

# Synthesis of a new class of aminocyclitol analogues with the conduramine D-2 configuration

Latif Kelebekli, Yunus Kara<sup>\*</sup> and Murat Celik<sup>\*</sup>

Full Research Paper		Open Access
Address:	Beilstein Journal of Organic Chemistry 2010, 6, No. 15.	
Atatürk University, Faculty of Science, Department of Chemistry,	doi:10.3762/bjoc.6.15	
TR-25240, Erzurum, Turkey		
	Received: 28 September 2009	
Email:	Accepted: 07 February 2010	
Yunus Kara <sup>*</sup> - yukara@atauni.edu.tr; Murat Celik <sup>*</sup> -	Published: 15 February 2010	
mcelik@atauni.edu.tr		
	Associate Editor: J. N. Johnston	
* Corresponding author		
	© 2010 Kelebekli et al; licensee Beilstein-Institut.	
Keywords:	License and terms: see end of document.	
aminocyclitol: cyclitols: endoperoxide: oxidation: X-ray analysis		

# Abstract

A new class of aminocyclitol derivatives with the bicyclo[4.2.0]octane skeleton was synthesized starting from cyclooctatetraene. Photooxygenation of *trans*-7,8-diacetoxy- and *cis*-7,8-dichlorobicyclo[4.2.0]octa-2,4-diene afforded the bicyclic endoperoxides. Reduction of the latter with thiourea followed by a Pd(0) catalyzed ionization/cyclization reaction gave the corresponding oxazolidinone derivatives. Oxidation of the double bond with KMnO<sub>4</sub> or OsO<sub>4</sub> followed by acetylation gave the acetate derivatives, the exact configuration of which was determined by spectroscopic methods. Hydrolysis of the oxazolidinone rings and removal of the acetate groups furnished the desired aminocyclitols.

## Introduction

Among the myriad of naturally occurring compounds are the aminocyclitol-containing natural products, which represent a large family of sugar derived microbial secondary metabolites and include the clinically active aminoglycoside inhibitors [1-11], many of which are widely used for the treatment of diseases in humans, animals and plants [1-15].

Glycosidase and related enzymes are involved in the biosynthesis of the oligosaccharide chains [1-15]. Carba analogues of oligosaccharides (carbasugars), generated by replacing the endocyclic O-atom in a monosaccharide [1-11], are thought to be better drug candidates than natural sugars, since they are hydrolytically stable. Spurred on by the heightened interest in the design of carbohydrate mimetics, which can be potent inhibitors of glycosidase (1-4) [11-16], we have developed a method for rapid entry to these compounds.



Page 1 of 7 (page number not for citation purposes)

Antibiotics containing an aminocyclitol unit have stimulated the development of synthetic methodologies [16] in the search for analogues with enhanced pharmacological profiles [6]. Balci and Kara [20-22] have synthesized the polyhydroxylated bicyclic molecule **5** having the bicyclo[4.2.0]octane skeleton bis-homoinositol. Furthermore, Trost et al. [23,24] have reported a regio- and stereoselective Pd(0) catalyzed reaction of diols in the presence of *p*-toluenesulfonyl isocyanate for the introduction of the amino alcohol functionality.

We are currently interested in the synthesis of cyclitols and their derivatives [25]. As a part of our program directed towards the synthesis of potential glycosidase inhibitors we used a bicyclo[4.2.0]octane framework for OH, chlorine and NH<sub>2</sub> groups as an intriguing carbohydrate alternative [26-30].

Herein, we report the synthesis of the new aminocyclitol analogues 6 and 7 from cyclooctatetraene.



