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Selective biochlorination of hydroxyquinolines by a flavin-dependent halogenase

Fuchao Xu^a, Amanda Merkley^a, Dayu Yu^b, Jixun Zhan^{a,*}

^a Department of Biological Engineering, Utah State University, 4105 Old Main Hill, Logan, UT 84322, USA ^b Viablife Biotech Co., Ltd, 1 Jing'yi Road, Hangzhou, Zhejiang 311113, China

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ABSTRACT

Rdc2 is a flavin-dependent halogenase from *Pochonia chlamydosporia*. Through the introduction of a His₆-tag to both the N- and C-termini, the isolation yield of Rdc2 from *Escherichia coli* using Ni-NTA affinity chromatography was increased by three-fold. In vitro reaction of Rdc2 and a flavin reductase (Fre) with seven different hydroxyquinolines revealed that 3-hydroxyquinoline (**3**), 5-hydroxyquinoline (**5**), 6-hydroxyquinoline (**6**), and 7-hydroxyquinoline (**7**) can be specifically halogenated. These products were prepared by incubating the corresponding substrates with IPTG-induced *E. coli* BL21(DE3)/Rdc2. They were respectively characterized as 3-hydroxy-4-chloroquinoline (**3**a), 5-hydroxy-6-chloroquinoline (**5**a), 5-chloro-6-hydroxyquinoline (**6**a), and 7-hydroxy-8-chloroquinoline (**7**a) by NMR and MS analyses. This work represents the first enzymatic preparation of chlorohydroxyquinolines and provides a 'green' method to synthesize this group of medicinally important compounds.

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Introduction

Quinolines are common building blocks of numerous bioactive molecules, such as camptothecin (anticancer),¹ chloroquine (antimalarial),² quinine (antimalarial),³ quinidine (antiarrhythmic),⁴ and montelukast (anti-asthma) shown in Figure 1.⁵ Hydroxyquinolines are used as versatile chelating agents and precursors to many pharmaceuticals and pesticides.⁶ For example, the derivatives of 8hydroxyquinoline can be used as multi-target compounds for the treatment of Alzheimer's disease.⁷ Hydroxyquinolines can be chemically synthesized from quinoline (1, Fig. 2).

Halogenated compounds represent about 20% of pharmaceuticals on the market and a quarter of those in the development pipeline. Chlorohydroxyquinolines are known to possess important medicinal properties.⁸ For example, it has been reported that 5chloro-8-hydroxyquinoline has significant antimycoplasma activity.⁹ However, preparation of halogenated molecules relies on chemical methods. Chemical halogenation typically requires harsh reaction conditions including high temperatures and toxic reagents (such as chlorine gas, hydrochloric acid, and hydrogen peroxide). This is often accompanied by poor selectivity, which results in unspecific halogenation and undesired byproducts, creating difficulties in the purification process.¹⁰ For example, a previous research showed that chlorination of quinoline at 160–190 °C yielded a mixture of five chlorinated products, including 3,4dichloroquinoline, 3,4,6-trichloroquinoline, 3,4,8-trichloroquinoline, 3,4,6,8-tetrachloroquinoline, and 3,4,6,7,8-pentachloroquinoline.¹¹

Biological halogenation is considered as a better approach for its high selectivity. Flavin-dependent halogenases are major players in the introduction of halogen in the biosynthesis of halogenated natural products such as radicicol.¹² Most reported flavindependent halogenases are prokaryotic tryptophan halogenases. The use of these enzymes in the preparation of halogenated molecules was limited due to their strict substrate specificity. A recent work reported that a tryptophan 6-halogenase (Th-Fal) from a thermophilic halotolerant Streptomyces violaceusniger can halogenate several tryptophan derivatives and aromatic substrates, including methyltryptophan, 5-hydroxytryptophan, kynurenine, anthranilic acid, and anthranilamide.¹³ However, there is still a lack of a highly flexible halogenase that can be used for a broad range of substrates. Rdc2 is the first flavin-dependent halogenase identified from fungi. This enzyme is involved in the biosynthesis of radicicol, specifically chlorinating monocillins (a group of resorcylic acids) at C-6. Rdc2 was found to have flexible substrate specificity and can work on similar macrolactones and other structurally different compounds such as oxytetracycline and curcumin.¹⁴ A homology model of Rdc2 was built, which indicates that the space at the active site of this enzyme is large enough to accommodate different substrates. Two hydroxyisoquinolines were converted by Rdc2 to two new halogenated derivatives.¹⁵ However, specific enzymatic preparation of chlorohydroxyquinolines has not been







