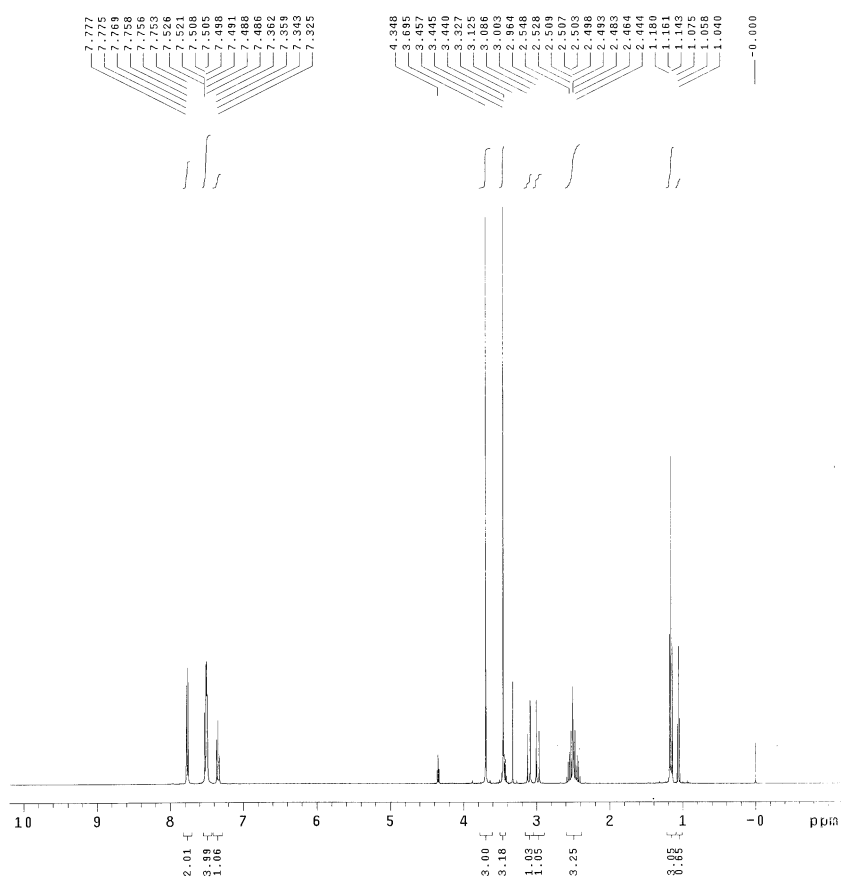
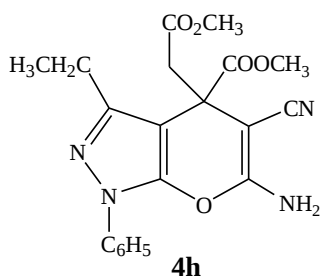




xinjiang daxue varian inova-400 1H-4N
zai-5-174-abulajiang in DMSO

exp1 s2pu1

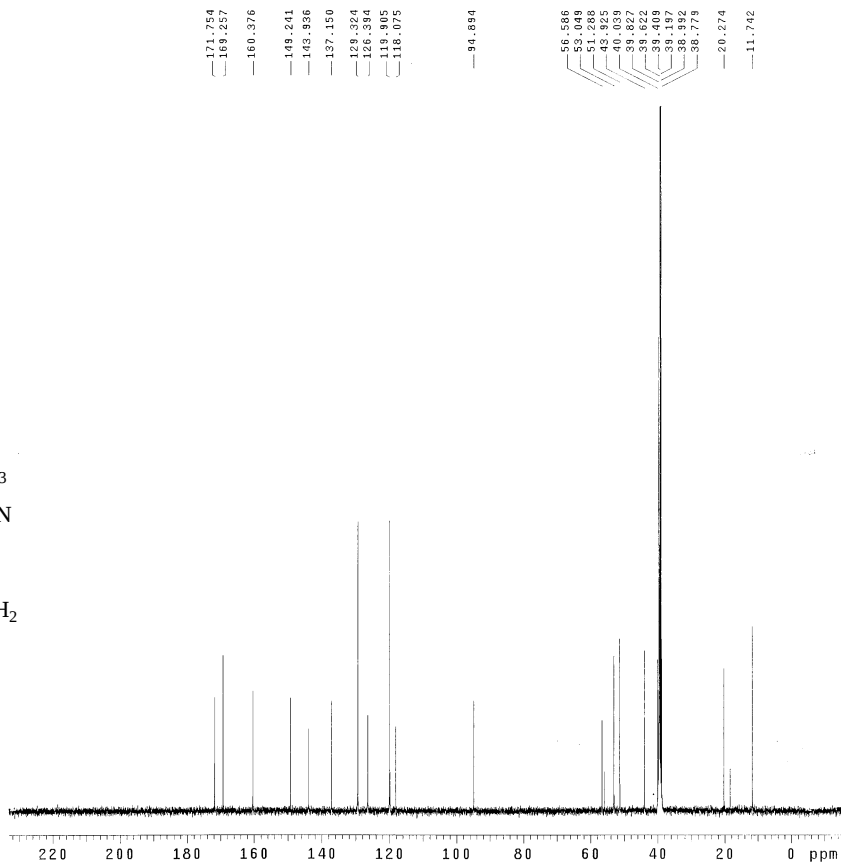
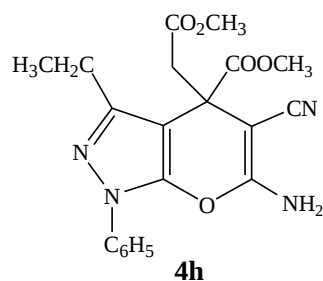
SAMPLE		SPECIAL	
date	May 24 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		hst	0.008
sw	6399.7	pw90	10.800
at	3.744	alpha	6.600
np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	20	hs	nn
ct	20	PROCESSING	
TRANSMITTER		fn	not used
tn	H1	DISPLAY	
sfreq	399.741	sp	-484.5
tof	400.0	wp	4561.3
tpwr	59	rf1	1091.4
pw	5.000	rpf	0
DECOUPLER		rp	-85.1
dn	C13	lp	-15.7
dof	0	PLOT	
dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	123
dmf	20056	th	5
	nm	ph	

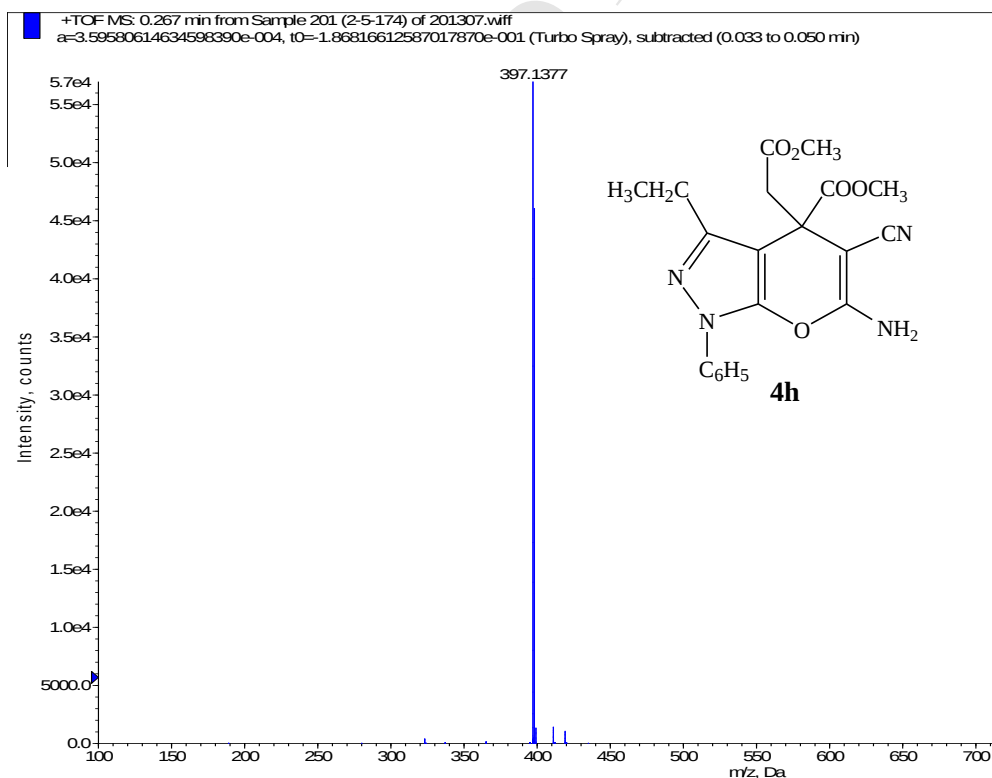
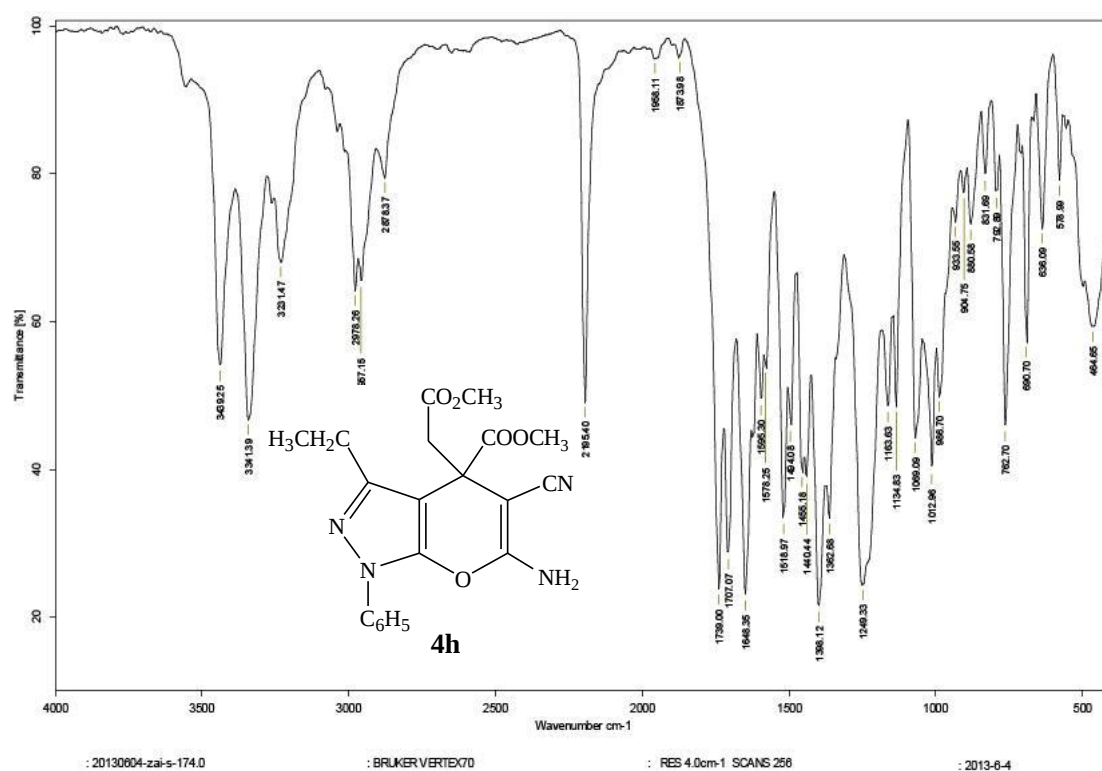


xinjiang daxue varian inova-400 13C-4N
zainai-puguli-zai-5-174-abulajiang
in DMSO

exp2 s2pu1

SAMPLE		SPECIAL	
date	May 29 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	0.008
sw	25000.0	pw90	10.800
at	1.199	alpha	20.000
np	58968	FLAGS	
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	1000	hs	nn
ct	1000	PROCESSING	
TRANSMITTER		fb	1.00
tn	C13	fn	not used
sfreq	100.525	DISPLAY	
tof	1500.0	sp	-1567.1
tpwr	58	wp	24999.2
pw	5.000	rf1	1567.0
DECOUPLER		rp	0
dn	H1	lp	-6.5
dof	400.0	PLOT	
dm	yyy	wc	180
dmm	w	sc	0
dpwr	42	vs	237
dmf	13400	th	12
	nm	ph	

