Synthesis and Biological Evaluation of Bupropion Analogues as Potential Pharmacotherapies for Cocaine Addiction

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A series of bupropion (1a) analogues (1b-1ff) were synthesized, and their *in vitro* and *in vivo* pharmacological properties evaluated with the goal of developing a 1a analogue that had better properties for treating addictions. Their *in vitro* pharmacological properties were examined by [³H]dopamine ([³H]DA), [³H]serotonin ([³H]5HT), and [³H]norepinephrine ([³H]NE) uptake inhibition studies, and by binding studies at the dopamine, serotonin, and norepinephrine transporters using [¹²⁵I]RTI-55 in cloned transporters. Several analogues showed increased [³H]DA uptake inhibition with reduced or little change in [³H]5HT and [³H]NE uptake inhibition relative to bupropion. Thirty-five analogues were evaluated in a 1 h locomotor activity observation test and 32 in an 8 h locomotor activity observation test and compared to the locomotor activity of cocaine. Twenty-four analogues were evaluated for generalization to cocaine drug discrimination after i.p. administration, and twelve analogues were tested in a time course cocaine discrimination study using oral administration. 2-(*N*-Cyclopropylamino)-3-chloropropiophenone (1x) had the most favorable *in vitro* efficacy and *in vivo* pharmacological profile for an indirect dopamine agonist pharmacotherapy for treating cocaine, methamphetamine, nicotine, and other drugs of abuse addiction.

Introduction

Cocaine abuse continues to be a significant medical problem in the United States with an estimated 2.1 million current users of the drug by age 12 and older.¹ In addition to its direct effects, cocaine abuse has also contributed to the increase of the spread of human immunodeficiency virus (HIV) infection and drug-resistant tuberculosis. Even though considerable effort has been devoted to the development of a pharmaco-therapy to treat patients addicted to cocaine, no effective medication is yet available for use in the clinic.² Indirect dopamine agonists^{3–10} are one class of compounds

Indirect dopamine agonists^{3–10} are one class of compounds that have received considerable attention for treatment of cocaine addiction. Studies directed toward the development of indirect dopamine agonists have involved structurally diverse classes of compounds including analogues of 3-phenyltropane, 1,4-dialkylpiperazines (GBR), phenylpiperidine, benztropine, methylphenidate, and mazindol.^{3,4,8,11}

Compound **1a** [(\pm)-2-*tert*-butylamino-3'-chloropropiophenone] is a well-known antidepressant. A sustained release (SR^{*a*}) formulation of **1a** has proven to be highly useful for treating nicotine abuse.^{12–14} Despite its use as an antidepressant for almost 30 years,¹⁵ the neurochemical mechanism(s) underlying its action is still not well-defined.^{16,17} Compound 1a inhibits the reuptake of dopamine (DA) and norepinephrine (NE) but, unlike many other antidepressants, has very little effect on inhibiting serotonin (5HT) reuptake.^{17,18} Its antidepressant effects have been attributed to its effects on the noradrenergic system.^{17,19,20} This is based in part on the fact that **1a** is metabolized to an active metabolite (2S,3S)-3,5,5trimethyl-2-phenylmorpholin-2-ol (2), which is a relatively more potent NE uptake inhibitor and is present at higher steady-state concentrations than those of 1a.¹⁷ However, some reports suggest that **1a** is a more potent DA reuptake inhibitor than an NE reuptake inhibitor.²¹ Microdialysis studies showed that acute 1a administration increased extracellular DA.²⁰ In animal behavioral pharmacology studies, 1a induced locomotor activity,^{22,23} generalized to cocaine and amphetamine in drug discrimination studies,^{24,25} produced conditioned place preference (CPP),²⁶ and is self-administered in both rats²⁷ and nonhuman primates.²⁵ In addition, 1a increased responding on a fixed interval (FI) schedule stimulus-shock termination study in squirrel monkey.²⁸ These effects likely reflect 1a's action as a DA reuptake inhibitor, and thus, 1a has the properties of an indirect dopamine agonist.

In a clinical trial of 1a for cocaine abuse, an exploratory analysis suggested that patients with depression may have benefited.²⁹ In another clinical study, 1a augmented contingency management for cocaine dependence in methadone-maintained patients, but there was no evidence for efficacy of 1a alone.³⁰ Compound 1a has even greater

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^{*a*} Abbreviations: DA, dopamine; 5HT, serotonin; NE, norepinephrine; CPP, condition place preference; FI, fixed internal; SR, sustained release; DEG, diethylene glycol; PPA, polyphosphoric acid; ABSA, acetamidobenzenesulfonyl azide; DBU, 1,4-diazabicyclo[5.4.0]undec-7ene; HEK, human embryonic kidney; DAT, dopamine transporter, SERT, serotonin transporter; NET, norepinephrine transporter.

efficacy for methamphetamine dependence. In one study, **1a** reduced methamphetamine-induced subjective effects and cue-induced craving.³¹ In another study, **1a** reduced methamphetamine use in relatively light users.^{32,33}

Given 1a's dopmainergic properties and its success in dependence studies for cocaine, methamphetamine, and nicotine, it is surprising that very few 1a analogues have been synthesized and studied to identify an analogue with better overall pharmacological properties. The study presented here further examines the pharmacology of previously reported 1a analogues and reports the pharmacology of several new 1a analogues. The results are compared to the pharmacological properties of 1a and cocaine. The study's goal was to develop a 1a analogue with increased dopaminergic properties as determined by its DA uptake with reduced noradrenergic properties as determined by NE uptake properties to reduce the potential of cardiotoxicity.34 Since the increased dopaminergic properties could also increase abuse potential, the desired analogue would need to have slow onset and long duration of action properties that are believed to reduce abuse potential.^{35–38} Specifically, this study describes the synthesis of a number of 1a analogues (1a-1ff, 3-7) and reports their monoamine transporter binding properties, functional monoamine uptake inhibition efficacy, locomotor activity, and drug discrimination properties. 2-(N-tert-Butyl)-3'-chlorobutyrophenone (10), 2-(N-tert-butyl)-3'-chloropentanophenone (1p), and 2-(*N*-tert-butyl)-3'-chlorohexanophenone (1q) were 31, 29, and 3.6 times more potent than 1a as DA uptake inhibitors. 2-(N-Cyclopropylamino)-3-chloropropiophenone (1x), which was 4 times more potent as a DA uptake inhibitor, was also more potent than **1a** as a 5HT uptake inhibitor. When tested in vivo, 10, 1p, and 1x showed effects of long duration in a test of locomotor activity. When compared for generalization to cocaine in a drug discrimination test after oral administration, 1x also had a slower onset and longer duration of action than 1a.



Chemistry

Scheme 1 describes the general synthesis used for 1a-1ff. In general, the original procedure used to prepare $1a^{39}$ and modified by Chenard and co-workers⁴⁰ was followed. This

procedure provided an efficient way to the bromo intermediates 10 and the subsequent 1a analogues 1b-1ff. The procedure used to convert nitriles 8 to the corresponding ketones was similar to that of Birch and co-workers⁴¹ or Bailey and coworkers.⁴² 2-(*N*-tert-Butylamino)-1-(2'-thienyl)-1-propanone (3) was synthesized using a procedure similar to that used to synthesize the 1a analogue 1bb starting with ketone 11 (see Scheme 2). Syntheses of 1b, 1d, 1f, 1h, 1j, 1o, 1p, and 1w were reported in a communication as intermediates used in the synthesis of other compounds.⁴³ However, no experimental details and no characterization of the compounds were given.⁴³ The hydrochloride salt of 1b has been reported.⁴⁴ This compound was characterized as the fumarate salt in this study.

Scheme 3 shows the procedure used for the synthesis of 4. Subjection of 3'-chloropropiophenone (10a) to Mannich reaction conditions with aqueous formaldehyde and dimethylamine gave 13. The methiodide 14 was obtained by alkylation of 13 with iodomethane. Treatment of 14 with *tert*-butylamine gave first a mixture of 15 and the desired 4. Subjection of the mixture to excess *tert*-butylamine provided the desired target compound.

Compound 5 was synthesized by the route shown in Scheme 4. Ketone 17 was obtained in moderate yield by the addition of ethylmagnesium bromide to acid 16 in the presence of the catalyst bis(diphenylphosphinoethane)dichloronickel(II). Bromination of the ketone gave dibromo compound 18 and a small amount of monobromo derivative 19. Treatment of the crude mixture with *tert*-butylamine in refluxing toluene gave the desired target compound 5.

Scheme 5 outlines the synthesis of **1a** analogue **6**. Treatment of **20** with hydrazine and potassium hydroxide in refluxing diethylene glycol (DEG) afforded **21**. Compound **21** was cyclized to **22** using polyphosphoric acid (PPA). Subjection of **22** to acetamidobenzenesulfonyl azide (ABSA) in acetonitrile containing 1,4-diazabicyclo[5.4.0]undec-7-ene (DBU) yielded the diazo compound **23**. Treatment of **23** with *tert*-butylamine in dry toluene in the presence of ruthenium acetate provided **6**.

Compound 7 was prepared by reduction of 1a as previously reported.⁴³

Biology

The competition binding assays were performed using (h)DAT, (h)SERT, and (h)NET, stably expressed in HEK293 cells, and the nonselective radioligand [¹²⁵I]RTI-55 for the analogues **1a–1ff** and **3–7** (Table 1).⁴⁵ The HEK-(h)DAT, -(h)SERT, and -(h)NET cells were also used to evaluate the compounds' ability to block the reuptake of [³H]dopamine ([³H]DA), [³H]serotonin ([³H]5HT), and [³H]norepinephrine ([³H]NE) (Table 1).⁴⁵

The **1a** analogues were evaluated *in vivo* for their ability to increase locomotor activity of mice and to generalize to the discriminative stimulus effects of cocaine in rats, according to protocols established by the NIDA Cocaine Treatment Discovery Program. Each compound was first evaluated in a locomotor activity test^{46–48} lasting 1 h, and compounds were reconsidered in a second test (lasting 8 h) if necessary to assess durations of effect longer than 60 min. The locomotor test results for each compound are listed in Table 2 in terms of the ED₅₀ in mg/kg, each compound's maximal stimulant effect as a percent of cocaine's maximal effect, the 30 min time period in which the maximum locomotor stimulation occurred, and the duration of the locomotor stimulation.

Generalization to cocaine was subsequently evaluated for a subset of the **1a** analogues using rats trained to discriminate

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Scheme 1^a



^{*a*} Reagents: (a) RCH₂MgBr, $(C_2H_5)_2O$ or RLi, pentane/ $(C_2H_5)_2O$; (b) Br₂, CH₂Cl₂; (c) R₁, R₂NH.

Scheme 2^{*a*}



^a Reagents: (a) Br₂, CH₂Cl₂; (b) H₂NC(CH₃)₃.

Scheme 3^{*a*}





cocaine from saline in a lever choice procedure. 47,49,50 Table 3 lists the percent of cocaine-appropriate responding following various intraperitoneal doses of the analogues along with ED₅₀ values for cases when such responding reached at least 80%, suggesting generalization. Analogues were also evaluated for the time course of their generalization to cocaine when dosed p.o. in a volume of 1 mL/kg at 45, 90, 180, or 360 min before the generalization test. Table 4 lists the percent of cocaine-appropriate responding following various oral doses of the analogues at each time point, along with ED₅₀ values for cases when such responding reached at least 80%.

Results and Discussion

Compound 1a is an approved drug for treating depression¹⁵ and as a smoking cessation drug.¹²⁻¹⁴ In addition, it has

Scheme 4^{*a*}



^{*a*} Reagents: (a) EtMgBr, Ni(dppe)Cl₂, THF; (b) HCl/H₂O; (c) Br₂, CH₂Cl₂; (d) (CH₃)₃CNH₂, toluene, reflux.