#### Results and Discussion

Diacetoxydiene 9 was synthesized in 99% yield from cyclooctatetraene (8) by the addition of mercury(II) acetate [31]. Tetraphenylporphyrin sensitized photooxygenation of diacetoxydiene 9 with singlet oxygen gave the expected endoperoxide 10. Reduction of the peroxide bond in 10 was performed with thiourea under very mild conditions to give the *cis*-diol 11 in 99% yield. The introduction of the amino alcohol functionality was achieved by a regio- and stereoselective Pd(0)

catalyzed reaction of diol **11** and TsNCO [32]. Thus treatment of the *cis*-diol **11** in THF with 2 equiv of *p*-toluenesulfonyl isocyanate gave the corresponding bis-carbamate **12** which was subsequently added to a solution of 5 mol % of the catalyst, prepared by stirring a mixture of ligand (triisopropylphosphite) and tris(dibenzylideneacetone)–dipalladium-chloroform complex in THF. Subsequent purification by column chromatography gave oxazolidinone **13** in 48% yield (Scheme 1). The structure of **13** was assigned by <sup>1</sup>H and <sup>13</sup>C NMR and later by X-ray analysis of product **15**.

The observed regio- and stereoselectivity was remarkable since the leaving groups are diastereotopic.

The metal-olefin complexation is a likely source of the stereoselectivity. Mechanistically, only palladium-olefin complexation *anti* to the leaving group will lead to the product **13** [33,34], which is inconsistent with a steric preference for the metal approaching the double bond in **12** from the side of the four-membered ring to form complex **14** (Scheme 2).

Since the double bond is not symmetrically disubstituted, palladium can theoretically form two complexes **15** and **16** after ionization. We assume that the formation of complex **15** is hindered due to the presence of an acetate group in the *endo* position.

*cis*-Dihydroxylation of **13** with KMnO<sub>4</sub> at -15 °C gave a single diol **17**, which was converted into the tetraacetate by treatment with acetic anhydride/CH<sub>3</sub>COONa [35] (Scheme 3). Careful examination of the reaction mixture did not reveal the formation of the other isomer. The stereochemical course of the hydroxylation may be *syn* or *anti* with respect to the oxazolidinone and cyclobutane rings. NMR spectroscopic







studies did not allow the assignment of the exact orientation of the hydroxyl groups. X-ray analysis of **18** (Figure 1) revealed the exact configuration of the compound. This also confirms the configurations of endoperoxide **10**, oxazolidinone **13** and *cis*-hydroxylation product **17**.

The all *cis*-configuration of the four acetate and amino groups in aminocyclitol **18** [25] attached to the six-membered ring resembles the configuration of conduramine D-2 [36-38]. Hydrolysis of the acetate groups with  $H_2SO_4$  proceeded smoothly to deliver aminocyclitol **6** in 84% yield.

For the synthesis of dichloro derivative, we replaced the acetoxy groups in **9** with *cis*- configured chlorine atoms. This provides a route for the synthesis of other haloaminocyclitol derivatives.



*cis*-Dichlorobicyclooctadiene **19** was synthesized from cyclooctatetraene **8** by the addition of chlorine following the literature procedure [39]. Photooxygenation of *cis*-dichlorobicylooctadiene **19** with singlet oxygen gave the expected endoperoxide **20** [19-21] (Scheme 4). Since the dichlorobicyclooctadiene **19** has no plane of symmetry, singlet oxygen approaches the diene unit exclusively from the less crowded side of the molecule in accord with previous reports [20,21]. Reduction of the peroxide bond in **20** with thiourea under very mild conditions gave the *cis*-diol **21** in 95% yield. Diol **21** in THF was treated with 2 equiv of *p*-toluenesulfonyl isocyanate to give the intermediate bis-carbamate **22** which was then treated as described above for **12** with the same Pd(0) catalyst to afford, after chromatography on a silica gel with hexane/ethyl acetate (3:1) as eluant, the oxazolidinone **23** in 61% yield.

The structure of **23** was assigned by <sup>1</sup>H NMR and <sup>13</sup>C NMR spectroscopy. The double bond in **22** is symmetrically disubstituted, and therefore palladium can form only one complex **25** after ionization (Scheme 5).