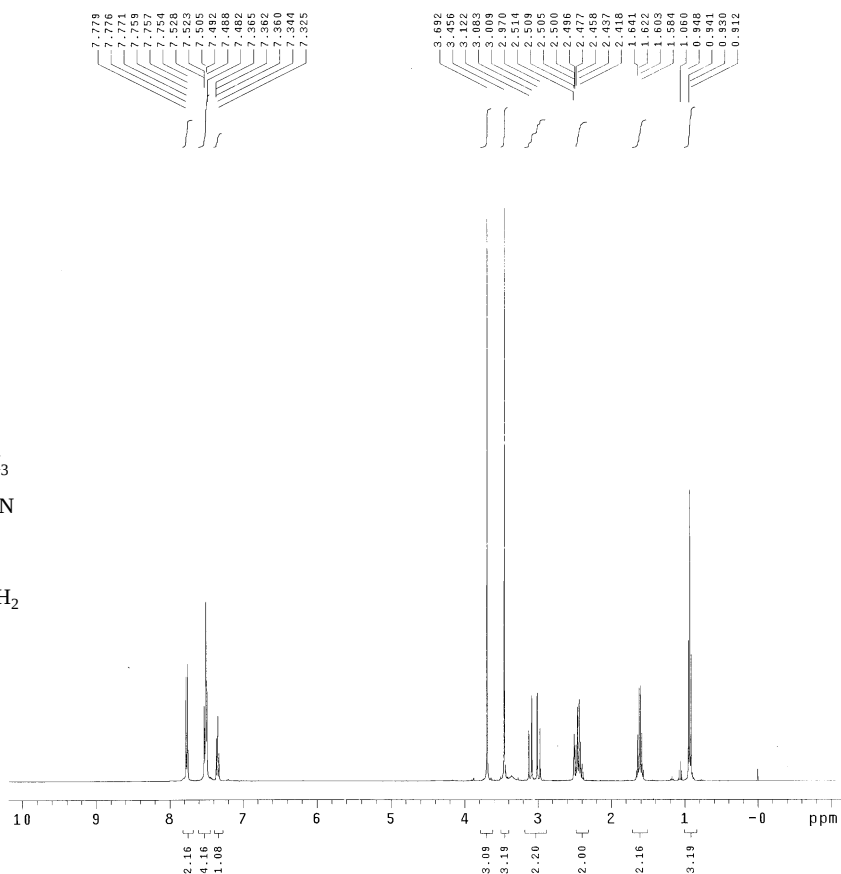
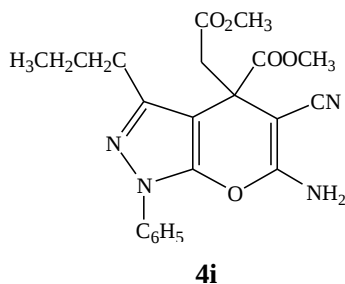




xinjiang daxue varian inova-400 1H-4N
zai-5-173-abulajiang in DMSO

exp1 s2pu1

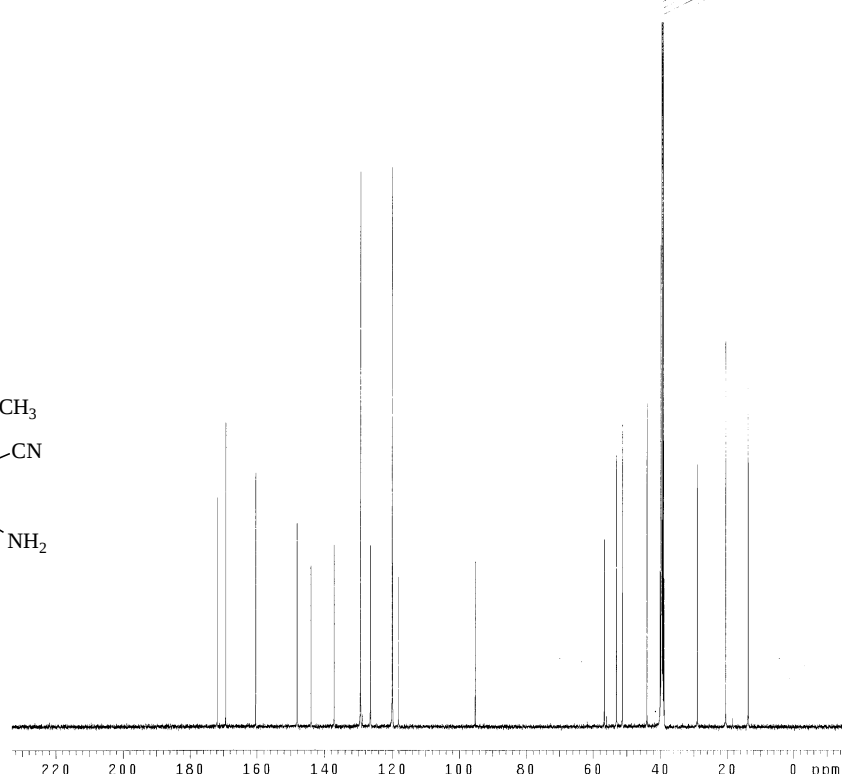
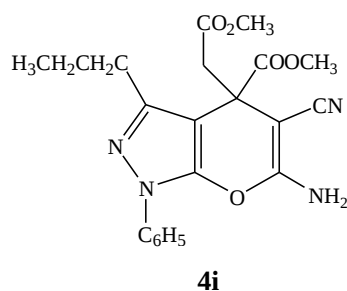
SAMPLE		SPECIAL	
date	May 20 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		SPECIAL	
sw	639.7	hst	0.008
at	639.7	pw90	10.800
np	52412	alpha	6.600
fb	4000	FLAGS	
bs	16	il	n
d1	3.000	in	n
nt	40	dp	y
ct	40	hs	nn
TRANSMITTER		PROCESSING	
tn	H1	fn	not used
sfrq	399.742	sp	DISPLAY
tof	800.0	wp	-465.9
tpwr	59	rfl	4538.9
pw	5.000	rfl	690.6
DECOUPLER		PLOT	
dn	C13	lp	-80.7
dof	0	lp	-14.1
dm	nmn	wc	180
dmm	c	sc	0
dpwr	45	vs	122
dmf	20056	th	4
	nm	ph	

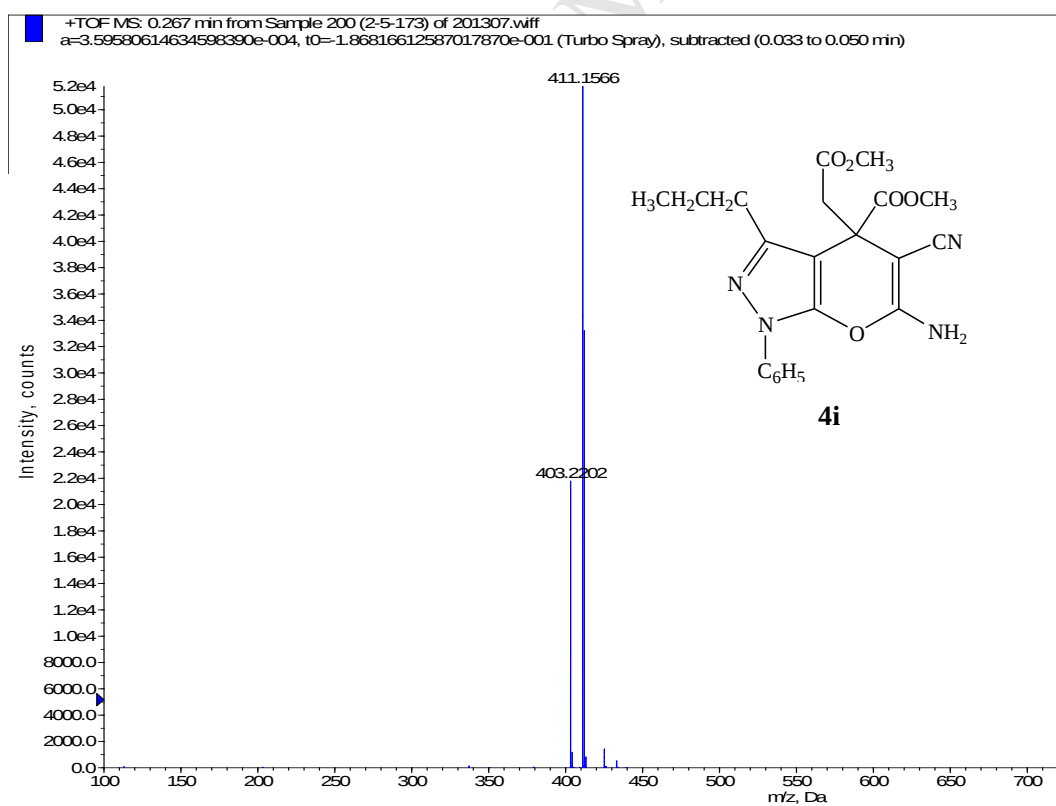
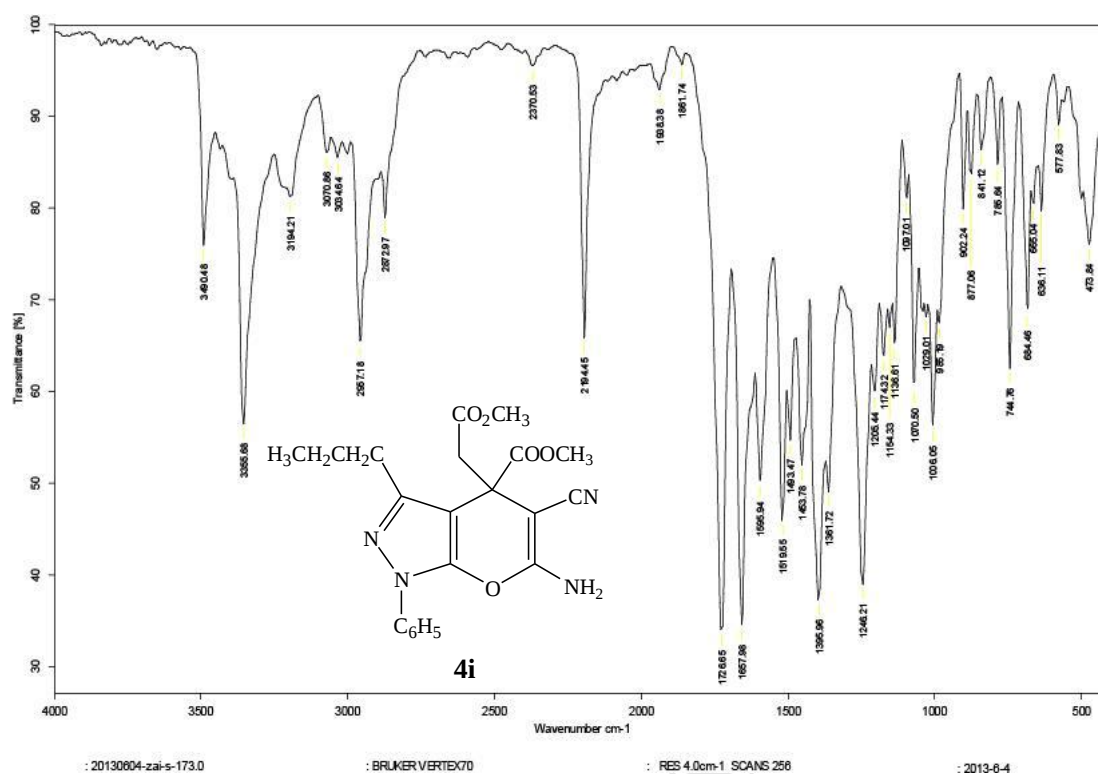


xinjiang daxue varian inova-400 13C-4N
zai-5-173-abulajiang in DMSO

exp2 s2pu1

SAMPLE		SPECIAL	
date	May 29 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	0.008
sw	25000.0	pw90	10.800
at	1.199	alpha	20.000
np	58968	FLAGS	
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	3000	hs	nn
ct	3000	PROCESSING	
tn	C13	lb	1.00
sfrq	100.525	fn	not used
tof	1500.0	sp	DISPLAY
tpwr	50	wp	-1567.1
pw	5.000	rfl	24999.2
DECOUPLER	H1	rfl	1567.8
dn	400.0	lp	-16.7
dof	yyy	lp	-254.7
dm	w	wc	180
dmm	42	sc	0
dpwr	13400	vs	237
dmf		th	12
	nm	ph	