Scheme 5^a



^{*a*} Reagents: (a) H_2NNH_2 , KOH, DEG; (b) PPA; (c) ABSA, DBU, CH₃CN; (d) Rh₂(OAc)₄, CH₃C₆H₅, (CH₃)₃CNH₂.

shown some positive effects in clinical trials for treatment of cocaine and methamphetamine dependence.^{29–33} While **1a**'s precise mechanism of action is unknown, it is a weak inhibitor of DA uptake and has been shown to increase DA transmission in both the nucleus accumbens and the prefrontal cortex.⁵¹ In addition, **1a** is a locomotor stimulant,^{22,23} generalizes to cocaine,^{24,25} produces CPP,²⁶ and is self-administered²⁷ in animal behavioral pharmacology studies. PET imaging studies show that administration of **1a** results in relatively low DAT occupancy.^{52–54} Thus, the modest

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Table 1. Compar	



								binding, ^{<i>a</i>} K_i (nM)		ſ'n	ptake, ^a IC ₅₀ (nM)	
compd	R	${f R}_1$	${f R}_2$	×	Y	Ζ	DAT	SERT	NET	[³ H]DA	[³ H]5-HT	[³ H]NE
cocaine							272 ± 60	601 ± 130	830 ± 147	267 ± 47	318 ± 57	385 ± 40
1 a	CH_3	Н	C(CH ₃) ₃	C	Η	Η	871 ± 126	$> 10 \mu { m M}$	6970 ± 2620	945 ± 213	c	443 ± 245
1b	CH ₃	Η	$C(CH_3)_3$	Η	Η	Η	5730 ± 480	$> 10 \mu M$	5700 ± 150	2310 ± 750	c	8700 ± 1200
lc	CH ₃	Н	$C(CH_3)_3$	Ĺ	Η	Η	4510 ± 460	$> 10 \mu M$	$> 10 \mu M$	1460 ± 220	c	c
1d	CH_3	Н	C(CH ₃) ₃	Br	Η	Η	4200 ± 1200	$> 10 \mu M$	$> 7140 \pm 640$	950 ± 210	c	6500 ± 1000
le	CH_3	Н	C(CH ₃) ₃	CH_3	Η	Η	$> 10 \mu M$	$> 10 \mu { m M}$	$> 10 \mu M$	с	c	c
lf	CH_3	Н	$C(CH_3)_3$	Η	Ū	Η	2195 ± 151	$> 10 \mu M$	$> 10 \mu M$	2319 ± 429	c	c
1_{g}	CH ₃	Н	$C(CH_3)_3$	Η	\mathbf{Br}	Η	1918 ± 221	4170 ± 730	$> 10 \mu M$	1295 ± 375	2520 ± 610	c
1h	CH_3	Н	$C(CH_3)_3$	Η	CH ₃	Η	$> 10 \mu { m M}$	$> 10 \mu{ m M}$	6840 ± 1368	с	c	4100 ± 860
li	CH_3	Н	C(CH ₃) ₃	ц	Ц	Η	$> 10 \mu M$	$> 10 \mu M$	$> 10 \mu M$	с	c	c
1j	CH_3	Н	C(CH ₃) ₃	C	ū	Η	472 ± 81	1480 ± 310	5400 ± 1200	271 ± 96	$> 10 \mu M$	2100 ± 380
1k	CH_3	Н	C(CH ₃) ₃	Ū	CH ₃	Η	1150 ± 370	2100 ± 510	5100 ± 860	650 ± 150	400 ± 190	900 ± 130
11	CH_3	Н	C(CH ₃) ₃	CH_3	Br	Η	1740 ± 440	1215 ± 90	4600 ± 601	950 ± 310	473 ± 74	1623 ± 35
1m	CH_3	Н	C(CH ₃) ₃	ĹĻ	Η	Ц	5660 ± 490	$> 10 \mu { m M}$	$> 10 \mu M$	5600 ± 1800	c	c
1n	CH_3	Η	C(CH ₃) ₃	C	Η	Ū	$> 10 \mu { m M}$	$> 10 \mu M$	$> 10 \mu M$	с	c	c
10	C_2H_5	Н	C(CH ₃) ₃	Ū	Η	Η	459 ± 50	$> 10 \mu { m M}$	3195 ± 145	31 ± 9	c	969 ± 410
1p	C_3H_7	Н	C(CH ₃) ₃	Ū	Η	Η	96 ± 20	$> 10 \mu M$	1171 ± 260	33 ± 7	c	472 ± 93
1q	C_4H_9	Η	C(CH ₃) ₃	C	Η	Η	350 ± 100	$> 10 \mu M$	3190 ± 850	69 ± 23	c	400 ± 190
1r	C ₅ H ₁₁	Н	C(CH ₃) ₃	Ū	Η	Η	709 ± 8.0	$> 10 \mu M$	4300 ± 1300	1570 ± 570	c	2000 ± 1000
1s	$C_{6}H_{13}$	Н	C(CH ₃) ₃	C	Η	Η	510 ± 120	$> 10 \mu { m M}$	2580 ± 700	135 ± 80	c	4890 ± 580
1t	(CH ₃) ₂ CHCH ₂	Н	C(CH ₃) ₃	Ū	Η	Η	140 ± 14	6200 ± 1800	2300 ± 700	440 ± 180	$> 10 \mu M$	360 ± 190
1u	$(C_6H_{11})CH_2CH_2$	Н	C(CH ₃) ₃	Ū	Η	Η	1700 ± 780	6200 ± 1500	3800 ± 1700	$> 10 \mu M$	$> 10 \ \mu M$	$> 10 \mu { m M}$
1v	CH ₃	Н	$CH_2CH_2CH_3$	G	Η	Η	$> 10 \mu M$	$> 10 \mu M$	$> 10 \mu M$	c	c	c
1w	CH ₃	Н	$CH(CH_3)_2$	Ū	Η	Η	3980 ± 230	$> 10 \mu M$	$> 10 \mu M$	2000 ± 516	c	c
1x	CH_3	Η	$CH(CH_2CH_2)$	C	Η	Η	1150 ± 370	3420 ± 260	4000 ± 1200	265 ± 94	3180 ± 170	2150 ± 850
1y	CH_3	Η	$CH(CH_2CH_2CH_2)$	Ū	Η	Η	343 ± 72	3450 ± 610	1800 ± 110	258 ± 46	185 ± 49	86 ± 32
1z	CH_3	Н	CH(CH ₂ CH ₂ CH ₂ CH ₂)	ū	Η	Η	2200 ± 490	$> 10 \mu M$	5700 ± 400	980 ± 340	c	221 ± 94
1aa	$C2H_5$	Н	$C(CH_3)_3$	Ū	0	Η	278 ± 43	860 ± 230	2240 ± 510	175 ± 55	790 ± 320	135 ± 4.9
1bb	$C3H_7$	Η	$C(CH_3)_3$	Ū	Ū	Η	43.7 ± 6.5	842 ± 50	520 ± 100	84 ± 28	1580 ± 560	43 ± 14
1cc	CH ₃	CH_3	$C(CH_3)_3$	Ū	Η	Η	6400 ± 1200	$> 10 \mu M$	$> 10 \mu M$	2060 ± 340	c	c
1dd	CH ₃	CH_3	CH_3	Ū	Η	Η	4133 ± 548	$> 10 \mu M$	3090 ± 740	1534 ± 222	c	1260 ± 290
lee	CH_3	CH_2CH_3	CH_2CH_3	C	Η	Η	2214 ± 308	$> 10 \mu M$	4476 ± 107	1744 ± 288	c	4603 ± 986
1ff	CH ₃	CH_2	;CH2CH2CH2CH2	Ū	Η	Η	1148 ± 298	5479 ± 824	4760 ± 1540	1033 ± 287	970 ± 178	4798 ± 947
e							$> 10 \mu M$	$> 10 \mu M$	$> 10 \mu M$	с	c	c
4				C	Η	Η	$> 10 \mu { m M}$	$> 10 \mu M$	$> 10 \mu M$	с	c	c
S						Η	$> 10 \mu { m M}$	$> 10 \mu M$	$> 10 \mu M$	c	c	c
9							$> 10 \mu M$	$> 10 \mu M$	$> 10 \mu M$	c	c	c
7							$> 10 \mu { m M}$	4000 ± 1400	$> 10 \mu M$	с	1240 ± 210	c
^a Value	as for the mean \pm stand	ard error of t	hree independent experimen	its, each co	onducted	with tri	plicate determinati	on. ^b Data taken f	rom ref 43. c Not det	ermined.		

Table 2. Comparison of Locomotor Stimulant Effects for 1a Analogues

compd	$\mathrm{ED}_{50}{}^a\mathrm{mg/kg}$	95% conf ^b	% peak ^c cocaine	time of max effect ^{d} (min)	duration ^e (min)
cocaine	7.8 ± 0.45^{f}		100	0-30	40-100
1a	5.5	(3.4 - 8.9)	87	0-30	130-270
1b	11.2	(6.0 - 21.4)	95	0-30	120
1c	17.5	(11.0-27.5)	80	0-30	150
1d	13.5	(8.5 - 21.4)	57	0-30	90-120
1e	31.0	NC ^g	76	10-40	50
1f	NE^{h}				
1g	29.5	(18.6 - 46.8)	54	140-170	160
1h	3.8	(1.8 - 8.4)	26	30-60	60-140
1i	54.6	NC^{g}	58	10-40	60
1j	26.9	(9.4 - 76.8)	61	270-300	140-240
1k	\geq 35.5 ⁱ		\geq 77 ^{<i>i</i>}	10-40	60
11	12.6	(6.4 - 25.1)	36	50-80	240
1m	35.0	(22.4 - 54.3)	84	0-30	280
1n	NE^h				
10	18.6	(10.7 - 31.6)	62	0-30	360
1p	15.5	(10.5 - 22.9)	108	90-120	210-480
1q	10.2	(5.5 - 19.0)	136	0-30	40-460
1r	12.9	(5.5 - 29.5)	90	60-90	280-460
1s	5.6	(2.4 - 12.9)	94	110-140	130-400
1t	25.7	(8.7 - 74.1)	89	260-290	60
1u	14.1	(5.8 - 33.9)	33	60-90	160
1v	NE^{h}				
1w	6.2	(2.8 - 14.1)	46	20-50	50-160
1x	16.6	(11.9 - 22.9)	82	0-30	350
1z	12.3	(9.1-16.2)	107	0-30	330-340
1aa	38.0	(14.8-95.5)	69	0-30	60
1bb	38.0	(24.0 - 61.7)	93	0-30	470
1cc	11.7	(8.3-16.6)	95	80-110	200-340
1dd	9.3	NC^{g}	64	90-120	180-190
1ee	16.6	(10.5 - 26.3)	90	20-50	130-240
1ff	34.7	(16.2 - 74.1)	45	320-350	30-150
3	NE^{h}				
4	NE^{h}				
5	NE^{h}				
6	2.6	NC^{g}	41	150-180	390-480
7	NE^{h}				

^{*a*} Dose producing 50% of the compound's maximal effect. ^{*b*} 95% confidence interval based on logistic fit to ascending portion of dose response. ^{*c*} Compound's maximal effect as a percent of cocaine's maximal effect. ^{*a*} The 30 min period in which the maximal effect occurred. ^{*c*} Duration of stimulant effect for one or more doses on ascending portion of dose response. ^{*f*} Mean \pm SD for 28 evaluations of cocaine's stimulant effect. ^{*s*} Could not be calculated. ^{*h*} No stimulant effect detected. ^{*f*} Peak effect not determined.

effect of **1a** could be in part due to its weak dopaminergic effects.