Hydrolysis of oxazolidinone **23** with  $K_2CO_3$  gave alcohol **26**, which was subsequently converted into acetate **27** by treatment with Ac<sub>2</sub>O/NaOAc [35] (Scheme 6).

*cis*-Hydroxylation of **27** with OsO<sub>4</sub> at 0°C gave the corresponding diol **28**, which was further converted into triacetate **7** with  $Ac_2O/NaOAc$  (Scheme 6). The exact configuration of triacetate **7** was confirmed by differential <sup>1</sup>H NMR NOE measurements







(Figure 2). Irradiation of H<sup>3</sup> (H<sup>3</sup>–CNHTs) at  $\delta$  3.85 caused signal enhancements of H<sup>1</sup>, H<sup>2</sup>, and H<sup>4</sup> at  $\delta$  = 5.45, 5.28 and 5.15, respectively. This result is consistent with a *cis*-relationship between H<sup>1</sup>, H<sup>2</sup>, H<sup>3</sup> and H<sup>4</sup> protons.

We assume that the stereochemical course of the hydroxylation of **27** proceeds through *syn* addition as previously observed in the hydroxylation of **13**, a quite similar structure to **27**.

The all *cis*-configuration of the acetate and amino groups attached to the six-membered ring resembles the configuration of conduramine D-2 [31,40,41]. The cyclic polyhdroxylated amines, also known as aminocyclitols, possess a wide variety of biological activities [42-45]. In conclusion, we have outlined the synthesis of a new family of aminocyclitols analogues **6** and 7 based on the bicyclo[4.2.0]octane frame work, with stereocontrol during the formation of all the stereogenic centres.

#### Experimental

Melting points were determined on a Büchi 539 capillary melting apparatus and are uncorrected. Infrared spectra were obtained from KBr or film on a Mattson 1000 FT-IR spectro-photometer. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on 200 (50) and 400 (100) MHz Varian spectrometer and are reported in  $\delta$  units with SiMe<sub>4</sub> as internal standard. Thin layer chromatography (TLC) was performed on E. Merck Silica Gel 60 F<sub>254</sub> plate (0.2 mm). All column chromatography was performed on silica gel (60 mesh, Merck). Elemental analyses were carried out on a Carlo Erba 1108 model CHNS-O analyzer.



## Supporting Information

Supporting Information File 1

Experimental Section The experimental section describes the synthesis, purification and characterization data of all substances given in this article. [http://www.beilstein-journals.org/bjoc/content/ supplementary/1860-5397-6-15-S1.pdf]

#### Acknowledgements

The authors are indebted to the Department of Chemistry (Atatürk University), state planning organization of Turkey (DPT) for purchasing a NMR, GS-MS and The Scientific and Technical Research Council of Turkey (TUBITAK, TBAG-105T162).