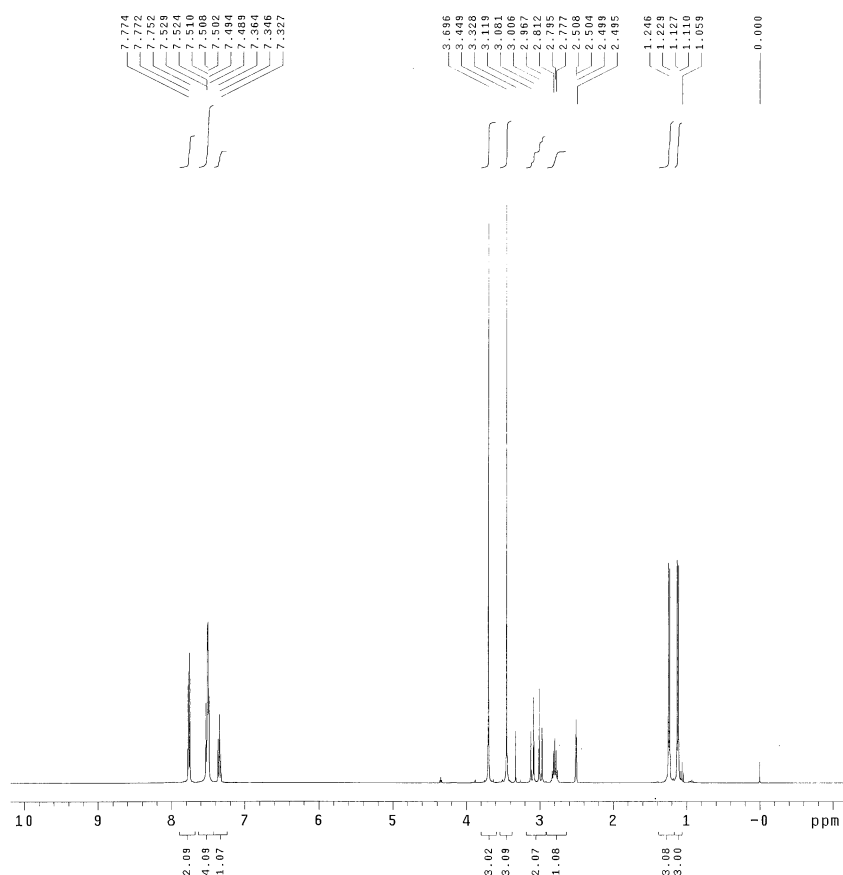
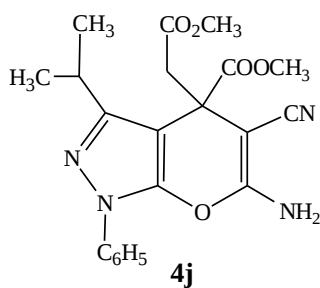




xinjiang daxue varian inova-400 1H-4N
zai-5-172-abulajiang in DMSO

exp1 s2pul

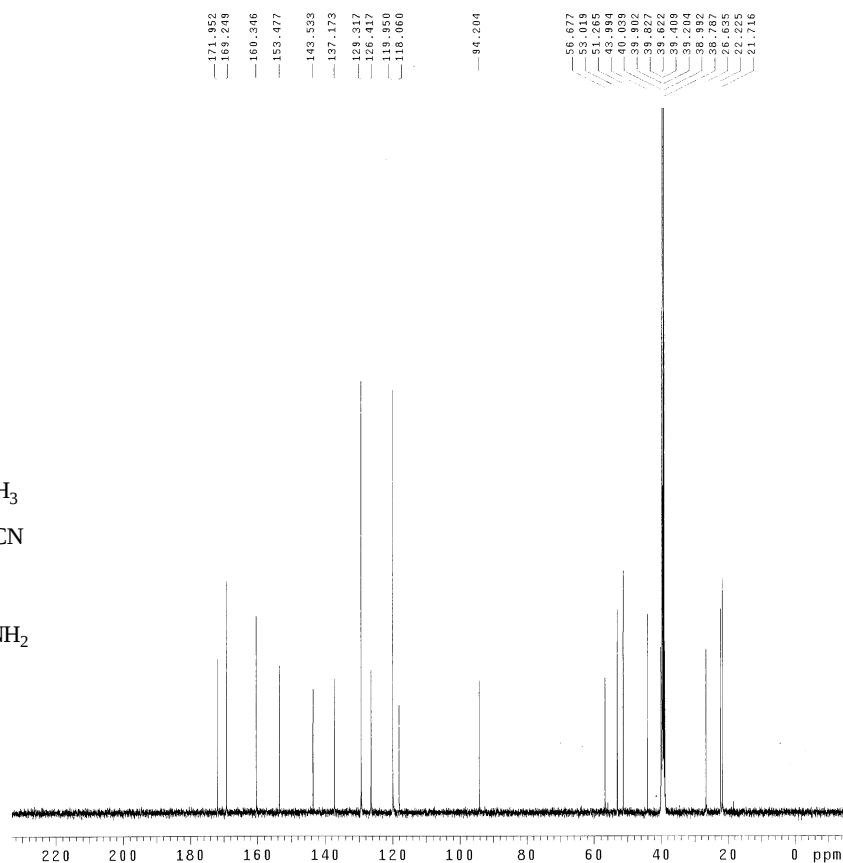
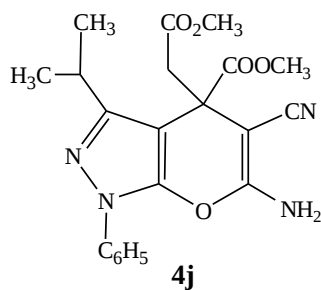
SAMPLE		SPECIAL	
date	May 20 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		hst	0.008
sw	6999.7	pw90	10.800
at	3.744	alpha	6.600
np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	40	hs	nn
ct	40	PROCESSING	
TRANSMITTER		fn	not used
tn		DISPLAY	
sfrq	399.742	sp	-466.1
tof	800.0	wp	4538.9
tpwr	59	rfl	690.8
pw	5.000	rfp	0
DECOUPLER		rp	-84.3
dn	C13	lp	-10.6
dof	0	PLOT	
dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	123
dmf	20056	th	4
	nm	ph	

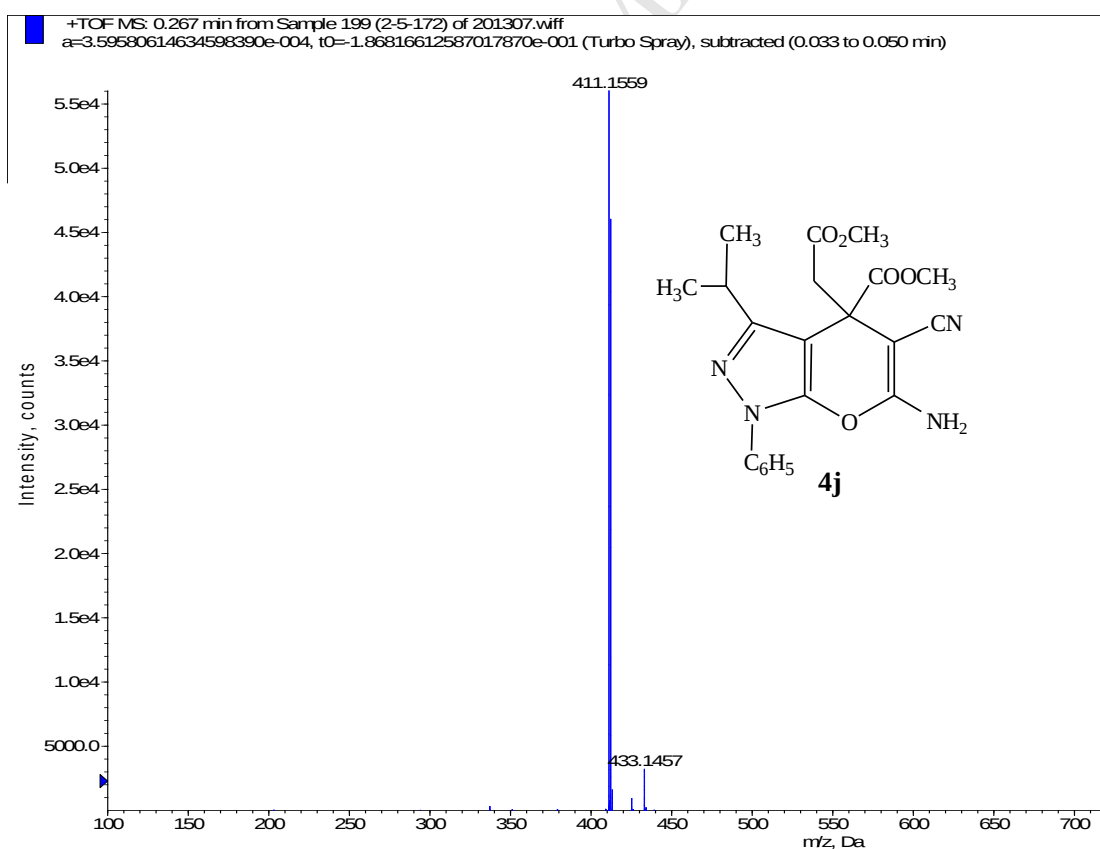
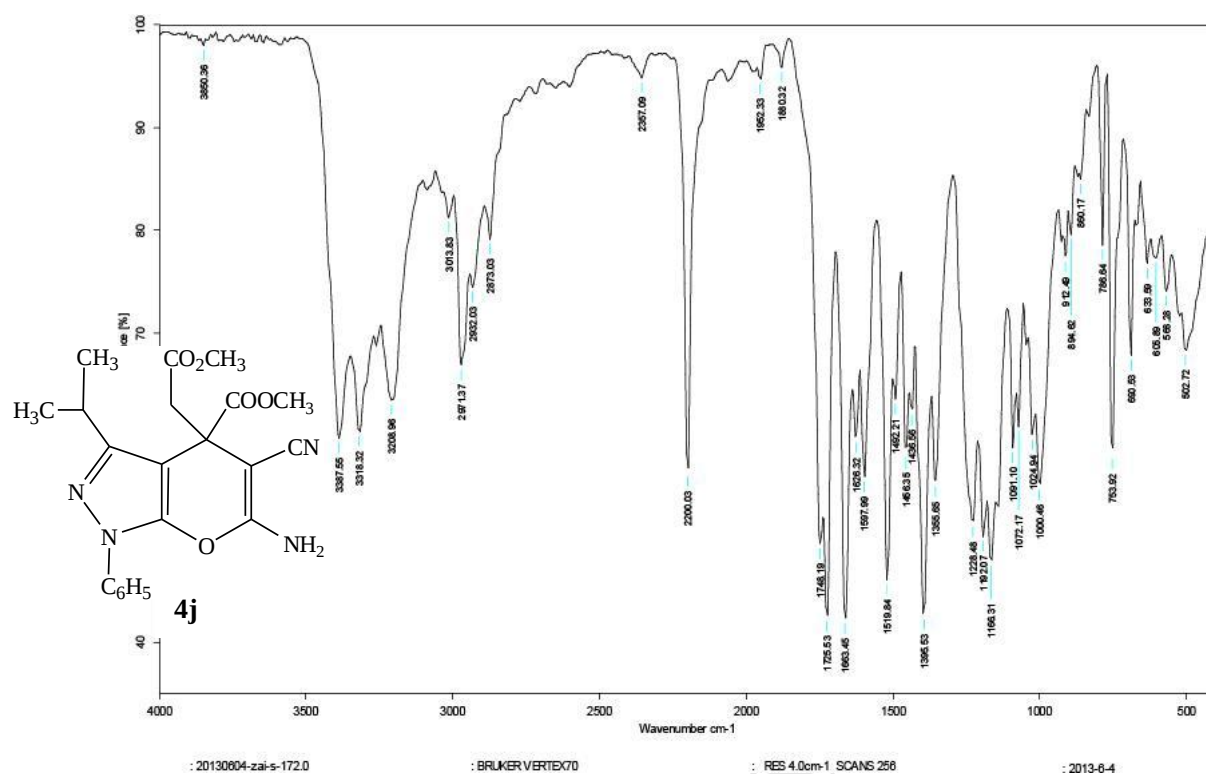