^{*} Corresponding author. Tel.: +1 435 797 8774; fax: +1 435 797 2769. *E-mail address:* jixun.zhan@usu.edu (J. Zhan).



Figure 1. Quinoline-containing bioactive molecules. The quinoline moiety is shaded.

previously reported. In this work, we systematically tested the halogenation capability of Rdc2 using seven monohydroxylated quinolines (**2–8**, Fig. 2), including 2-hydroxyquinoline (**2**), 3-hydroxyquinoline (**3**), 4-hydroxyquinoline (**4**), 5-hydroxyquinoline (**5**), 6-hydroxyquinoline (**6**), 7-hydroxyquinoline (**7**), and 8-hydroxyquinoline (**8**).

Results and discussion

The reaction of Rdc2 requires a partner enzyme, flavin reductase (Fre), which reduces FAD to yield FADH₂. We previously constructed pZJ54 (pET28a-*rdc2*) and pZJ62 (pET28a-*fre*) to purify N-His₆-tagged enzymes for in vitro reactions.¹⁴ In order to get a higher Rdc2 yield, we designed a construct with a His₆-tag at both N- and C-terminal ends. To this end, the intron-free *rdc2* gene was amplified from the previously constructed pZJ54 by PCR with Phusion High-Fidelity DNA Polymerase (New England Biolabs) using a pair of specific primers, 5'-aa<u>CATATG</u>TCGGTACCCAAGTCTTG-3' (the *Ndel* site is underlined) and 5'-aa<u>AAGCTT</u>AACTTTGTTGAGGC-CAA-3' (the *Hin*dIII site is underlined). The amplified gene was ligated into the cloning vector pJET1.2 to yield pFC55. The Rdc2 gene was then excised from pFC55 with *Ndel* and *Hin*dIII and ligated into pET28a between the same sites to generate pFC56.

The plasmid pFC56 was introduced into *E. coli* BL21(DE3) for protein expression. The resulting strain was grown at 37 °C in Luria–Bertani (LB) medium supplemented with 50 μ g mL⁻¹ kanamycin. Protein expression was induced at OD₆₀₀ 0.4–0.6 with 200 μ M isopropyl-1-thio- β -p-galactopyranoside (IPTG). The double







Figure 3. HPLC analysis (at 250 nm) of the in vitro reactions of Rdc2 with hydroxyquinolines (at 30 °C for 2 h) in the presence of Fre, NADH, FAD, O_2 and NaCl. (a) Inactivated Rdc2+2; (b) Rdc2+2; (c) inactivated Rdc2+3; (d) Rdc2+3; (e) inactivated Rdc2+4; (f) Rdc2+4; (g) inactivated Rdc2+5; (h) Rdc2+5; (i) inactivated Rdc2+6; (j) Rdc2+6; (k) inactivated Rdc2+7; (l) Rdc2+7; (m) inactivated Rdc2+8; (n) Rdc2+8.

His₆-tagged Rdc2 was purified by Ni-NTA column at a yield of 18.5 mg L⁻¹, which represents a three-fold increase compared to the N-terminal His₆-tagged Rdc2.¹⁴ The plasmid pZJ62 was introduced into *E. coli* BL21-CodonPlus (DE3)-RIL for the purification of Fre, as previously reported.¹⁴ Both enzymes were stored in 50% glycerol at -20 °C.