Much research, particularly in animals, suggests that repeated administration of cocaine produces disruption in brain dopamine (DA) functions that can lead to enhanced cocaine-seeking behavior. Reversal of these changes in DA activity can attenuate these cocaine-induced neurochemical and behavioral effects.55,56 In the early 1990s, we and others hypothesized that a compound having good potency and selectivity for the dopamine transporter (DAT) combined with a slow onset and long-duration of activity relative to cocaine would reverse these changes in DA activity and would, therefore, be useful pharmacotherapy for cocaine addiction. An optimal compound would have no or low abuse potential itself.^{4,6,9,55,56} While not exactly analogous, this is similar to the use of methadone for opiate addiction and varenicline or nicotine replacement therapy for tobacco smoking (nicotine addiction). Therefore, a 1a analogue possessing increased dopaminergic properties may be able to reverse dopaminergic deficits in chronic cocaine users better than 1a. This study identified 1a analogues that had increased dopaminergic properties combined with slow onset and long duration of activity as determined

by their DA uptake, locomotor, and drug discrimination properties.

In this study, we report that **1a** analogues with better DAT binding (lower K_i values) and [³H]DA uptake (lower IC₅₀ values) were obtained by (a) replacing the methyl group α to the ketone group with medium-sized alkyl groups; (b) changing the type and number of substituents on the 3-chlorophenyl ring; and (c) replacing the *N*-tert-butyl group with other *N*-alkyl groups. Since for the most part the rank order potency of the binding assays mirrors those of the uptake values, only the monoamine uptake values will be discussed.

Compound **1a** has IC₅₀ values of 945 and 443 nM for DA and NE uptake inhibition, respectively (Table 1). Since the K_i value for binding to the SERT was > 10 μ M, the 5HT uptake IC₅₀ was not determined. Thus, **1a** is 3.5 times less potent as an inhibitor of DA uptake than cocaine. Compound **1a** and cocaine have almost equal potency for NE uptake, and cocaine with an IC₅₀ value of 318 nM for 5HT uptake is much more potent than **1a**. Analogues **1o**-**1q** obtained by replacing the α -methyl group in **1a** with an ethyl, propyl, or butyl group had IC₅₀ values of 31, 33, and 69 nM, respectively, compared to 945 nM for **1a**, and were the most DA efficacious analogues. Analogues **1s**-**1t** with larger hexyl and isobutyl

 Table 3. Effect of 1a Analogues in Rats Trained to Discriminate Cocaine after IP Administration

			dos	se (mg/k	cg), % c	ocaine-l	ever resp	ponding				
compd	pretreatment Time (min)	vehicle	cocaine	1	2.5	5	10	25	50	100	$ED_{50} \ (mg/kg)$	comments ^a
1a	15	0	83	24.1	0.7	0	67.1	66.3	66.7			А
1b	15	0	100		16	32.8	83.3	83.4			5.9	В
											(3.8-9.1)	
1c	15	1	83		17.6	0	50.2	100			10.0	В
											(0.6 - 182.0)	
1d	30	0	100		0.3	34.4	21	83.6			11.6	В
16	15	0	0.2			0.1	22.7	10.1	0	22.2	(6.0 - 22.6)	D
11	15	0	83			0.1	32.7	18.1	0	33.3		В
lg	80	2	100			0.0	1./	16./	20	0.5	20 (В
Ih	15	0	83			0.6	0	16./	66./	100	39.6	C
1:	45	0	100				17	17	47	62.0	(29.3-33.7)	C
11	45	0	100				1.7	167	47	03.9		B
1j 1)	50	0	100				0	0	16.7	66.9		D C
1m	30	0	100		0	43.6	16.7	50	83.3	00.9	17.3	B
	50	0	100		0	12.0	10.7	50	00.0		(6.6-45.0)	D
10	15	19	83		0.6	33.4	50	100			7.8	С
											(4.9 - 12.5)	-
1p	15	0	100		0.2	17.2	33.3	83.3			12.2	D
											(7.4 - 20.1)	
1q	30	11	100		0	17.2	66.6	100			7.8	Е
-											(5.7 - 10.7)	
1r	45	17	100				0	1	50.9			F
1s	75	18	99			0	0	48.5	49.9	67.2		В
1u	60	0	100			0.2	0.7	69.6				\mathbf{B}^{b}
$1w^c$	15	1	90	21.2	27	43.8	53.3	78.6	73.5	94.9	10.0	
d											(3.6 - 27.4)	
$1x^a$	15	1	100	21.5	39.5	67.2	66.7	66.3				F
1z	30	1	100	0.9	23.9	0	66.2	83.4			8.8	G
	20	0	100	0		16.5	10	100			(6.1-12.5)	
	30	0	100	0	33.3	16.7	10	100			10.9	В
100	30	0	100	0	41.8	33.3	33.3	100			0.3	В
1.00	15	0	100			0	40.4	75	667		(2.3-15.5)	F
1dd	15	2	100		0	0 82.2	49.4	100	00.7		2.0	F G
Tuu	15	3	100		0	03.3	63.2	100			(2, 5-6, 1)	U
100	15	2	83		0	16.7	98.6	74.8			(2.3-0.1)	н
100	15	4	05		U	10.7	20.0	70			(3.6-9.7)	11
1ff	335	0	100				7	16.5	32.7	33.7	(3.0).1)	В
cocaine	555	4.5	89.1	14.7	39.2	60.6	, 84.9	10.0	52.7	55.1	3.2	2
(n = 66)			0,			00.0	0				(2.7 - 3.8)	
(()	

^{*a*} Response rate comments: A = The average response rate was increased relative to vehicle control following 5 to 25 mg/kg with a maximum effect at 5 mg/kg (127% of vehicle control). The average response rate decreased to 30% of vehicle control following 50 mg/kg **1a**. B = Response rate failed to show significant change. C = Response rate was decreased following 100 mg/kg. D = Response rate increased following 25 mg/kg. E = Response rate was increased following 50–100 mg/kg. G = Response rate was decreased following 25 mg/kg. H = Response rate was decreased following 25 mg/kg. ^{*b*} Adverse effects were seen at 50 mg/kg. ^{*c*} Compound **1w** was tested at 0.5 mg/kg and showed 0.3% cocaine-lever responding. ^{*d*} Compound **1x** was tested at 0.1, 0.25, and 0.5 mg/kg and showed % cocaine-lever responding of 9%, 24%, and 63%. ^{*e*} Confidence interval could not be calculated.

 α substituents had IC₅₀ values of 135 and 440 nM and, thus, were also better DA uptake inhibitors than **1a**. Replacement of the α -methyl group in **1a** with a much larger 2-(cyclohexyl)ethyl α substituent to give **1u** resulted in complete loss of DA uptake inhibition (IC₅₀ value of > 10 μ M).

Changing the aromatic substituent pattern of **1a** also led to analogues with better IC_{50} values for DA uptake inhibition. For example, the 3,4-dichlorophenyl analogue **1j** and the 3chloro-4-methylphenyl analogue **1k** with IC_{50} values of 271 and 650 nM, respectively, were 3.5 and 2 times more potent inhibitors than **1a**. The 3-bromophenyl and 4-bromo-3-methylphenyl analogues **1d** and **1l** with IC_{50} values of 950 nM were as potent as **1a**. Replacing the α methyl group of **1j** with an ethyl or propyl group gave analogues **1aa** and **1bb**, which had slightly lower IC_{50} values than **1j**. Replacement of the 3-chlorophenyl ring with a thiophene ring led to **3**, which had no efficacy for DA uptake inhibition.

Replacement of the *N-tert*-butyl group with an *N*-cyclopropyl or *N*-cyclobutyl group gave 1x and 1y with IC₅₀ values of 265 and 258 nM for DA uptake inhibition, which are 3.6 and 3.7 times more potent than 1a. The *N*-cyclopentyl analogue 1z had an IC₅₀ of 980 nM, almost identical to that of 1a. The *N*-isopropyl analogue 1w with an IC₅₀ value of 2000 nM was 2 times less potent than 1a. Surprisingly, the *N*-propyl analogue 1v was inactive. None of the *N*,*N*-disubstituted analogues 1c-1ff had high efficacy for DA uptake inhibition. Of these, the most potent compound was the *N*-piperidino analogue 1ff, which had an IC₅₀ value of 1033 nM.

Overall, the DA uptake results show that changing the α -methyl group of **1a** to a larger ethyl (**1o**), *n*-propyl (**1p**), or

Table 4. Drug Discrimination Effects of 1a and 1a Analogues in Rats (p.o.) in a Time Course Study

compd	pretreat-ment time	vehicle	2.5	5	10	25	50	100	200	$ED_{50} \left(mg/kg\right)$	comments ^a
1a	45	0		33	17	50	83			22.8	А
	90			0	1	0	50			(12.0 - 43.2)	
	180			2.5	0	22	17				
	360			0	0	1	0				
1b	45	11	6	0	0	51	83			25.2	А
	90		1	0	0	33	33			(26.5 - 38.4)	
	180		0	33	3	20	50				
	360		0	0	0	34	4				
1c	45	0		17	17	49	67	95		25.3	В
	90			0	0	16	33	75		(13.7 - 46.9)	
	180			0	33	0	18	83			
	360			0	0	0	0	50			
1d	45	0	13	0	0	17	51	67	100	52.9	C, D
	90		0	15	0	0	21	56	100	(33.0 - 84.7)	
	180		0	0	0	0	0	34	5		
	360		33	7	0	0	0	14	34		
1m	45		0	31	0	4	67	18		41.5	E
	90	0	20	7	33	25	58	84		(23.4 - 73.6)	
	180		0	0	0	0	20	17			
	360		0	0	3	7	0	0			
1p	45	0	0	17	33	33 ^b	50^c	100^{d}		23.2	F
	90		0	0	0	33 ^b	17^c	100^{d}		(11.9 - 45.4)	
	180		0	22	33	0 %	17^{c}	50^{d}			
	360		0	0	0	0^{b}	16^{c}	1^d			
1q	45	0		17	0	0	83			46.9^{e}	А
	90			0	0	0	17				
	180			26	29	0	17				
	360			0	0	0	0				
1s	45	0			0	0	83			46.9 ^e	G
	90				0	0	17				
	180				0	0	25				
	360				0	0	0				
$1x^e$	45		34	34	74	67				12.8	Н
	90	0	0	0	33	83				(8.2 - 19.9)	
	180		34	0	33	99					
	360		0	0	0	0					
1z	45	1	17	50	60	67	100			6.6	A, I
	90		NT	0	33	67	83			(3.0 - 14.6)	
	180		NT	0	0	66	33				
	360	. –	NT	0	0	34	16				
laa	45	17	33	17	34	50	83			15.2	А
	90		17	1	0	33	67			(4.5 - 51.2)	
	180		50	0	0	2	50				
an f	360		17	0	2	33	16			•• • • <i>P</i>	
1bb/	45	. –	0	0	33	58				22.8°	А
	90	17	0	13	0	84					
	180		0	0	5	17					
	360	~	0	0	0	0				5.0	
ldd	45	0	19	38	83					5.3	А
	90		0	0	100					(3.1-9.1)	
	180		33	0	33						
	360		0	0	18						

^{*a*} Response rate comments: A = No significant change in response rate. B = Response rate was reduced at 50 and 100 mg/kg. C = Response rate was decreased at 200 mg/kg. D = Four of six rats failed to complete the first fixed ratio at 180 min following 200 mg/kg. E = Response rate was decreased 90 min following 50 mg/kg. F = Response rate was decreased following 2.5 and 5 mg/kg at 45 min. G = Response rate was increased relative to vehicle control 45 min following 10 mg/kg. H = Response rate failed to show significant change at the 90 min pretreatment interval. I = Decreased food consumption was observed following 25 mg/kg (1/24 rats) and 50 mg/kg (1/24 rats). ^{*b*} Dose = 20 mg/kg. ^{*c*} Dose = 40 mg/kg. ^{*d*} Dose = 80 mg.kg. ^{*e*} Confidence intervals could not be calculated. ^{*f*} Compound **1x** was also studied at other doses. At 0.25 mg/kg, the % cocaine lever responding was 0 for all time points. ^{*g*} Compound **1f** was tested at 1 mg/kg and showed 0% lever pressing at all time points.

n-butyl (**1q**) group gave the greatest increase in [³H]DA uptake inhibition relative to **1a**. In addition, adding a 4'-chloro (**1j**) group to **1a** gave an increase in [³H]DA uptake inhibition relative to **1a**. Compounds with an α -ethyl or α -*n*-propyl substituent combined with a 4'-chloro substituent (**1aa**, **1bb**) did not show increased [³H]DA uptake inhibition relative to **10** and **1p**, respectively. Replacement of the *N*-tert-butyl group with an *N*-cyclopropyl (**1x**) or *N*-cyclopentyl (**1y**) resulted in a small increase in [³H]DA uptake inhibition. Overall, 11 **1a** analogues from this study had lower IC₅₀ values for DA

uptake inhibition ($IC_{50} = 31 \text{ nM}$ to 650 nM) than **1a** ($IC_{50} = 945 \text{ nM}$). The conformationally restricted **1a** analogues **5** and **6** had no efficacy at all three transporters.