xinjiang daxue varian inova-400 13C-4N
zai-naipuguli-zai-5-172-abulajiang
in DMSO

exp2 s2pul

SAMPLE		SPECIAL	
date	May 28 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	0.008
sw	25000.0	pw90	10.800
at	1.189	alpha	20.000
np	59968	FLAGS	
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	1300	hs	nn
ct	1300	PROCESSING	
TRANSMITTER		lb	1.00
tn		fn	not used
sfrq	100.525	DISPLAY	
tof	1500.0	sp	-1567.1
tpwr	56	wp	24999.2
pw	5.000	rfl	1567.8
DECOUPLER		rfp	0
dn	H1	rp	-33.0
dof	400.0	lp	-272.6
dm	yyy	PLOT	
dmm	w	wc	180
dpwr	42	sc	0
dmf	13400	vs	251
	nm	ph	12

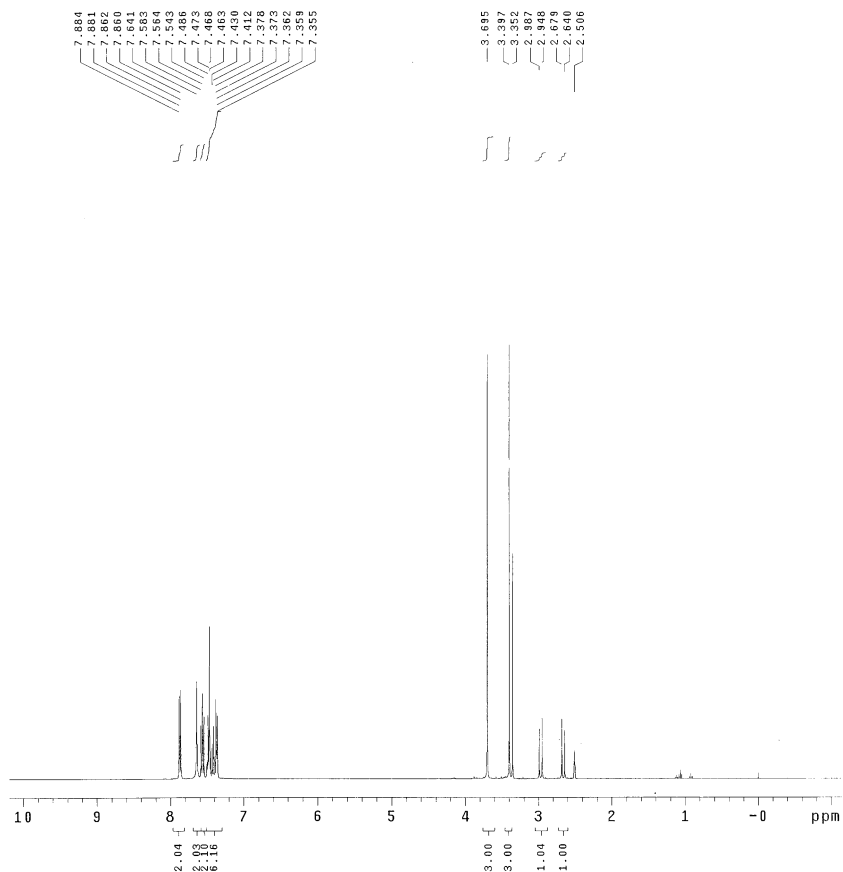
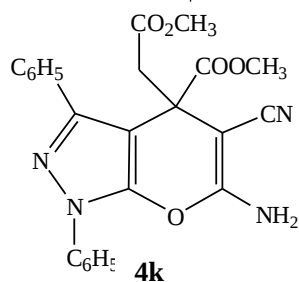




xinjiang daxue varian inova-400 1H-4N
zai-5-171-abulajiang in DMSO

exp1 s2pul

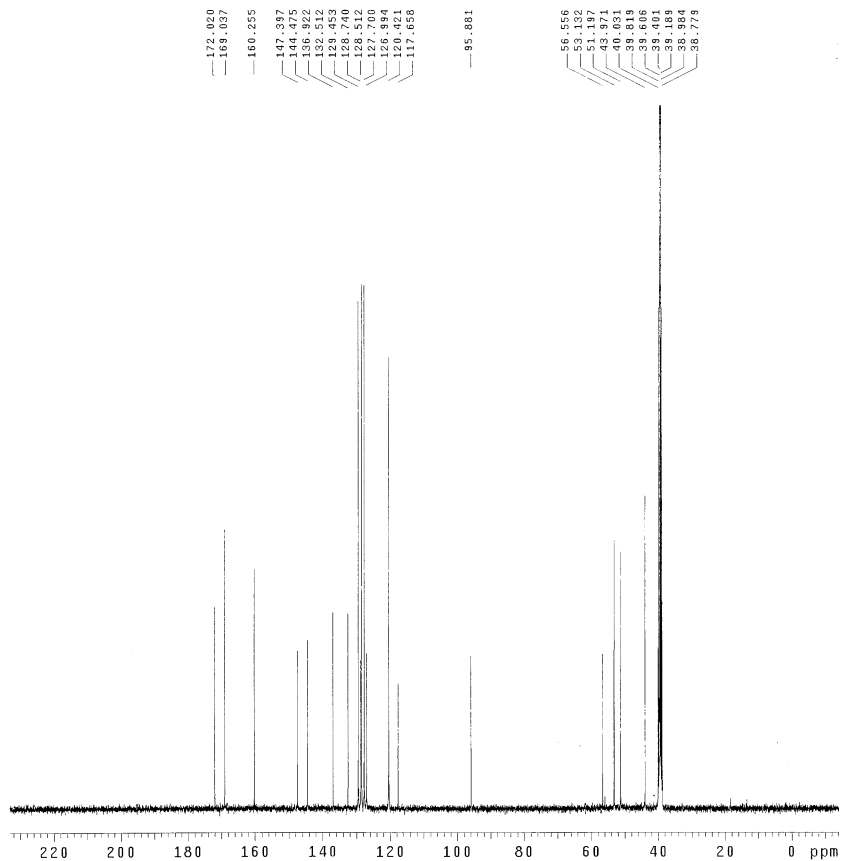
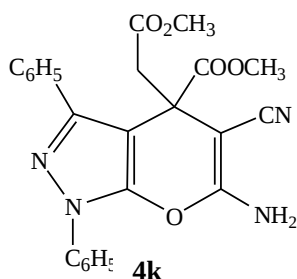
SAMPLE		SPECIAL	
date	May 20 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		hst	
sw	6999.7	pw90	0.008
at	3.744	alpha	10.800
np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	40	hs	nn
ct	40	PROCESSING	
tn	H1	fn	not used
sfrq	399.742	sp	-476.4
tof	800.0	wp	4550.2
tpwr	59	rfl	690.0
pw	5.000	rfp	0
DECOUPLER		rp	-79.6
dn	C13	lp	-16.5
dof	0	PLOT	
dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	92
dmf	20056	th	6
	nm	ph	