We then conducted in vitro enzymatic assays with Rdc2, Fre, NADH, FAD, NaCl, O₂, and seven different hydroxyquinolines 2-8 as substrates. The reactions were incubated at 30 °C for 2 h and the negative controls were maintained under the same conditions with inactivated Rdc2. As shown in Figure 3, HPLC analysis revealed that **3**, **5**, **6**, and **7** can be converted to less polar products 3a, 5a, 6a, and 7a by Rdc2, respectively. By contrast, no products were synthesized from 2, 4, and 8. A comparison of the UV spectra of 3a, 5a, 6a, and 7a with the corresponding substrates showed a bathochromic shift (Fig. S1), indicating that these products are chlorinated derivatives of 3, 5, 6, and 7. ESI-MS of all these products showed the $[M+H]^+$ quasimolecular peaks at m/z 182 and 180, with a ratio of 3:1 (Fig. S2), representing a characteristic isotope pattern of monochlorinated compounds. All the hydroxyquinolines are isomers with a molecular weight of 145, while the molecular weight of **3a**, **5a**, **6a**, and **7a** is 179, which is 34 mass units larger than the substrates, indicating that they are indeed the chlorinated products.

To obtain sufficient amounts of chlorinated products for NMR spectroscopic analysis, we used a whole-cell biotransformation approach. A total of 15 mg of **3**, **5**, **6**, and **7** were separately fed into 500 mL of IPTG-induced fermentation broth of *E. coli* BL21(DE3)/ pFC56, from which 4 mg of **3a**, 3.2 mg of **5a**, 4.5 mg of **6a**, and 4.3 mg of **7a** were isolated in pure form, respectively.

The purified products were dissolved in methanol- d_4 and the NMR spectra were acquired on a Bruker AvanceIII HD Ascend-500 NMR instrument. To determine the chlorination position for

Table 1		
¹ H (500 MHz) N	MR data for 3a, 5a ,	6a , and 7a (CD ₃ OD, δ in ppm, J in Hz)
	_	

Position	3a	5a	6a	7a
2	8.62 (1H, s)	8.83 (1H, d, J = 3.0)	8.80 (1H, d, <i>J</i> = 4.4)	8.84 (1H, d, J = 4.2)
3	_	7.56 (1H, dd, J = 8.5, 4.0)	7.59 (1H, dd, <i>J</i> = 8.4, 4.1)	7.44 (1H, dd, J = 8.1, 4.4)
4	_	8.72 (1H, d, J = 8.5)	8.49 (1H, d, J = 8.4)	8.31 (1H, d, J = 8.1)
5	8.17 (1H, d, J = 7.9)	_	_	7.79 (1H, d, J = 8.9)
6	7.63-7.70 (1H, overlap)	_	_	7.34 (1H, d, J = 8.9)
7	7.63-7.70 (1H, overlap)	7.55 (1H, d, 9.0)	7.58 (1H, d, J = 9.0)	_
8	7.99 (1H, d, <i>J</i> = 7.9)	7.68 (1H, d, 9.0)	7.94 (1H, d, <i>J</i> = 9.0)	_

Table 2

¹³C (125 MHz) NMR data for **3a**, **5a**, **6a**, and **7a** (CD₃OD, δ in ppm)

Position	3a	5a	6a	7a
2	142.5 (CH)	149.9 (CH)	147.9 (CH)	150.2 (CH)
3	148.2 (C)	120.6 (CH)	122.1 (CH)	119.2 (CH)
4	102.9 (C)	131.7 (CH)	130.4 (CH)	137.3 (CH)
4a	107.3 (C)	126.5 (C)	126.9 (C)	124.1 (C)
5	122.5 (CH)	156.4 (C)	112.0 (C)	127.3 (CH)
6	127.9 (CH)	94.8 (C)	151.2 (C)	118.8 (CH)
7	126.9 (CH)	119.8 (CH)	121.3 (CH)	155.3 (C)
8	128.2 (CH)	130.7 (CH)	129.6 (CH)	113.8 (C)
8a	102.2 (C)	143.8 (C)	144.0 (C)	145.2 (C)



Figure 4. Selected ¹H–¹H COSY and HMBC correlations of 3a, 5a, 6a and 7a.

each product, ¹H (Figs. S3–S6), DEPTQ-135 (Figs. S7–S10), HSQC, HMBC, and ¹H–¹H COSY spectra were recorded. The proton and carbon signals were assigned and are listed in Tables 1 and 2.