Similarly to 1a, most of the 1a analogues from this study showed little efficacy for 5HT uptake inhibition. The 3-chloro-4-methylphenyl, 3-methyl-4-bromophenyl, and *N*-cyclopentyl analogues 1k, 1l, and 1y with IC₅₀ values of 400, 473, and 185 nM for 5HT uptake inhibition, respectively, were the most potent.

Since there is concern that a pharmacotherapy having norepinergic activity might enhance the cardiotoxicity of cocaine, we hoped that our **1a** analogues would have reduced $[^{3}H]NE$ uptake inhibition potency compared to that of **1a**. Several of the potent DA uptake inhibitors (**1o**-**1q**, **1s**, and **1x**) had lower efficacy for NE uptake inhibition than **1a**. However, analogues **1bb**, **1y**, and **1aa** with IC₅₀ values of 43, 86, and 135 nM for NE uptake inhibition, respectively, were 34, 17, and 10 times better as NE uptake inhibitors than **1a**. While these analogues are not of interest for development of a cocaine pharmacotherapy, they would be of interest for the development of **1a** analogues as potential treatment for smoking cessation or possibly as antidepressants.

Meltzer and co-workers⁵⁷ used the DAT inhibitor pyrovalerone (**24a**) as a lead structure to develop new monoamine uptake inhibitors. In his study, Meltzer pointed out that pyrovalerone has structural features similar to **1a**. The *N*-cyclopentyl **1a** analogue **1z** in the present study is most similar to pyrovalerone. However, **1z** was not as potent as pyrovalerone as a DA and NE uptake inhibitor. Meltzer synthesized several analogues of **24a**, where the aromatic substituent, the amine group, and the substituent α to the carbonyl were changed. Several analogues had lower IC₅₀ values for DA and NE uptake inhibition than **24a**.



Replacement of the cyclopentyl group in 1z with a cyclobutyl or cyclopropyl group to give 1y and 1x, respectively, resulted in significant changes in monoamine uptake properties. Compound 1z had an IC50 value of 980 nM for DA uptake inhibition compared to 258 and 265 nM for 1y and 1x, respectively. Surprisingly, the IC_{50} value for NE uptake inhibition for 1z of 221 nM improved to 86 nM for the cyclobutyl analogue 1y but increased to 2150 nM for the cyclopropyl analogue 1x. The cyclopentyl analogue 1z was totally inactive as a 5HT uptake inhibitor, whereas the cyclobutyl analogue 1y has an IC₅₀ of 185 nM for this transporter, which is better than the other 1a analogues studied (Table 1). Changing the cyclopentyl in 1z to the cyclopropyl group in 1x also resulted in improved 5HT uptake inhibition but only to an IC_{50} value of 3180 nM. Meltzer⁵⁷ found that opening the cyclopentyl ring of 24d to give 25 resulted in a large loss in uptake inhibition at all three transporters. We found that 1v, which can be viewed as the open ring analogue of 1x, was inactive at all three transporters. Reduction of the carbonyl of 24a to give alcohol 26

resulted in total loss of affinity at all three transporters. We found that reduction of the carbonyl of **1a** analogue **1b** to give 7 resulted in loss of affinity at the DAT and NET but gave a significant increase in potency for 5HT uptake from $> 10\,000$ nM in **1a** to 1240 nM in **7**.

Thirty-five 1a analogues were evaluated for their ability to increase locomotor activity, and the results were compared to results obtained for cocaine in separate studies conducted on a monthly basis. Cocaine had an ED₅₀ value of 7.8 ± 0.45 mg/kg in 28 studies conducted over the period in which the 1a analogues were evaluated, whereas the ED_{50} value for 1a was 5.5 mg/kg. Cocaine's locomotor stimulant efficacy (maximum locomotor activity minus activity of vehicle control) was set at 100%, and efficacy of each analogue was expressed relative to that for cocaine as determined during the same month. Four analogues, 1q, 1s, 1w, and 1dd, yielded ED_{50} values similar to that of cocaine ($ED_{50} =$ 5.6–10.2). The ED₅₀ values for 28 compounds ranged from 2.6 mg/kg for 6 to 54.6 mg/kg for 1i, whereas 7 analogues failed to yield locomotor stimulant effects at any time within 8 h following injection. Compound 1a and ten analogues, 1b, 1m, 1p, 1r, 1s, 1t, 1z, 1bb, 1cc, and 1ee, showed locomotor stimulant efficacy similar to cocaine (84-107%). Compound 1q with 136% was more stimulatory than cocaine. The remaining compounds had peak stimulant effects of 26-82% of that of cocaine. Several compounds, 1g, 1j, 1p, 1r, 1s, 1t, 1u, 1cc, 1dd, 1ff, and 6 had periods of maximum stimulatory effect at times of 1 h or greater following injection. The period of maximum stimulatory effect of 1a was 0-30 min, which was similar to that of cocaine and analogues 1b-1d, 1m, 1o, 1q, and 1x-1bb. Compounds 1j, 1t, and 1ff yielded a very slow onset of stimulatory activity, with peak effects delayed by 260-320 min following injection. Compound 1a had a stimulant effect of approximately 2-4.5 h duration. Sixteen of the analogues, 1b, 1j, 1l, 1m, 1o, 1p, 1q, 1r, 1s, 1x, 1z, 1bb, 1cc, 1dd, 1ee, and 6, had duration of locomotor activity greater than 3 h. Analogues 10-1q, which were 31, 29, and 14 times more potent than **1a** as DA uptake inhibitors, yielded very long durations of locomotor activity stimulation (360 to 480 min). In addition, analogue 1x, which was 3.6 times more potent than 1a as a DA uptake inhibitor with increased potency as a 5HT uptake inhibitor, possessed a duration of locomotor effect of 350 min, which is longer than 1a. Analogue 10 also had a much longer duration of effect compared with 1a. It is also interesting to note that the conformationally rigid analogue 6 yielded a weak stimulant effect (31% of cocaine) that was of very slow onset and long duration.

Meltzer and co-workers⁵⁷ evaluated pyrovalerone analogues **14b** and **24c** for locomotor activity, which had IC₅₀ values of 67.9 and 40 nM, respectively, for DA uptake inhibition. Compounds **24b** and **24c** had ED₅₀ values of 0.21 and 2.2 mg/kg with long durations of action. The cyclopentyl **1a** analogue **1z** had an ED₅₀ value of 12.3 mg/kg with long durations of action (Table 2).

Twenty-four **1a** analogues were evaluated using drug discrimination tests for generalization to cocaine after i.p. administration in rats using standard 2-lever operant chambers, with pretreatment intervals adjusted based on studies of locomotor activity (Table 3). Twelve analogues were also tested in a time course study using oral administration (Table 4). Fourteen **1a** analogues showed full generalization to cocaine with ED_{50} values of 3.9 mg/kg for **1dd** to 39.6 mg/kg for **1h** in the i.p. cocaine discrimination study. Compound **1a** substituted only partially for the discrimination stimulus effects of cocaine. The lowest dose yielding maximum substitution (10 mg/kg) resulted in 67% cocaine-appropriate responding. As pointed out in the Introduction, others have reported that **1a** shows full generalization in cocaine discrimination studies in animals including rats after i.p. administration.^{24,58,59} However, response rate was decreased relative to vehicle control following 5-25 mg/kg doses. The failure to obtain full generalization was likely due to the greater response suppression observed in this sample of rats.

Compound 1dd with an $ED_{50} = 3.9 \text{ mg/kg}$ was the most potent of the analogues, producing full generalization at doses of 5 to 25 mg/kg, even though its potency at all three transporters was relatively weak. Analogue 1b fully generalized at 10 and 25 mg/kg. Analogues 1b, 1c, 1d, 1o, 1p, 1q, 1z, 1aa, 1bb, and 1dd all showed full generalization at a dose of 25 mg/kg. Analogue 1m showed full generalization at a dose of 50 mg/kg and analogues 1h and 1w at 100 mg/kg. The *N*-cyclopropyl analogue 1x produced partial generalization from 5 to 25 mg/kg, 1cc from 10 to 50 mg/kg, and 1u at 25 mg/ kg. Higher doses were not tested due to decreases in response rate or adverse effects. Analogue 1s produced partial generalization from 25 to 100 mg/kg, 1i from 50 to 100 mg/kg, and 11 at 100 mg/kg. Doses greater than 100 mg/kg were not tested in the i.p. studies.

All 12 analogues tested in the time-course study showed full generalization in at least one time point. ED_{50} values were calculated for the 45 min pretreatment with the exception of 1m, 1x, and 1bb, which were calculated at the 90 min pretreatment. ED₅₀ values ranged from 5.3 for 1dd to 52.9 mg/kg for 1d. Compound 1a showed full generalization at 45 min following a 50 mg/kg dose and partial generalization at 90 min after a 50 mg/kg dose. The ED₅₀ for drug-appropriate responding 45 min following 1a was 22.8 mg/kg. Analogue 1dd showed full generalization at 45 and 90 min after a 10 mg/kg dose. Analogue 1x showed partial generalization at 45 min following a 10 mg/kg and 25 mg/kg dose and full generalization at 90 and 180 min at a 25 mg/kg dose. Analogue 1x was the only compound to show full generalization at 180 min. Analogue 1bb also showed full generalization at 90 min at a dose of 25 mg/kg. Similar to 1a, analogues 1b, 1q, 1s, and 1aa showed full generalization at one time point at a dose of 50 mg/kg. Analogue 1z showed full generalization at 45 and 90 min following 50 mg/kg. Analogues 1c and 1p showed full generalization at a dose of 100 mg/ kg and **1d** at a dose of 200 mg/kg.

Even though most of the analogues that had low IC₅₀ values for DA uptake also had low ED₅₀ values for cocaine generalization, there were several exceptions. For example, the most potent analogue in the cocaine discrimination studies was 1dd with an ED₅₀ value of 3.9 mg/kg in the i.p. study and 5.3 mg/kg in the p.o. time course study. However, the IC₅₀ value for 1dd for DA uptake inhibition was 1534 nM compared to 943 nM for 1a. Analogue 1j with an IC₅₀ value of 271 nM was 3.5 times more potent as a DA uptake inhibitor than 1a but did not even show partial generalization to cocaine. In the Introduction, it was pointed out that 1a is metabolized to yield the active metabolite 2 in vivo that likely contributes to its therapeutic effects. The nature of the metabolites associated with 1dd, 1j, and possibly other analogues is unknown, but it is conceivable that metabolites with different profiles of activity could account for the imperfect correlation between the in vitro and in vivo studies.