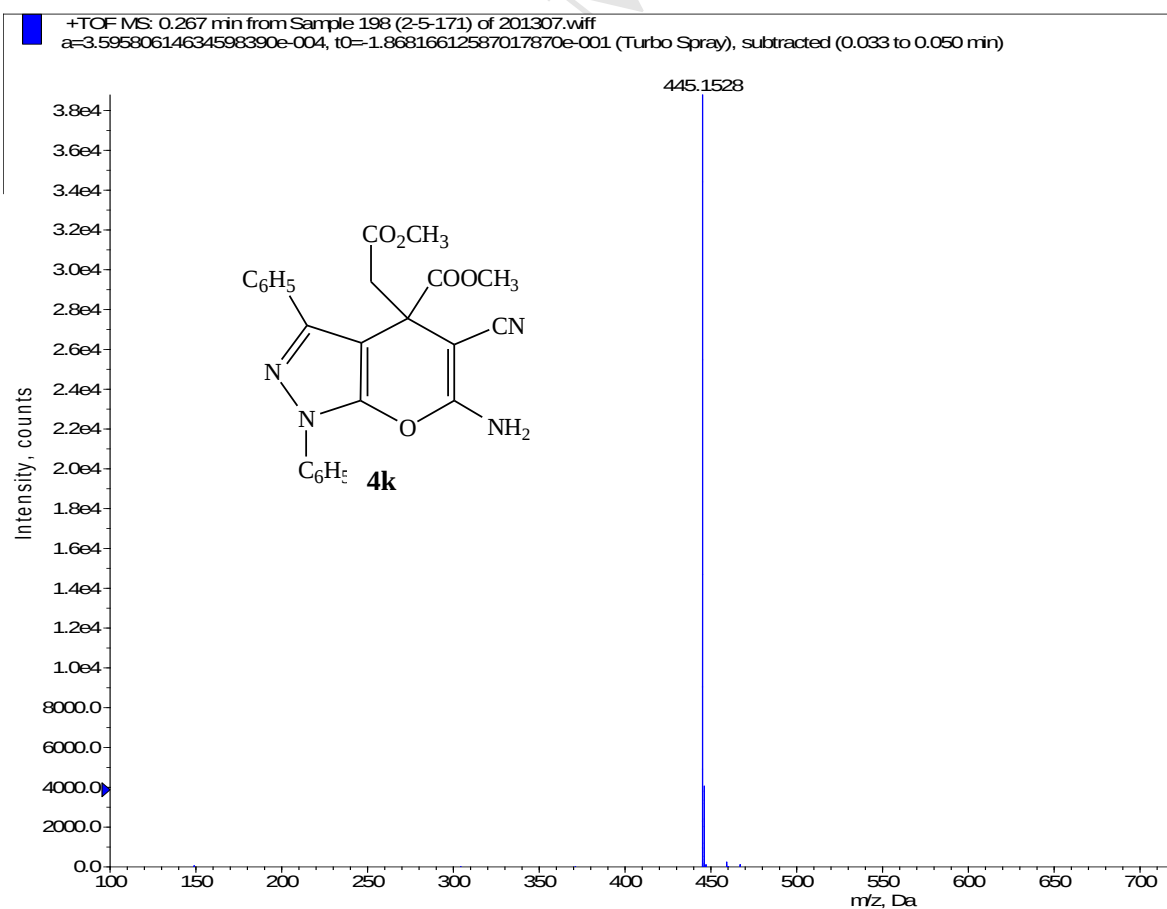
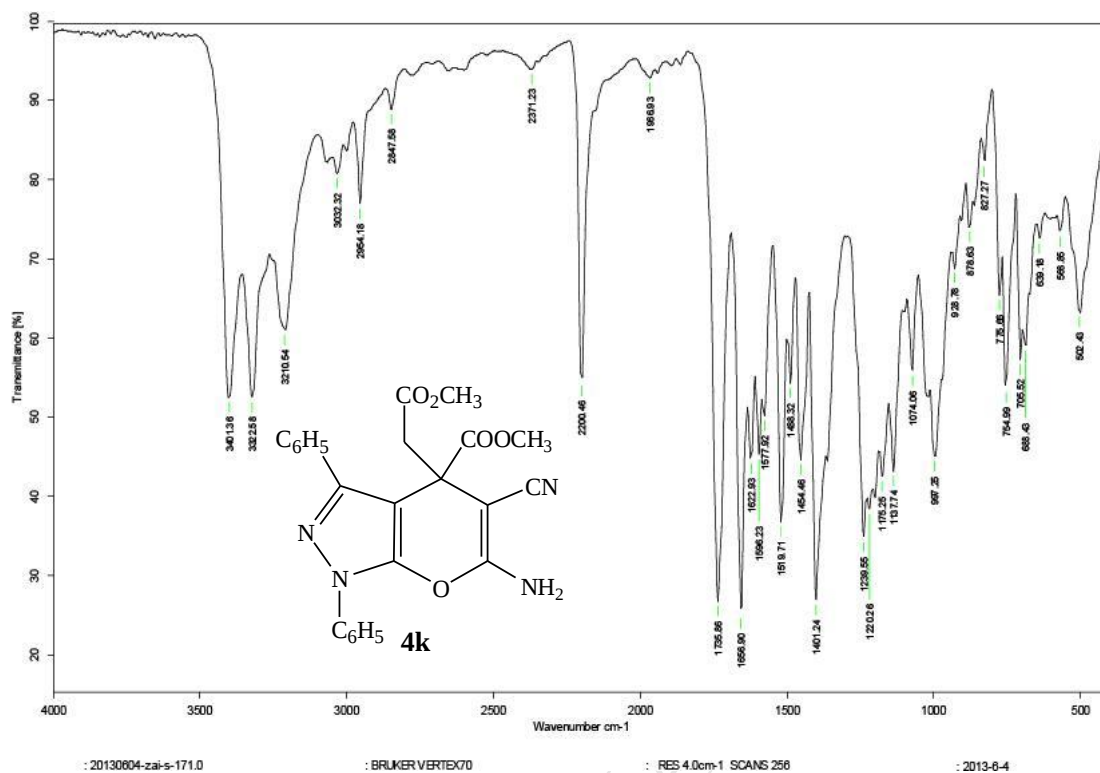


xinjiang daxue varian inova-400 13C-4N
zai-5-171-abulajiang in DMSO

exp2 s2pul

SAMPLE		SPECIAL	
date	May 28 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	
sw	25000.0	pw90	0.008
at	1.199	alpha	20.000
np	59968	il	n
fb	14000	in	n
bs	16	dp	y
d1	3.000	hs	nn
nt	1400	PROCESSING	
ct	1400	lb	1.00
tn	C13	fn	not used
sfrq	100.525	sp	-1567.1
tof	1500.0	wp	24999.2
tpwr	56	rfl	1567.6
pw	5.000	rfp	0
DECOUPLER		rp	-11.5
dn	H1	lp	-262.6
dof	400.0	PLOT	
dm	yyy	wc	180
dmm	42	sc	0
dpwr	13400	vs	251
dmf		th	12
	nm	ph	

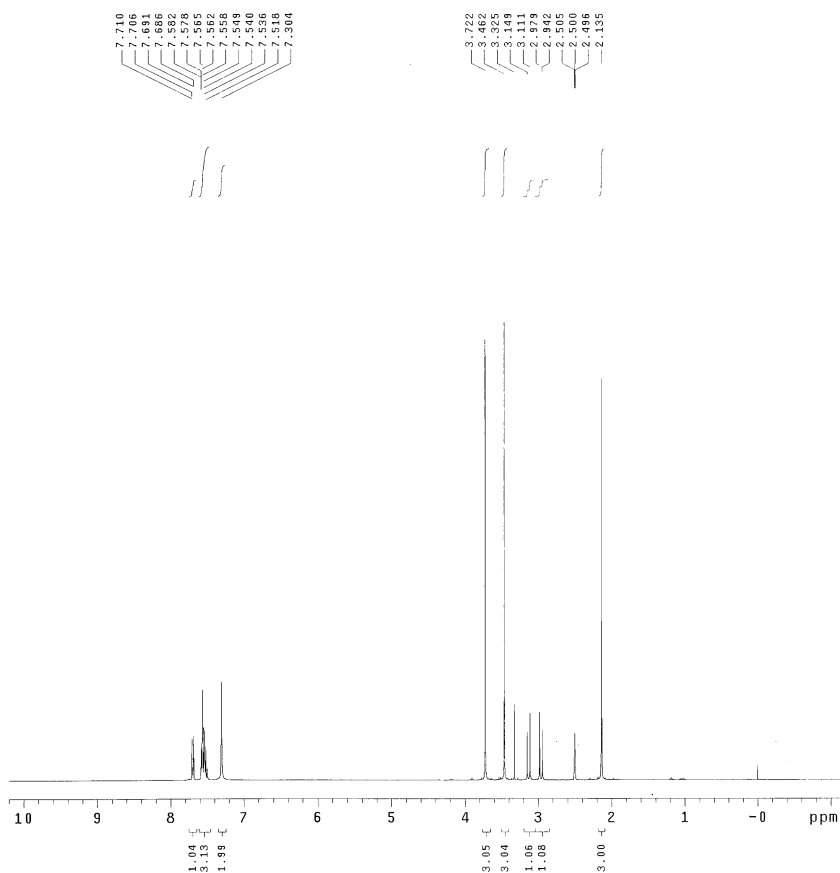
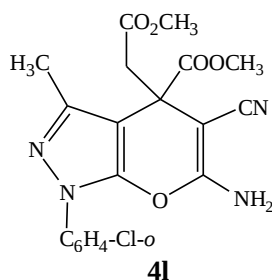




xinjiang daxue varian inova-400 1H-4N
zainaipuguli-zai-6-193-abulajiang
in DMSO

exp1 s2pu1

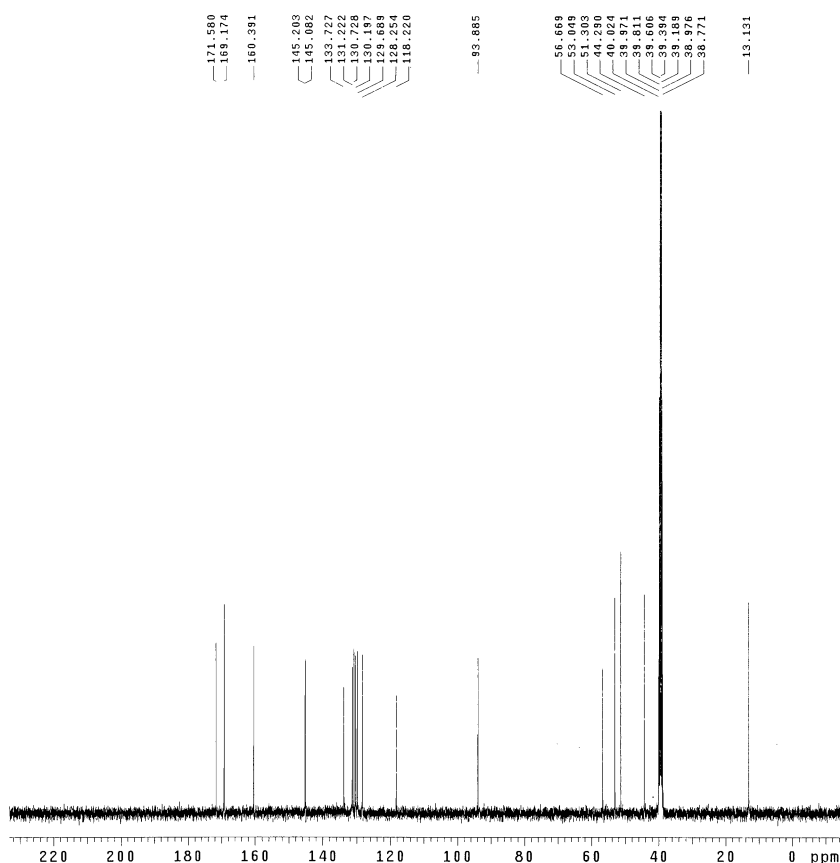
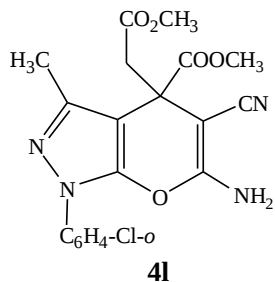
SAMPLE		SPECIAL	
date	Jun 14 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		hst	0.008
sw	6999.7	pw90	10.600
at	3.744	alpha	6.600
np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	20	hs	nn
ct	20	PROCESSING	
TRANSMITTER		fn	not used
tn		DISPLAY	
sfrq	399.742	sp	-468.5
tof	800.0	wp	4546.2
tpwr	59	rfl	693.6
pw	3.000	rpf	0
DECOUPLER		rp	-80.9
dn	C13	lp	-11.1
dof	0	PLOT	
dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	97
dmf	20056	th	4
	nm	ph	