#### References

- 1. Posternak, T. The Cyclitols; Holden Day: San Francisco, CA, 1962.
- 2. Anderson, L. The Cyclitols. In *The Carbohydrates*, 2nd ed.;
- Pigman, W.; Horton, D., Eds.; Academic Press: New York, 1972; Vol. 1A, pp 519–579.
- Arjona, O.; Gómez, A. M.; López, C.; Plumet, J. Chem. Rev. 2007, 107, 1919. doi:10.1021/cr0203701
- 4. Asano, N. Glycobiology 2003, 13, 93R. doi:10.1093/glycob/cwg090
- Heightman, T. D.; Vasella, A. T. Angew. Chem., Int. Ed. 1999, 38, 750. doi:10.1002/(SICI)1521-3773(19990315)38:6<750::AID-ANIE750>3.0. CO;2-6
- Shan, M.; O'Doherty, G. A. Org. Lett. 2008, 10, 3381. doi:10.1021/ol801106r
- Shan, M.; O'Doherty, G. A. Synthesis 2008, 19, 3171. doi:10.1055/s-2008-1067262
- Chakraborty, C.; Vyavahare, V. P.; Dhavale, D. D. *Tetrahedron* 2007, 63, 11984. doi:10.1016/j.tet.2007.09.011
- Frigell, J.; Cumpstey, I. *Tetrahedron Lett.* 2007, 48, 9073. doi:10.1016/j.tetlet.2007.10.138
- Sardinha, J.; Guieu, S.; Deleuze, A.; Fernández-Alonso, M. C.; Rauter, A. P.; Sinaÿ, P.; Marrot, J.; Jiménez-Babero, J.; Sollogoub, M. *Carbohydr. Res.* 2007, 342, 1689. doi:10.1016/j.carres.2007.05.021
- 11. Plumet, J.; Gomez, A. M.; Lopez, J. C. *Mini-Rev. Org. Chem.* **2007**, *4*, 201. doi:10.2174/157019307781369931
- Chiasson, J.-L.; Josse, R. G.; Hunt, J. A.; Palmason, C.; Roger, N. W.; Ross, S. A.; Ryan, E. A.; Tan, M. H.; Wolever, T. M. S. *Ann. Intern. Med.* **1994**, *121*, 928.
- 13. Balci, M. *Pure Appl. Chem.* **1997,** *69*, 97. doi:10.1351/pac199769010097
- 14. Hudlicky, T.; Reed, J. W. Advances in Asymmetric Synthesis; JAI Press: Greenwich, 1995; Vol. 1, p 271.
- 15. Carless, H. A. J. *Tetrahedron: Asymmetry* **1992**, *3*, 795. doi:10.1016/S0957-4166(00)82174-6
- Gultekin, M. S.; Celik, M.; Balci, M. Curr. Org. Chem. 2004, 8, 1159. doi:10.2174/1385272043370069
- 17. El Ashry, E. S. H.; Rashed, N.; Shobier, A. H. S. *Pharmazie* **2000**, *55*, 251.