In summary, a number of 1a analogues showed monoamine uptake inhibition efficacy and an animal behavior profile that suggested they might be better indirect dopamine agonists than 1a. Analogues 10-1q and 1x had the best overall profiles, with 1x being the most interesting. Compound 1x was four times more potent than 1a in the DA uptake inhibition test and was more selective for DA uptake relative to NE uptake inhibition than 1a. Compounds 1o-1q were 30, 29, and 14 times more potent than 1a as DA uptake inhibitors raising concern that they may have too much dopaminergic activity for treatment of cocaine addiction. Unlike 1a and 1o-1q, 1x also has some efficacy as a 5HT uptake inhibitor. Studies from our laboratory as well as others have reported animal behavioral studies that show that reduction of cocaine self-administration can be enhanced by 5HT uptake inhibition.⁶⁰ Compound 1x in the initial drug discrimination study showed partial generalization to cocaine at the dose of 5, 10, and 25 mg/kg compared to 1a, which showed partial generalization at a dose of 10 and 25 mg/kg. Analogues 10-1q all showed full generalization at 25 mg/kg and, thus, were more cocaine-like. More importantly, 1x was more potent than 1a in the time-course discrimination study and had a slower onset and longer duration of action. The in vitro efficacy and animal behavioral properties thought to be necessary for an indirect dopamine agonist pharmacotherapy for treating abuse of cocaine, methamphetamine, and nicotine are both better for 1x than for 1a.

Experimental Section

Nuclear magnetic resonance (¹H NMR and ¹³C NMR) spectra were recorded on a 300 MHz (Bruker AVANCE 300) or 500 MHz (Varian Unity ANOVA) spectrometer. Chemical shift data for the proton resonances were reported in parts per million (δ) relative to internal (CH₃)₄Si (δ 0.0). Elemental analyses were performed by Atlantic Microlab, Norcross, GA. Purity of compounds (>95%) was established by elemental analysis. Analytical thin-layer chromatography (TLC) was carried out on plates precoated with silica gel GHLF (250 μ M thickness). TLC visualization was accomplished with a UV lamp or in an iodine chamber. All moisture-sensitive reactions were performed under a positive pressure of nitrogen maintained by a direct line from a nitrogen source. Anhydrous solvents were purchased from Aldrich Chemical Co. or VWR.

Synthesis of 2-(*N-tert*-Butylamino)-3-chlorobutanophenone (10) Fumarate. The synthesis of the title compound is given as a typical example used for the synthesis of 1a analogues 1b-1ff. Experimental details for the synthesis of each analogue can be found in Supporting Information.

Step 1. 3'-Chlorobutanophenone (90). 3-Chlorobenzonitrile 8d (3.0 g, 0.022 mol) and THF (75 mL) were placed in a 250 mL flask equipped with a magnetic stir bar. The flask was cooled to 0 °C with an ice-water bath. Propylmagnesium chloride (26.2 mL, 2 M in Et₂O) was syringed in over a 10 min period. The reaction was stirred under nitrogen at room temperature. After 96 h, the flask was cooled to 0 °C. The reaction was quenched by adding 0.1 M hydrochloric acid (75 mL). After stirring for 1 h at room temperature, the solution was transferred to a separatory funnel. Water (50 mL) and ammonium hydroxide (2 mL) were added to basify the reaction, and the aqueous layer was extracted $3 \times$ with methylene chloride. The organic layer was dried (Na2SO4) and filtered. The solvent was removed under reduced pressure to give 3.51 g (88%) of 90 as a light-yellow oil. ¹H NMR (CDCl₃) δ 7.95 (s, 1H), 7.81-7.87 (d, 1H), 7.50-7.56 (d, 1H), 7.38-7.40 (t, 1H), 2.90-2.95 (t, 2H), 1.72-1.81 (m, 2H), 0.99-1.05 (t, 3H).

Step 2. 2-Bromo-3'-chlorobutanophenone (100). Ketone **90** (3.51 g, 0.01 mol) and methylene chloride (75 mL) were placed

in a 500 mL flask equipped with a magnetic stir bar. The solution was stirred under nitrogen, and bromine (0.98 mL, 0.019 mol) was syringed into the flask. A small amount of bromine was added initially to catalyze the reaction. After reaction started, the remaining bromine was added over a 10 min period. After stirring for 14 h, the solution was transferred to a separatory funnel. A saturated sodium bicarbonate solution was added to basify the reaction. The aqueous layer was washed with a 1 M sodium thiosulfate solution and extracted $3 \times$ with methylene chloride. The organic layer was dried (Na2SO4) and filtered. The solvent was removed under reduced pressure to give 5.20 g of an oil. The orange oil was purified by flash chromatography on silica gel using 5:1 hexane-methylene chloride as eluent to afford 4.04 g (80%) of 10o as a colorless oil. ¹H NMR (CDCl₃) δ 8.00 (s, 1H), 7.86-7.91 (d, 1H), 7.54-7.59 (d, 1H), 7.41-7.48 (t, 1H), 4.99-5.05 (t, 1H), 2.07-2.30 (m, 2H), 1.07-1.12 (t, 3H).

Step 3. 2-(*N*-tert-Butylamino)-3'-chlorobutanophenone (10) Fumarate. Intermediate 10o (3.90 g, 0.015 mol) and tert-butylamine (7.84 mL, 0.075 mol) were placed in a pressure tube equipped with a magnetic stir bar. The tube was sealed and heated at 75 °C with an oil bath. After 2 h, the reaction mixture was cooled to room temperature and transferred to a separatory funnel. A saturated sodium bicarbonate solution was added to basify the reaction, and the aqueous layer was extracted with methylene chloride (3×). The organic layer was dried (Na₂SO₄), and the solvent was removed under reduced pressure. The oil was dissolved in methanol, and the solvent was removed under reduced pressure to afford 3.51 g (93%) of 1o as a pale-yellow oil. ¹H NMR (CDCl₃) δ 7.95 (s, 1H), 7.84–7.89 (d, 1H), 7.53– 7.58 (d, 1H), 7.40–7.48 (t, 1H), 4.04–4.10 (m, 1H), 2.04–2.24 (m, 2H), 1.02 (s, 9H), 0.92–0.99 (t, 3H).

Amine **10** was converted to a fumarate salt by adding 1 equiv of fumaric acid to an Et₂O solution of **10**. Recrystallization from methanol and Et₂O afforded 2.64 g of **10** · fumarate as a white solid: mp 155–156 °C. ¹H NMR (CD₃OD) δ 8.20 (s, 1H), 8.10–8.15 (d, 1H), 7.76–7.81 (d, 1H), 7.60–7.68 (t, 1H), 6.70 (s, 2H), 5.20–5.25 (t, 1H), 2.01–2.11 (m, 2H), 1.37 (s, 9H), 1.15–1.22 (t, 3H). Anal. (C₁₈H₂₄NO₅) C, H, N.

2-Methyl-3-(*N*,*N*-dimethylamino)-3'-chloropropiophenone (13). 3'-Chloropropiophenone 10a (5.0 g, 0.03 mol) and methanol (50 mL) were placed in a pressure tube equipped with a magnetic stir bar. Aqueous formaldehyde (2.44 mL, 37% by weight) and aqueous dimethylamine (4.09 mL, 40% by weight) were added. The tube was sealed and refluxed in an oil bath at 75 °C. After 18 h, the reaction mixture was cooled to room temperature. Hydrochloric acid (4 mL) was added, and the reaction mixture was stirred for 2 h. The solvent was removed under reduced pressure. The reaction mixture was transferred to a separatory funnel and extracted $3 \times$ with methylene chloride. The organic layer was dried (Na_2SO_4) and filtered. The solvent was removed under reduced pressure to afford 5.73 g (86%) of 13 as a paleyellow oil. ¹H NMR (CDCl₃) δ 7.94 (s, 1H), 7.82–7.87 (d, 1H), 7.51-7.56 (d, 1H), 7.39-7.46 (t, 1H), 3.60-3.70 (q, 1H), 2.75-2.85 (m, 1H), 2.30-2.39 (m, 1H), 2.23 (s, 6H), 1.18-1.22 (d. 3H).

2-Methyl-3-(trimethylammonia)-3'-chloropropiophenone Iodide (14). Amine 13 (5.73 g, 0.025 mol) and methanol (50 mL) were placed in a pressure tube equipped with a magnetic stir bar. Iodomethane (1.74 mL, 0.03 mol) was added. After 2 days, more iodomethane (0.6 mL) was added. After stirring for 7 days at room temperature, the solution was filtered through a fritted funnel and washed with methanol. The solvent was removed under reduced pressure. The salt was recrystallized from isopropanol and Et₂O. The solution was filtered and washed with cold Et₂O. The solvent was removed under reduced pressure to afford 5.56 g (60%) of 14 as light-yellow crystals. ¹H NMR (CDCl₃) δ 8.16 (s, 1H), 8.11–8.16 (d, 1H), 7.69–7.74 (d, 1H), 7.56–7.62 (t, 1H), 4.24–4.34 (m, 2H), 3.42–3.52 (m, 1H), 3.15 (s, 9H), 1.31–1.34 (d, 3H). 2-Methylene-3'-chloropropiophenone (15). Quaternary amine salt 14 (4.0 g, 0.011 mol), sodium carbonate (1.27 g, 0.01 mol), and DMF (50 mL) were placed in a pressure tube equipped with a magnetic stir bar. The mixture was stirred to dissolve the sodium carbonate. After 30 min, *tert*-butylamine (1.26 mL, 0.01 mol) was added. The tube was sealed and heated in an oil bath at 70 °C. After 5 h, the tube was cooled to room temperature, and the reaction mixture was transferred to a separatory funnel. Water (50 mL) and ammonium hydroxide (10 drops) were added to basify the reaction, and the aqueous layer was extracted 3× with Et₂O. The organic layer was dried (Na₂SO₄) and filtered. The solvent was removed under reduced pressure to give 3.4 g of 14 and 15 as a light-yellow oil. ¹H NMR (CDCl₃): 2× as many aromatic peaks (thus, two compounds), alkene shown by δ 5.96 (s, 1H), 5.64 (s, 1H), 2.05 (s, 3H).

2-Methyl-3-(*N-tert*-butylamino)-3'-chloropropiophenone (4) Fumarate. The mixture of 14/15 (3.4 g) and an excess of *tert*butylamine (6 mL) were placed in a pressure tube, and the reaction mixture was stirred at room temperature. The solution was transferred to a separatory funnel and basified with a saturated sodium bicarbonate solution. The aqueous layer was extracted $3\times$ with methylene chloride. The organic layer was dried (Na₂SO₄) and filtered. The solvent was removed under reduced pressure to give 2.14 g (78%) of **4** as a light-yellow oil. ¹H NMR (CDCl₃) δ 7.94 (s, 1H), 7.82–7.87 (d, 1H), 7.52–7.57 (d, 1H), 7.39–7.46 (t, 1H), 3.52–3.60 (m, 1H), 3.01–3.10 (m, 1H), 2.60–2.69 (m, 1H), 1.20–1.25 (d, 3H), 1.11 (s, 9H).

Amine 4 was converted to a fumarate salt using the same procedures as for 10. Recrystallization from methanol and Et₂O afforded 2.53 g of 4 · fumarate as a white solid: mp 128–129 °C. ¹H NMR (CD₃OD) δ 8.01 (s, 1H), 7.94–7.99 (d, 1H), 7.63–7.68 (d, 1H), 7.51–7.58 (t, 1H), 6.65 (s, 2H), 3.85–3.96 (m, 1H), 3.50–3.60 (m, 1H), 3.04–3.11 (m, 1H), 1.41 (s, 9H), 1.27–1.30 (d, 3H). Anal. (C₁₈H₂₄ClNO₅) C, H, N.