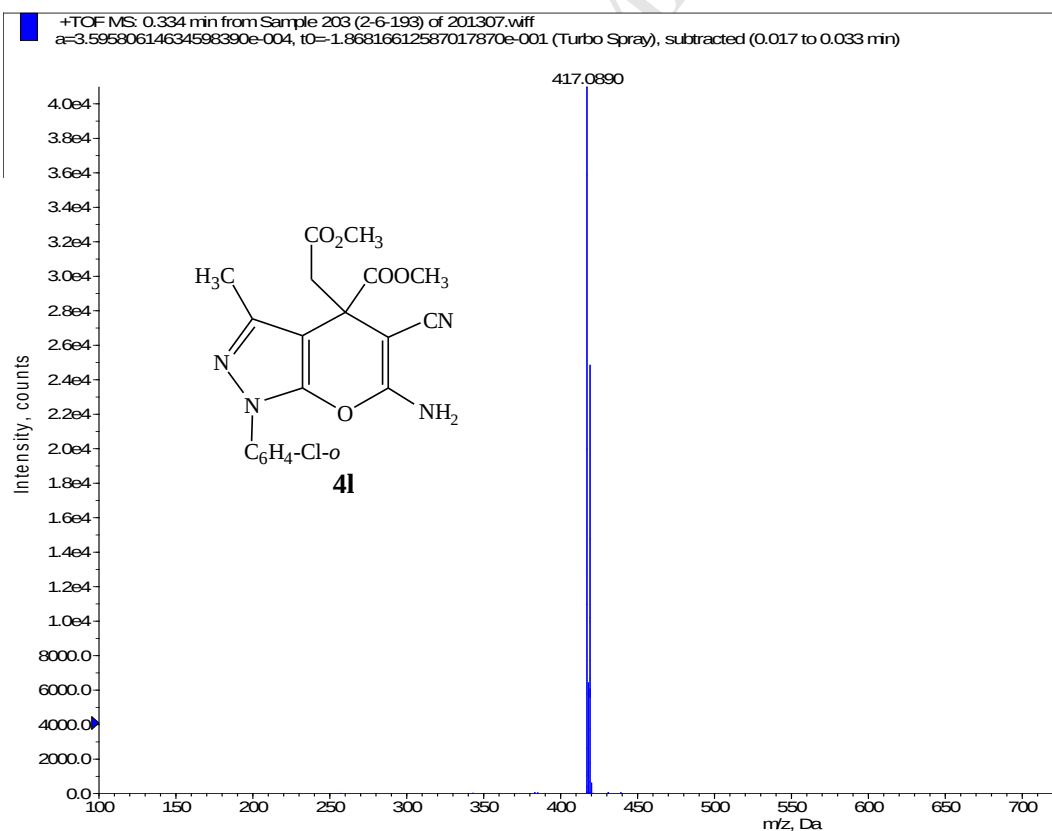
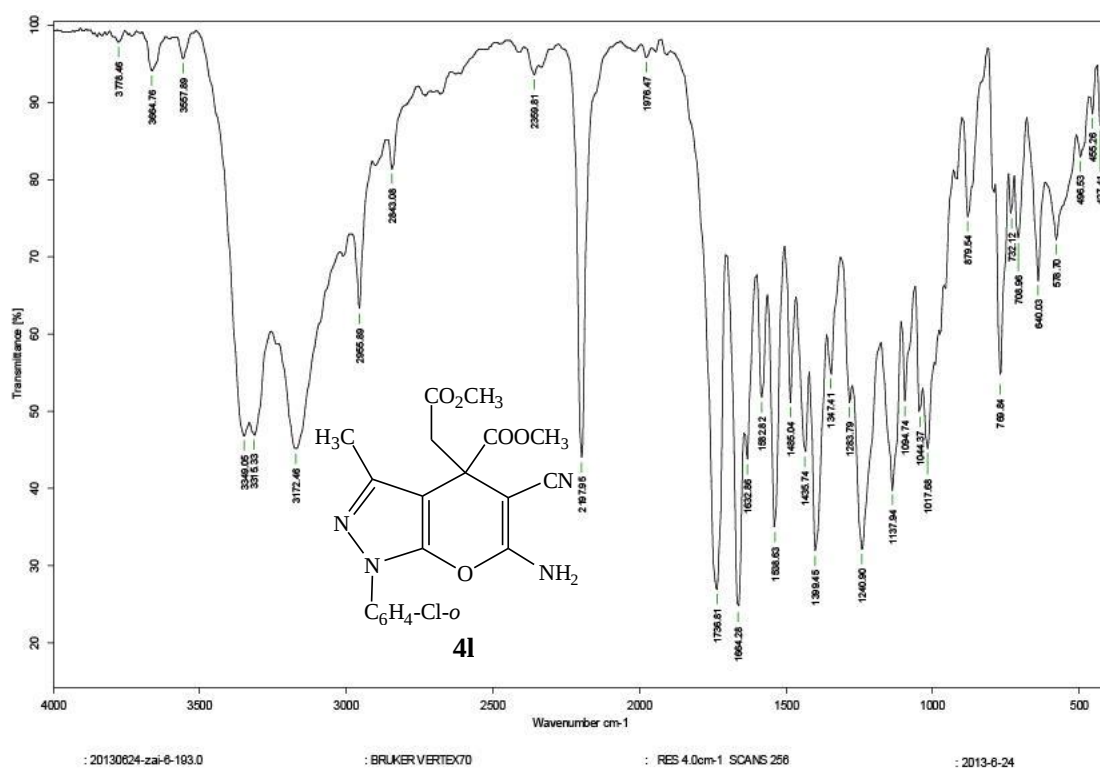


xinjiang daxue varian inova-400 13C-4N
zainaipuguli-zai-6-193-abulajiang
in DMSO

exp2 s2pu1

SAMPLE		SPECIAL	
date	Jul 18 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	0.008
sw	25000.0	pw90	10.600
at	1.195	alpha	20.000
np	59968	FLAGS	
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	300	hs	nn
ct	300	PROCESSING	
TRANSMITTER		lb	1.00
tn		fn	not used
sfrq	100.525	DISPLAY	
tof	1500.0	sp	-1567.1
tpwr	56	wp	24999.2
pw	5.000	rfl	1567.8
DECOUPLER		rp	0
dn	H1	rp	-46.8
dof	400.0	lp	-242.0
dm	yyy	PLOT	
dmm	w	wc	180
dpwr	42	sc	0
dmf	13400	vs	208
	nm	th	14
		ph	

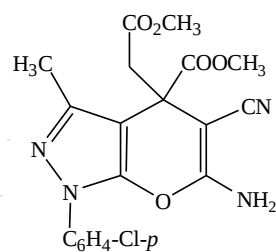




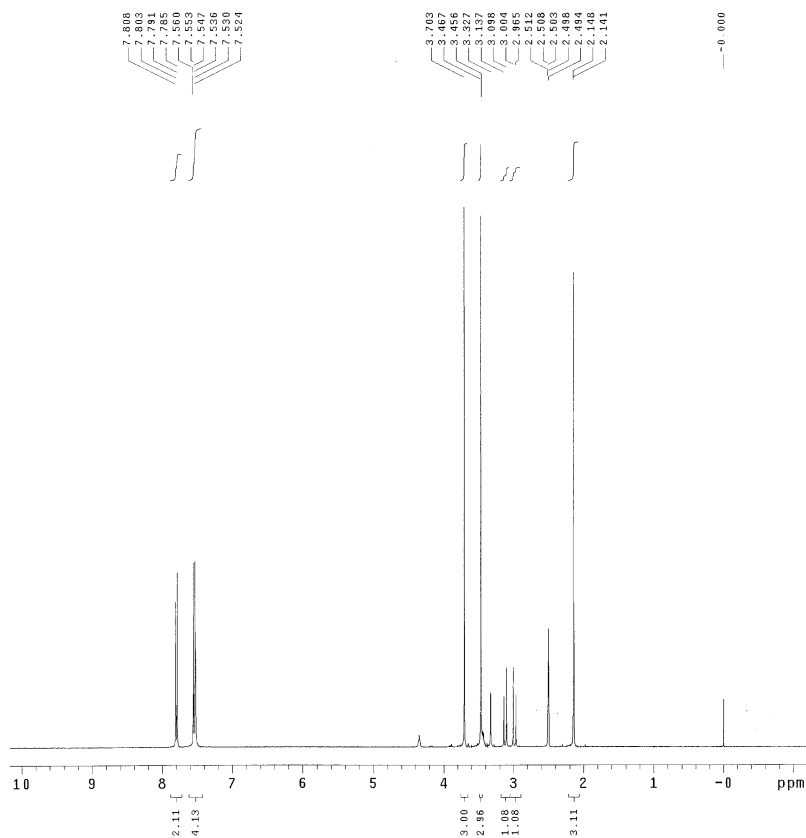
xinjiang daxue varian inova-400 1H-4N
zainaipuguli-2-zai-4-161-abulajiang
in DMSO

exp1 s2pu1

SAMPLE		SPECIAL	
date	May 2 2013	temp	not used
solvent	DMSO	gain	30
file	exp	spin	20
ACQUISITION		hst	0.008
sw	6399.7	pw90	10.800
at	3.744	alfa	6.600
np	52412	FLAGS	
fb	4000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	20	hs	nn
ct	20	PROCESSING	
TRANSMITTER		fn	not used
tn	H1	DISPLAY	0
sfrq	399.741	sp	-495.6
tof	400.0	wp	4561.3
tpwr	59	rfl	1081.1
pw	5.000	rff	0
DECOUPLER		rp	-115.5
dn	C13	lp	-15.3
dof	0	PLOT	
dm	nnn	wc	180
dmm	c	sc	0
dpwr	45	vs	120
dmf	20056	th	4
	nm	ph	



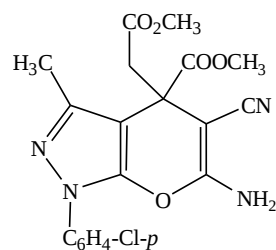
4m



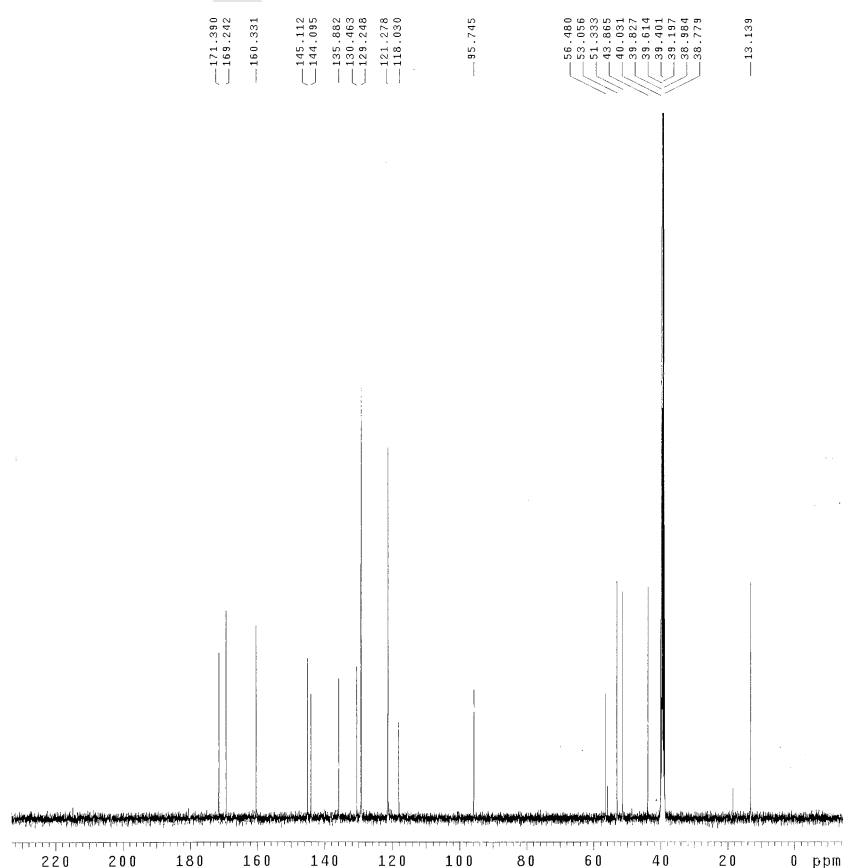
xinjiang daxue varian inova-400 13C-4N
zainaipuguli-2-zai-5-161-abulajiang
in DMSO