The ¹H NMR spectrum of **3a** showed five proton signals, including two doublet proton signals (δ 7.99 and 8.17), an overlapped signal (2H, δ 7.63–7.70), and a singlet (δ 8.62). This is one proton less than the substrate, indicating that one of the CH of 3-hydroxyquinoline was chlorinated. The ¹H–¹H COSY spectra showed a spin system that consists of the two doublet signals and two overlapped proton signals, which suggested that the halogenation did not occur on the benzene ring and thus one of the protons on the pyridine ring was substituted by a chlorine atom. This was supported by the presence of the singlet signal, which was assigned to H-2 according to the HMBC correlations (Fig. 4) of this proton to C-4 (δ 102.9) and C-8a (δ 102.2). Thus, it can be deduced that the halogenation occurred at C-4 position, which is ortho to the hydroxyl group (Fig. 4). This was further supported by the HMBC correlations of H-5 (δ 8.17) to C-4. Thus, **3a** was identified as 3-hydroxy-4-chloroquinoline.

The ¹H and DEPTQ-135 NMR of **5a** also showed five CH signals, confirming that one proton of the substrate **5** was chlorinated. The signal of H-2 can be easily located in the low field at δ 8.83. Two spin systems were observed in the ¹H-¹H COSY spectrum (Fig. 4). The CH–CH–CH system belongs to the pyridine ring, which includes H-2 (δ 8.83, d), H-3 (δ 7.56, dd), and H-4 (δ 8.72, d). The CH–CH system consists of two protons from the benzene ring,

including the proton signals at δ 7.55 (d) and δ 7.68 (d). Thus, we concluded that that one of the protons of the benzene ring was substituted by a chlorine atom. The large coupling constant of 9.0 Hz indicated that the two CH groups are *ortho* to each other. Accordingly, the possibility of C-7 halogenation can be ruled out. Based on the HMBC correlations of both proton signals at δ 7.55 and 7.68 to C-8a at δ 143.8 (Fig. 4), the chlorination site was determined to be at C-6. Thus, this product was identified as 5-hydroxy-6-chloroquinoline.

The ¹H and DEPTQ-135 NMR spectra of **6a** are similar to those of **5a**. The signal of H-2 was located in the low field at δ 8.80. With this signal, the spin system of H-2/H-3/H-4 was identified by the ¹H-¹H COSY correlations (Fig. 4). The other spin system consists of two doublet signals from the benzene ring that belongs to H-7 and H-8, as suggested by the large coupling constant of 9.0 Hz. Accordingly, it can be deduced that C-5 of 6 was chlorinated. This was confirmed by the HMBC correlations of H-4 (δ 8.49, d) and H-7 (δ 7.58, d) to C-5 at δ 112.0. Therefore, **6a** was characterized as 5-chloro-6-hydroxyquinoline.

The ¹H NMR spectrum of **7a** showed five proton signals. The presence of a CH–CH–CH spin system in the ¹H–¹H COSY spectrum (Fig. 4) indicated that none of the protons on the pyridine ring was chlorinated. The ¹H–¹H COSY correlations of H-5 to H-6, which are doublet signals at δ 7.79 and 7.34 with a coupling constant of 8.9 Hz, respectively, indicated that Rdc2 introduced a chlorine atom to C-8. The structure was thus characterized as 7-hydroxy-8-chloroquinoline, which was supported by the HMBC correlations shown in Figure 4. **7a** is a new compound.