3'-Chloro-6'-methylpropiophenone (17). To a stirred solution of 3-chloro-6-methylbenzoic acid 16 (12.5 g, 0.073 mol) in dry THF (100 mL) was added bis(diphenylphosphinoethane)dichloronickel(II) catalyst (794 mg, 1.5 nmol). After stirring under nitrogen at 0 °C for 15 min, ethylmagnesium bromide (147 mL, 1 M in THF) was added via a cannula. After the initial exothermic reaction, the reaction mixture turned green, then black. Subsequently, the remaining ethylmagnesium bromide (100 mL, 3 M in THF) was added. After stirring for 16 h at room temperature, the reaction was quenched with 10% hydrochloric acid/water and extracted with Et₂O. The organic layer was washed with sodium bicarbonate, dried (Na₂SO₄), and filtered. The solvent was removed under reduced pressure to give 13.02 g of an oil. The yellow oil was purified by flash chromatography on silica gel using 9:1 hexane-Et₂O as eluent to afford 6.87 g (51%) of 17 as a pale-yellow oil. ¹H NMR (CDCl₃) δ 7.57 (s, 1H), 7.30-7.34 (d, 1H), 7.16-7.19 (d, 1H), 2.85-2.93 (q, 2H), 2.44 (s, 3H), 1.15–1.23 (t, 3H).

2-Bromo-3'-chloro-6' a-bromomethylpropiophenone (18). To a stirred solution of ketone **17** (4.79 g, 0.026 mol) in chloroform (100 mL) was added bromine (3.4 mL, 0.066 mol). The reaction mixture was heated with a heat gun until hydrogen bromide evolution began. Upon completion, the solvent was removed under reduced pressure to give 6.15 g (69%) of **18** as an orange oil. ¹H NMR (CDCl₃) δ 7.71 (s, 1H), 7.48–7.54 (d, 1H), 7.23–7.27 (d, 1H), 5.11–5.19 (q, 1H), 4.66–4.76 (dd, 2H), 1.91–1.93 (d, 3H). The monobrominated product **19** was also formed. ¹H NMR (CDCl₃) δ 7.55 (s, 1H), 7.34–7.38 (d, 1H), 7.24–7.27 (d, 1H), 5.08–5.16 (q, 1H), 2.44 (s, 3H), 1.87–1.90 (d, 3H).

2-tert-Butyl-6-chloro-3-methyl-4-oxo-1,2,3,4-tetrahydroisoquinoline (5) Hydrochloride. To a solution of **18** (6.15 g, 0.018 mol) in toluene (300 mL) was added *tert*-butylamine (4 mL, 38 mmol). After refluxing 4 h under nitrogen, the resulting slurry was basified with ammonium hydroxide and extracted with methylene chloride. The organic layer was dried (Na_2SO_4) and filtered. The solvent was removed under reduced pressure, and toluene was azeotroped with ethanol and water (5:3:1) to give the cyclized amine as an oil. The orange oil was purified by flash chromatography on silica gel using 9:1 hexane–Et₂O as eluent to afford 1.26 g (28%) of **5** as a pale-yellow oil. ¹H NMR (CDCl₃) δ 7.95 (s, 1H), 7.43–7.47 (d, 1H), 7.17–7.20 (d, 1H), 4.00–4.17 (m, 3H), 1.23–1.26 (d, 3H), 1.15 (s, 9H).

The isoquinoline **5** was converted to a hydrochloride salt and was recrystallized from methanol– Et_2O to afford 740 mg of **5**·HCl as an off-white solid: mp 182 °C (dec). ¹H NMR (CDCl₃) δ 8.06 (s, 1H), 7.68–7.71 (d, 1H), 7.35–7.38 (d, 1H), 4.88–4.95 (d, 1H), 4.55–4.62 (d, 1H), 4.42–4.44 (q, 1H), 1.94–1.97 (d, 3H), 1.47 (s, 9H). Anal. (C₁₄H₁₉C₁₂NO) C, H, N.

4-(4-Chlorophenyl)butyric Acid (21). A heterogeneous mixture of 4-(4-chlorophenyl)-4-oxobutanoic acid (5.00 g, 0.0235 mol), potassium hydroxide (3.5 g of 85%, 0.0522 mol), hydrazine monohydrate (2.57 g, 0.0514 mol), and diethylene glycol (21 mL) were heated in a flask equipped with a Dean-Stark trap and condenser. The mixture became homogeneous on heating. The heating of both was maintained at 120-130 °C for 1.5 h and then raised to 180 °C for 3 h. The reaction mixture was cooled to ambient temperature, diluted with water (25 mL), and poured into 2.5 M hydrochloric acid (30 mL). The mixture was allowed to stand for 16 h and the white amorphous solid collected by filtration. To remove the residual diethylene glycol, the solid was dissolved in saturated potassium carbonate (50 mL) and diluted with water (100 mL). The clear solution was poured carefully into stirred 2.5 M hydrochloric acid (50 mL). White crystals formed immediately and were collected by filtration, washed with water $(2 \times 200 \text{ mL})$, and dried under vacuum. This resulted in 3.9 g (83%) of **21**. ¹H NMR (CDCl₃) δ 7.26 (s, 2H), 7.12 (s, 1 H), 2.56 (t, 2H, J = 6 Hz), 2.37 (t, 2H, J = 9 Hz), and 2.13 (p, 2H, J = 6 Hz).

7-Chloro-1-oxo-1,2,3,4-tetrahydronaphthalene (22). Polyphosphoric acid (20 g, excess) was place in a beaker and heated to 90 °C on a steam bath. 4-(4-Chlorophenyl)butyric acid (27, 0.017 mol) was added in portions. The mixture was stirred for 5 min. An additional portion of polyphosphoric acid (20 g, excess) was added and heated to 90 °C for 5 min. The thick, homogeneous viscous orange oil was removed from the steam bath and cooled to 60 °C before water (200 mL) was added. When the reaction was complete (all the orange oil gone) and the mixture had cooled to ambient temperature, the mixture was extracted with ether $(2 \times 100 \text{ mL})$. The ethereal extracts were washed with water ($2 \times 100 \text{ mL}$), 1 N sodium hydroxide ($2 \times 100 \text{ mL}$) mL), water (100 mL), aqueous acetic acid (100 mL of 3%), saturated sodium bicarbonate (100 mL), and finally with water (100 mL). The ethereal layer was dried (MgSO₄) and concentrated to give 2.52 g (82%) of 22 as a white amorphous solid. ¹H NMR (CDCl₃) δ 7.85 (d, 1H, J = 6 Hz), 7.42 (d, 1H, J = 6 Hz), 7.21 (d, 1H, J = 9 Hz), 2.94 (m, 2H), 2.63 (m, 2H), 2.15 (m, 2H).

7-Chloro-2-diazo-l-oxo-l,2,3,4-tetrahydronaphthalene (23). 7-Chloro-1-oxo-1,2,3,4-tetrahydronaphthalene 22, (2.0 g, 0.011 mol) and acetamidobenzenesulfonyl azide (5.33 g, 0.022 mol) were dissolved in acetonitrile (50 mL) and cooled in an ice bath. A solution of 1,4-diazabicyclo[5.4.0]undec-7-ene (3.37 g, 0.0221 mol) in acetonitrile (5 mL) was added dropwise. The temperature was maintained at 0 °C for 2 h. The mixture was allowed to slowly warm to room temperature and stirred a total of 18 h. The purple solution was poured into 1 N sodium hydroxide (100 mL) and extracted with Et_2O (3 × 100 mL). The combined organic fractions were dried (MgSO₄) and concentrated. The black solid was passed through silica gel eluting with hexane and gradually increasing the polarity by adding ethyl acetate (until a 1:1 mixture was obtained). Concentration of the product fraction afforded 1.6 g of **23** as a bright-yellow solid. ¹H NMR (CDCl₃) δ 8.26 (s, 1H), 7.39 (d, 1H, J = 6 Hz), 7.13 (d, 1H, J = 15 Hz), 2.99 (s, 4H).

2-(*tert*-**Butylamino**)-**7-**chlorotetralone (6) Fumarate. Ruthenium acetate (0.8 g, 0.0018 mol) and *tert*-butylamine (5 g, 0.668 mol) were dissolved in dry toluene (100 mL). The mixture

was warmed in a 115 °C oil bath, creating a homogeneous lightpurple solution. A solution of 7-chloro-2-diazo-1-oxo-1,2,3,4tetrahydronaphthalene (1.50 g, 0.007 mol) in toluene (20 mL) was added dropwise over 20 min. The resulting dark-purple solution was cooled to ambient temperature and poured into 10% hydrochloric acid (50 mL), and the layers were separated. The aqueous layer was made alkaline by pouring it into a slurry of ice (20 g) and conc. ammonium hydroxide (20 mL). The resulting pink slurry was extracted with $Et_2O(2 \times 50 \text{ mL})$, dried (MgSO₄), and concentrated to give 0.382 g of yellow resin, which quickly discolored to purple when exposed to air. The resin was promptly dissolved in acetone and poured into a solution of fumaric acid (0.176 mg, 0.0015 mol) in warm acetone. The resulting precipitate was collected by filtration, triturated with acetone $(2 \times 150 \text{ mL})$, and then dried, resulting in 461 mg of $6 \cdot$ fumarate: mp softens 146 °C (dec) 196–199 °C. ¹H NMR (D₂O) δ 7.68 (s, 1H), 7.36 (d, 1H, J = 8 Hz), 7.09 (d, 1H, J = 8 Hz), 6.35 (s, 2H), 4.22 (dt, 1H, J = 10 Hz, J = 4.5 Hz), 2.99-2.87 (m, 2H), 2.3-2.07 (m, 2H), 1.14 (s, 9H). Anal. (C₁₈H₂₂ClNO₅) C, H, N.

Inhibition of Radioligand Binding of [125 I]RTI-55 to hDAT, hSERT, or hNET in Clonal Cells. Cell Preparation. HEK293 cells expressing hDAT, hSERT, or hNET inserts are grown to 80% confluence on 150 mm diameter tissue culture dishes and serve as the tissue source. Cell membranes are prepared as follows. Medium is poured off the plate, and the plate is washed with 10 mL of calcium- and magnesium-free phosphate-buffered saline. Lysis buffer (10 mL; 2 mM HEPES with 1 mM EDTA) is added. After 10 min, cells are scraped from plates, poured into centrifuge tubes, and centrifuged $30\,000 \times g$ for 20 min. The supernatant fluid is removed, and the pellet is resuspended in 12-32 mL of 0.32 M sucrose using a Polytron at setting 7 for 10 s. The resuspension volume depends on the density of binding sites within a cell line and is chosen to reflect binding of 10% or less of the total radioactivity.

Assay Conditions. Each assay tube contains 50 μ L of membrane preparation (about 10–15 μ g of protein), 25 μ L of unknown, compound used to define nonspecific binding, or buffer (Krebs-HEPES, pH 7.4; 122 mM NaCI, 2.5 mM CaCl₂, 1.2 mM MgS0₄, 10 μ M pargyline, 100 μ M tropolone, 0.2% glucose, and 0.02% ascorbic acid, buffered with 25 mM HEPES), $25 \,\mu$ L of [¹²⁵I]RTI-55 (40–80 pM final concentration), and additional buffer sufficient to bring up the final volume to $250\,\mu$ L. Membranes are preincubated with unknowns for 10 min prior to the addition of the $[^{125}I]RTI-55$. The assay tubes are incubated at 25 °C for 90 min. Binding is terminated by filtration over GF/C filters using a Tomtec 96-well cell harvester. Filters are washed for 6 s with ice-cold saline. Scintillation fluid is added to each square, and radioactivity remaining on the filter is determined using a Wallac μ - or β -plate reader. Specific binding is defined as the difference in binding observed in the presence and absence of 5μ M mazindol (HEK-hDAT and HEK-hNET) or $5 \,\mu$ M imipramine (HEK-hSERT). Two or three independent competition experiments are conducted with duplicate determinations. GraphPAD Prism is used to analyze the ensuing data, with IC₅₀ values converted to K_i values using the Cheng-Prusoff equation $(K_i = IC_{50}/(1 + ([RTI - 55]/K_d RTI - 55))))$

Filtration Assay for Inhibition of [³H]Neurotransmitter Uptake in HEK293 Cells Expressing Recombinant Biogenic Amine Transporters. Cell Preparation. Cells are grown to confluence as described above. The medium is removed, and cells are washed twice with phosphate buffered saline (PBS) at room temperature. Following the addition of 3 mL Krebs-HEPES buffer, the plates are warmed in a 25 °C water bath for 5 min. The cells are gently scraped and then triturated with a pipet. Cells from multiple plates are combined. One plate provides enough cells for 48 wells, which is required to generate data on two complete curves for the unknowns.