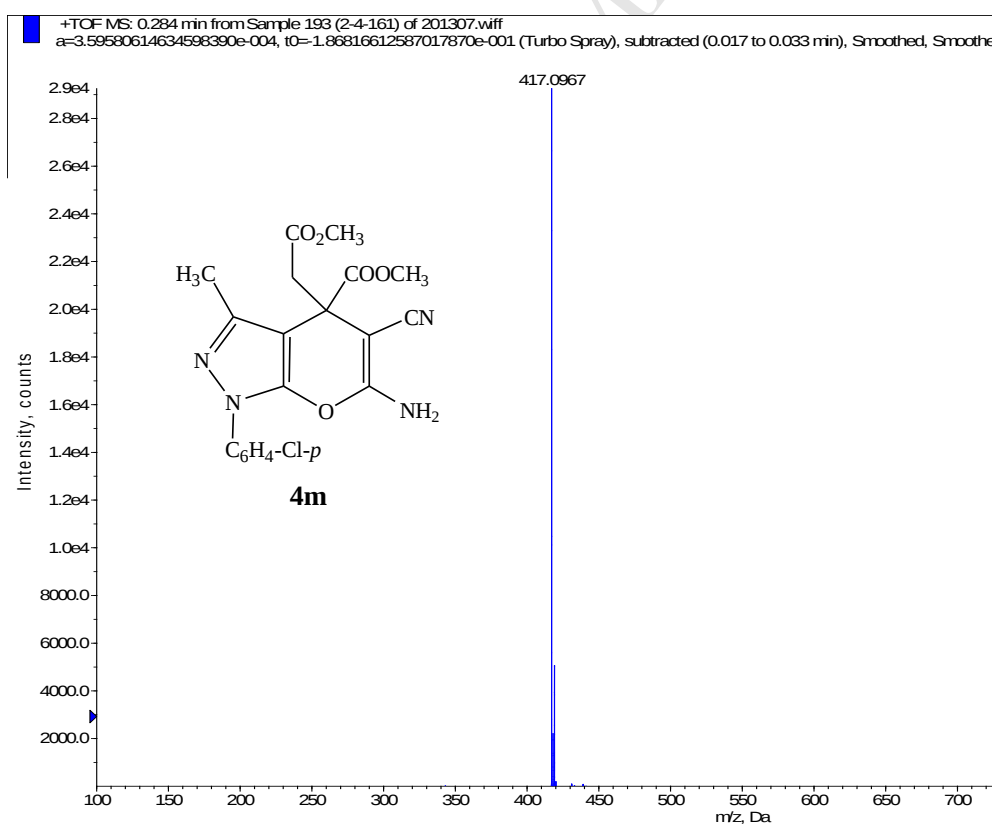
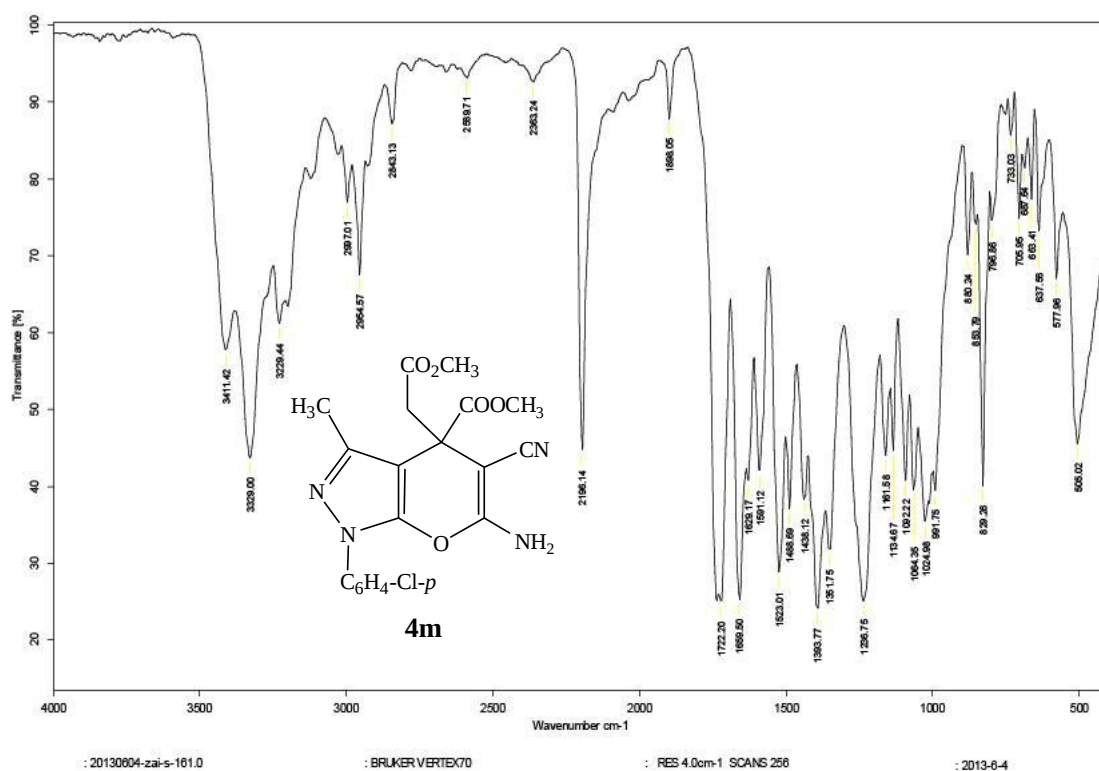
exp2 s2pu1

SAMPLE		SPECIAL	
date	May 28 2013	temp	not used
solvent	DMSO	gain	50
file	exp	spin	20
ACQUISITION		hst	0.008
sw	25000.0	pw90	10.800
at	1.199	alfa	20.000
np	59868	FLAGS	
fb	14000	il	n
bs	16	in	n
d1	3.000	dp	y
nt	1000	hs	nn
ct	1000	PROCESSING	
TRANSMITTER		lb	1.00
tn	C13	fn	not used
sfrq	100.525	DISPLAY	0
tof	1500.0	sp	-1567.1
tpwr	56	wp	24999.2
pw	5.000	rff	1567.8
DECOUPLER		rp	0
dn	H1	rp	-36.3
dof	400.0	lp	-261.4
dm	yyy	PLOT	
dmm	w	wc	180
dpwr	42	sc	0
dmf	13400	vs	296
	nm	th	12
		ph	



4m





5. X-ray crystallographic data data for compound 4a.

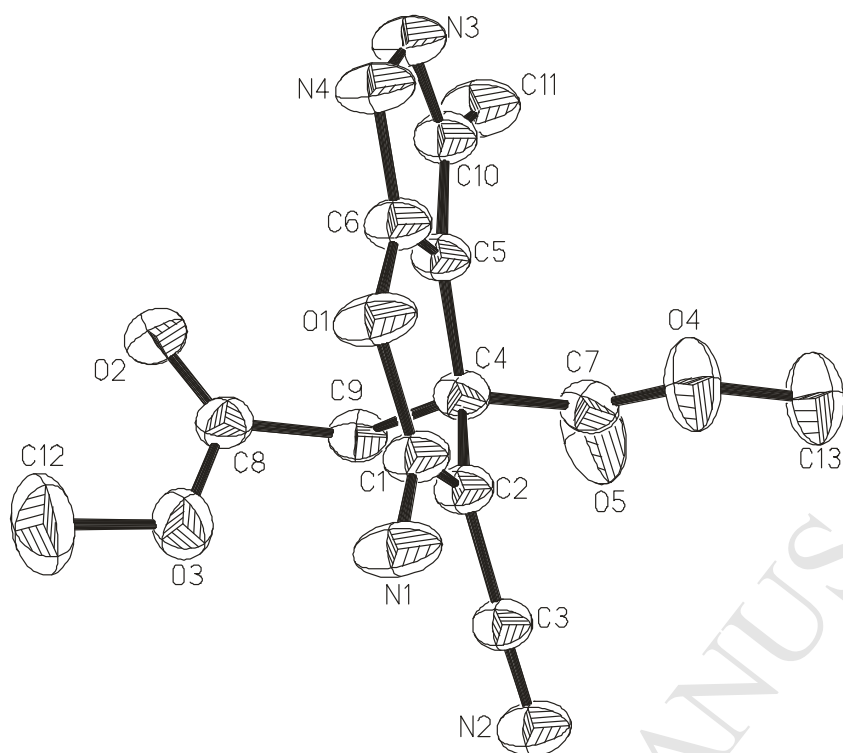


Figure 1. ORTAP diagram of compound 4a

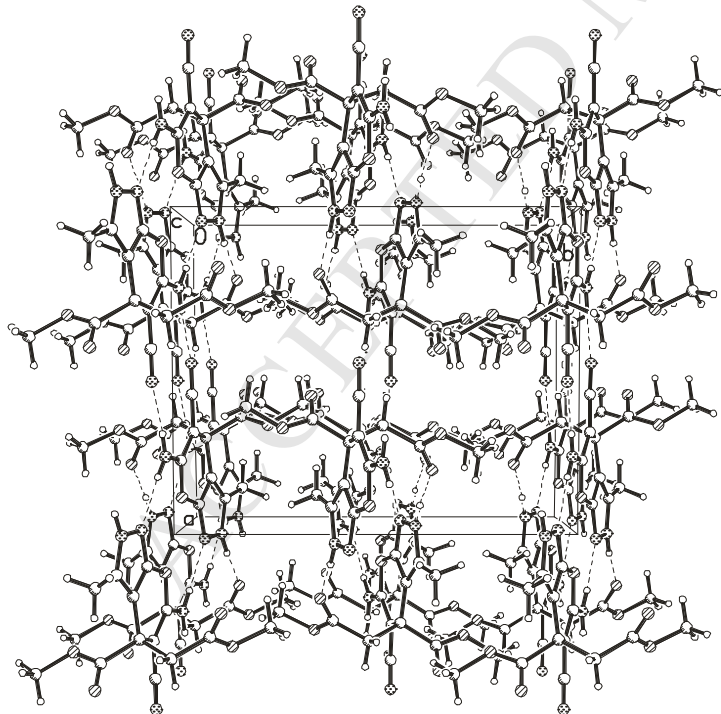


Figure 2. Packing diagram of compound 4a

Table 1. Crystal data and structure refinement for A.

Identification code	a
Empirical formula	C ₁₃ H ₁₄ N ₄ O ₅
Formula weight	306.28
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P2(1)/c
Unit cell dimensions	a = 11.4834(17) Å alpha = 90 deg. b = 14.134(3) Å beta = 97.363(4) deg. c = 9.8205(15) Å gamma = 90 deg.
Volume	1580.8(4) Å ³
Z, Calculated density	4, 1.287 Mg/m ³
Absorption coefficient	0.101 mm ⁻¹
F(000)	640
Crystal size	0.79 x 0.65 x 0.26 mm
Theta range for data collection	3.26 to 27.47 deg.
Limiting indices	-14 ≤ h ≤ 14, -18 ≤ k ≤ 18, -12 ≤ l ≤ 12
Reflections collected / unique	13236 / 3585 [R(int) = 0.0358]
Completeness to theta = 27.47	99.1 %
Absorption correction	Empirical
Max. and min. transmission	0.9607 and 0.9281
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	3585 / 0 / 204
Goodness-of-fit on F ²	1.006
Final R indices [I > 2σ(I)]	R ₁ = 0.0771, wR ₂ = 0.2307
R indices (all data)	R ₁ = 0.1227, wR ₂ = 0.3179
Extinction coefficient	0.077(15)
Largest diff. peak and hole	0.377 and -0.343 e.Å ⁻³

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for A. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
O(1)	1108(2)	-149(2)	1038(2)	68(1)
C(3)	4310(2)	-387(2)	1675(3)	51(1)
N(4)	-284(2)	-630(2)	2406(3)	67(1)
C(1)	2272(2)	-143(2)	893(3)	54(1)
N(1)	2423(2)	164(2)	-353(3)	70(1)
N(2)	5277(2)	-387(2)	1552(3)	72(1)
N(3)	-199(2)	-995(2)	3707(3)	66(1)
O(2)	2016(2)	1382(2)	4720(3)	93(1)
C(2)	3121(2)	-409(2)	1916(3)	49(1)
C(4)	2914(2)	-633(2)	3400(3)	51(1)
C(5)	1621(2)	-770(2)	3370(3)	50(1)
C(6)	827(2)	-508(2)	2246(3)	57(1)
C(7)	3585(2)	-1532(2)	3866(3)	59(1)
C(8)	2769(3)	1082(2)	4113(3)	60(1)
C(9)	3391(2)	173(2)	4390(3)	57(1)
O(3)	3158(3)	1565(2)	3113(3)	91(1)
O(4)	3335(3)	-2244(2)	3057(3)	96(1)
O(5)	4277(3)	-1579(2)	4870(4)	131(2)
C(10)	915(2)	-1084(2)	4315(3)	59(1)
C(11)	1197(3)	-1469(3)	5715(4)	89(1)
C(13)	3967(4)	-3127(3)	3389(5)	110(2)
C(12)	2625(6)	2475(4)	2770(7)	149(2)

Table 3. Bond lengths [Å] and angles [deg] for A.

