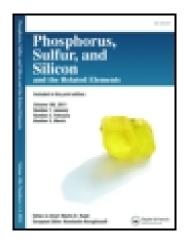
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Synthesis of Four- and Five-Membered Heterocycles Derived from an Iminophosphorane

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SYNTHESIS OF FOUR- AND FIVE-MEMBERED HETEROCYCLES DERIVED FROM AN IMINOPHOSPHORANE

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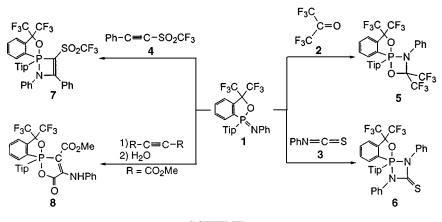
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Reactions of an iminophosphorane bearing the Martin ligand with a ketone, an isothiocyanate, and an alkyne gave the corresponding cycloadducts, $1,3,2\lambda^5$ -oxazaphosphetidine, $1,3,2\lambda^5$ -diazaphosphetidine-4-thione, and $1,2\lambda^5$ -azaphosphetine, respectively, while that with dimethyl acetylenedicarboxylate (DMAD) and successive hydrolysis afforded the $1,2\lambda^5$ -oxaphosphole-(2H)-one. Thermal reactions of the cycloadducts were also studied.

Keywords: Cycloaddition; heterocycles; iminophosphorane; phosphorane; thermolysis; x-ray analysis

There has been much interest in the chemistry of four-membered heterocycles containing both a pentacoordinate phosphorus atom and other heteroatom(s) in view of their unique structures and reactivities.¹ In the course of our study on such heteracyclobutanes, we have achieved the syntheses of $1,2\lambda^5$ -oxaphosphetanes² and $1,2\lambda^5$ -azaphosphetidines,³ that is, the intermediates of Wittig reactions, by taking advantage of the Martin ligand. On the other hand, an iminophosphorane, which is a nitrogen analog of the phosphorus ylide, is expected to construct a variety of four-membered ring compounds by the cycloaddition with multiple bond compounds. We report here the syntheses and thermolyses of novel heterocycles bearing a pentacoordinate phosphorus atom derived from an iminophosphorane bearing the Martin ligand and a 2,4,6-triisopropylphenyl (denoted as Tip hereafter) group.

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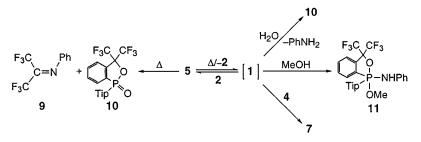


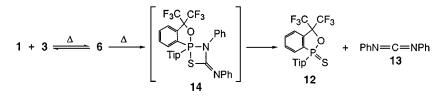
SCHEME 1

RESULTS AND DISCUSSION

Reactions of iminophosphorane 1 with hexafluoroacetone 2, phenyl isothiocyanate 3, and phenylethynyl trifluoromethyl sulfone 4 gave the corresponding cycloadducts, $1,3,2\lambda^5$ -oxazaphosphetidine $5,^4$ $1,3,2\lambda^5$ -diazaphosphetidine-4-thione 6, and $1,2\lambda^5$ -azaphosphetine 7, respectively. Interestingly, successive treatment of 1 with dimethyl acetylenedicarboxylate (DMAD) and water afforded not the corresponding 1,2-azaphosphetine, but the $1,2\lambda^5$ -oxaphosphol-5(2*H*)-one 8^5 (Scheme 1). The structures of these products were established by x-ray crystallographic analyses. It is worth mentioning that the oxygen analog of 1,2-azaphosphetine, 1,2-oxaphosphete, is only an intermediate in the reactions of phosphine oxides and DMAD.⁶

The thermolysis of the four-membered ring compound **5** revealed the formation of the Wittig-type products, the corresponding imine **9** and cyclic phosphinate **10**. The existence of the equilibrium with the starting materials on the thermal conditions was found by the thermolyses





SCHEME 3

of 5 in the presence of water, methanol, and 4, which gave 10, methoxyphosphorane 11, and 7, respectively (Scheme 2).

The thermolysis of $1,3,2\lambda^5$ -diazaphosphetidine-4-thione **6**, afforded not only the starting materials, **1** and **3**, but also cyclic thiophosphinate **12** and diphenylcarbodiimide **13**. The formation of **12** and **13** strongly suggests the generation of the corresponding $1,3,2\lambda^5$ thiazaphosphetidine **14** as an intermediate (Scheme 3).

 $1,2\lambda^5$ -Azaphosphetine **7** is thermally quite stable, in contrast to **5** and **6**. It did not decompose even upon heating at 180°C. It is inactive toward the Wittig reactions with benzaldehyde and phenyl isocyanate.

In summary, we have revealed that $1,3,2\lambda^5$ -oxazaphosphetidine **5** and $1,3,2\lambda^5$ -diazaphosphetidine-4-thione **6**, which were obtained from an iminophosphorane **1**, have two reactivities including the regeneration of **1**. Meanwhile, $1,2\lambda^5$ -azaphosphetine **7** is quite thermally stable.

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