Uptake Inhibition Assay Conditions. The assay is conducted in 96 1-mL vials. Krebs-HEPES (350 μ L) and unknowns, compounds used to define nonspecific uptake, or buffer (50 μ L) are added to vials and placed in a 25 °C water bath. Specific uptake is defined as the difference in uptake observed in the presence and absence of 5 μ M mazindol (HEK-hDAT and HEK-hNET) or 5 μ M imipramine (HEK-hSERT). Cells (50 μ L) are added and preincubated with the unknowns for 10 min. The assay is initiated by the addition of [³H]dopamine, [³H]serotonin, or [³H]norepinephrine (50 μ L, 20 nM final concentration). Filtration through Whatman GF/C filters presoaked in 0.05% polyethylenimine is used to terminate uptake after 10 min. The IC₅₀s are calculated applying the GraphPAD Prism program to triplicate curves made up of six drug concentrations each. Two or three independent determinations of each curve are made.

Locomotor Activity Studies.⁶¹ Locomotor activity of mice within 10-min epochs was measured under dim illumination using a Digiscan apparatus (model RXYZCM-16, Omnitech Electronics, Columbus, OH) consisting of 40 testing chambers $(40.5 \times 40.5 \times 30.5 \text{ cm}^3)$ each surrounded by a panel of infrared beams and photodetectors. Compound 1a analogues were initially evaluated for ability to increase locomotor activity of mice during a test lasting 1 h, following i.p. injection of the vehicle or 1 of 4-10 doses of the test compound (n = 8 mice per dose group). Incremental doses were tested until (i) a locomotor stimulant effect was evident and a peak or plateau of the dose-effect curve could be defined, (ii) a locomotor depressant dose-response could be defined, or (iii) there was no effect in doses up to 100 mg/kg. For analysis of stimulant potency and efficacy, the earliest 30 min period in which a maximal stimulant effect first appeared as a function of dose was considered. The maximal effect, measured in locomotor activity counts, was estimated by fitting a 3-parameter logistic peak or transition function (Tablecurve 2D v2.03; Jandel Scientific, San Rafael, CA) with the constant set to the mean activity counts of the vehicle control. Stimulant efficacy of each analogue (maximal activity counts-vehicle control counts) is reported as a percentage of that calculated for cocaine as determined monthly in a separate study. An ED₅₀ was estimated as the center of a logistic dose-response function fit to the activity count data for the time period of maximal effect. A majority of the 1a analogues yielded stimulant effects lasting longer than 1 h and were reconsidered in studies lasting a total of 8 h. The duration of the stimulant effect was estimated for all doses yielding a statistically significant stimulant effect that was equal to or less than the maximal effect, and reported as a range in Table 2.

Drug Discrimination Studies. These studies were conducted using standard behavior-testing chambers (Coulbourn Instruments, Allentown, PA) interfaced to computers programmed with MED-PC IV (Med Associates, East Fairfield, VT) for the operation of the chambers and collection of data. All rats were first trained to discriminate cocaine (10 mg/kg) from saline using a two-lever choice methodology. Ten minutes prior to each training session, the rats received an injection of either saline (S) or cocaine (C) and were placed in the test chamber which contained two response levers. A food pellet became available after every 10 responses on only one of the levers (a designated cocaine- or saline-appropriate lever). Each training session lasted until the rats earned 20 food pellets or for a maximum of 20 min. The rats received training sessions in a double alternating fashion (i.e., C-C-S-S-C, etc.) until they met a criterion of 85% or greater injection-appropriate responding during 9 of their last 10 sessions. All rats had received approximately 60 training sessions before they were used in the generalization experiments. In contrast to the training sessions, both levers were active during generalization tests, such that 10 consecutive responses on either lever yielded a pellet. Cocaineand saline-appropriate responding was reconfirmed in standard training sessions conducted in between generalization tests. Different doses of **1a** analogues were injected i.p. prior to the generalization tests, using pretreatment times and starting doses adjusted based on the results of locomotor activity studies. Different doses of a given analogue were tested incrementally in the same group of six rats until (i) full generalization was evident, (ii) the rate of lever responding in the group was decreased to 20% of vehicle control, or (iii) toxicity was evident. Drug discrimination data were expressed as the mean percentage of responses on the cocaine-appropriate lever occurring in each generalization test, whereas the rates of responding were expressed as a function of the number of responses made divided by the total session time. Full generalization was defined as $\geq 80\%$ cocaine-appropriate lever responding and partial generalization as $\geq 40\%$ and < 80% cocaine-appropriate responding. For all compounds yielding full generalization, an ED₅₀ was estimated as the center of a logistic dose-response transition function fit to the percentage of cocaine-appropriate lever responding for all doses tested.

In studies of the time-course of generalization to cocaine, rats that had been trained for i.p. cocaine discrimination were administered **1a** analogues p.o., by gavage, in a volume of 1 to 6 mL/kg body weight. Generalization tests were then performed 45, 90, 180, or 360 min later in separate groups of 3-6 rats. Different doses of each analogue were tested incrementally beginning with starting doses determined by the i.p. discrimination studies, until full substitution was evident at one or more time points. A total of 6 rats were tested at each time point for the highest dose tested, and for vehicle and all doses tested at the shortest time point yielding full generalization (for which an ED₅₀ value was calculated).

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Supporting Information Available: Experimental details for the synthesis of target compounds **1b–1ff** and results from elemental analysis. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- Substance Abuse and Mental Health Services Administration-Office of Applied Studies. Results from the 2007 National Survey on Drug Use and Health: National Findings. In Rockville, MD, 2008.
- (2) Herman, B. H.; Elkashef, A.; Vocci, F. Medications for the treatment of cocaine addiction: Emerging candidates. *Drug Discovery Today: Therapeutic Strategies* 2005, 2, 87–92.
- (3) Trudell, M. L.; Izenwasser, S. Dopamine transporters. Chemistry, biology, and pharmacology; John Wiley & Sons, Inc.: New York, 2008; Chapters 6–9, p 444.
- (4) Carroll, F. I.; Howard, J. L.; Howell, L. L.; Fox, B. S.; Kuhar, M. J. Development of the dopamine transporter selective RTI-336 as a pharmacotherapy for cocaine abuse. *AAPS J.* 2006, 8, E196–E203.
- (5) Runyon, S. P.; Carroll, F. I. Dopamine transporter ligands: recent developments and therapeutic potential. *Curr. Top. Med. Chem.* 2006, *6*, 1825–1843.
- (6) Grabowski, J.; Shearer, J.; Merrill, J.; Negus, S. S. Agonist-like, replacement pharmacotherapy for stimulant abuse and dependence. *Addict. Behav.* 2004, 29, 1439–1464.
- (7) Howell, L. L.; Wilcox, K. M. The dopamine transporter and cocaine medication development: drug self-administration in nonhuman primates. J. Pharmacol. Exp. Ther. 2001, 298, 1–6.
- (8) Carroll, F. I.; Howell, L. L.; Kuhar, M. J. Pharmacotherapies for treatment of cocaine abuse: Preclinical aspects. J. Med. Chem. 1999, 42, 2721–2736.

- (9) Mello, N. K.; Negus, S. S. Preclinical evaluation of pharmacotherapies for treatment of cocaine and opioid abuse using drug selfadministration procedures. *Neuropsychopharmacology* **1996**, *14*, 375–424.
- (10) Rothman, R. B.; Glowa, J. R. A review of the effects of dopaminergic agents on humans, animals, and drug-seeking behavior, and its implications for medication development. *Mol. Neurobiol.* 1995, *11*, 1–19.
- (11) Mishra, M.; Kolhatkar, R.; Zhen, J.; Parrington, I.; Reith, M. E.; Dutta, A. K. Further structural optimization of cis-(6-benzhydrylpiperidin-3-yl)-benzylamine and 1,4-diazabicyclo[3.3.1]nonane derivatives by introducing an exocyclic hydroxyl group: interaction with dopamine, serotonin, and norepinephrine transporters. *Bioorg. Med. Chem.* 2008, *16*, 2769–2778.
- (12) Wilkes, S. The use of bupropion SR in cigarette smoking cessation. Int. J. Chron. Obstruct. Pulmon. Dis. 2008, 3, 45–53.
- (13) Tong, E. K.; Carmody, T. P.; Simon, J. A. Bupropion for smoking cessation. A review. *Comprehensive Therapy* 2006, *32*, 26–33.
- (14) Hurt, R. D.; Sachs, D. P.; Glover, E. D.; Offord, K. P.; Johnston, J. A.; Dale, L. C.; Khayrallah, M. A.; Schroeder, D. R.; Glover, P. N.; Sullivan, C. R.; Croghan, I. T.; Sullivan, P. M. A comparison of sustained-release bupropion and placebo for smoking cessation. *N. Engl. J. Med.* **1997**, *337*, 1195–1202.
- (15) Fava, M.; Rush, A. J.; Thase, M. E.; Clayton, A.; Stahl, S. M.; Pradko, J. F.; Johnston, J. A. 15 years of clinical experience with bupropion HCl: from bupropion to bupropion SR to bupropion XL. Prim. Care Companion J. Clin. Psychiatry 2005, 7, 106–113.
- (16) Ascher, J. A.; Cole, J. O.; Colin, J. N.; Feighner, J. P.; Ferris, R. M.; Fibiger, H. C.; Golden, R. N.; Martin, P.; Potter, W. Z.; Richelson, E.; et al. Bupropion: A review of its mechanism of antidepressant activity. J. Clin. Psychiatry 1995, 56, 395–401.
- (17) Dhillon, S.; Yang, L. P.; Curran, M. P. Bupropion: a review of its use in the management of major depressive disorder. *Drugs* 2008, 68, 653–689.
- (18) Ferris, R. M.; Cooper, B. R. Mechanism of antidepressant activity of bupropion. *J. Clin. Psychiatry Monogr.* **1993**, *11*, 2–14.
 (19) Cryan, J. F.; Dalvi, A.; Jin, S. H.; Hirsch, B. R.; Lucki, I.; Thomas,
- (19) Cryan, J. F.; Dalvi, A.; Jin, S. H.; Hirsch, B. R.; Lucki, I.; Thomas, S. A. Use of dopamine-beta-hydroxylase-deficient mice to determine the role of norepinephrine in the mechanism of action of antidepressant drugs. J. Pharmacol. Exp. Ther. 2001, 298, 651–657.
- (20) Li, S. X.; Perry, K. W.; Wong, D. T. Influence of fluoxetine on the ability of bupropion to modulate extracellular dopamine and norepinephrine concentrations in three mesocorticolimbic areas of rats. *Neuropharmacology* **2002**, *42*, 181–190.
- (21) Kelley, J. L.; Musso, D. L.; Boswell, G. E.; Soroko, F. E.; Cooper, B. R. (2*S*,3*S*,5*R*)-2-(3,5-Difluorophenyl)-3,5-dimethyl-2-morpholinol: A novel antidepressant agent and selective inhibitor of norepinephrine uptake. J. Med. Chem. **1996**, 39, 347–349.
- (22) Nielsen, J. A.; Shannon, N. J.; Bero, L.; Moore, K. E. Effects of acute and chronic bupropion on locomotor activity and dopaminergic neurons. *Pharmacol., Biochem. Behav.* **1986**, *24*, 795–799.
- (23) Nomikos, G. G.; Damsma, G.; Wenkstern, D.; Fibiger, H. C. Effects of chronic bupropion on interstitial concentrations of dopamine in rat nucleus accumbens and striatum. *Neuropsychopharmacology* **1992**, *7*, 7–14.
- (24) Jones, C. N.; Howard, J. L.; McBennett, S. T. Stimulus properties of antidepressants in the rat. *Psychopharmacology (Berl.)* 1980, 67, 111–118.
- (25) Lamb, R. J.; Griffiths, R. R. Self-administration in baboons and the discriminative stimulus effects in rats of bupropion, nomifensine, diclofensine and imipramine. *Psychopharmacology (Berl.)* **1990**, *102*, 183–190.
- (26) Ortmann, R. The conditioned place preference paradigm in rats: effect of bupropion. *Life Sci.* 1985, 37, 2021–2027.
- (27) Tella, S. R.; Ladenheim, B.; Cadet, J. L. Differential regulation of dopamine transporter after chronic self-administration of bupropion and nomifensine. J. Pharmacol. Exp. Ther. 1997, 281, 508– 513.
- (28) Spealman, R. D.; Madras, B. K.; Bergman, J. Effects of cocaine and related drugs in nonhuman primates. II. Stimulant effects on schedule-controlled behavior. *J. Pharmacol. Exp. Ther.* **1989**, 251, 142–149.
- (29) Margolin, A.; Kosten, T. R.; Avants, S. K.; Wilkins, J.; Ling, W.; Beckson, M.; Arndt, I. O.; Cornish, J.; Ascher, J. A.; Li, S. H.; et al. A multicenter trial of bupropion for cocaine dependence in methadone-maintained patients. *Drug Alcohol Depend.* **1995**, *40*, 125– 131.
- (30) Poling, J.; Oliveto, A.; Petry, N.; Sofuoglu, M.; Gonsai, K.; Gonzalez, G.; Martell, B.; Kosten, T. R. Six-month trial of bupropion with contingency management for cocaine dependence in a methadone-maintained population. *Arch. Gen. Psychiatry* 2006, 63, 219–228.