O(1)-C(1)	1.363(3)
O(1)-C(6)	1.366(3)
C(3)-N(2)	1.132(3)
C(3)-C(2)	1.416(3)
N(4)-C(6)	1.317(3)
N(4)-N(3)	1.370(4)
C(1)-N(1)	1.331(4)
C(1)-C(2)	1.360(4)
N(1)-H(1A)	0.8600
N(1)-H(1B)	0.8600
N(3)-C(10)	1.346(4)
N(3)-H(3A)	1.01(4)
O(2)-C(8)	1.189(4)
C(2)-C(4)	1.539(4)
C(4)-C(5)	1.494(3)
C(4)-C(7)	1.526(4)
C(4)-C(9)	1.550(4)
C(5)-C(10)	1.381(4)
C(5)-C(6)	1.389(4)
C(7)-O(5)	1.187(4)
C(7)-O(4)	1.291(4)
C(8)-O(3)	1.320(4)
C(8)-C(9)	1.479(4)
C(9)-H(9A)	0.9700
C(9)-H(9B)	0.9700
O(3)-C(12)	1.445(5)
O(4)-C(13)	1.460(4)
C(10)-C(11)	1.476(5)
C(11)-H(11A)	0.9600
C(11)-H(11B)	0.9600
C(11)-H(11C)	0.9600
C(13)-H(13A)	0.9600
C(13)-H(13B)	0.9600
C(13)-H(13C)	0.9600
C(12)-H(12A)	0.9600
C(12)-H(12B)	0.9600
C(12)-H(12C)	0.9600
C(1)-O(1)-C(6)	115.9(2)
N(2)-C(3)-C(2)	176.3(3)
C(6)-N(4)-N(3)	101.9(2)
N(1)-C(1)-C(2)	127.1(2)

N(1)-C(1)-O(1)	110.1(2)
C(2)-C(1)-O(1)	122.7(2)
C(1)-N(1)-H(1A)	120.0
C(1)-N(1)-H(1B)	120.0
H(1A)-N(1)-H(1B)	120.0
C(10)-N(3)-N(4)	113.5(2)
C(10)-N(3)-H(3A)	120(2)
N(4)-N(3)-H(3A)	126(2)
C(1)-C(2)-C(3)	119.0(2)
C(1)-C(2)-C(4)	125.0(2)
C(3)-C(2)-C(4)	115.6(2)
C(5)-C(4)-C(7)	111.1(2)
C(5)-C(4)-C(2)	106.4(2)
C(7)-C(4)-C(2)	108.8(2)
C(5)-C(4)-C(9)	112.2(2)
C(7)-C(4)-C(9)	107.4(2)
C(2)-C(4)-C(9)	110.9(2)
C(10)-C(5)-C(6)	103.8(2)
C(10)-C(5)-C(4)	134.1(3)
C(6)-C(5)-C(4)	122.0(2)
N(4)-C(6)-O(1)	119.5(3)
N(4)-C(6)-C(5)	114.7(3)
O(1)-C(6)-C(5)	125.9(2)
O(5)-C(7)-O(4)	122.7(3)
O(5)-C(7)-C(4)	123.6(3)
O(4)-C(7)-C(4)	113.6(3)
O(2)-C(8)-O(3)	121.7(3)
O(2)-C(8)-C(9)	125.6(3)
O(3)-C(8)-C(9)	112.6(3)
C(8)-C(9)-C(4)	113.5(2)
C(8)-C(9)-H(9A)	108.9
C(4)-C(9)-H(9A)	108.9
C(8)-C(9)-H(9B)	108.9
C(4)-C(9)-H(9B)	108.9
H(9A)-C(9)-H(9B)	107.7
C(8)-O(3)-C(12)	117.4(3)
C(7)-O(4)-C(13)	117.7(3)
N(3)-C(10)-C(5)	106.2(3)
N(3)-C(10)-C(11)	122.0(3)
C(5)-C(10)-C(11)	131.8(3)
C(10)-C(11)-H(11A)	109.5
C(10)-C(11)-H(11B)	109.5
H(11A)-C(11)-H(11B)	109.5
C(10)-C(11)-H(11C)	109.5

H(11A)-C(11)-H(11C)	109.5
H(11B)-C(11)-H(11C)	109.5
O(4)-C(13)-H(13A)	109.5
O(4)-C(13)-H(13B)	109.5
H(13A)-C(13)-H(13B)	109.5
O(4)-C(13)-H(13C)	109.5
H(13A)-C(13)-H(13C)	109.5
H(13B)-C(13)-H(13C)	109.5
O(3)-C(12)-H(12A)	109.5
O(3)-C(12)-H(12B)	109.5
H(12A)-C(12)-H(12B)	109.5
O(3)-C(12)-H(12C)	109.5
H(12A)-C(12)-H(12C)	109.5
H(12B)-C(12)-H(12C)	109.5

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for A. The anisotropic displacement factor exponent takes the form:

$$-2\pi^2 [h^2 a^{*2} U_{11} + \dots + 2hka^*b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
O(1)	34(1)	108(2)	61(1)	28(1)	4(1)	7(1)
C(3)	40(1)	61(2)	53(2)	7(1)	3(1)	1(1)
N(4)	38(1)	99(2)	64(2)	16(1)	6(1)	-3(1)
C(1)	35(1)	68(2)	58(2)	9(1)	4(1)	1(1)
N(1)	39(1)	111(2)	60(2)	29(1)	6(1)	6(1)
N(2)	44(1)	94(2)	77(2)	13(1)	8(1)	1(1)
N(3)	42(1)	94(2)	62(2)	13(1)	6(1)	-3(1)
O(2)	83(2)	80(2)	126(2)	-7(2)	52(2)	2(1)
C(2)	35(1)	59(2)	52(2)	8(1)	4(1)	1(1)
C(4)	36(1)	58(2)	56(2)	9(1)	-2(1)	1(1)
C(5)	36(1)	61(2)	53(2)	7(1)	0(1)	-4(1)
C(6)	40(1)	72(2)	59(2)	11(1)	4(1)	-2(1)
C(7)	46(2)	65(2)	64(2)	13(1)	-3(1)	0(1)
C(8)	50(2)	63(2)	67(2)	-10(1)	9(2)	-7(1)
C(9)	47(2)	68(2)	53(2)	1(1)	-3(1)	-4(1)
O(3)	121(2)	69(2)	89(2)	16(1)	41(2)	15(1)
O(4)	119(2)	62(2)	93(2)	-5(1)	-35(2)	24(1)
O(5)	143(3)	78(2)	145(3)	10(2)	-83(2)	18(2)
C(10)	49(2)	71(2)	57(2)	9(1)	3(1)	-2(1)
C(11)	72(2)	126(3)	72(2)	34(2)	15(2)	1(2)
C(13)	137(4)	63(2)	122(4)	6(2)	-10(3)	30(2)
C(12)	218(7)	82(3)	155(5)	40(3)	58(5)	46(4)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for A.

	x	y	z	U(eq)
H(1A)	3120	211	-581	84
H(1B)	1825	316	-930	84
H(9A)	3325	-20	5325	68
H(9B)	4217	265	4316	68
H(11A)	2033	-1478	5962	134
H(11B)	895	-2101	5744	134
H(11C)	846	-1078	6351	134
H(13A)	3694	-3599	2721	164
H(13B)	3829	-3337	4284	164
H(13C)	4792	-3025	3380	164
H(12A)	2978	2753	2031	223
H(12B)	2744	2882	3558	223
H(12C)	1798	2394	2494	223
H(3A)	-880(30)	-1110(30)	4250(40)	86(11)

Table 6. Torsion angles [deg] for A.

C(6)-O(1)-C(1)-N(1)	175.9(3)
C(6)-O(1)-C(1)-C(2)	-4.8(4)
C(6)-N(4)-N(3)-C(10)	0.5(4)
N(1)-C(1)-C(2)-C(3)	-1.3(5)
O(1)-C(1)-C(2)-C(3)	179.6(3)
N(1)-C(1)-C(2)-C(4)	171.5(3)
O(1)-C(1)-C(2)-C(4)	-7.7(5)
N(2)-C(3)-C(2)-C(1)	-180(100)
N(2)-C(3)-C(2)-C(4)	7(5)
C(1)-C(2)-C(4)-C(5)	15.8(4)
C(3)-C(2)-C(4)-C(5)	-171.2(2)
C(1)-C(2)-C(4)-C(7)	135.5(3)
C(3)-C(2)-C(4)-C(7)	-51.5(3)
C(1)-C(2)-C(4)-C(9)	-106.5(3)
C(3)-C(2)-C(4)-C(9)	66.5(3)
C(7)-C(4)-C(5)-C(10)	53.1(4)
C(2)-C(4)-C(5)-C(10)	171.3(3)
C(9)-C(4)-C(5)-C(10)	-67.2(4)

C(7)-C(4)-C(5)-C(6)	-131.5(3)
C(2)-C(4)-C(5)-C(6)	-13.2(4)
C(9)-C(4)-C(5)-C(6)	108.2(3)
N(3)-N(4)-C(6)-O(1)	179.8(3)
N(3)-N(4)-C(6)-C(5)	-0.3(4)
C(1)-O(1)-C(6)-N(4)	-172.9(3)
C(1)-O(1)-C(6)-C(5)	7.1(4)
C(10)-C(5)-C(6)-N(4)	0.0(4)
C(4)-C(5)-C(6)-N(4)	-176.7(3)
C(10)-C(5)-C(6)-O(1)	179.9(3)
C(4)-C(5)-C(6)-O(1)	3.3(5)
C(5)-C(4)-C(7)-O(5)	-118.3(4)
C(2)-C(4)-C(7)-O(5)	124.9(4)
C(9)-C(4)-C(7)-O(5)	4.7(4)
C(5)-C(4)-C(7)-O(4)	62.2(3)
C(2)-C(4)-C(7)-O(4)	-54.6(3)
C(9)-C(4)-C(7)-O(4)	-174.7(3)
O(2)-C(8)-C(9)-C(4)	99.4(4)
O(3)-C(8)-C(9)-C(4)	-82.2(3)
C(5)-C(4)-C(9)-C(8)	-54.1(3)
C(7)-C(4)-C(9)-C(8)	-176.5(2)
C(2)-C(4)-C(9)-C(8)	64.7(3)
O(2)-C(8)-O(3)-C(12)	-0.6(6)
C(9)-C(8)-O(3)-C(12)	-179.0(4)
O(5)-C(7)-O(4)-C(13)	-1.8(6)
C(4)-C(7)-O(4)-C(13)	177.7(3)
N(4)-N(3)-C(10)-C(5)	-0.6(4)
N(4)-N(3)-C(10)-C(11)	-179.6(3)
C(6)-C(5)-C(10)-N(3)	0.4(3)
C(4)-C(5)-C(10)-N(3)	176.3(3)
C(6)-C(5)-C(10)-C(11)	179.3(4)
C(4)-C(5)-C(10)-C(11)	-4.7(6)

Table 7. Hydrogen bonds for A [A and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
N(1)-H(1A)...N(2)#1	0.86	2.19	3.043(3)	169.1
N(1)-H(1B)...N(4)#2	0.86	2.18	3.043(4)	177.3
N(3)-H(3A)...O(2)#3	1.01(4)	1.79(4)	2.804(4)	176(4)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y,-z #2 -x,-y,-z #3 -x,-y,-z+1