Structural identification of the products **3a**, **5a**, **6a**, and **7a** suggested that among the seven tested hydroxyquinolines, four were taken by Rdc2 as substrates to form the corresponding monochlorinated products. The chlorination reaction is highly specific as the position where the chlorine atom was introduced is *ortho* to the hydroxyl group of the four substrates. This is consistent with the results for isoquinolines in our previous studies. A number of pharmaceutically important molecules contain a hydroxylated or chlorinated quinoline moiety, such as quinine, quinidine, chloroquine, and montelukast shown in Figure 1. Thus, the hydroxyl group and chlorine atom attached to quinoline represent important functional groups. In this work, we generated four quinoline derivatives that contain both functional groups at different positions.

In summary, the expression and purification Rdc2 in E. coli was improved in this work. In vitro reactions of Rdc2 with seven hydroxyquinolines revealed that this halogenase can halogenate four of these substrates. The corresponding products were then prepared through fermentation of the Rdc2-expressing E. coli strain with supplemented substrates. The four products were isolated and structurally characterized based on the UV, MS, and NMR spectra. This work represents the first enzymatic preparation of chlorohydroxyquinolines. The results further expand the reservoir of the acceptable substrates of Rdc2, which makes it a useful biological halogenating agent in the preparation of structurally diverse halogenated molecules.

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Supplementary data

Supplementary data (experimental procedures, UV and ESI-MS spectra) associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetlet.2016.10.044.

References and notes

- 1. Efferth, T.; Fu, Y.-J.; Zu, Y.-G.; Schwarz, G.; Badireenath, K. V. S.; Wink, M. Curr. Med. Chem. 2016, 14, 2024-2032.
- 2 Foley, M.; Tilley, L. Pharmacol. Ther. 1998, 79, 55-87.
- Achan, J.; Talisuna, A. O.; Erhart, A.; Yeka, A.; Tibenderana, J. K.; Baliraine, F. N.; 3 Rosenthal, P. J.; D'Alessandro, U. Malar. J. 2011, 10, 144.
- 4. Shen, X. T.; Antzelevitch, C. *Circulation* **1986**, 73, 1342–1353.
- Leff, J. A.; Busse, W. W.; Pearlman, D.; Bronsky, E. A.; Kemp, J.; Hendeles, L.; 5 Dockhorn, R.; Kundu, S.; Zhang, J.; Seidenberg, B. C.; Reiss, T. F. N. Engl. J. Med. 1998. 339. 147-152.
- 6. Prachayasittikul, V.; Prachayasittikul, S.; Ruchirawat, S.; Virapong, P. Drug Des. Dev. Ther. 2013, 7, 1157-1178.
- 7. Prati, F.; Bergamini, C.; Fato, R.; Soukup, O.; Korabecny, J.; Andrisano, V.; Bartolini, M.; Bolognesi, M. L. ChemMedChem 2016, 11, 1-13.
- Heseltine, W. W.; Campbell, P. J. J. Trop. Med. Hyg. **1960**, 63, 163–165. 8
- Cosgrove, R. F.; Baines, S. Antimicrob. Agents Chemother. **1978**, 13, 540–541. Samant, B. S.; Saraf, Y. P.; Bhagwat, S. S. J. Colloid Interface Sci. **2006**, 302, 207– Q
- 10. 213
- 11. van Pee, K. H.; Patallo, E. P. Appl. Microbiol. Biotechnol. 2006, 70, 631-641.
- 12. Heemstra, J. R., Jr.; Walsh, C. T. J. Am. Chem. Soc. 2008, 130, 14024–14025.
- Menon, B. R. K.; Latham, J.; Dunstan, M. S.; Brandenburger, E.; Klemstein, U.; 13 Leys, D.; Karthikeyan, C.; Greaney, M. F.; Shepherd, S. A.; Micklefield, J. Org. Biomol. Chem. 2016, 14, 9354-9361.
- Zeng, J.; Zhan, J. ChemBioChem 2010, 11, 2119-2123. 14
- Zeng, J.; Lytle, A. K.; Gage, D.; Johnson, S. J.; Zhan, J. Bioorg. Med. Chem. Lett. 15. **2013**, 23, 1001–1003.