- (31) Newton, T. F.; Roache, J. D.; De La Garza, R.; Fong, T.; Wallace, C. L.; Li, S. H.; Elkashef, A.; Chiang, N.; Kahn, R. Bupropion reduces methamphetamine-induced subjective effects and cueinduced craving. *Neuropsychopharmacology* **2006**, *31*, 1537–1 544.
- (32) Elkashef, A. M.; Rawson, R. A.; Anderson, A. L.; Li, S. H.; Holmes, T.; Smith, E. V.; Chiang, N.; Kahn, R.; Vocci, F.; Ling, W.; Pearce, V. J.; McCann, M.; Campbell, J.; Gorodetzky, C.; Haning, W.; Carlton, B.; Mawhinney, J.; Weis, D. Bupropion for the treatment of methamphetamine dependence. *Neuropsychopharmacology* **2008**, *33*, 1162–1170.
- (33) Shoptaw, S.; Heinzerling, K. G.; Rotheram-Fuller, E.; Steward, T.; Wang, J.; N., S. A.; De La Garza, R.; Newton, T.; Ling, W. Randomized, placebo-controlled trial of bupropion for the treatment of methamphetamine dependence. *Drug Alcohol Depend*. 2008, 96, 222–232.
- (34) Zugibe, F. T.; Breithaupt, M.; Costello, J. Cardiotoxic mechanisms and interrelationships of cocaine: including a single case depicting several of these mechanisms. J. Clin. Forensic Med. 1998, 5, 140– 146.
- (35) Volkow, N. D. Stimulant medications: how to minimize their reinforcing effects? Am. J. Psychiatry 2006, 163, 359–361.
- (36) Volkow, N. D.; Ding, Y. S.; Fowler, J. S.; Wang, G. J.; Logan, J.; Gatley, J. S.; Dewey, S.; Ashby, C.; Liebermann, J.; Hitzemann, R.; et al. Is methylphenidate like cocaine? Studies on their pharmacokinetics and distribution in the human brain. *Arch. Gen. Psychiatry* **1995**, *52*, 456–463.
- (37) Quinn, D. I.; Wodak, A.; Day, R. O. Pharmacokinetic and pharmacodynamic principles of illicit drug use and treatment of illicit drug users. *Clin. Pharmacokinet.* **1997**, *33*, 344–400.
- (38) Gorelick, D. A. The rate hypothesis and agonist substitution approaches to cocaine abuse treatment. Adv. Pharmacol. 1998, 42 (Catecholamines), 995–997.
- (39) Mehta, N. B. The chemistry of bupropion. J. Clin. Psychiat. 1983, 45, 56–59.
- (40) Chenard, B. L.; Bordner, J.; Butler, T. W.; Chambers, L. K.; Collins, M. A.; De Costa, D. L.; Ducat, M. F.; Dumont, M. L.; Fox, C. B.; Mena, E. E.; Menniti, F. S.; Nielsen, J.; Pagnozzi, M. J.; Richter, K. E. G.; Ronau, R. T.; Shalaby, I. A.; Stemple, J. Z.; White, W. F. (1*S*,2*S*)-1-(4-Hydroxyphenyl)-2-(4-hydroxy-4phenylpiperidino)-1-propanol: A potent new neuroprotectant which blocks *N*-methyl-D-aspartate responses. *J. Med. Chem.* **1995**, *38*, 3138–3145.
- (41) Birch, S. F.; Dean, R. A.; Fidler, F. A.; Lowry, R. A. The preparation of the C₁₀ monocyclic aromatic hydrocarbons. *J. Am. Chem. Soc.* **1949**, *71*, 1362–1369.
- (42) Bailey, W. F.; Punzalan, E. R. Convenient general method for the preparation of primary alkyllithiums by lithium–iodine exchange. *J. Org. Chem.* **1990**, *55*, 5404–5406.
- (43) Musso, D. L.; Mehta, N. B.; Soroko, F. E. Synthesis and evaluation of the anticonvulsant activity of a series of 2-amino-1-phenyl-1-propanols derived from the metabolites of the antidepressant bupropion. *Bioorg. Med. Chem. Lett.* **1997**, *7*, 1–6.
- (44) Foley, K. F.; Cozzi, N. V. Novel aminopropiophenones as potential antidepressants. *Drug Develop. Res.* **2003**, *60*, 252–260.
- (45) Eshleman, A. J.; Carmolli, M.; Cumbay, M.; Martens, C. R.; Neve, K. A.; Janowsky, A. Characteristics of drug interactions with recombinant biogenic amine transporters expressed in the same cell type. J. Pharmacol. Exp. Ther. 1999, 289, 877–885.
- (46) Froimowitz, M.; Wu, K. M.; Moussa, A.; Haidar, R. M.; Jurayj, J.; George, C.; Gardner, E. L. Slow-onset, long-duration 3-(3',4'dichlorophenyl)-1-indanamine monoamine reuptake blockers as potential medications to treat cocaine abuse. J. Med. Chem. 2000, 43, 4981–4992.
- (47) Katz, J. L.; Agoston, G. E.; Alling, K. L.; Kline, R. H.; Forster, M. J.; Woolverton, W. L.; Kopajtic, T. A.; Newman, A. H. Dopamine transporter binding without cocaine-like behavioral effects: synthesis and evaluation of benztropine analogs alone and in combination with cocaine in rodents. *Psychopharmacology* (*Berl.*) 2001, 154, 362–374.
- (48) Gonzalez, L. A.; Gatch, M. B.; Taylor, C. M.; Bell-Horner, C. L.; Forster, M. J.; Dillon, G. H. Carisoprodol-mediated modulation of GABAA receptors: in vitro and in vivo studies. *J. Pharmacol. Exp. Ther.* 2009, 329, 827–837.
- (49) Carroll, F. I.; Fox, B. S.; Kuhar, M. J.; Howard, J. L.; Pollard, G. T.; Schenk, S. Effects of dopamine transporter selective 3phenyltropane analogs on locomotor activity, drug discrimination, and cocaine self-administration after oral administration. *Eur. J. Pharmacol.* **2006**, *553*, 149–156.
- (50) Gatch, M. B.; Taylor, C. M.; Flores, E.; Selvig, M.; Forster, M. J. Effects of monoamine oxidase inhibitors on cocaine discrimination in rats. *Behav. Pharmacol.* 2006, 17, 151–159.

- (51) Rau, K. S.; Birdsall, E.; Hanson, J. E.; Johnson-Davis, K. L.; Carroll, F. I.; Wilkins, D. G.; Gibb, J. W.; Hanson, G. R.; Fleckenstein, A. E. Bupropion increases striatal vesicular monoamine transport. *Neuropharmacology* **2005**, *49*, 820–830.
- (52) Meyer, J. H.; Goulding, V. S.; Wilson, A. A.; Hussey, D.; Christensen, B. K.; Houle, S. Bupropion occupancy of the dopamine transporter is low during clinical treatment. *Psychopharmacology* (*Berl.*) 2002, *163*, 102–105.
- (53) Learned-Coughlin, S. M.; Bergstrom, M.; Savitcheva, I.; Ascher, J.; Schmith, V. D.; Langstrom, B. In vivo activity of bupropion at the human dopamine transporter as measured by positron emission tomography. *Biol. Psychiatry* **2003**, *54*, 800–805.
- (54) Argyelan, M.; Szabo, Z.; Kanyo, B.; Tanacs, A.; Kovacs, Z.; Janka, Z.; Pavics, L. Dopamine transporter availability in medication free and in bupropion treated depression: a ^{99m}Tc-TRODAT-1 SPECT study. J. Affect. Disord. 2005, 89, 115–123.
- (55) Dackis, C. A.; Gold, M. S. New concepts in cocaine addiction: The dopamine depletion hypothesis. *Neurosci. Biobehav. Rev.* 1985, 9, 469–477.
- (56) Volkow, N. D.; Wang, G. J.; Fowler, J. S.; Logan, J.; Gatley, S. J.; Hitzemann, R.; Chen, A. D.; Dewey, S. L.; Pappas, N. Decreased

striatal dopaminergic responsiveness in detoxified cocaine-dependent subjects. *Nature* 1997, *386*, 830–833.
(57) Meltzer, P. C.; Butler, D.; Deschamps, J. R.; Madras, B. K. 1-(4-0.0000)

- (57) Meltzer, P. C.; Butler, D.; Deschamps, J. R.; Madras, B. K. 1-(4-Methylphenyl)-2-pyrrolidin-1-yl-pentan-1-one (Pyrovalerone) analogues: a promising class of monoamine uptake inhibitors. *J. Med. Chem.* 2006, *49*, 1420–1432.
- (58) Broadbent, J.; Michael, E. K.; Riddle, E. E.; Apple, J. B. Involvement of dopamine uptake in the discriminative stimulus effects of cocaine. *Behav. Pharmacol.* **1991**, *2*, 187–197.
 (59) Broadbert J. Correct J. T. J. D. F. Correct and Complexity of the state of the state
- (59) Broadbent, J.; Gaspard, T. M.; Dworkin, S. I. Assessment of the discriminative stimulus effects of cocaine in the rat: lack of interaction with opioids. *Pharmacol., Biochem. Behav.* **1995**, *51*, 379– 385.
- (60) Howell, L. L.; Carroll, F. I.; Votaw, J. R.; Goodman, M. M.; Kimmel, H. L. Effects of combined dopamine and serotonin transporter inhibitors on cocaine self-administration in rhesus monkeys. J. Pharmacol. Exp. Ther. 2007, 320, 757–765.
- (61) de Fiebre, C. M.; de Fiebre, N. E.; Coleman, S. L.; Forster, M. J. Comparison of the actions of gamma-butyrolactone and 1,4butanediol in Swiss-Webster mice. *Pharmacol., Biochem. Behav.* 2004, 77, 705–710.