- 18. El Ashry, E. S. H.; Rashed, N.; Shobier, A. H. S. *Pharmazie* **2000**, *55*, 321.
- 19. El Ashry, E. S. H.; Rashed, N.; Shobier, A. H. S. *Pharmazie* **2000**, *55*, 403.
- Kelebekli, L.; Kara, Y.; Balci, M. Carbohydr. Res. 2005, 340, 1940. doi:10.1016/j.carres.2005.05.021
- Kara, Y.; Balci, M. *Tetrahedron* 2003, *59*, 2063. doi:10.1016/S0040-4020(03)00209-6
- Sahin, E.; Kelebekli, L.; Kara, Y.; Celik, M.; Balci, M. Acta Crystallogr. 2007, E63, o2464. doi:10.1107/S1600536807017011
- 23. Trost, B. M.; Van Vranken, D. L.; Birgel, C. J. Am. Chem. Soc. **1992**, *114*, 9327. doi:10.1021/ja00050a013
- 24. Trost, B. M.; Murphy, D. J. *Organometallics* **1985**, *4*, 1143. doi:10.1021/om00125a039
- 25. Kelebekli, L.; Celik, M.; Kara, Y.; Balci, M. *Tetrahedron Lett.* **2006**, *47*, 7031. doi:10.1016/j.tetlet.2006.07.108
- Curti, C.; Zanardi, F.; Battistini, L.; Sartori, A.; Rassu, G.; Auzzas, L.; Roggio, A.; Pinna, L.; Casiraghi, G. *J. Org. Chem.* **2006**, *71*, 225. doi:10.1021/jo0520137
- Contelles, Y.; de Opaza, E. J. Org. Chem. 2003, 67, 3705. doi:10.1021/jo0111107
- 28. Blériot, Y.; Giroult, A.; Mallet, J.-M.; Rodriguez, E.; Vogel, P.; Sinaÿ, P. *Tetrahedron: Asymmetry* **2002**, *13*, 2553. doi:10.1016/S0957-4166(02)00654-7
- 29. Honda, T.; Kimura, N. Org. Lett. 2002, 4, 4567. doi:10.1021/ol027192i
- Boyer, F. D. J.; Lallemand, J.-Y. Synlett 1992, 969. doi:10.1055/s-1992-21548
- 31. Cope, A. C.; Nelson, N. A.; Smith, D. S. J. Am. Chem. Soc. 1954, 76, 1100. doi:10.1021/ja01633a049
- 32. Andriuzzi, O.; Gravier-Pelletier, C.; Vogel, P.; Le Merrer, Y. *Tetrahedron* **2005**, *61*, 7094. doi:10.1016/j.tet.2005.05.066
- 33. Trost, B. M.; Van Vranken, D. L. J. Am. Chem. Soc. 1993, 115, 444. doi:10.1021/ja00055a013
- 34. Trost, B. M.; Van Vranken, D. L. Chem. Rev. 1996, 96, 395. doi:10.1021/cr9409804
- 35. Celik, M.; Balci, M. ARKIVOC 2007, No. viii, 150.
- Leung-Toung, R.; Liu, Y.; Muchowski, J. M.; Wu, Y.-L. Tetrahedron Lett. 1994, 35, 1639. doi:10.1016/0040-4039(94)88307-6
- Leung-Toung, R. Y.; Liu, J.; Muchowski, J. M.; Wu, Y.-L. J. Org. Chem. 1998, 63, 3235. doi:10.1021/jo971907r
- 38. Supplementary data in the form of CIFs have been deposited with the Cambridge Crystallographic Data Centre (CCDC 299509). Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 IEZ, UK [Fax: +44(0)-1223-336033 or e-mail: deposit@ccdc.cam.ac.uk]. Selected X-ray crystallographic data for **18** ( $C_{24}H_{27}NO_{12}S$ ): Space group: Orthorhombic, *Pbn*21; *a* = 11.6906(4) Å, b = 20.8899(10) Å, c = 22.1919(9) Å, V = 5419 Å^3, Z = 8, F(000) = 2320, D<sub>calc</sub> = 1.36 g cm<sup>-3</sup>, MoK<sub>α</sub> = 0.71073 Å, independent reflections 7705 (Rint = 0.0421),  $\lambda$  radiation observed reflections 7093 (I > 2rl), refinement method; full-matrix least-squares on *F*2, data/restraints/parameters 7093/1/695,  $R_1$  = 0.0557,  $R_w$  = 0.1186, goodness-of-fit on *F*2 = 1.21.
- Reppe, M.; Schlichting, O.; Klager, K.; Topel, T. Justus Liebigs Ann. Chem. **1948**, 560, 1–92. doi:10.1002/jlac.19485600102
- Mehta, G.; Ramesh, S. S. Can. J. Chem. 2005, 83, 581. doi:10.1139/v05-032
- 41. Mehta, G.; Pavolli, K. *Chem. Commun.* **2002,** *23*, 2828. doi:10.1039/b208918a

- 42. Lysek, R.; Vogel, P. *Tetrahedron* **2006**, *62*, 2733. doi:10.1016/j.tet.2005.11.046
- 43. Miyabe, H.; Nishiki, A.; Naito, T. *Chem. Pharm. Bull.* **2003**, *51*, 100. doi:10.1248/cpb.51.100
- 44. McIntosh, M. C.; Weinreb, S. M. J. Org. Chem. 1993, 58, 4823. doi:10.1021/jo00070a016
- Secen, H.; Sütbeyaz, Y.; Balci, M. *Tetrahedron* **1990**, *46*, 3715. doi:10.1016/S0040-4020(01)90509-5

# License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<u>http://creativecommons.org/licenses/by/2.0</u>), which permits unrestricted use, distribution, and reproduction in

any medium, provided the original work is properly cited. The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions:

(http://www.beilstein-journals.org/bjoc)

The definitive version of this article is the electronic one which can be found at: doi:10.3762/bjoc.6.15